

# **Development of Carbon Molecular Sieves Hollow Fiber Membranes based on Polybenzimidazole Doped with Polyprotic Acids with Superior H<sub>2</sub>/CO<sub>2</sub> Separation Properties**

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NETL CO<sub>2</sub> Capture Technology Project Review Meeting  
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# Project Overview

**Award number:** DE-FE0031636

**Project period:** 10/1/18 to 9/30/21

**Program manager:** Andrew O'Palko

**Project Objective:** Develop CMS hollow fiber membranes with H<sub>2</sub> permeance of 1000 GPU and H<sub>2</sub>/CO<sub>2</sub> selectivity of 40 at 200-300 °C, enabling membrane-based systems capturing 90% CO<sub>2</sub> from coal-derived syngas with 95% CO<sub>2</sub> purity at a cost of electricity 30% less than baseline capture approaches.

Team Members	Federal Share	Cost-share	Total	Roles
UB	\$534,999	\$202,225	\$737,224	Materials development
LANL	\$200,000	\$0	\$200,000	Membrane development
Trimeric	\$ 65,000	\$0	\$ 65,000	Techno-economic analysis
Total:	\$799,999	\$202,225	\$1,002,224	

# **Project Scope in Each Budget Period**

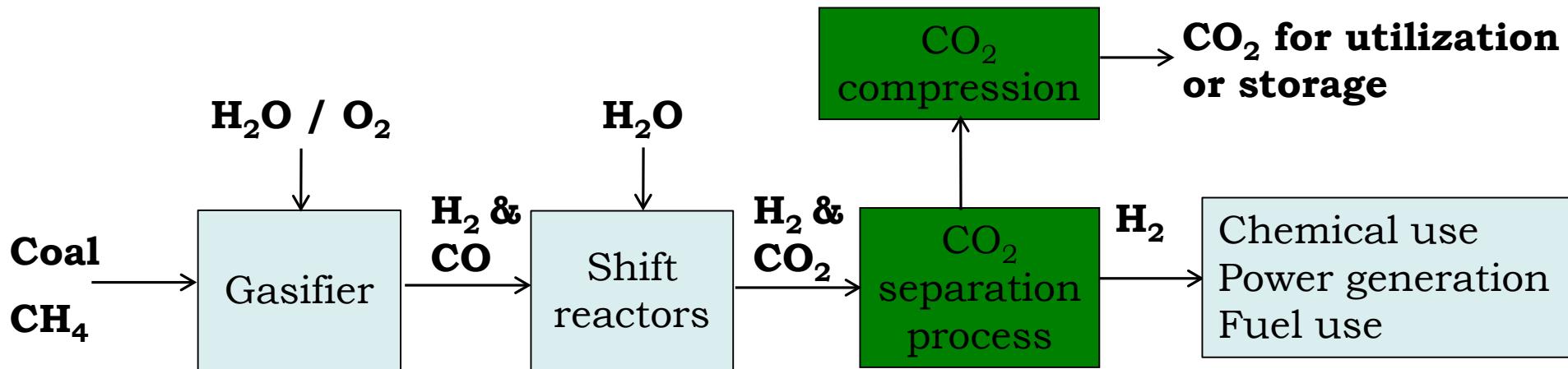
## **BP 1 Materials development (10/1/18 – 3/31/20; 18 months)**

- Optimize CMS materials with an H<sub>2</sub> permeability of 200 Barrers and H<sub>2</sub>/CO<sub>2</sub> selectivity of 40 with simulated syngas; and
- Optimize the hollow fiber membranes based on PBI doped with polyprotic acids.

## **BP 2 Membrane development (4/1/20 – 9/30/21; 18 months)**

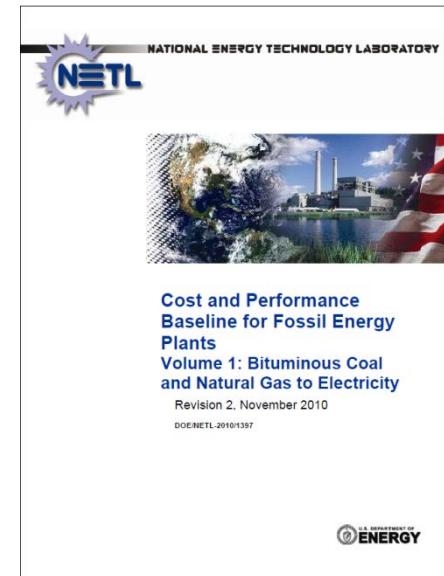
- Optimize membranes achieving the targeted H<sub>2</sub>/CO<sub>2</sub> separation performance;
- Test membranes using simulated syngas containing H<sub>2</sub>S, CO and water vapor;
- Determine the efficiency of the membrane reactors for the WGS reaction; and
- Conduct the techno-economic analysis.

# **CO<sub>2</sub> separation is energy-intensive and expensive**



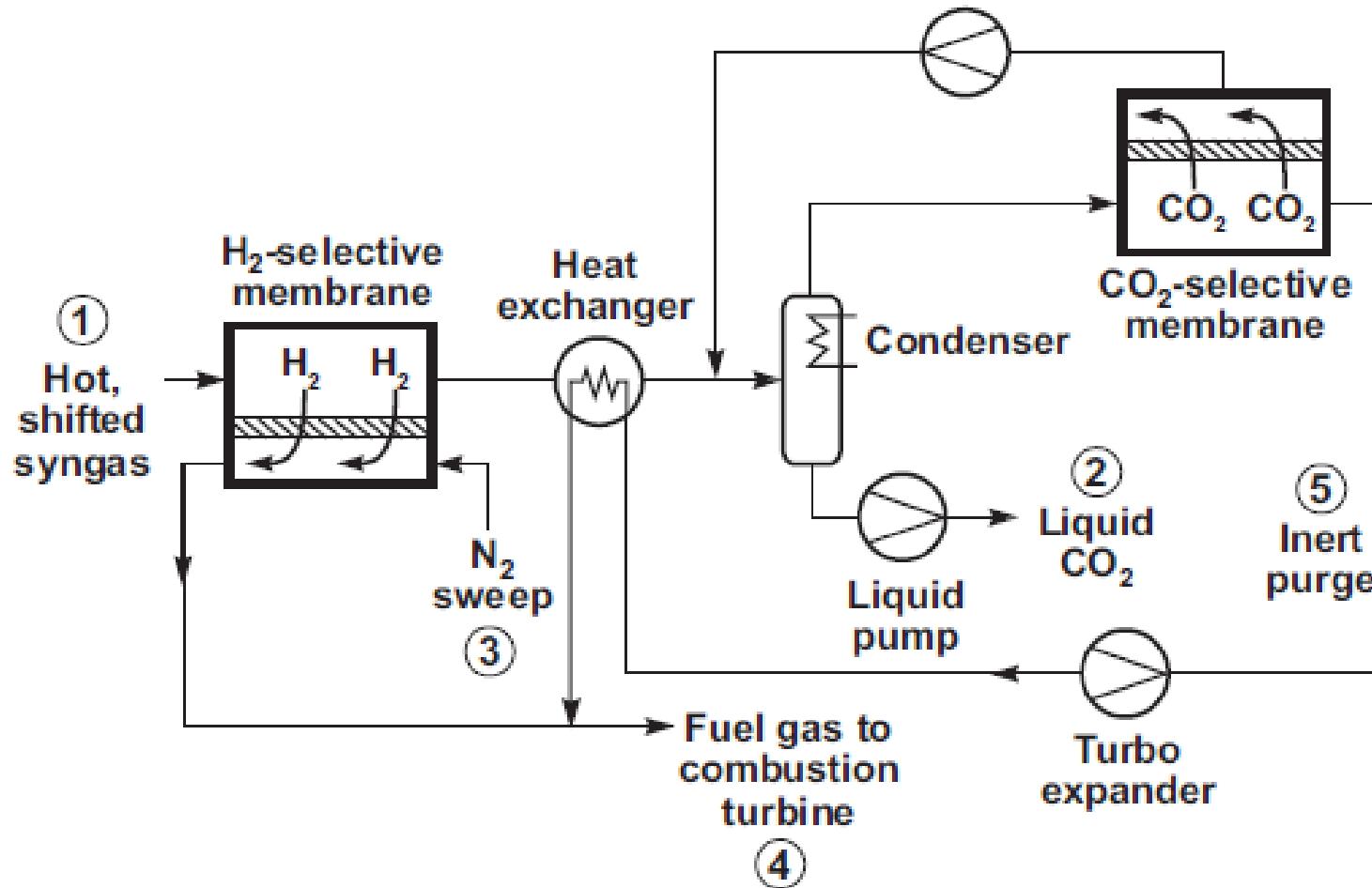
## **GEE IGCC/Selexol 543 MWe plant (Case 2)**

