

# LOAD-Z

**L**ow-NOx, **O**perable **A**mmonia Combustor **D**evelopment for **Z**ero-Carbon Power



FT4000® Aeroderivative Dual Fuel Gas Turbine Engine.



COLLINS AEROSPACE | PRATT & WHITNEY | RAYTHEON

### **RTX Technology Research Center**

**DOE/NETL University Turbine Systems Research (UTSR) Meeting September 25, 2024**

*Prime Contractor*: RTX Technology Research Center (RTRC) *Subcontractor*: University of Connecticut (UConn)

**RTRC** 

**RTX Technology Research Center** 



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# **Key Project Goals …. WHAT / WHY**

**L**ow-NOx **O**perable **A**mmonia-Combustor **D**evelopment (LOAD-Z)



- $-$  Fundamental NH<sub>3</sub> flame data relevant to turbines:
	- P, T > ambient  $\rightarrow$  relevant to compressor exit conditions .....
	- strained & turbulent flames …. …. ….

o *Targeted outcome*: expand published data w/ new, useful data (previously unreported)







- Predictive capability for  $NH<sub>3</sub>$  combustion & emissions
	- $\bullet$  NO<sub>X</sub> formation kinetics integral w/ NH<sub>3</sub> comb. kinetics .....
	- CFD of turb. NH<sub>3</sub> flames w/ NO<sub>x</sub> & NH<sub>3</sub> slip ( $\eta$ <sub>COMB</sub>) prediction
		- o *Targeted outcome*: capability for GT combustor design



- Develop & test NH<sub>3</sub> gas-turbine combustor " $@$  scale"
	- $\bullet$  Single-nozzle-rig (SNR) scale demo.  $@$  high P, T ...
	- Pure  $NH_3$  combustion @ 75% 100% power
		- o *Targeted outcome*: < 30ppm NOx\*\* & >99.99% efficiency



\*\*Note recent ETN recommendations for NOx reporting with hydrogen-containing fuels



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# **Approach & Work …. HOW**



- **4 Lab Experiments -&- 1 Combustor Test:** *̶ Ongoing: 2 laminar / fundamental rigs*
- **RTRC RTX Technology** ... ...................<br>esearch Cente **UCONN**
- **3 "Kinetics" Models -&- 1 CFD Model:** *̶ Kinetics validation: Counterflow-flame & Flat-flame*
- *̶ Start Q4 & Q1: 2 turbulent / applied rigs* → *complex fluid mechanics*
	-
	-
- 
- *̶ Combustor design: Chem. Reactor Network & CFD w/species transport*
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# **Fundamental Counterflow Data .... UCONN EXPERIMENTS**

**L**ow-NOx **O**perable **A**mmonia-Combustor **D**evelopment (LOAD-Z)

- (1)
- UConn UConn<br>Experiments
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# Lab-Scale Experiment: Counterflow Flame Rig *Laminar (2D)*



– Stringent test of kinetic mechanisms, for comb. model development

Oxidizer

– Canonical representation of turbulent "flamelet"



# Counterflow Experiments – Measure Flow Uniformity & Strain



# Counterflow Exp. – Global vs. Local Strain Rate in NH<sub>3</sub> flames



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# Counterflow Experiments – Procedure for Measuring Extinction



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UCONN

- Sequence:  $-$  Establish stable, premixed NH<sub>3</sub> / air flames @ initial D<sub>SEPARATION</sub> …… twin flames fixed in space
- At fixed P,  $\phi$ : uniformly increase U<sub>JET</sub> & observe D<sub>SEPARATION</sub>  $\downarrow$ experiments
- $\frac{g}{g}$  modeling - Quasi-steady approach to  $\bigcup_{EXT}$  at <u>extinction</u>  $\bigrightarrow \frac{4U_{EXT}}{L}=a_{EXT\_GLOBAL}$

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# Counterflow Experiments – Extinction Strain Rate vs.  $\phi$ , p, T



# Counterflow Experiments – Press/Temp Effect on Extinction  $(3 \phi's)$



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# UConn Next Steps – ... Turbulent Flame Speed Rig for NH<sub>3</sub>

### Q4- Fabricate **Turb. Flame Speed** Rig:

Turbulent intensity range: 15−25%

- 1. NH<sub>3</sub> Bunsen Burner Outlet
- 2. Jet In Crossflow Port
- 3. Sharp-Edged Orifice Plate
- 4.  $H<sub>2</sub>$  Pilot Burner Plate
- **NH PLIF for turbulent flame structure imaging**
- **High-speed PIV for turbulent flow-velocity characterization**
- **Turbulent intensity enhancement utilizing jet-in-crossflow & contraction section [e.g. Michigan\*/Lund\*\* Hi-Pilot/DRZ]** \*J.Driscoll, Univ. of Mich.; \*\*M.Alden, Lund Univ.





