

Electro-Driven Microbial CO₂ Utilization

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NETL – Research & Innovation Center



U.S. Department of Energy

National Energy Technology Laboratory

Mastering the Subsurface Through Technology Innovation, Partnerships and Collaboration:
Carbon Storage and Oil and Natural Gas Technologies Review Meeting

August 29, 2019

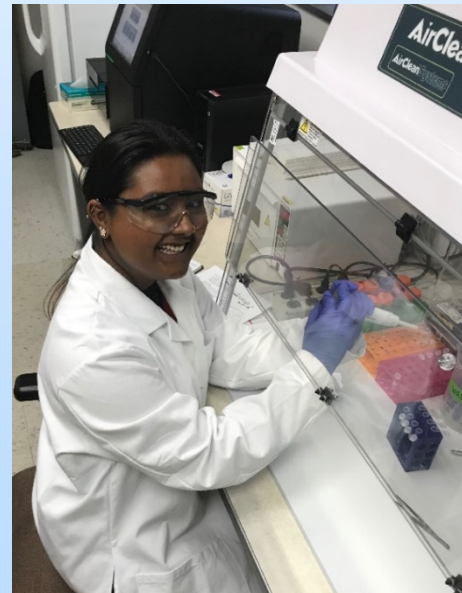
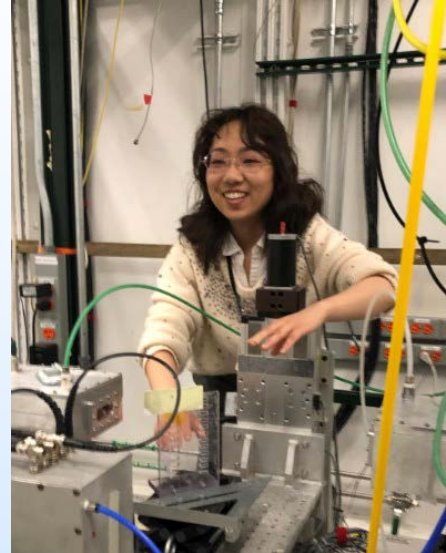


Presentation Outline

- Overview FY17
 - Research Team
 - Task Goals and Structure
- Technical Status
- Lessons Learned
- Project Summary
- Synergistic Opportunities
- Accomplishments to Date

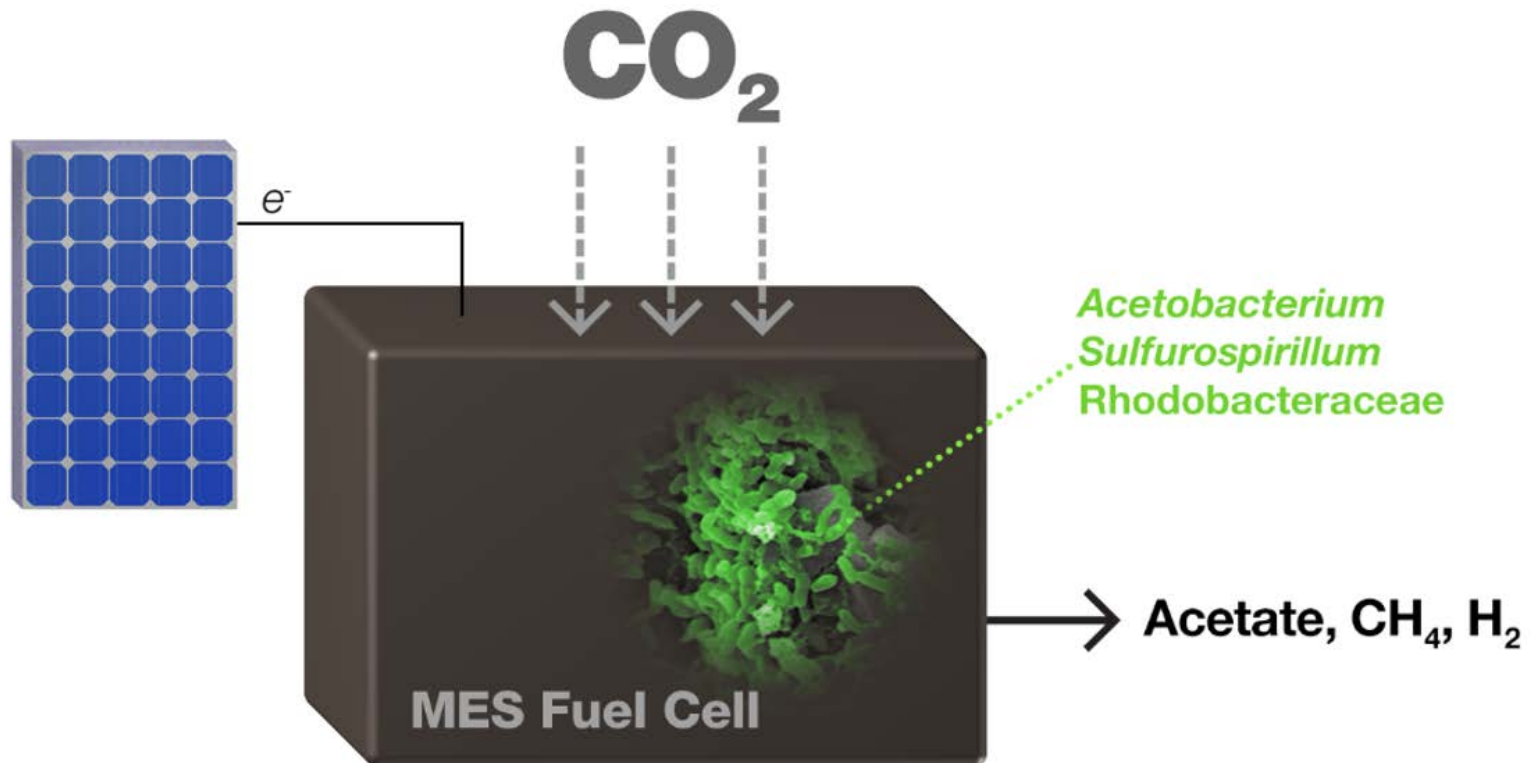
FY2019 Research Team

- Dan Ross, Leidos-RIC
- Mengling Stuckman, Leidos-NETL
- Chris Marshall, Marquette University
- Kara Tinker, ORISE
- Preom Sarkar, ORISE
- Joe Moore, ORISE
- Ashley Miles, ORISE



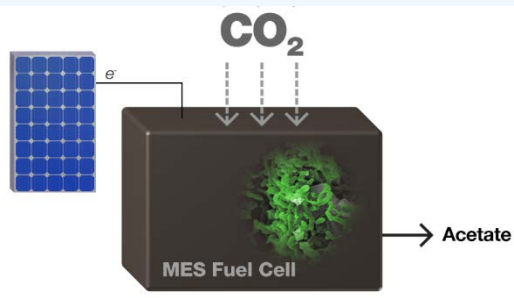
Task 3 (FY19): Biological CO₂ Utilization Goals

Grow a novel microbial community capable of converting CO₂ to value-added products. This microbial community (the biocatalyst) will be used to (1) optimize CO₂-to-acetate conversion rates, and (2) further upgrade CO₂-derived intermediate products such as acetate, carbon monoxide, and formate.