	<b>CO<sub>2</sub> capture</b>
<b>Power consumption</b>	50 MWe
<b>Capital cost</b>	\$252 MM



**Lower cost and more energy efficient separation technology is needed.**

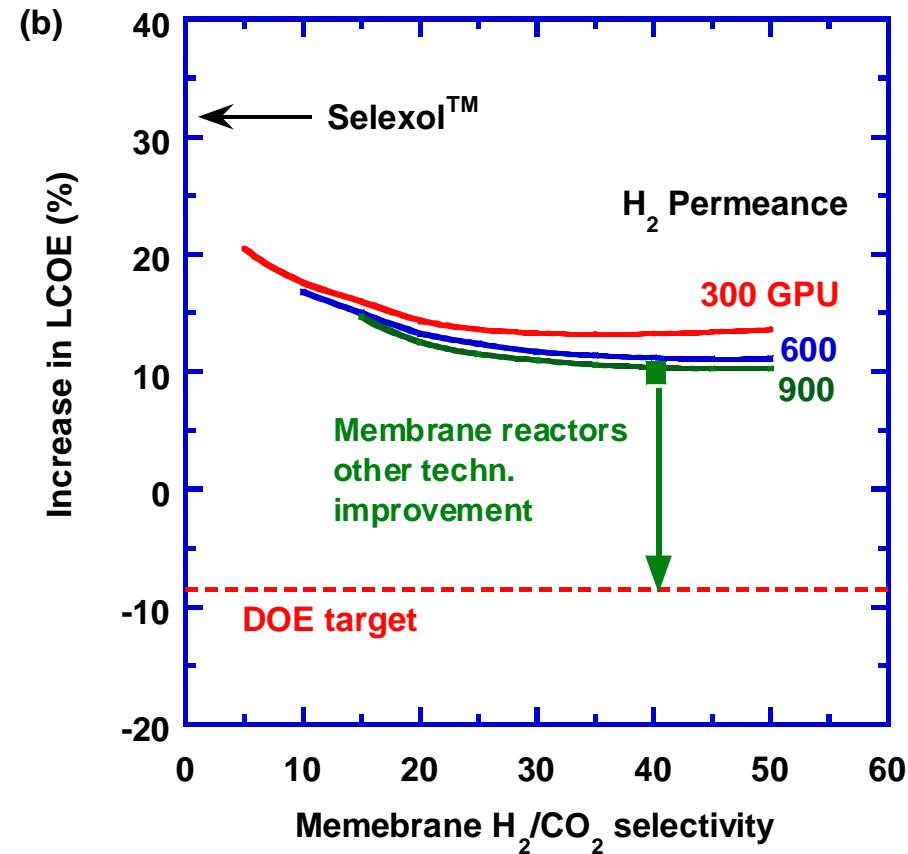
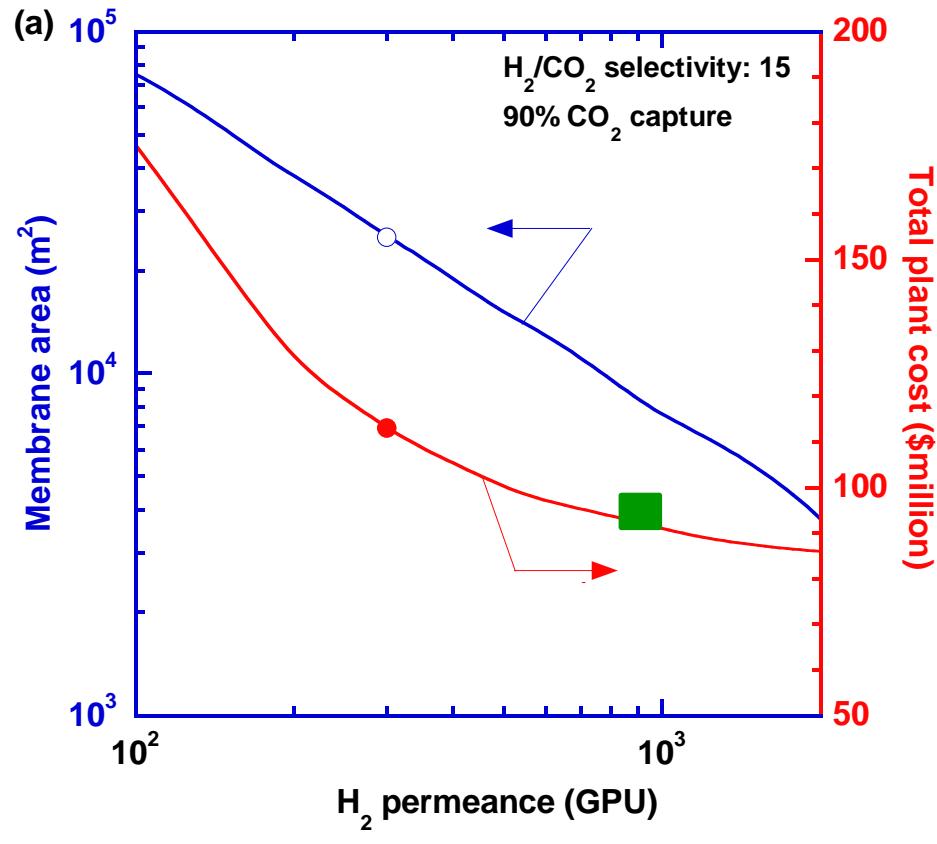
# MTR's Membrane Process Design



Merkel, Zhou and Baker, *J. Membr. Sci.*, 389, 442 (2012)

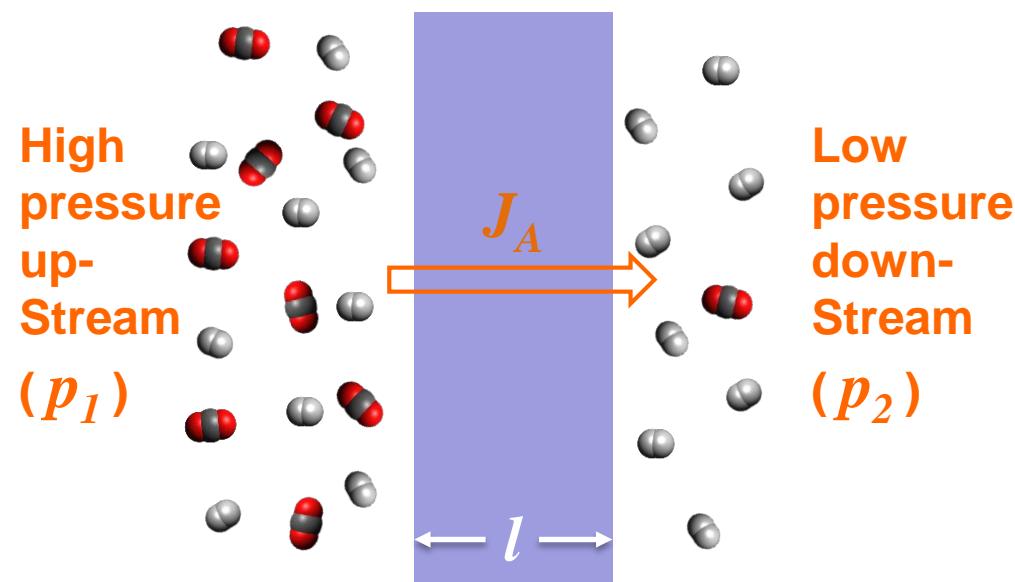
Merkel, et al., *NETL CO<sub>2</sub> Capture Technology Meeting*, 2011.

