
Generation and Utilization of NH_3/H_2 Fuel Blends

Colin Wolden

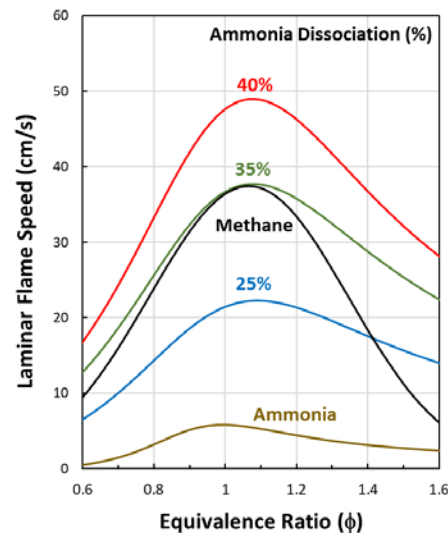
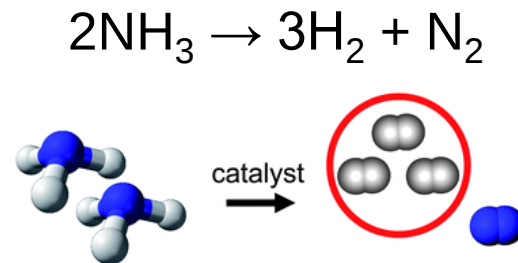
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Ammonia Combustion Technology Webinar: 7/8/2025

Ammonia-Hydrogen: The Goldilocks Blend

Fuel	ΔH_c (kJ/mol)	LFS (cm/s)	MIE (mJ)	Flammability
NH ₃	382.6	7	680	15 – 28%
H ₂	285.8	>200	0.017	4 – 75%
CH ₄	890.4	35	0.28	5 – 15%

NH ₃ /H ₂	~300	Tunable: Whatever you want
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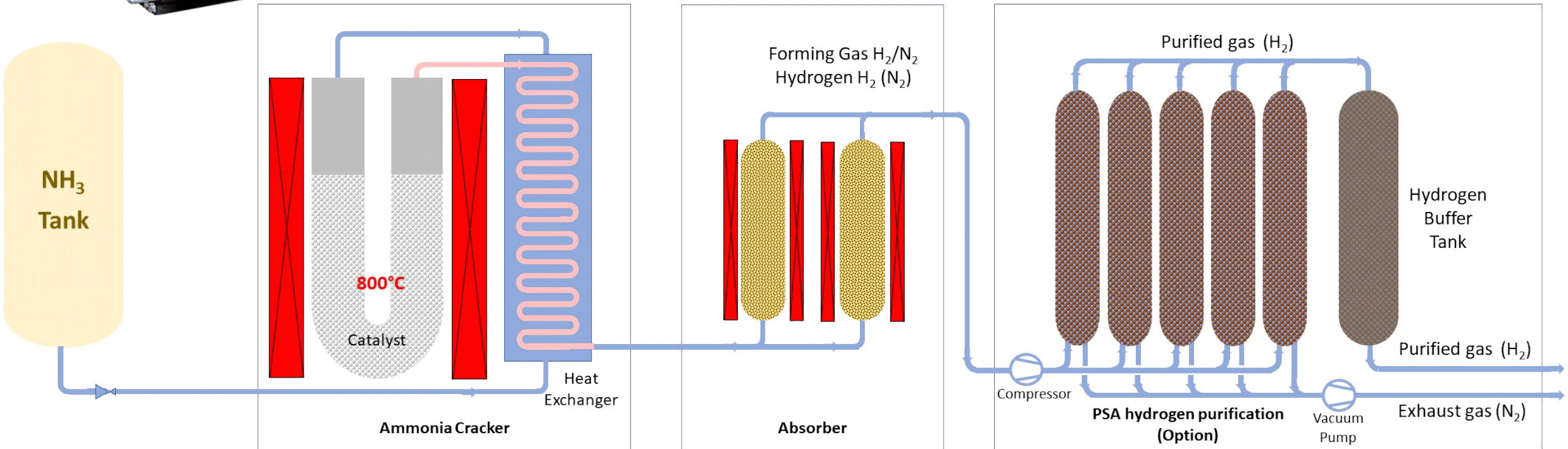


Current Technology for NH_3 cracking

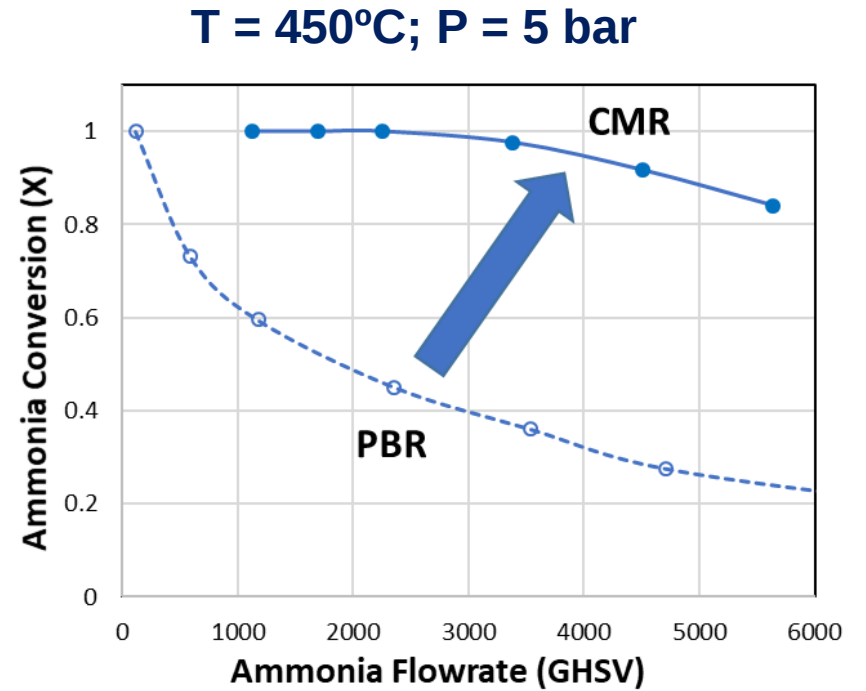
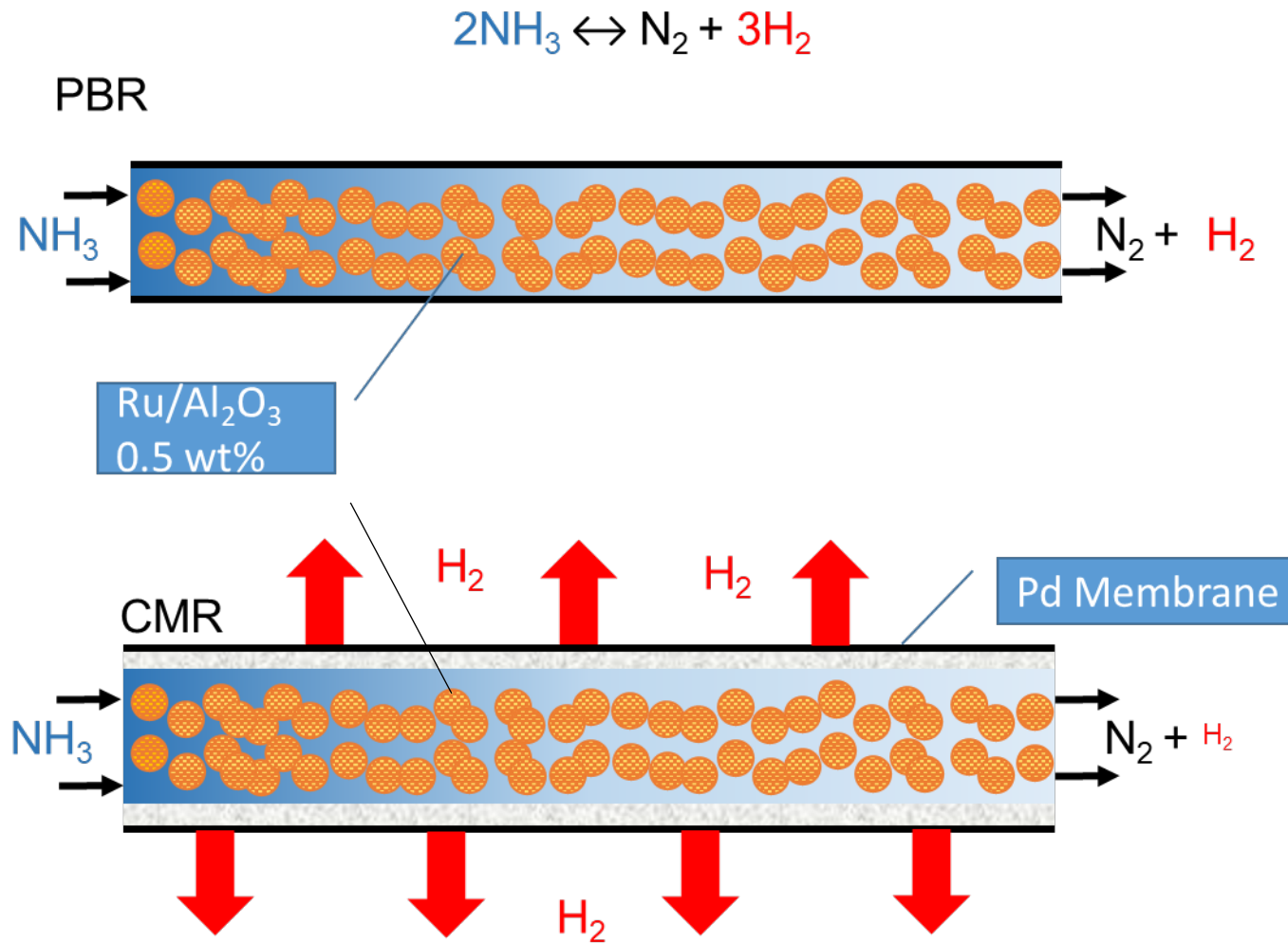


