

Effect of Impurities on Supercritical Carbon Dioxide Compatibility (FWP-FEAA144)

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 - Tenaris
 - EPRI
 - Sam Sham, INL (US heat, alloy 709)

Supercritical CO₂ Allam cycle has been demonstrated!

NetPower 25MWe test facility (Texas)

Exelon, Toshiba, CB&I, 8Rivers Capital: \$140m



The prototype NET Power plant near Houston, Texas, is testing an emission-free technology designed to compete with conventional fossil power.

CHICAGO BRIDGE & IRON

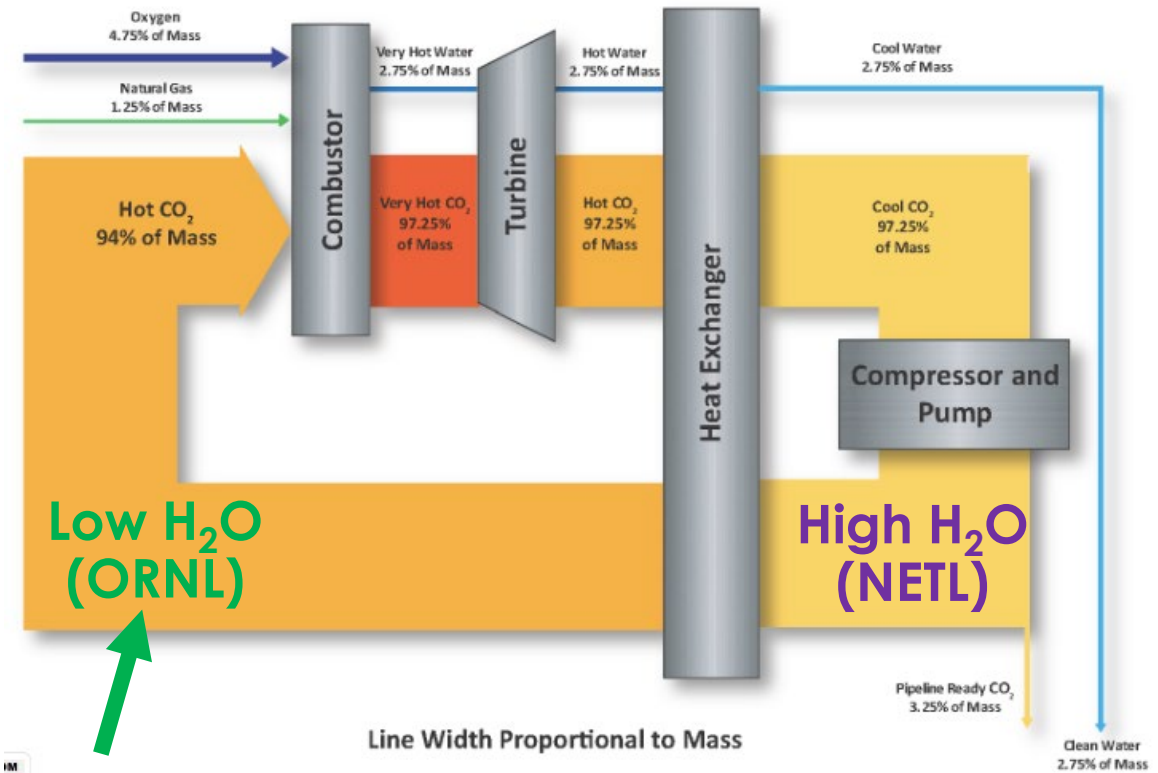
November 2021: plant connected to grid (announced plans for 280MW)

Material challenges:

Combustor: 900°C (initially 1100°C)

Turbine exit: 750°C/300 bar

Combustion impurities: O₂, H₂O, SO₂?



Low H₂O (ORNL)

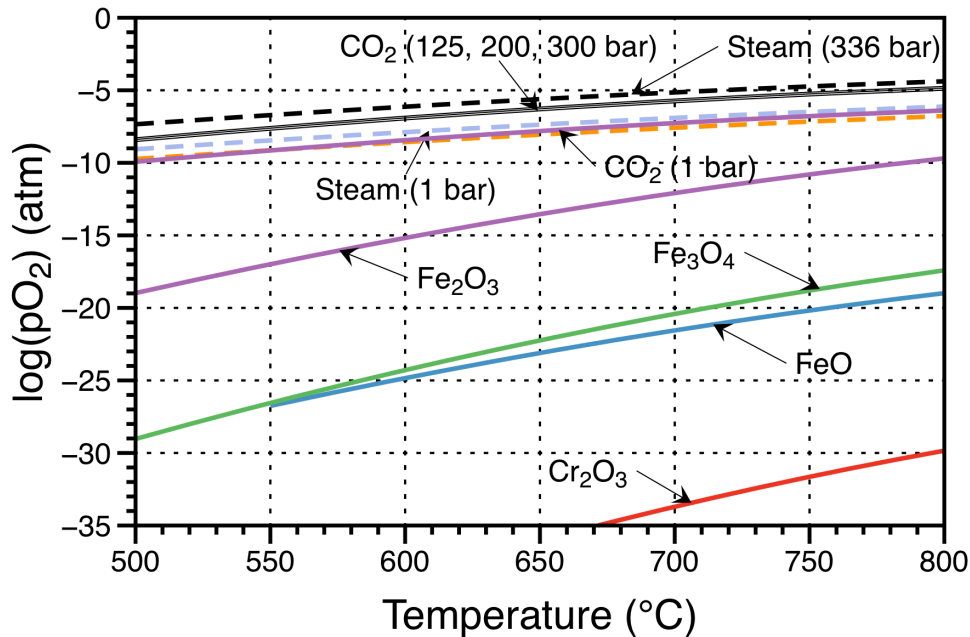
High H₂O (NETL)

Burning natural gas in sCO₂ creates impurities...what is effect?

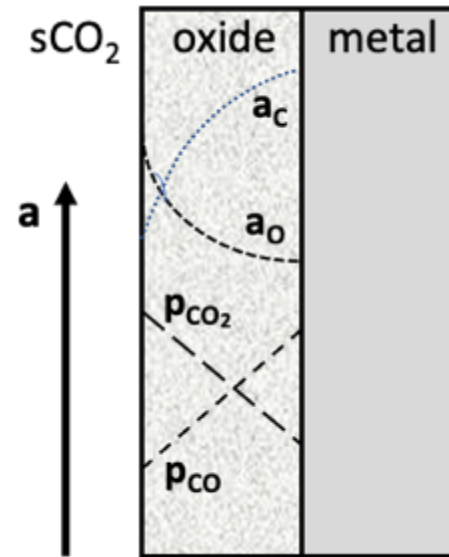
Thermodynamics: Oxygen levels similar in steam/CO₂

Concern about high C activity at m-o interface: not "inert"

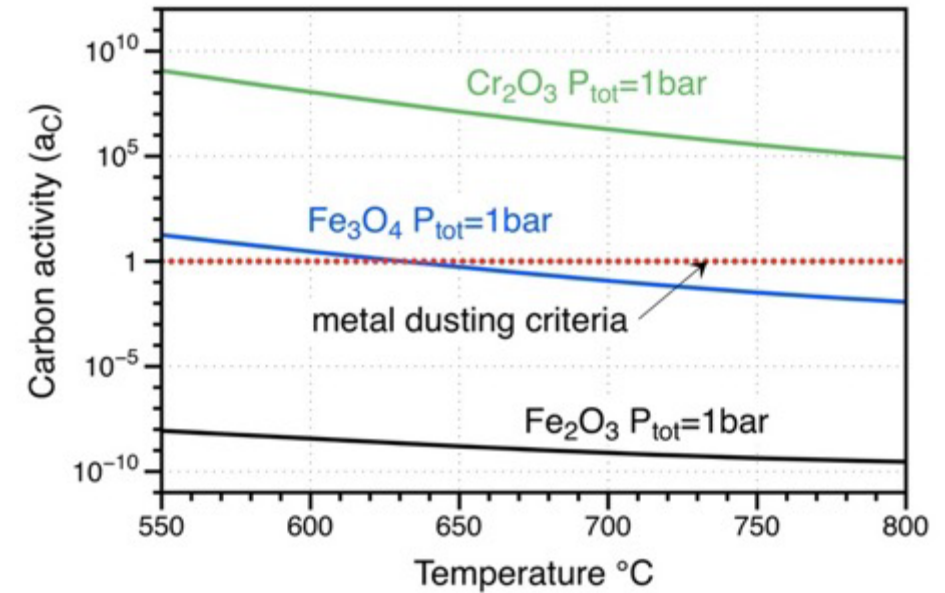
Factsage calculations:



All of our favorite oxides are stable!



Young et al., 2011



High carbon activity at $P_{total} = 1$ bar (What is $P_{interface}$?)

ORNL sCO₂ project wrapping up in 2024 (originally 2023)

- Focus on 4 steels
 - BCC: 9Cr (G91), 12Cr (VM12)
 - FCC: 316H (16Cr-10Ni), 709 (20Cr-25Ni)
- Completed tasks
 - Reaction kinetics with and without impurities (1%O₂+0.1%H₂O)
 - Degraded protective Cr-rich oxide on 316H and 709
 - Evaluated 3 pack cementation coatings: G91+Cr, G91+Al, 316H+Cr
 - Cr: internal carburization observed in both alloys, less protective in sCO₂+O₂+H₂O
 - Al: ~20wt.%Al not sufficient to form protective Al-rich oxide in sCO₂ at 600°/650°C
- Tasks wrapping up
 - Creep testing Ni-base alloy foils (sCO₂ HX application)
 - Foils can show decreased strength compared to bulk material
 - More strongly affected by environment (example 1000 h 800°C air+10%H₂O)
 - Modeling C ingress
 - Determine maximum use temperature for steels to avoid C embrittlement

ORNL steel project started in August 2019

Test matrix & progress

Temperature	RG sCO ₂	+1%O ₂ +0.1%H ₂ O
450°C (842°F)	2,000 h	1,000 h
550°C (1022°F)	2,000 h	1,000 h
650°C (1202°F)	2,000 h	1,500 h



**Autoclave: 300 bar sCO₂
500-h cycles**



~5 cm² alloy coupons + tensile specimens

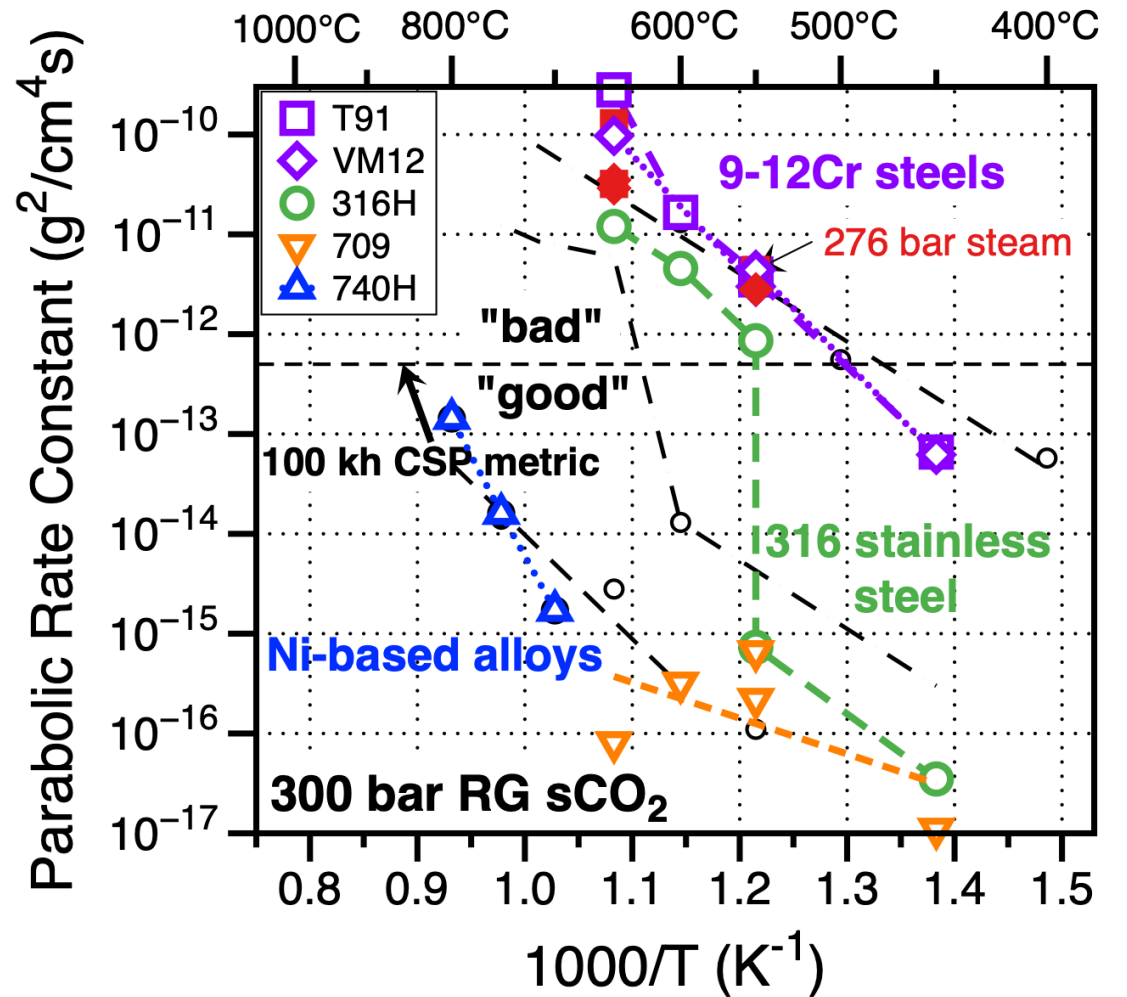
- Four primary alloys in test matrix
 - T91: Fe-9Cr-1Mo, creep strength enhanced steel
 - VM12: Fe~12Cr-Co-W
 - 316H: conventional stainless steel
 - 709: advanced austenitic, 20Cr-25Ni+Nb
- 10 specimens of each alloy
- With and without impurities (open vs. closed)