# MTR's Techno-Economic Analysis



# Membrane: Energy-efficient Separation

## Solution-diffusion model



- (1) Sorption on upstream side
- (2) Diffusion down partial pressure gradient
- (3) Desorption on downstream side

## Productivity - Permeability

$$P_A = S_A \times D_A$$

## Purity - Gas selectivity

$$\alpha_{\text{H}_2/\text{CO}_2} = \frac{P_{\text{H}_2}}{P_{\text{CO}_2}} = \left( \frac{S_{\text{H}_2}}{S_{\text{CO}_2}} \right) \times \left( \frac{D_{\text{H}_2}}{D_{\text{CO}_2}} \right)$$

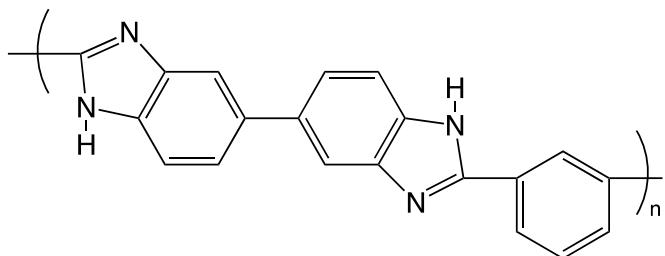
solubility selectivity

diffusivity selectivity

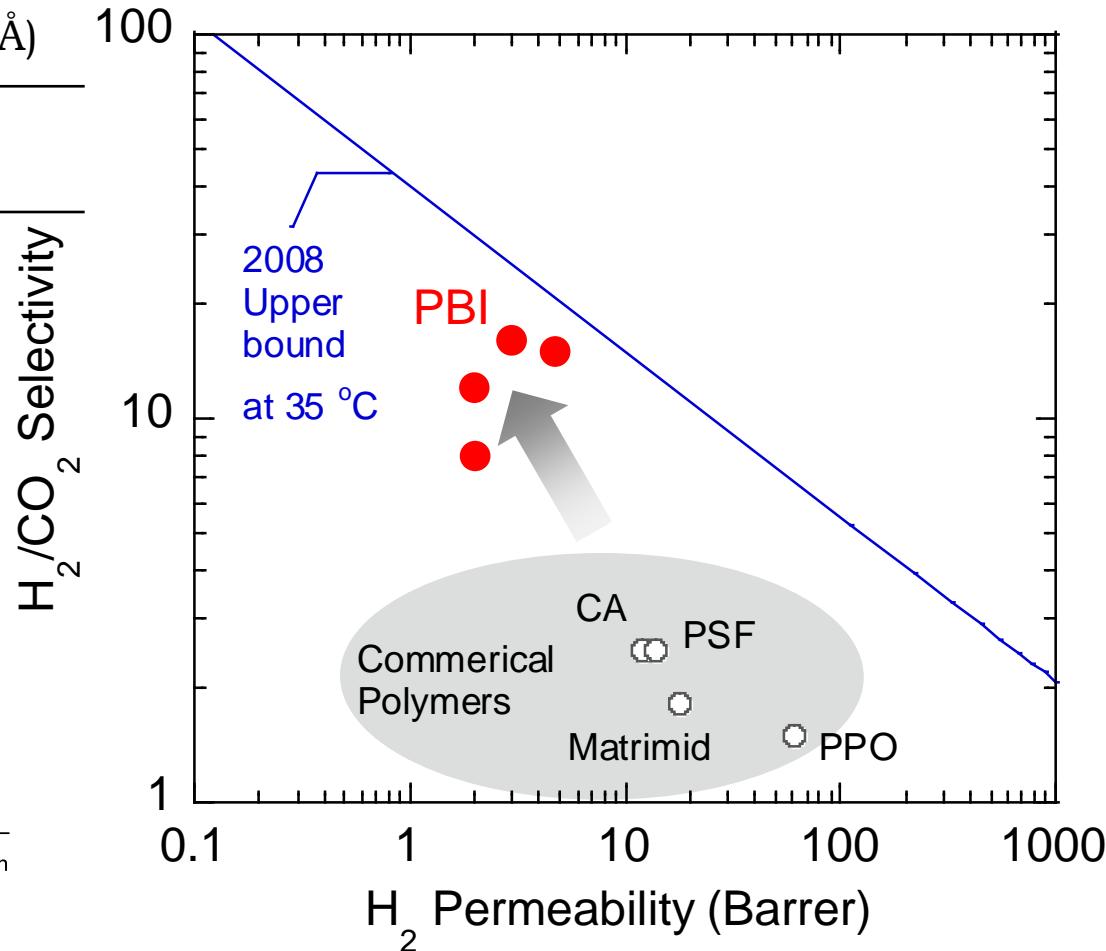
# Permeability/Selectivity Tradeoff

	Critical temperature (K)	Kinetic diameter (Å)
H <sub>2</sub>	33	2.89
CO <sub>2</sub>	304	3.3

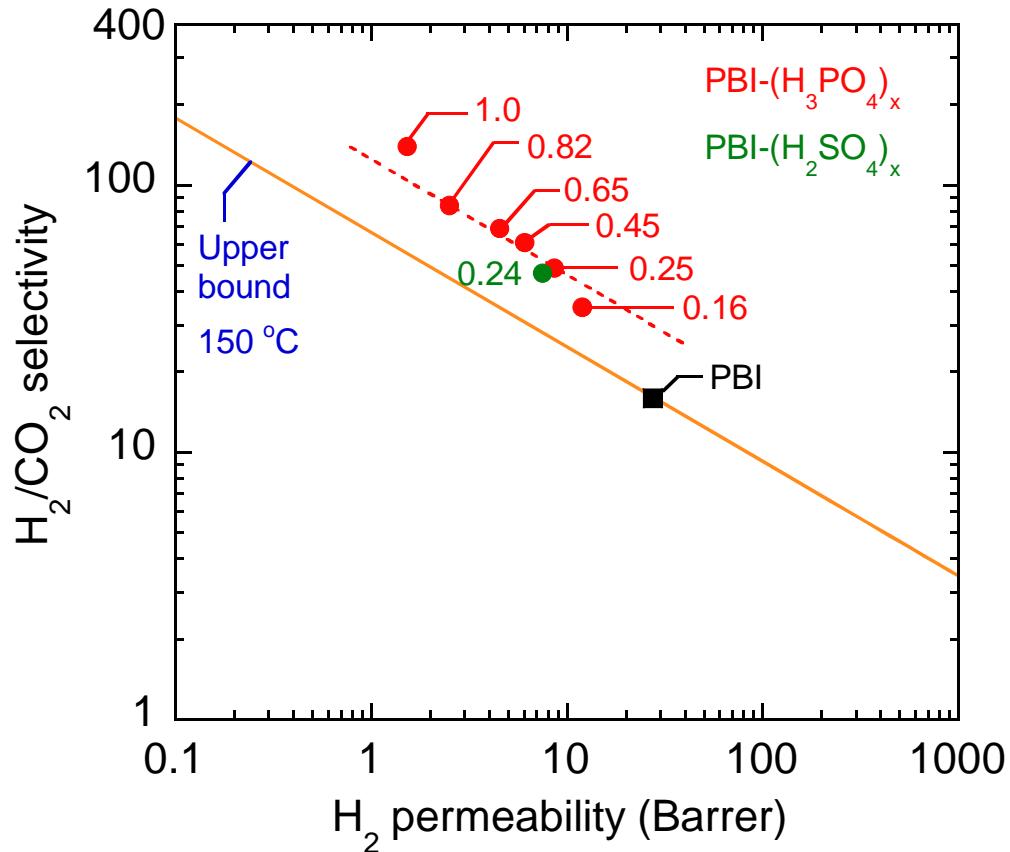
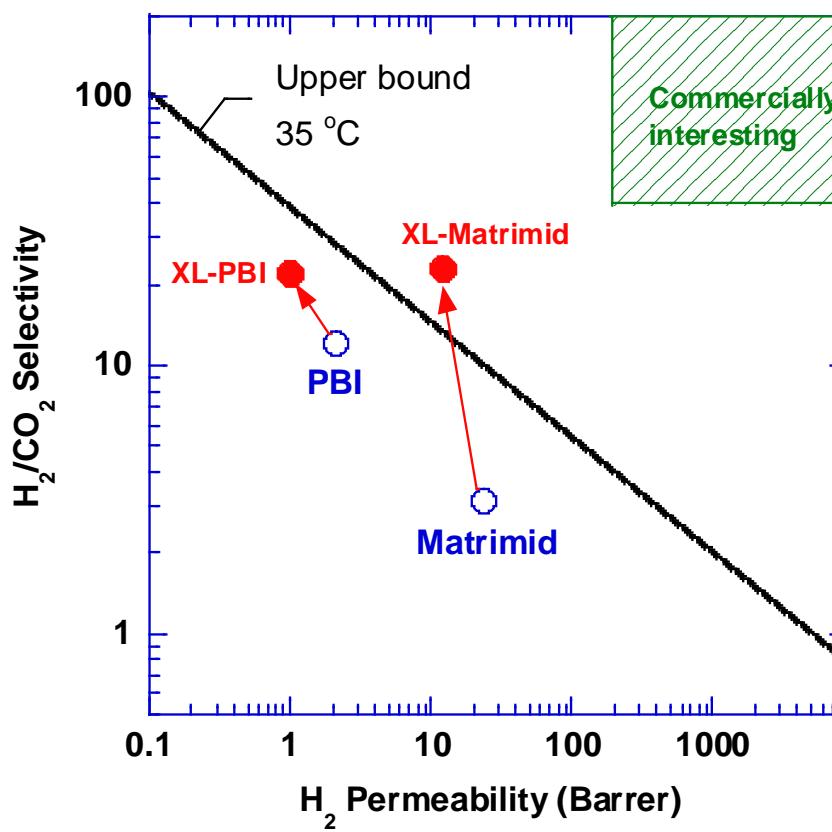
$$\frac{S_{H_2}}{S_{CO_2}} \ll 1 \text{ and } \frac{D_{H_2}}{D_{CO_2}} \gg 1$$



