Characterizing Shales as Seals for CO₂ Containment and Shales as Reservoirs for Geologic Storage of CO₂

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U.S. Department of Energy

National Energy Technology Laboratory Mastering the Subsurface Through Technology Innovation, Partnerships and Collaboration: Carbon Storage and Oil and Natural Gas Technologies Review Meeting

August 13-16, 2018

Tuesday, August 14th 2:30 PM

Presentation Outline

Research Theme: Shales, Characterizing Interactions of CO₂ with Shale for Potential Storage Needs and Seal Integrity

• Task 17.0 CO₂-Shale Interactions- Macroscopic

• Team Members: Johnathan Moore, Sarah Brown, Ernest Lindner, Michael Hannon, Yael Tucker, Leebyn Chong, Eugene Myshakin, Guanyi Lu, Andrew Bunger, Dustin Crandall

• Task 18.0 CO₂-Shale Interactions- Microscopic

• Team Members: Angela Goodman, Sean Sanguinito, Barbara Kutchko, Jeff Culp, Mary Tkach, Sittichai Natesakhawat, Dustin Crandall, Patricia Madden

Shale Response to CO₂ Literature Review

- Over 350 pages of detailed reviews and over 300 references
 - Lindner, E. N. Review of the Effects of CO₂ on Very-Fine-Grained Sedimentary Rock/Shale – Part III: Shale Response to CO₂; NETL-TRS-11-2017; NETL Technical Report Series; U.S. Department of Energy, National Energy Technology Laboratory: Morgantown, WV, 2017; p 228.
 - Lindner, E. N. Review of the Effects of CO₂ on Very-Fine-Grained Sedimentary Rock/Shale – Part II: Clay Mineral & Shale Response to Hydration; NETL-TRS-10-2016; NETL Technical Report Series; U.S. Department of Energy, National Energy Technology Laboratory: Morgantown, WV, 2016; p 68.
 - Lindner, E. N. Review of the Effects of CO₂ on Very-Fine-Grained Sedimentary Rock/Shale – Part I: Problem Definition; NETL-TRS-1-2016; NETL Technical Report Series; U.S. Department of Energy, National Energy Technology Laboratory: Morgantown, WV, 2016; p 68.



Shale swelling due to CO₂

- Inconclusive experiments have led to MD simulations





Representation of the clay/CH₄/CO₂ system. The color coding of atoms is purple, red, white, gray, yellow, pink, and green for K, O, H, C, Si, Al, and Mg, respectively. Chong, L., Myshakin, E.M. (2018) Molecular simulations of competitive adsorption of carbon dioxide ⁴ – methane mixture on illitic clay surfaces, Fluid Phase Equilibria, Vol 472, 2018

Full Immersion Pressure Pulse Decay

- Apply pressure disturbance to entire outer surface area.
- Assume uniform but distinct permeabilities in radial and axial directions.
- Numerical simulation to predict pressure response
- Experimental tests have shown excellent agreement with traditional measurements.
- Simultaneously determine both permeabilities from a single test!



Stress Corrosion in Shales

- The goal of the proposed research is to observe and quantify the impact of stress corrosion on shale permeability.
 - Stress corrosion is the time dependent failure due to stresses that are insufficient to instantaneously break the rock.
- Development of a coupled numerical model to account for this behavior, benchmarked on experimental observations, will enable greater seal characterization for long term stress behavior of shale sealing formations.
- Just started experiments in July.







Task Technical Approach and Project Relevancy

- <u>Objective</u>: Quantify CO₂ interactions with shale at the nano- and micro-scale and provide quantitative inputs at the macro-scale to understand flow properties and inputs for reservoir simulations of large-scale storage activities in shale.
 - <u>Question:</u> How does CO₂ interact with fracturing fluid and shale for storage, as an alternative fracturing agent, and as an EOR agent?



Fig. 2. —A comparison of quartz and feldspar, carbonate content, and clay conte References for data are provided in Table 2.