Task 3 (FY19): Biological CO₂ Utilization Goals

MES



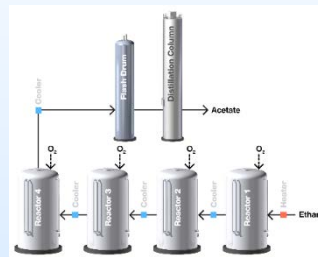
Cost: \$1.79/kg

Methanol Carbonylation



Cost: \$0.32/kg

Ethane Direct Oxidation



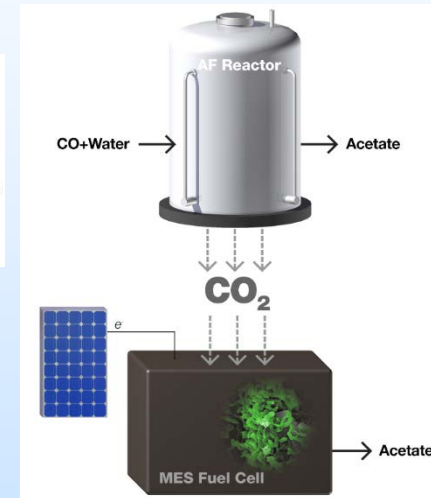
Cost: \$0.14/kg

AF



Cost: \$5.13/kg

AF+MES



Cost: \$0.30/kg

Acetic Acid selling price: \$0.60/kg

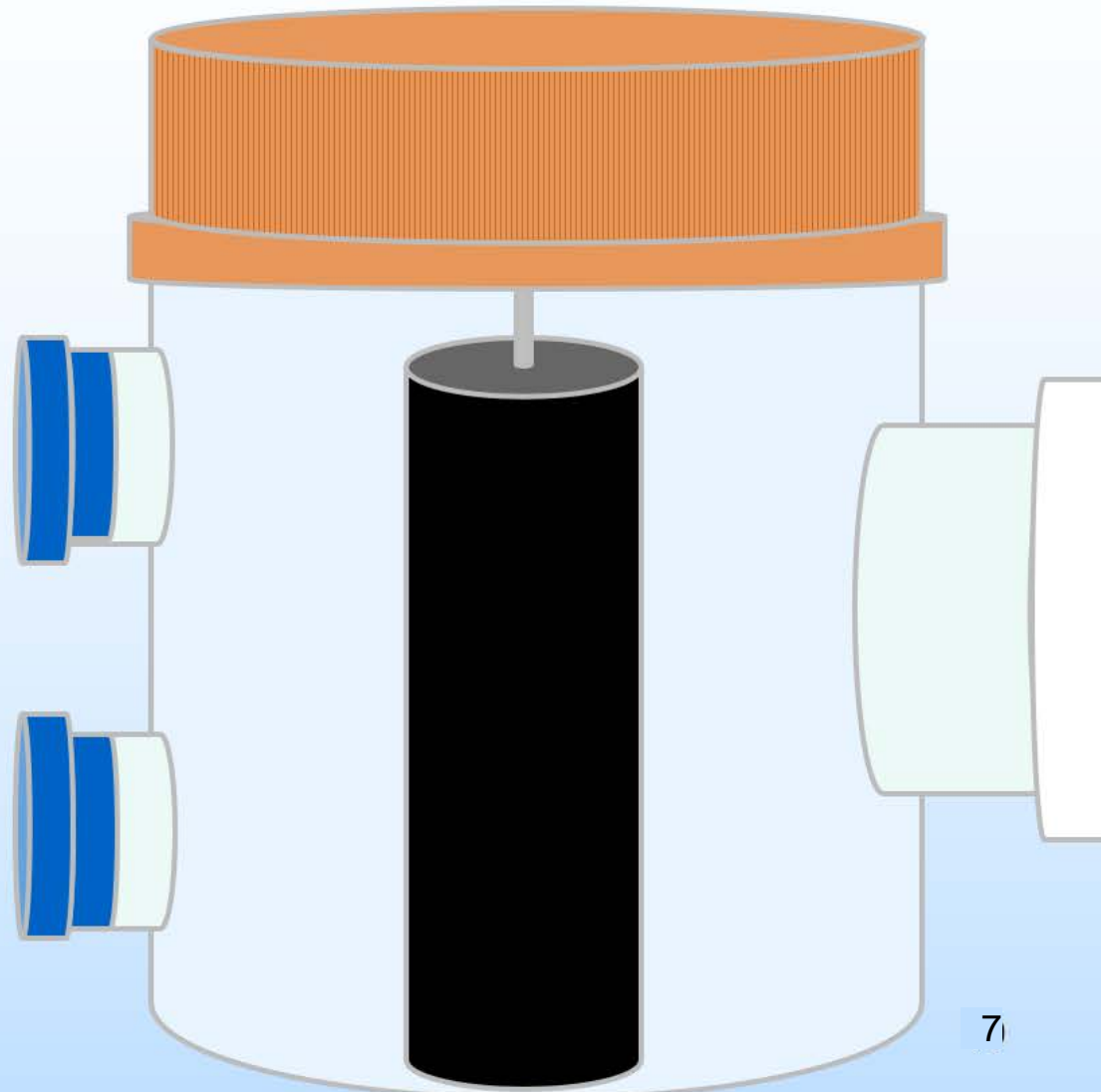
Takeaway Points

- MES may already be cost competitive if designed to be integrated in other systems
- Stand alone MES will be more competitive with increased acetic acid production or with other value-added commodity production
- MES may be more efficient at CO₂ conversion with a concentrated CO₂ input stream compared to photosynthetic processes