Polybenzimidazole (PBI)  
(Celazole®)

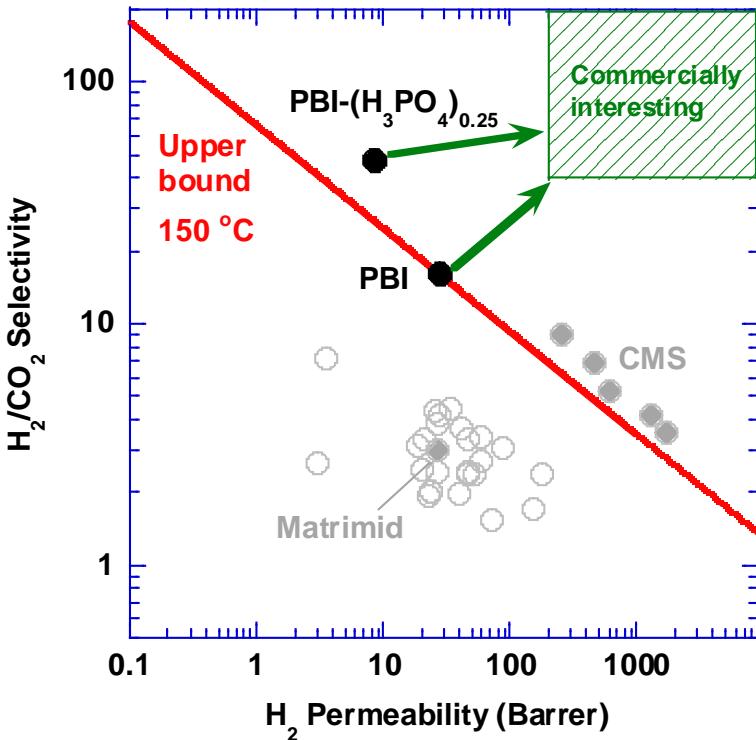


# Polymeric Membranes for H<sub>2</sub>/CO<sub>2</sub> Separation



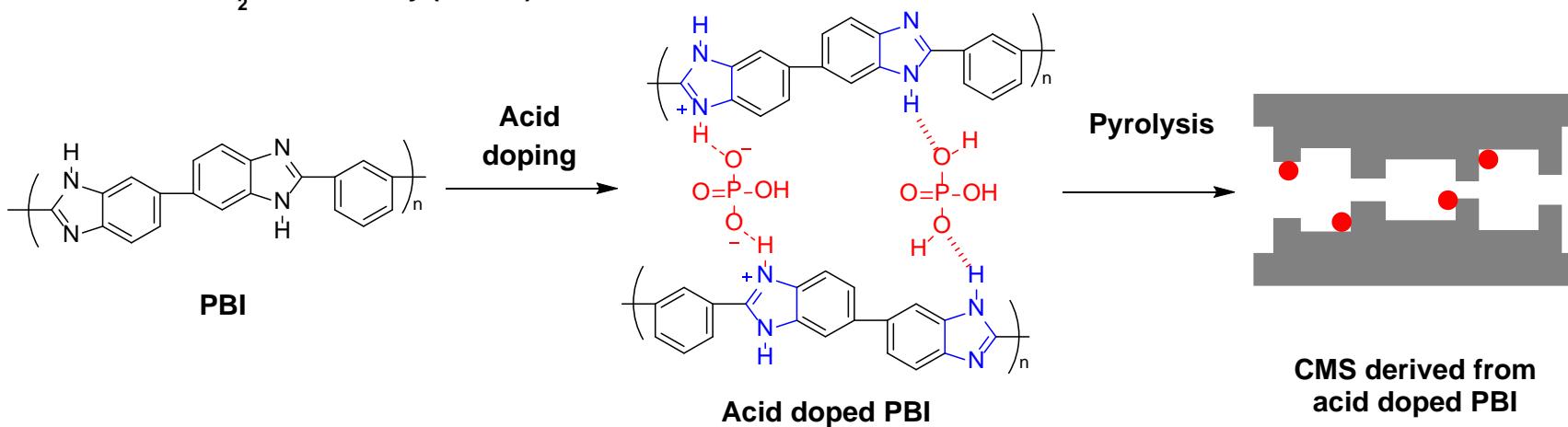
- Shao, Low, Chung, and Greenberg *J. Membr. Sci.* 2009, 327, 18-31.  
Zhu, Swihart, and Lin *J. Mater. Chem. A* 2017, 5 (37), 19914-19923.  
Zhu, Swihart and Lin, *Energy Environ. Sci.* 2018, 11 (1), 94-100.

# Our Approach: Carbonizing PBI/acid to Enhance H<sub>2</sub>/CO<sub>2</sub> Separation Performance

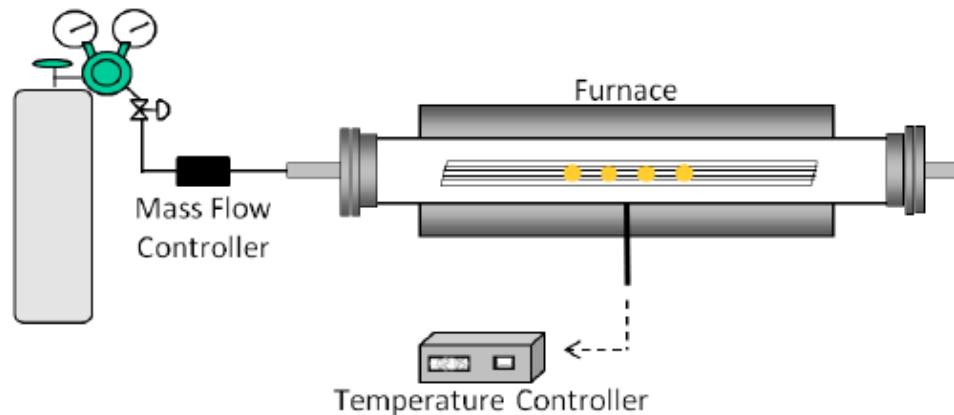
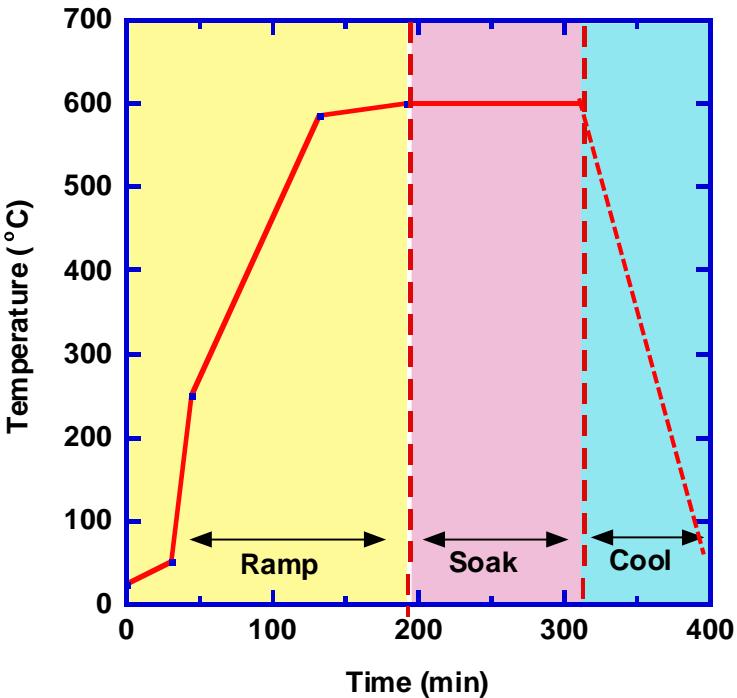


