Regional Resource Assessment for CO₂ Storage in New Mexico and Surrounding Areas: Identification, Characterization, and Evaluation of In-Situ Mineralization Site/Complex

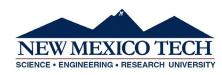
Carbon Conversion FOA 2614: AOI 4 DE-FE0032257 Project Performance Dates: 09/04/2023 – 09/03/2025

Sai Wang, PhD New Mexico Tech - PRRC

U.S. Department of Energy National Energy Technology Laboratory 2024 FECM / NETL Carbon Management Research Project Review Meeting August 8, 2024



Project Participants





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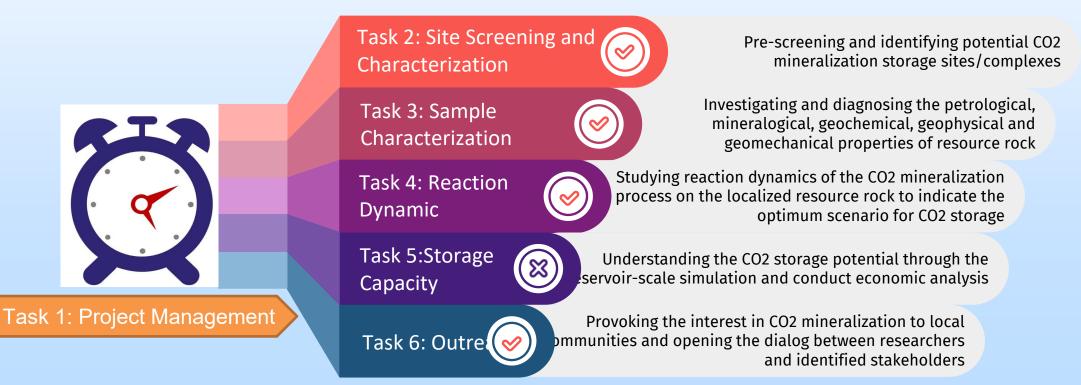




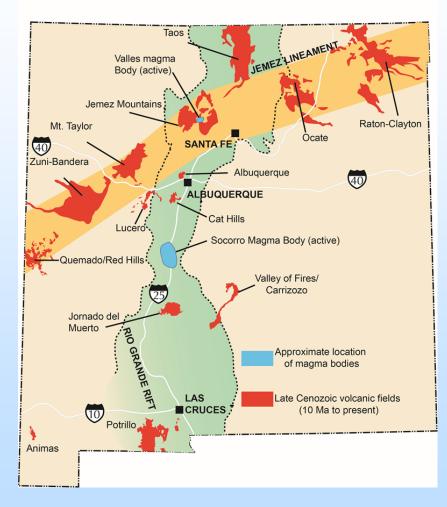
Project Objective and Goals

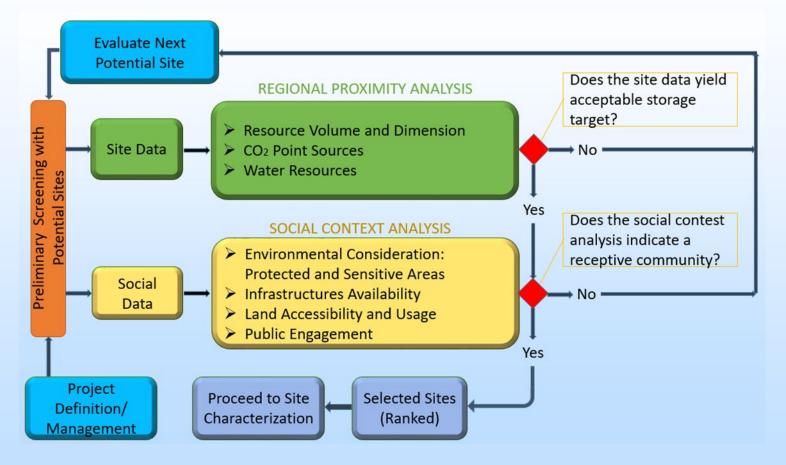
 Project Objective: Identify and access statewide resources for potential CO2 storage via mineralization processes, including near surface and subsurface basalt formations and related stratigraphic units, and/or mining wastes in the state of New Mexico, as well as identify and characterize potential targeted storage sites/complexes to provide insights on storage capacity.

• Tasks:



Site Selection-Near-surface Basalt

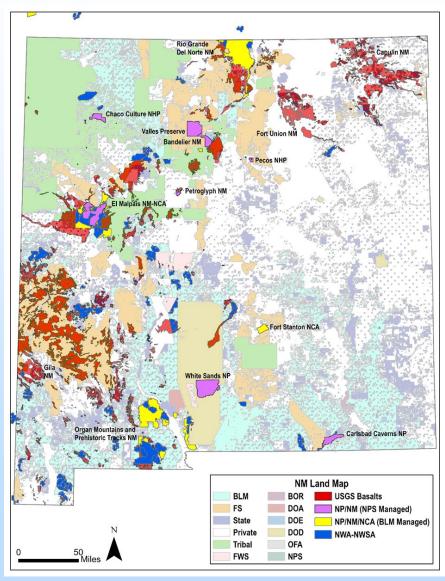


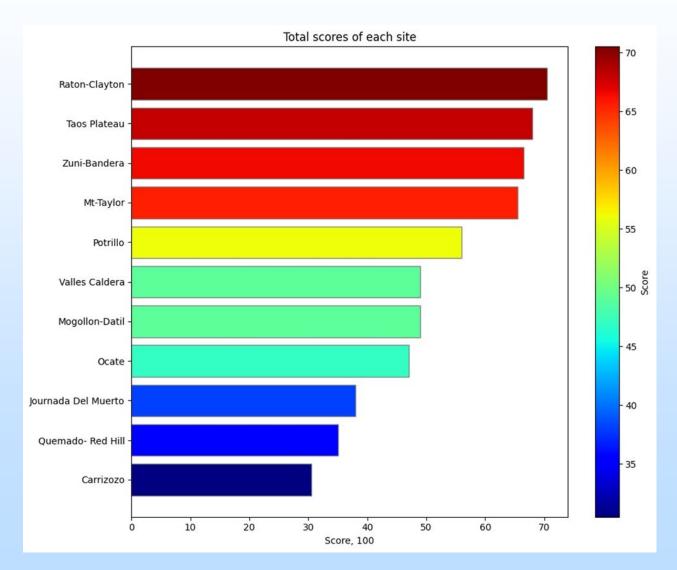


Site Suitability: Decision criteria are relevant to the specifics of CCUS via mineralization projects, such as: geologic formation volume, presence of divalent cation, proximity to sensitive areas, land access, CO2 sources, surrounding water resources, infrastructure availability and public engagement, etc.

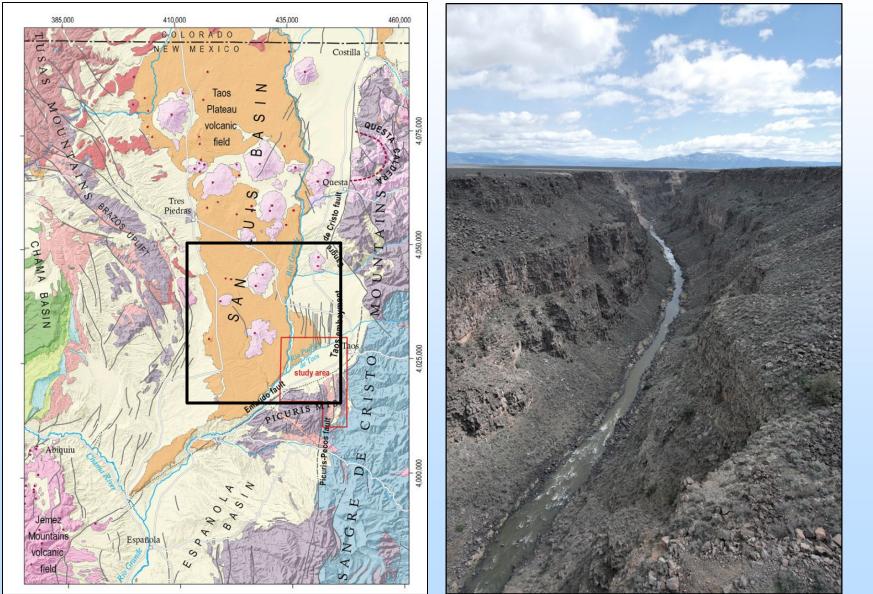
Geographical distribution of the basaltic rock in New Mexico

Site Selection-Near-surface Basalt





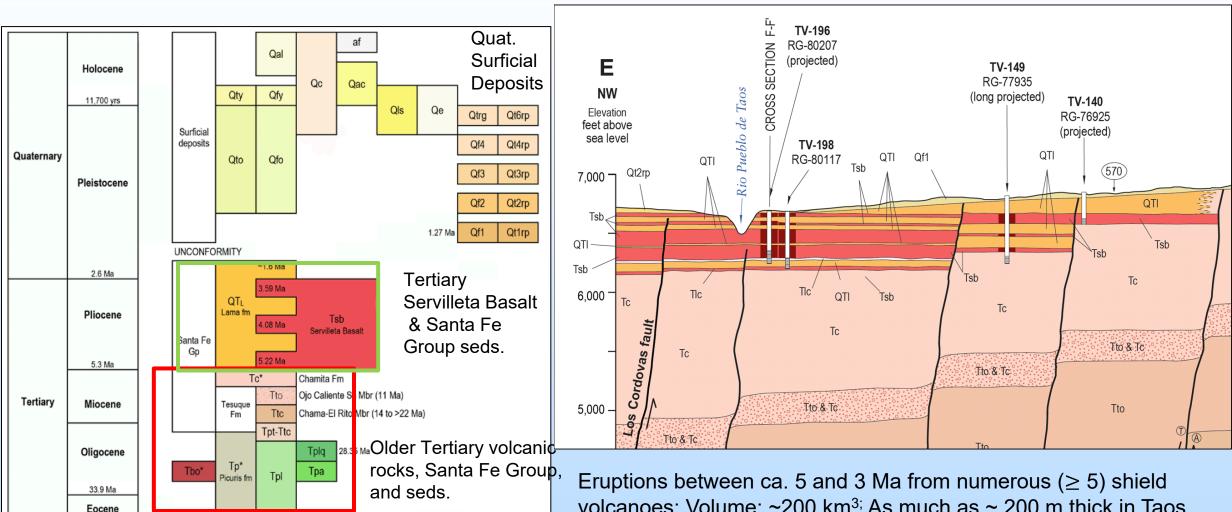
Taos Plateau Volcanic Field - Introduction



>6.0 Ma to 1.0 Ma ~2,500 km² ~250 km³ 35-50 exposed volcanoes(more buried) Compositionally diverse Late Cenozoic field in NM