Alloy	UNS	Cr	Ni	Mn	Si	C	N	Other
Gr.91	K90901	8.6	0.3	0.5	0.4	.10	.05	0.9Mo,0.2V
VM12	12CrCoW	11.5	0.4	0.4	0.4	.12	.04	1.6W,1.5Co
316H	S31609	16.3	10.0	0.8	0.5	.04	.04	2.0Mo,0.3Co
NF709	S31025	20.1	25.2	0.9	0.4	.06	.15	1.5Mo,0.2Nb

Baseline of research grade (RG) CO₂: ≤ 5 ppm H₂O and ≤ 5 ppm O₂

Measured rates in sCO₂ consistent with the literature

- Metric developed for Solar CSP
 - Slow rate = OK for 100kh life
- Ni-based alloys all “good”
 - Lifetime model: $\leq 800^{\circ}\text{C} = 100\text{kh}$
- **Steel limitations**
 - Ferritic-martensitic alloys $< 500^{\circ}\text{C}$
 - Austenitic alloys $< 600^{\circ}\text{C}$
 - Obvious jump in kinetics
 - Advanced austenitics (709), better
 - Value in 20-25%Cr, 20-25%Ni
 - Low values hard to measure

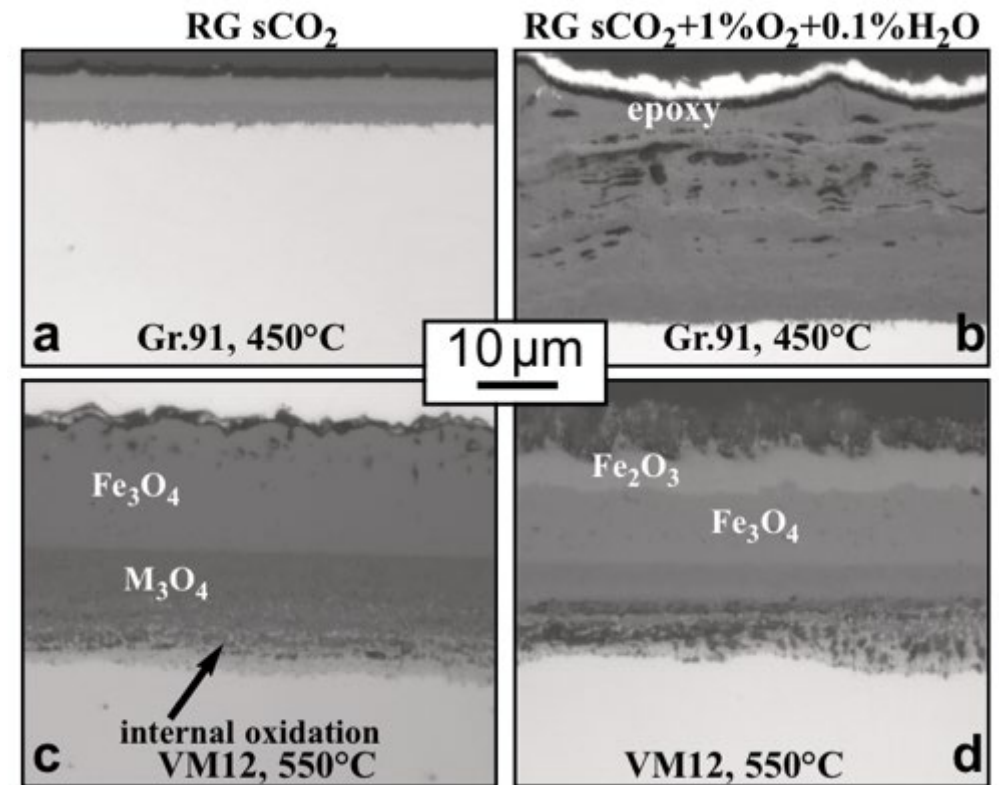
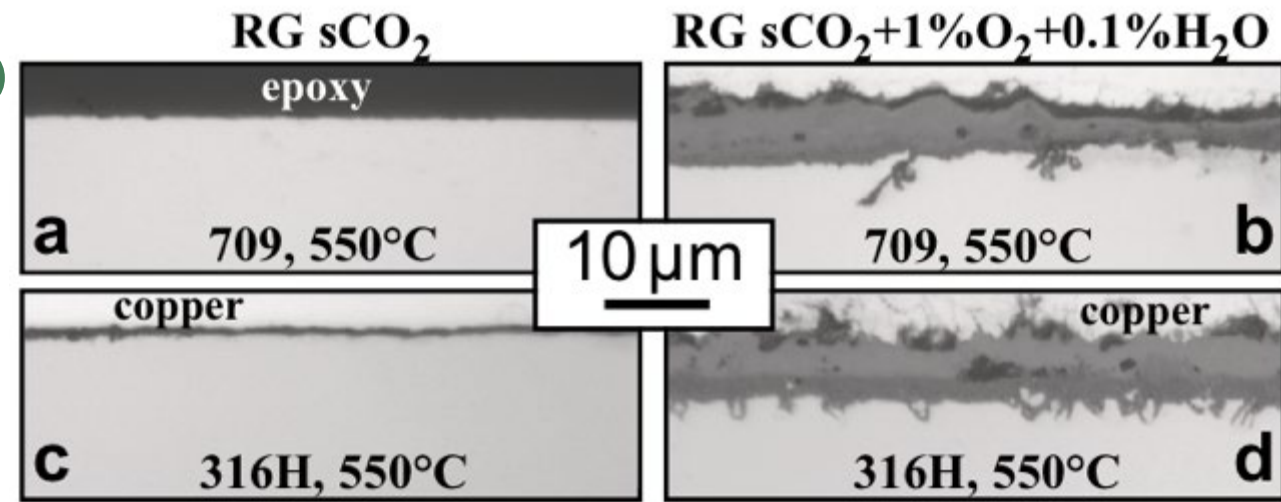


9-12Cr steels have similar rates in 276 bar steam

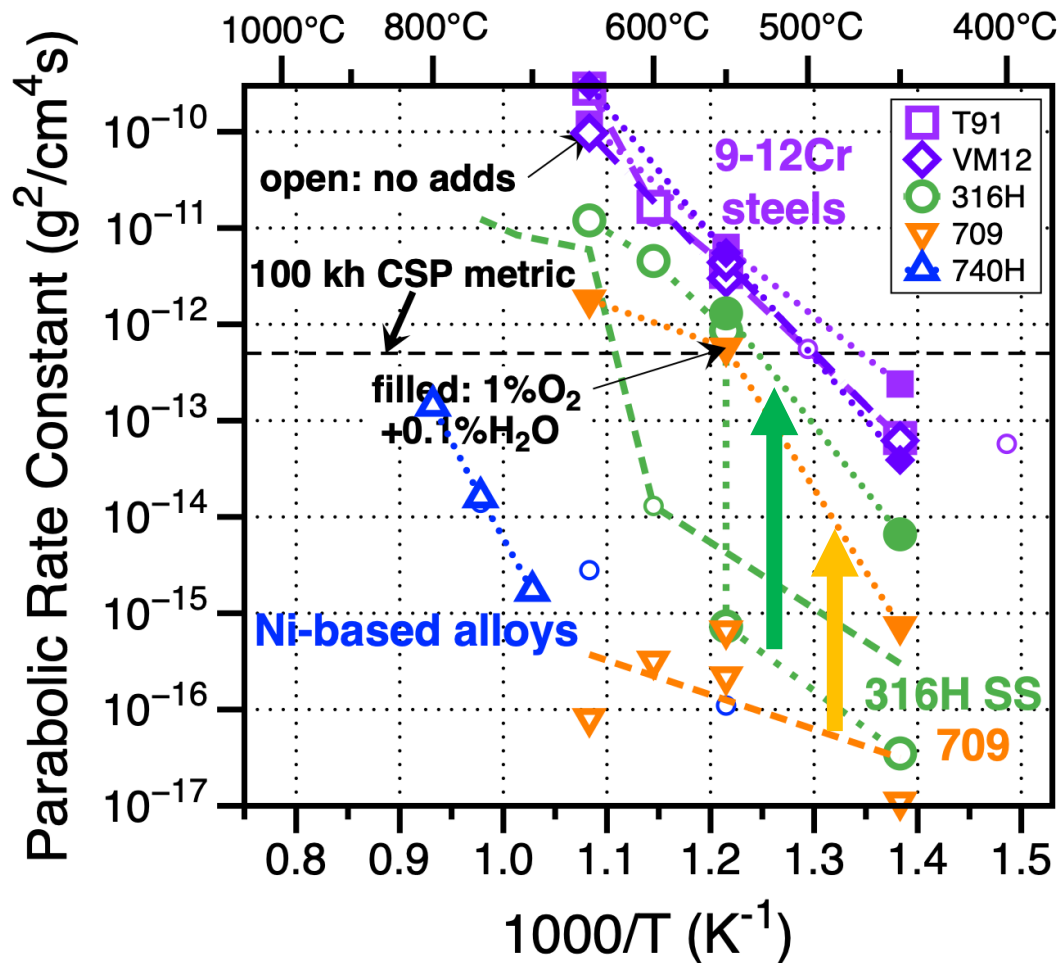
SS: faster attack with O_2/H_2O

FeCr: Fe_2O_3 formation

- Stainless steel (316/709)
 - RG sCO_2 : thin Cr-rich scale
 - Good C barrier
 - Impurities: duplex scale
 - Common with steels
 - Now is C ingress possible?
- 9-12Cr steels
 - 450°C: increased T91 attack
 - 550°-650°C
 - Clear duplex scale in both cases
 - With 1% O_2 form Fe_2O_3 layer
 - VM12: no benefit of higher Cr content after 1,000 h



Addition of 1%O₂+0.1%H₂O: accelerates SS mass gains!