## Poly(benzimidazole) (PBI)

- Commercially available
- High  $T_g$  (417 °C),  $T_d$ : 550 °C



# Protocol to Prepare Carbon Molecular Sieve (CMS) Materials

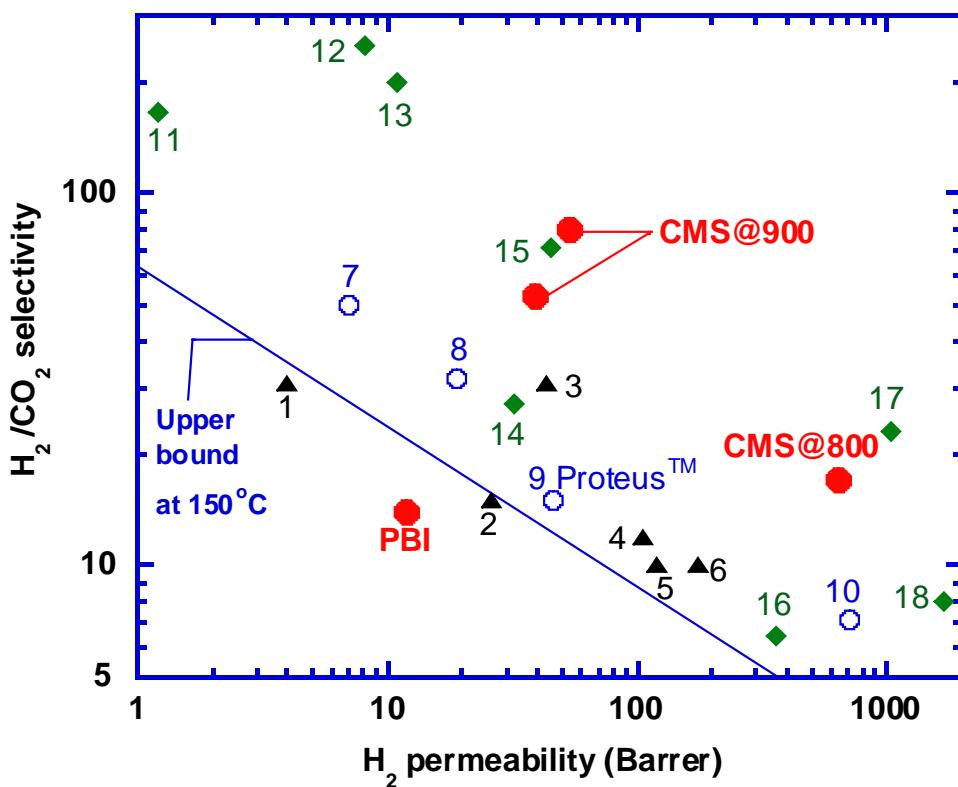
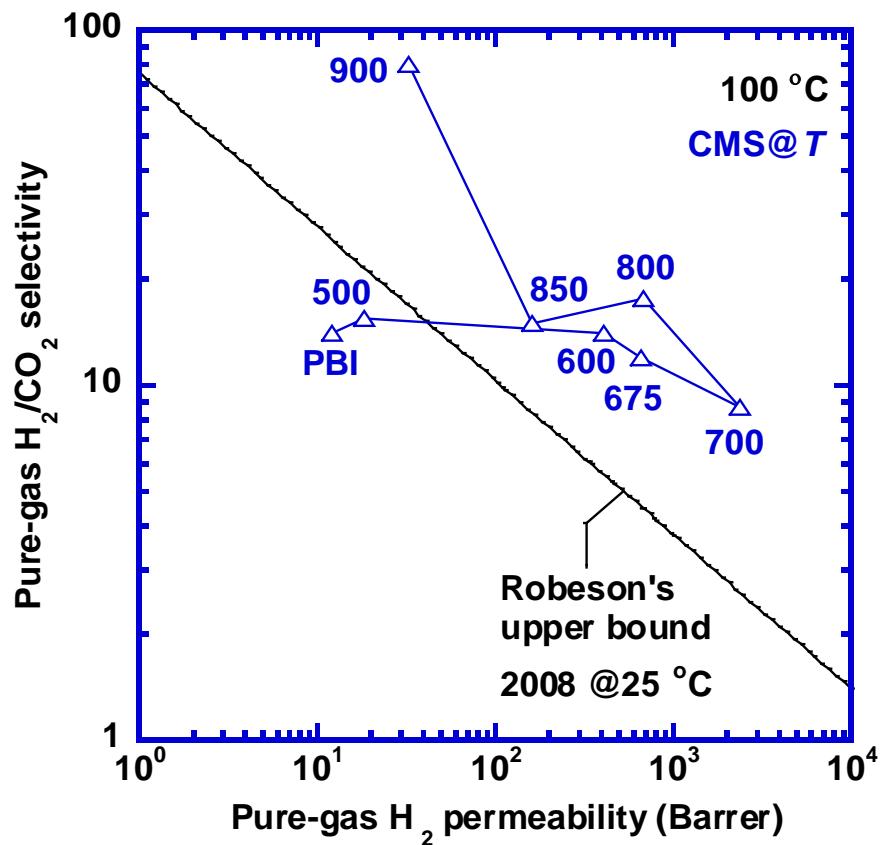


