



DOE Contract No. DE-FE0029787

# High Energy Systems for Transforming CO<sub>2</sub> to Valuable Products

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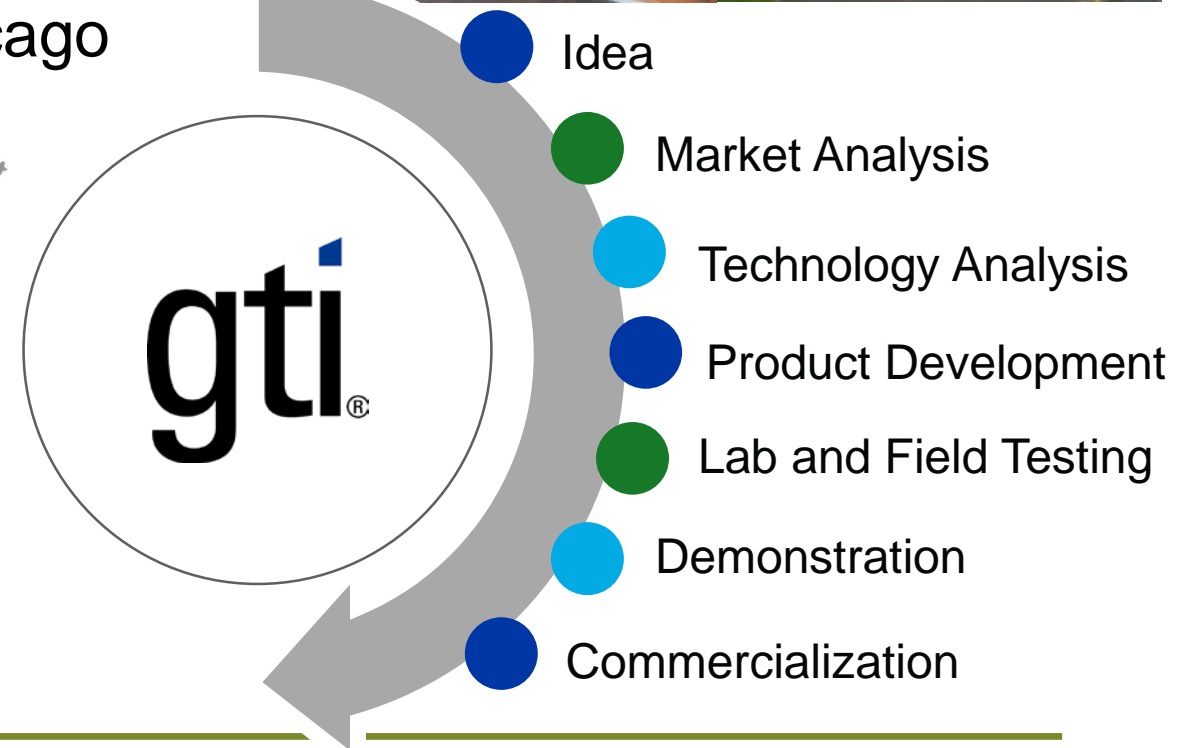
NETL CO<sub>2</sub> Capture Technology Project Review Meeting, Pittsburgh, PA, August 26-30, 2019

# Outline

- **Project Overview**
- **Technology Background**
- **Technical Approach Discussion**
- **Progress and Current Status**
- **Plans for Future**

# Introduction to GTI

- Research organization, providing energy and environmental solutions to the government and industry since 1941
- Facilities: 18 acre campus near Chicago



