High Temperature Ceramic-Carbonate Dual-Phase Membrane **Reactor for Pre-combustion Carbon Dioxide Capture**

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1 Introduction

• CO₂ Capture technology in IGCC process

Post-combustion capture

Inherent energy penalty in concentrating a flue gas at atmospheric pressure containing only 15% vol. CO_2 into pure CO_2 at high pressure.

3 Recent progress in ceramic-carbonate dual-phase MRs

• Ceramic-carbonate dual-phase (CCDP) membrane

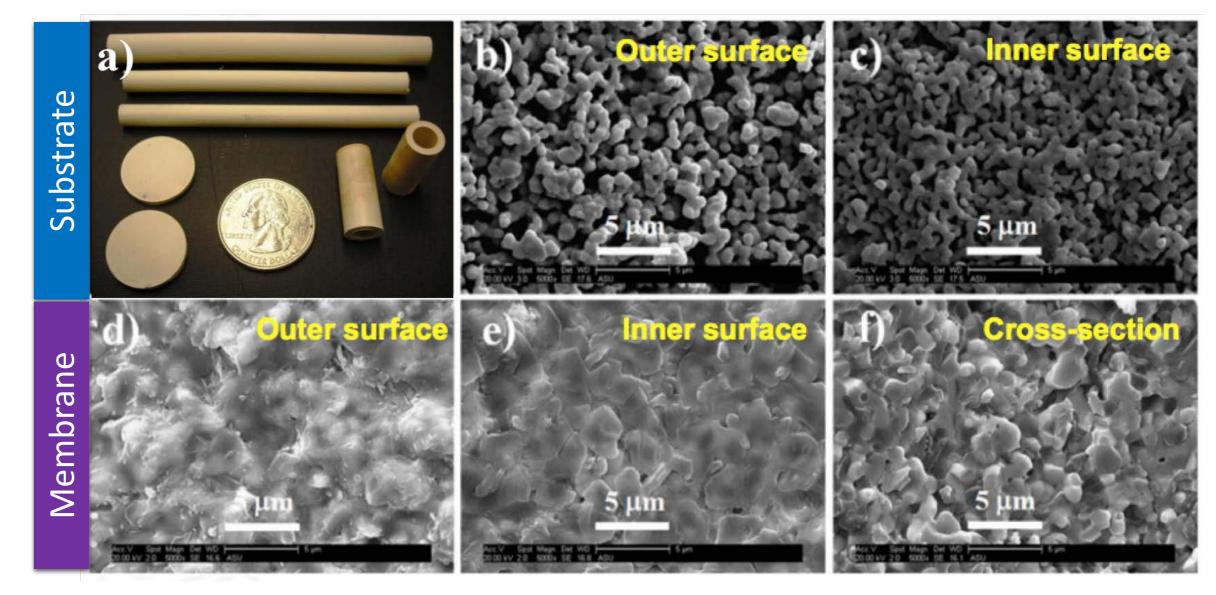
Substrate geometry

Disk membrane Tubular membrane

Membrane:

Symmetric thick membrane

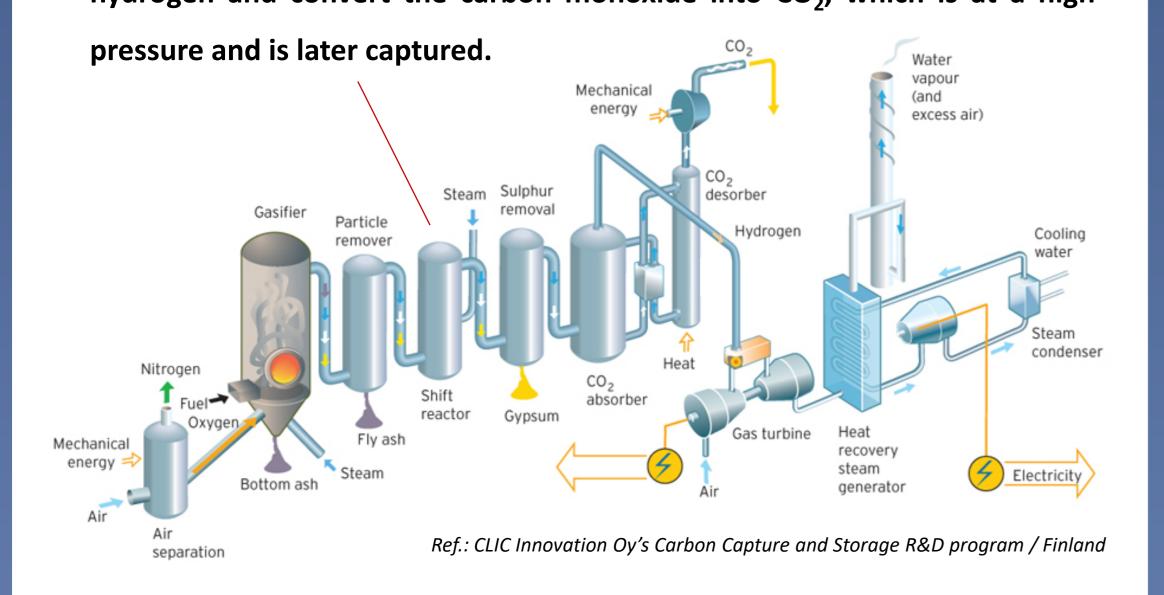
Morphology of dual-phase membrane



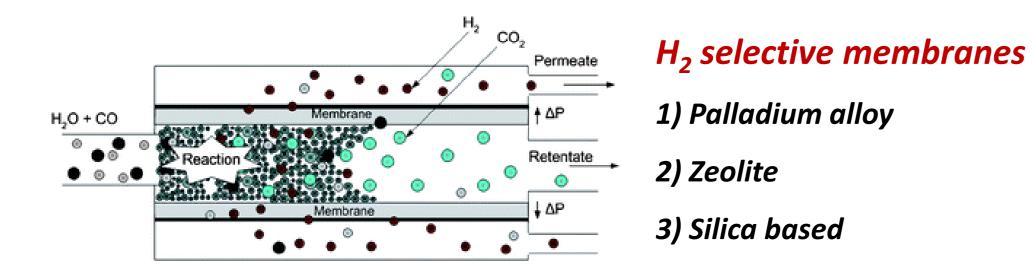


Pre-combustion capture

Water-gas-shift (WGS) reaction: syngas is shifted to produce additional hydrogen and convert the carbon monoxide into CO₂, which is at a high



- Membrane reactor technology for WGS reaction
 - ✓ equilibrium shift and CO conversion enhancement
 - \checkmark CO₂ is concentrated at high pressure (30–35 bar)