## Pyrolysis Protocols: 200 cc N<sub>2</sub>/min

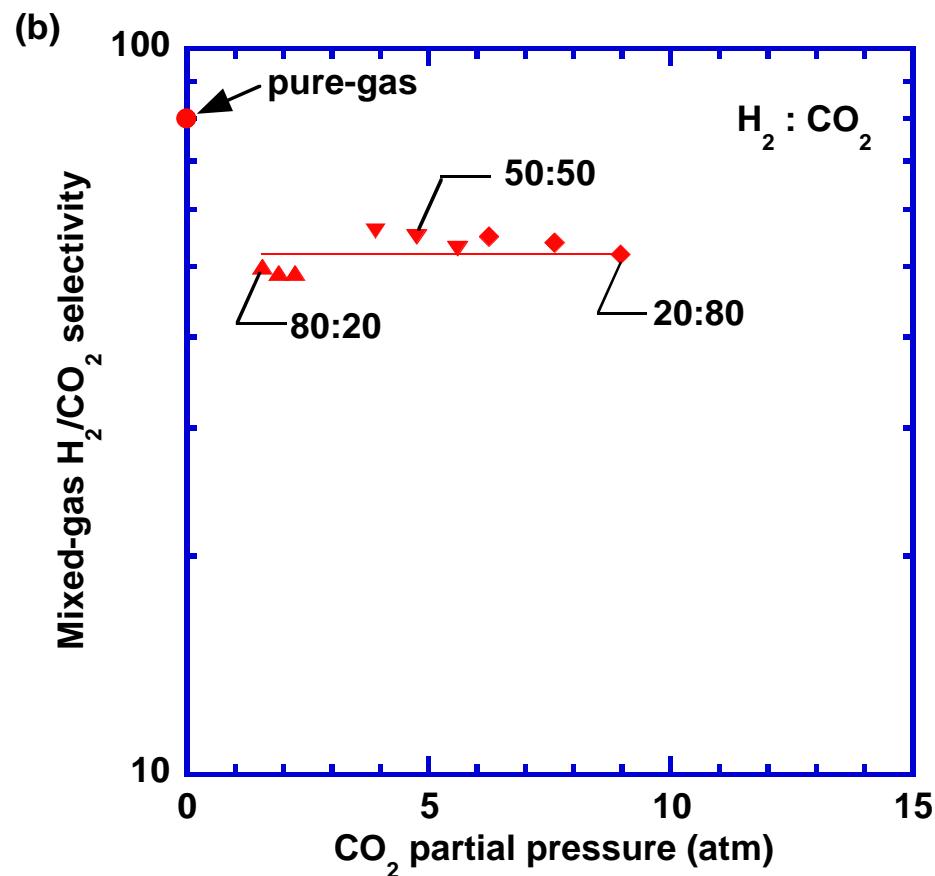
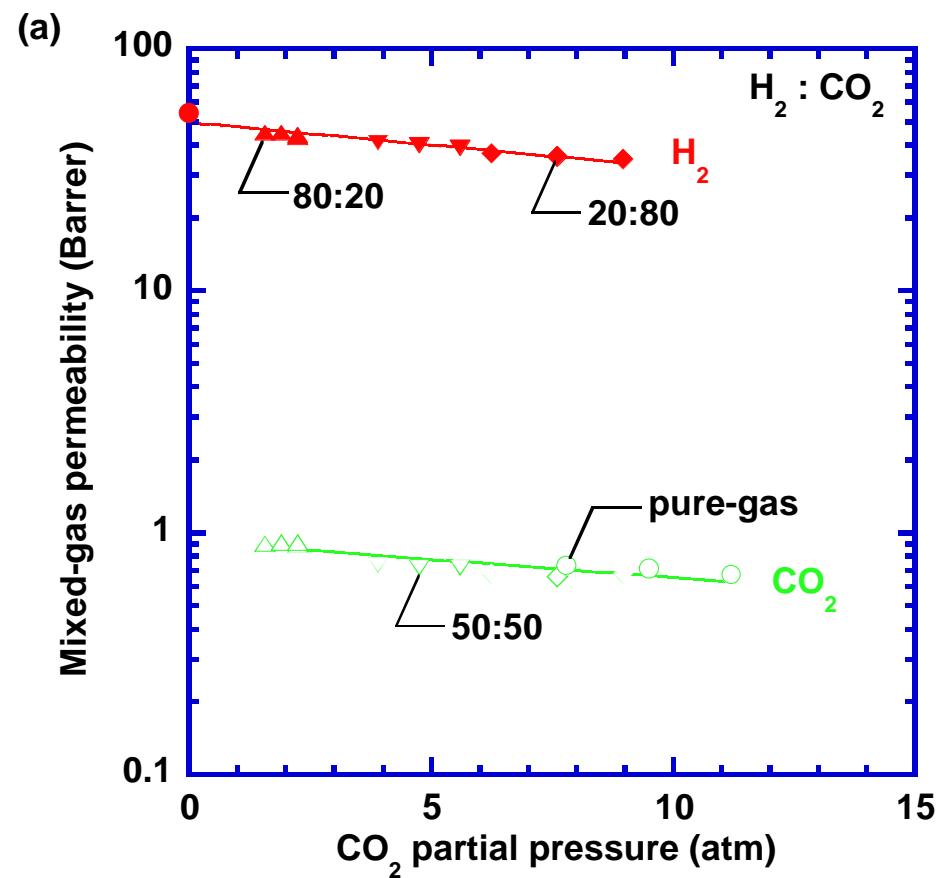
- Isothermal at 25 °C for 30 min
- 25 °C to 50 °C with ramp rate of 0.83 °C/min
- 50 °C to 250 °C with ramp rate of 13.3 °C/min
- 250 °C to ( $T_{max}$  - 15 °C) with ramp rate of 3.85 °C/min
- ( $T_{max}$  - 15 °C) to  $T_{max}$  with ramp rate of 0.25 °C/min
- Stay at  $T_{max}$  for 2 hours and cool down under N<sub>2</sub> flow

$T_{max}$ : 600-900 °C

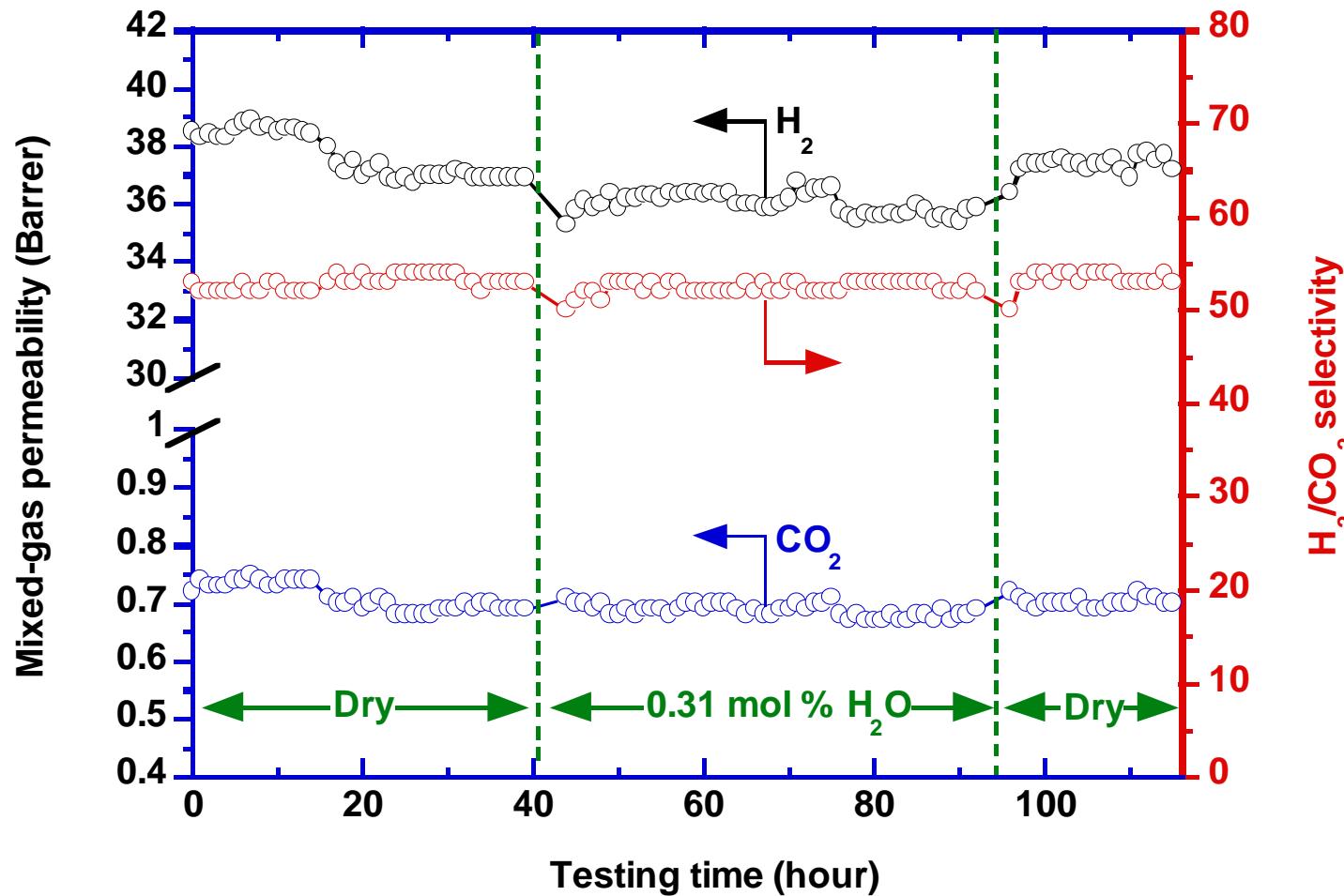
# Super H<sub>2</sub>/CO<sub>2</sub> Separation Performance of PBI-CMS



