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## UNDUNIVERSITY OF NORTH DAKOTA

## Electromagnetic Energy-Assisted Approaches to Convert Fossil Fuels to Low Cost Hydrogen

Grant Number: DE-FE-0032061 2024 FECM / NETL Spring R&D Project Review Meeting PI: Johannes G. van der Watt (University of North Dakota) *PM: Heather Hunter (DOE/NETL)* 





## **Program Overview**

<u>Opportunity</u>: Produce H<sub>2</sub> from methane rich sources without CO<sub>2</sub> formation

**Problem:** Carbon deposition and catalyst deactivation

•  $CH_4(g) \xrightarrow{catalyst} C(s) + 2H_2(g)$ 

• Catalyst loss – either reactivity or physical loss

Goal: Extend H<sub>2</sub> production through targeted catalyst regeneration

Solution approach: Use an in-situ electromagnetic (EM) energy-assisted mechanism to regenerate catalysts

# **Potential Significance – Catalyst Regeneration**

Reduce overall catalyst replenishment cost

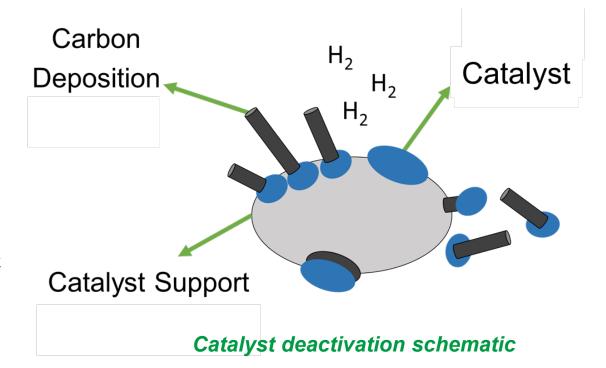
- *Purer* carbon products less catalyst poisoning
- Low-cost & CO<sub>2</sub>-free process for H<sub>2</sub> production
- Enable wider adoption of H<sub>2</sub>-related technologies from fossil resources
- Applicable to other catalyst-based processes coking issues

# **Background – CH<sub>4</sub> Decomposition**

Thermo-catalytic decomposition of CH<sub>4</sub>

- $CH_4(g) \rightarrow C(s) + 2H_2(g) \Delta H_{rxn} = 37 \text{ kJ/mol-H}_2$ 
  - Electrolysis 285 kJ/mol-H<sub>2</sub>
  - Steam methane reforming (SMR) 41 kJ/mol-H<sub>2</sub>
- Typically conducted at 500 800°C

- Typical catalysts include: Transition metals, even C
- Without catalyst: 1200°C requirement



# Background – Catalyst "Cleaning" Approaches

- Physical removal attrition
  - C remains same

- Potential catalyst loss & impure catalysts with C
- Chemical removal Combust/Gasify carbonaceous deposits
  - Combustion:  $O_2 + C \rightarrow CO_2$
  - Gasification:  $H_2 + C \rightarrow "CH_4"$  and  $H_2O + C \rightarrow CO + H_2$
  - Chemical Acid leaching to remove and recover metal catalysts

# Approach

Technique:

- In-situ carbon removal using an Electromagnet (EM) Energy-Assisted Thermocatalytic Process
- Mild process applicable to catalyzed reaction temperatures
- Target periodic cleaning
- Mechanism chemical and physical C removal

How:

- Conduct proof-of-concept experiments
- Supplement with computational modeling
- Assess large-scale applicability of approach (integrated setups)

# **Technical Summary**

Task 1 – PMP and Technology Maturation Plan

- Task 2 Catalyst Preparation and Performance Testing
- Task 3 Computational Fluid Dynamics (CFD) Modeling of Conversion System

Task 4 – Hydrocarbon Conversion Testing using EM Energy-Assisted Thermocatalytic Process

Task 5 – Component Identification for Future Work

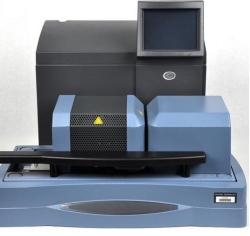
Task 6 – PMP (North Dakota Industrial Commission - NDIC)

- **Task 7 Evaluation and Characterization of Carbon Products**
- Task 8 Process Model of Greener H<sub>2</sub> Production System

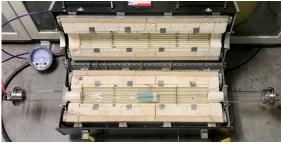
Extended Scope

- Subtask 2.1 Catalyst Preparation
- Catalyst/Support structures identification
- Combinations of different supports tested
  - SiO<sub>2</sub>, C, Al<sub>2</sub>O<sub>3</sub>, and Aerogel Supports
- Catalysts
  - Ni, C and Fe

