



Reimagining the carbon ecosystem

# Dual Function Materials for Direct Air Capture of CO<sub>2</sub>

DOE SBIR Grant: DE-SC0020795

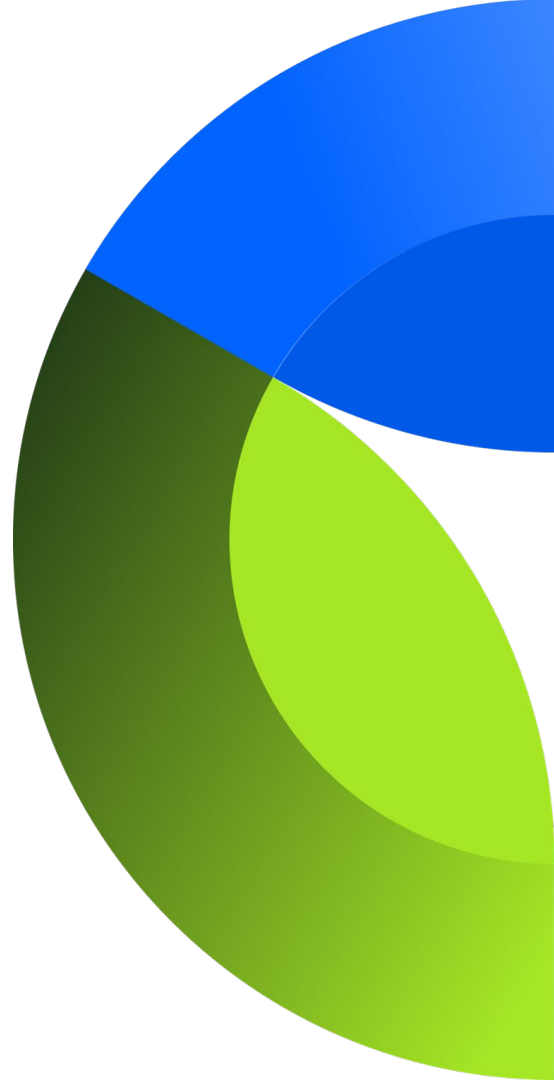
Presented by  
Jonathan Peters, Principal Investigator

Federal Project Manager  
Zachary Roberts

August 6, 2024



[www.susteoninc.com](http://www.susteoninc.com)



# Project Overview

DOE SBIR Grant: DE SC0020795	Start Date	End Date	Award Amount
Phase I	6/29/2020	3/28/2021	\$250,000
Phase II	8/23/2021	8/22/2023	\$1,600,000
Phase IIA	8/28/2023	2/27/2025	\$1,150,000

## Key Susteon Personnel



**Jonathan Peters**  
Sr. Process Engineer



**Monica Abdallah**  
Research Engineer



**Jian-Ping Shen**  
Senior Principal Scientist



**Raghubir Gupta**  
CEO/Co-Founder

## Partners



SASOL



COLUMBIA UNIVERSITY  
IN THE CITY OF NEW YORK



AngloAmerican

CORNING



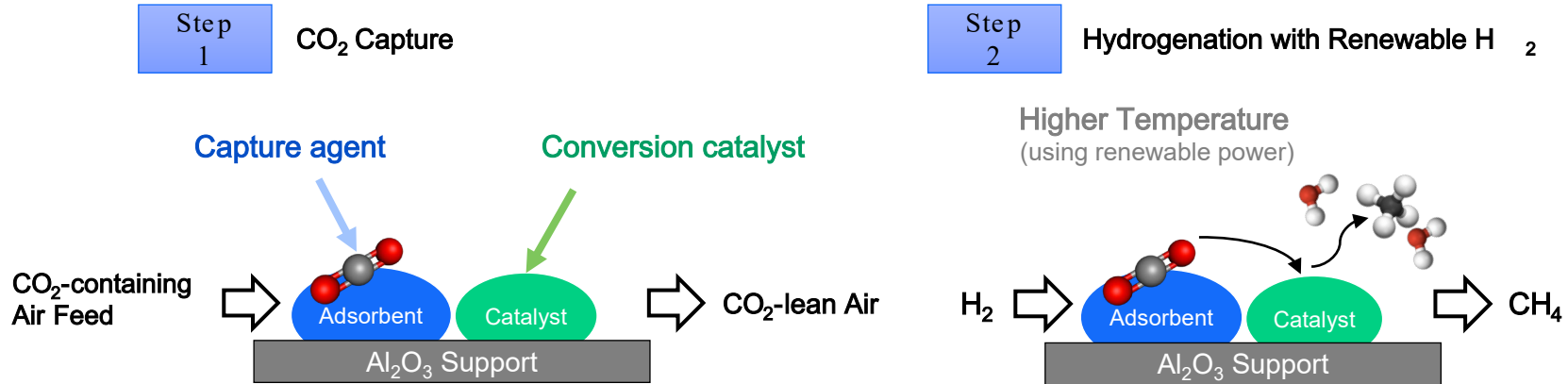
NGK INSULATORS, LTD.



TotalEnergies

# Reactive Carbon Capture Platform: Dual Function Materials

**Dual function materials (DFMs)** are sorbent-catalyst combinations that can capture CO<sub>2</sub> from a point source or DAC and directly convert it into a value-added product.



The DFM process is a **Power-to-X platform** for converting **renewable power into methane** that is compatible with the current pipeline infrastructure.

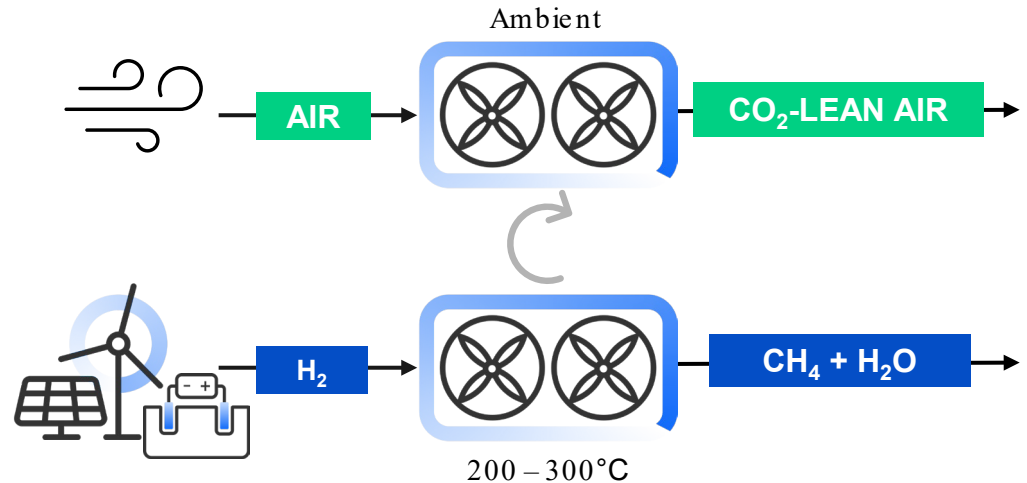
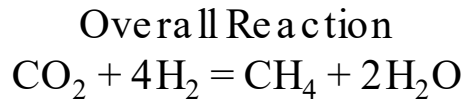
# The DAC-DFM Process for Reactive Capture