Taos Plateau Volcanic Field - Stratigraphy

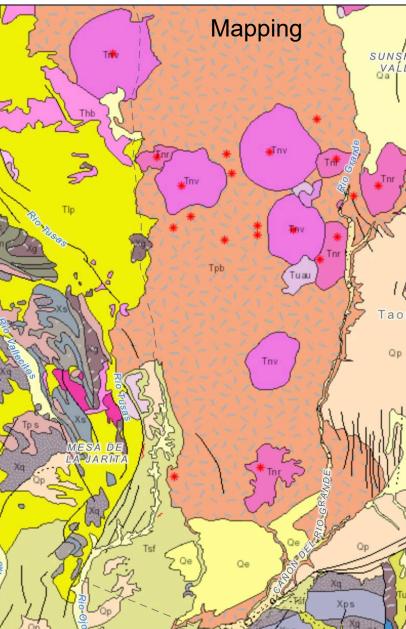
Servilleta Basalt

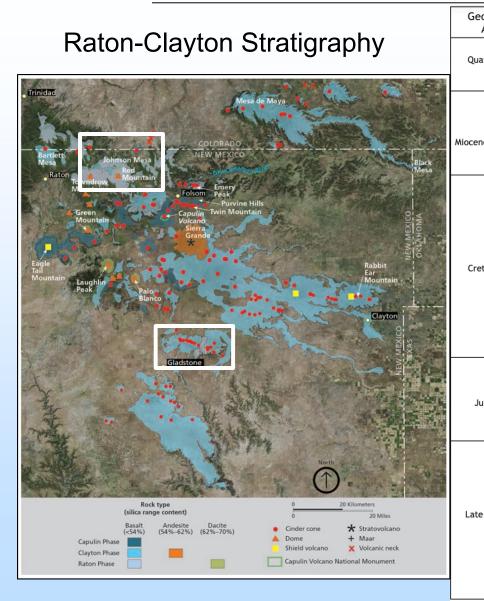


volcanoes; Volume: ~200 km^{3;} As much as ~ 200 m thick in Taos Gorge; thins to 1-2 m at margins of the field; Three subunits: Lower, Middle, Upper; Composed of thin Pahoehoe flows



- Porous
- Some extensive clay layers have been identified between the basalt flows. These clays layers act as impermeable zones that appear to be the seal zone candidate.
- Notable features includes variable fractures related to cooling, extensive vesicle pipes, and the diktytaxitic texture of the groundmass





eologic Age	S	tratigraphic Unit	General Rock Avg. Type(s) Thicknes				
			Eolian sand sheets, dune and alluvial deposits.	~0-30 m	1		
aternary	Raton-Clayton- Capulin volcanics		Dark gray to black basalts, cinder cones and fissure vents ranging in age from - 36 ka 9 Ma. Incl. Sierra Grande: med. gray andesite, -2.6-3.8 Ma.		$\begin{array}{c} \begin{array}{c} & & & & & & \\ \hline & \uparrow & \downarrow & & \uparrow & \downarrow \\ \downarrow & \uparrow & \downarrow & \downarrow & \uparrow \\ \downarrow & \uparrow & \land & \downarrow & \uparrow \\ \downarrow & \downarrow & \land & \downarrow & \uparrow \\ \downarrow & \downarrow & \uparrow & \downarrow & \downarrow \\ \downarrow & \downarrow & \downarrow & \downarrow & \downarrow \\ \downarrow & \downarrow & \downarrow$		
ne-Pliocene	Ogallala Fm.		Reddish-brown to tan coarse-grained sand with local lenses of pebble to cobble conglomerate. Heavily bioturbated. Locally capped by well-developed calcrete.	0 - 200 m			
	Smoky Hill Marl (Niobrara Fm.)		Dark gray silty to sandy shale with thin beds of limestone and marl.	305 m			
		Ft. Hays Ls.	Pale gray medium bedded limestone	15 m			
	Carlile Shale		Dark gray shale with thin limestone beds in upper section.	61 m			
etaceous		Greenhorn Ls.	Gray shale and pale gray medium-bedded micrite beds.	9 m			
	Graneros Shale		Medium gray shale with thin fossiliferous limestone beds.	38 m			
		Romeroville Ss.	Yellowish-gray medium-grained, locally pebbly sandstone.	0-8 m			
	ota	Pajarito Shale	Medium gray shale.	10-20 m			
	Dakota Groun	Mesa Rica Ss.	Brownish-yellow persistent medium grained, cross-bed- ded sandstone.	33 m			
	Glencairn Fm.		Gray to dark gray shale, siltstone and sandstone.	22 m			
	Lytle Ss.		Light gray conglomeratic cross-bedded sandstone.	10-20 m			
urassic	Morrison Fm.		Gray-green and red mudstone with locally thick medium to coarse-grained sandstone and thin micrite beds.	52-168 m			
			Dark brown mudstone with nodules of alabaster.	0-8 m			
	Exeter Sandst.		White to pale pink cross-bedded sandstone.	0-24 m)		
e Triassic		Sheep Pen Ss.	Light-brown, thin-bedded sandstone.	0-33 m			
		Sloan Canyon Fm.	Red and pale gray-green mudstone with lenses of medium-grained sandstone.	0-46 m			
	Dockum Group	Travesser Fm.	Reddish-brown siltstone and sandstone with local intraformational conglomerate lenses.	75-168 m			
		Baldy Hill Fm.	Purple, red and green mottled mudstone with lenses of coarse-grained sandstone. Base not exposed.	> 30 m			

