

A Scalable Process for Upcycling Carbon Dioxide (CO₂) and Coal Combustion Residues into Construction Products

Gabriel Falzone,¹ Iman Mehdipour,¹ Gaurav Sant,¹ Brian Turk,² Raghubir Gupta²

¹ UCLA Samueli School of Engineering, University of California, Los Angeles (UCLA), Boelter Hall, 420 Westwood Plaza, Los Angeles, CA 90095, USA.
² Susteon Inc., 10 Placid Court, Durham, NC 27713, USA (<http://susteon.com/>).

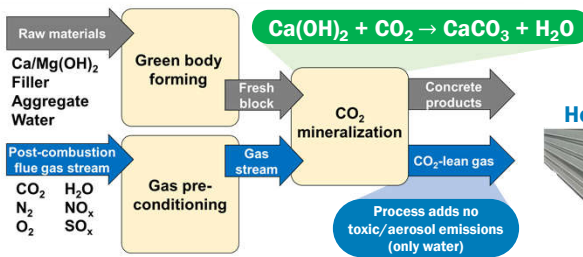
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Motivation and project objectives

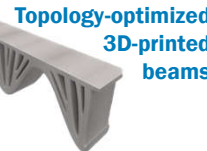
Concrete, a mixture of portland cement, aggregate, and water is indispensable in construction (> 30 billion tons produced / year).^A But nearly 1 ton of CO₂ is emitted for each ton of portland cement produced (> 4 billion tons / year),^A accounting for around 8 % of global CO₂ emissions.^B The vast concrete market provides an impactful sink for CO₂ emissions, which may be fixed within solid products by thermodynamically favorable CO₂ mineralization reactions.

- 1. Upcycle industrial wastes and CO₂** – Produce low-carbon CO₂Concrete products from coal combustion residues, flue gas CO₂, and low-grade waste heat
- 2. Design CO₂ mineralization system** – Produce data supporting heat and mass balances for design of a “bolt-on” system at coal-fired power plants
- 3. Field test system using real flue gas** – Fabricate and field test a CO₂ mineralization system to consume about 100 kg of CO₂ per day from coal flue gas

CO₂ mineralization process



Concrete masonry units



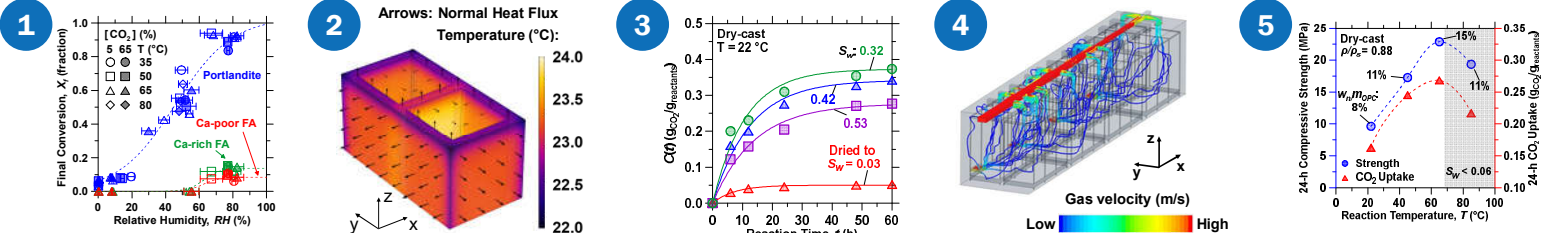
Process may be tailored to produce a variety of precast concrete/concrete masonry products Initial product is hollow-core block (concrete masonry units), which follow ASTM C90 performance criteria:

- Compressive Strength (> 13.8 MPa)
- Density (determines weight classification)
- Water absorption (< 320 kg/m³)
- Physical appearance/dimensional tolerances

Preliminary lifecycle analysis (LCA) indicates ~ 65 % CO₂ emissions reduction relative to conventional CMU

- Portlandite (hydrated lime, Ca(OH)₂) is a highly efficient reactant for CO₂ mineralization (CO₂ uptake 0.59 g/g) that is also abundant and near cost-equivalent to cement
- Fillers may include coal combustion residues (CCRs) such as ASTM C618 non-compliant fly ashes, which are not typically usable in concrete mixtures
- “Green bodies” are non-strengthened, but shape-stable components that may have their surfaces exposed to flue gas to promote CO₂ mineralization reactions
- Flue gas pre-conditioning is limited to changing the temperature and/or relative humidity of the gas stream – CO₂ enrichment/capture or pressurization are unneeded