# Project Overview

# High Energy Systems for Transforming CO<sub>2</sub> to Valuable Products

- **Sponsor**



DE-FE0029787

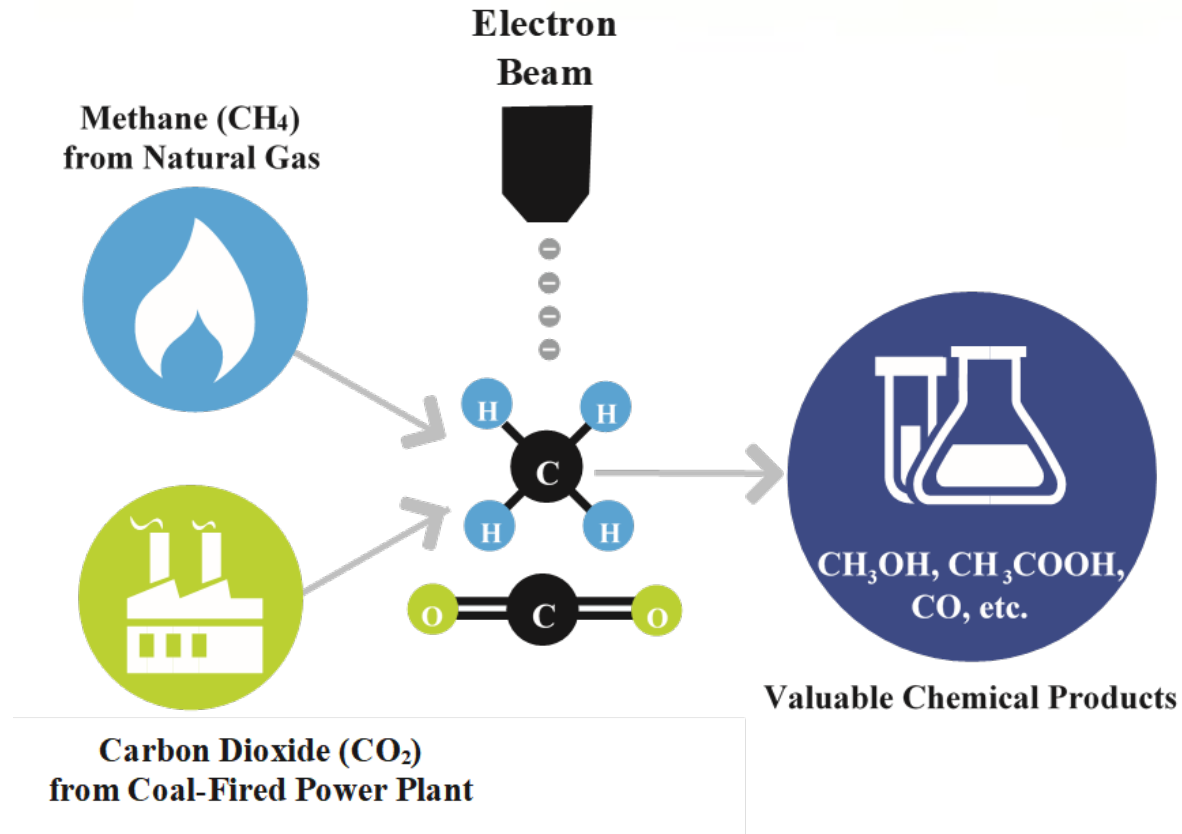
- **Funding**: Federal: \$799,997, Cost-share: \$206,000, Total: \$1,005,997

- **Duration**: 39 months

BP1: 5/1/2017 – 10/31/2019    BP2: 11/1/2019 – 7/31/2020

- **Objective**: Develop a direct electron beam synthesis (DEBS) process to produce valuable chemicals such as acetic acid, methanol, and carbon monoxide, using carbon dioxide captured from a coal-fired power plant and natural gas.

# Project Objectives



- Develop the Direct E-Beam Synthesis (DEBS) process




Use high-energy electron beams from an accelerator to break chemical bonds

- Produce valuable chemicals, such as acetic acid, methanol, and carbon monoxide, at relatively low severity (pressure near one atmosphere and temperatures <150°C)

Utilize near-pure CO<sub>2</sub> captured from a pulverized coal (PC)-fired power plant and methane, imported as natural gas