Bad
Good

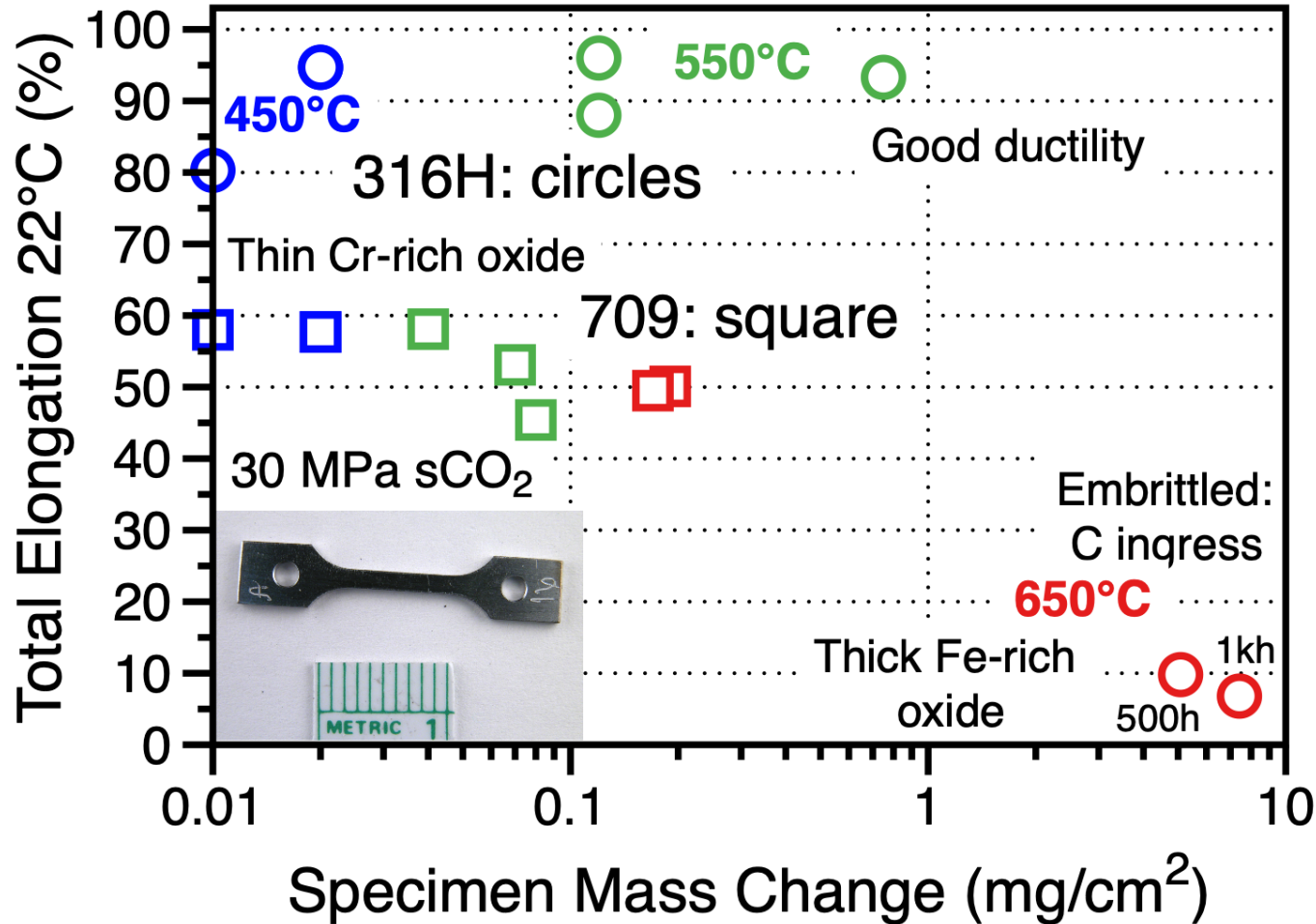
Performance metric from CSP

Open symbol: RG sCO₂
Shaded: RG sCO₂ + 1%O₂ + 0.1%H₂O
 - no change for Gr91 and VM12
 - higher rates for 316H and 709

Based on CSP (solar) metric: all limited to <550°C with impurities
 Rates for 709 (UNS S31025): may not reflect steady state at 1000 h

RG sCO₂: combination of mass change & 25°C ductility to illustrate issue with 316H at 650°C (709 not affected)

High ductility & low mass gain



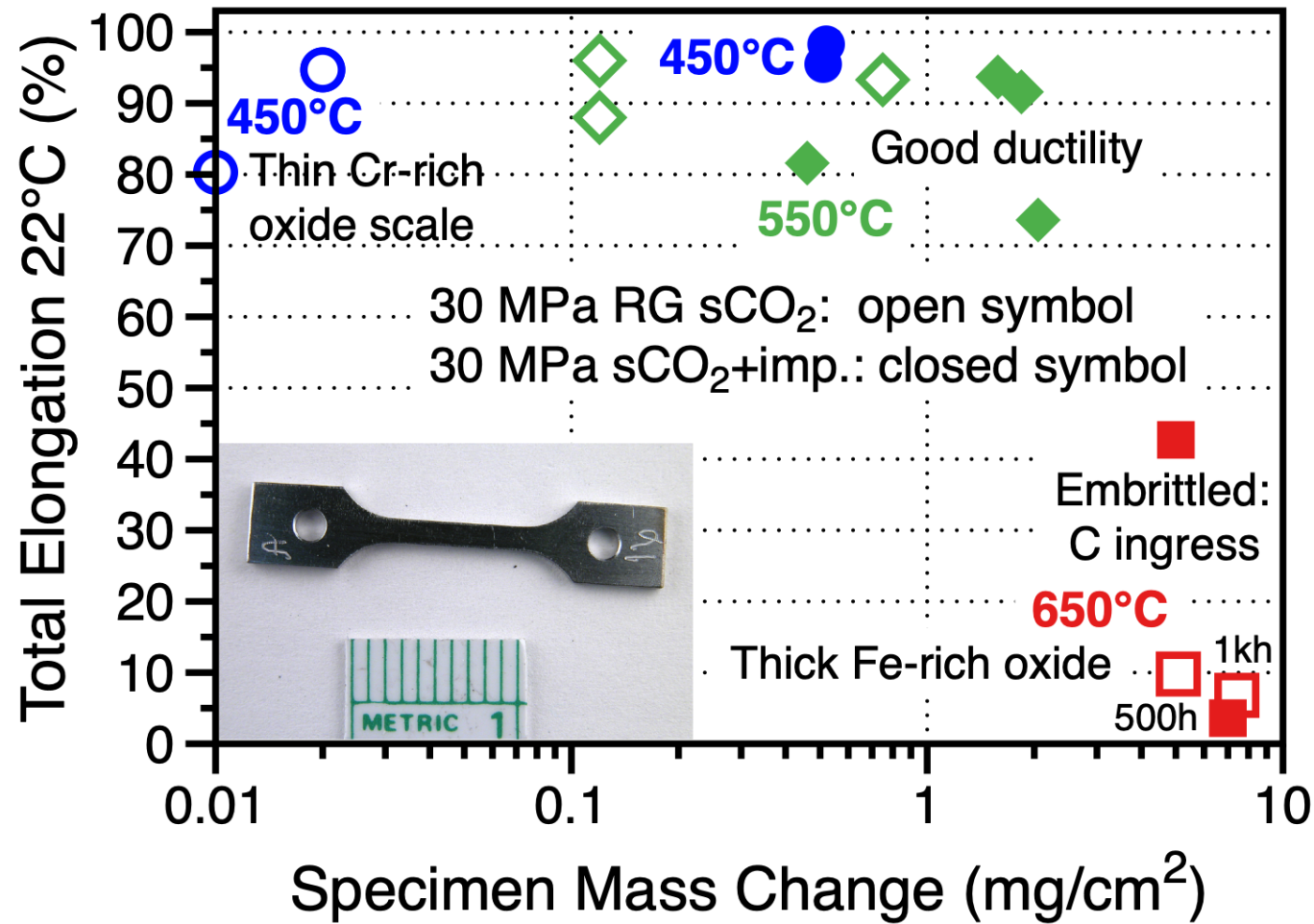
500h & 1000 h measurements for each stainless steel

Low ductility & high mass gain

316H: not much change in 25°C ductility with impurities

More mass gain, similar embrittlement at 650°C

High ductility & low mass gain

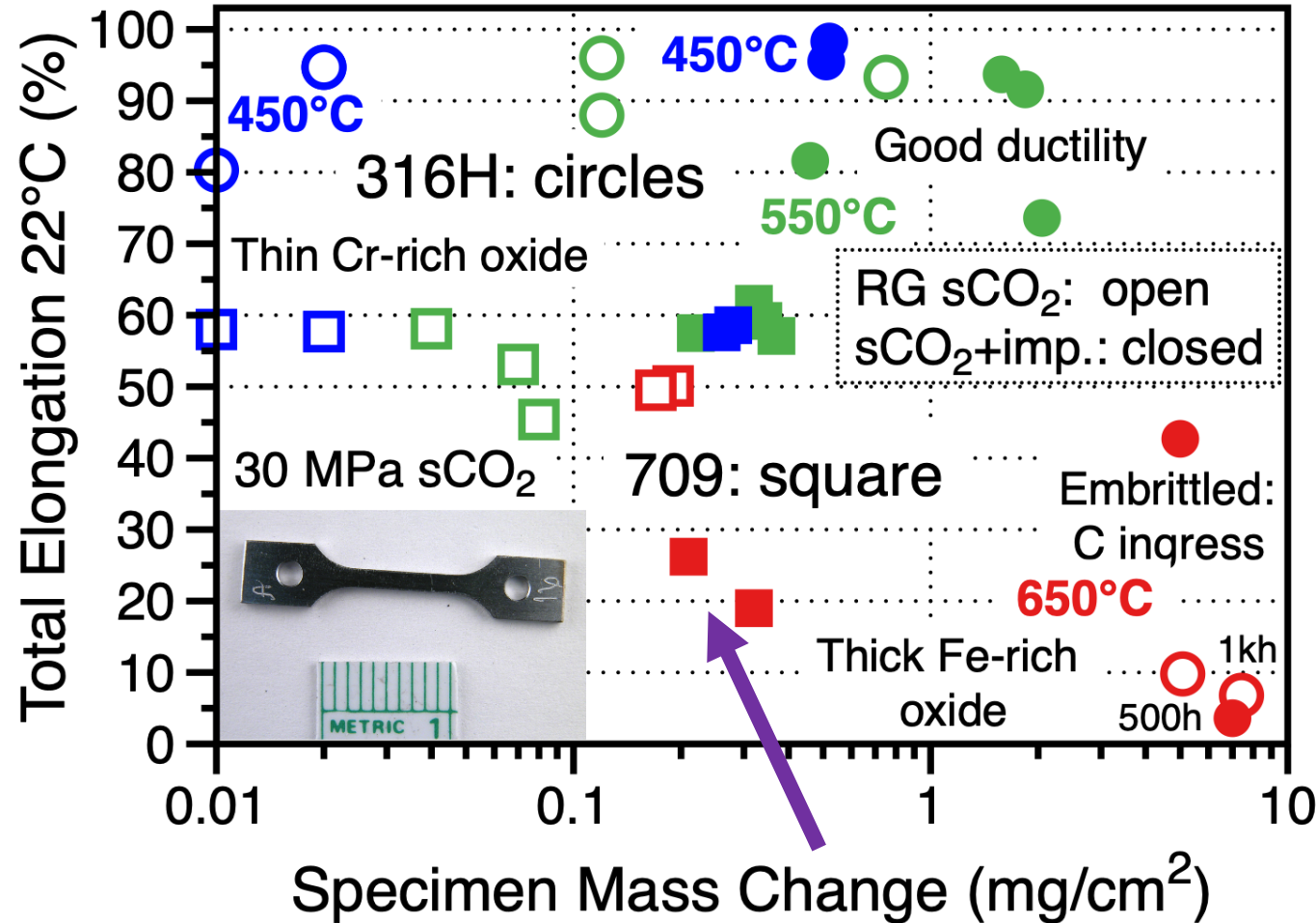


500h & 1000 h measurements for 316H

Low ductility & high mass gain

709 also showed drop in 25°C ductility at 650°C with the addition of O₂/H₂O impurities

