# Energy-Efficient Direct Air Capture System for High-Purity CO<sub>2</sub> Separation

**DE-FE0032128** 

Joo-Youp Lee University of Cincinnati

## Project Overview

- 1) Funding: DOE share \$1,499,999 and cost share \$393,650
- 2) Overall Project Performance Dates: 10/1/2021 12/31/2024
- 3) Project Participants: University of Cincinnati (UC), BASF, Trimeric
- 4) Overall Project Objectives: demonstrate Recipient's DAC sorbent technology to capture CO<sub>2</sub> from ambient air and separate it at high purity.

# Technology Background

- Modified amine-based sorbent technology with low desorption energy requirement and resistance to thermal and oxidative degradations
- Sorbents with high capacities, fast kinetics, small mass-transfer resistances, and low desorption energy requirement
- State-of-the-art coating technology for scale-up from powdered form to monolithic form
- High throughput of air flow with minimum pressure drop through sorbent-coated monolith in passive air contactor
- Passive air contactor without energy requirement during adsorption

# Technical Approach/Project Scope

- Task 3: Manufacture CO<sub>2</sub> sorbent (UC)
  - ✓ Manufacture CO<sub>2</sub> sorbent
  - ✓ Evaluate long-term lab-scale performance
- Task 4: Manufacture sorbent-washcoated monolith structure (UC + BASF)
  - Develop sorbent-washcoated monolith
  - Determine cell size and length
- Task 6: Evaluate performances of sorbent-washcoated monolith in air contactor system (UC)
  - ✓ Measure temperature, humidity, velocity, and CO₂ concentration
  - Evaluate performances
- Task 7: TEA and LCA (UC + Trimeric)
  - Development of TCM and LCI
  - ✓ Assessments of DAC technology

# Technical Approach/Project Scope

## **Major milestones**

Task	Milestone Title & Description	Planned Completion Date	
3	Manufacture of 10 kg of CO <sub>2</sub> sorbent	PSD+23 months	
4	Manufacture of two sorbent-washcoated monolith prototypes	PSD+24 months	
6	CO <sub>2</sub> capture efficiency, energy requirements, and overall volumetric CO <sub>2</sub> productivity	PSD+39 months	

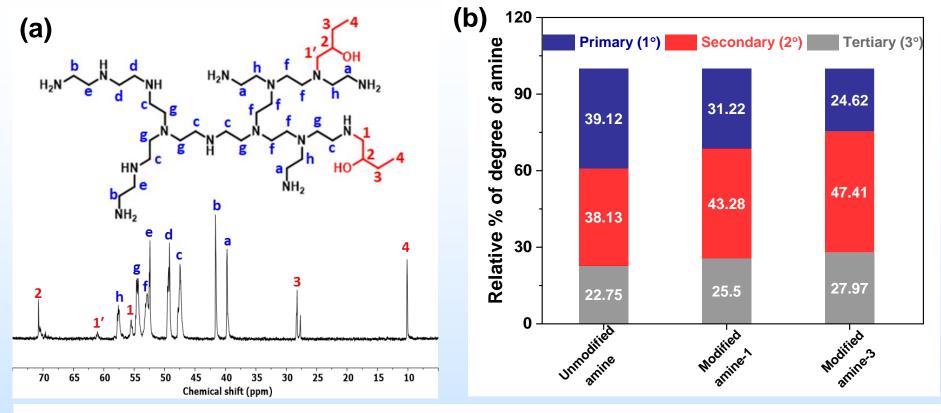
### **Major Success criteria**

- 70% average CO<sub>2</sub> capture efficiency in passive air contactor with monolith with pressure drop of <100 Pa and <75 kJ/mol CO<sub>2</sub>
- Overall volumetric productivity of ~2 (gmol CO<sub>2</sub>/(hr x V(l))

## **Project Risks and Mitigation Strategies**

- Low performance of sorbent-coated monolith: BASF will attempt many different coating formulations
- Low DAC system performance: parametric testing will be carried out to investigate the effects of parameters (i.e., operating conditions, materials, etc.)<sup>5</sup>