# High Energy Systems for Transforming CO2 to Valuable Products

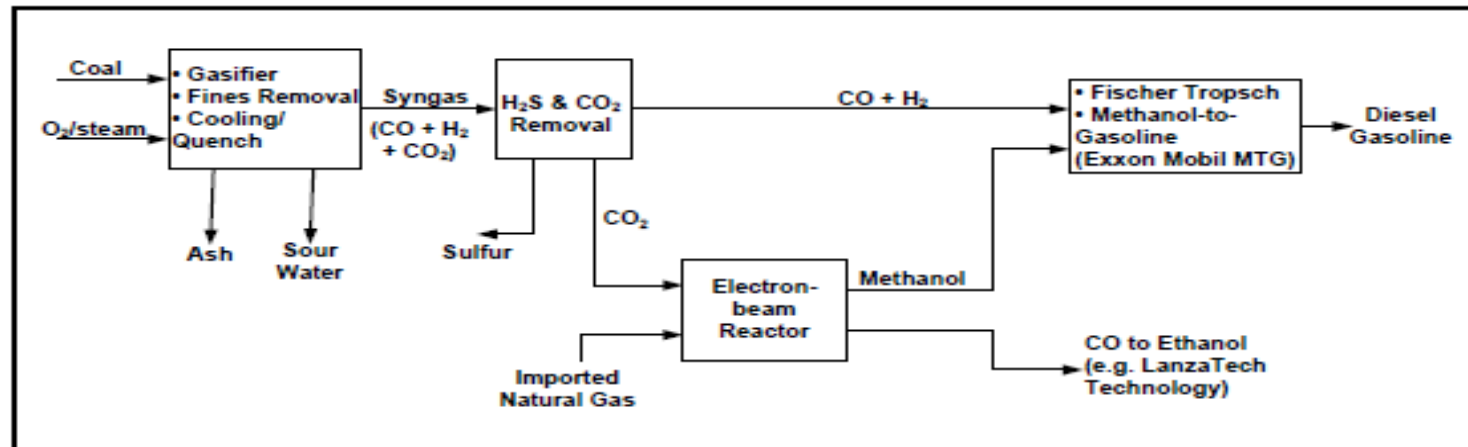
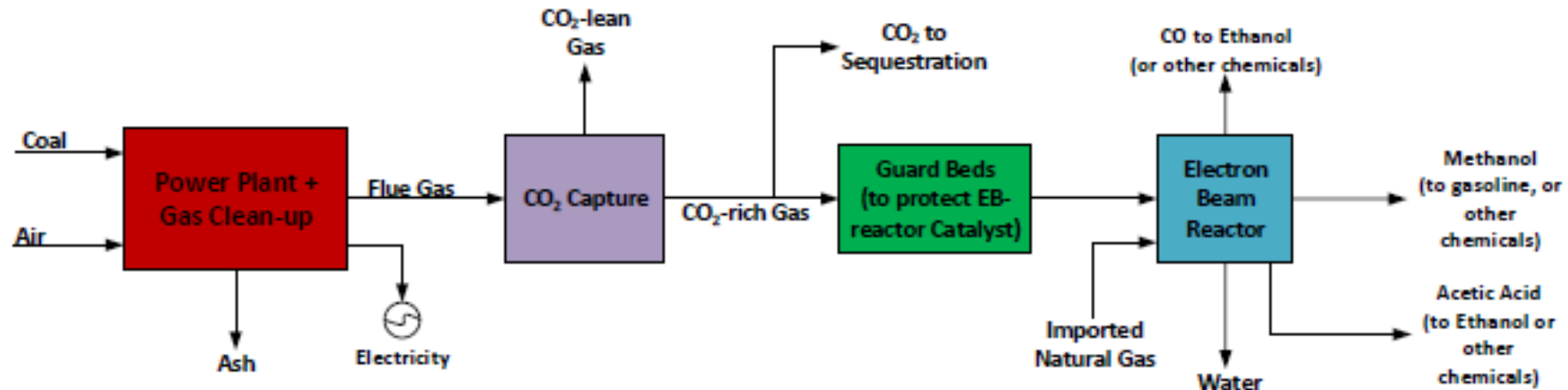
## ■ Team:

Member	Role
	<ul style="list-style-type: none"><li>• Overall project integration and management</li><li>• Design and construct E-Beam reactor and testing unit</li><li>• Conceptual design for coal-fired power plants with DEBS</li></ul>
	<ul style="list-style-type: none"><li>• Provide guidance in E-Beam reactor design and E-Beam accelerator for testing</li><li>• Provide commercial size electron accelerator design and costing</li></ul>
	<ul style="list-style-type: none"><li>• Develop kinetic model for the E-Beam reactor</li></ul>