- Identified promising Ni-SiO<sub>2</sub> catalyst
  - Added Cu-promotor disperses Ni more evenly



TGA Tests (TA SDT Q600)



2-inch tube furnace



Fluidized/Fixed bed testing in 3/8-inch tube reactor

Kinetics for CFD

- Conduct kinetic study using TGA
- Ensure sufficient gas flow mitigate external mass transfer limitations (initially)
- Initial tests determine optimum flow rate & sample mass
  - 500-650°C & 30%-50% CH<sub>4</sub> (N<sub>2</sub> balance)
  - Eventually use CFD to develop geometries complementing EM-assisted process



Pre- and post-test with Ni-SiO<sub>2</sub> catalyst

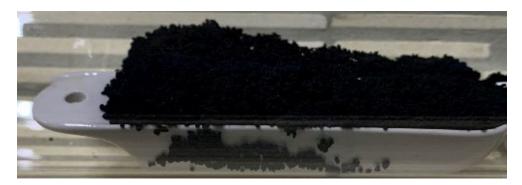
Subtask 2.2 – Initial High Temperature Catalyst Performance Testing

- TGA testing scaled up to tube furnace
- Tube furnace easy integration EM Energyassisted thermocatalytic process
- Just under 70% CH<sub>4</sub> conversion at 650°C

Key result - moldable catalyst with suitable performance



**Pre-test material** 



Post-test material with carbon buildup

- Structured catalyst testing
- Required flat surface for C-deposition and C-removal
- Decided on disk-shape (compatible with tube furnace
- Placed perpendicular to gas flow
- Run short durations for C-growth

 Excessive run time – carbon growth destroys surface



**Tube Furnace Testing - Structured** 

# Task 3: Task 3 – Computational Fluid Dynamics (CFD) Modeling of Conversion System

#### Subtask 3.1 – Establish Baseline CFD and Kinetic Model

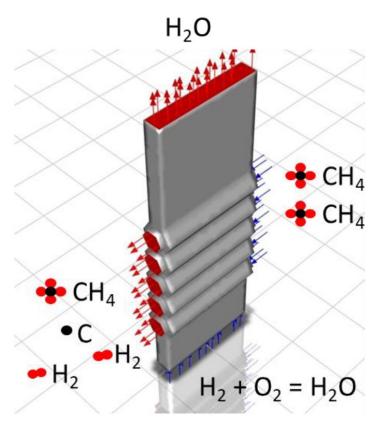
Capture decomposition reaction behavior

Subtask 3.2 – Update CFD and Kinetic Model to Include Effects of the Electromagnetic Energy-Assisted Mechanism

• Assess geometries/setups enhancing removal technique

Results: 2D & 3D models

- Depicting zones of high C-deposition
- Temperature profiles (heat addition points)
- Locational aspects of EM Energy-Assisted Mechanism



**Reaction Modeling** 

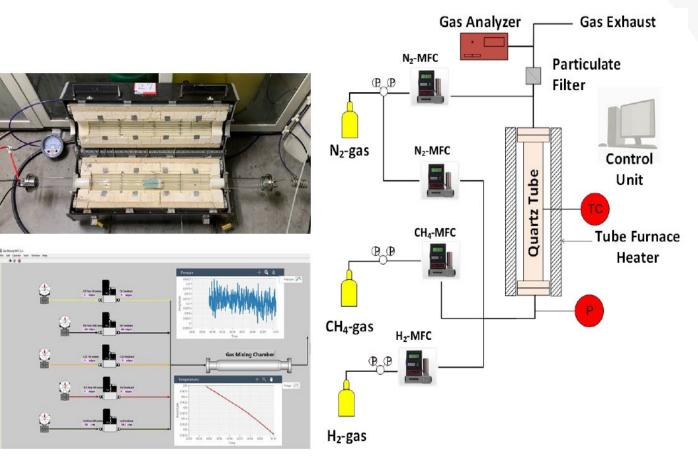
#### Series of experiments devoted to

- Decomposition testing
- Surface regeneration

#### Assembled integrated test unit

- 2"-ID glass reactor
- Mass flow controllers
- Gas analyzer

EM Energy-Assisted Module (not shown)



**Experimental Unit Setup** 

- Structured catalyst/support testing
- Initial tests failed (inset images)
  - Aggressive CH<sub>4</sub> decomposition (too long)
  - Refined binding technique

- Process refined: samples subjected to ~10 minutes of CH<sub>4</sub> decomposition
- Thin layer approach lines up with CFD and viable C-removal technique