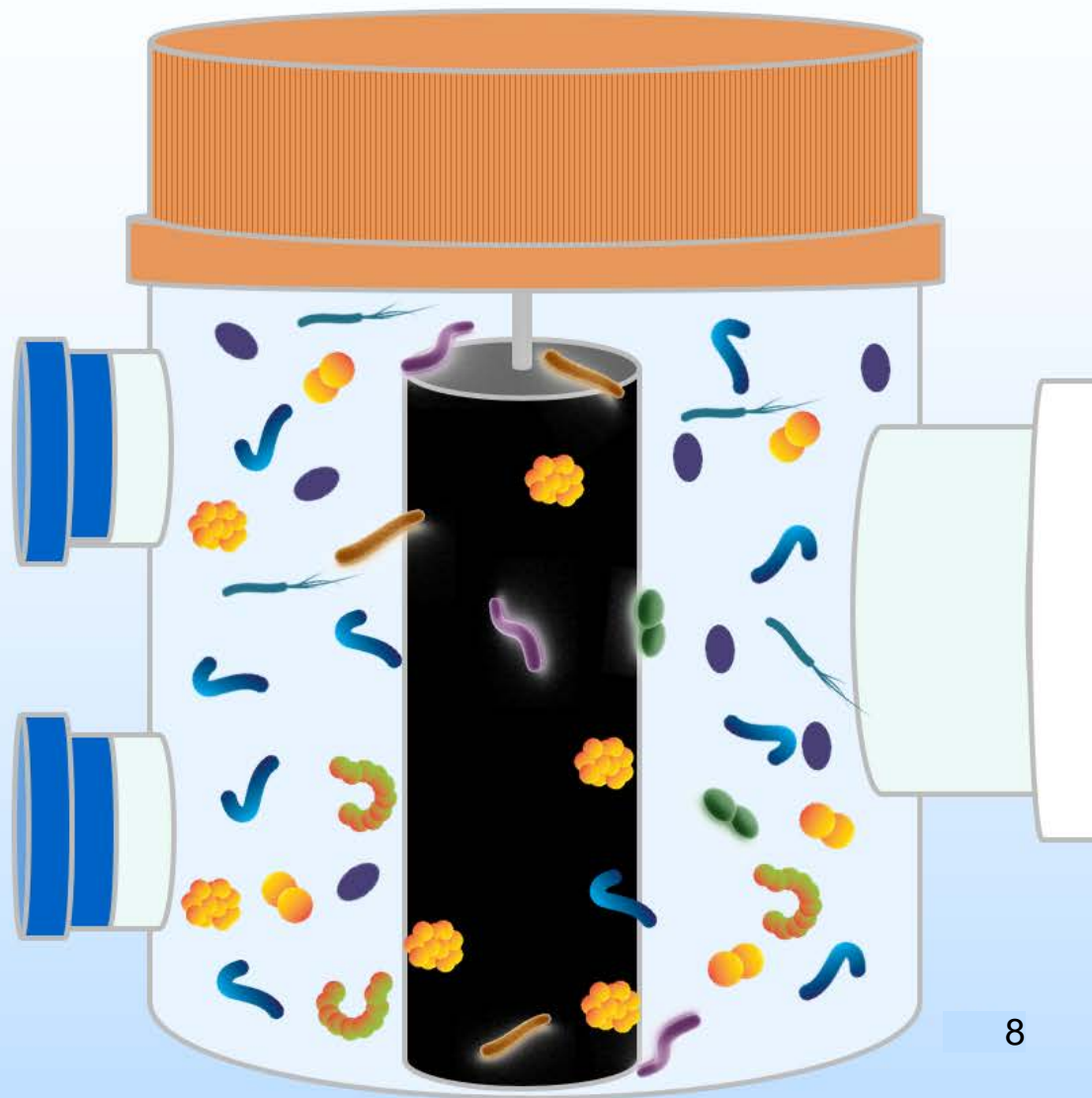
Task 3 (FY19): Biological CO₂ Utilization Goals

- **Subtask 3.1 Biological Electro-Synthesis Technologies (BEST)**
 - Growth of biocatalyst from new samples capable of 50% CO₂ and H₂ conversion to acetic acid.
 - Demonstrate CO₂ conversion with newly grown biocatalyst in bench-scale electrochemical reactor
- **Subtask 3.2 Biological Upgrading of CO₂ Derived Chemical Intermediates**
 - Biocatalyze CO₂-derived intermediates, such as acetate and carbon monoxide, into longer chain carbon molecules such as ethanol, butyrate, and/or butanol.