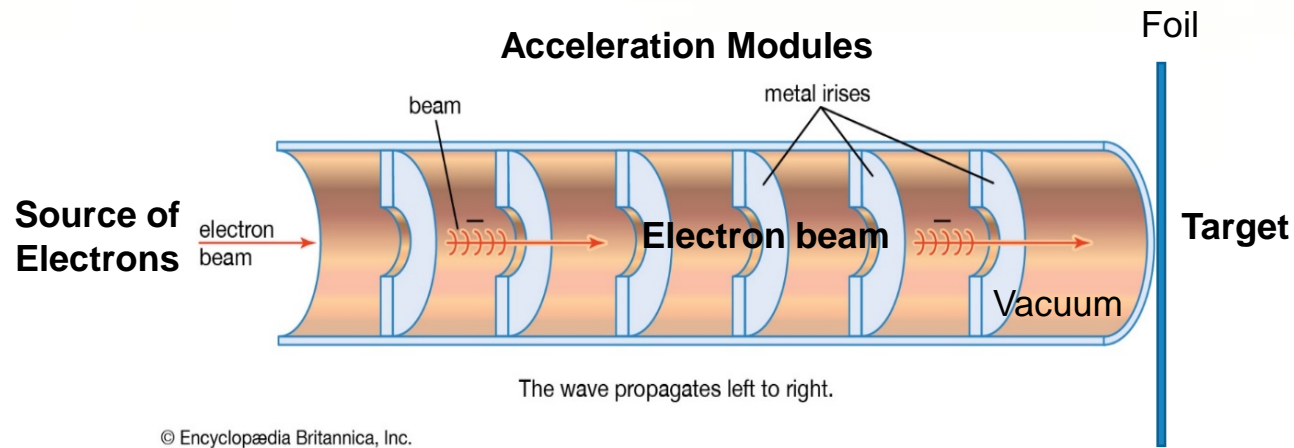
# Technology Background



# DEBS Process for Post- and Pre-combustion



# Electron Beam Fundamentals



## 200keV & 20mA E-Beam:

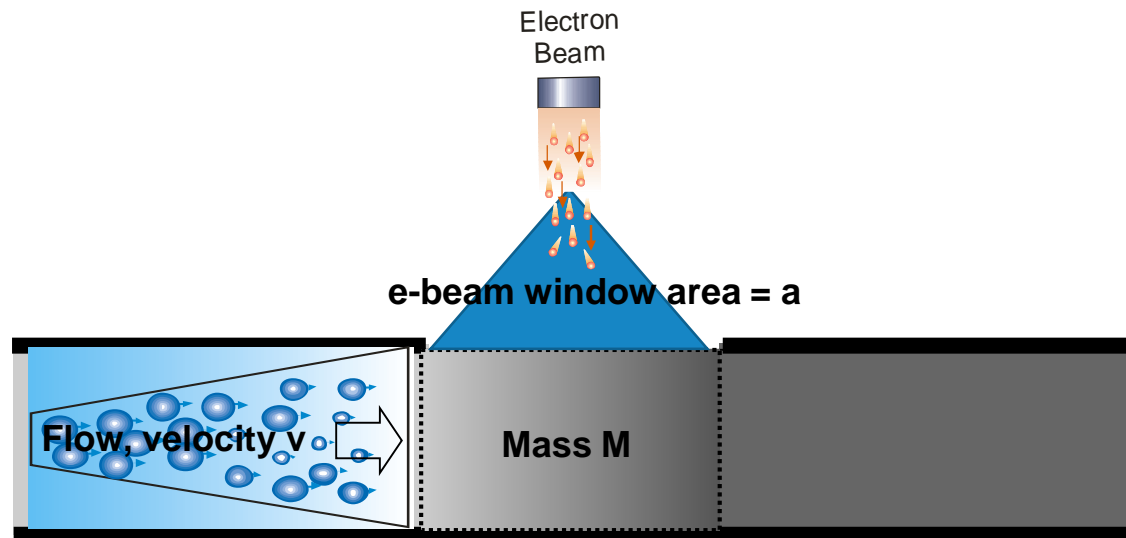
E-Beam power = 4000 watt (4000 J/sec)

## Each electron will have:

$3.2 \times 10^{-14}$  J of energy

## E-Beam will have:

$1.25 \times 10^{17}$  electrons per second



Bond dissociation energy (kJ/mol):

C-H	337.2
C-O	1076.5

Each electron has the potential to break:

~60,000 C-H

~20,000 C-O

# Comparison of Plasma Conversion Technologies

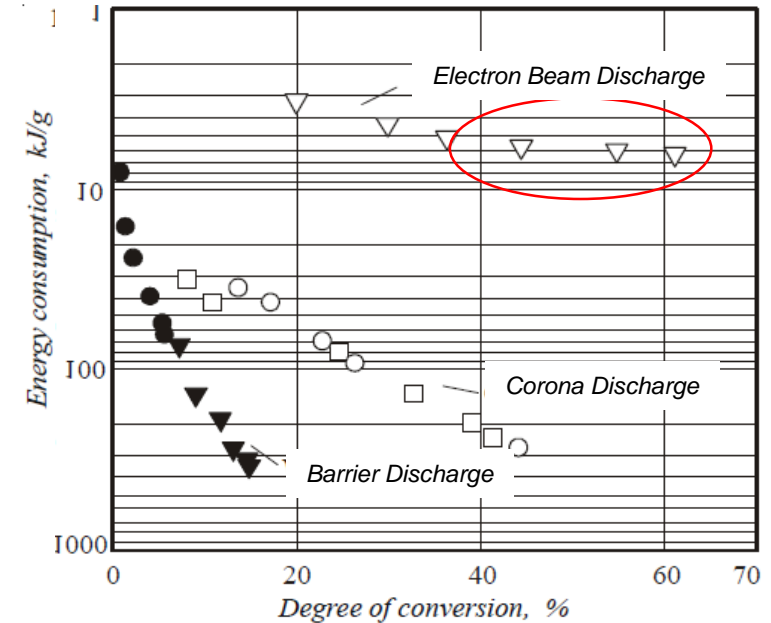
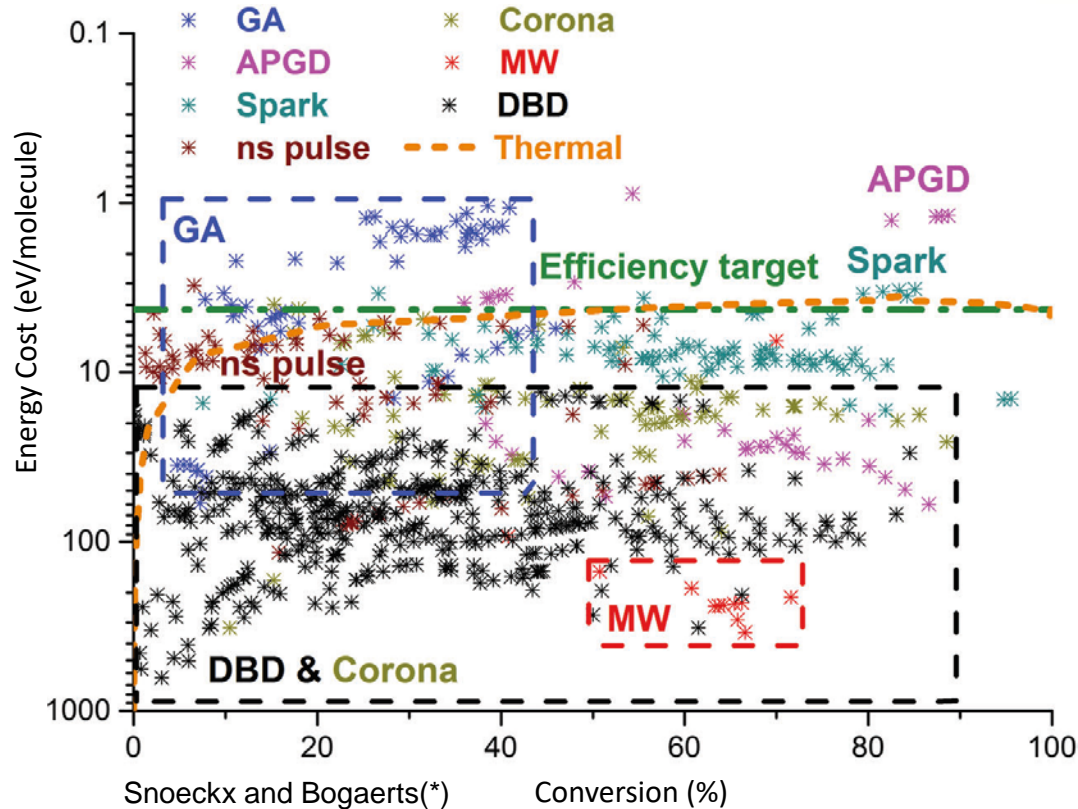


Fig. 1. Energy consumption vs. degree of conversion of methane using different methods of activation: ●, □, △ : 100% CH<sub>4</sub>; ○, ▲: mixture (1:1) of CH<sub>4</sub> and CO<sub>2</sub>.

Vinokurov(\*\*)

# Advantages Over Traditional Processes

- The DEBS process uses **high-energy electron beams** to break chemical bonds, allowing production of the desired chemicals at **near-ambient pressure and temperatures**
- Valuable chemical production by DEBS technology applied to CO<sub>2</sub> captured from coal-fired power plant will provide:
  1. **Low pressure / low temperature single reactor operation**
  2. **More energy-efficient CO<sub>2</sub> utilization**
  3. **Lower capital and operating costs**