**Objective:** Lower the cost of DAC through development of advanced dual-functional materials (DFM) and production of renewable natural gas (RNG) from CO<sub>2</sub>

## Two Step Process Cycle

**Step 1:** Adsorb CO<sub>2</sub> from air onto DFM at ambient conditions

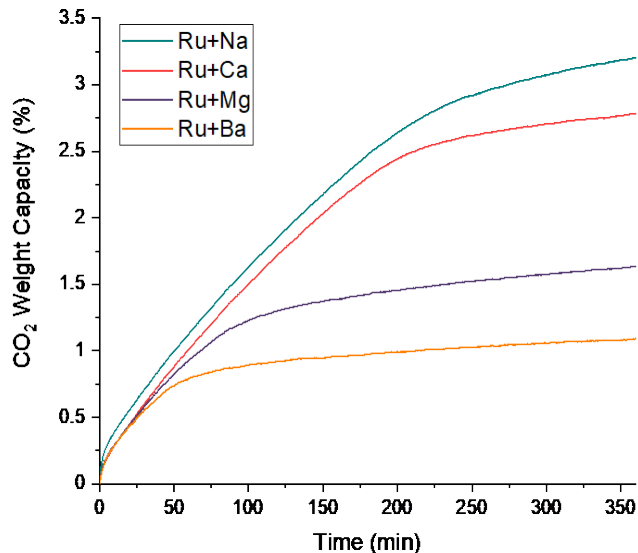
**Step 2:** Add renewable H<sub>2</sub> and heat to regenerate the sorbent to directly produce methane



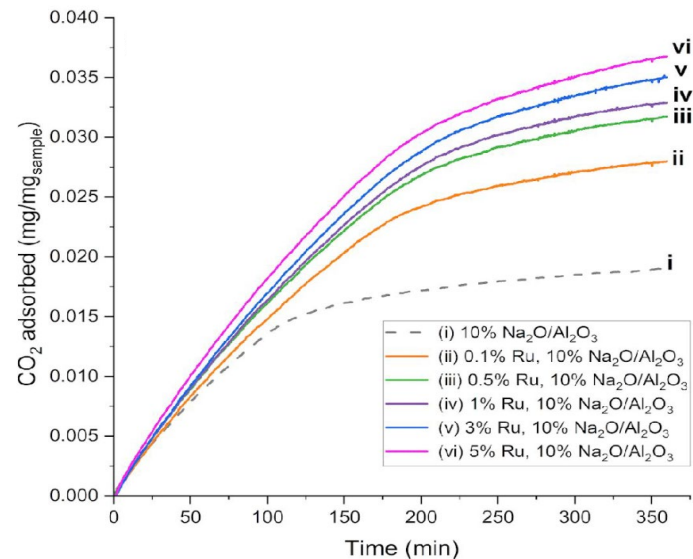
This is a **Power-to-Gas** technology using atmospheric CO<sub>2</sub>.

# Phase I Achievements: DFM Formulation Ru + Na sorbent

1% Ru, 10% Na<sub>2</sub>O/Al<sub>2</sub>O<sub>3</sub> (green) shows the highest CO<sub>2</sub> capture capacity (~3 wt.%).



**Additional insight** : Ru enhances CO<sub>2</sub> capture capacity of sorbent (solid v. dotted lines).



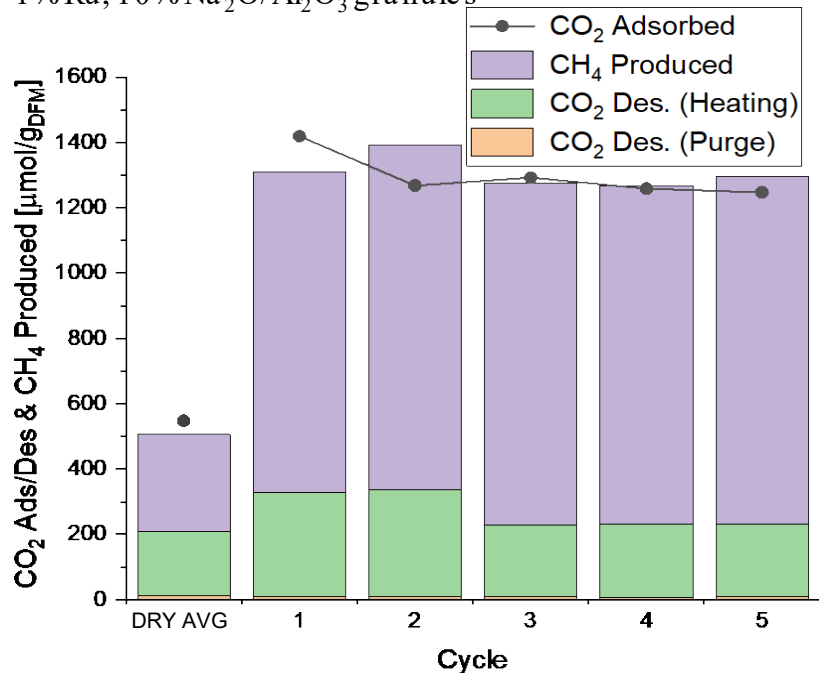
**Thermal gravimetric analysis:** Adsorption @ 25°C on 1% Ru, 10% sorbent/Al<sub>2</sub>O<sub>3</sub> granules with 375 ppm CO<sub>2</sub>/air

Jeong-Potter, et al. *Industrial & Engineering Chemistry Research*. 2022. DOI:10.1021/acs.iecr.2c00364  
Jeong-Potter, et al. *Applied Catalysis B: Environmental*. 2022. DOI:10.1016/j.apcatb.2021.120990

# Phase I Achievements: DFM Formulation Ru + Na sorbent

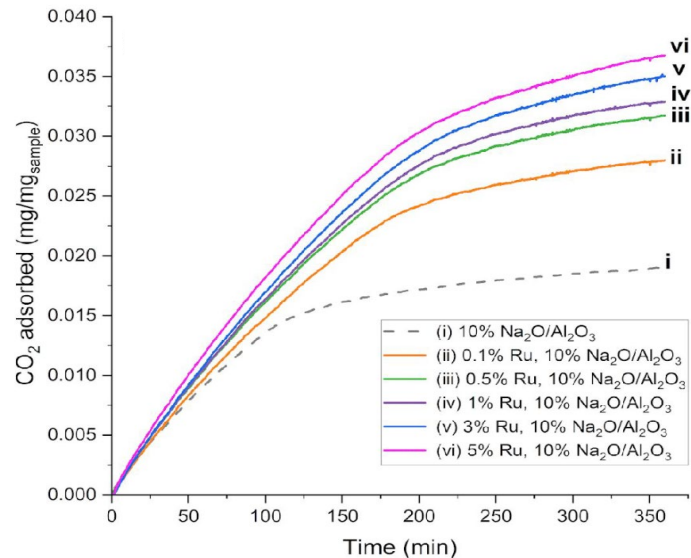
## Concept Feasibility in Lab -Scale Reactor

