# **Bench-scale Development of a Transformational Switchable-hydrophilicity Solvent**enabled Absorption Process for Energy-efficient CO<sub>2</sub> Capture and Fixation

Xiansen Li<sup>1</sup>, Hua Zhu<sup>1</sup>, David Atwood<sup>2</sup>, Nour Alwohoush<sup>3</sup>, Bikram Bhatia<sup>3</sup> <sup>1</sup> Thermisoln, LLC, Lexington, KY <sup>2</sup> Department of Chemistry, University of Kentucky, Lexington, KY <sup>3</sup> Department of Mechanical Engineering, University of Louisville, Louisville, KY

# MOTIVATION

A new switchable-hydrophilicity solvent (SHS) based post-combustion CO<sub>2</sub> capture technology enabling both the carbon capture and fixation can significantly improve the energy efficiency of CO<sub>2</sub> capture process



# **\_ENGES IN CCS WITH CONVENTIONAL AMINE ABSORBENTS**

- □ Absorption reaction/extraction coupling to break up chemical

deliver energy-efficient carbon

### CO<sub>2</sub> ABSORPTION

Solvent	Amine content (M)	Absorption T (°C)	Absorption time (min)	Aqu. vol. (%)	Aqu. CO <sub>2</sub> loading capacity (mol/kg)
SHS-1	3.5	40	10	60.40	1.039
			20	72.45	1.605
			30	73.58	2.040
			60	76.27	2.533
MEA	5.0	40	10	100	0.615
			20	100	1.226
			30	100	1.981
			60	100	2.211

✓ Aqueous-phase carbon loadings are enriched due to liquid-liquid phase separation (LLPS) ✓ Aqueous-phase volume increases over time  $\checkmark$  CO<sub>2</sub> absorption capacity shows advantage over 30 wt.% MEA solvents under the same operating conditions

After regeneration, aqueous phase accounts for 60.6 vol.%

Regeneration

10 min @ 65 °C

- Aqueous-phase carbon loading capacity takes up 97.7%
- Regenerated solvent behaves like a fresh chemical

**RICH SOLVENT REGENERATION** 

Spontaneous LLPS promotes a deep regenerability of rich solvents under mild operating conditions

## **GYPSUM UPCYCLING INTO LIMESTONE**



- ✓ FGD gypsum was fully transformed into well-faceted calcite within 10 min at 40 °C
- 1 kg of FGD gypsum was completely converted into calcite within 10 min at 60 °C

### **KEY COMPONENTS OF CO<sub>2</sub> CAPTURE PROCESS**







### **CARBON CAPTURE PERFORMANCE METRICS**

Metrics	Values
CO <sub>2</sub> loading capacity at 40 °C	≥ 2 mol/kg
Overall cyclic CO <sub>2</sub> -equivalent loading capacity	≥ 1.8 mol/kg
CO <sub>2</sub> absorption time	≤ 10 min
ER of rich solvents at $T \le 65 ^{\circ}\text{C}$	≥ 95%
Regeneration reaction time	≤ 10 min
Gypsum waste purity	≥ 95%
Gypsum conversion efficiency	≥ 95%
Gypsum carbonation time at $T \le 65 ^{\circ}\text{C}$	≤ 10 min
Limestone purity	≥ 95%
CO <sub>2</sub> removal efficiency	≥ 90%
Final CO <sub>2</sub> purity in vapor phase	≥ 95%
Increase in cost of electricity generation	≤ 35%
Overall CO <sub>2</sub> capture cost	~\$30/tonne

### **FUTURE WORK**

- Upgrade the whole carbon capture system to realize its fullscale operations at a bench level
- Design and assemble the atomizers to enable an overall solvent flowrate up to 2 GPM
- A 2<sup>nd</sup>-generation scrubber was built and commissioned inhouse with novel packing materials in place
- CO<sub>2</sub> absorption kinetics and overall CO<sub>2</sub> cyclic loading capacity will be further enhanced
- Seamlessly integrate the key three-unit operations to enable continuous operational processes
- More exhaustive TEA of this carbon capture technology will be conducted to gauge its viability for future scale-up

### ACKNOWLEDGEMENTS

This project is financially supported by the DOE SBIR/STTR Programs (DE-SC0022734) with Mariah I. Young as the PM, and by the KY Cabinet for Economic Development (2022-002-006). The authors would also like to thank LG&E and KU Energy, LLC for its generous FGD gypsum offerings.