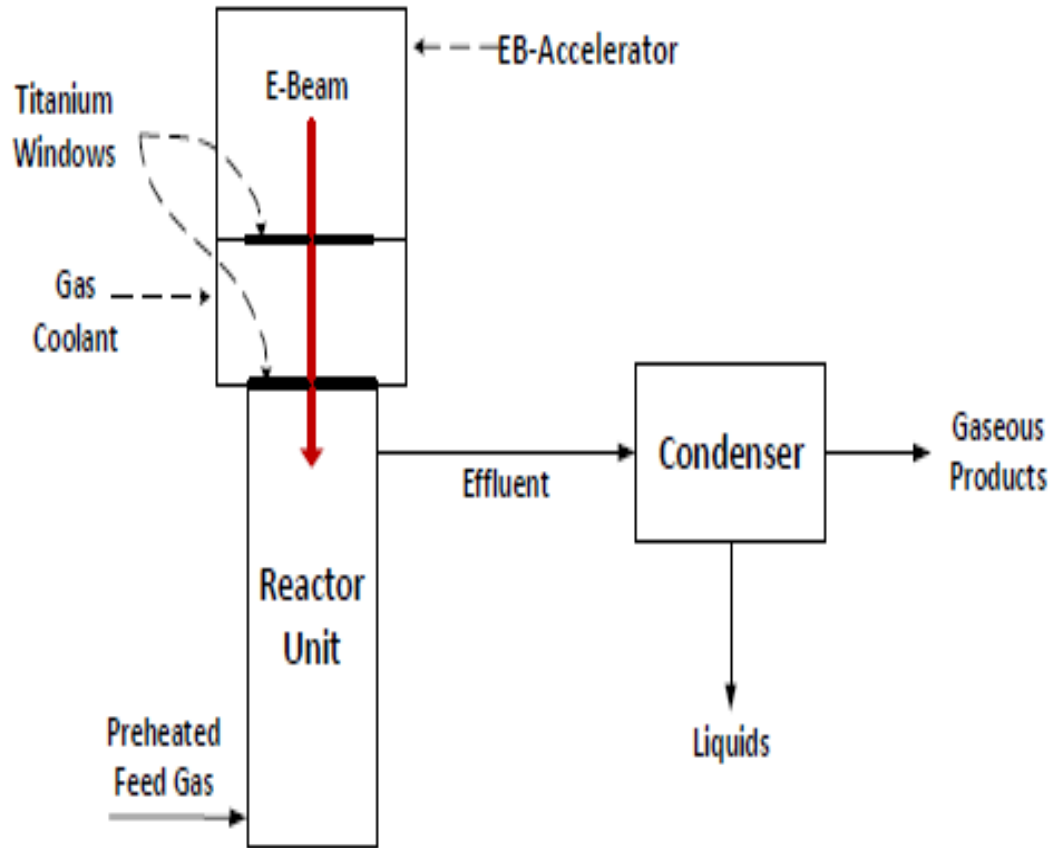
# Challenges

- Technology Challenge:
  - Delivering maximum e-beam dose
  - Determining which products are more probable
  - Minimizing power use by E-Beam Accelerator

# Technical Approach



# Experimental Design & Key Experimental Parameters



- E-Beam dose, (kJ/gm)
- Gas residence time (ms)
- E-Beam energy : 80-200 (keV), 20 (ma)
- Use of a promoter
- Use of catalyst(s) to promote desired reactions

# Milestone Schedule

Budget Period	Milestone Number	Title or Brief Task Description	Task Completion Date		
			Original Planned	Revised Planned	Actual
1A	1.1	Update Project Management Plan	4/30/17	6/1/17	6/27/17
1A	1.2	Kickoff Meeting	4/30/17	6/13/17	7/13/17
1B	2.1	Complete Final Design	5/1/17	8/1/19	8/1/19
1B	1.3	Submit Continuation Application	1/1/18	8/1/19	8/21/19
1B	7.1	Develop Preliminary Kinetic Model	12/31/17	10/31/19	
2	5.1	Start Parametric Testing	2/1/18	11/15/19	
2	5.2	Determine key operating parameters that would Maximize per pass CO <sub>2</sub> Conversion	3/31/18	12/31/19	
2	6.1	Identify Operating Conditions and Catalyst Combinations for Chemical Production	7/31/18	12/31/19	
2	7.2	Develop Kinetic Model	2/28/19	7/31/20	
2	8.1	Report Analysis of Experimental Data	2/28/19	7/31/20	
2	8.2	Complete Economic Analysis	2/28/19	7/31/20	
2	1.4	Submit Final Technical Report	4/1/19	8/31/20	



# Success Criteria

Decision Point	Date	Success Criteria
<b>Go / No-Go</b>	10/31/2019	<ul style="list-style-type: none"> <li>• Complete design and manufacture of testing skid with E-Beam reactor.</li> <li>• Successful commissioning of a viable reactor system and testing unit:               <ul style="list-style-type: none"> <li>Verify gas flow meter control by measuring the vent using a dry test meter</li> <li>Operate chiller for condenser to achieve less than -20°C in the condenser</li> <li>Verify detection limit of acetic acid and methane using RGA at 100ppmv</li> </ul> </li> <li>• Identify at least two catalysts to control the recombination and increase the yields for more valuable products</li> <li>• Complete design and cost estimate for the modification to a commercial electron accelerator shield housing</li> </ul>
<b>Completion of the project</b>	7/31/2020	<ul style="list-style-type: none"> <li>• 85% acetic acid, 15% methanol and CO selectivity</li> <li>• Higher than 25% CO<sub>2</sub> conversion per pass</li> <li>• Development of kinetic model</li> <li>• Reduce the COE by at least 50% compared to DOE Case 12-B</li> <li>• Achieve no net GHG emissions in production of products</li> </ul>

