

# Direct Air Capture Work at NIST: Reference Materials to Advance DAC Technologies

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## NIST Climate Program

NIST addresses climate change in impactful ways, from measurements and modeling of greenhouse gas (GHG) emissions to research and tools to build more resilient communities and alternative energy infrastructure. NIST programs advance research and measurements for energy efficiency and sustainability, as well as standards, frameworks and other resources for enabling metrology.

### Climate Measurements and Monitoring

- Traceability of GHG measurements
- GHG measurement technology
- Ensuring climate data quality and standardization

### Decarbonization of the Economy

- Built environment
- Energy infrastructure
- Carbon Dioxide Removal (CDR), Carbon Capture Utilization and Sequestration (CCUS)
- Manufacturing

### Adaptation and Resilience

- Disaster and failure studies
- Wildland-Urban Interface fires
- Community resilience
- Connected systems resilience

### Life Cycle Analysis, Carbon Accounting

NIST provides key tools to mitigate and adapt to climate change in the form of:

- Measurement methods, calibrations and performance metrics,
- Reference materials and data,
- Unique facilities and testbeds, and
- Technical guidance and resources for communities and researchers.

## Benchmark Materials

Develop critical measurement science to accelerate innovation and enable timely, effective, and scalable direct air capture

NIST is developing a series of DAC benchmark materials as Research Grade Test Materials (RGTM's)

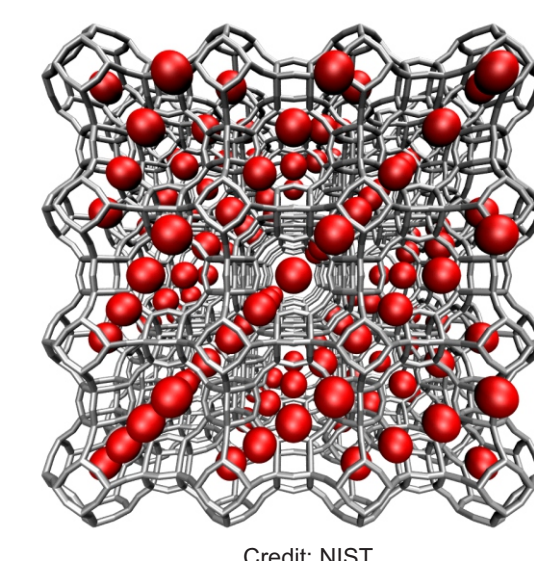
✓ Seeking industry feedback on measurements and prioritizing materials of interest through workshop

✓ Coordinating with DOE's DAC facilities to support measurement needs

✓ Collaborating with large-scale manufacturers in RGTM process

### Key questions

- How much CO<sub>2</sub> is adsorbed and desorbed?
- How fast does it happen?
- How much energy does it take?
- How does it perform in real-world conditions?
- How long will it last?



## Benchmark Measurements

Anticipated Range of Performance Conditions

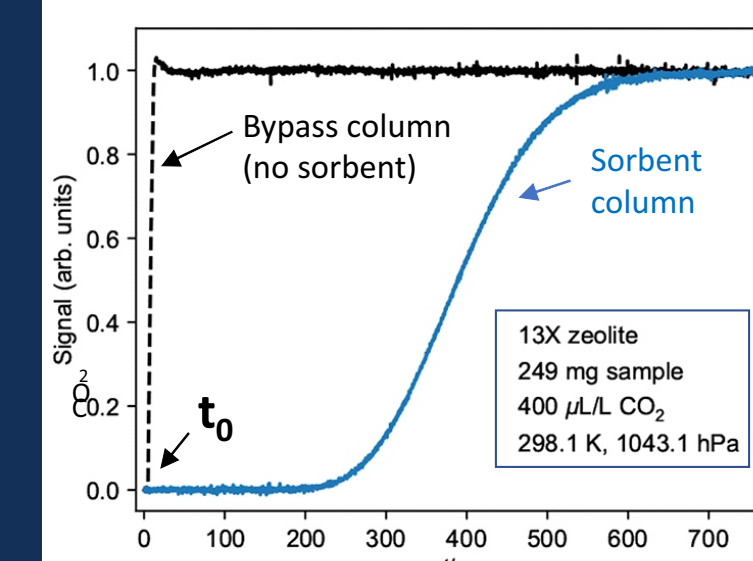
Figure of Merit	Range	Units
Temperature	-233.1 to 323.1	K
Pressure	300 -1050	hPa
Relative Humidity	0 - 80	%
CO <sub>2</sub> Concentration	400	mL/L