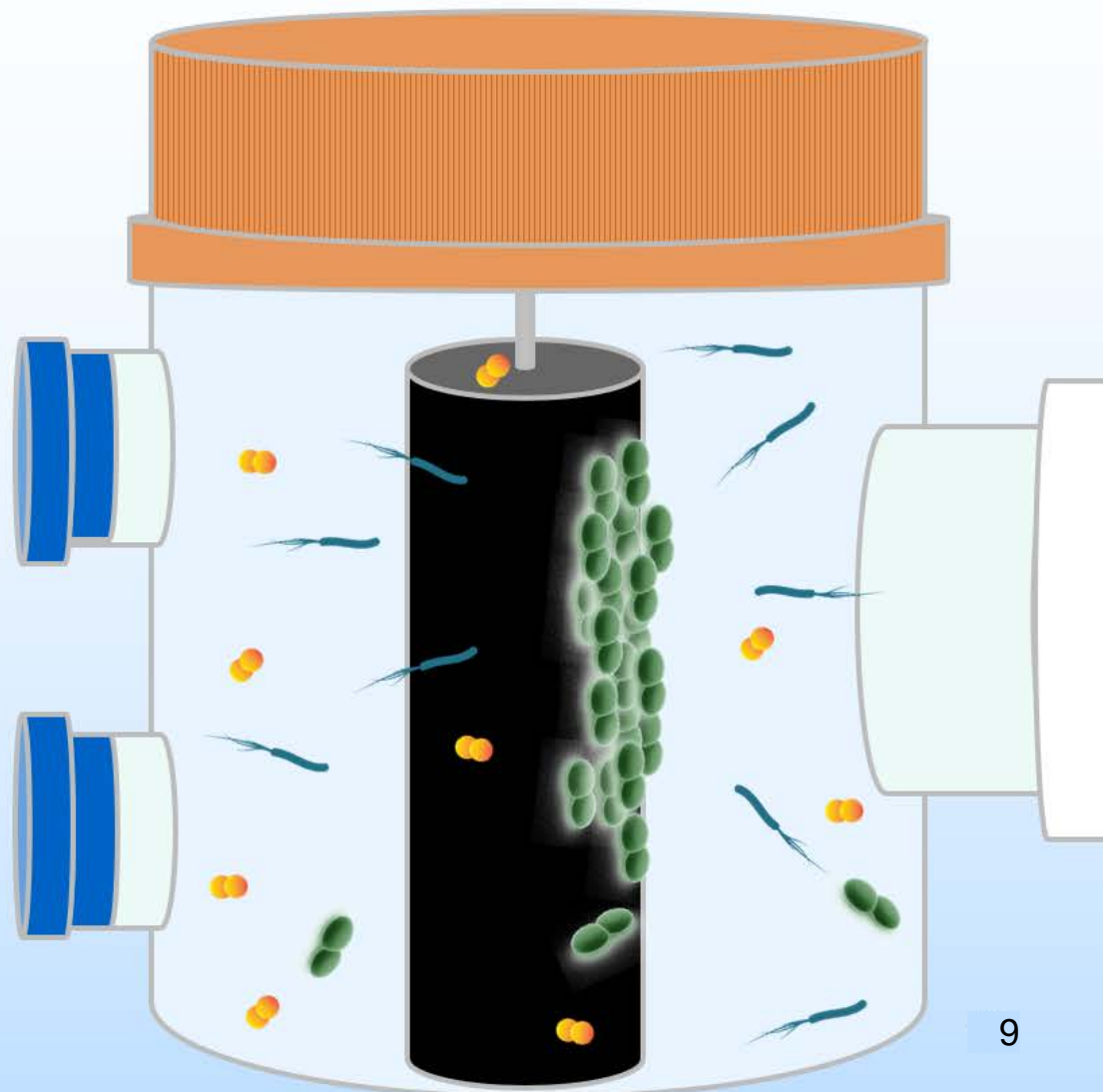
Background



Background



Background

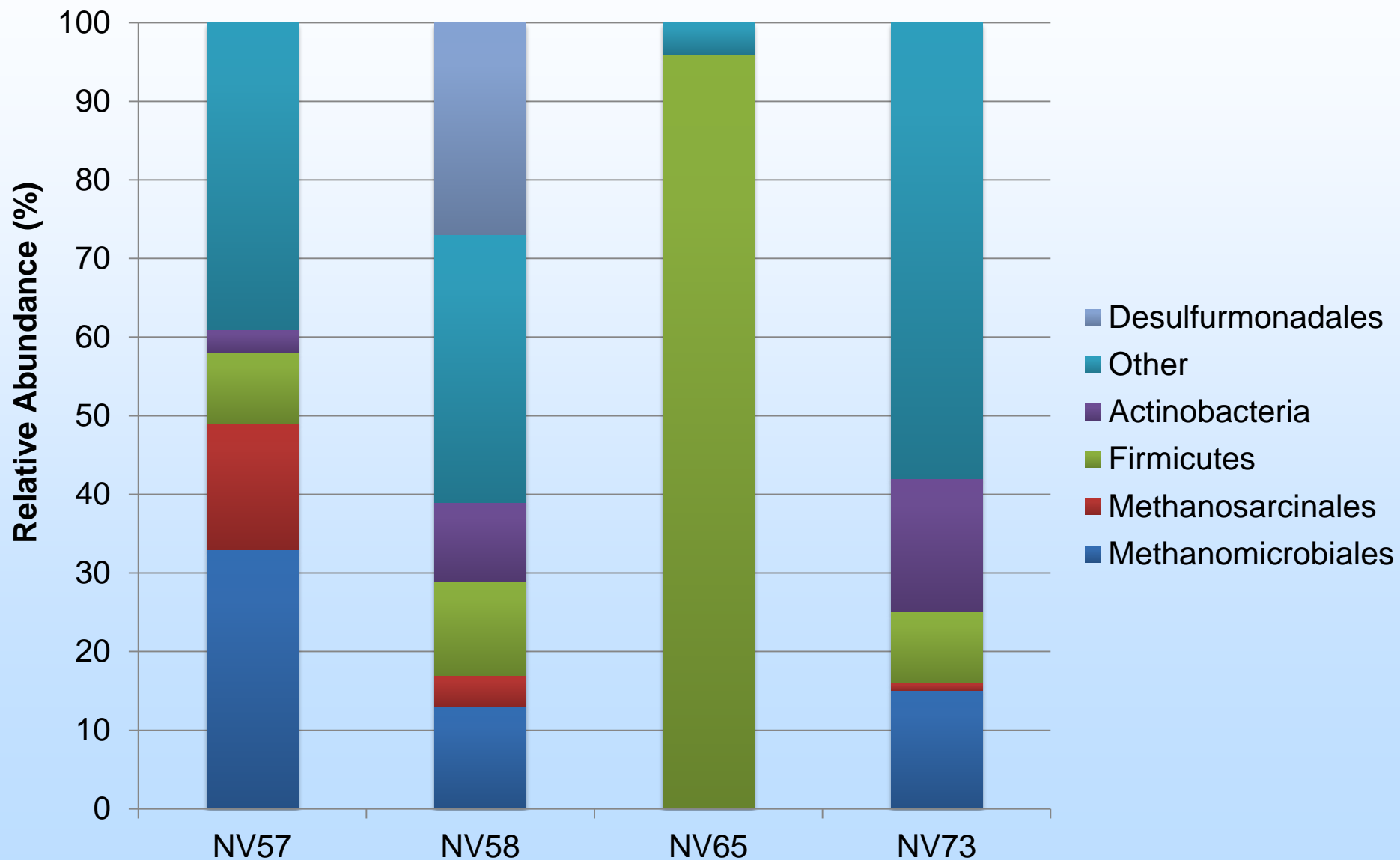


Technical Status

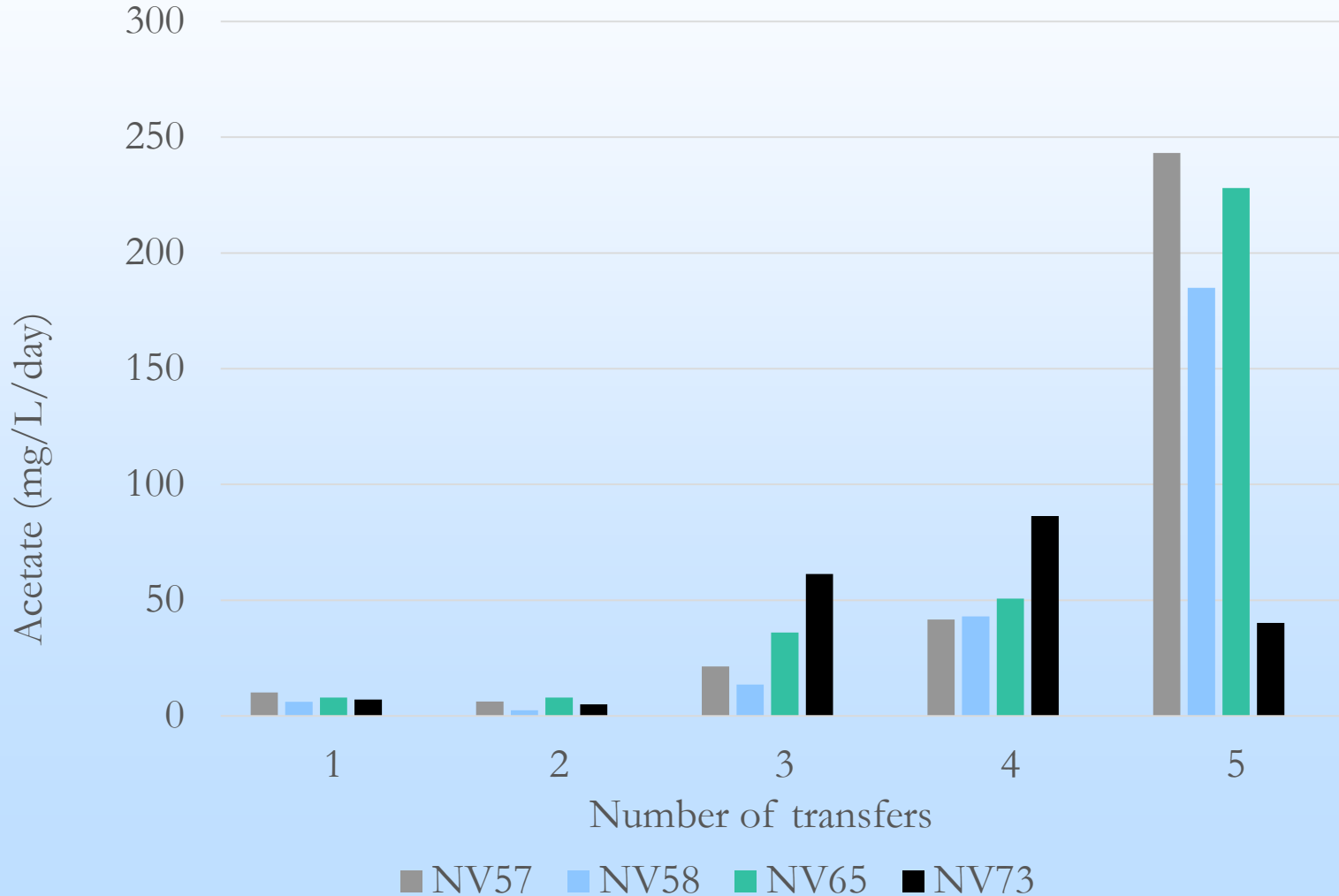
Initial microbial communities were recovered from unconventional oil and gas reservoirs and coalbeds