# Risk Status

## Initial Risks:

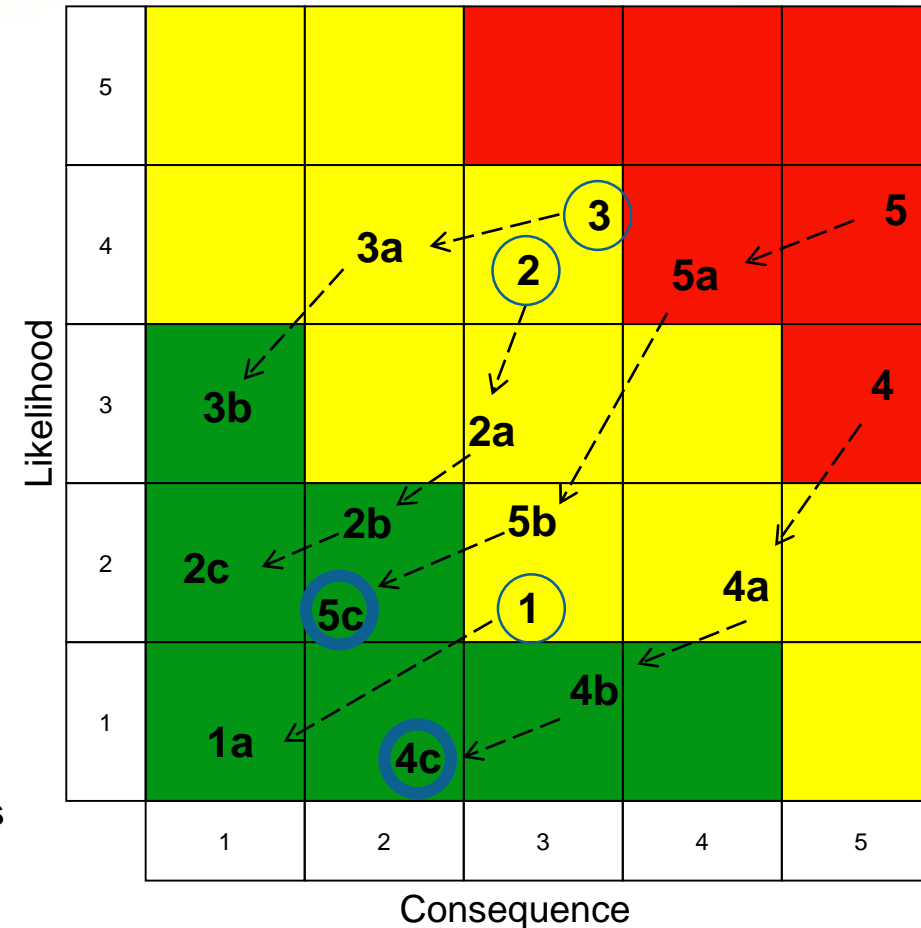
1. Reactor size too small for practical use in testing unit
  - 1a. Reduce E-Beam power and increase reactor size
2. Recombination reactions occur too quickly
  - 2a. Decrease residence time in reactor
  - 2b. Include a “recombination chamber”
  - 2c. Change location of catalyst
3. Reactions produce unidentified products
  - 3a. Increase analytical diagnostic capability
  - 3b. Change catalyst

## Resource Risk:

4. Accelerator provider not able to perform project
  - 4a. Reserved “beam time” for GTI’s experiments
  - 4b. Collaborate with other accelerator facilities
  - 4c. Operate accelerator at GTI

## Resource Issue:

5. Accelerator provider not able to perform experiments
  - 5a. Identify other facilities with similar capability
  - 5b. Collaborate with other accelerator facilities
  - 5c. Working with PCT-Ebeam Integration



# Progress and Current Status of Project

# Electron Accelerator

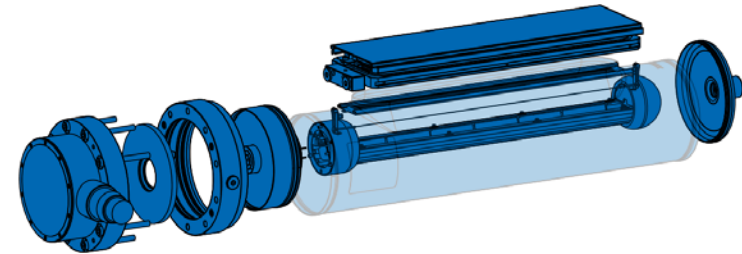
- Electron accelerator provided by PCT E-Beam Integration
- Uses a seal lamp unit from COMET
- Custom made accelerator and reactor housing