The following methods were identified as top priority for the NIST DAC Benchmark Materials:

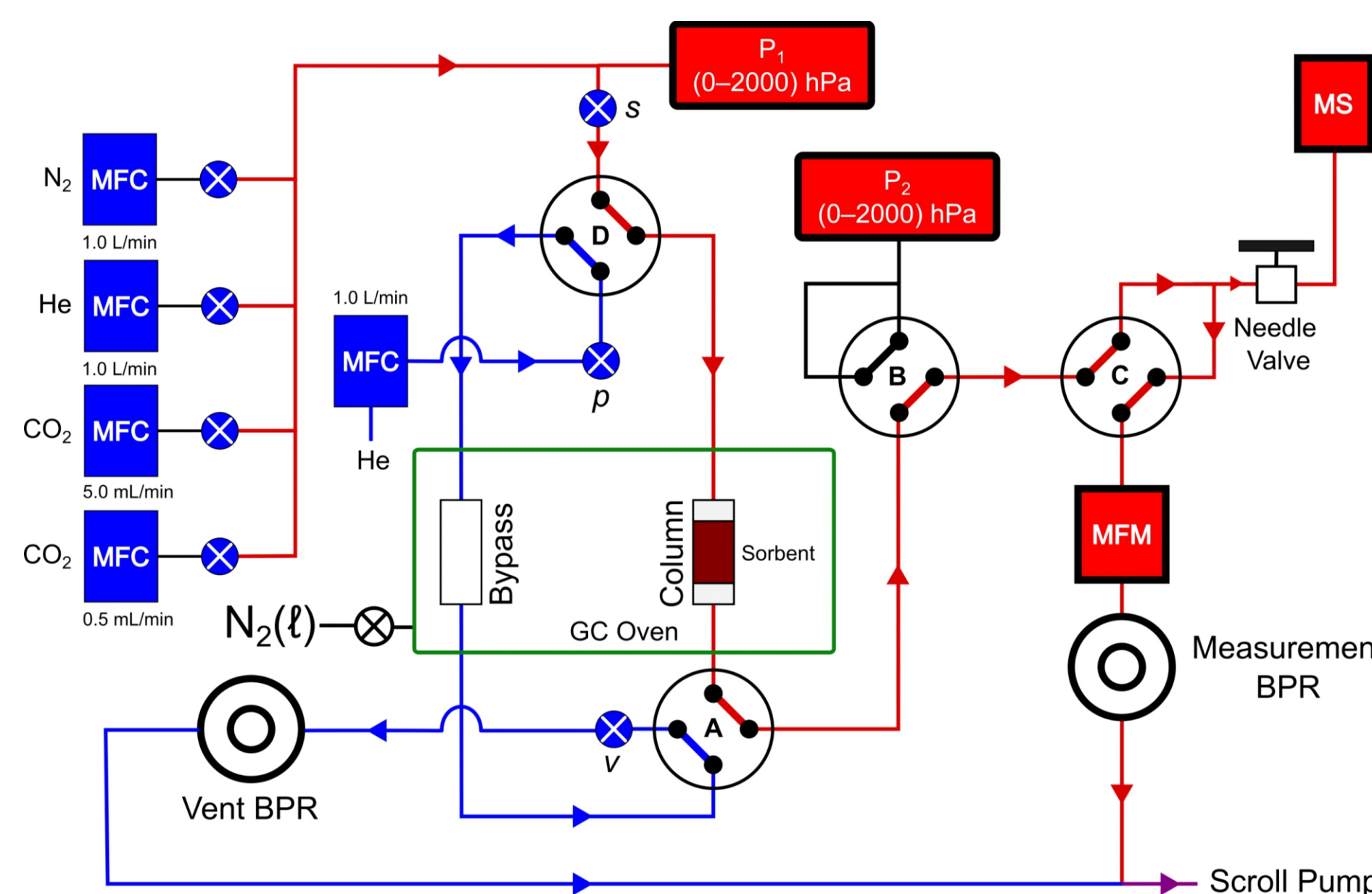
- Breakthrough
- Adsorption Isotherms
- Density/Porosity
- Microstructure Characterization (SEM)
- Heat Capacity