CapEx & OpEx Intensive

- Very high temperature for complete dissociation
- Compressors, PSA for purification



Catalytic Membrane Reformer (CMR)

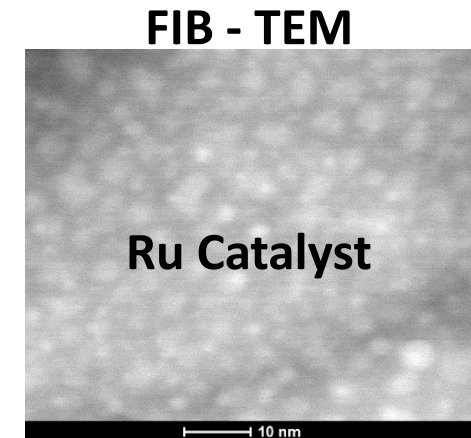
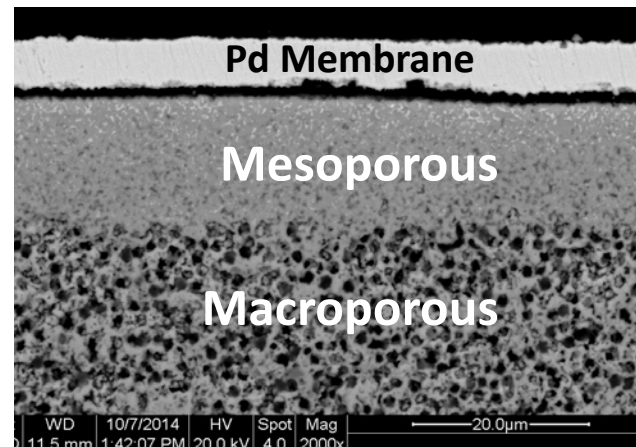
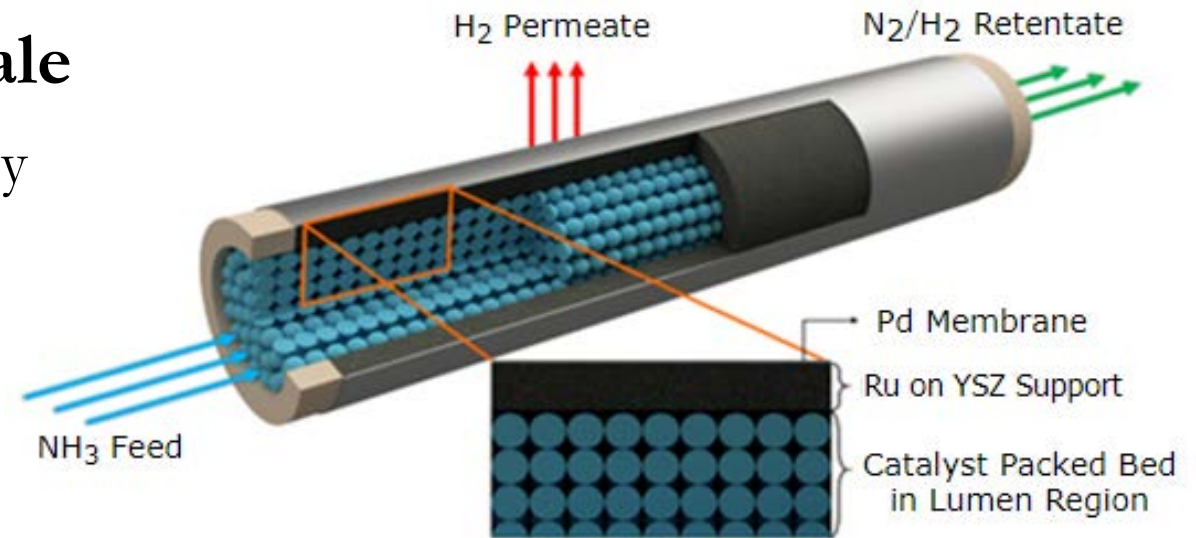


$$r = k_f \left[\left(\frac{P_{\text{NH}_3}^2}{P_{\text{H}_2}^3} \right)^\beta - \frac{P_{\text{N}_2}}{K_{eq}} \left(\frac{P_{\text{H}_2}^3}{P_{\text{NH}_3}^2} \right)^{1-\beta} \right]$$

Initial Target: UHP Hydrogen

Attributes fully demonstrated at lab scale

- Patented catalytic membrane reformer geometry
- Innovation fabrication
- Achieved record productivity
- Reduce $T < 450^{\circ}\text{C}$

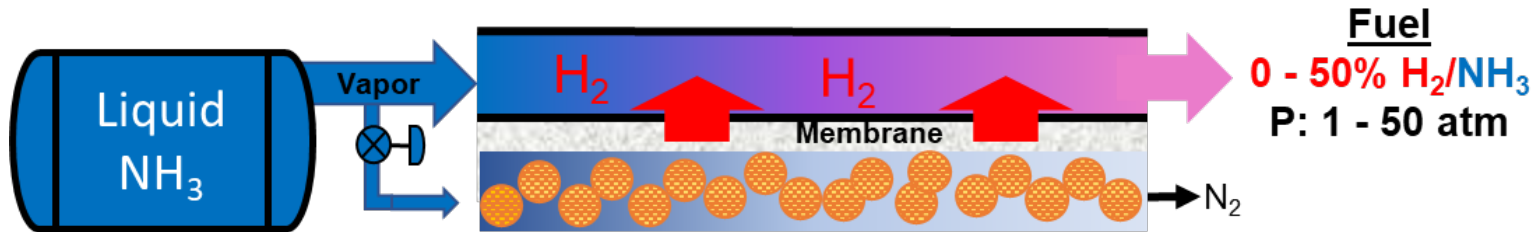


R. Sitar, J. Shah, Z. Zhang, H. Wikoff, J. D. Way and C. A. Wolden, "Compact ammonia reforming at low temperature using catalytic membrane reactors", *Journal of Membrane Science* **644**, 120147 (2022).