 Lithology Key
 Sandstone

 Basalt
 Limestone

 Conglomerate/
 Siltstone

 Siltstone
 Conglomerate/

Eruptions between ca. 9 Ma and 37 ka

- ~140 vents (mostly cinder cones)
- 7,000-10,000 km2
- ~100-200 km3

Approximately 1,200 feet thick

Basalt-Andesite-basalt

Two subregions of interest: Johnson Mesa & Don Carlos Hills

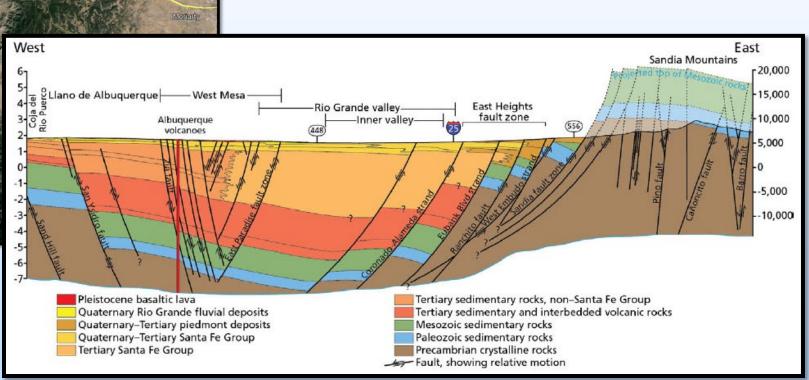
Sandia Cre

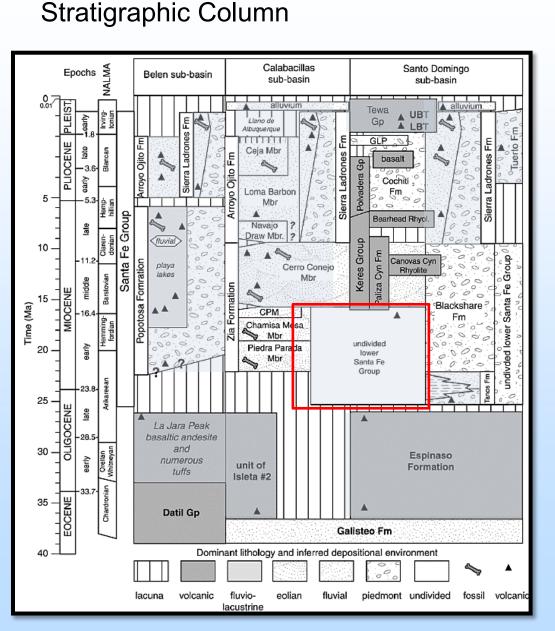
Sandia

Subsurface Basalt – Albuquerque Basin



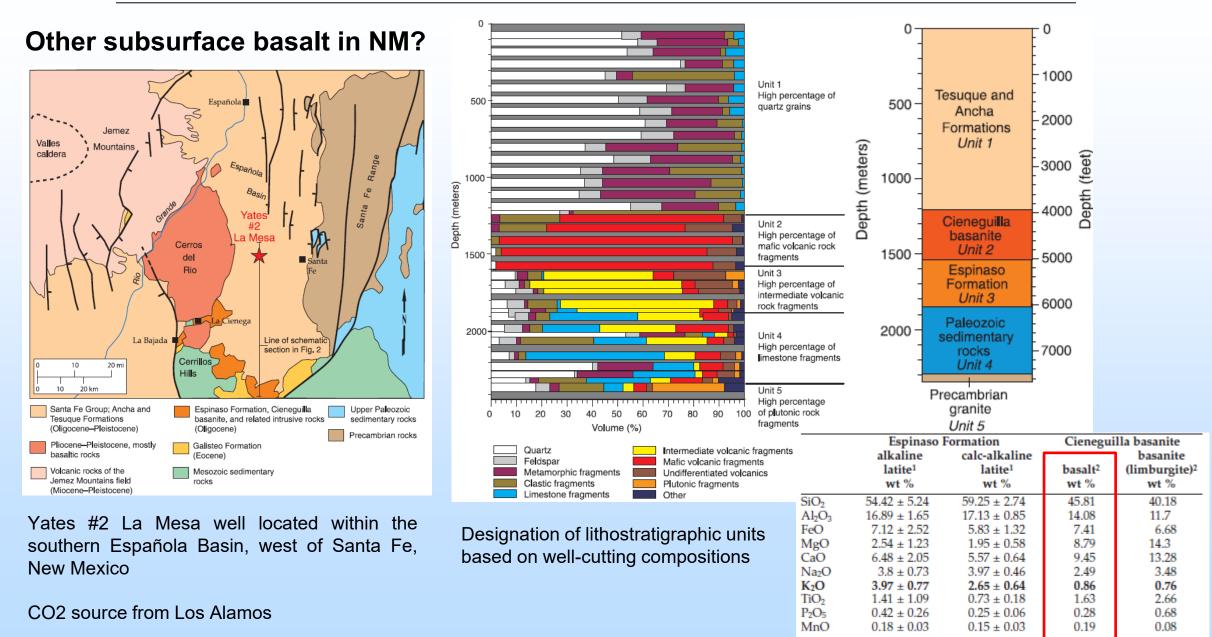
- Albuquerque Basin is a sub-basin in the Rio Grande Rift
- Wells with potential Basalt zones
- Most of the wells have cuttings or core available
- Shell #2 Isleta & Carpenter #1 Atrisco Grant (Arrows) may have the greatest potential