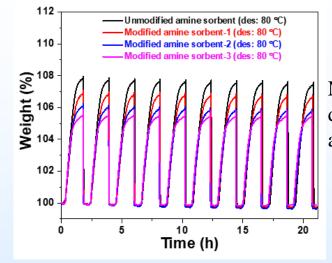
### **Modified PEI Sorbent**



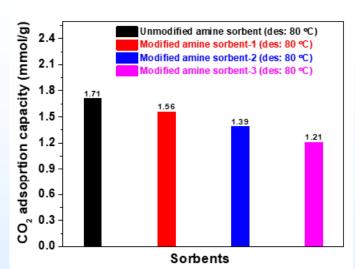
<sup>13</sup>C NMR spectrum for (a) modified amine and (b) distribution of amine states

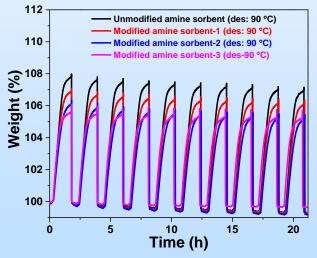
- 1) Increased resistance to thermal and oxidative degradations by lowering the basicity and increasing steric hindrance.
- 2) Reduced bonding strengths leading to low heat of desorption with decreased capacity
- 3) UC filed U.S. patent/PCT for 10 modifications.

## Performance Evaluations in TGA: 10 cycles

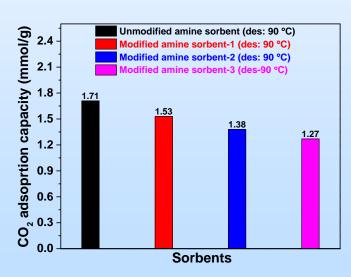


N<sub>2</sub> desorption at **80** °C





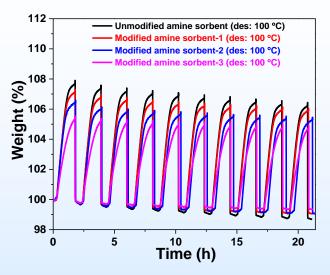
N<sub>2</sub> desorption at 90 °C



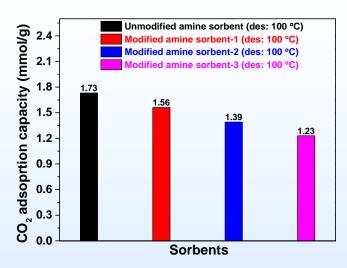
Capacity decreases with an increase in level of modification.

Adsorption: 400 ppm CO<sub>2</sub> in dry air at 30 °C and Desorption: N<sub>2</sub> at (a) 80 and (b) 90 °C

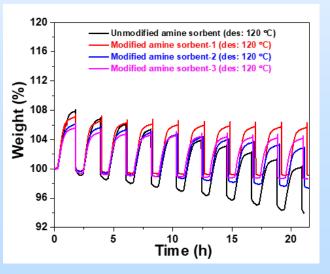
## Performance Evaluations in TGA: 10 cycles