Pivot: Use sweep to generate NH_3/H_2 Blends

CMR Operation

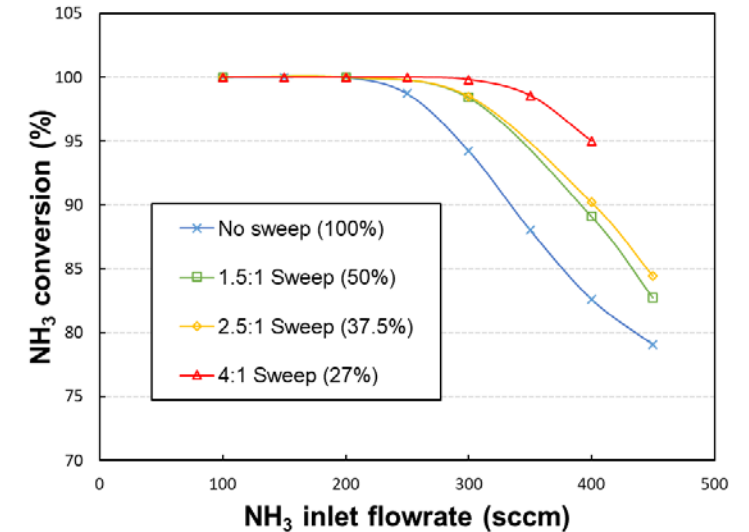
- Send a fraction to reformer, completely decompose
- Extract all the hydrogen into sweep gas, reject N_2
- Enables 100% H_2 recovery



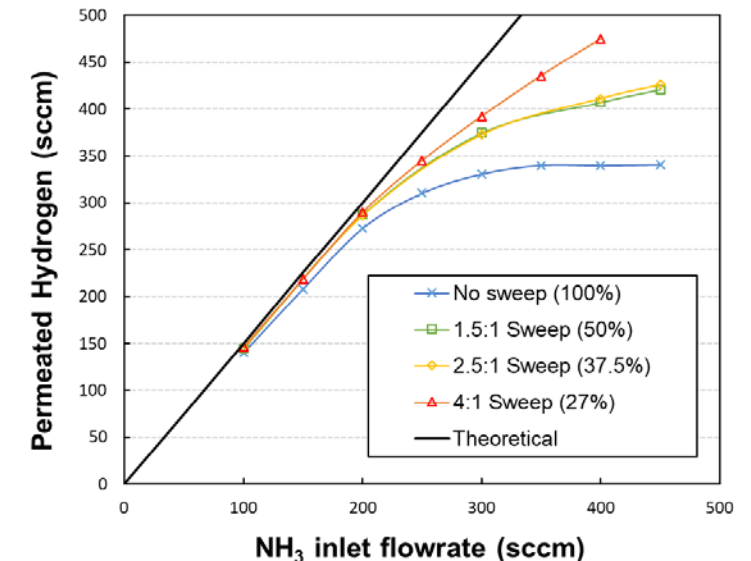
Sweep Gas Options

- Ammonia: Tunable H_2/NH_3 Blends
- Methane: Hydrogen enriched natural gas (HENG)
- Steam: Humidified Hydrogen

Higher Conversion



Improved Recovery



R. Sitar, J. Shah, J. D. Way and C. A. Wolden, "Efficient generation of H_2/NH_3 fuel mixtures for clean combustion", *Energy & Fuels* **36**, 9357-9364 (2022).

CMR Advantage: N₂ Rejection

$$S_L = S_{LO} T_n^\alpha P_n^\beta \gamma$$

$$S_{LO} = f(\phi, x_{H_2})$$

$$\alpha = f(\phi, x_{H_2})$$

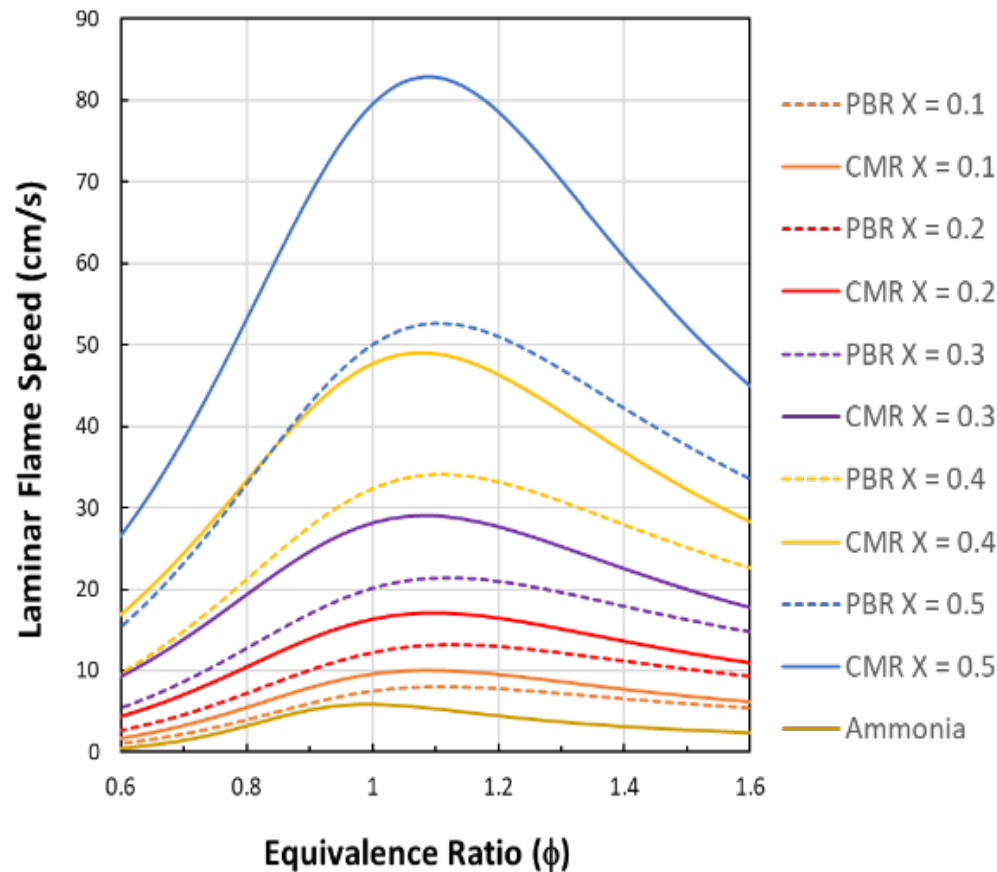
$$\beta = f(\phi, x_{H_2})$$

$$\gamma = f(\phi, x_{N_2}, x_{H_2}, T_n, P_n)$$

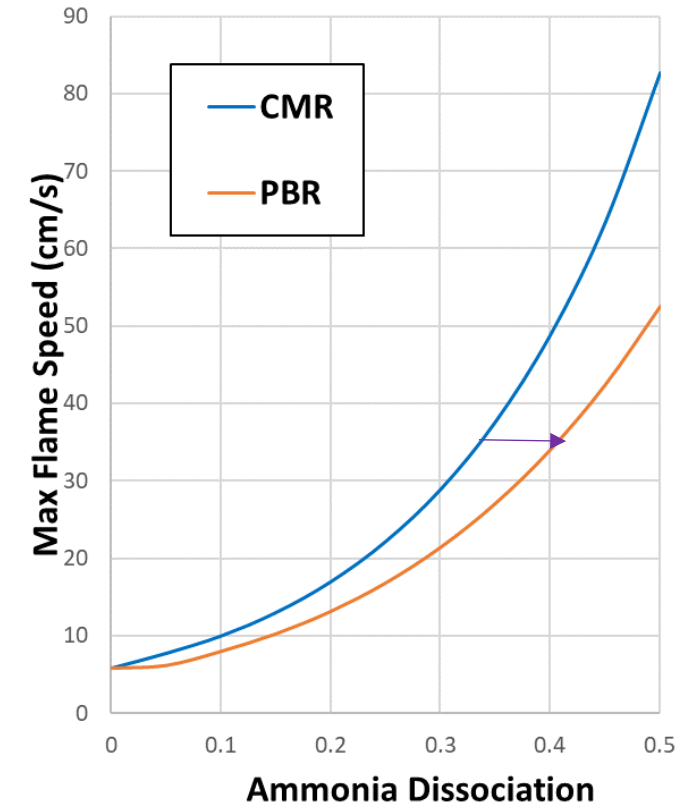
Benefits

- CMR: 25 – 60% higher LFS
- Match Methane
 - CMR requires 35% conversion
 - PBR requires 42% conversion
 - 20% more energy
 - Higher adiabatic flame temperature

CMR: Solid Lines (NH₃/H₂)
PBR: Dashed Lines (NH₃/H₂/N₂)



CMR: NH₃/H₂
PBR: NH₃/H₂/N₂



X = Ammonia Conversion

A. Goldmann and F. Dinkelacker, "Approximation of laminar flame characteristics on premixed ammonia/hydrogen/nitrogen/air mixtures at elevated temperatures and pressures," *Fuel* **224**, 366-378 (2018).