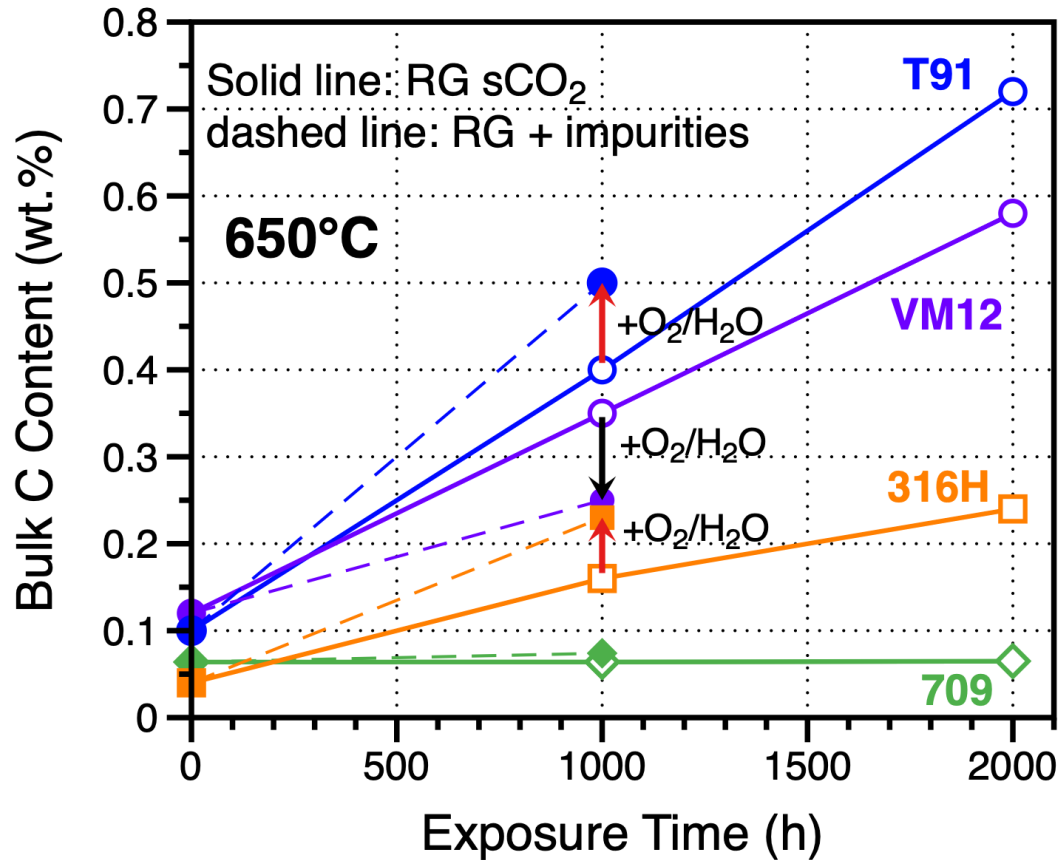
High ductility & low mass gain



500h & 1000 h measurements for each stainless steel

Low ductility & high mass gain

Bulk C measurements: Fe-rich oxides allow C ingress



RG sCO₂: No impurities
open symbols

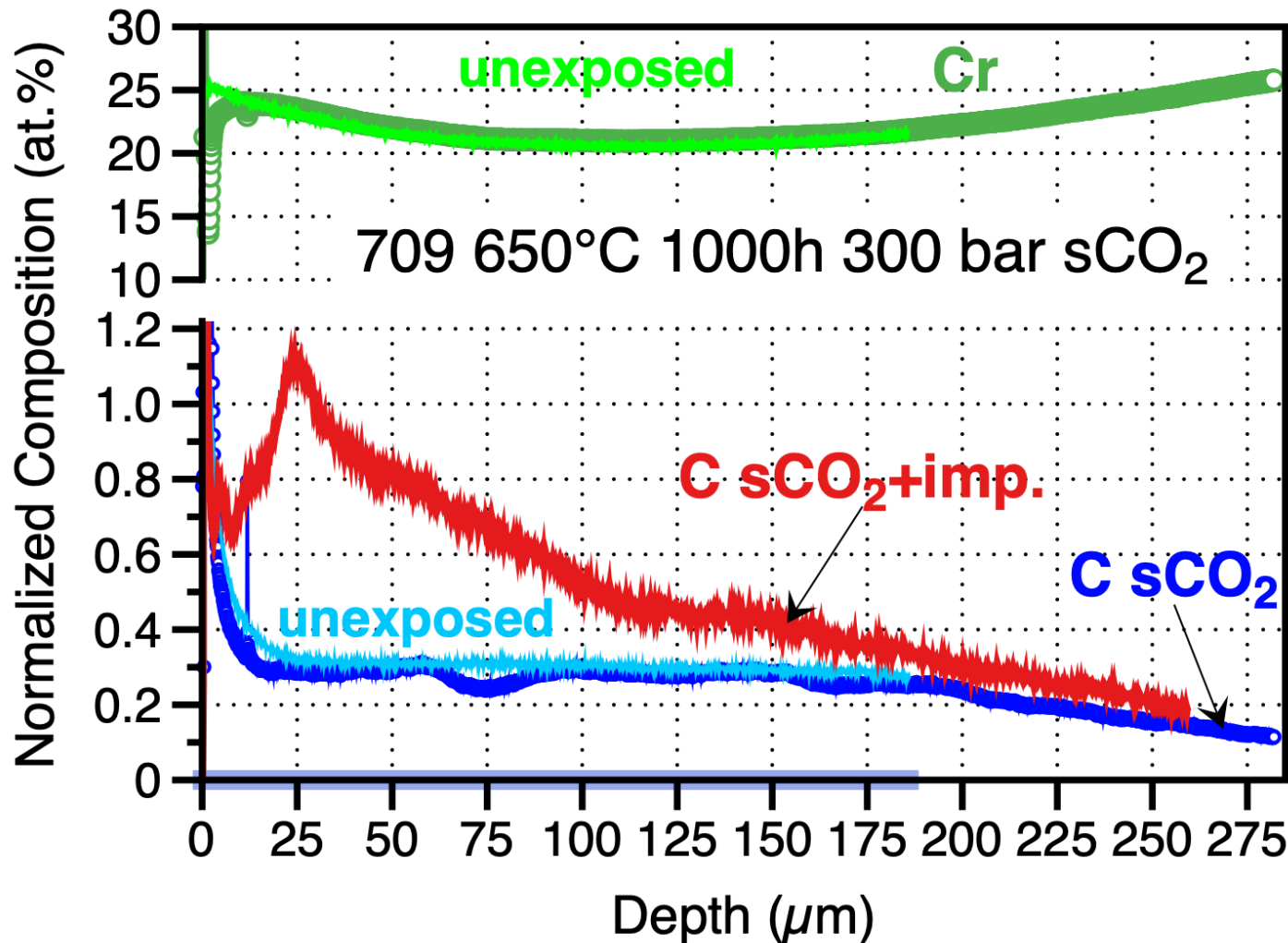
O₂/H₂O add: closed symbols

Relatively unaffected

- Measurements by combustion analysis, increasing with time
- Focus on 650°C results, less ingress at 600° and 550°C
- sCO₂ impurities tend to increase C ingress (but not all cases)

GDOES of 709 specimens after 1000 h

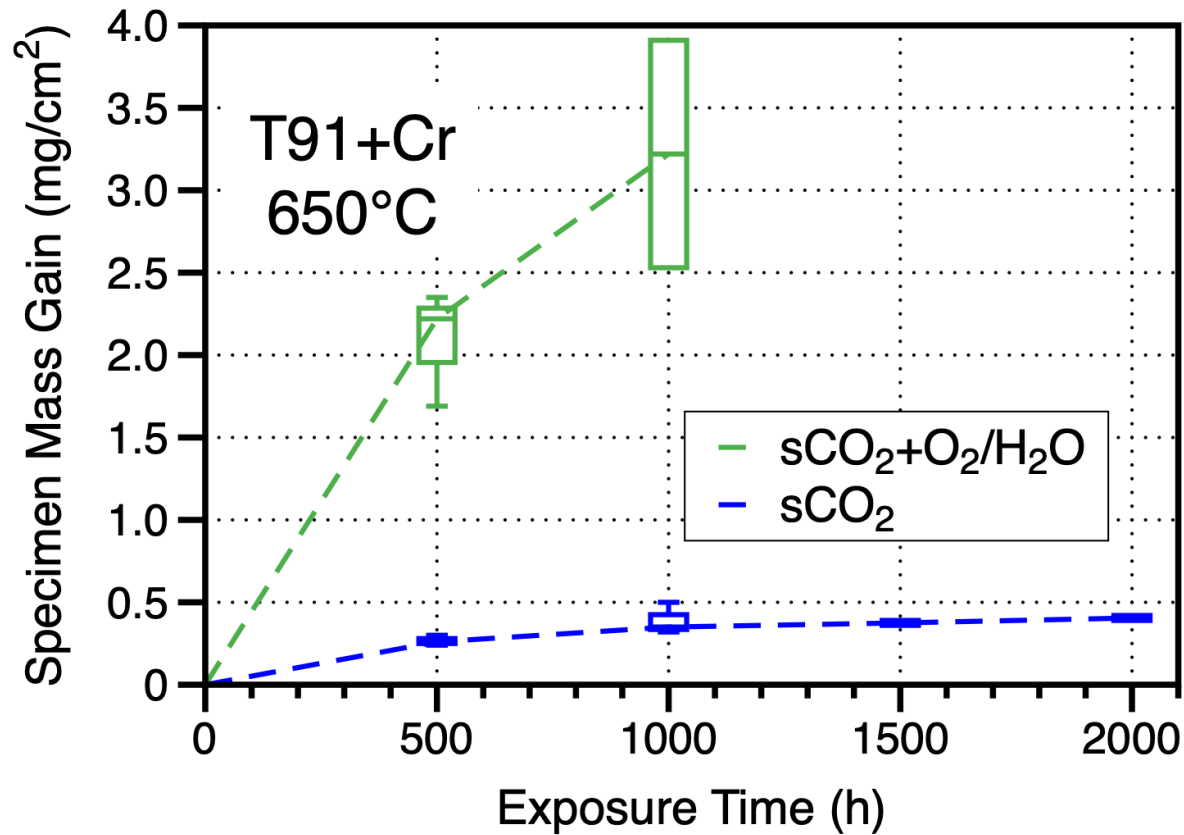
GDOES: Glow discharge optical emission spectroscopy



650°C: more C ingress with impurities

GDOES: signal normalized based on starting measured composition

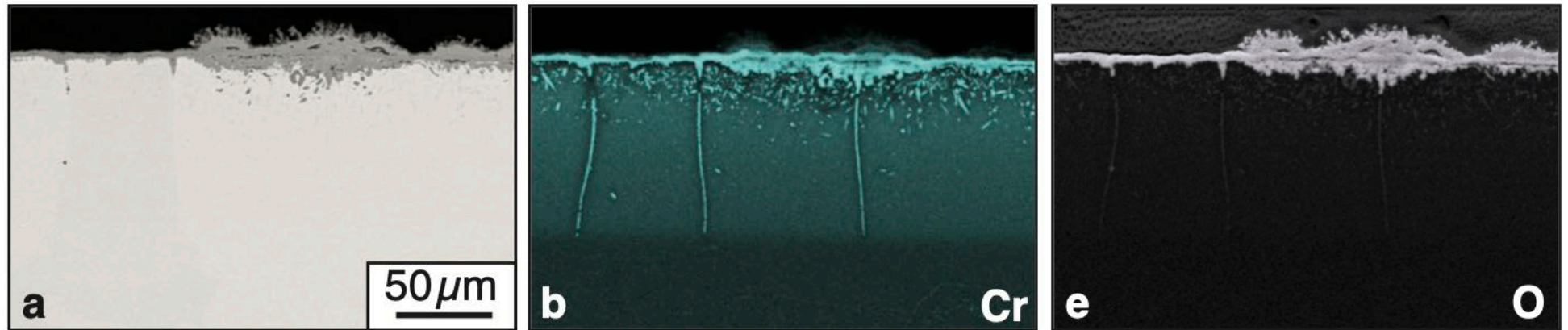
Proprietary Cr pack coatings on Gr.91 tested at 650°C



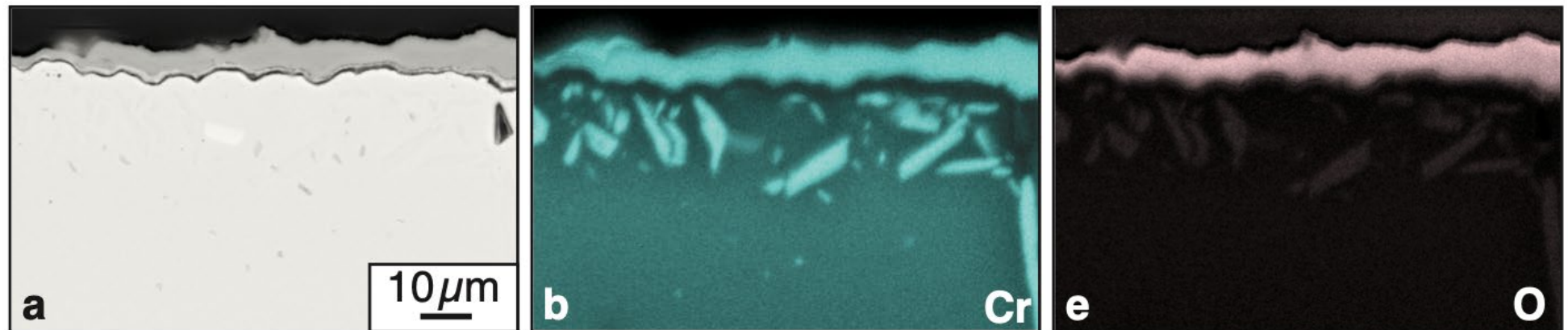
- Four specimens tested at each condition
- Low mass gain maintained in RG sCO₂ to 2000 h (2 specimens)
- Much higher mass gain in sCO₂ with 1%O₂-0.1%H₂O