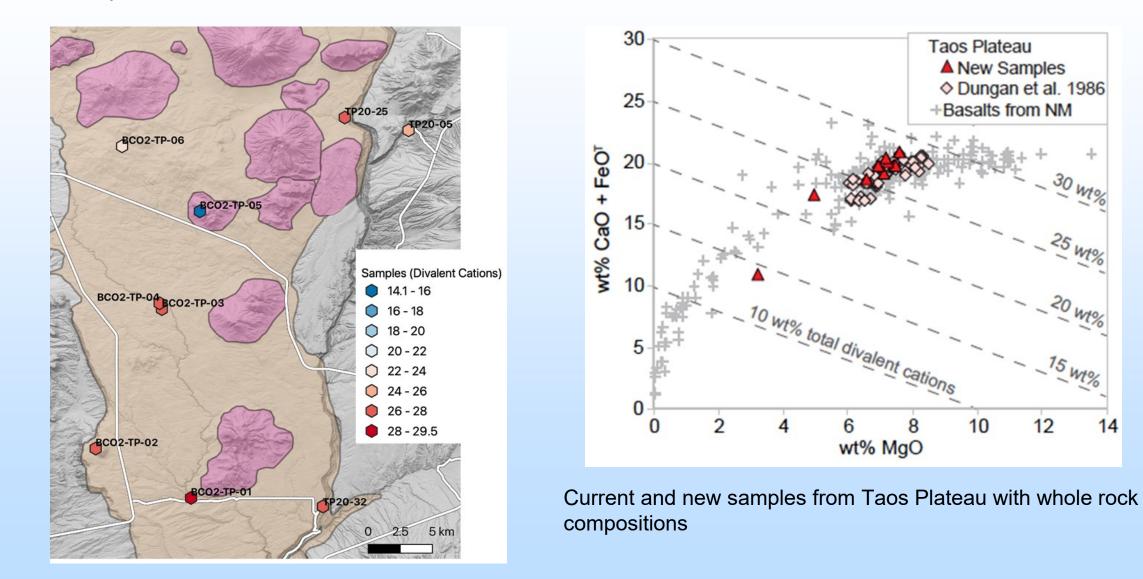


API	Well	Location	Depth (ft)	Basalts	Depths
30-001-05004	Carpenter #1Atrisco Grant	28-10N-1E	6,652	Basalts	1,570-1,580;
					3,350-3,455 sill?;
					4,130-4,140;
					4,175-4,185 volcanic cave?;
					4,560-4,600;
					4,910-4,970 sill?;
					5,170-5,210 sill?;
					5,500-5,670;
					6,320-6,360
30-061-20004	Shell #2 Santa Fe (cutting)	29-6N-1W	14,305	Basalts	14,700-14,800;
					14,900-15,200
30-061-20008	Shell Isleta Central (cutting)	7-7N-2E	16,346	Basalts	3,300-3,520;
					6,950-7,000;
					7,300
30-061-20031	Grober #1 Fuqua	19-5N-3E	6,500		Not readable
30-001-20002	Trans Ocean Isleta #1	8-8N-3E	10,378		Not readable
30-001-20003	Shell Isleta #2 (core, cutting)	16-20N-13W	21,266	Basalt	1,570-1,600 vesicular;
					4,100-4,200 tuffs;
					5,500-5,600 scoria
30-001-20004	Shell West Mesa Federal #1-	24-11N-1E	19,375	Basalt	13,850-13,900;
	24 (core)				14,700-14,800;
					14,900-15,200 abundant
					fragments

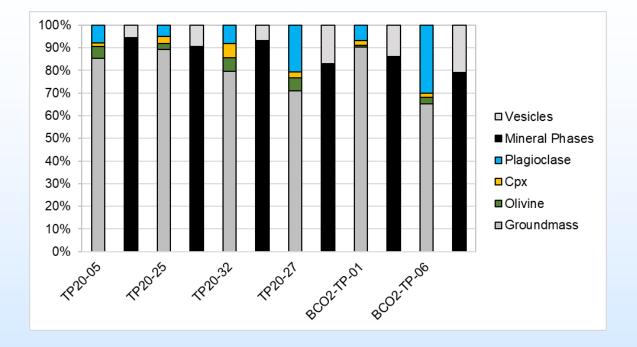
- Santa Fe Group deposits: Basalt flows interbedded with sedimentary basin fills (Sandstones, siltstones, shales, etc). Some volcanic sills as well.
- Strip logs were used initially to identify horizons with basaltic clasts.
- Listed wells that contained basalts, their depths, and there are core and/or cuttings at the NMBGMR.