 Method: Examining petrophysical characteristics including <u>reaction</u> <u>mechanism, precipitation, dissolution, surface area, porosity, permeability,</u> <u>and mineralogy</u> of the host formation

Results: Infrared Spectroscopy of the CO₂-Shale Interface



Results: Infrared Spectroscopy of the CO₂-Fluid-Shale Interface

- Infrared frequencies consistent with CO₂ dissolution in water in high pressure geological storage conditions
- <u>CO₂ not interacting with Shale, Kerogen or Clays</u>

Sample	Infrared Frequency (cm ⁻¹) Wet	Spectral Assignment	
Marcellus Shale: Oatka Creek Member	2342	CO ₂ dissolution	
Marcellus Shale: Union Springs Member	2342	CO ₂ dissolution	
Utica Shale: Flat Creek Member	2342	CO ₂ dissolution)is
Illite-Smectite	2342	CO ₂ dissolution	solu
Illite	2342	CO ₂ dissolution	utio
Kaolinite	2342	CO ₂ dissolution	Ď
Chlorite	2342	CO ₂ dissolution	
Extracted kerogen from the New Albany Shale	2342	CO ₂ dissolution	



Utica Shale (US-1)

carbonate CO₂-Shale 874 6.9 MPa A _{0.8}. -5.7 MPa Overall - intensity of the carbonate b 4.2 MPa • 2.9 MPa bands increased with pressure -1.4 MPa 0 0.6 0 MPa Buffering observed ullet1440 1420 1400 b 712 a _{0.4} n **Adsorbed Water** с Utica-1 e 0.2 carbonate Carbonate 6 1.4 MPA Utica Shale (US-1) 712 Elapsed Time 1400 1000 800 1600 1200Wavenumber (cm⁻¹) 0 Mir 1424 2 Days 8 Days S Water 0 12 Days 15 Davs 874 r 26 Days h 35 Days а CO₂-Fluid-Shale n с е Indicate carbonate formation and 1760 1600 1440 1280 \bullet 0.5 dissolution and mineral dissolution $V_{3}(CO_{2})$ 2343 **Buffering occurring** ullet3600 2000 1600 1200 800 400 CO₂ Dissolution Wavenumber (cm⁻¹)

Results: Infrared Spectroscopy of Changes in Carbonate Chemistry

Results: Scanning Electron Microscopy of Utica Shale

BEFORE

- Long-term effects of carbon dioxide and fluid remaining in the subsurface
- Changes in porosity are observed after reaction with CO₂ and fluid

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Quantifying dry supercritical CO₂-induced changes of the Utica Shale Sien Sanguinto¹⁰, Angela Goodman,⁴ Mary Tsach², Barbara Kuthao¹, Jeffrey Culg²⁺¹, Sittschai Sateschhward¹, Jim Fazdo¹⁰, Itsis Fuka¹¹,² Dustin Crandal¹¹ Tada and Spear et al. (State State States) and and an analysis of the State States and and and the states of the state States and and an analysis of the State States and States and States and States (States and States) and and and states and states and states (States and States) and and states and states and states (States and States) and and states and states and states (States) and and states and states



Potential to alter flow pathways and impact hydrocarbon production

Results: BET surface area and pore size analysis of Utica Shale

		Unexposed	CO2-exposed	Wet CO2-exposed
Sample amount (g)		0.3030	0.3042	0.3091
BET surface area (m ² /g)		6.8	6.3	5.7
Micropore surface area (m ² /g)		1.4	0.4	0.0
Micropore volume from N2-77K (cm ³	/g)	0.00071	0.00030	0.00000
Ultramicropore volume from CO2-273K	cm ³ /g)	0.00127	0.00141	0.00127
Total micropore volume (cm ³ /g)		0.00197	0.00171	0.00127
Mesopore volume (cm ³ /g)		0.01177	0.01239	0.01539
Fitting error for N2-77K (%)		0.570	0.978	1.553
Fitting error for CO2-273K (%)		0.179	0.257	0.255
Micropore surface area and volume decreased significantly after CO ₂ exposure, possibly altering overall permeability and fracture networks	dV(d) (cm ³ /nm/g)	.015 .010 .005 .000 0.1	Unexposed US Dry CO ₂ - Wet 1.0	-1 exposed US-1 t CO ₂ exposed US-1 A 10.0
			Pore Width	n (nm)

100.0

Results: Quantitative CO₂ Adsorption isotherms for Utica Shale

- Isotherms are similar to the BET isotherms
- Filling of micro-pores at low pressure followed by meso-pores
 - Sharp initial uptake of ~ 1 cm³/g CO₂ in the lower P/P₀ range of the isotherm followed by a shallow positive slope over the remainder of the isotherm.
- Individual components show greater uptake than shale
 - Constituents inaccessible even when finely ground?
- Significant uptake of CO₂ by kerogen
 - Higher fraction of micropores favorable for CO₂ interaction