## Dynamic Column Breakthrough

NIST has developed instrumentation and measurement methods to characterize the adsorption of gaseous species



**Example Data.** DCB curves for CO<sub>2</sub> on 13X zeolite exposed to 400 μL/L CO<sub>2</sub> in He at 500 cm<sup>3</sup>/min. Shown are the respective mass spectral *m/z* 44 signals (CO<sub>2</sub> parent peak) for the empty bypass column (black dashed curve) and the sorbent column (blue solid curve). *t*<sub>0</sub> indicates the initiation time of each experiment. The measured capacity under these conditions is 0.23 mol/kg; static adsorption isotherm measurements yield 0.22 mol/kg.



Schematic of the NIST DCB apparatus. Valve orientations and gas flow paths (red for the sample and blue for the purge) correspond to those at the initiation of a sorbent column adsorption experiment (i.e., *t*<sub>0</sub>). Bold letters identify two-way, four-port rotary valves, and crossed circles are on-off valves, with some labeled for reference in the text. MFC indicates a mass flow controller, and MFM is a mass flowmeter. MS is the mass spectrometer, P1 and P2 indicate manometers of the total system pressure, and BPR refers to a back pressure regulator.

## References

McGivern, W., Nguyen, H. and Manion, J. (2023), Improved Apparatus for Dynamic Column Breakthrough Measurements Relevant to Direct Air Capture of CO<sub>2</sub>, *Industrial and Engineering Chemistry Research*, <https://doi.org/10.1021/acs.iecr.2c04050>

Baumann, A., Yamada, T., Ito, K., Snyder, C., Hoffman, J., Brown, C., Stafford, C. and Soles, C. (2024), Measuring the Influence of CO<sub>2</sub> and Water Vapor on the Dynamics in Polyethyleneimine To Understand the Direct Air Capture of CO<sub>2</sub> from the Environment, *Chemistry of Materials*, 2024.

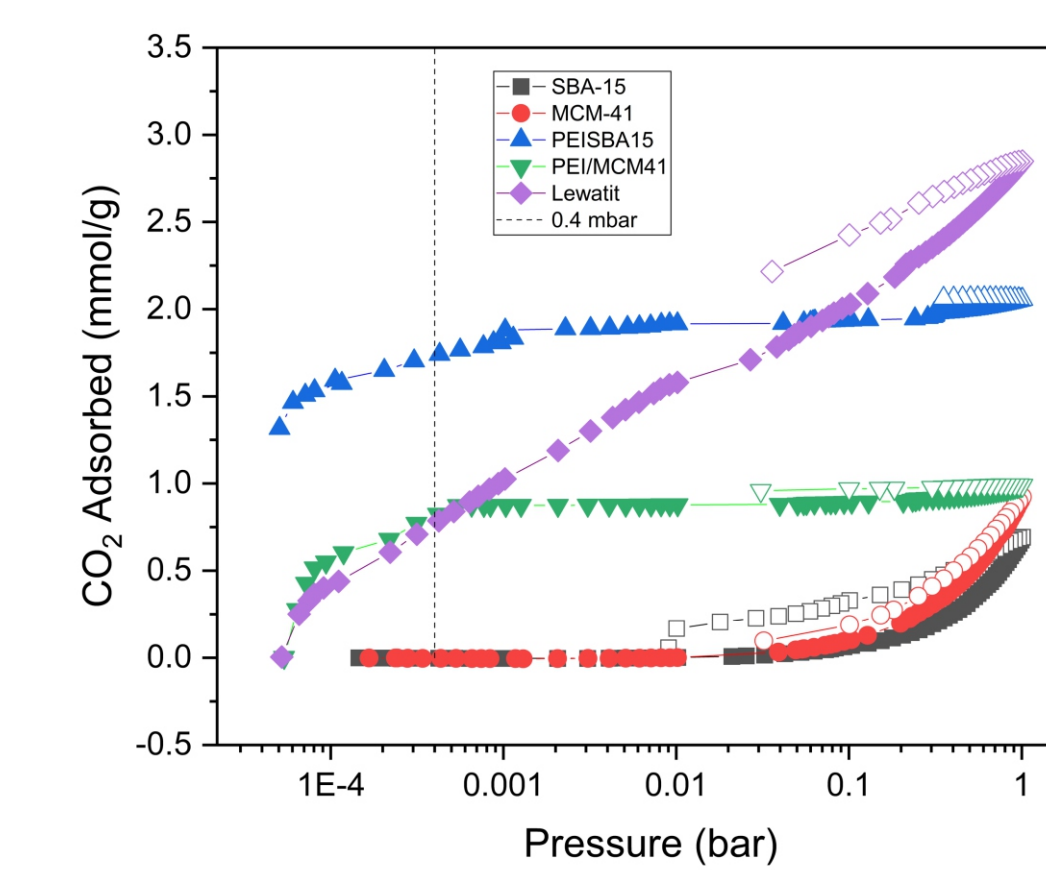
Hoffman, J., Baumann, A. and Stafford, C. (2024), Thickness Dependent CO<sub>2</sub> Adsorption of Poly(ethyleneimine) Thin Films for Direct Air Capture, *Chemical Engineering Journal*, <https://doi.org/10.1016/j.cej.2023.148381>, 2024.