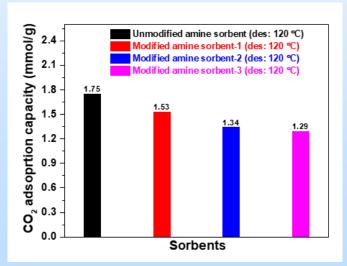
N<sub>2</sub> desorption at 100 °C



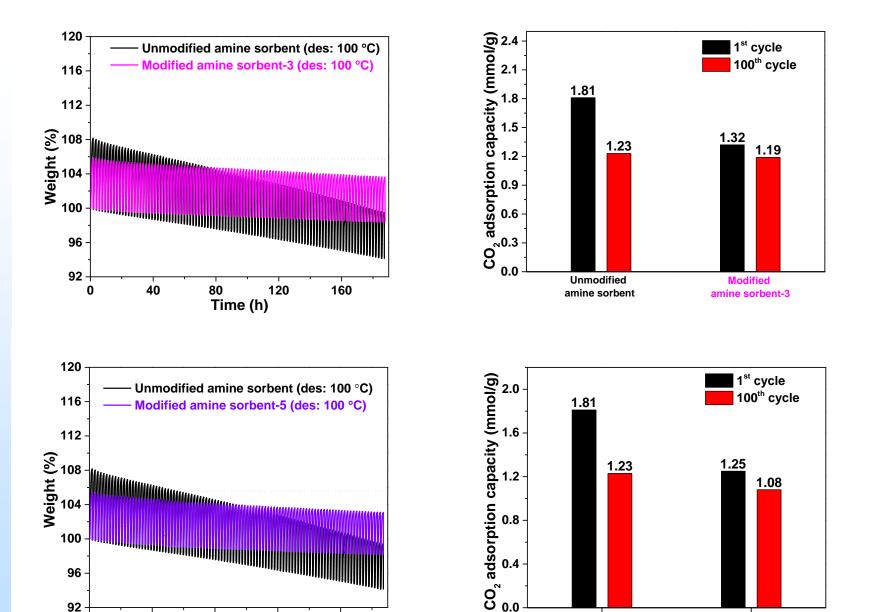
Capacity decreases with an increase in level of modification.



N<sub>2</sub> desorption at 120 °C



Adsorption: 400 ppm CO<sub>2</sub> in dry air at 30 °C and Desorption: N<sub>2</sub> at (a) 100 and (b) 120 °C



Adsorption: 400 ppm CO<sub>2</sub> in Dry Air at 30 °C and Desorption: N<sub>2</sub> at 100 °C

Unmodified

amine sorbent

Modified

amine sorbent-5

92

0

40

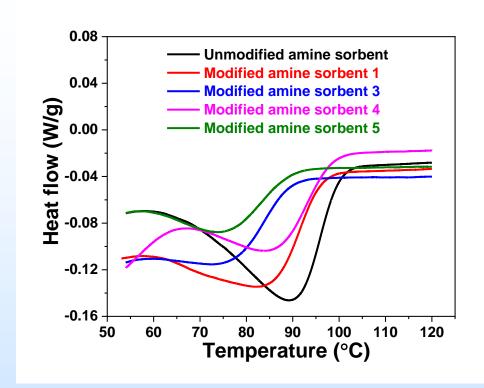
80

Time (h)

120

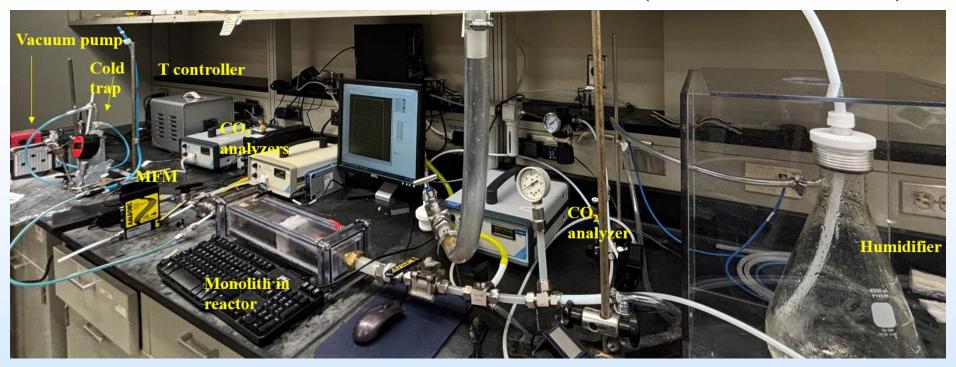
160

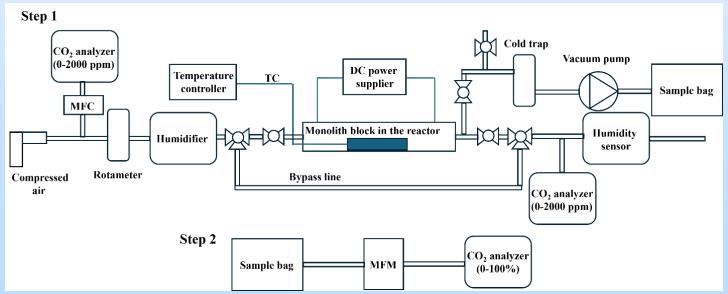
## **Heat of Desorption for Powdered Sorbents**