Potential Benefits of Nitrogen Removal

Modified standard SI engine (Frigo lab in Italy)

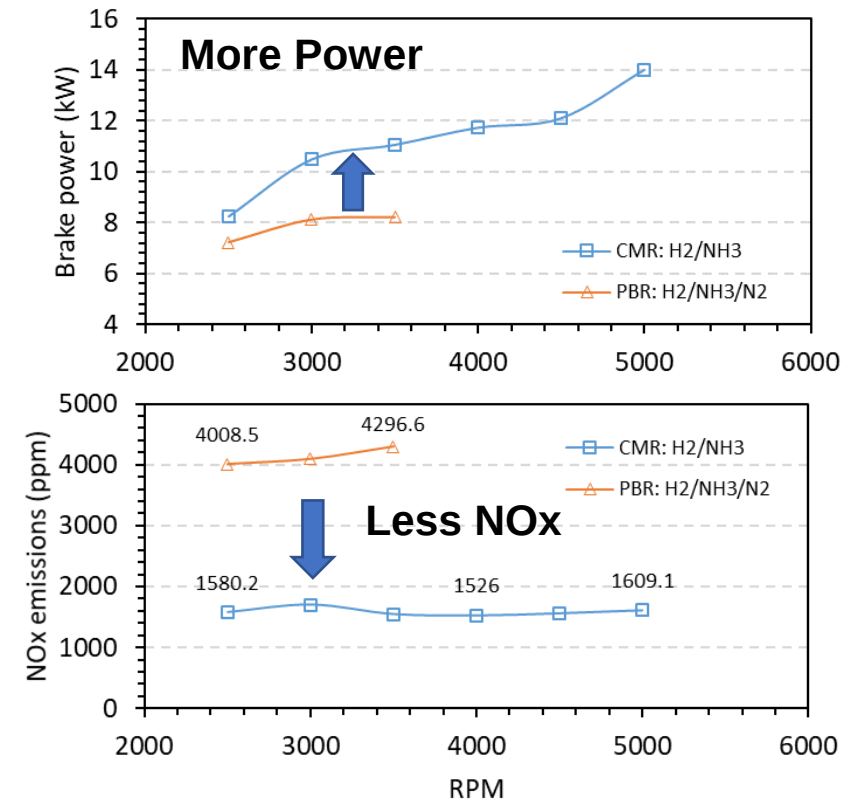
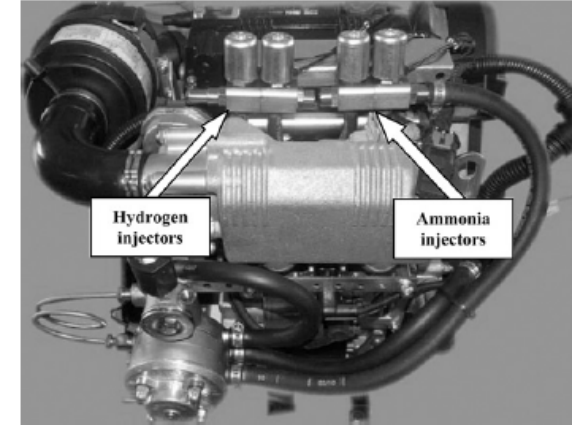
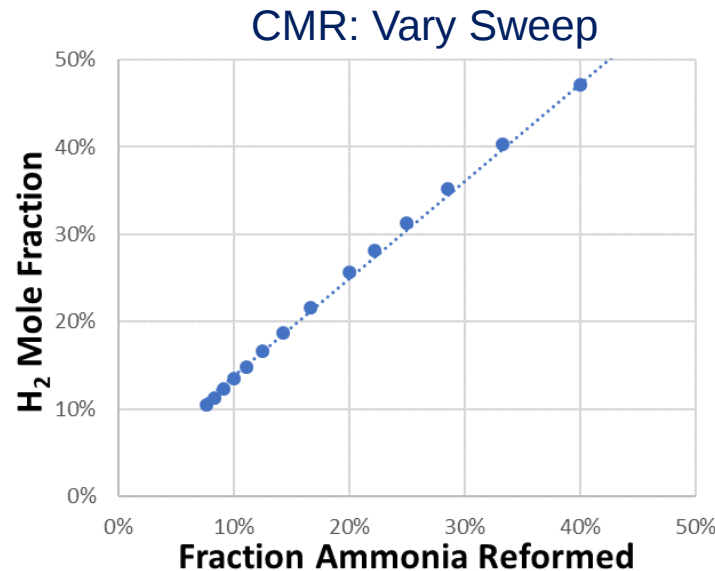
- Compared NH_3/H_2 (CMR) vs. $\text{NH}_3/\text{H}_2/\text{N}_2$ (PBR)
- Benefits of N_2 removal:
 - Achieved $\sim 35\%$ more power, greater range of stable operation
 - Produced $>65\%$ less NO_x (less than gasoline operation)

Requires Dynamic Composition Control

- $\sim 50\%$ H_2 at cold start
- $\sim 10\%$ H_2 under full load
- Easily achieved in CMR

M. Comotti and S. Frigo, "Hydrogen generation system for ammonia-hydrogen fuelled internal combustion engines," *International Journal of Hydrogen Energy* **40**, 10673 (2015).