Fuel 226 (2015) 54-64		
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ELSEVIER	journal homepage: www.elsevier.com/locate/fuel	
Full Length Article		
Quantifying dry s	upercritical CO2-induced changes of the Utica Shale	
Sean Sanguinito ^{a,b,*} , A Sittichai Natesakhawa	ngela Goodman ^a , Mary Tkach ^a , Barbara Kutchko ^a , Jeffrey Culp ^{a,b} , ^{Pa,c} , Jim Fazio ^{a,b} , Isis Fukai ^{a,d} , Dustin Crandall ^a	Queca lair opticidae
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Accomplishments to Date

– Task 17.0 CO₂-Shale Interactions- Macroscopic

- Publication of detailed literature review of CO₂/shale interactions to guide future research.
- Fundamental understanding of swelling behavior examined from MD simulations and laboratory experiments.
- Development of faster techniques to measure permeability of low permeability shales.
- Examine and upscale pressure corrosion relationships relevant to shales as CO₂ seals.

– Task 18.0 CO₂-Shale Interactions- Microscopic

- Quantified interactions between CO₂, and fluid with Utica and Marcellus Shale.
 (Sanguinito et al, *Quantifying dry supercritical CO₂-induced changes of the Utica Shale*, *Fuel*, 2018, 54-64)
- Beginning measurements to understand changes in dynamic conditions.

Lessons Learned

- Some take away points from the literature review -

1 – Mineralogy of shale/mudstone is critical to understanding interactions with CO_2 . Clay content, clay type and organic content should be measured when testing.

2 – Chemical interactions vary with time and occur over long-durations. We're not doing a good job on long term testing in the laboratory.

3 – Water content influences sorption and swell. Laboratory content and *in-situ* content should be understood to provide context to presented results.

4 – Micro-fabric of shale influences and controls CO₂ reactions. Laboratory testing will always incur some disturbance; understanding and minimizing this is important.

5 – Lab-scale swell tests with CO_2 indicate minor deformation (a couple of percent), but clay mineral layer spacing swell is shown up to 60%. More work linking these scales is required over longer exposure periods.

 $6 - CO_2$ sorption in shales does not follow typical Langmuir or BET isotherms at higher pressures (>5 MPa). Increased sorption up to pressures ~ 7 MPa, but decreases substantially after ~10 MPa.

7 – Fracture dynamics with CO₂ complex and poorly understood. Channeling, dissolution, and the impact of surface alteration affect flow properties. Flow velocity and exposure duration play an important role in this response.

8 – Near-well vs. far-field environments drastically different. Dry out may occur near to injection point and induce shrinkage. Carbonic acid far-field will dominate reactions. Chemical interactions need field studies to identify these boundaries.

Synergy Opportunities

We are very happy to utilize the skills and resources at NETL RIC to further the mission(s) of FE across portfolios. The number of ongoing collaborative studies is numerous, and if you identify places where we can help with your studies, please let us know.



Task 18.0 CO₂-Shale Interactions- Microscopic

Pursuing SANS/USANS measurements with NIST to further probe pore changes 16 at the micro and nano scale

Project Summary

Task 17.0 CO₂-Shale Interactions- Macroscopic

- Comprehensive literature review of CO_2 /shale interactions published.
- MD determined excess adsorption isotherms for single-component carbon dioxide and methane show that carbon dioxide demonstrates greater affinity to the illite surface.

Task 18.0 CO₂-Shale Interactions- Microscopic

- Changes in pores are occurring on the micron (opening) and nano (closing) scale after reaction with CO₂ and fluid
- Further investigate pore changes to understand possible alteration of overall permeability and fracture networks

Appendix

These slides will not be discussed during the presentation, but are mandatory.

Benefit to the Program

Task 17.0CO2-Shale Interactions- Macroscopic

- Carbon Storage Program Major Goals
 - Develop technologies to improve reservoir storage efficiency while ensuring containment effectiveness.
 - Support industry's ability to predict CO₂ storage capacity in geologic formations to within ±30 percent.
- Project Benefits Statement:
 - Publication of detailed literature review of CO₂/shale interactions to guide future research.
 - Fundamental understanding of swelling behavior examined from MD simulations and laboratory experiments.
 - Development of faster techniques to measure permeability of low k shales.
 - Examine and upscale pressure corrosion relationships relevant to shales as CO₂ seals.