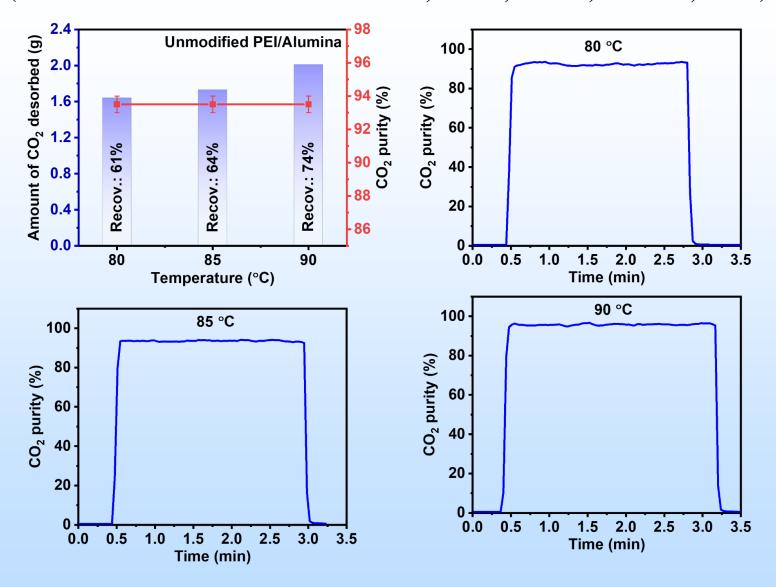
Sorbents	Heat of desorption (kJ/gmol CO <sub>2</sub> )	Energy savings relative to unmodified amine sorbent (%)*	
Unmodified amine sorbent	73.74	0	
Modified amine sorbent 1	56.87	-22.9	
Modified amine sorbent 3	34.77	-52.8	
Modified amine sorbent 4	35.83	-51.4	
Modified amine sorbent 5	31.82	-56.8	

## Performance Evaluations of Small Block (3 cm×3 cm×15 cm)



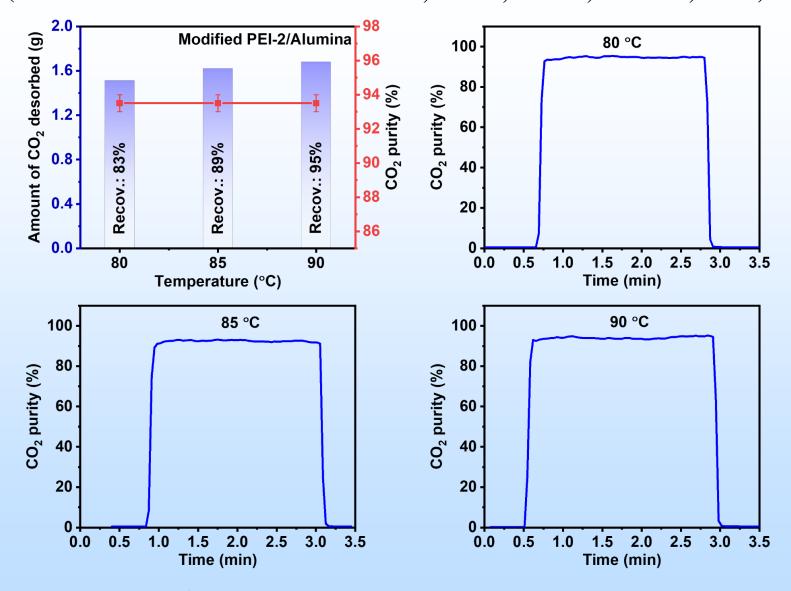


## CO<sub>2</sub> Desorption by JH between 80-90 °C from Unmodified PEI/Alumina Block (Block was saturated under 175 LPM air, SV: 77,778 hr<sup>-1</sup>, 50% RH, 20 °C)