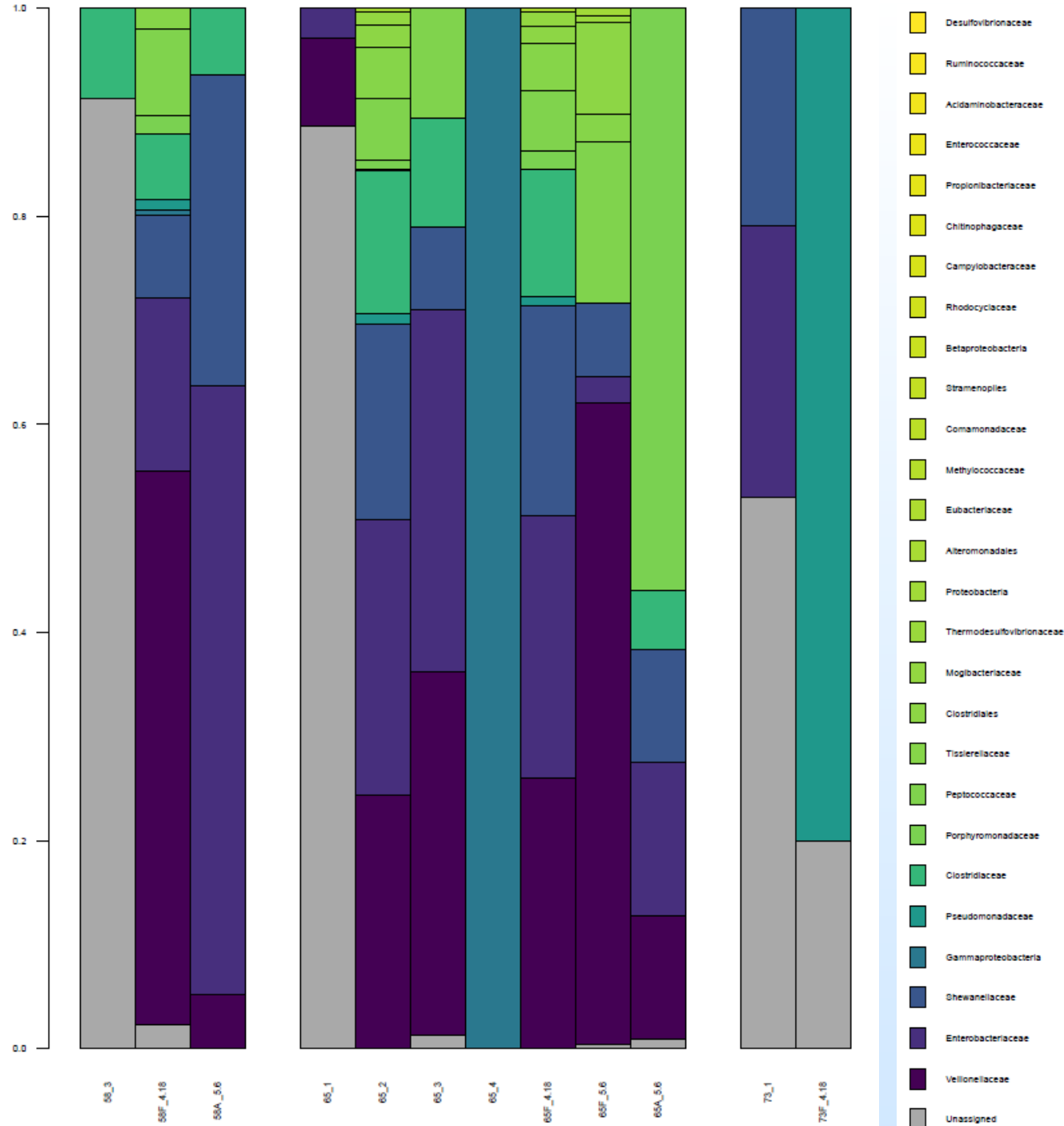


Technical Status

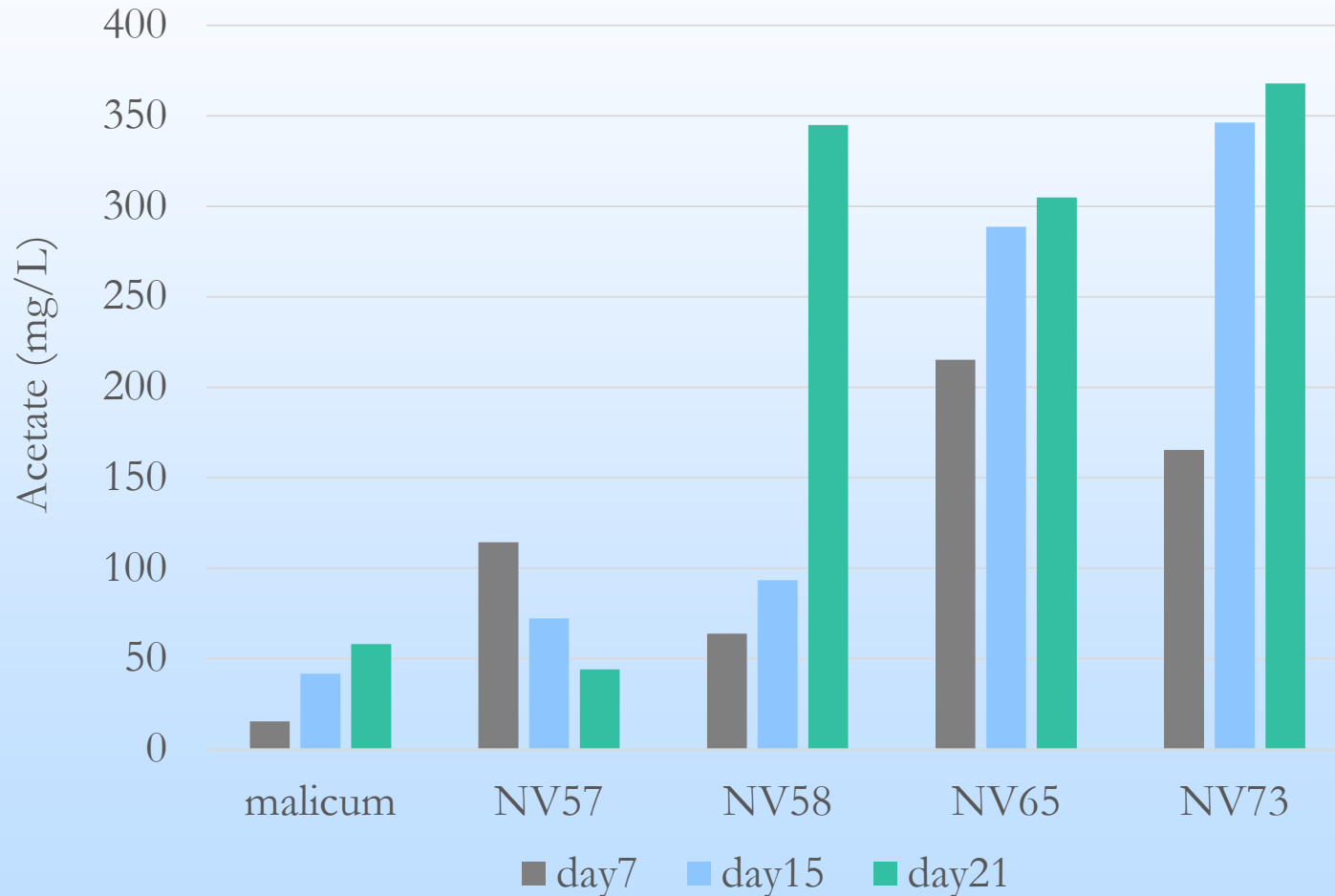


Technical Status: CO₂/H₂

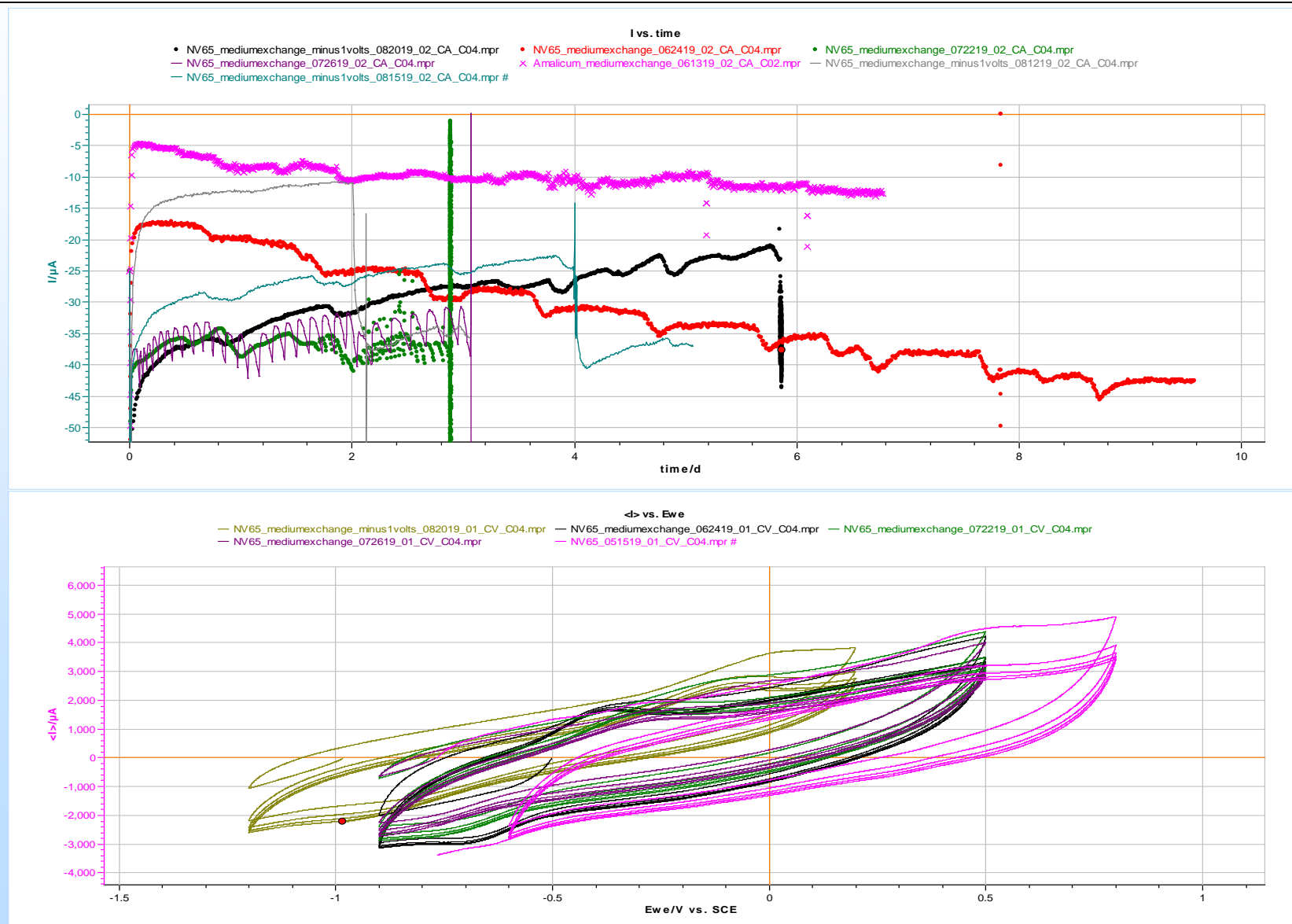