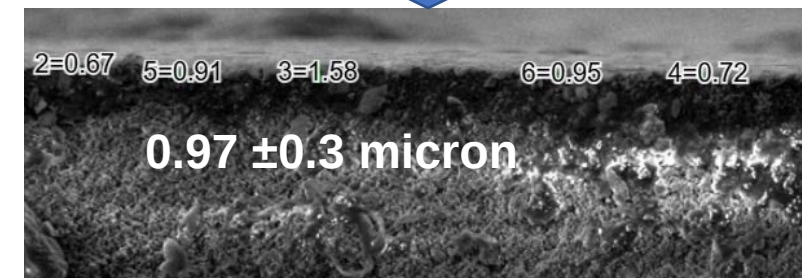
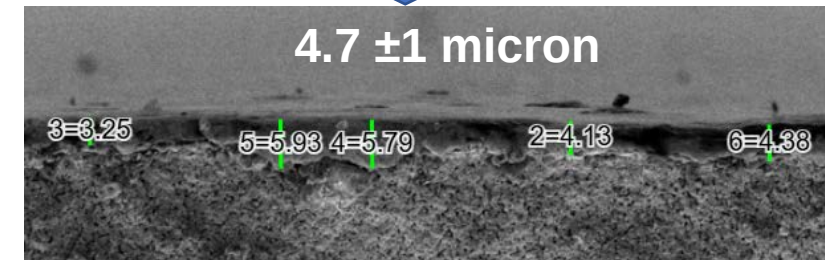
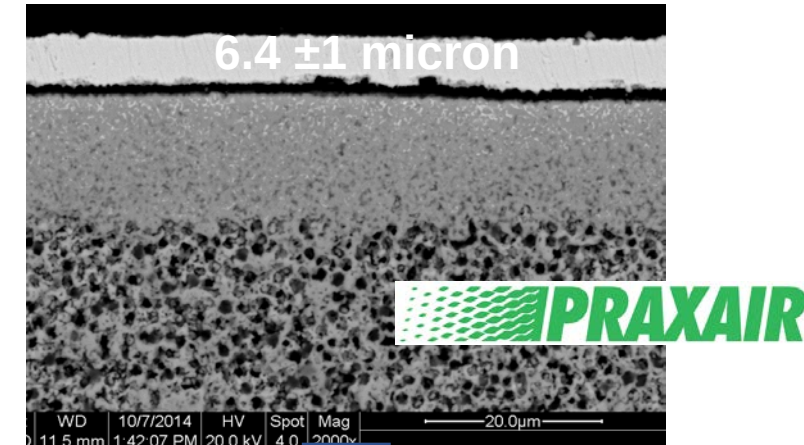
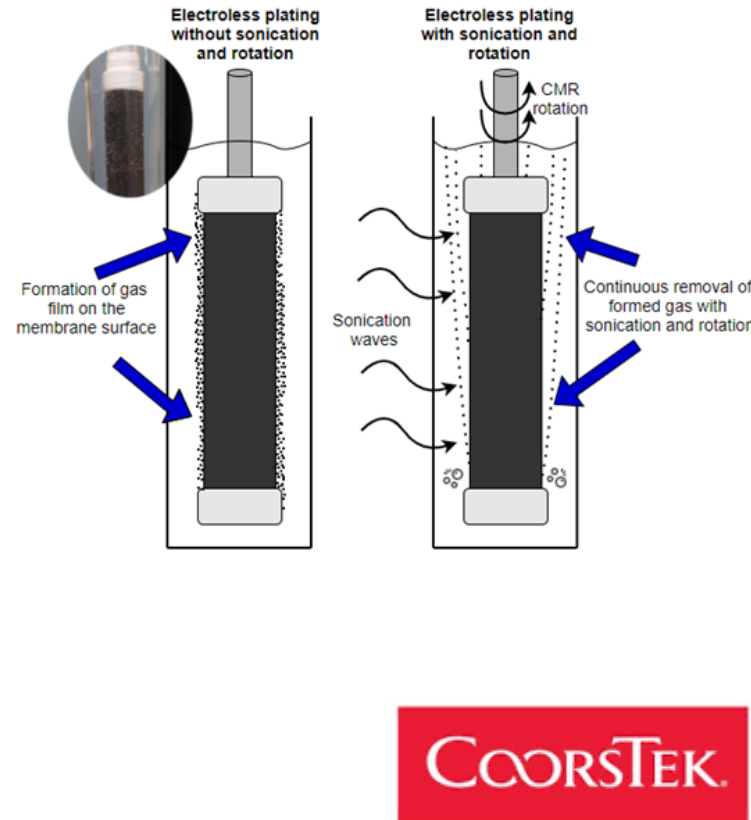
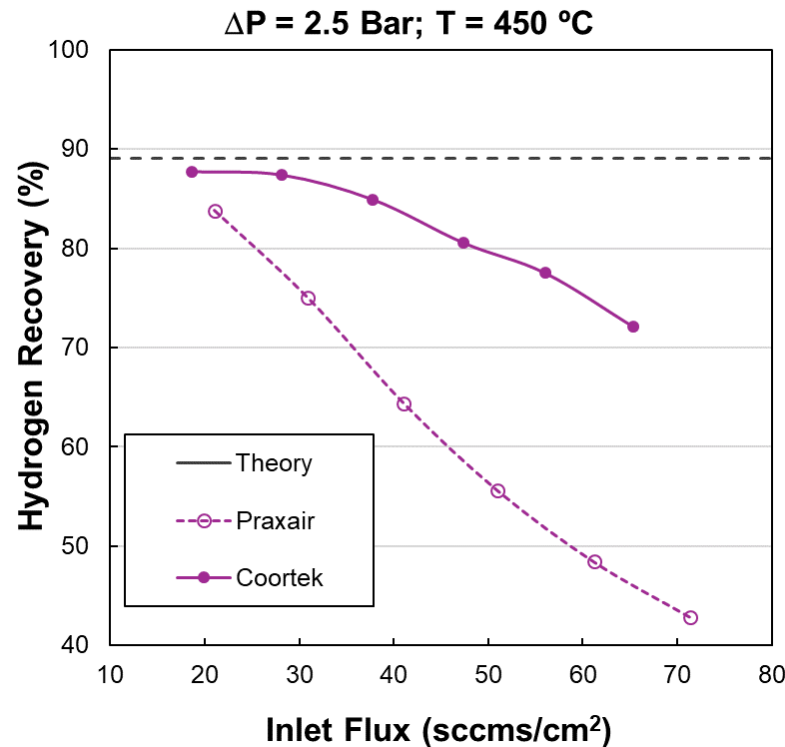
S. Frigo and R. Gentili, "Analysis of the behaviour of a 4-stroke SI engine fuelled with ammonia and hydrogen," *International Journal of Hydrogen Energy* **38**, 1607 (2013).



Goals of Current Project: Better Performance & Cheaper

Better Membranes

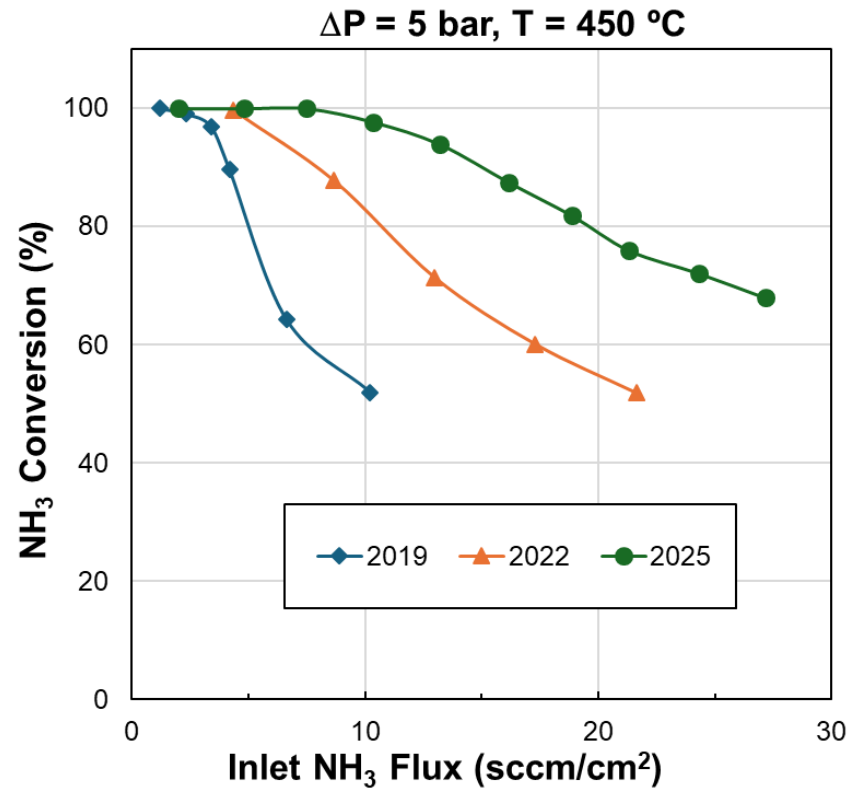
- Performance Limited by Flux
- Reduce Pd from 6.4 to $< 1 \mu\text{m}$
- Reduce resistance of membrane support
- Recovery from 75% $\text{H}_2/25\% \text{N}_2$