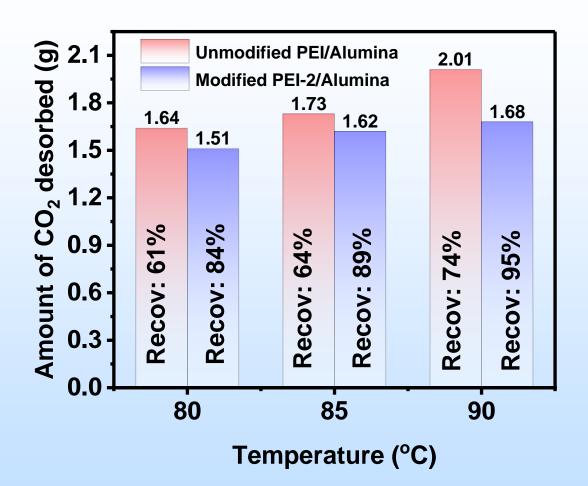
- Vacuum pressure of -28" Hg (full vacuum: -29.9" Hg)
- Initial air present inside the reactor: ~6.3%

CO<sub>2</sub> desorption by JH between 80-90 °C form Modified PEI-2/Alumina Block (Block was saturated under 175 LPM air, SV: 77,778 hr<sup>-1</sup>, 50% RH, 20 °C)



- Vacuum pressure of -28" Hg (full vacuum: -29.9" Hg)
- Initial air present inside the reactor: ~6.3%

# Working Capacity and CO<sub>2</sub> Recovery by Joule Heating between 80-90 °C



- Vacuum pressure of -28" Hg (full vacuum: -29.9" Hg)
- Initial air present inside the reactor: ~6.3%
- RH: ~50%; CO<sub>2</sub> purity: ~93-94%; SV: 77,778 hr<sup>-1</sup>

## CO<sub>2</sub> desorption between 80-90 °C

#### **Modified PEI-2/Alumina block**

Des	Adsorption capacity			Working	Residual CO <sub>2</sub>	Total CO <sub>2</sub>	G.0	CO <sub>2</sub>
T (°C)	mmol CO <sub>2</sub> / g sorbent	mmol CO <sub>2</sub> / monolith	Working capacity by JH (mmol)	capacity (mmol CO <sub>2</sub> / g sorbent)	desorbed by N <sub>2</sub> (mmol)	desorbed (mmol)	CO <sub>2</sub> recovery from JH (%)	purity (%)
80	1.19	31.96	<b>34.42</b> (1.51 g)	1.28	6.76	41.18	84	94
85	1.21	32.45	<b>36.76</b> (1.62 g)	1.37	4.52	41.28	89	93
90	1.17	31.35	<b>39.11</b> (1.68 g)	1.42	1.95	41.06	95	94

#### PEI/Alumina block

Des	Adsorption capacity			Working	Residual CO <sub>2</sub>	Total CO <sub>2</sub>		CO <sub>2</sub>
T (°C)	mmol CO <sub>2</sub> / g sorbent	mmol CO <sub>2</sub> / monolith	Working capacity by JH (mmol)	capacity (mmol CO <sub>2</sub> / g sorbent)	desorbed by N <sub>2</sub> (mmol)	desorbed (mmol)	CO <sub>2</sub> recovery from JH (%)	purity (%)
80	1.68	46.66	<b>37.21</b> (1.64 g)	1.34	23.76	60.97	61	93
85	1.67	46.31	<b>39.24</b> (1.73 g)	1.41	21.78	61.02	64	93
90	1.66	46.07	<b>45.26</b> (2.01 g)	1.69	15.09	60.85	74	94