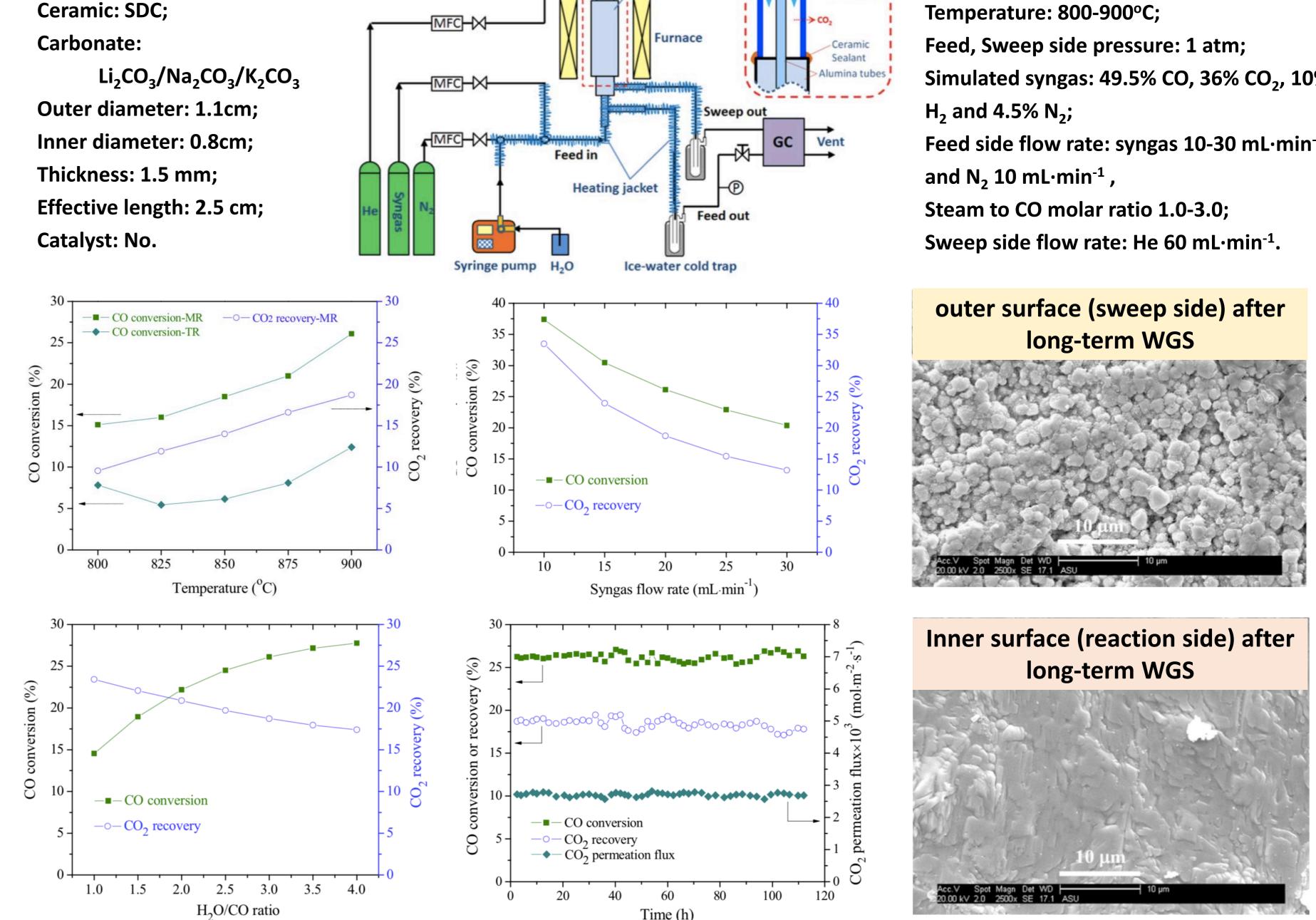


Asymmetric thin membrane

Ceramic phase materials

 $La_{0.6}Sr_{0.4}Co_{0.8}Fe_{0.2}O_{3-\delta}$ (LSCF) **Yttria-stabilized zirconia (YSZ)** $Sm_{0.2}Ce_{0.8}O_{1.9-\delta}$ (SDC) $Bi_{1.5}Y_{0.3}Sm_{0.2}O_3$ (BYS) $La_{0.85}Ce_{0.1}Ga_{0.3}Fe_{0.65}Al_{0.05}O_{3-\delta}$ (LCGFA)

• Tubular CCDP-MRs for high-temperature WGS Ref.: J. Membr. Sci., 2016, 520, 907-913.



Reaction conditions: Simulated syngas: 49.5% CO, 36% CO₂, 10% Feed side flow rate: syngas 10-30 mL·min⁻¹



Ref.: Energy Environ. Sci., 2010, 3, 268-278

4) Carbon molecular sieve

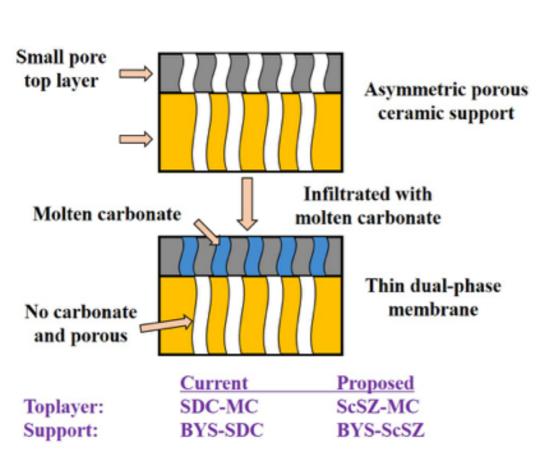
Major issues:

- relatively low H₂ permeance and H₂/CO₂ selectivity
- limited H₂ purity and recovery in the permeate stream
- Objective
 - developing a novel tubular membrane reactor (MR) made of **CO₂ permeable membranes (CO₂ permeance > 2000 GPU** and CO₂ selectivity > 500) for high-temperature and highpressure WGS reaction.
 - producing high-pressure H₂ and CO₂ streams with a purity over 90% and 99%, respectively.