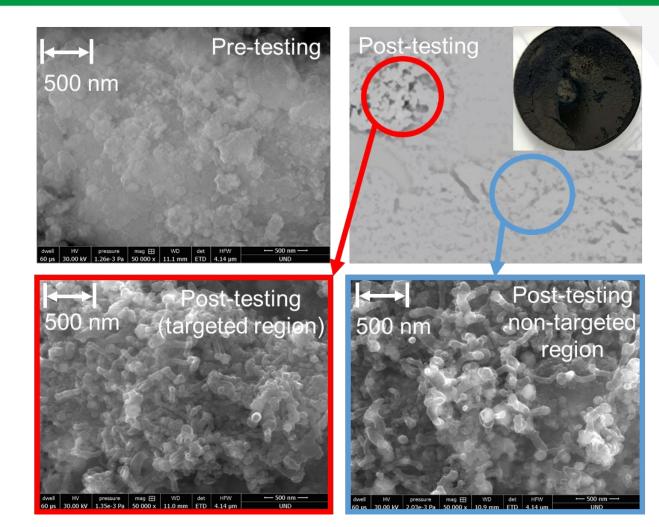


Pre-decomposition structured catalyst/support disk

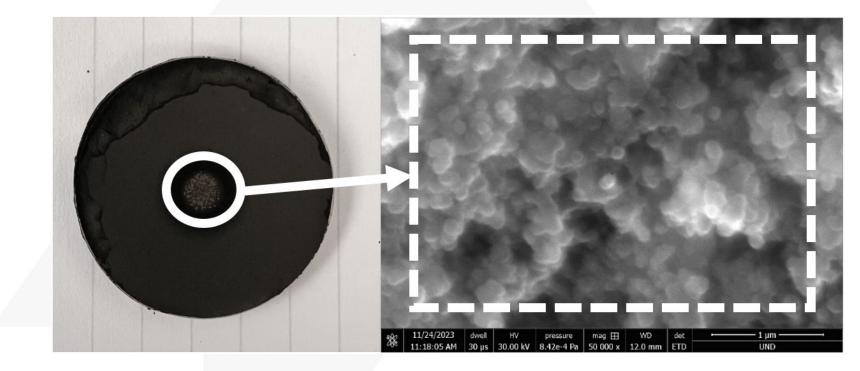


Post-decomposition structured catalyst/support disk

- EM-assisted mechanism on structured catalyst
- Ambient testing no effect
- At 650°C visible surface changes
- SEM-EDX on removal/non-removal area
- Tubular-shaped C visually detected both cases
- EDX inconclusive (rough surface)



- Increased regeneration intensity following CH<sub>4</sub> decomposition
- Post cleaning surface indicating lack of carbonaceous material
- Excessive cleaning  $\rightarrow$  catalyst inactivity



Catalyst surface after decomposition, regeneration and additional decomposition cycle

#### Change operating regime

• Alter regeneration intensity (<5 min.)

Start with:

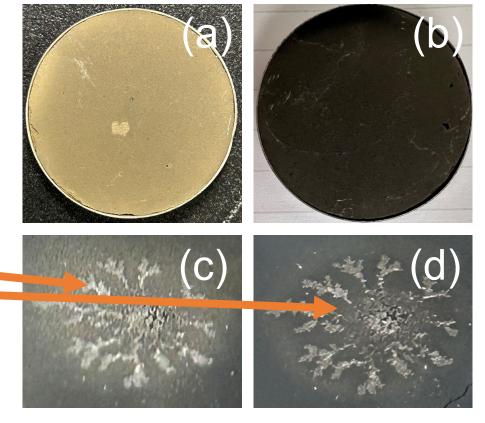
• Structured catalyst (a)

- Decompose methane (b)
- Surface after catalyst regeneration (c)
- Surface after additional decomposition stage (d)

High intensity regions no activity

Surrounding areas darkened (carbon deposition)

Ongoing: Assess surfaces for carbon deposition postregeneration (TGA, SEM, TC/TIC)

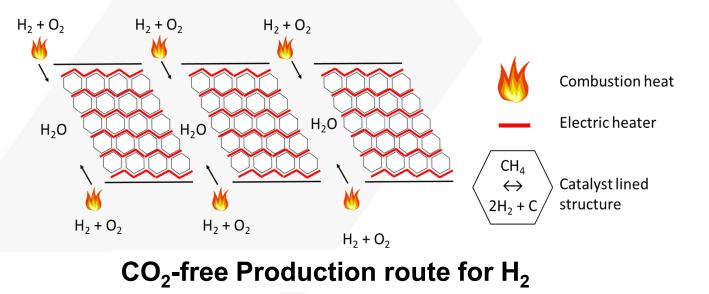


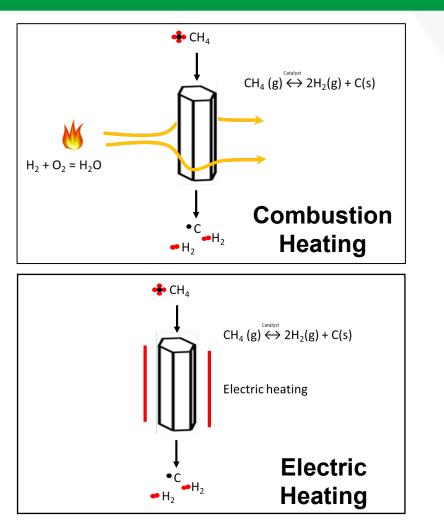
Carbon buildup and removal sequences

## Task 5: Component Identification for Future Work

### System Design:

- Design large-scale system for CDM process
- CO<sub>2</sub>-free process (with renewable electricity)
- 1 tonne H<sub>2</sub> per hour