### **L**ow-NOx **O**perable **A**mmonia-Combustor **D**evelopment (LOAD-Z) **Modeling NH3 Kinetics …. COUNTERFLOW & CHEM. REACTOR NETWORKS**

- $-$  Fundamental NH<sub>3</sub> flame data relevant to turbines:
	- $\bullet$  P, T  $\gg$  ambient  $\to$  relevant to compressor exit conditions .....
	- **Extrained & turbulent flames** …. …. ….
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- $-$  Predictive capability for NH<sub>3</sub> combustion & emissions
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- Develop & test NH<sub>3</sub> gas-turbine combustor "@ scale"
	- **Single-nozzle-rig (SNR) scale demo. @ high P, T ....**
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*FOCUS HERE:*

# **Computational Methods w/ Detailed Kinetics**

- Cantera, open-source computational framework developed by Dave Goodwin at Caltech
- Models developed for flame speed, chemical reactor networks (CRN), and counterflow flames (premixed & non-premixed)
- Additional tools developed to perform sensitivity analysis, including feature sensitivity (e.g. extinction strain rate), reaction path, chemical mechanism reduction and other diagnostic tools



**(iii) NH<sup>3</sup> Counterflow Flames**: Premixed and Non-Premixed



### **Premixed Counterflow Flames Non-Premixed Counterflow Flames**



**Lantera** 

# **Chemical Kinetic Mechanisms**

*Selected published, comprehensive N/H chemical mechanisms*

• Open-source computational framework developed by Dave Goodwin at Caltech Cantera

### • **Glarborg et al. Mechanism (2018 & 2022 versions)**

 [Glarborg, Miller, Ruscic, Klippenstein: Modeling nitrogen chemistry in combustion, Prog. Energy Combust. Sci. (2018) 31-68] [Glarborg: The NH<sub>3</sub>/NO<sub>2</sub>/O<sub>2</sub> system: Constraining key steps in ammonia ignition and N<sub>2</sub>O formation, Combust. Flame, Vol. 257 (2023)]

### • **Stagni et al. Mechanism (2020)**

[Stagni, Cavallotti, Arunthanayothin, Song, Herbinet, Battin-Leclerc, Faravelli: React. Chem. Eng. 5 (2020) 696–711]

### • **Powell & Papas et al. Mechanism- (2010 & 2011 versions)** → **RTRC**

[Powell, Papas, Dreyer: Hydrogen- and C<sub>1</sub>-C<sub>3</sub> Hydrocarbon-Nitrous Oxide Kinetics in Freely, Propagating and Burner Stabilized Flames, Shock Tubes, and Flow Reactors, Combust. Sci. Tech. 182 (2010) 252-283] [Powell, Papas, Dreyer: Flame Structure measurements of NO in Premixed Hydrogen-Nitrous Oxide Flames, Proc. Combust. Inst. 33 (2011) 1053-1062] **Ammonia Oxidation Pathway Schematic** 

• Two "Powell" mechanisms differ by only 3 different rate expressions for amine radical reactions:

 $(3) NH_3 + NH_2 \leftrightarrow N_2H_3 + H_2$  $(2) NH_2 + NO \leftrightarrow NNH + OH$  $(1) NH_2 + NO \leftrightarrow N_2 + H_2O$ 



# **RTRC Predictions vs. UConn Measurements**



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NH<sub>3</sub>/Air

# **Chemical Reactor Network (CRN) Model** → **ROM Design Tool**



▪ Overall "theoretical" NOx levels <30 ppm for a RQL architecture appear feasible





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\*\*  $NH<sub>3</sub>$  slip <1E-5 ppm at exhaust

 $*$  15% O<sub>2</sub> dry

### **L**ow-NOx **O**perable **A**mmonia-Combustor **D**evelopment (LOAD-Z) **Modeling NH3 Combustor …. CFD for NH3 NOx Estimates & Combustor Design**

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# **Modeling Activities & Next Steps ...... Toward GT Design**