Wentz, C., Tsinas, Z. and Forster, A. (2023), A Synthetic Methodology for Preparing Impregnated and Grafted Amine-Based Silica-Composites for Carbon Capture, *Journal of Visualized Experiments*.

Wentz, C. *et al.* Long-Term Stability of Amine-Grafted Silica for Direct Air Capture via Accelerated Aging. Submitted 2024.

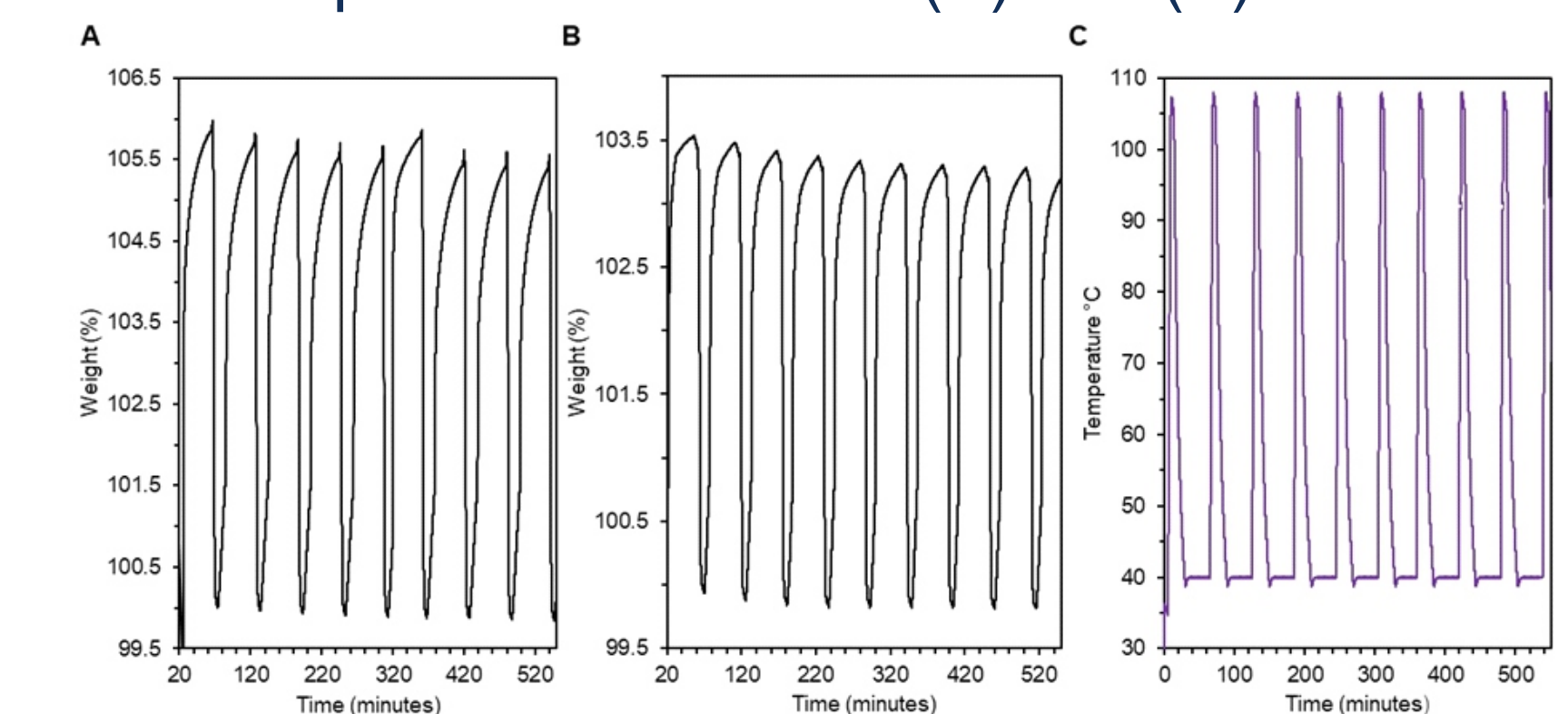
Carter, M. *et al.* Progress in Development of Characterization Capabilities to Evaluate Candidate Materials for Direct Air Capture Applications. Submitted 2024.