Task 18: CO₂-Shale Interactions – Microscopic

- Carbon Storage Program Major Goals
 - Develop and validate technologies to ensure for 99 percent storage permanence.
 - Develop technologies to improve reservoir storage efficiency while ensuring containment effectiveness.
 - Support industry's ability to predict CO₂ storage capacity in geologic formations to within ±30 percent.
- Project Benefits Statement:
 - Determine how CO₂ and fluid induced alterations of Marcellus, Utica, Mancos, Eagleford, and Barnett shale affect flow pathways in terms of precipitation, dissolution, and pore space alteration.
 - Identify key properties of shale formations (carbonate rich vs silicate rich) at the nano- and micro-scale needed for quantifying CO₂ storage and seal activities.

Project Overview

Goals and Objectives

Task 17: CO₂-Shale Interactions – Macroscopic

- **Objective:** The product of this work is to deliver a literature review of CO2/shale interactions and a novel full immersion pulse decay device that measures shale properties under stressed conditions. The objective of this task is to develop an understanding of the interaction of CO2 with shale. Direct measurements of shale properties under stressed conditions and analysis of interactions of fractured shale with CO2 are the primary thrusts of this project.
- **Benefit:** Shale formations are widespread throughout the United States and understanding how they could be properly used in a national storage strategy is important. Development of techniques to rapidly measure shale properties, identification of shale mineralogy that will interact poorly with CO2 and understanding of how shale could fracture under low consistent stresses will all directly benefit geologic CO2 storage efforts.
- **Challenges:** Several R&D challenges exist for the development of CO2/shale interactions. As identified in a comprehensive literature review of CO2/shale interactions (Lindner 2017) there is wide variability in the reported behavior of shale and CO2. In part, a poorly defined understanding of what constitutes a shale is at fault; mineralogical content as opposed to organic content in the context of gas plays. Focusing efforts on fundamental mineralogical descriptions of shale instead of shale play names should alleviate this challenge. A second challenge is replication of subsurface conditions: (1) shale acquired from depth has been altered by the removal of stresses and fluids and (2) shale acquired from outcrops has undergone long term changes due to its uplift and exposure. By combining in situ conditions for laboratory testing and simulating subsurface environments best practices will be used to overcome these challenges.
- Approach: Coupled experimental and numerical modeling of phenomena.

Project Overview

Goals and Objectives

Task 18: CO₂-Shale Interactions – Microscopic

- Task Technical Approach and Project Relevancy (PI: Angela Goodman)
- Objective: Quantify CO₂ interactions with shale at the nano- and micro-scale and provide quantitative inputs at the macro-scale to understand flow properties and inputs for reservoir simulations of large-scale storage activities in shale.
- Benefit: Reaction of CO₂ with in situ fluids, fracturing fluids, and reactive shale interfaces may generate new reactive surfaces or intermediates that may alter the properties of the formation. A fundamental understanding of the reactivity of CO₂ with shale and fluid interfaces will help in identifying how storage in shale formations plays a role in CCS activities.
- Challenges:
 - Fundamental research examining the geochemical interactions of CO₂ and fluids with shale is limited.
 - Limited understanding of how reaction of CO₂ with in situ fluids, fracturing fluids, and reactive shale interfaces may generate new reactive surfaces or intermediates that may alter the properties of the formation.
- Approach:
 - Complete experimental analysis of CO₂, fluid, and shale interactions at the nano- and micro-scale for a suite of shales suitable for potential CO₂ storage activities.
 - Relate shale attributes to CO₂ storage potential.
 - Scale results observed at the nano- and micro-scale to the macro-scale to understand flow properties and inputs for reservoir simulations
 of large-scale storage activities in shale.

Organization Chart

Research Theme: Shales: Characterizing Interactions of CO_2 with Shale for Potential Storage Needs and Seal Integrity

• Task 17.0 CO₂-Shale Interactions- Macroscopic

Team Members: Johnathan Moore, Sarah Brown, Ernest Lindner, Michael Hannon, Yael Tucker, Leebyn Chong, Eugene Myshakin, Guanyi Lu, Andrew Bunger, Dustin Crandall

• Task 18.0 CO₂-Shale Interactions- Microscopic

Team Members: Angela Goodman, Sean Sanguinito, Barbara Kutchko, Jeff Culp, Mary Tkach, Sittichai Natesakhawat, Dustin Crandall, Patricia Madden





CO₂-Shale Interactions - Macroscopic



Timeline Overview – Sub-Task 17.1.1



Not applicable, project ending in 2018.