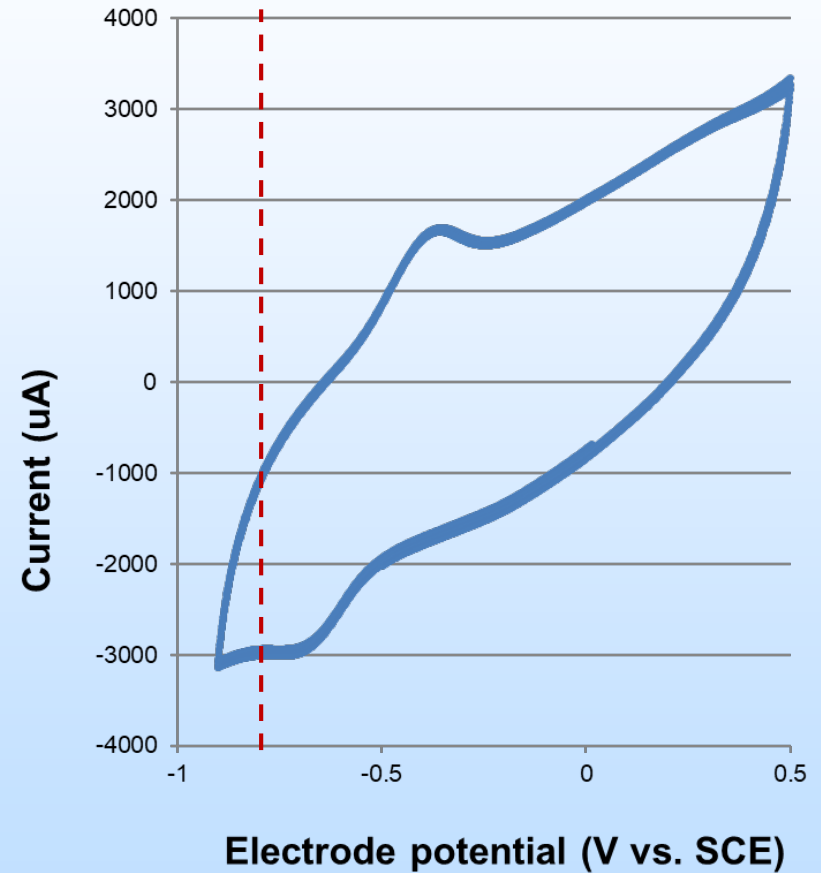
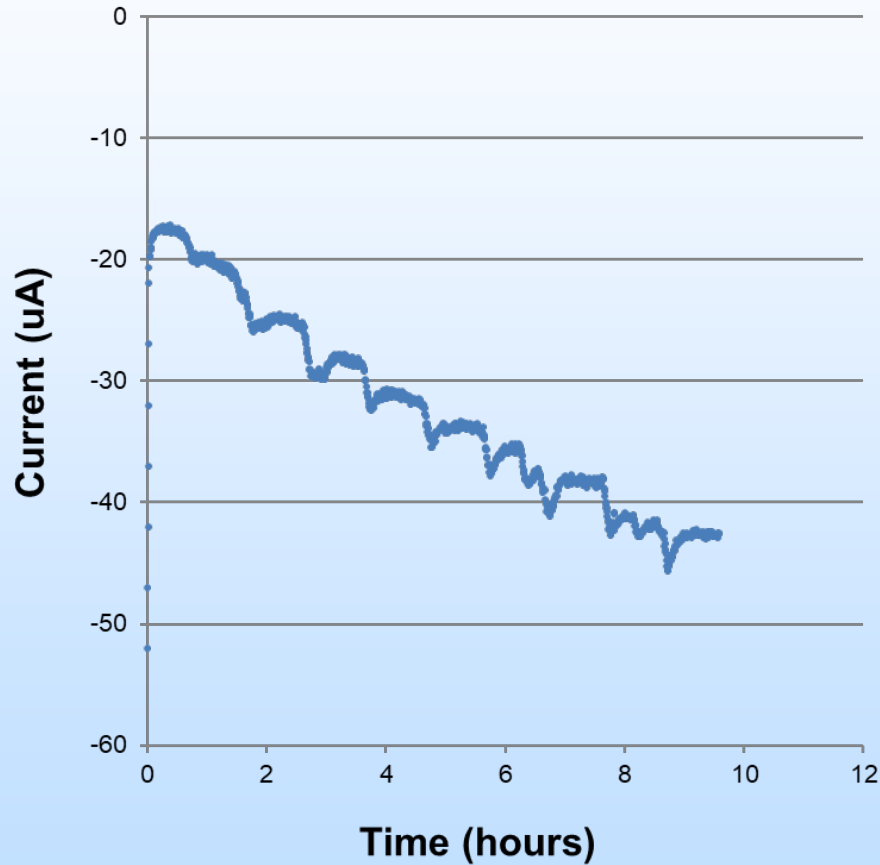
Technical Status: Electrochemical Reactor