1% Ru, 10% Na<sub>2</sub>O/Al<sub>2</sub>O<sub>3</sub> granules



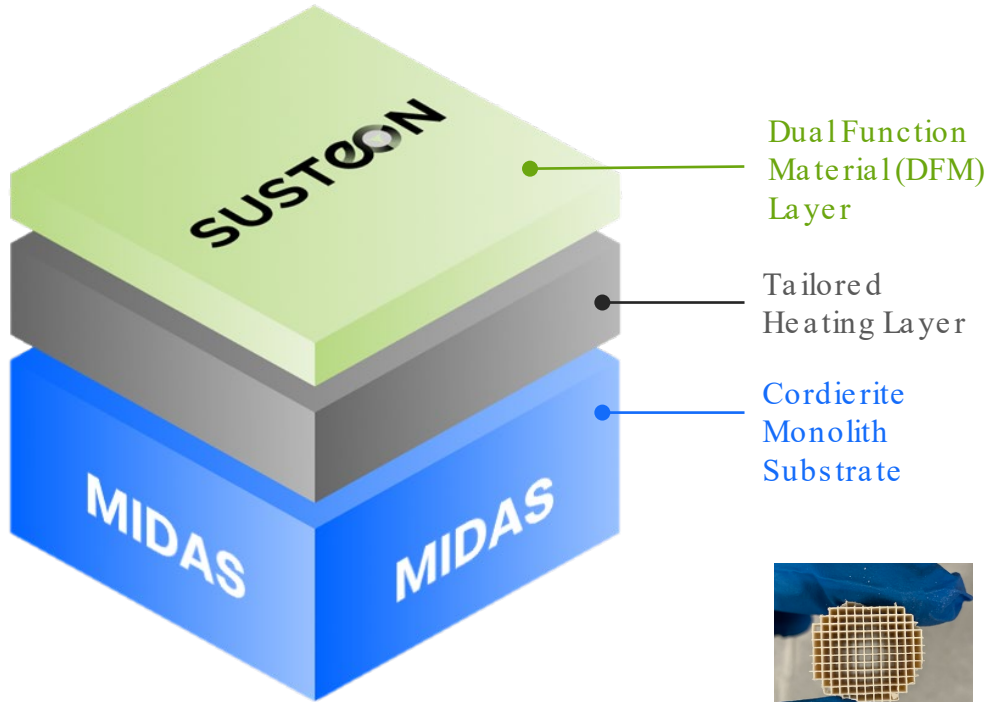
Avg. CO <sub>2</sub> Captured	Avg. CH <sub>4</sub> Produced
1300 μmol/g <sub>DFM</sub>	1040 μmol/g <sub>DFM</sub>

**Additional insight** : Ru enhances CO<sub>2</sub> capture capacity of sorbent (solid v. dotted lines).



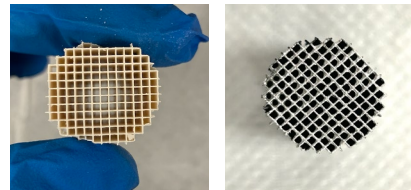
Jeong-Potter, et al. *Industrial & Engineering Chemistry Research*. 2022. DOI:10.1021/acs.iecr.2c00364  
 Jeong-Potter, et al. *Applied Catalysis B: Environmental*. 2022. DOI:10.1016/j.apcatb.2021.120990

# Phase II: Electrification via Structured Material Assembly (SMA)



Our SMA integrates the heating layer, sorbent, catalyst, support and monolith substrate. **Key advantages:**

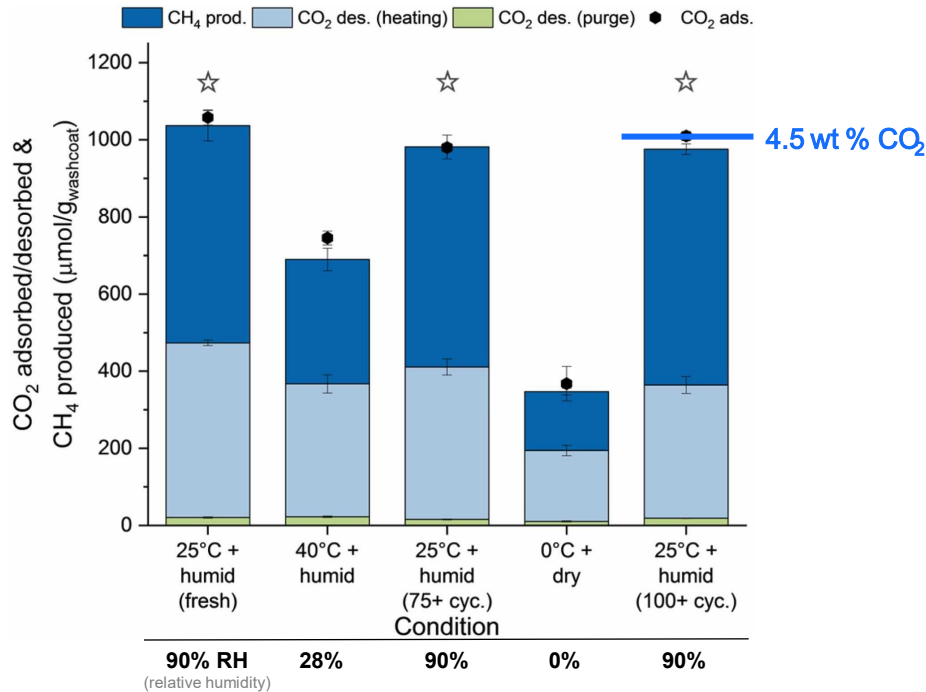
- ✓ **Increases productivity** by enabling fast CO<sub>2</sub> adsorption and fast conversion
- ✓ **Lowers the energy utilization** by reducing pressure drop during adsorption and energy losses during methanation
- ✓ **Rapidly heats** between adsorption and conversion
- ✓ **Powered by low -carbon electricity** for maximum conversion efficiency



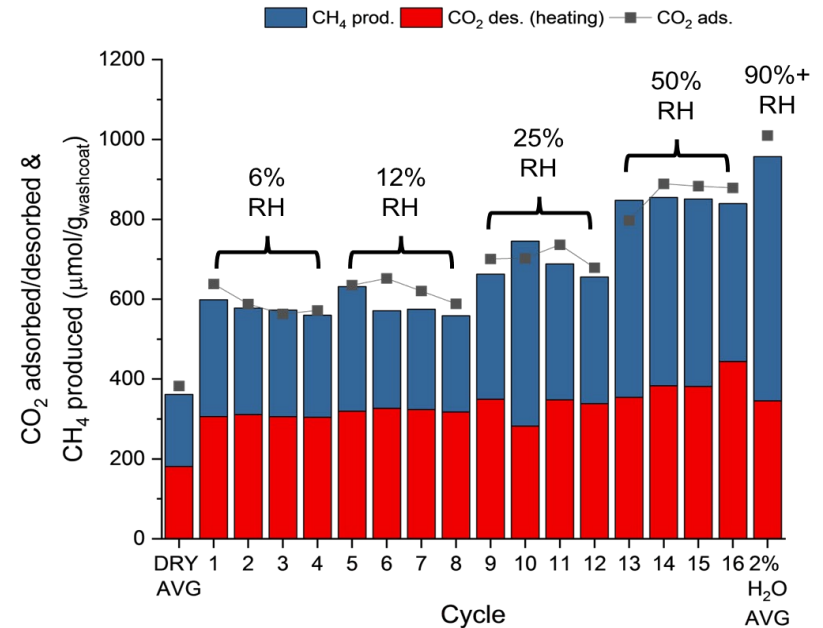
Before washcoating After washcoating

# Phase II: DFM Monolith Showed Long-Term Stability >100 Cycles

CO<sub>2</sub> capacity **returns to 4.5 wt.%** after exposed to extreme climate conditions over 100+ cycles.