 $-$  Single-nozzle high-pressure combustor, fired w/NH<sub>3</sub> fuel

### **Turbulent combustion regime for NH<sup>3</sup> -fueled gas turbines:**

- v' &  $\ell_0$  ~independent of fuel type (only aero dependent)
- $v_{L}$   $\downarrow$  w/ NH<sub>3</sub> fuel
- $\ell_{\text{\tiny L}} \, \uparrow$  w/ NH<sub>3</sub> fuel
- shift by  $\Sigma$ 5 10x

### **CFD modeling of NH<sup>3</sup> -fueled gas turbine combustor:**

- Challenges:
	- Efficient flamelet models uncertain (regime => stirred)
	- NOx cannot be post-processed (integral to comb. rxns)
- For efficient GT design calculations, *possible approach*:
	- Reduced NH<sub>3</sub>/NOx kinetics w/transport, e.g. EDC\*\*
	- Steady RANS turbulence model saves computational power for chemistry/transport



# **Developing CFD Approach for NH<sup>3</sup> Combustor Aero Design**

### **Begin CFD methodology evaluation w/H<sup>2</sup> combustion case from existing RTRC exp. rig:**

- Change *from* LES turbulence model *to* steady-RANS (realizable k-eps.)
- Change *from* 2-mixture-fraction/Zimont combustion model *to* EDC w/species transport
- Start w/complete  $H_2$  combustion chemical mechanism (10-species)
- Then add NOx mechanism (coupled w/H<sub>2</sub> mechanism) & evaluate emiss. predict. (12-species)
- Then add NH<sub>3</sub> combustion mechanism (31-species)



LES w/out NOX LES w/out NOA<br>prediction capability<br>prediction capability prediction capability<br>
(no species transport)

*RANS/EDC/LLNL H2/Air Mech. & Stagni NOx*



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# *RANS/EDC/LLNL H2/Air Mechanism* within factor<br>1 to 1.5 for three<br>1 to 1.5 FARs...

#### GT2024-121321

Proceedings of ASME Turbo Expo 2024 **Turbomachinery Technical Conference and Exposition** GT2024

June 24-28, 2024, London, United Kingdom



FIGURE 5. FT4000 Single Nozzle Combustor Rig (SNR).



FIGURE 7. CFD analysis of engine (a) and rig (b) combustor velocity flowfields. 19

# **Initial CFD Simulation of NH<sup>3</sup> Combustion in Existing Geom.**

#### *RANS/EDC/LLNL H2/Air Mech. & Stagni NOx*



#### *RANS/EDC/Stagni Mech. for Combustion & NOx*



#### GT2024-121321

Proceedings of ASME Turbo Expo 2024 **Turbomachinery Technical Conference and Exposition** GT2024 June 24-28, 2024, London, United Kingdom



#### FIGURE 5. FT4000 Single Nozzle Combustor Rig (SNR).



FIGURE 7. CFD analysis of engine (a) and rig (b) combustor velocity flowfields.



# **Fundamental Flat-Flame Data …. RTRC EXPERIMENTS**

**L**ow-NOx **O**perable **A**mmonia-Combustor **D**evelopment (LOAD-Z)

(1)

RTRC

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- RTRC<br>Experiments **RTRC RTX Technology Research Center**

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# **High-Pressure Flat-Flame Burner Rig**

### Goal: Characterize  $NO<sub>x</sub>$  emissions from laminar flame at high pressure for model validation

*Scarce data available on NO<sup>X</sup> formation in NH<sup>3</sup> flames at >>10 atm Target system operation >20 atm*

*Allows evaluation of staged (RQL) NH<sup>3</sup> combustion*



**D=1.54"**

perforation pattern:  $d = 0.5$  mm  $l = 0.7$  mm

**Burner Surface ~2400 holes, D=0.020"**









*Methane Combustion*



*Ammonia Combustion*





# **Facility Preparations for High-Pressure Operation**

### **RTRC configuration & capability:**

- 425 to >500+ K preheat capability (>25-atm w/new heater install.)
- 10 to >20-atm pressure vessel & feeds ... fuel, air, & N<sub>2</sub> chamber flow

### **Facility Preparations – Completed**

- High-pressure vessel w/ pass-throughs
- Ventilated experiment enclosure
- Interlocked PLC control system
- High-pressure air &  $N_2$  supply
- Inlet air heater (low pressure)
- Remote flow control system for air,  $\mathsf{CH}_4$ , NH<sub>3</sub>, and N<sub>2</sub>
- Backpressure valve for chamber pressure control

### **Facility Preparations – In Progress**

- *Inlet air heater (high-pressure)*
- **High-pressure liquid ammonia** *pumping cart w/ refrigerated pumps*
- Heater controllers for air and  $NH<sub>3</sub>$ heaters