Completion

Timeframe

CO₂-Shale Interactions



- Publication of detailed literature review of CO₂ and H₂O interactions with shale. (Completed November 2017) 1.
- 2. Presentation of literature review at a scientific conference. (June 2018)
- 3. Submission of peer-reviewed publication describing unknowns of CO₂ and H₂O interactions with shale. (September 2018)

Impact		
Key Accomplishments/Deliverables	Value Delivered	
 2016-2017: Literature review of CO₂ and water interactions with shale Lindner, E. Review of the Effects of CO₂ on Very-Fine-Grained Sedimentary Rock/Shale – Part III: Shale Response to CO₂; NETL-TRS-11-2017; NETL Technical Report Series; U.S. Department of Energy, National Energy Technology Laboratory: Morgantown, WV, 2017; p 228. Lindner, E. Review of the Effects of CO₂ on Very-Fine-Grained Sedimentary Rock/Shale – Part II: Clay Mineral & Shale Response to Hydration; NETL-TRS-10-2016; NETL Technical Report Series; U.S. Department of Energy, National Energy Technology Laboratory: Morgantown, WV, 2017; p 68. Lindner, E. Review of the Effects of CO₂ on Very-Fine-Grained Sedimentary Rock/Shale – Part I: Problem Definition; NETL-TRS-1-2016; NETL Technical Report Series; U.S. Department of Energy, National Energy Technology Laboratory: Morgantown, WV, 2017; p 68. 	 There has been decades of research in multiple disciplines that examine the interactions of CO₂, H₂O and shale; even with this research there are some knowledge gaps in how CO₂ and shale will interact during geologic carbon storage. By performing a detailed literature review across disciplines, we are illuminating the most relevant research problems that remain. To date, a three part literature report of over 350 pages was completed, but this needs to condensed to reach a broader audience. Delivering this report in a consumable fashion will enable the value of these reports to be utilized. 	
	TRL Score Go / No-Go Project Milestone 2	

Timeline Overview – Sub-Task 17.1.2



3





- 1. EERC Collaborative tests complete. (June 2018)
- 2. Molecular simulations of CH4/CO2 and clay under different water saturations complete. (July 2018)
- 3. Molecular simulations of CH4/CO2 retention in kerogen under dry conditions and with residual water complete. (December 2018)
- 4. Paper with a tentative title: "Molecular modeling of ternary CO2-CH4-H2O mixture adsorption on illitic clay surfaces in shales" submitted to a peer-reviewed international journal. (December 2018)
- 5. Three fracture flow experiments complete. (September 2019)
- 6. Six fracture flow experiments complete and relationship between shale content and fracture permeability developed. Submission of results to peer reviewed journal. (June 2020)

December 2018 – Are there cases tested with molecular dynamics simulations at representative GCS conditions where the amount of swelling would indicate that fracture closure likely?

Key Accomplishments/Deliverables	Value Delivered
<u>2017</u> : Through collaborative studies with the EERC, CO ₂ permeability tests in ractured and unfractured Bakken shale have been performed. The results are mixed. Fractured shale permeability has been shown to decrease, but no consistent trend or observable deformation of structure has been observed.	 The interaction of CO₂ and shale can be complex, and a basic understanding of how various shale formations will interact as seals or potential reservoir formations must be described in order to provide reliable long term estimates of storage potential

Timeline Overview – Sub-Task 17.1.3



Low Permeability Shale Measurements



Milestones

Go / No-Go

- 1. Compare and contrast traditional pulse decay with high resolution steady state gas permeability measurements. (Completed December 2017)
- 2. Peer reviewed manuscript of modification to high-resolution steady-state permeability measurements with low downstream pressure submitted to journal. (September 2018)
- 3. Assemble a bench scale full immersion pulse decay device and shakedown apparatus. Compare/contrast measurements to traditional pulse decay. (December 2018)
- 4. Develop a full immersion pulse decay permeability technique for fractured shales. Examine this technique numerically to confirm measurements can be obtained, and perform shakedown on experimental prototypes. (June 2019)

September 2019 - Test full immersion pulse decay permeability technique for fractured shales experimentally and show match numerical expectations. If successful continue towards commercialization with partners.