Technical Status: Electrochemical Reactor



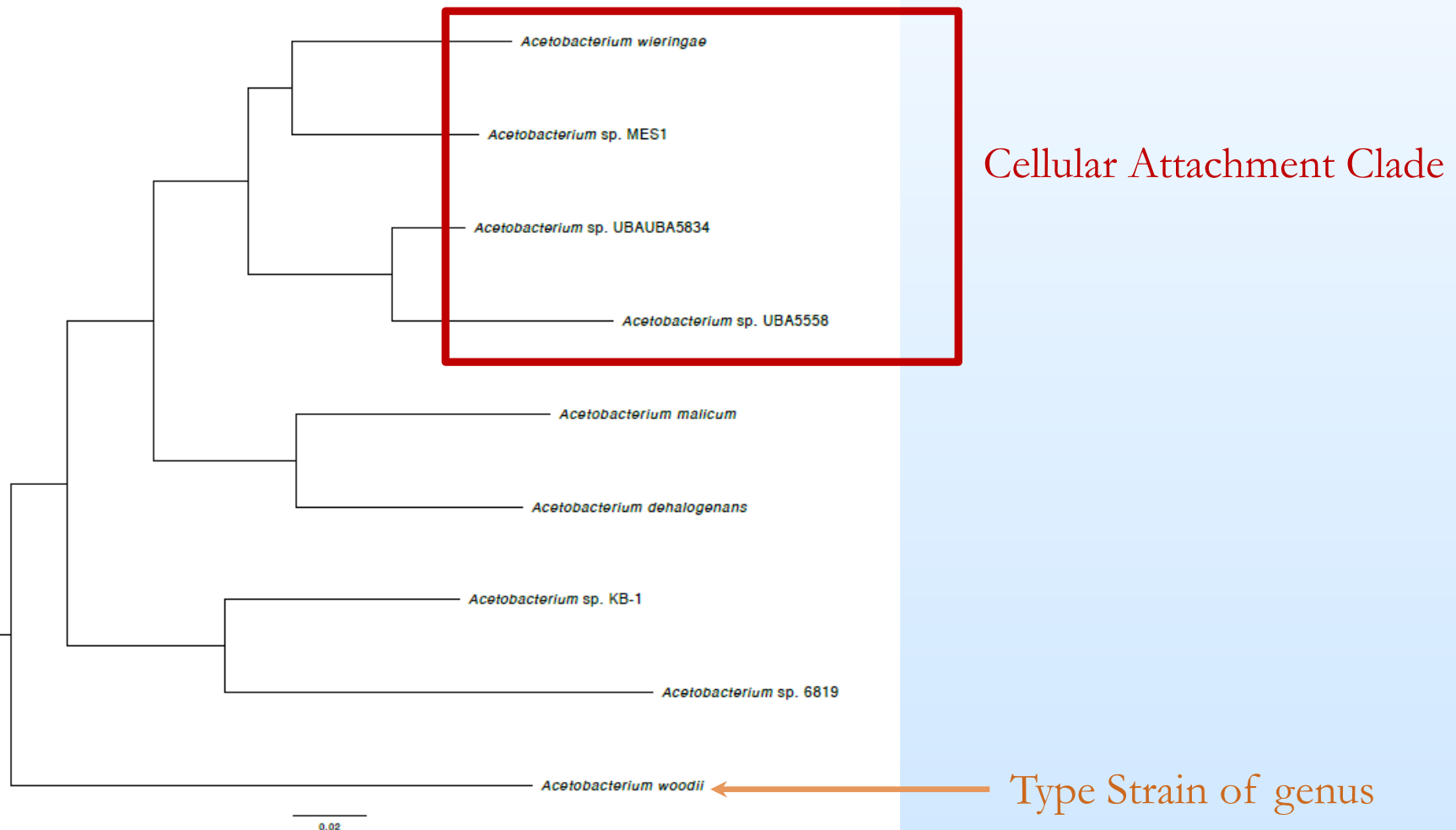
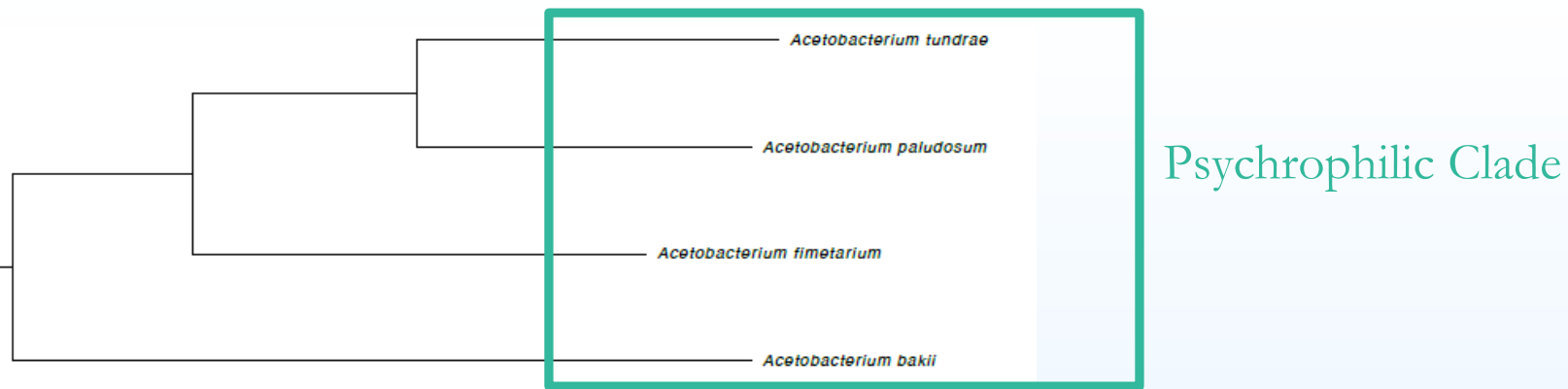
Technical Status: Electrochemical Reactor