## Adsorption and Temperature Effects



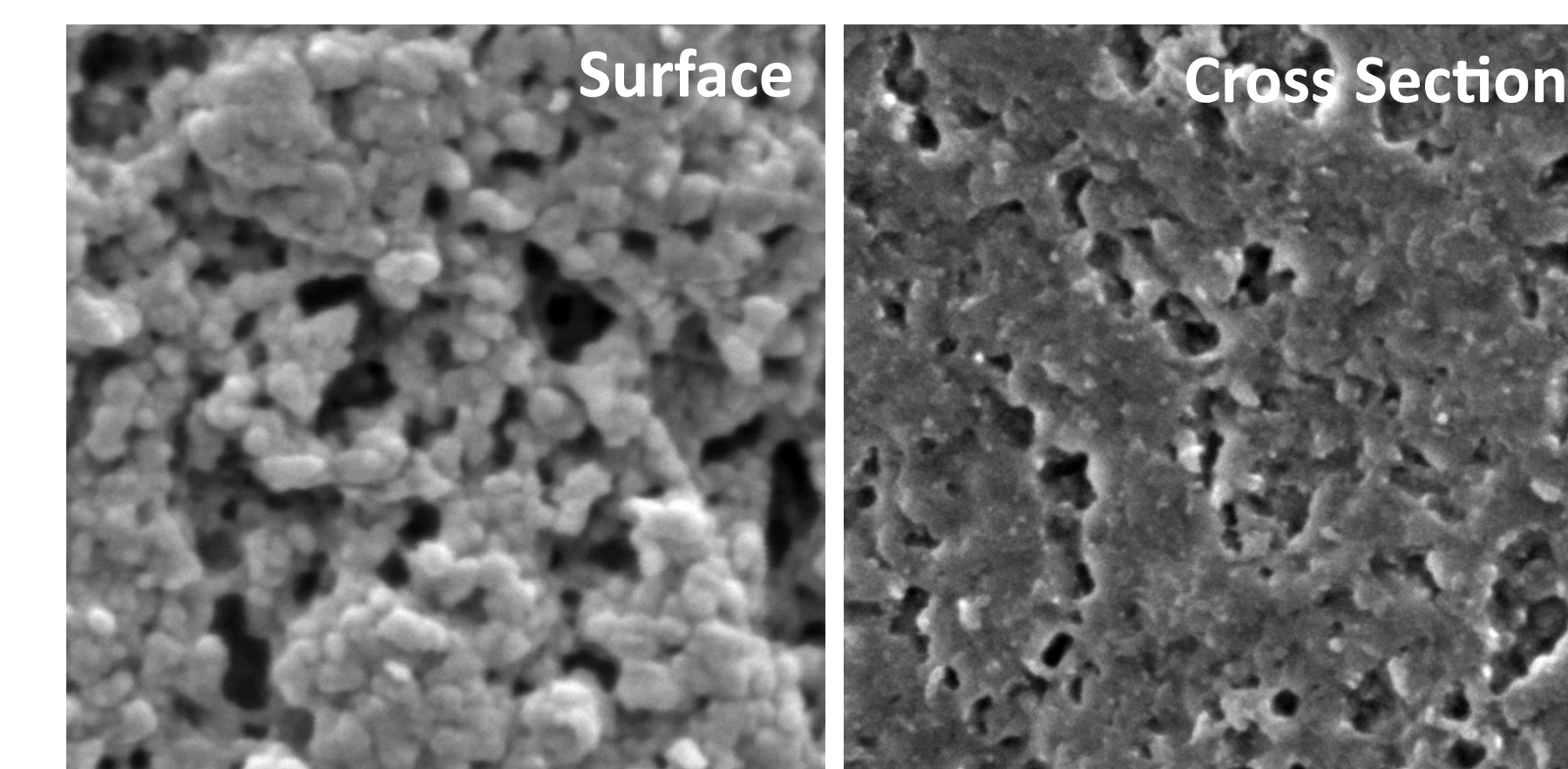
Representative excess pure CO<sub>2</sub> sorption isotherms at 25 °C (298 K) for various DAC materials. The dotted line marks CO<sub>2</sub> uptake at 0.4 mbar (equivalent partial pressure to 400 ppm). Filled symbols are for adsorption; open symbols are for desorption. Uptake is given as mmol of CO<sub>2</sub> per g of dry sample.