Project

Completion

Milestone

Go / No-Go

Timeframe

TRL Score

5. Commercialization of full immersion pulse decay technique with external partner. (December 2019)

Impact

Key Accomplishments/Deliverables	Value Delivered
2017 : Numerical simulations and preliminary experimental measurements have shown that the full immersion pulse decay technique (patent pending) is able to measure shale permeability faster than any other reliable method on the market. Discussions with industry and applications for technology commercialization funds are underway.	• Shale permeability measurements that are accurate, can be obtained efficiently, and with a process that is openly described will provide greater confidence in measurements for stakeholders and regulators that need to quantify carbon storage seal integrity.



Timeline Overview – Task 17.1.4

NATIONAL ENERGY TECHNOLOGY LABORATORY

Stress Corrosion of Shales



- 1. Modifications to existing CT and flow systems approved for stress corrosion testing. (June 2018)
- 2. Successful test of shale microstructure cracking observed in laboratory due to stress corrosion. (December 2018)
- 3. Complete parametric study demonstrating acoustic emission and specimen failure rate at various levels of stress, fluid pressure, and with various fluid/rock combinations. (December 2018)
- 4. Observe microstructural changes during stress corrosion cracking and complete core-scale experiments. (September 2019)
- 5. Implement geomechanically-coupled numerical simulations and benchmark to core-scale experiments. (December 2020)

September 2019 – Successful observation of microstructural changes due to stress corrosion cracking with laboratory experiments.

Chart Key

Project

Completion

Go / No-Go

Timeframe

RL Score

5

Milestone

Impact		
Key Accomplishments/Deliverables	Value Delivered	
2017 : New start in 2018. Prior work in mechanical deformation of rocks in the Carbon Storage, NRAP, and Onshore Unconventional FWPs, using the computed tomography scanner facility at NETL and stress corrosion measurements in shale by personnel at the University of Pittsburgh will be leveraged to accomplish this task.	 The goal of the proposed research is to observe and quantify the impact of stress corrosion on shale permeability. Stress corrosion is the time dependent failure due to stresses that are insufficient to instantaneously break the rock. Development of a coupled numerical model to account for this behavior benchmarked on experimental observations, will enable greater seal characterization for long term stress behavior of shale sealing formation 	



Task 18: Project Timeline Overview

CO₂-Shale Interactions- Microscopic (PI: Angela Goodman)



Dependent upon quantitative results with the Utica and

Go / No-Go

Timeframe

#

TRL Score

Project

Completion

Milestone

Marcellus samples

- 1. M.8.1 Quantify the geochemical impact of CO₂ and fluid interactions on Utica and Marcellus shale at the nano- and micro-scale.
- 2. Quantify the potential changes in flow properties of Utica Shale that has been modified with CO_2 and fluid.
- 3. Quantify the geochemical impact of CO₂ and fluid interactions on Barnett, Eagleford, and Mancos Shale at the nano- and micro-scale.
- 4. Quantify the potential changes in flow properties of Barnett, Eagleford, and Mancos Shale that has been modified with CO₂ and fluid.
- 5. Initial database of shale properties at the nano-, micro-, and macro-scale needed for quantifying CO₂ storage potential.

Imp	act
Key Accomplishments/Deliverables	Value Delivered
2015: Quantified CO ₂ interaction with Na+ exchanged montmorillonite 2016 : Effects of short and long term geochemical reactions of fracturing fluid with Marcellus Shale and Huntersville Chert 2017: Quantified interactions of CO ₂ with Utica Shale	 Determine how CO₂ and fluid induced alterations of Marcellus, Utica, Mancos, Eagleford, and Barnett shale affect flow pathways in terms of precipitation, dissolution, and pore space alteration. Identify key properties of shale formations (carbonate rich vs silicate rich) at the nano- and micro-scale needed for quantifying CO₂ storage and seal activities
	Chart Key

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Bibliography

Task 18.0CO2-Shale Interactions- Microscopic

Estimating the Prospective CO₂ Storage Resource and Quantifying CO₂ Induced Changes of Shales

In-situ Surface Chemistry, Surface Topography, Surface Area, and Porosity



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Task 18.0CO2-Shale Interactions- Microscopic

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