SEM EDS: coating performance not promising

RG sCO₂
+1%O₂
+0.1%H₂O

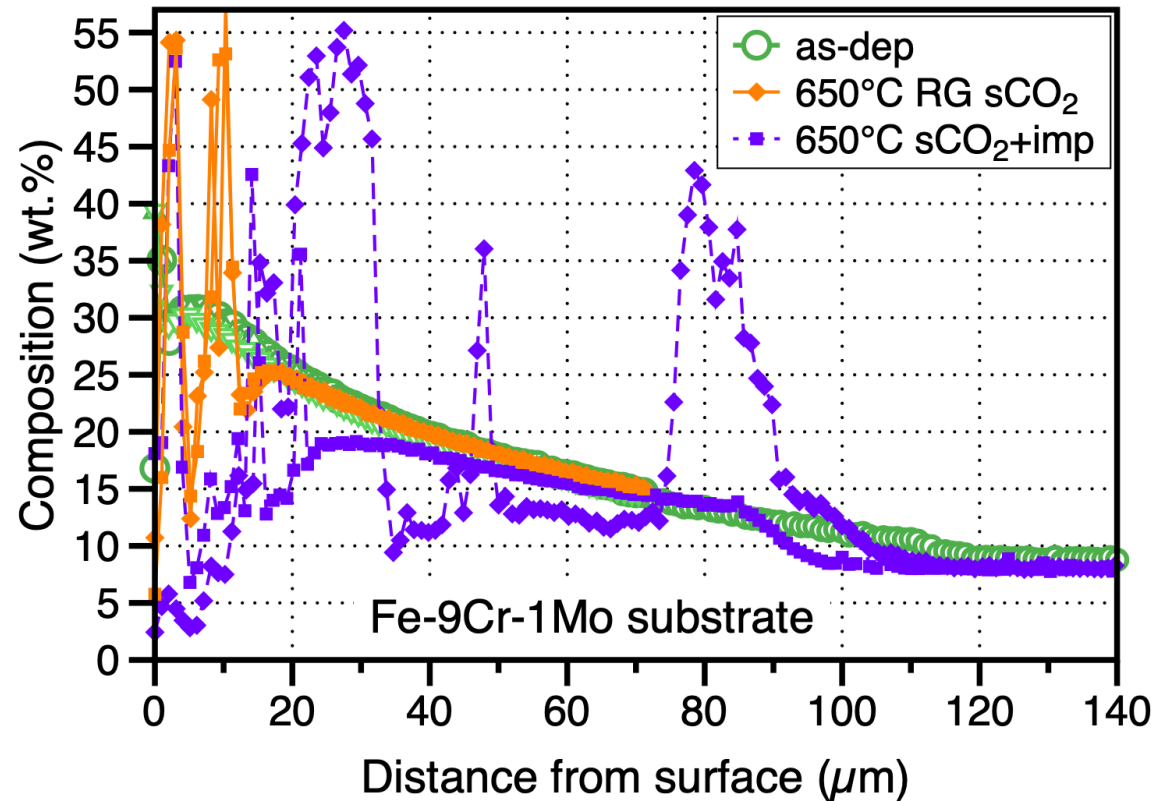


RG sCO₂
No impurities



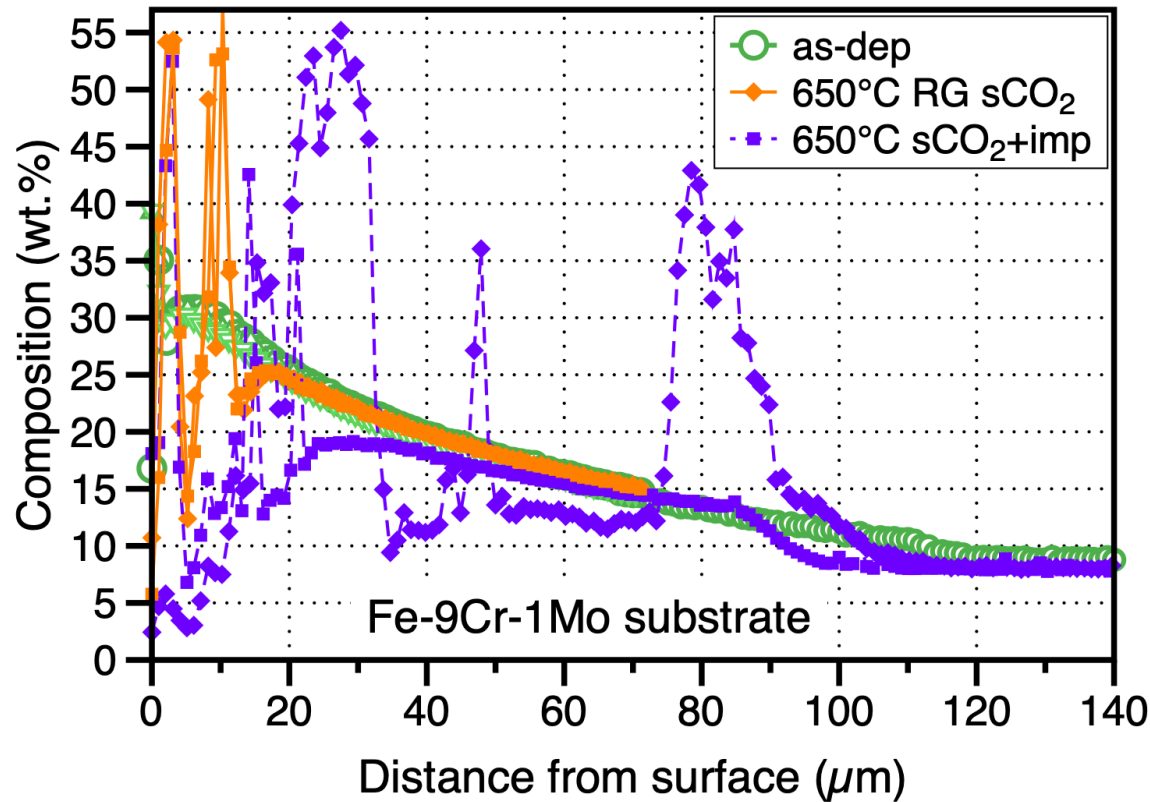
- Unable to form protective Cr-rich oxide with 1%O₂+0.1%H₂O
- Cr-rich precipitates in coating (carbides by EPMA)

sCO₂ ± impurities: significant Cr consumption by C or O

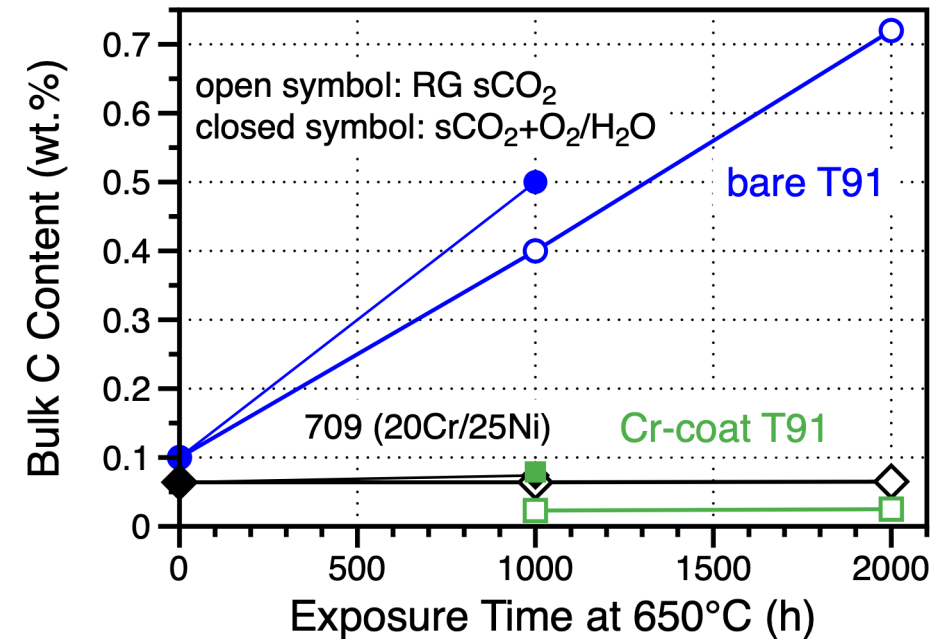


- 1000 h/650°C: higher Cr consumption with impurities
- 650°C: temperature too high for ~110 μm thick Cr pack coating

sCO₂ ± impurities: significant Cr consumption by C or O

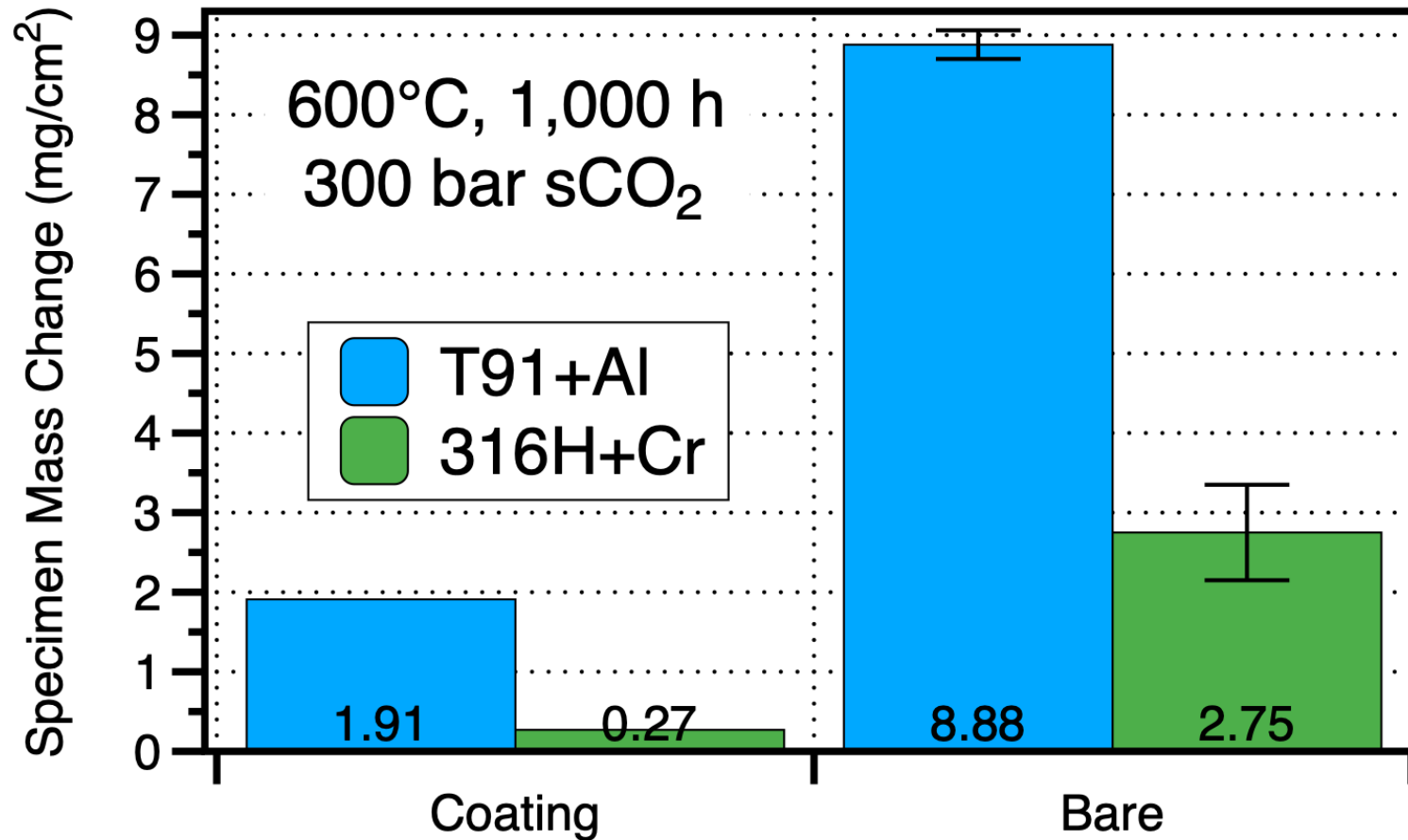


But coating blocked C!