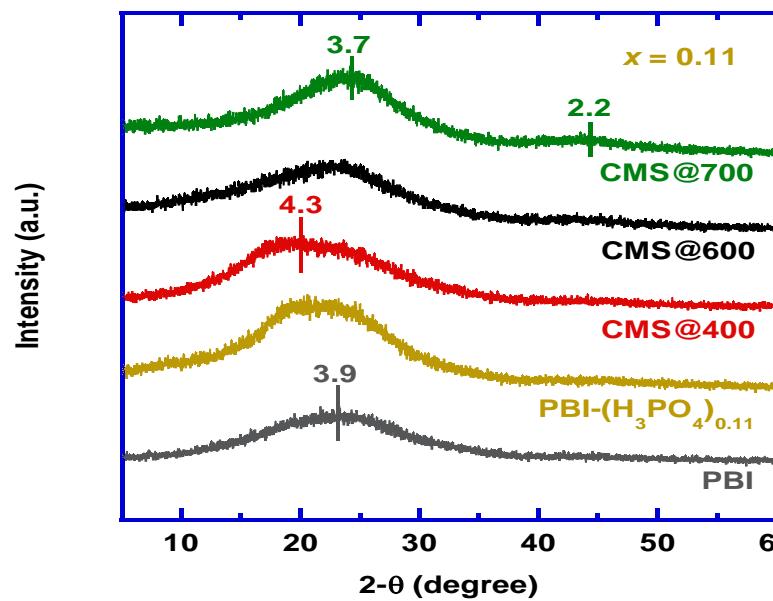
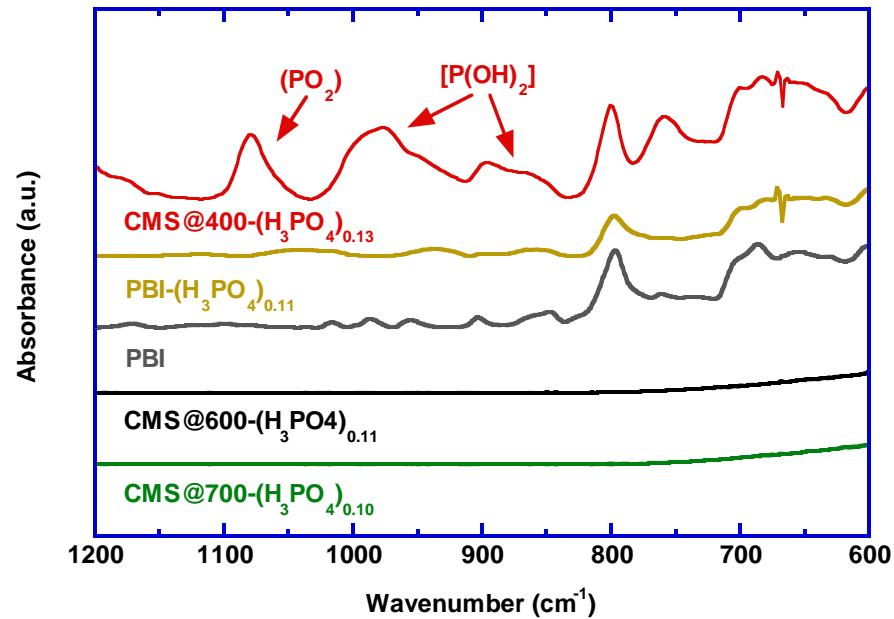
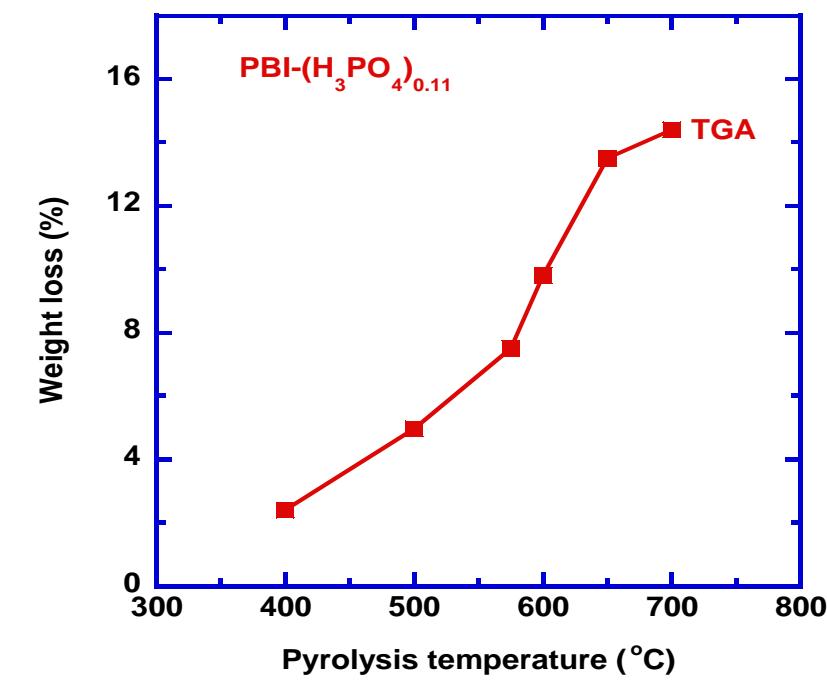
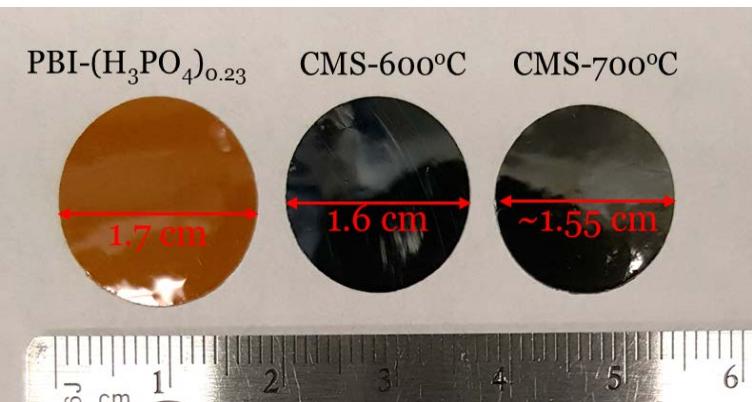
# Mixed-gas H<sub>2</sub>/CO<sub>2</sub> Separation Performance of PBI-CMS@900 at 100 °C



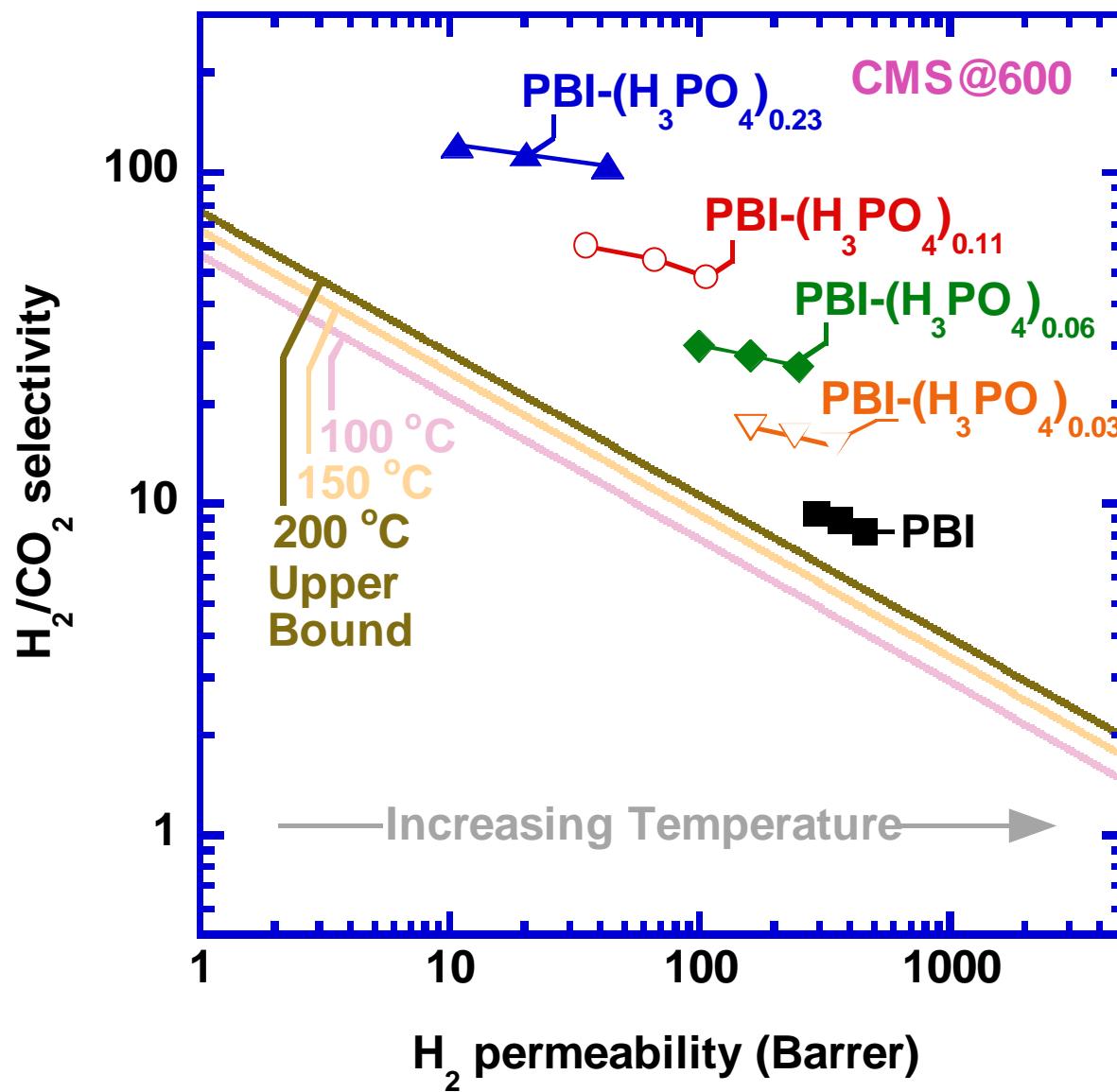
# Stable H<sub>2</sub>/CO<sub>2</sub> Separation Performance of PBI-CMS@900 at 100 °C