Goals of Current Project: Better Performance & Cheaper

Performance Evolution

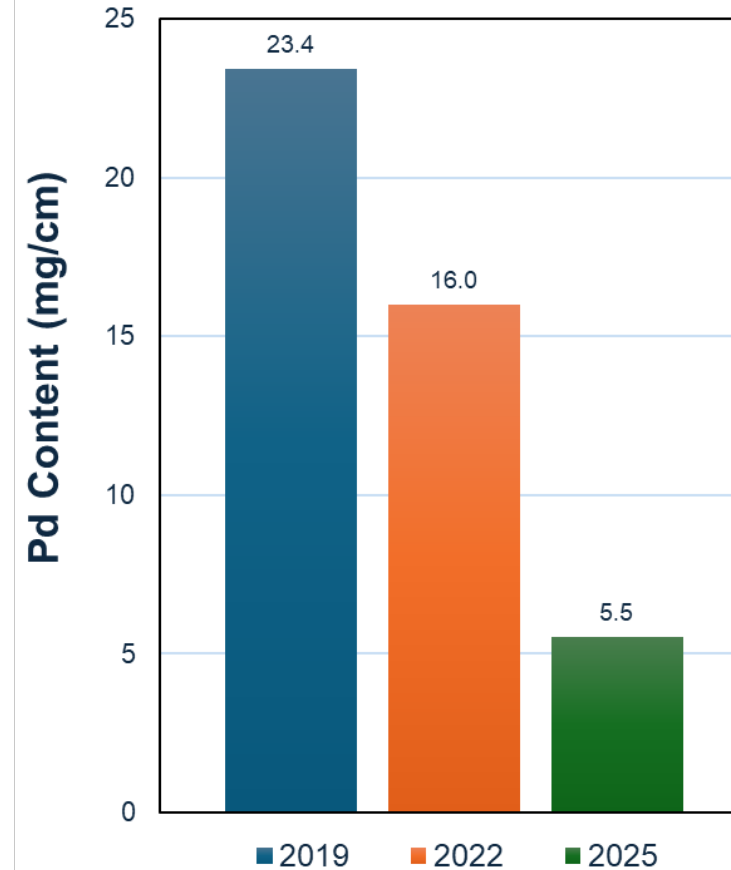
- No catalyst in lumen
- 4X enhancement in throughput



Z. Zhang, S. Liguori, T. F. Fuerst, J. D. Way, and C. A. Wolden, "Efficient ammonia decomposition in a catalytic membrane reactor to enable hydrogen storage and utilization," *ACS Sustainable Chemistry & Engineering* **7**, 5975-5985, (2019).

Pd Inventory

- >4X reduction

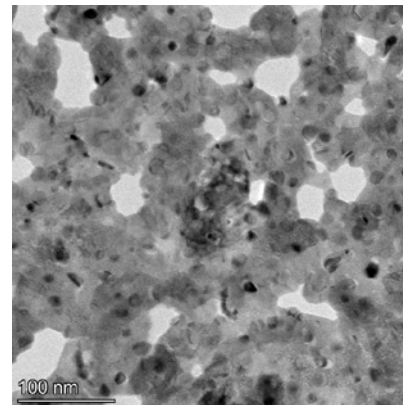
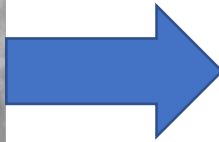
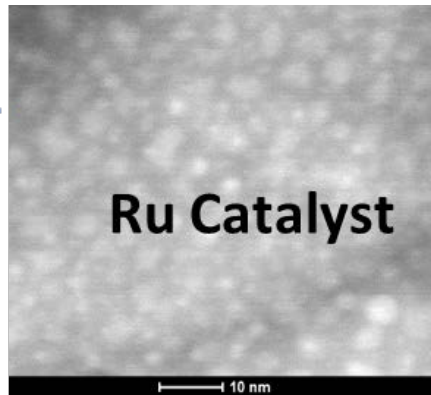


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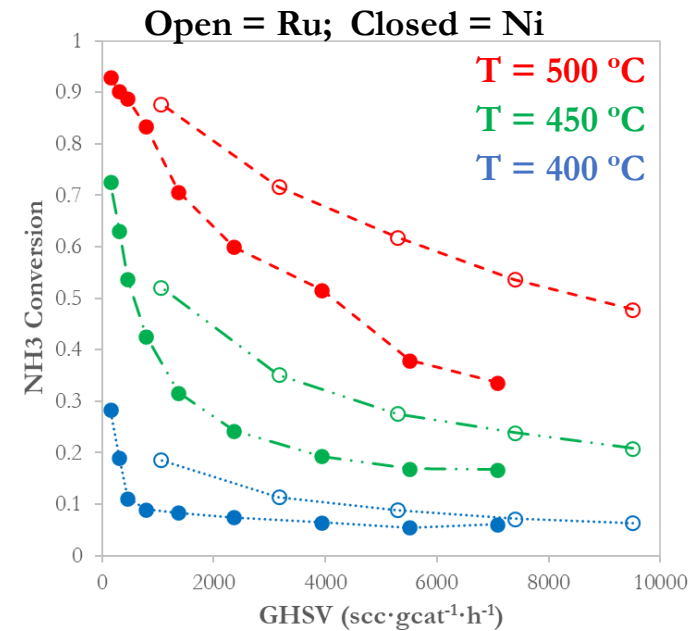
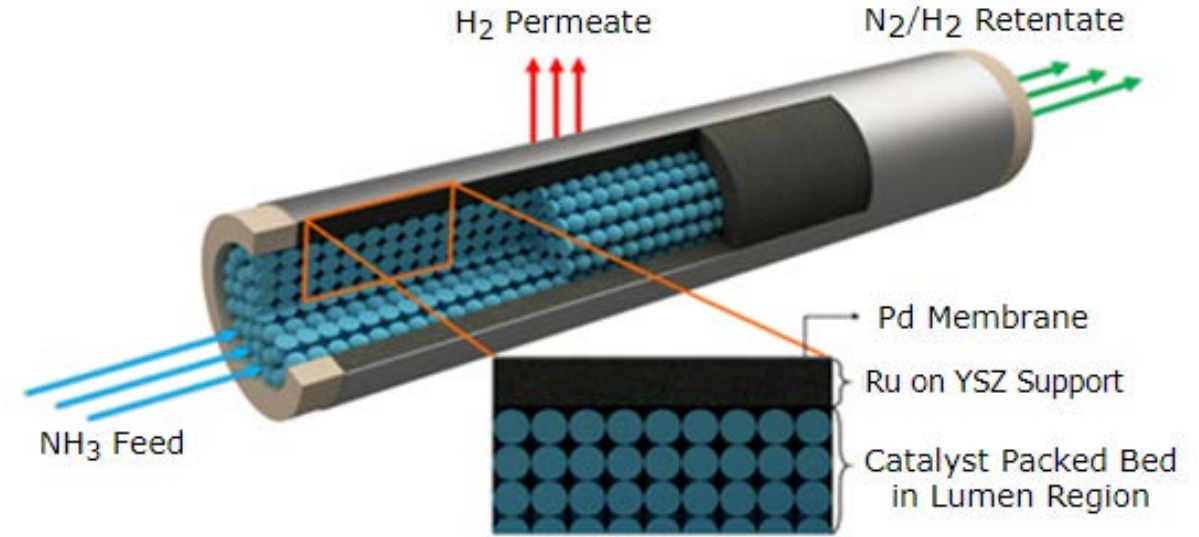
Goals of Current Project: Better Performance & Cheaper

Cheaper Catalysts

- Ru costs > Pd Costs
- Significant Ru in Lumen
- Developing Nanoscale Ni Catalyst
- Approaching performance of Ru Control (BASF)