# COMET Sealed Lamp Accelerator

- 200 keV, 20 ma electron beam
- Beam window is 40mm x 400mm



# Completed Accelerator Housing

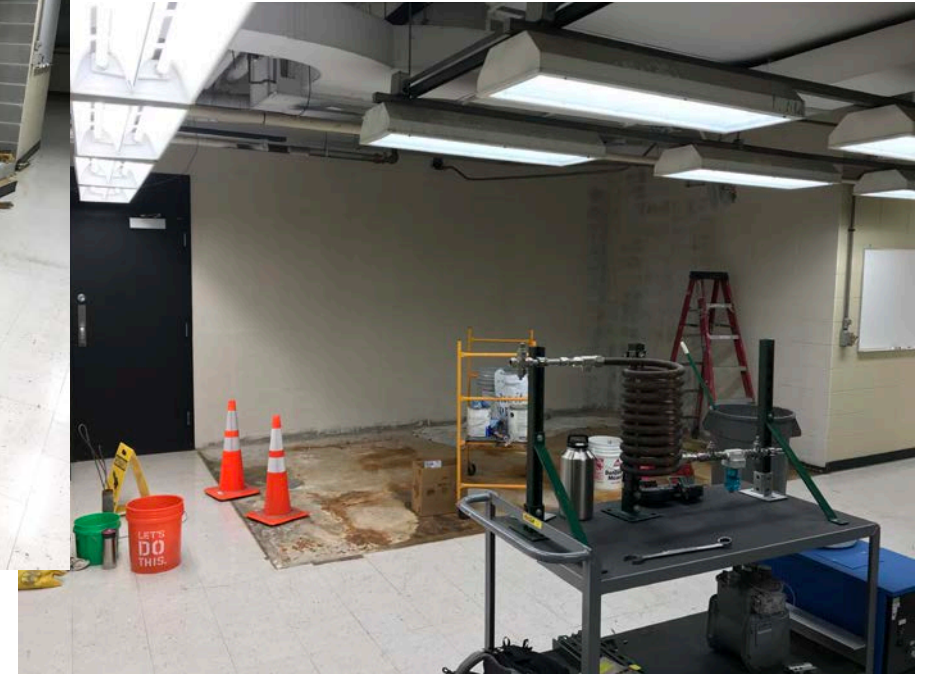
- Construction and completed accelerator housing during Factory Acceptance Test at PCT E-Beam Integration.





# GTI Facility Modification

- Remodeled a section of existing laboratory to accept electron accelerator.
- Upgraded ventilation to allow for high volume of combustible gas use.



# Electron Accelerator Delivered to GTI





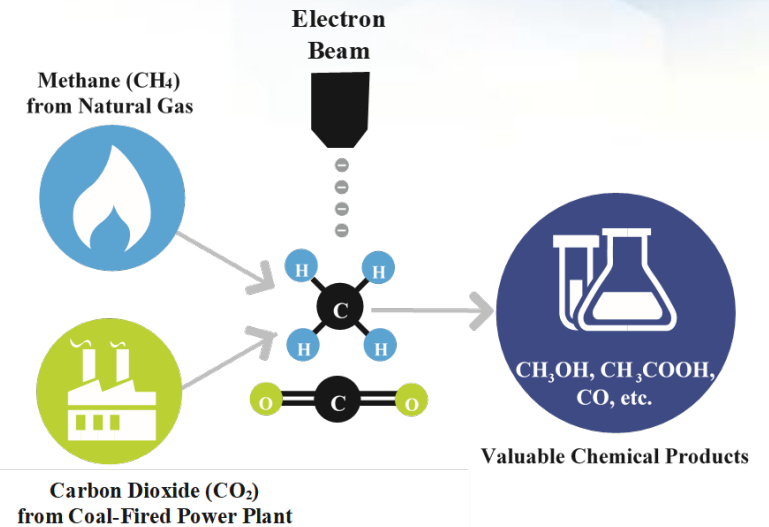
# Plans for Future Testing

# Plans for future testing/development

- **Finish reactor and testing skid fabrication**
- **Begin testing at GTI with new reactor and accelerator**
- **Kinetic model verification**
- **Techno-economic analysis**
- **Scaling up accelerator and reactor is not expected to be an issue:**
  1. **Available beam coverage from existing equipment is large**
  2. **Multiple accelerators can be connected to increase beam coverage if necessary**

# Summary

- Objective is to develop a commercially viable non-equilibrium process that breaks bonds directly unlike conventional chemistry that requires heating the entire molecule
- Irradiation of  $\text{CH}_4$  and  $\text{CO}_2$  mixture has been modeled for over 200 compounds with over 1600 reactions
- E-Beam reactor designed and constructed
- Electron accelerator is delivered to GTI and being commissioned for parametric testing



# Acknowledgements

- Financial Support



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# Disclaimer

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