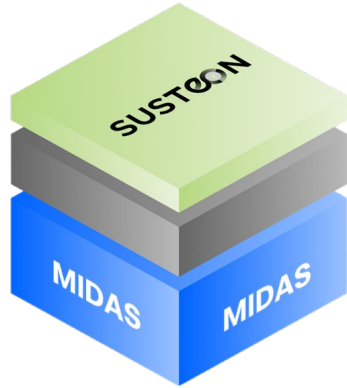


CO<sub>2</sub> capture is **enhanced** with higher moisture content in the feed



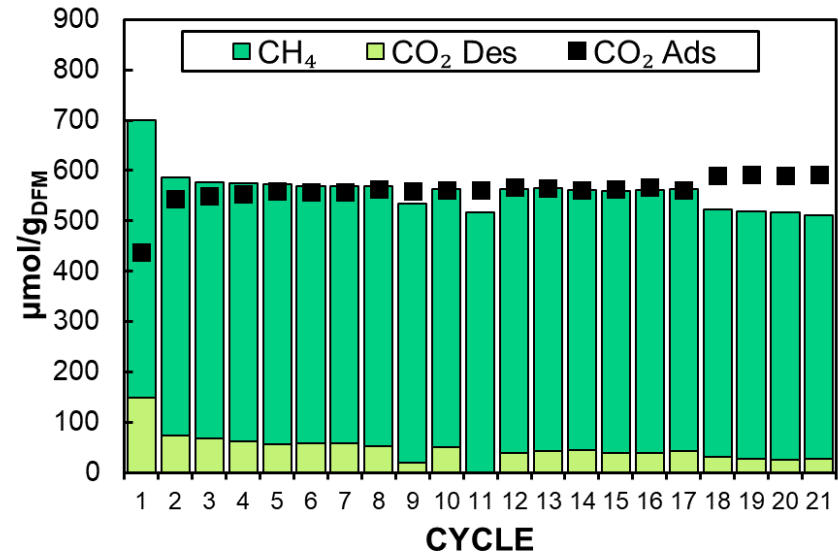


# Phase II: Demonstrated Joule Heated DFM Cycles

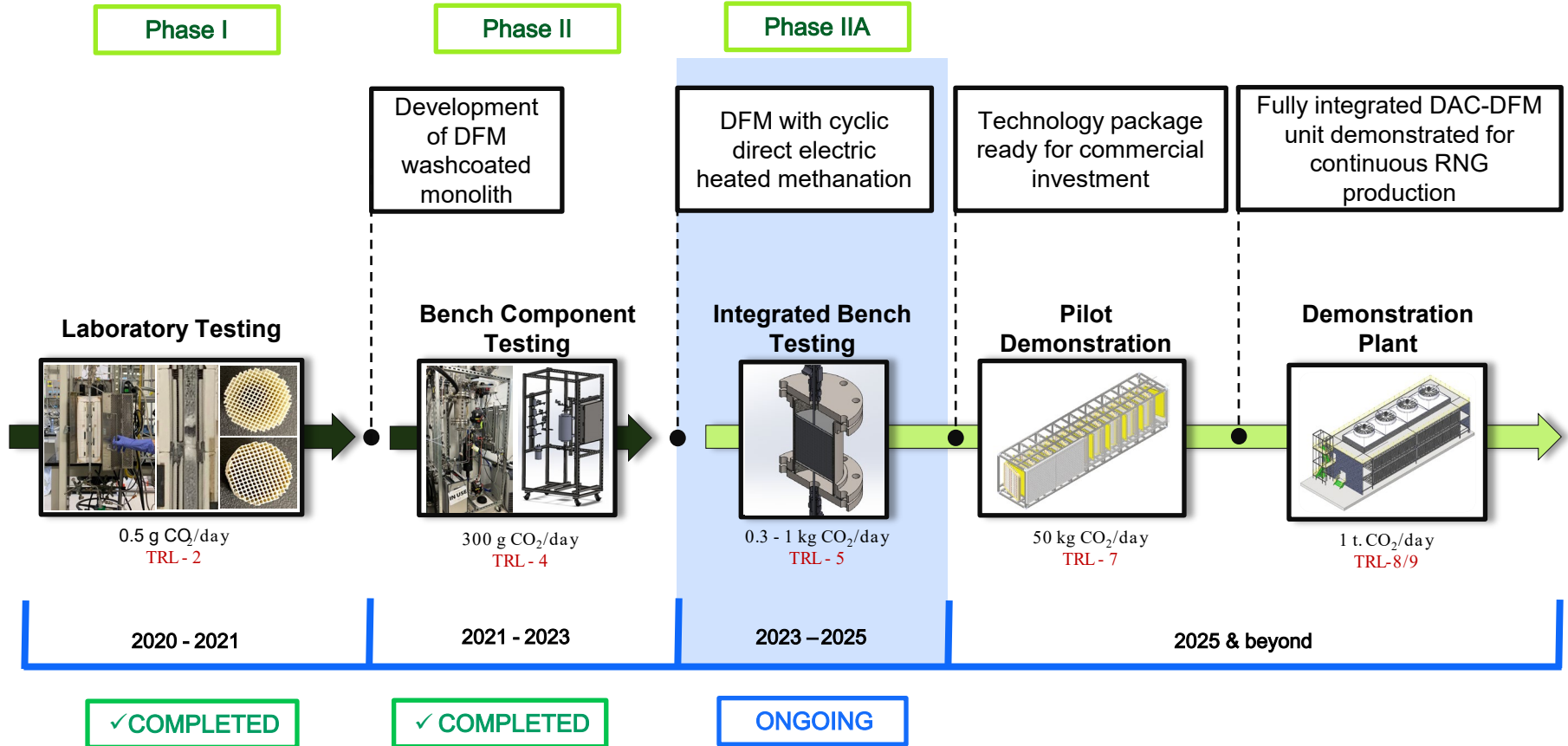


Reliable, cyclic CO<sub>2</sub> conversion is achieved using the DFM SMA with Joule heating layer