Cyclic TGA adsorption and desorption study of (A) PEI impregnated MCM-41, (B) DAS grafted MCM-41, and (C) temperature profile of the procedural runs for (A) and (B).



Cross instrument temperature cycling via TGA and DCB have been used to probe stability and viability of DAC materials.

## Microstructural and Heat Capacity



Preliminary data and analysis revealed similar porosity between the surface and internal structure of the candidate RGTM.

Through image analysis, the % porous area can be calculated for the material.

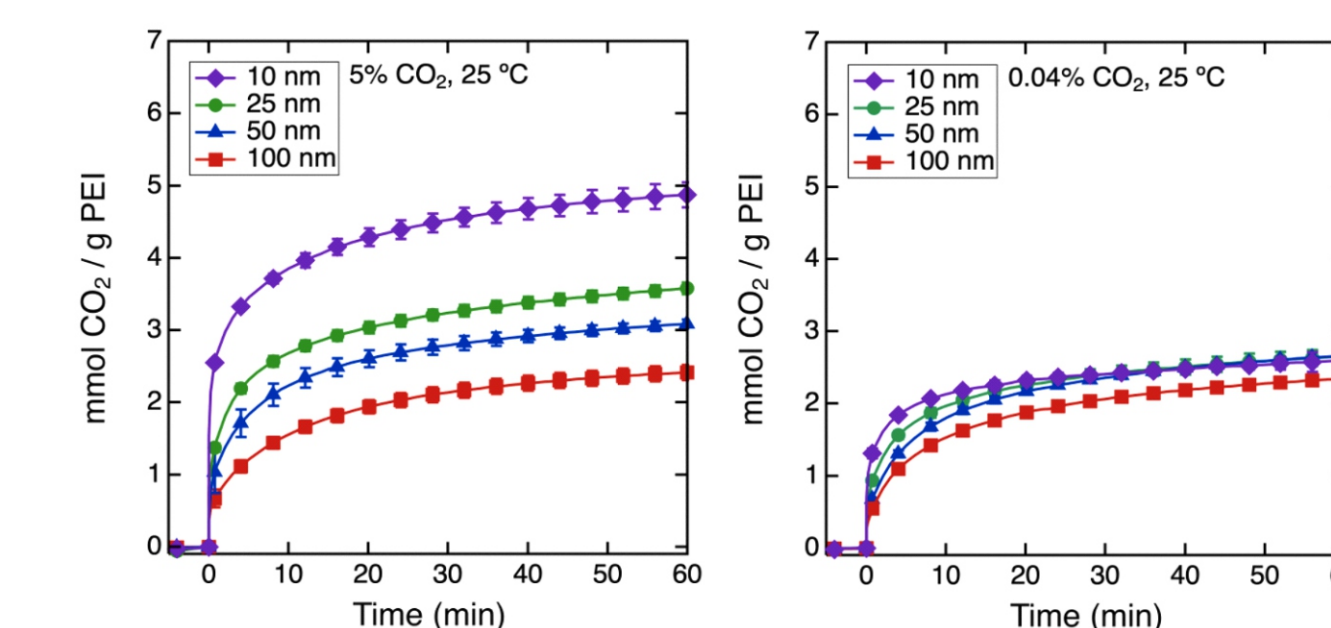
Sample	% Porous Area
Surface	5.682
Cross Section	4.636

The heat capacity of the RGTM candidate was determined using 2 methods: (a) the MDSC mode (reversing heat capacity), and (b) the Direct/Standard DSC mode (total heat capacity).