Sample characterization



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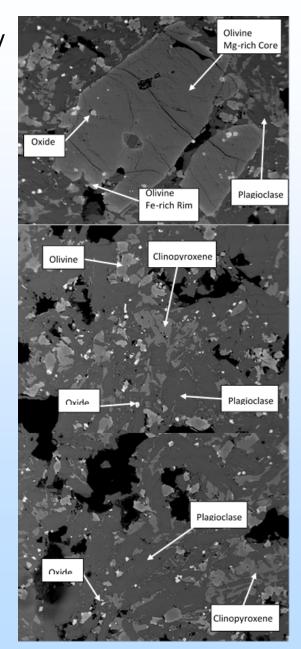


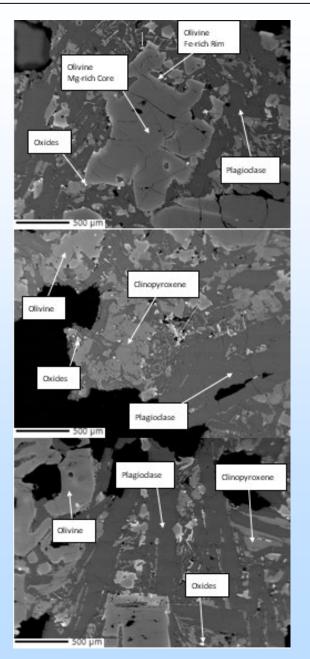
Results of point counts, are shown as bar graphs for mineral and groundmass assemblages (left bar, blue, orange, green and grey) and for the sample vesicularity (right bar, black and grey). bar).

- point counts to describe the overall mineralogy of the samples
- Samples have phenocryst abundances that range from ~10-25%, with assemblages consisting of olivine + clinopyroxene + plagioclase + Fe-Ti oxides. In general, plagioclase is the most abundant mineral in all assemblages followed by either olivine (e.g., TP20-05) or by pyroxene (e.g., TP20-32).
- Samples range in vesicularity from ~6% vesicles to ~20%.



Sample TP20-27. Point counting of this thin section had results of 17.05% vesicles and 82.95% crystalline material. Out of the crystalline material, 70.85% was groundmass, 20.79% was plagioclase, 5.84% was olivine.





Electron microprobe analyses(EMA) on the mineral phases within samples. The backscatter electron images of major mineral phases for all the samples shows:

- Olivine tends to be normally zoned, where the cores are magnesium rich and the faint rims are iron rich.
- Small olivine crystals also make up portions of the groundmass assemblage in all samples.
- Fe-Ti oxides occur as inclusions in olivine and also occur in the groundmass in all samples.

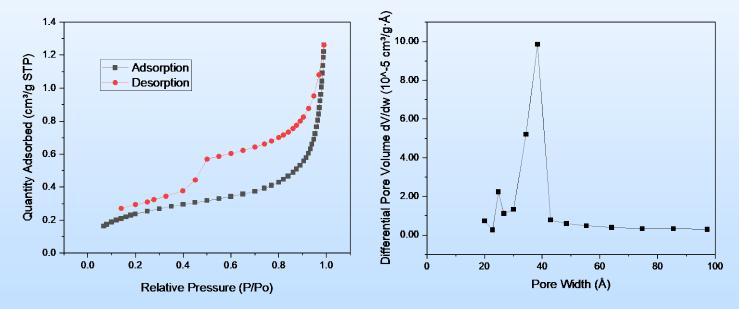
Geophysical Properties

The BET-specific surface area was analyzed for basaltic rock samples from Taos using Micrometrics ASAP 2020 Plus 2.0.

Method:

- Adsorbs gas molecules (e.g., nitrogen) onto the material's surface.
- Measures gas adsorption at various relative pressures.
- Uses the BET equation to calculate surface area.





- The specific surface area for this study was $0.9178 \pm 0.0055 \text{ m}^2/\text{g}$.
- The hysteresis loop is a type IV isotherm as shown in figure 1.
- The particle size width is 40 nm and corresponds to a mesopore.

Geomechanical Properties

Objective: Obtain mechanical properties of candidate resource rocks and changes in mechanical properties as a result of CO₂ injection.

Progress: Experimental test plan is outlined, sample prep in progress

Planned experiments and mechanical properties:

Planned Experiment	Triaxial shear*	Uniaxial Compressive Strength (UCS)*	Indirect tension	Triaxial compression	Notched 3-ptbend*
Measured	friction at confinement	cohesion,	tensile	internal	fracture and material
parameter		internal friction	strength	friction	toughness

*test will be run on reacted and unreacted samples

Reaction Dynamic





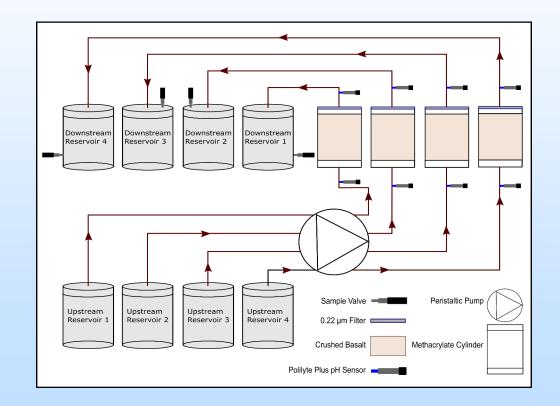
- Conduct the batch-type experiments to address dissolution and precipitation kinetics (Ambient T&P, Insitu P&T).
- GEM-Selektor code package based on the Gibbs energy minimization method will be used for thermodynamic and kinetic simulations of fluid-rock reactions during CO2 mineralization.
- Geochemist Workbench to simulate the dissolution and precipitation kinetics

- Multi-element analysis, determine which elements we could reliably detect
- Analyzed for major elements: Ca, Mg, Na, K, Si, Fe, Ba,