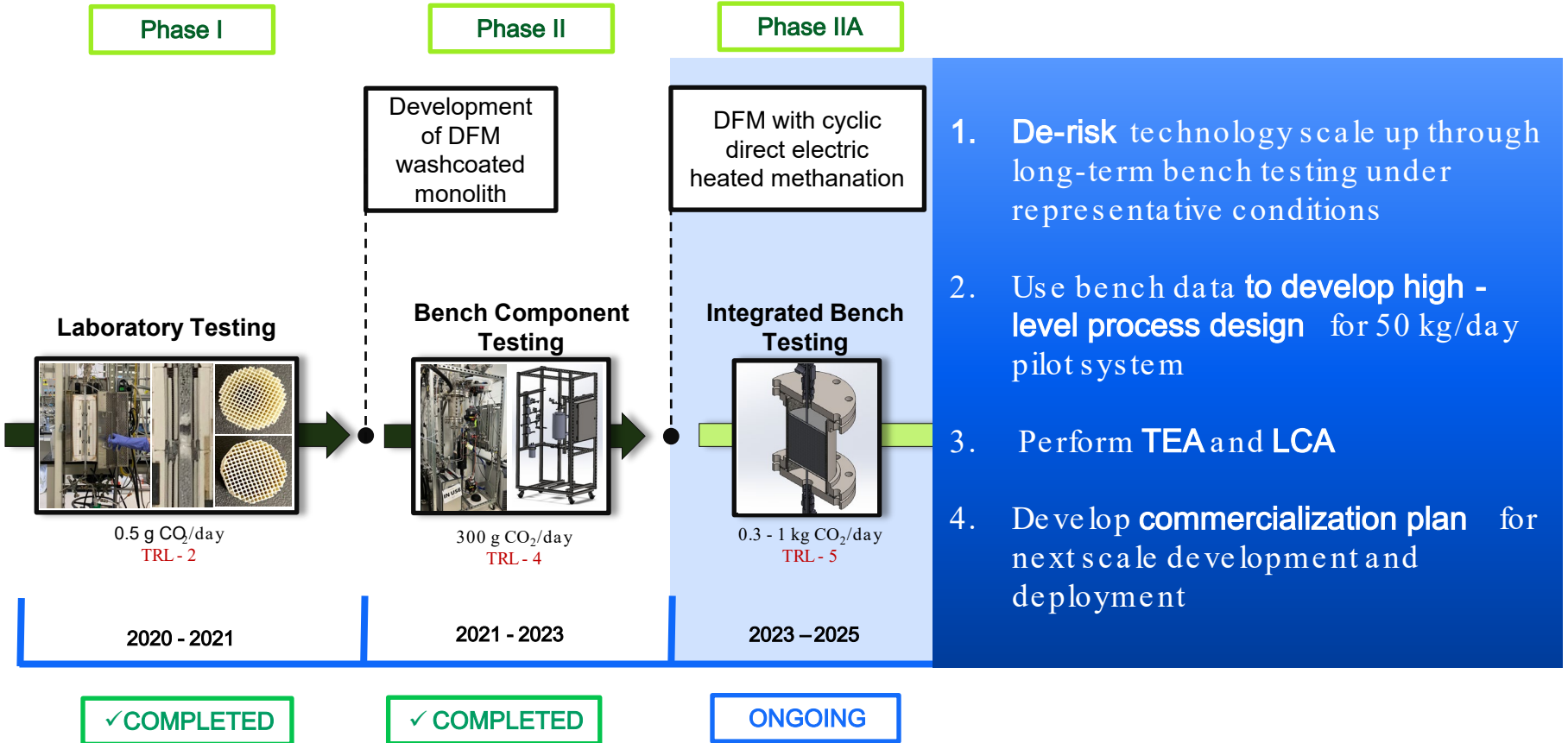
Additional screening tests showed over 70% conversion of CO<sub>2</sub> to CH<sub>4</sub> during 300+ hours of continuous testing