# Carbonization of $\text{H}_3\text{PO}_4$ Doped PBI

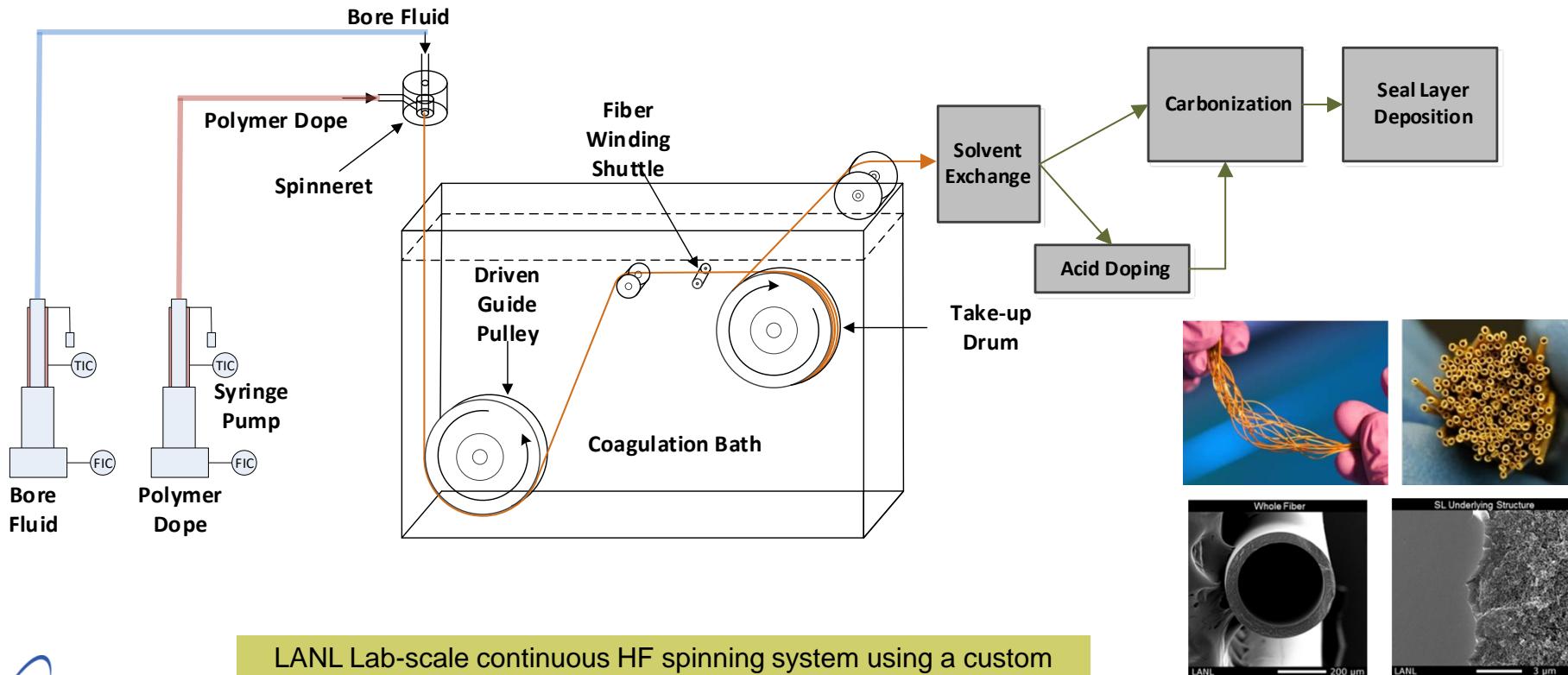


# Effect of $\text{H}_3\text{PO}_4$ Doping and Pyrolysis Temperature on $\text{H}_2/\text{CO}_2$ Separation at 100 °C



# Hollow Fiber (HF) Membrane

- High membrane surface area platform
  - | The H<sub>2</sub> permeability of the PBI-based CMS membrane materials mandates ultra-thin selective layers and high surface area membrane deployment platforms
- Approach: PBI HF membranes carbonization to obtain PBI-CMS HF membranes
  - | Liquid-liquid demixing based spinning process developed to obtain high performance base PBI HF membranes with thin selective layer



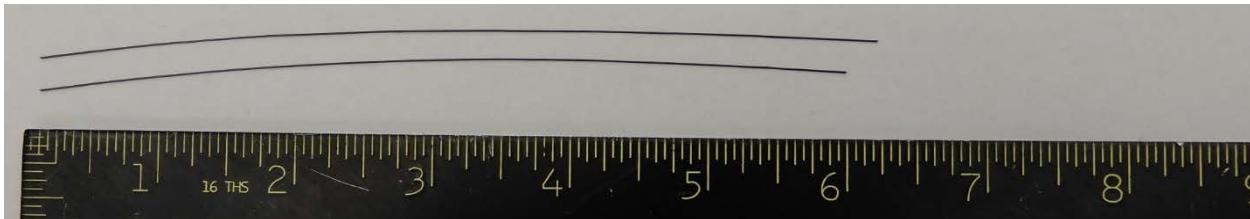
Fiber Diameter: 200 to 500  $\mu\text{m}$   
SL Thickness: 150 to 500 nm

# PBI-CMS HF Membrane Fabrication

- Goal: Development of base PBI HF membranes having variety of morphologies including selective layer thickness and support layer structure
  - | *In-situ* formation of an integrally skinned HF using commercially available PBI material leveraging LANL PBI spinning capability and process understanding
- Initiated base PBI HF membrane fabrication and pyrolysis to obtain PBI-CMS HF membranes
  - | Utilized spinning conditions known to produce PBI HF membranes with thin selective layer and porous support structure
  - | Pyrolyzed at 580 °C for 1 hour in inert atmosphere



Base PBI HFM



Carbonized PBI HFM

# Path Forward

- HF fabrication, acid doping and carbonization
  - | Fabrication optimization of PBI and acid doped PBI HF membranes to achieve defect minimized thin selective layer membranes
    - ▀ Milestone BP1 Q2FY20: Acid doped PBI HF membranes exhibiting  $H_2/CO_2 \approx 40$ .
  - | Pyrolysis condition optimization to achieve high performance PBI-CMS HF membranes with minimized porous support structure collapse during carbonization
  - | Demonstrate fabrication consistency via performance demonstration of fibers from multiple, replicate spinning campaigns
- Sealing Layer Development & Integration
  - | Develop and optimize post-carbonization defect seal layer deposition process for PBI-CMS HF membranes
- Demonstration and Validation of Developed Materials and Methods
  - | Evaluate PBI, acid doped PBI and PBI-CMS HF membranes for  $H_2/CO_2$  separation properties at syngas relevant process conditions
  - | Conduct parametric tests of PBI-CMS HF membranes in CO and H<sub>2</sub>S containing simulated syngas at process relevant conditions

# BP1 Techno-Economic Analysis (TEA)

- Cursory analysis during BP1
  - Identify key cost drivers, sensitivities for key variables
  - Identify opportunities for process optimization
  - Bottom-up analysis during BP2
- UB to provide
  - Conceptual design, process simulation and material/energy balances
  - Data on membrane permeance and selectivity
- Key data
  - Inlet pressure, membrane pressure drop
  - Compositions of product streams from membrane
  - Fate of contaminants (CO, H<sub>2</sub>S, H<sub>2</sub>O)
  - Membrane costs
- Develop preliminary CAPEX, OPEX and energy analysis
- Due 3/31/20



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# Summary