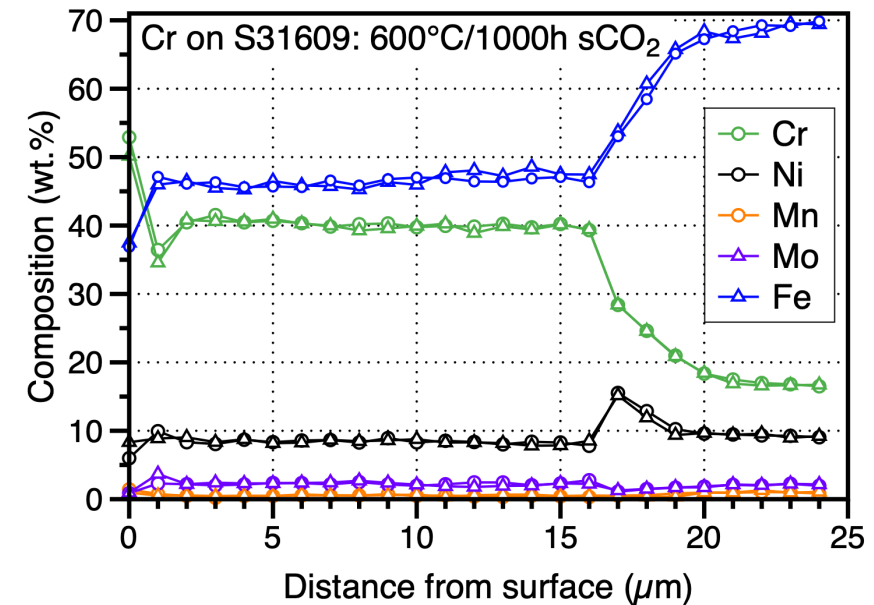
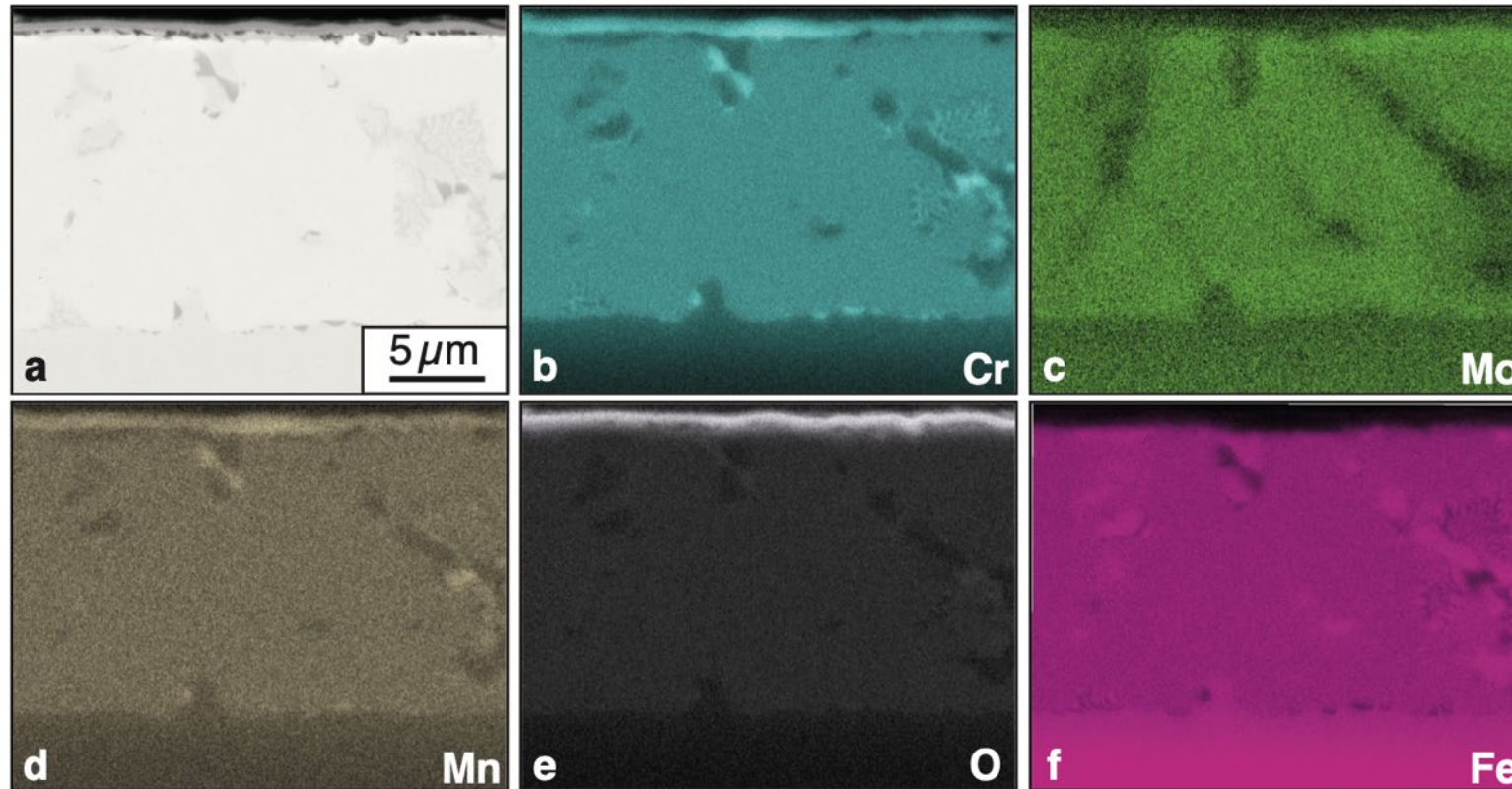
- 1000 h/650°C: higher Cr consumption with impurities
- 650°C: temperature too high for ~110 μm thick Cr pack coating

2nd batch of pack coatings also tested at 600°C in sCO₂



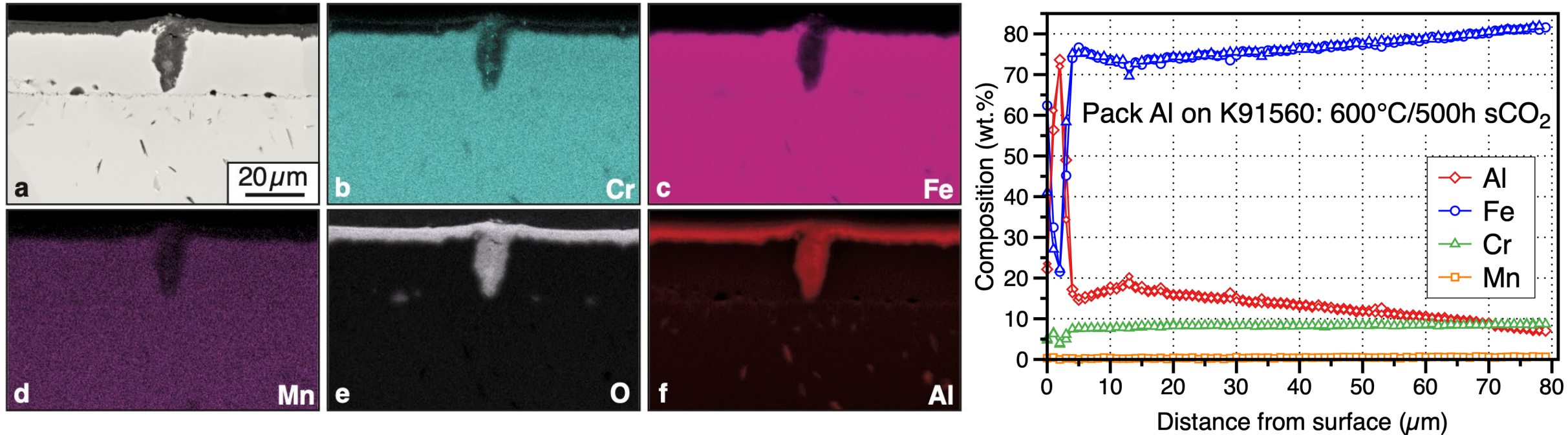
- Cr on 316H: similar benefit as on T91
- Aluminized T91: mass gain not consistent with Al₂O₃ formation
 - Even more difficult for selective oxidation at 600°C

Chromized 316H: thin Cr-rich scale formed + precipitates



- Thinner coating formed on FCC 316H substrate
 - Again, coating made by company, no details on process
- Beneficial effect in RG sCO₂ at 600°C after 1,000 h

Aluminized Gr.91: thick oxide after 500 h

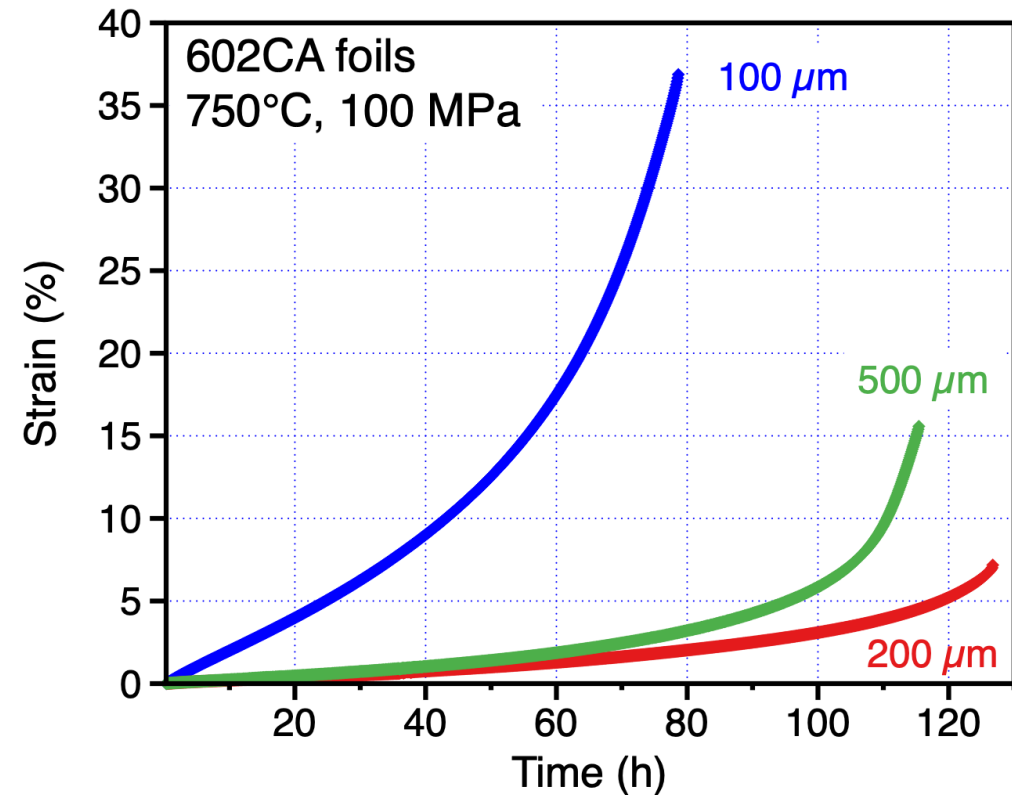
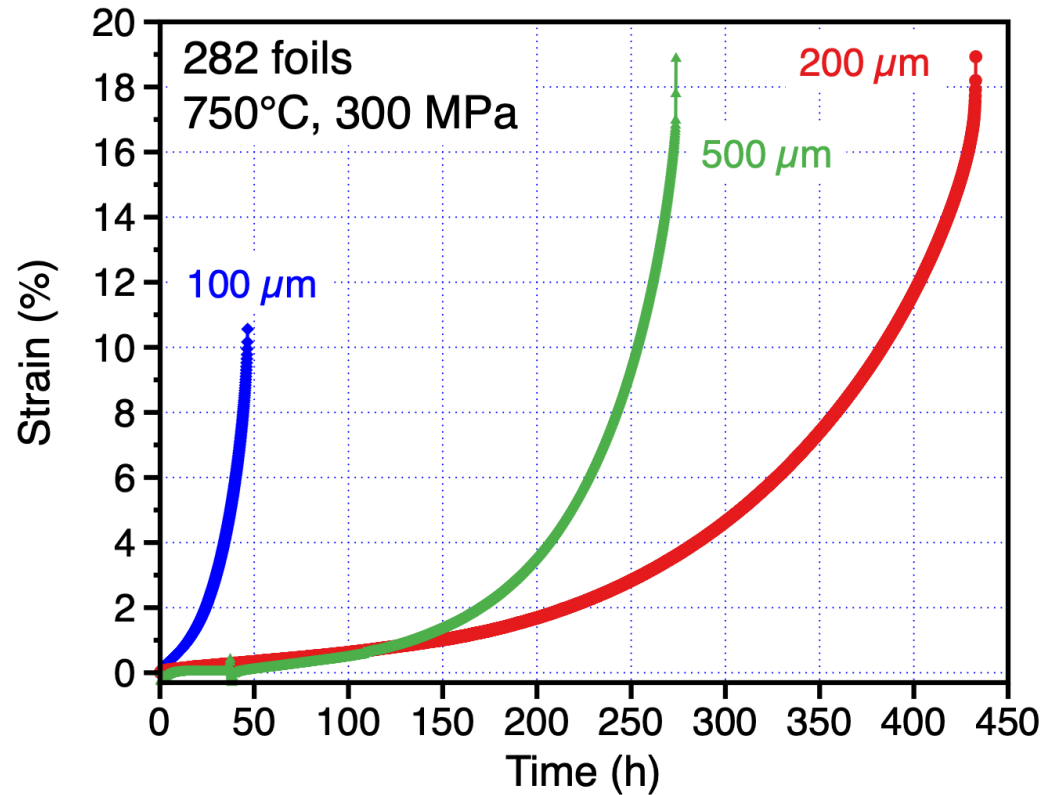


- ~100 μm thick Al diffusion profile into substrate
- Surface oxide Al-rich but thicker than expected after 500 h at 600°C in sCO₂