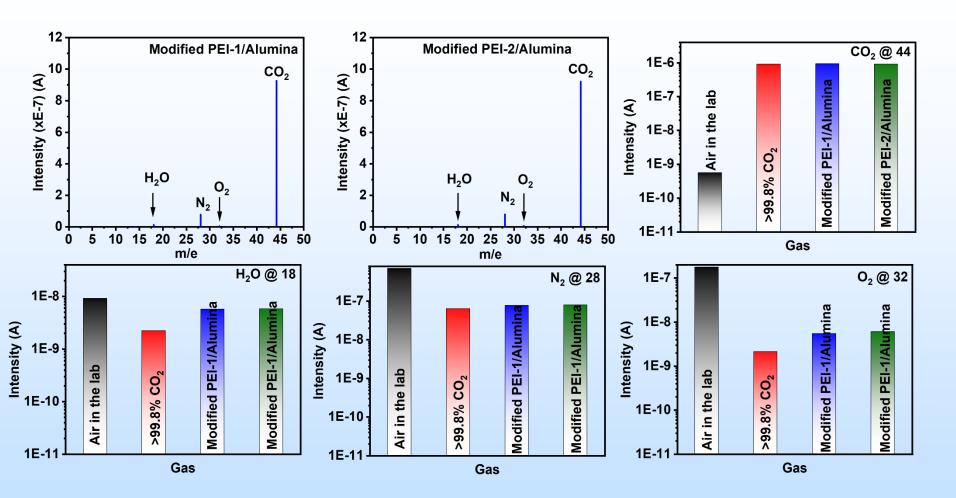
• PEI block has ~30% higher saturation capacity than modified PEI-2 block.

• Adsorption T: 20 °C

• **RH:** 50%

• **SV:** 77,778 hr<sup>-1</sup>

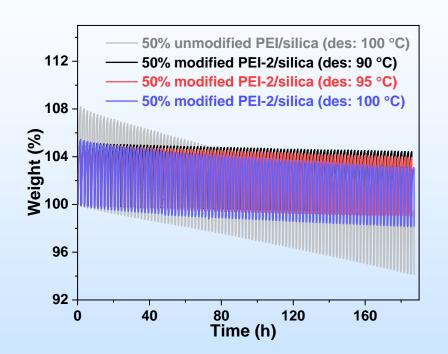
## MS Analysis of Gas Desorbed by Joule Heating at 90 °C

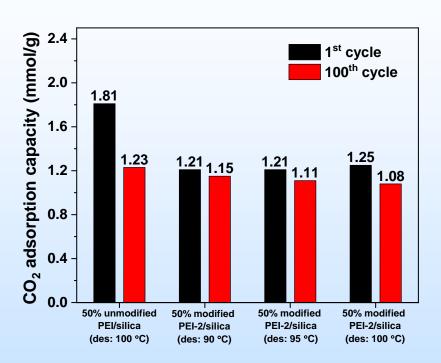


#### **Tech-grade CO<sub>2</sub> gas compositions:**

- CO<sub>2</sub>: >99.8%
- $O_2$ : < 20 ppm
- **Moisture:** < 20 ppm

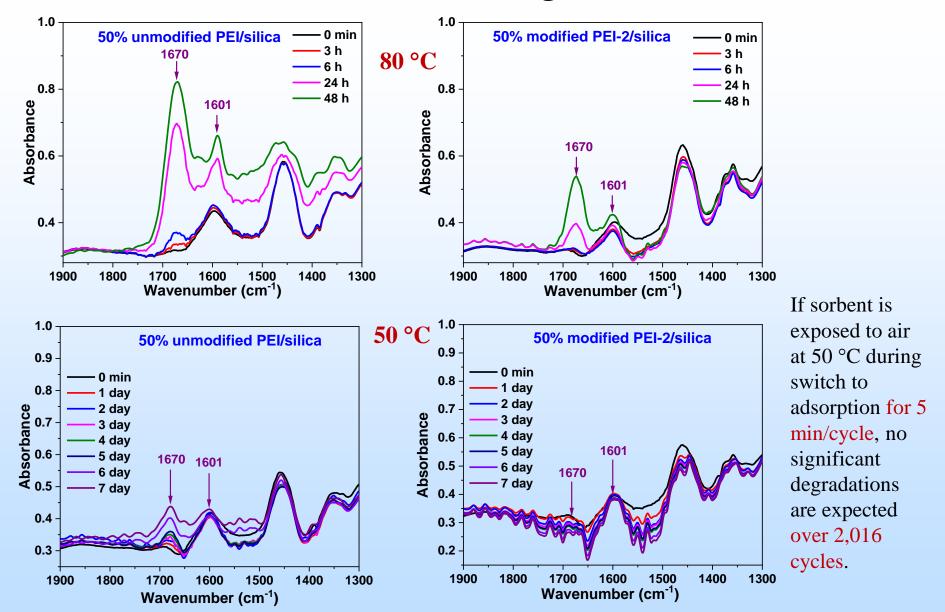
# Thermal Stability of Modified PEI-2 Sorbent between 90-100 °C over 100 Cycles