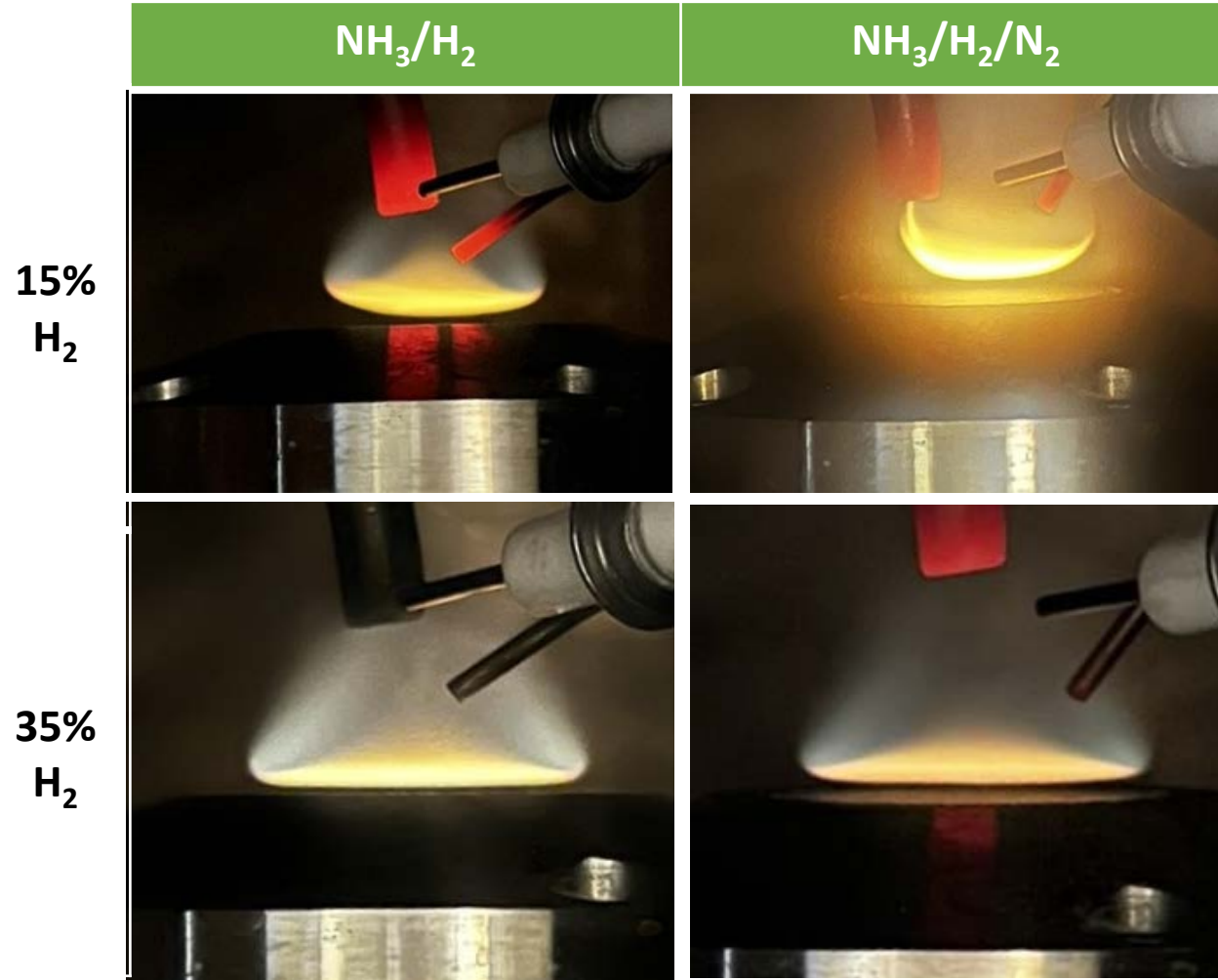
Ni/La₂O₃



Utilization of NH_3/H_2 : Laminar Flame Burner

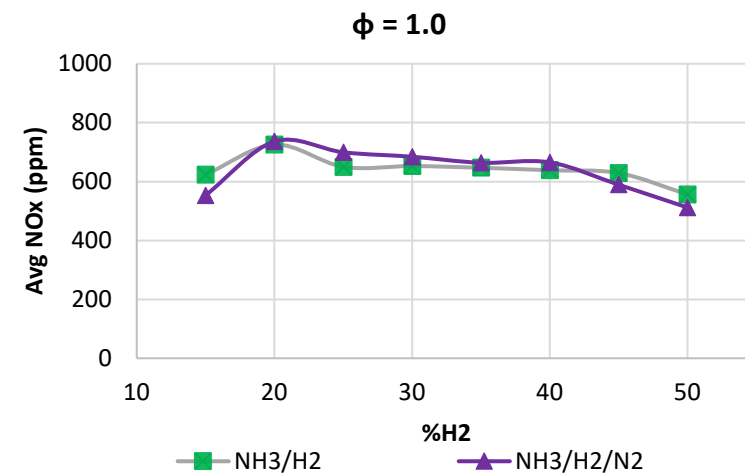
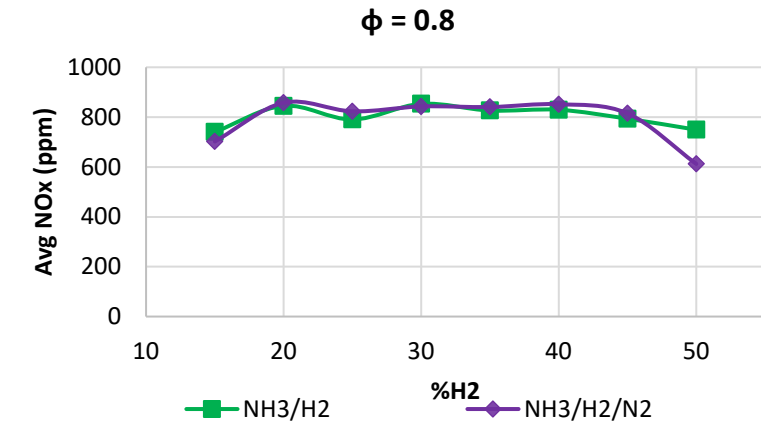
Any Benefit to N_2 Removal?

Explored % H_2 and ϕ



Improved Flame Stability

Negligible Impact on NO_x



Utilization of NH_3/H_2 : Generator

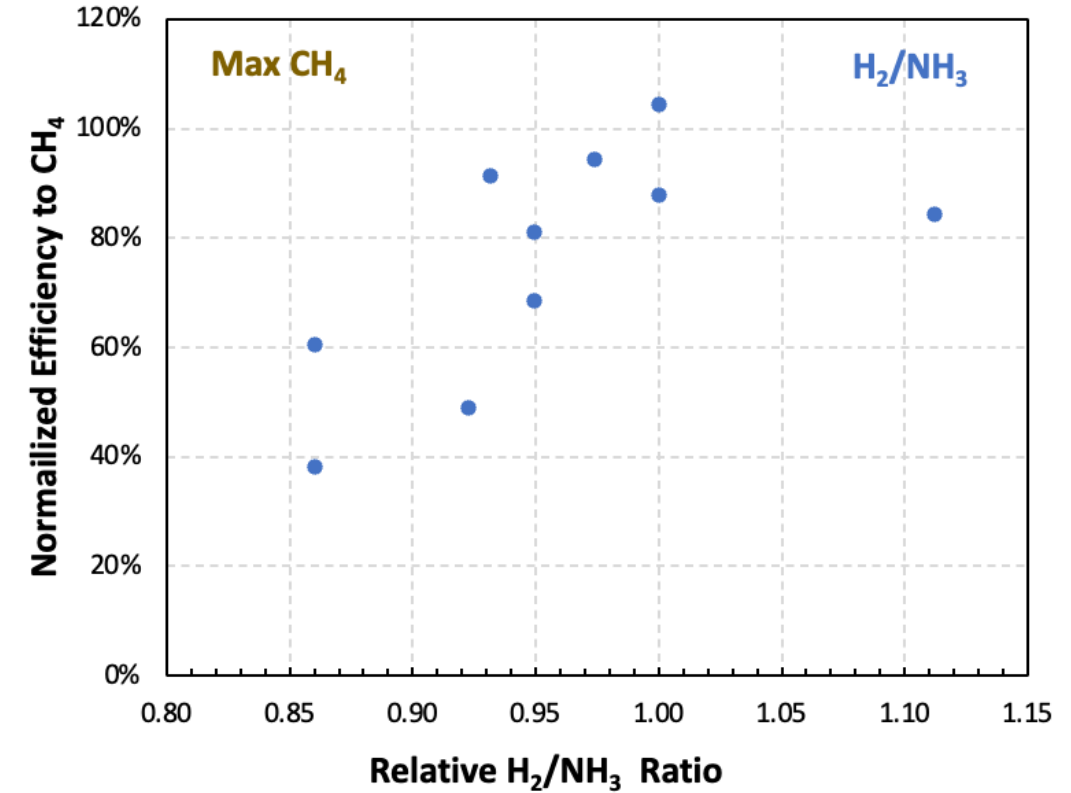
Is it a true drop-in Fuel?