Thin walled heat exchanger: reduced 750°C creep lifetime for 282 and 602CA foils

282: Ni-19Cr-10Co-8Mo-1.5Al-2.2Ti-0.06C
602CA: Ni-25Cr-9Fe-2.4Al-0.2Ti-0.2C

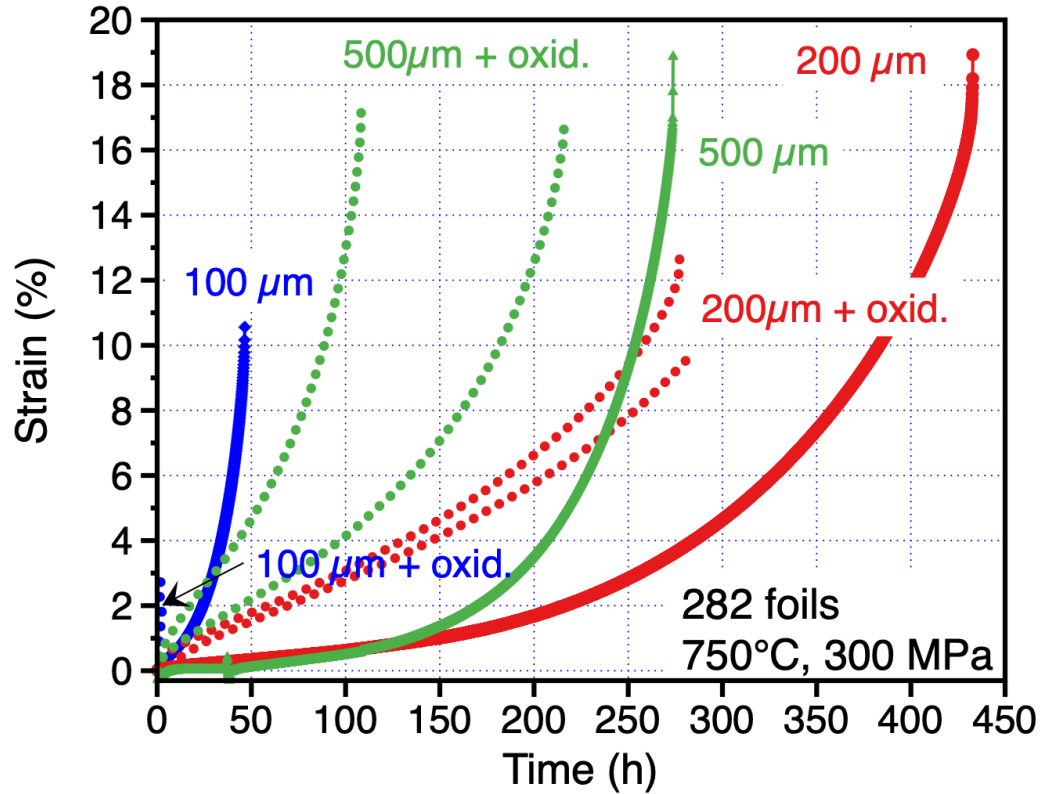
Expected lifetime for wrought = ~1500h



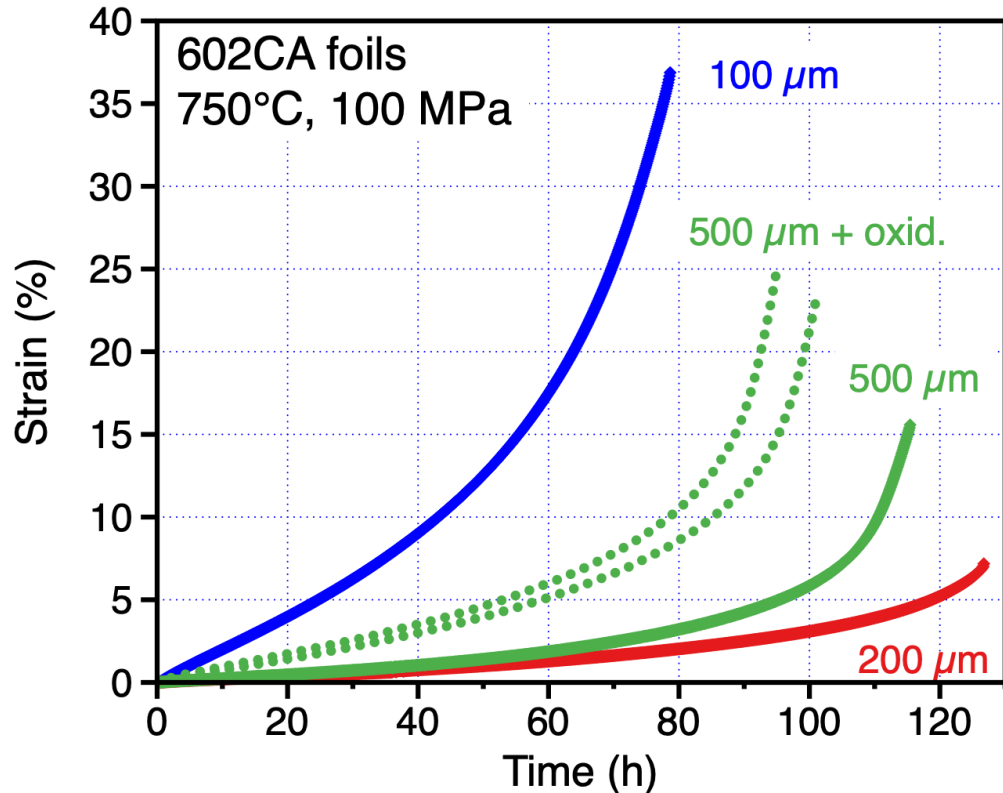
- Lower stress (100 MPa) for weaker 602CA material
- Lowest lifetime for 100μm foils as expected
- But 500μm foil exhibited shorter life than 200μm!

Does corrosion affect foil lifetime? Yes!

Foil creep specimens exposed 1,000 h at 800°C in wet air



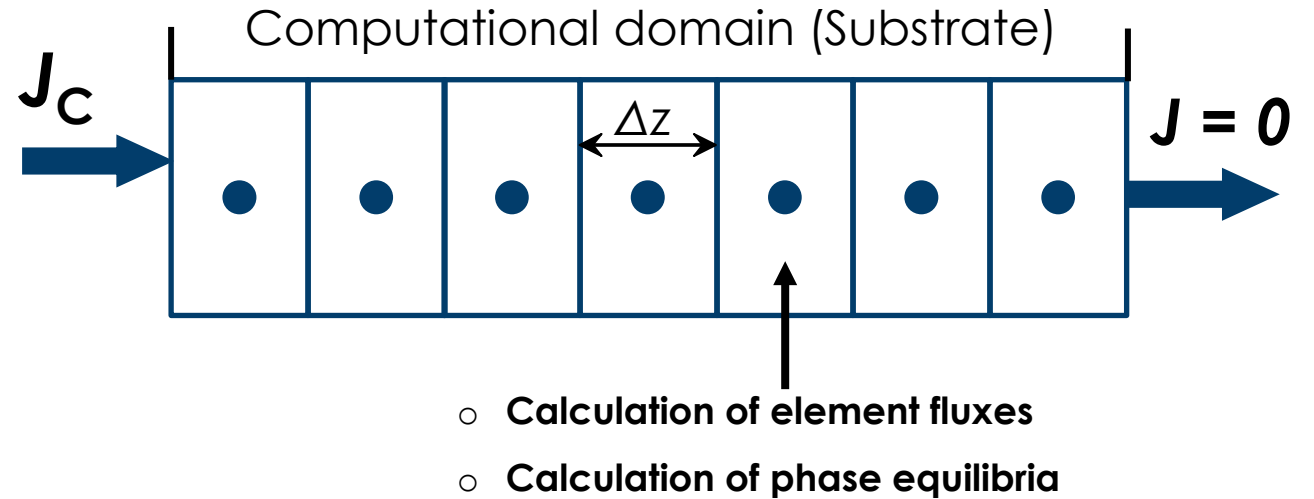
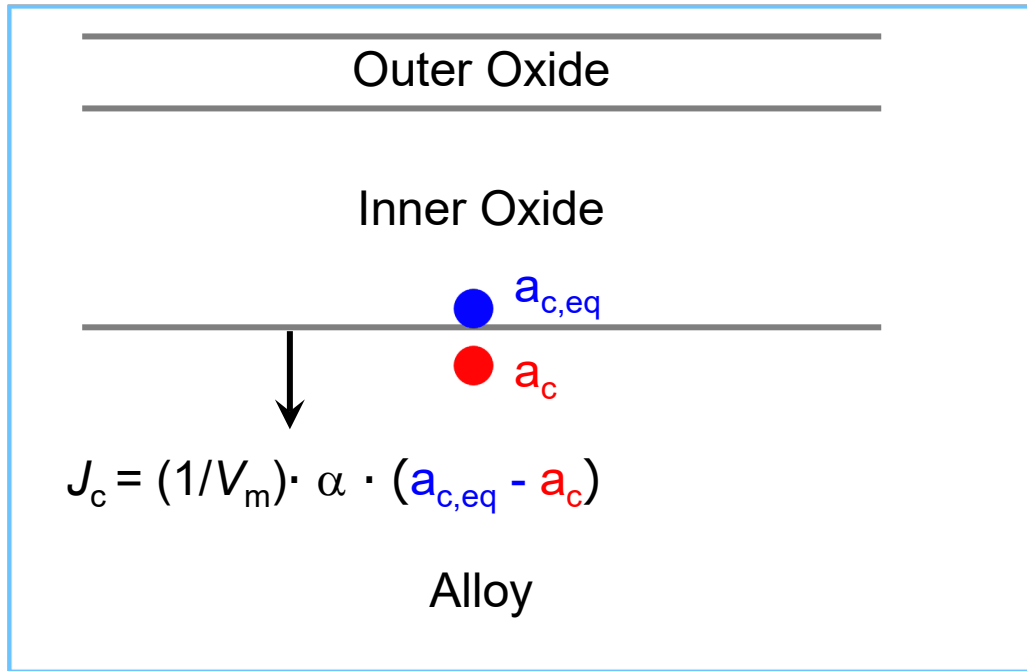
Expected lifetime for wrought = ~1500h



282: Ni-19Cr-10Co-8Mo-1.5Al-2.2Ti-0.06C
602CA: Ni-25Cr-9Fe-2.4Al-0.2Ti-0.2C

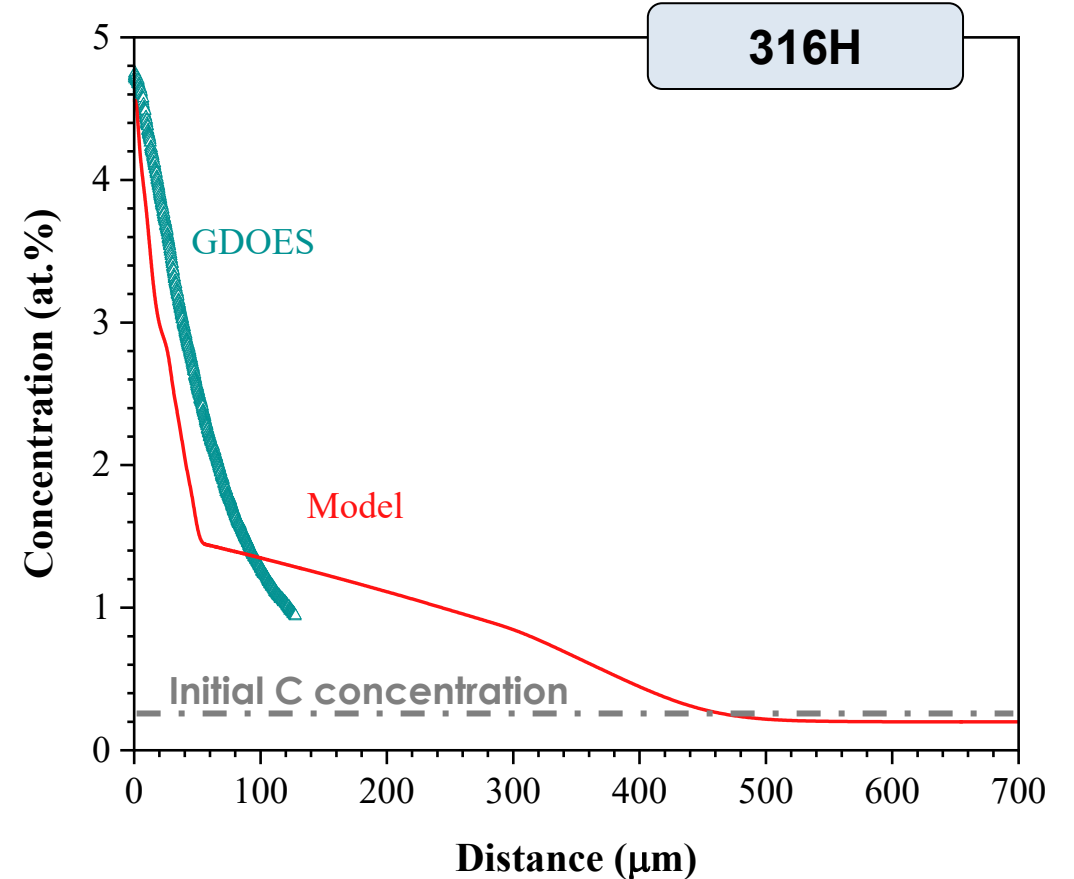
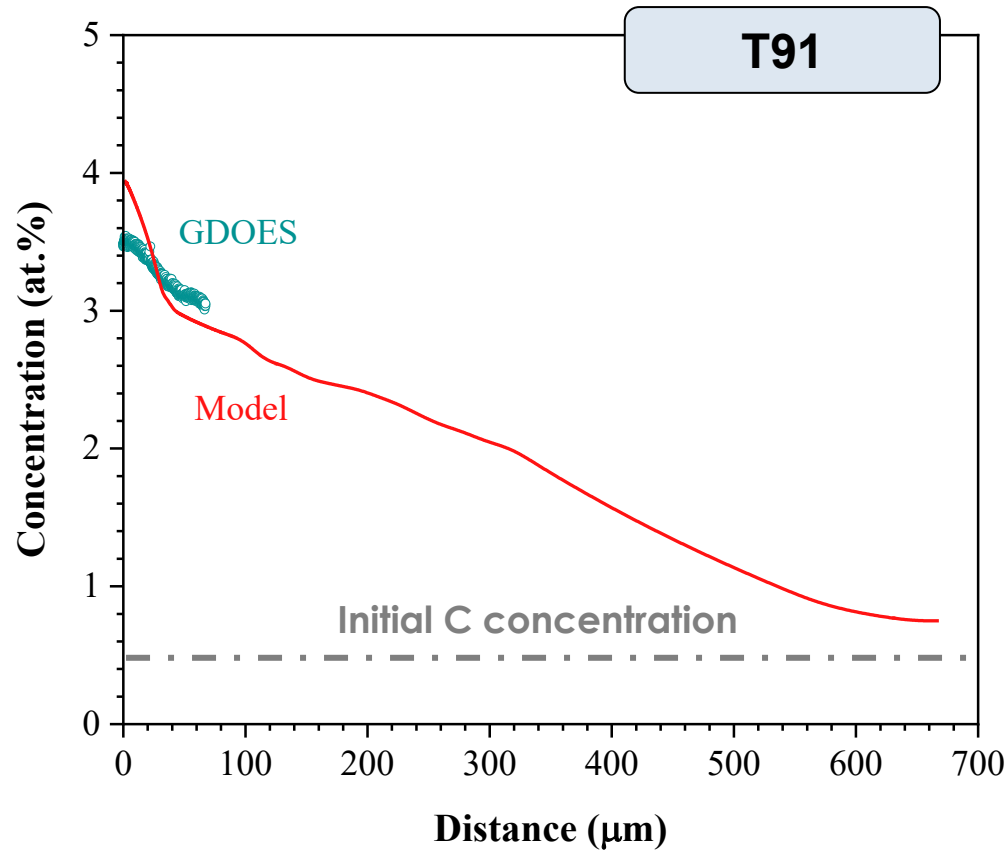
- Post-test characterization in progress
- 282: internal Al + Ti oxidation reduces γ' strengthening
- 602CA: Cr oxidation impacts carbide strengthening

