the Energy to Lead

Nano-engineered catalyst for the utilization of CO₂ in dry reforming to produce syngas DOE Contract No. DE-FE0029760

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Project overview

- **Performance period**: July 1, 2017 June 30, 2020
- **Funding**: \$799,807 DOE (\$200,000 co-funding)
- **Objectives**: Develop nano-engineered catalyst supported on highsurface-area ceramic hollow fibers for the utilization of CO_2 in dry reforming of methane ($CO_2 + CH_4 \rightarrow 2 H_2 + 2 CO$) to produce syngas

• <u>Team:</u>

Member	Roles
gti.	 Project management and planning Quality control, reactor design and testing Techno-economic analysis (TEA) and life cycle analysis (LCA)
MISSOURI	 Catalyst development and testing

Introduction to GTI and Missouri S&T

OFFICE

- Idea Market Analysis Technology Analysis Product Development Lab and Field Testing Demonstration Commercialization
 - Not-for-profit research company, providing energy and natural gas solutions to government and industry since 1941





- **Co-educational research university** located in Rolla, Missouri
- Prof. Liang Group: expertise in atomic layer deposition thin film coatings, catalyst synthesis and testing



Background of dry reforming of methane using captured CO₂

- CH₄ + CO₂ → 2H₂ + 2CO with H₂/CO ratio <1 due to the reverse watergas shift reaction (CO₂ + H₂ ≈ CO + H₂O)
 - Different from methane steam reforming $(CH_4 + H_2O \rightarrow CO + 3 H_2)$ where H_2/CO ratio >3 due to water-gas shift reaction $(CO + H_2O \rightleftharpoons CO_2 + H_2)$
- Syngas: feedstock for fuels and chemicals production
- H₂/CO ratio determines the resulting products
 - Dry reforming syngas (H₂/CO ratio = 0.7 1) can be used for producing high yield C₅₊ hydrocarbons
 - Higher H₂/CO ratio can be achieved by blending with products from steam reforming
- Typical catalysts:
 - **<u>Precious metals</u>** (Pt, Rh, Ru): expensive
 - Low-cost Ni: issue of sintering of the Ni particles

Nano-engineered Ni catalyst prepared by atomic layer deposition (ALD) may resolve sintering issue

- ALD is a commercial process in semiconductor industry
- Advantages over traditional catalysts prepared by incipient wetness (IW)
 120
 - Higher activity
 - Better stability
 - Ni/γ-Al₂O₃particle
 - CO₂ and CH₄
 cylinder gases
 used in testing



Integration of the technology with coalfired power plants



Two conceptual process designs: 1) packed bed reactor, and 2) tube-shell transport reactor

 Packed bed reactor: the reactor is filled with nano-engineered catalyst supported on 1-2 cm long hollow fibers



Nano-engineered Ni catalyst prepared by ALD



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C Catalysts are calcined in air at 550 °C

X-ray photoelectron spectroscopy analysis of α -Al₂O₃ nanoparticles supported Ni catalysts



TEM image of α -Al₂O₃ nanoparticle-supported Ni catalysts

- Particle size: 2-6 nm, average 3.1 nm
 - Particles prepared by traditional methods are ~10-20 nm



Novel α -Al₂O₃ hollow fiber with high packing density is being used as catalyst substrate in current project



Commercial substrates

Catalyst Geometry	SA/V (m²/m³)	
1-hole	1,151	
1-hole-6-grooves	1,733	
4-hole	1,703	
10-hole	2,013	
Monolith	1,300	
4-channel ceramic hollow fibers	3,000	



Novel α -Al₂O₃ hollow fibers

- Four channels, 35 cm long
- OD of 3.2 mm and a channel inner diameter of 1.1 mm
- Geometric surface area to volume as high as 3,000 m²/m³

Dry reforming performance of the α -Al₂O₃ hollow fiber supported Ni catalysts (Ni/ α -Al₂O₃-HF)



- Higher activity due to highly dispersed nanoparticles: ~3.6 nm Ni particles compared to ~10-20 nm particles prepared by traditional method
- Better stability due to strong bonding between nanoparticles and substrates since the particles are chemically bonded to the substrate during ALD

Al₂O₃ ALD film increases Ni-support interaction, and thus improves catalytic performance





Dry reforming performance of the AI_2O_3 promoted Ni/α - AI_2O_3 -HF catalysts

800 °C, 15 psia, CO₂ and CH₄ cylinder gases used in testing

Catalyst	Conversion (%)	H ₂ /CO ratio	Methane reforming rate (Lh ⁻¹ g _{Ni} ⁻¹)
Ni/α-Al ₂ O ₃ -HF	88	0.85	2,500
$2AI_2O_3$ -Ni/ α -AI $_2O_3$ -HF	91	0.85	2,600
$5AI_2O_3$ -Ni/ α -AI $_2O_3$ -HF	90	0.84	2,600
$10AI_2O_3$ -Ni/ α -AI_2O_3-HF	88	0.85	2,500



CeO₂ promoted Ni/α-Al₂O₃-HF catalysts

- CeO₂ can potentially increase Ni-support interaction, and provide highly mobile oxygen to inhibit coking of the catalyst
- We improved the catalyst performance by CeO₂ coating prepared by impregnation method



ALD reactor modified for depositing catalysts onto 20-cm-long hollow fibers





Ni nanoparticles successfully deposited on 20-cm-long hollow fibers by ALD



Before Ni ALD

After Ni ALD



Dry reforming performance of the Ni ALD coated 20-cm-long hollow fibers

20-cm-long fibers were broken up into 1-cm-long fibers and tested in a packed bed reactor (CO_2 and CH_4 cylinder gases used in testing)



Tube-shell transport reactor designed, Ni coated 20-cm-long hollow fibers to be tested



Future plans

In this project



After the current project

Test the technology at a larger scale with captured CO₂





- Novel α-Al₂O₃ hollow fiber increases surface area, and enables tube-shell transport reactor configuration.
- ALD nano-engineered catalyst improves activity and stability for utilization of CO₂ in dry reforming of methane to produce syngas (compared to catalysts prepared by conventional incipient wetness method).
- Coating of Al₂O₃ or CeO₂ on Ni/α-Al₂O₃-HF catalysts further improves dry reforming performance.
- Uniform Ni was successfully coated on 20-cm-long hollow fibers using a modified ALD reactor.



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