# State of Technology & Objectives of Phase IIA

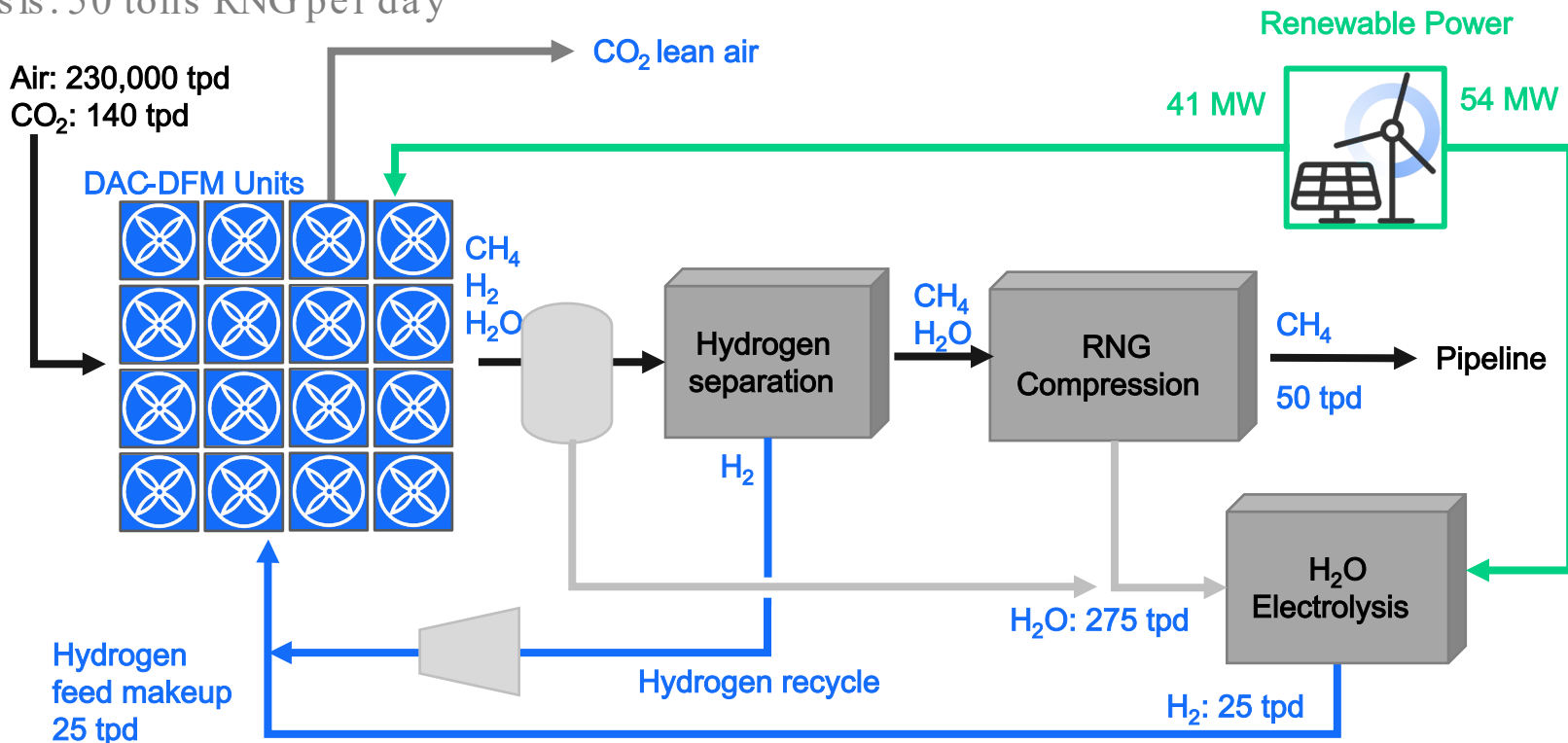


# State of Technology & Objectives of Phase IIA



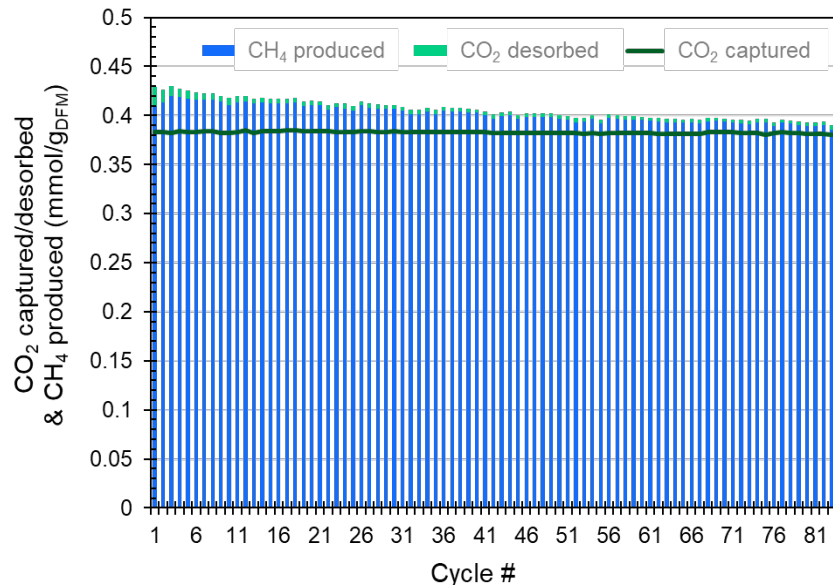
# Conceptual Process Design

Basis: 50 tons RNG per day

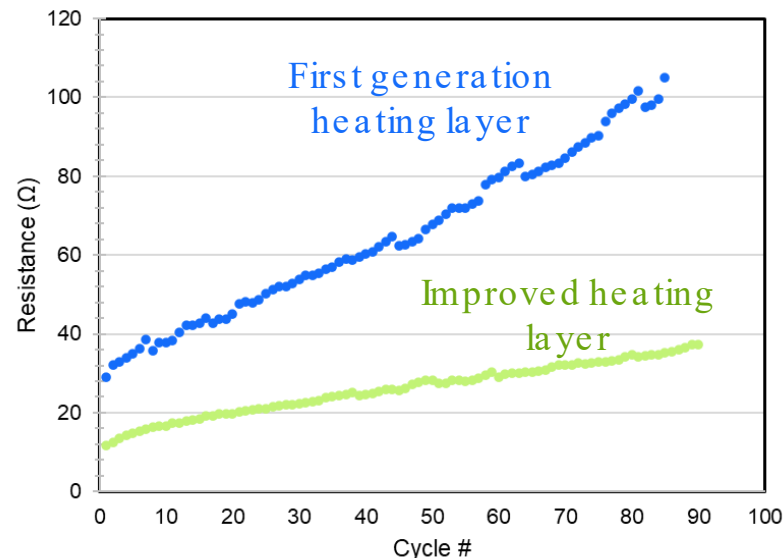


# Stable DFM Integrated with Improved Resistive Carbon

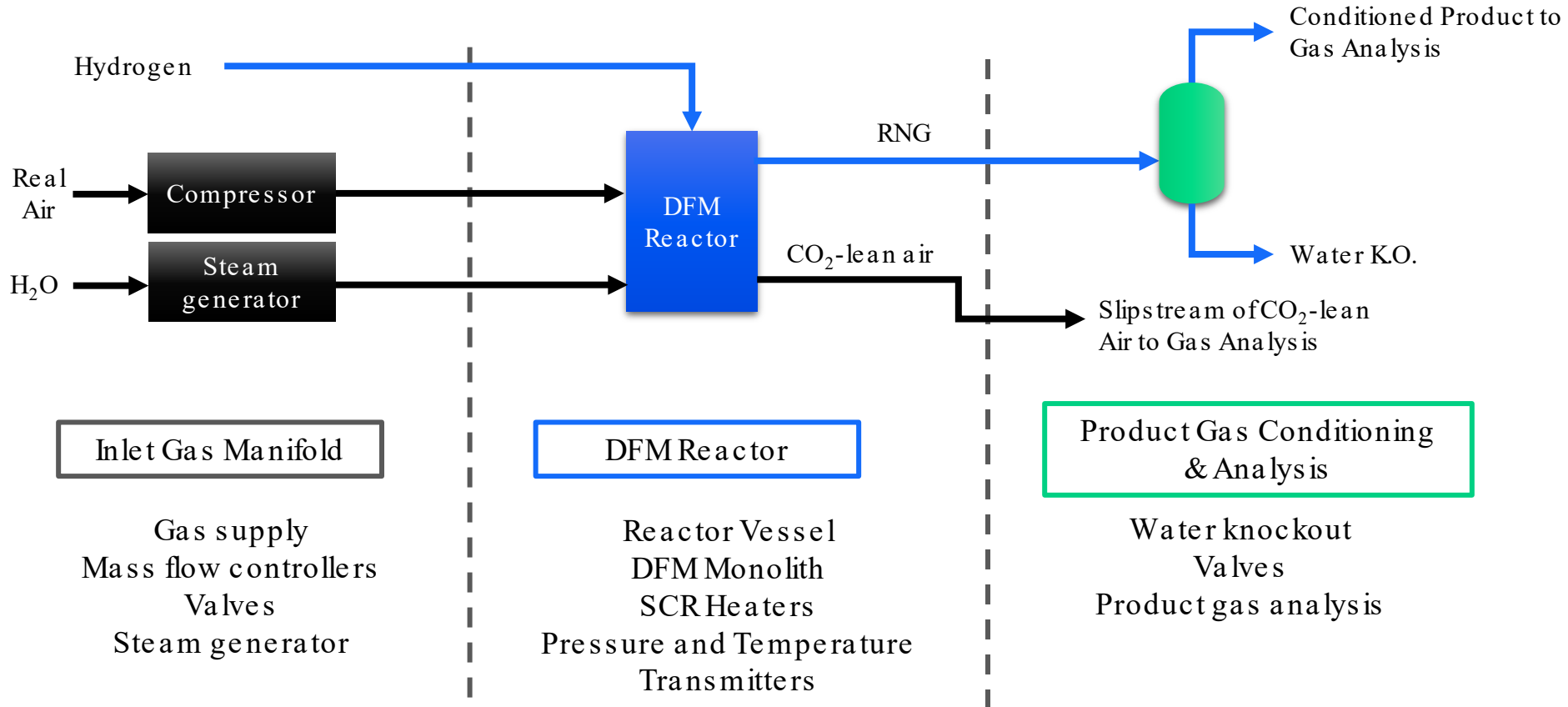
>90% conversion to CH<sub>4</sub> achieved using electrical heating platform (SMA)



Stability of heating layer resistance is improved with the new carbon synthesis method.