Firman Tri-Fuel Generator (Gasoline/LPG/ CH_4)

Compared CH_4 with $\text{NH}_3:\text{H}_2$ blends

Zero Modification

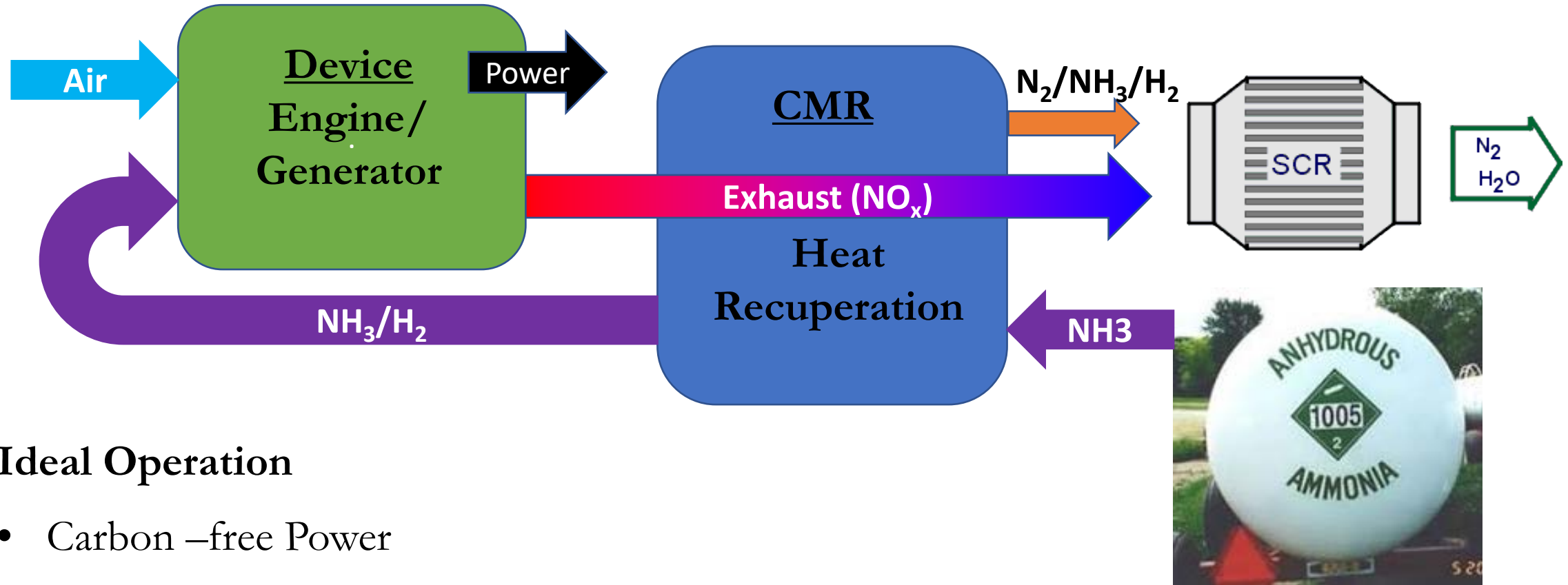
$$\eta = \frac{\text{Electrical Power}}{\dot{n}_{\text{fuel}} \times \text{LHV}}$$



NH_3/H_2 Generator

- Stable operation with varied load
- Fuel to power efficiency comparable to CH_4
- Exhaust temperature suitable for heat integration

Concept: Integrated Carbon-Free Power



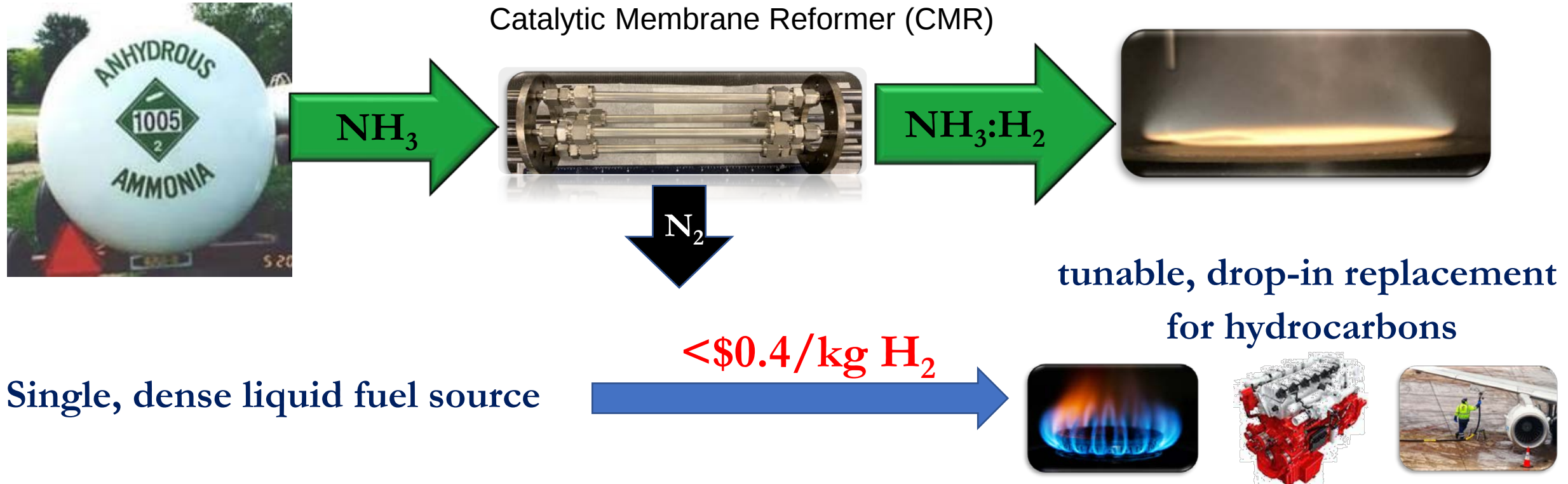
Ideal Operation

- Carbon –free Power
- No NO_x
- Adiabatic Operation



- CEO: Dr. Rok Sitar
- Email: rok@blazeenergytech.com
- Website: <https://blazeenergytech.com/>

Summary: Catalytic Membrane Reformer



Better Storage

- Existing infrastructure
- 25X lower storage costs
- Zero carbon

Better System

- Higher throughput ($>10\text{X}$)
- Energy Efficient ($T < 400^\circ\text{C}$)
- Low OpEx/CapEx

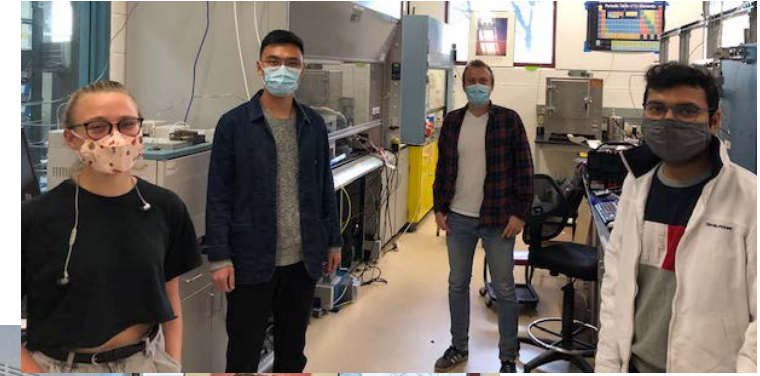
Better Fuel

- Dynamic control (H_2/NH_3)
- High pressure delivery
- No N_2 : More power

Acknowledgements

People

- Prof. Doug Way
- Alums: Dr. Zhenyu Zhang (Ampower)
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