One-dimensional Physics-based Modeling approach



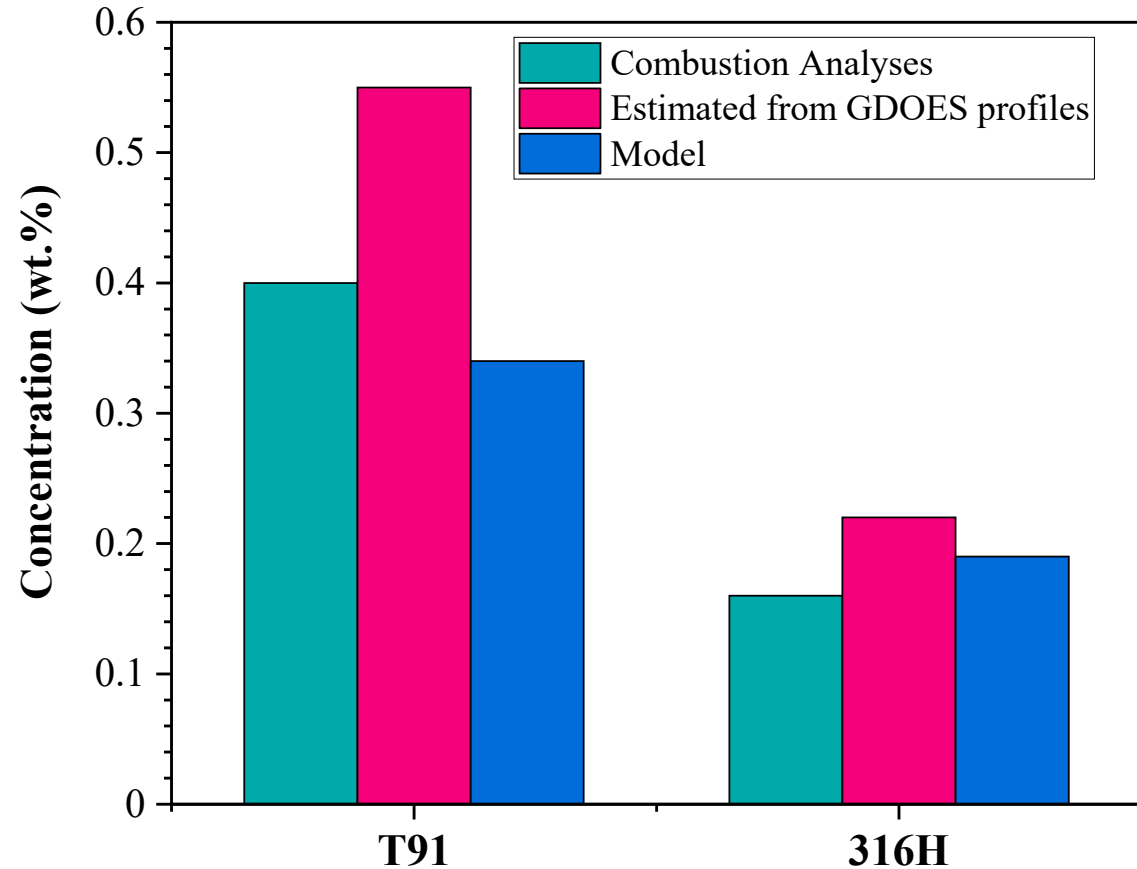
- α is mass transfer coefficient
- Assumed $a_{c,eq} = 1$ while a_c is carbon activity in the alloy at the oxide-alloy interface
- Use of independent thermodynamic-kinetic data (Thermo-Calc)
- Empirical description of grain boundaries (e.g., accelerated diffusion)
- Consideration of relevant elements & phases in commercial high temperature alloys and coating systems
- Thermodynamic calculations on multiple cores
 - 30,000h simulation of multicomponent-multiphase alloys in < 1 week

Model qualitatively predicted the observed differences in carburization behaviors between T91 and 316H after 1000 h at 650 °C in sCO₂



- **Almost complete carburization expected for a 1.5 mm T91 specimen**
- **About 0.4 mm carburization depth predicted for 316H**

Good agreement between the measured and predicted C-uptake for T91 and 316H after 1000h at 650 °C in sCO₂ (without impurities)



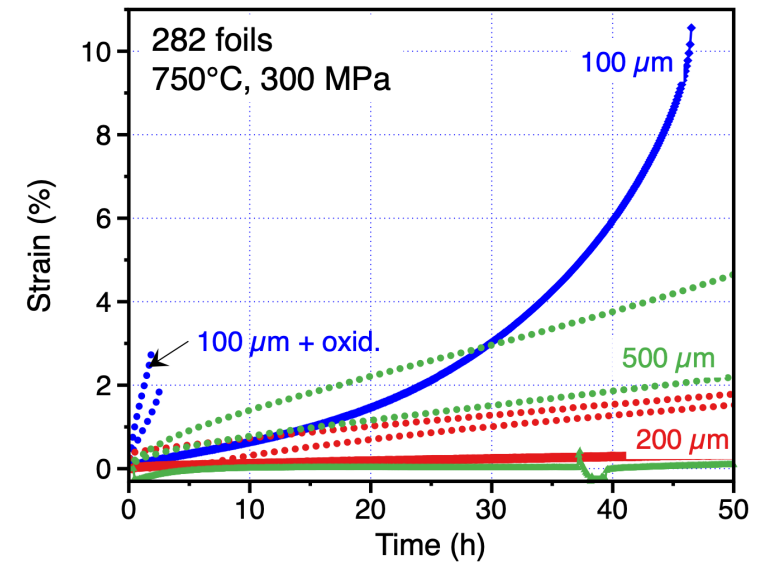
- Linear extrapolation of GDOES profile from measurement depth to specimen center is most likely resulting in overestimation of C uptake

✦ **How long can 316H provide carburization resistance at lower temperatures?**

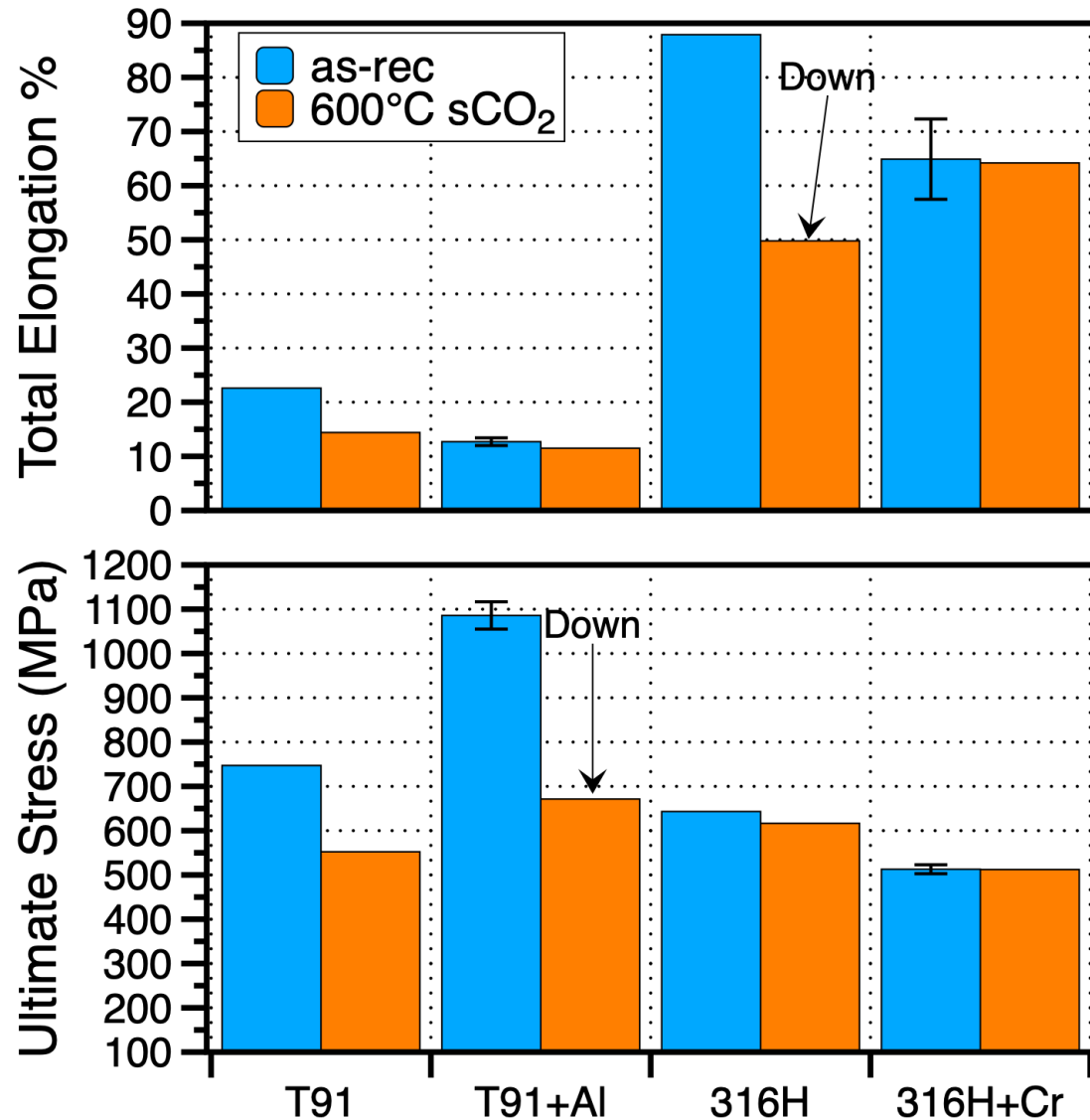
Summary: sCO₂ is a challenging environment

- Steels degrading at 550°-650°C in 300 bar sCO₂
 - Increased attack when O₂ and H₂O impurities in sCO₂
 - Even 709 (20Cr-25Ni) showed signs of embrittlement at 650°C with O₂+H₂O
- Opportunity for coatings?
 - Pack Cr and Al coatings evaluated on T91 and 316H at 650°C in sCO₂
 - Cr on T91: reduced mass gain but internal carbides + less protective with O₂+H₂O
 - Initial exposure at 600°C in sCO₂
 - Al on T91: ~20wt.% Al not sufficient to form protective Al-rich oxide
- At 650°-800°C, Ni-based alloys appear compatible
 - Thin wall HX: creep testing 100-500 μm foils of 282 and 602CA in air:
 - Reduced lifetime especially for 100 μm foils
 - Further degradation when oxidized (800°C in wet air to accelerate effect)

Questions?



Post-600°C sCO₂ exposure room temperature tensile testing



- 0.015/min strain rate
- **T91 HT: ~1050°C + ~700°C temper**
- Pack aluminized T91
 - Reduced ductility as coated
 - Drop in UTS vs. as-coated
 - 1050°C pack affected properties
 - Similar to normalization (temper in sCO₂)
- Cr-coated 316H
 - Lower properties after coating
 - No ductility loss after 600°C sCO₂