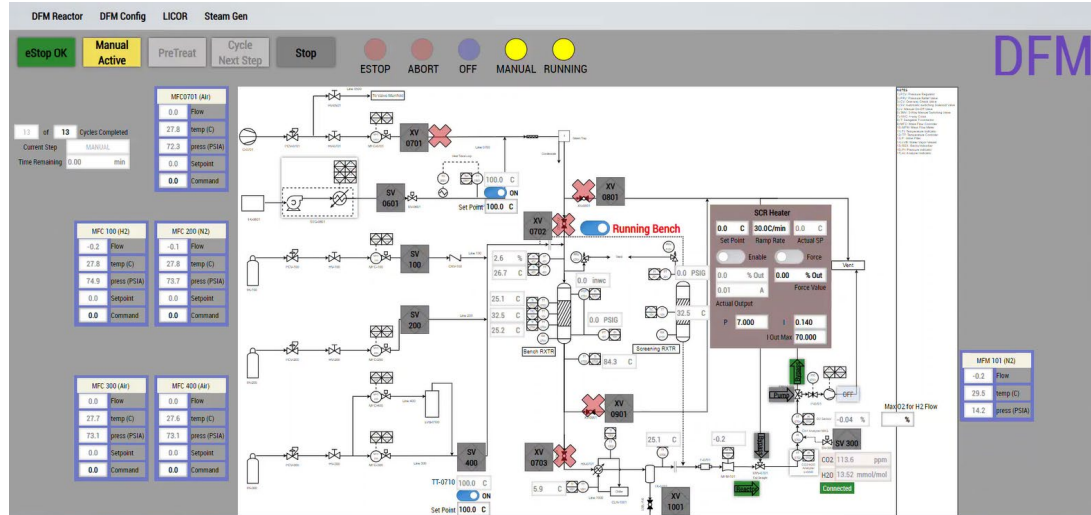
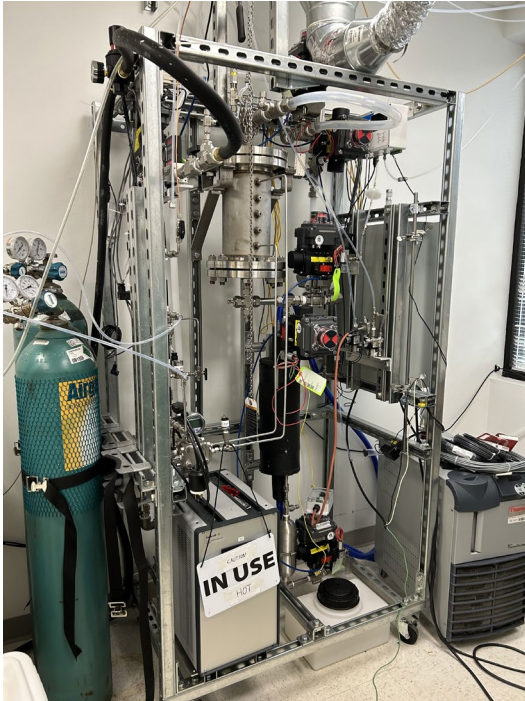


# Bench Unit Overview & Key Components



# Bench DFM Unit is Fully Commissioned

Reactor HMI displays the current system state, allows manual operation, and monitoring and intervention during automated cycling.



## Bench reactor

- Up to 300 g C<sub>Q</sub> captured & converted per day
- Includes Joule heating functionality for cyclic temperature swing
- Concept for pre-commercial scale

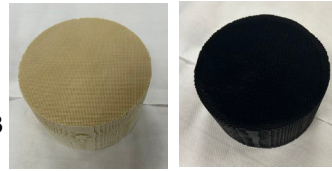
# Commissioning the Bench Unit with an Unoptimized DFM Brick