Dynamic Flow-through Experiments

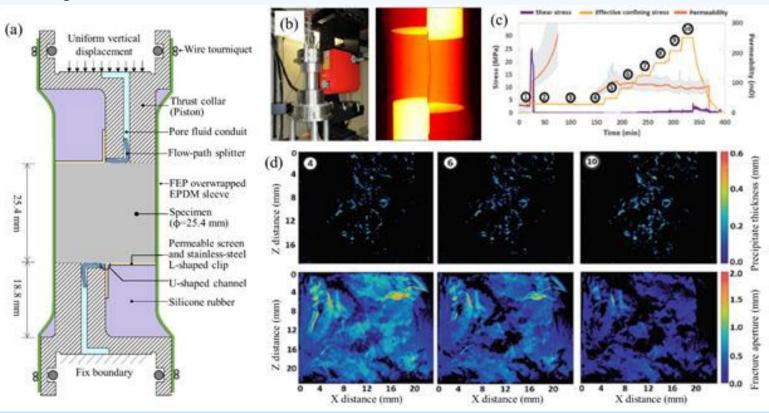
Objective: Measure rates of chemical reactions of potential reservoirs during continuous exposure of different waters as a function of grain size vs. fracture area, while monitoring for changes in mechanical properties.

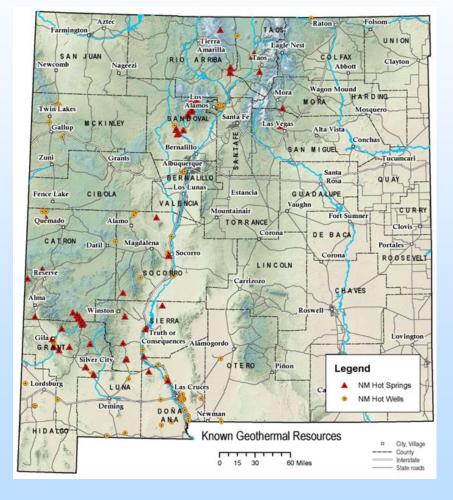
Table 1. Proposed Experimental conditions for reactive tests, B.E. is basalt equilibrated.					
Test	Basalt	Reactive	Experimental	Grain Size	
#	Туре	Mechanism	Solution	(um)	
1	Pristine	Dissolution	GW + CO2(a)	44 - 125	
2	Vesicular/Fractured	Dissolution	GW + CO2(a)	44 - 125	
3	Pristine	Precipitation	BE GW + CO2(a)	44 - 125	
4	Vesicular/Fractured	Precipitation	BE GW + CO2(a)	44 - 125	
5	Pristine	Dissolution	GW + CO2(a)	250 -500	
6	Vesicular/Fractured	Dissolution	GW + CO2(a)	250 -500	
7	Pristine	Precipitation	BE GW + CO2(a)	250 -500	
8	Vesicular/Fractured	Precipitation	BE GW + CO2(a)	250 -500	
9	Pristine	Dissolution	GW + CO2(a)	1000 -2000	
10	Vesicular/Fractured	Dissolution	GW + CO2(a)	1000 -2000	
11	Pristine	Precipitation	BE GW + CO2(a)	1000 -2000	
12	Vesicular/Fractured	Precipitation	BE GW + CO2(a)	1000 -2000	
13	Pristine	Recirculating	GW + CO2 (a)	whole	
14	Vesicular/Fractured	Recirculating	BE GW + CO2(a)	whole	



Progress: Experimental design is complete. In process of fabrication of flow cells.

- Perform the flow-through tests integrated with real-time X-ray microtomography to investigate sustainability of CO2 injection into the target basalt samples.
- Using Elevated Heat Flow to Enhance Reaction Rates.





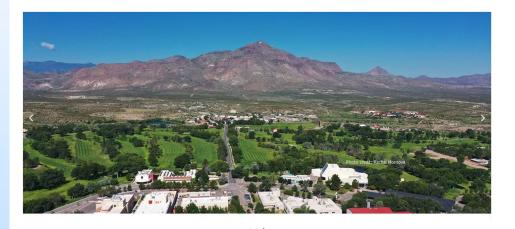
Evolution of fracture geometry and precipitate growth based on a triaxial directshear (TDS) experiment integrated with fluid flow and real-time X-ray micro tomography

Outreach Activity

Website in development

- Collaboration with Arizona Geological Survey to synchronize outreach in the region.
- Share project information
- Share CO₂ mineralization research https://co2rocks.net/

CO2 Mineralization RESEARCH STAKEHOLDER ENGAGEMENT NEWS



Harnessing Potential: Pioneering CO2 Storage through Strategic Resource Assessment

Welcome to the Regional Resource Assessment for CO2 Storage Project!

The "Regional Resource Assessment for CO2 Storage in New Mexico and Surrounding Areas" project aims to identify, characterize, and evaluate potential sites for CO2 storage through mineralization processes.