# **Extended Scope**

- Cost-share provided by North Dakota Industrial Commission (NDIC)
- Scope extended to:
  - Examining uses for carbonaceous products
  - Implementing strategy for producing H<sub>2</sub> using process at large scale
  - Focus on H<sub>2</sub> production without CO<sub>2</sub> formation
  - Project end date: 07/31/24 (DOE and NDIC)
- Added Tasks 6, 7 and 8 to project
- Task 6 PMP for NDIC

# Task 7: Evaluation and Characterization of Carbon Products

Task 7: Evaluation and Characterization of Carbon Products

- Produced bulk carbon sample
- Tested carbon purity
- Acid leached carbon

- Prepared coin cells and testing
- Compare to commercially procured carbon



**Bulk carbon production** 



**Coin cell production** 



Coin cell testing



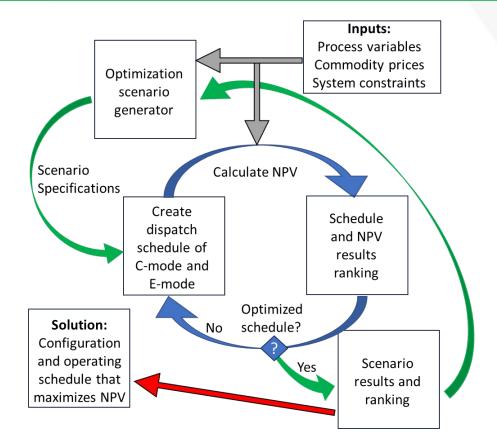
# Task 8: Process Model of Greener H<sub>2</sub> Production System

#### Approach

- CH<sub>4</sub> decomposition for H<sub>2</sub>
- C-mode and E-mode for production flexibility
- CO<sub>2</sub>-free process (with renewable electricity)

### Key Questions

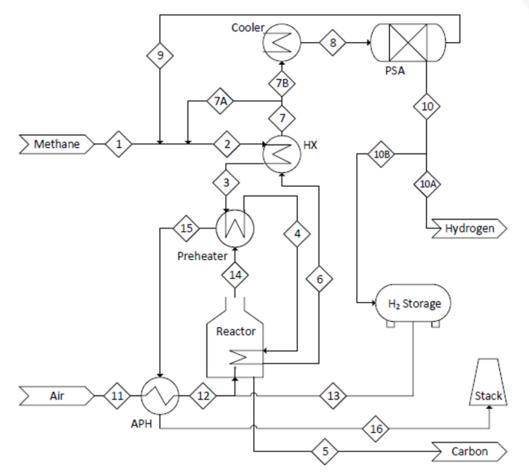
- Time ratio in each mode
- Impact of electricity and hydrogen prices
- Optimal hydrogen storage



Electric and Combustion Heating Loop Optimization Routine Depiction

# Task 8: Process Model of Greener H<sub>2</sub> Production System

- H<sub>2</sub> Production Scenarios:
  - ND-MISO
  - NREL future system scenario analysis
  - H<sub>2</sub> supply by pipeline or liquid tankers
- Next steps:
  - Getting Plant CAPEX
  - Process optimization (sizing variations)
- Inset figure general process layout
  - 1 tonne H<sub>2</sub> per hour



#### **CO<sub>2</sub>-free H<sub>2</sub> Production Pathway**

# **Future Work**

Task 1 – PMP and Technology Maturation Plan

Task 2 – Catalyst Preparation and Performance Testing

Task 3 – Computational Fluid Dynamics (CFD) Modeling of Conversion System

Task 4 – Hydrocarbon Conversion Testing using EM Energy-Assisted Thermocatalytic Process

Catalyst regeneration verification – SEM, TGA and TC/TIC

Task 5 – Component Identification for Future Work

Task 6 – PMP (NDIC)

Task 7 – Evaluation and Characterization of Carbon Products

Compare lab-grown C in batteries to commercially procured C

Task 8 – Process Model of Greener H<sub>2</sub> Production System

Update equipment costing from simulation results and input information for inner-outer loop approach for process optimization

# Disclaimer

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# Thank you Questions?

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