Reaction kinetics, heat/mass transfer, and component strength



Conversion limits and kinetics of CO₂ mineralization reactions:

Process design is informed by data describing the CO₂ uptake of alkaline solids (e.g., Ca(OH)₂ and fly ashes (FA)) in contact with simulated flue gases of varying temperature, relative humidity (RH), and CO₂ concentration (near atmospheric pressure).^{C,D,E}

Heat generation and transfer in CO₂Concrete components:

Exothermic portlandite carbonation and cement hydration reactions generate heat that contributes to temperature rise and vaporization of water. A finite element model (FEM) is being developed to predict gradients in block properties that may result.

Effects of microstructure and pore saturation on carbonation:

The liquid water saturation (S_w) in porous cementing microstructures influences the rate and extent of CO₂ uptake. Reducing S_w increases CO₂ uptake until a critical limit of ≈ 0.1 is reached, below which carbonation is water-limited.^F

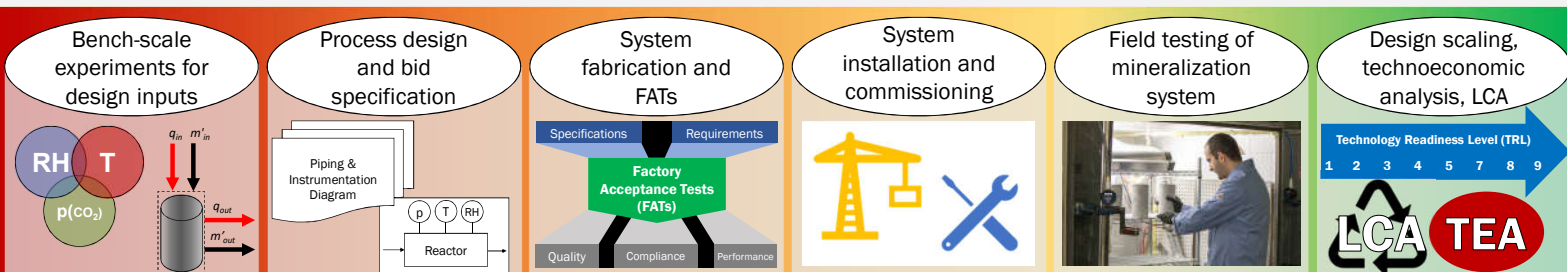
Computational fluid dynamics (CFD) modeling for reactor design:

The design of flue gas handling and distribution equipment within the CO₂ mineralization / curing chamber is informed by CFD simulations. In this way, the effects of various shelving and block arrangements on flow uniformity may be evaluated.

Compressive strength performance of dry-cast CO₂Concrete products:

The compressive strength of concrete is a critical performance metric. CO₂Concrete formulations exceed the strength requirements of relevant product standards (e.g., ASTM C90) immediately after processing by optimizing process conditions.^F

Project overview and milestones



References

A. Mehta, P. K.; Monteiro, P. J. *Concrete: Microstructure, Properties, and Materials*, 3rd ed.; McGraw-Hill Education, 2014.
B. Lehne, J.; Preston, F. Making Concrete Change: Innovation in Low-Carbon Cement and Concrete; Chatham House, The Royal Institute of International Affairs: London, UK, 2018; p 138
C. Vance, K.; et al. Direct Carbonation of Ca(OH)₂ Using Liquid and Supercritical CO₂: Implications for Carbon-Neutral Cementation. *Ind. Chem. Res.* **2015**, 54 (36), 8908–8918.
D. Wei, Z.; et al. Clinkering-Free Cementation by Fly Ash Carbonation. *J. CO₂ Util.* **2018**, 23, 117–127.
E. Falzone, G.; et al. The CO₂ Mineralization Kinetics and Conversion Limits of Alkaline Solid Reactants/Monoliths (Manuscript in preparation), 2019.
F. Mehdipour, I.; et al. How Microstructure and Pore Moisture Affect Strength Gain in Portlandite-Enriched Composites That Mineralize CO₂. *ACS Sustainable Chem. Eng.* **2019**, 7 (15), 13053–13061.