Technical Status: Acetobacterium

malicum	dehalogenans	UBA8819	KB-1	MES1	wieringae	UBA5834	UBA5558	finetarium	bakii	woodii	tundrae	paludosum	AAI
	96	85	86	87	87	73	72	74	74	79	74	74	<i>Acetobacterium malicum</i>
97		85	86	87	87	87	86	74	74	79	74	74	<i>Acetobacterium dehalogenans</i>
84	84		99	83	83	83	81	73	72	78	72	72	<i>Acetobacterium</i> sp. UBA8819
84	83	99		84	84	83	82	73	74	79	73	73	<i>Acetobacterium</i> sp. KB-1
84	84	81	82		97	97	96	74	73	78	73	73	<i>Acetobacterium</i> sp. MES1
84	84	81	82	97		97	96	73	73	77	73	73	<i>Acetobacterium wieringae</i>
84	84	81	81	97	97		100	74	73	78	73	73	<i>Acetobacterium</i> sp. UBA5834
84	84	81	81	97	98	100		72	72	77	72	72	<i>Acetobacterium</i> sp. UBA5558
80	79	79	78	78	79	78	78		72	73	80	81	<i>Acetobacterium finetarium</i>
80	78	79	80	77	78	77	78	80		72	73	73	<i>Acetobacterium bakii</i>
80	80	79	80	79	79	79	79	78	78		74	74	<i>Acetobacterium woodii</i>
78	78	78	78	77	77	77	77	81	79	80		94	<i>Acetobacterium tundrae</i>
78	78	78	78	78	78	77	77	81	79	78	95		<i>Acetobacterium paludosum</i>

ANI



Lessons Learned

- Each biocatalyst is a dynamic system
 - Optimal reactor conditions will be adjusted until equilibrium is reached
 - Sequencing difficulties due to reduced diversity
- Voltammetry does not correspond to acetate production
 - Might be due to production of other constituents
 - CH_4 measurements
 - Might be due to H_2 production
- Early biocatalyst “contenders” are often surpassed by “dark-horses”.



Project Summary

- **Subtask 3.1 Biological Electro-Synthesis Technologies (BEST)**
 - Biocatalyst enriched on $\text{H}_2:\text{CO}_2$
 - Coalbed microbial communities were most easily enriched on CO_2
 - Highest acetate concentration was 250 mg/L, reaching approximately 52% bioconversion from CO_2 .
 - One culture produced butyrate
 - Bioelectrochemical reactors poised at -560 mV vs SHE and produced up to 345 mg/L of acetate.
 - Acetobacteria pan-genome comparison demonstrate conserved Wood Ljungdahl pathway, 2 unique clades, and divergence from type strain
- **Subtask 3.2 Biological Upgrading of CO_2 Derived Chemical Intermediates**
 - Permit modification and ES&H approval completed

Project Summary

Next Steps

- Biocatalysts will continue to develop
- Minor adjustments with potential, electrode, and media
- Reactor optimization
- Taxonomic and metagenomic characterization of biocatalyst
- SEM imaging of electrodes

Synergy Opportunities

- Field sampling efforts
- Comparison with other related subsurface reactions
 - Unconventional resource systems
 - Coal systems
 - Carbon storage systems
- Electrochemistry expertise within NETL
- Communication with outside research groups
- TEA and LCA with Systems Engineering and Analysis
- Integration with Carbon Capture
- Communication with renewable energy experts

FY19 Accomplishments to Date

- FY19 Milestone: Growth of biocatalyst from new samples capable of 50 percent CO₂ and H₂ conversion to acetic acid.
 - Completed. Growth of a new biocatalyst from coalbed fluid was capable of 52% CO₂ conversion to acetic acid
- FY19 Milestone: Demonstrate bio-electrochemical CO₂ conversion with newly grown biocatalyst in bench-scale electrochemical reactor at a faradaic efficiency of at least 20 percent.
 - Ongoing
- FY19 Milestone: Growth of a biocatalyst in the electrochemical reactor with more than 50% faradaic efficiency.
 - Ongoing