# **Emissions Sampling System**

### **Emissions Sampling**

- Air-cooled sample probe
- Probe traversable using translation stage system
- Heated sample line to prevent condensation
- NO<sub>x</sub> analyzer system with ammonia scrubber & dryer
- Heated,  $NH<sub>3</sub>$  compatible FTIR system accepts wet sample
	- FTIR species:  $H_2O$ , NH<sub>3</sub>, NO, NO $_2$ , N $_2$ O
- Sample chain can be easily swapped from conventional analyzer to FTIR system
- Sample probe can be swapped with B-type TC-probe for heat loss characterization





*Conventional NO<sup>X</sup> Analyzer*



Wavenumber [cm]





# Flat-Flame NO<sub>X</sub> Emissions Results – Moderate Pressure<br>*Lines* = Stagni Mechanism Predictions



- **NO<sup>X</sup> analyzer used to collect emissions data across multiple pressure conditions**
- 425 K preheat, 9-15 cm/s burner velocity
- *x=0.67D*  1D sample-probe location
- Stable burner operation achieved across pressures and equivalence ratios at 425 K
- Adiabatic flame model used w/ Stagni mech.
- FTIR data at select 7-atm conditions shows high NH $_{\rm 3}$  comb. efficiency (<75 ppm NH $_{\rm 3})$

#### $NH<sub>3</sub>$  / air flames  $@P = 3$ -atm

- Experimental challenges observed when translating probe to
- Measured oxygen concentration increases at locations
- Downstream cooling air entrainment is likely occurring
- **Further system optimization needed to reduce pathways for entrainment before additional measurements**



 $\phi =$ 

ი = 1.0

# **Next Steps for High-Pressure NO<sub>X</sub> Measurements**

### **Next Steps for Improved Emissions Data**

- Improve sampling/burner system for reduced air entrainment
- Collect additional data using  $\mathsf{NO}_{\mathsf{X}}\!\mathsf{/O}_{\mathsf{2}}$  analyzer system to confirm reduced entrainment
- Collect temperature profile along centerline to characterize burner heat loss
- Collect additional emissions data using FTIR to measure other species

### **Improvements for Higher Pressure Operation**

- Integrate high-pressure liquid pumping cart to allow higherpressure gaseous NH $_3$  input
- Integrate liquid-pumping cart with existing PLC infrastructure
- Finish installing high-pressure-rated heater & controllers

### **Laminar burner operating well up to 7 atm**

**Further improvements underway to improve & validate data quality and achieve higher pressure operation >20 atm**







# **BACKUP**



UNIVERSITY OF CONNECTICUT



# **RTRC Predictions vs. UConn Measurements (1/2)**



NH<sub>3</sub>/Air

# **Counterflow NH<sup>3</sup> Flames w/Heat Loss** → **Impact & Lean Limit**



C-shaped curve for counterflow premixed fuel-lean, ammonia-air flames showing:

- **i.** adiabatic stretch-induced stretch rate  $\kappa_{\text{ext}}$  ad (solid line),
- **ii**. non-adiabatic stretch-induced stretch rates  $\kappa_{\text{ext}}$ <sub>S</sub> (dashed line), and
- iii. radiative-induced stretch rates  $\kappa_{\text{ext,R}}$  (+ symbols).

#### **Take-Away:**

- **Heat-loss weakens low-burning-rate NH<sup>3</sup> flames**
- **Stretched high-burning-rate flames more robust**



Fang, Papas, Sung, Stevens, & Smith, "Effects of radiative heat loss on extinction limits of counterflow premixed ammonia-air flames," *Proceedings of the Combustion Institute, Vol. 40, doi: 10.1016/j.proci.2024.105569 (2024).*

## **Feature Sensitivity to Extinction Strain Rate**

 $(1) NH_2 + NO \leftrightarrow N_2 + H_2O$  $(2) NH_2 + NO \leftrightarrow NNH + OH$ (3)  $NH_3 + NH_2 \leftrightarrow N_2H_3 + H_2$ 





### **Sensitive rate constants:**

*NH<sup>2</sup> /NO Interactions*  $(1 b) NH_2 + NO \leftrightarrow NNH + OH$  *NH<sup>i</sup> /N2H<sup>i</sup> Reactions*

 $(1a) NH_2 + NO \leftrightarrow N_2 + H_2O$   $(2a) N_2H_2 + M \leftrightarrow NNH + H + M$  $(2b) N_2 H_3 + M \leftrightarrow N_2 H_2 + H + M$ 



30

# **END**