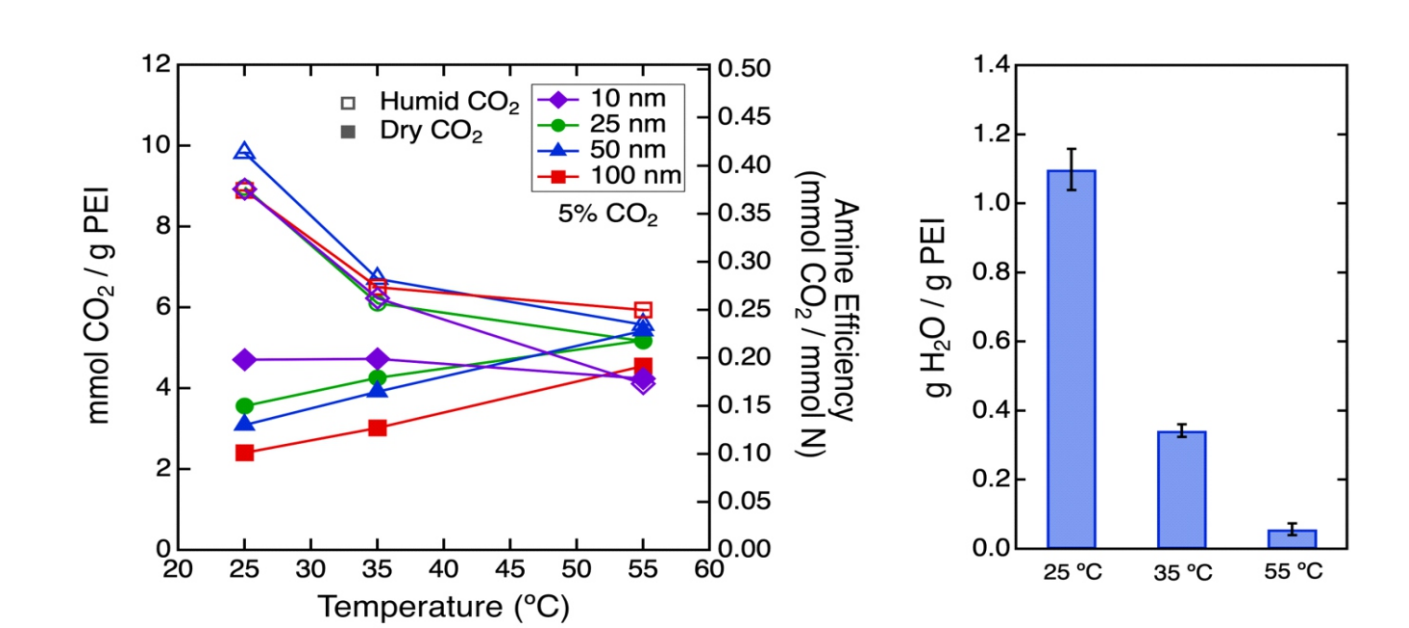
Temperature [°C]	MDSC - C <sub>p</sub> [J/(g·°C)]	Standard DSC - C <sub>p</sub> [J/(g·°C)]
10	1.292	1.349
25	1.395	1.453
50	1.542	1.732
100	1.702	3.189
180	1.957	2.046

## CO<sub>2</sub> Sorption in Thin Films

Dry and humid CO<sub>2</sub> sorption in PEI thin films species in polyethyleneimine films



- Thinner films exhibit greater CO<sub>2</sub> capture capacity per g of PEI, suggesting more amine sites are accessible due to lower diffusional resistance.
- This effect is less pronounced, but still true, when capturing 400 ppm CO<sub>2</sub>.



- Humidity enhances the CO<sub>2</sub> capacity and amine efficiency of the PEI, due to increased mobility and facilitated carbamate formation.
- Enhancement due to humidity decreases with temperature due to decrease in overall water sorption.