2 Dual-phase membranes

Characteristics

composed of a porous ceramic phase and a molten carbonate (MC) phase



4 Scope of the further research for CCDP-MRs

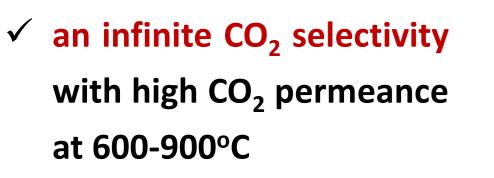
Research Tasks to be Performed

Task 1.0 Project Management and Planning (BP1 and BP2)

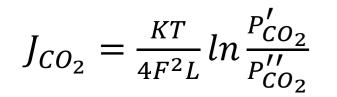
Task 2.0 Synthesis and Characterization of Dual-Phase Membranes (BP1)

Task 3.0 High Temperature, High Pressure CO₂ Permeation Studies (BP1)

Task 4.0 Development of Improved Ceramic-Carbonate Dual-Phase Materials and Membranes (BP1)

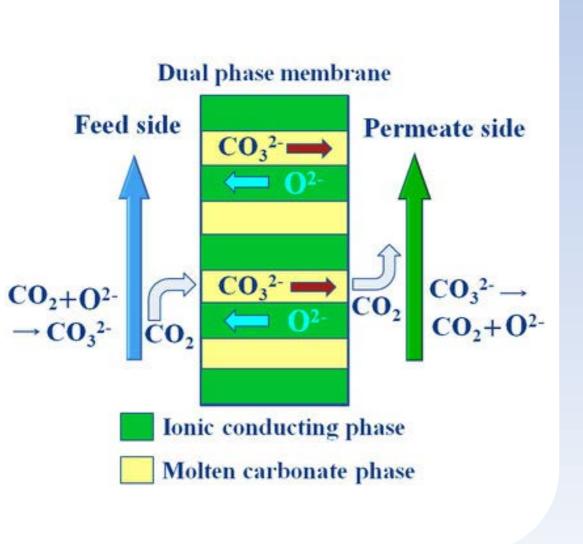


*CO*², *transport through dual-phase membrane*



F is Faraday constant, L is membrane thickness, and K, the total conductance is related to the ratio of volume fraction to tortuosity, carbonate or oxygen ionic conductivity of carbonate (sub c) and ionic-conducting ceramic (sub i) phase.

 $K = \frac{(\varepsilon/\tau)_c \sigma_c (\varepsilon/\tau)_i \sigma_i}{(\varepsilon/\tau)_c \sigma_c + (\varepsilon/\tau)_i \sigma_i}$



Task 5.0 Study on CO₂ Permeation Properties of CCDP Membranes (BP1)

Task 6.0 Fabrication and Characterization of CCDP Tubular Membranes (BP2)

Task 7.0 Modeling and analysis of CCDP membrane reactor for WGS (BP2)

Task 8.0 Studies on WGS in Improved Dual-Phase Membrane Reactors (BP2)

Task 9.0 Process Design and Techno-Economic Analysis (BP2)

5 Conclusions

 \square We propose a novel process for CO₂ pre-combustion capture that is applying a ceramic-carbonate dual phase

membrane reactor with CO_2 selective membrane for high-temperature WGS.

☑ Our work experimentally demonstrated that CCDP-MRs could offer a strong improvement in the CO conversion and

the CCDP membranes show good CO₂ permeation flux and high thermal and chemical stability under WGS reaction.

 \square Further studies will be mainly focused on the development of new membrane materials.





2018 NETL CO₂ Capture Technology Project Review Meeting, Pittsburgh, August 13-17, 2018



Ceramic phase materials

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