This initiative focuses on basalt formations, related stratigraphic units, and mining wastes to provide a comprehensive understanding of storage canacities and potential benefits

Our goal is to identify and assess New Mexico state resources for potential CO2 storage via mineralization, focusing on basalt formations and related stratigraphic units, as well as mining wastes

Workshop in preparation

- November 7th-8th, 2024 in Socorro, New Mexico
- The event gather around 100 energy stakeholders from NM
- Collaboration with the Consortium for Sustainable Energy and Advanced Management (CESAM)

https://nm-secm.org/outreach/

Consortium for Energy Sustainability and Advanced Management (CESAM)



REGISTRATION Coming soon

Contact information: jean-lucien.fonguergne@nmt.edu

Energy Research and Collaboration Outreach and Community Engagement Education and Workforce development



Consortium for Energy Sustainability and Advanced Management (CESAM)

Day 1, November 7th:

- Panel: Overview of NM Universities and National Laboratories Energy Research, Education and Outreach



- To be confirmed: Navajo Technical University Sandia National Lab University of New Mexico
- Panel: Overview of NM Energy Partnerships



- Four Corner Energy Alliance
- Panel: Solar, Wind and Storage
- Panel: Subsurface Energy and Storage
- Panel : Carbon Management

Day 2, November 8th:

- Panel: Mining Innovations and Challenges
- Panel: Nuclear Research in New Mexico
- Panel: The Role of Water in Energy
- Panel: Environmental Sustainability
- Panel: Education and Community Engagement

Outreach Activity

Article in development

- Will be released in the Outcrop magazine published by the Rocky Mountain Association of Geologist in November 2024. <u>https://www.rmag.org/</u>
- Another in preparation for Spring, 2024 for the Lite Geology Magazine (NMT)

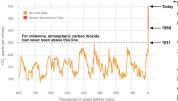


Turning CO₂ into Stone: The Potential and Challenges of CO₂ Mineralization for Carbon Sequestration

To combat climate change, humanity is turning to innovative technologies to reduce atmospheric carbon dioxide (CO₂) levels. Among these, CO₂ mineralization for Carbon Capture and Sequestration (CCS) stands out as a promising solution. This process not only captures CO₂ emissions from industrial sources and the atmosphere but also permanently stores them in solid minerals, preventing them from subsurface leakage and contributing to global warming. This article delves into the importance of CO₂ mineralization, highlights its implementation on an industrial scale, explains its principles, contrasts it with other sequestration methods, outlines its advantages and challenges, and presents a recent project in New Mexico, led by New Mexico Institute of Mining and Technology, aiming to harness this technology for environmental sustainability.

Why CO2 mineralization is a necessity

The concentration of CO₂ in the atmosphere has surged to levels not seen in millions of years, primarily due to human activities such as fossil fuel combustion and deforestation. This increase in greenhouse gases is a major driver of climate change, leading to extreme weather events, rising sea levels, and loss of biodiversity. Thus, reducing atmospheric CO₂ levels is crucial. CO₂ mineralization offers a way to effectively removing it from the atmosphere for millennia.



climate chan

"he atmospheric concentration of CO, has creased by 50% since the onset of dustrial times in the 18th century. Data om National Oceanic and Atmospheric dministration's Observatory shows a ontinuous rise in atmospheric CO, levels nce 1958, while ice core samples reveal CO, vels during Earth's last three glacial cycles. he historical increase in CO, from 365 parts er million (ppm) in 2002 to over 420 ppm, ighlights the impact of human activities on

Community Engagement

- Initial Assistance & Validation Meeting, November 9, 2023
- Outreach and engagement with land owners from the Don Carlos hill and Johnson Mesa, near the sites of interest. More than 15 landowners engaged via meetings, phone calls or mail.
- Engagement with Freeport, mining company.
- Contact with University of Eastern New Mexico for outreach event.
- Engagement with students and New Mexico communities at the Science Café in Socorro, NM.



Great outreach day today at the Science Café organized by the New Mexico Bureau of Geology Mineral Museum at New Mexico Institute of Mining and Technology !

It was a fantastic opportunity for everyone to dive into the world of science through fun and engaging activities.

We discussed our Carbon Capture and Storage (CCS) projects, the process of CO2 mineralization, and looked at beautiful calcite samples. Thanks to all who joined us and contributed to these conversations about our planet's future.

Kudo Cynthia Connolly and all the people involved in the organisation the event!

#ScienceForEveryone #Geology #CCS #CO2mineralization

Voir la traduction

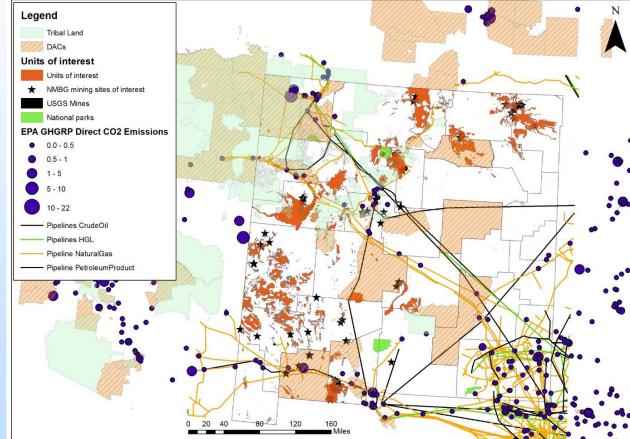




Figure 1: Evolution of CO, concentration in the atmosphere. Data source Reconstruction from ice cores. Credit NOAA (https://climate.nasa.gov/vital-signs/carbon-dioxide/?intent=121)

Future Plans

- Continue to study the geological and hydrogeological properties on the selected sites.
- Continue to characterize on the petrological, mineralogical, geochemical, geophysical and geomechanical properties of resource rock.
- Study reaction dynamics of the CO2 mineralization process on the localized resource rock in order to indicate the optimum scenario for CO2 storage and upscaling.
- Outreach activities. We will continue our efforts on engagement with Minority Serving Institutions and stakeholders in the project area for outreach event and develop outreach materials (Articles, website, social media, workshops...)



Acknowledgements

The project would like to thank DOE for the award opportunity through DE-FE0032257 and our partners.