## Sample specifications:

5.25" diameter x 2.83" length

0.31 wt.% Ru & 15.15 wt.%  $\text{Na}_2\text{CO}_3$

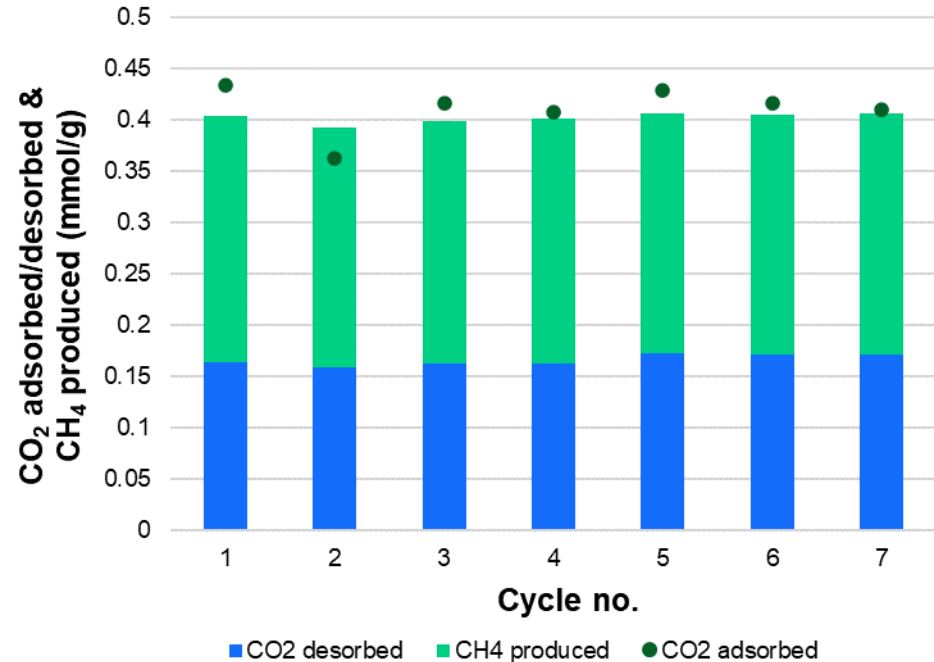
1.75 g/in<sup>3</sup> washcoat loading



Uncoated and coated monoliths

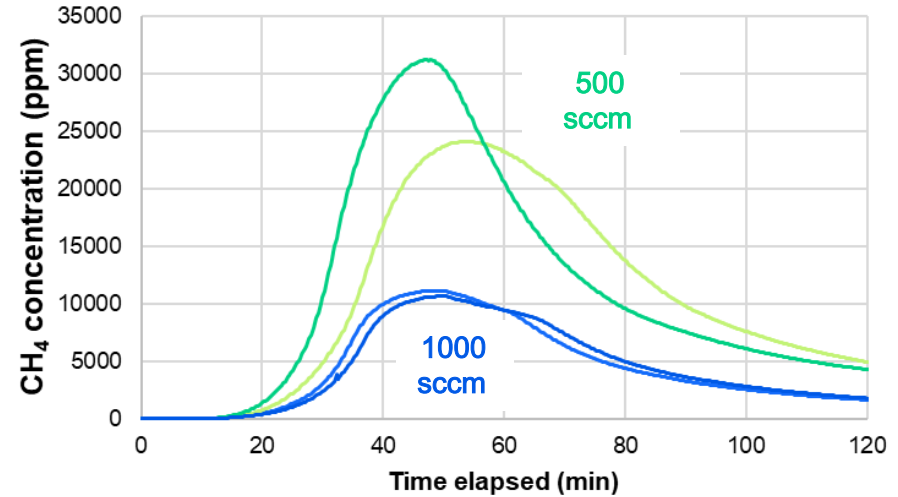
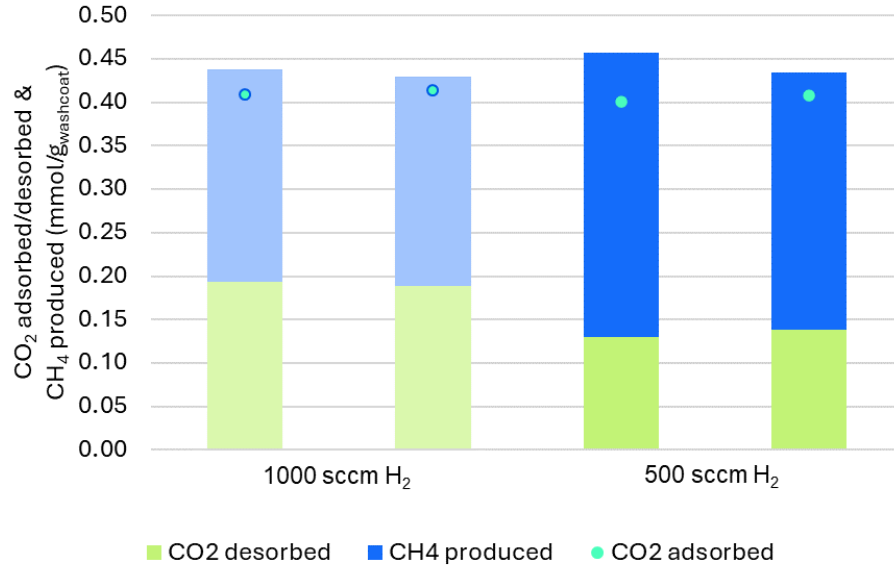
Air flow rate [WHSV]	Air	H <sub>2</sub> flow rate [WHSV]	Target heating rate & methanation temperature
100 SLPM [74 hr <sup>-1</sup> ]	Humid (~2% H <sub>2</sub> O)	1000 sccm [0.75 hr <sup>-1</sup> ]	30°C/min, 300°C

After successful commissioning, we are optimizing the DFM brick and operating conditions.

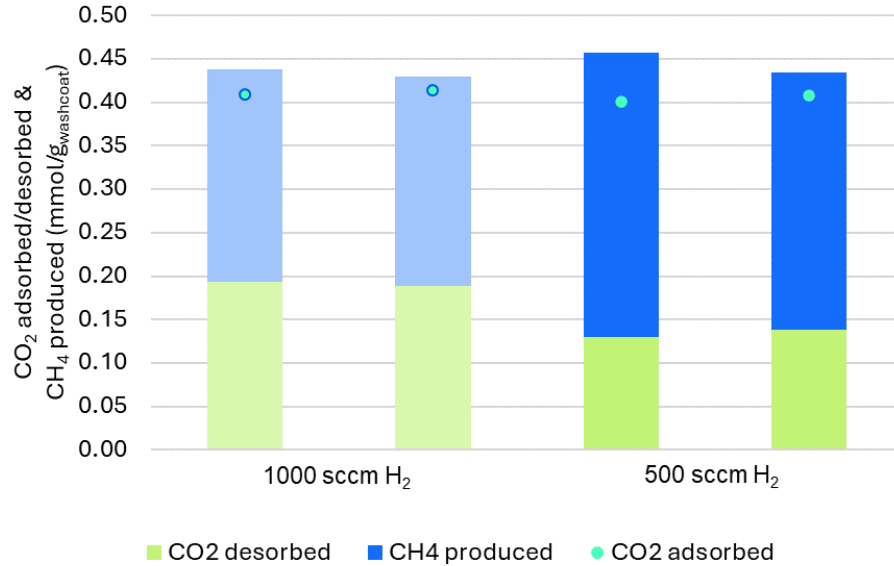




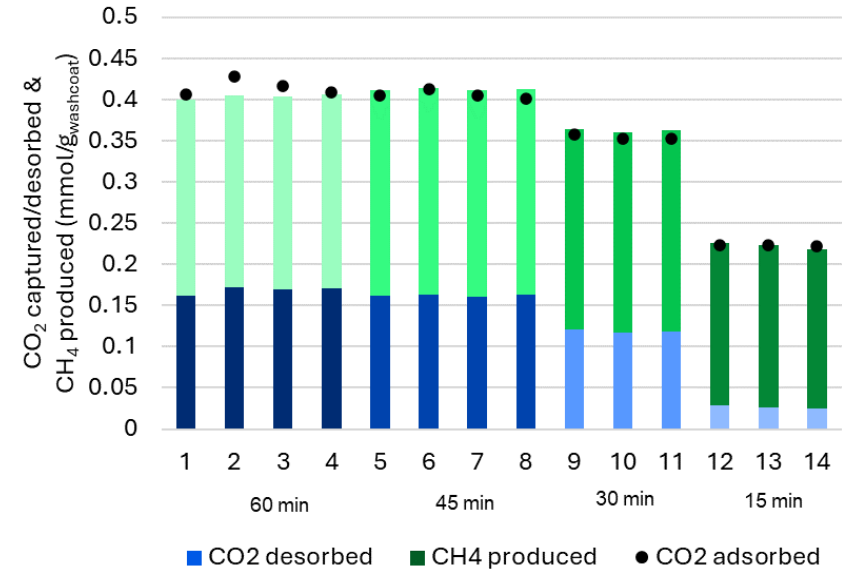
# Impact of H<sub>2</sub> Flow Rate on CO<sub>2</sub> Conversion



# Impact of H<sub>2</sub> Flow Rate on CO<sub>2</sub> Conversion



Conversion further improves by shortening the duration of capture step



# Lessons Learned

---

1. CO<sub>2</sub> can be captured and converted into methane in a single reactor.
2. Electric heating was successfully demonstrated to provide the heat for temperature swing and methanation light-off.
3. Process design developed for a commercial embodiment.
4. Robust DFM compositions successfully coated on a commercial monolith.
5. Mechanical design for long-term operation is being developed.
6. Low-cost, carbon-free hydrogen is critical for commercial viability of this technology.
7. Methane produced from the process can be designated as renewable natural gas (RNG), and incentives such as LCFS and 45Q credits will promote its early adoption.

# Budget Period 2: Planned Work

---

1. Conduct long-term bench-scale testing
  - Parametric testing to optimize operating conditions and productivity
2. Refine and validate the current process model
3. Develop a process design and technology package
4. Perform techno-economic assessment (TEA) and life cycle analysis (LCA)
5. Develop Business/Commercialization plan



**Jonathan E. Peters**

Senior Process Engineer

[jep@susteoninc.com](mailto:jep@susteoninc.com)

**Monica Abdallah, PhD**

Research Chemical Engineer

[mja@susteoninc.com](mailto:mja@susteoninc.com)

**Jian -Ping (JP) Shen, PhD**

Senior Principal Scientist

[jps@susteoninc.com](mailto:jps@susteoninc.com)

**Raghubir Gupta, PhD**

CEO/Co-Founder

[rg@susteoninc.com](mailto:rg@susteoninc.com)



[www.susteoninc.com](http://www.susteoninc.com)

Thank you