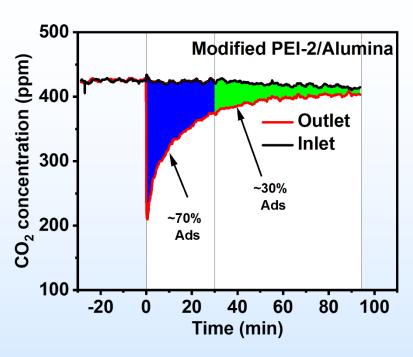
- Thermal degradation at ≤90 °C seems to be negligible for modified PEI-2 sorbent.
- Adsorption under 400 ppm CO<sub>2</sub> in dry air at 30 °C
- Desorption under N<sub>2</sub>

## Thermal and Oxidative Degradations in Air



- 1,670 cm<sup>-1</sup>: either imine (C=N) or carbonyl (C=O) group.
- 1,601 cm<sup>-1</sup>: NH<sub>2</sub> deformation of primary amines.

#### **Adsorption and Desorption Profiles of Small Block Testing**

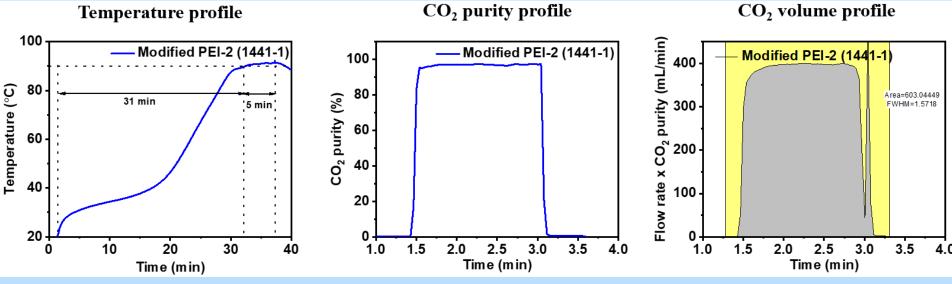


#### **Adsorption conditions:**

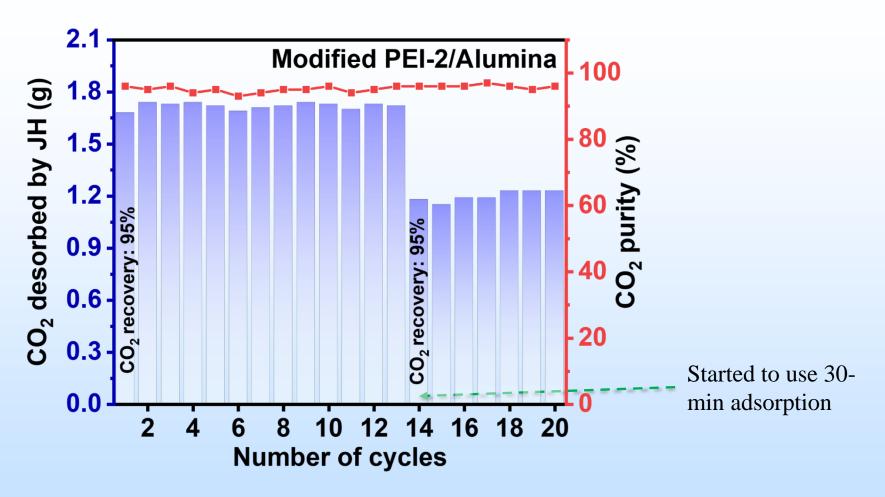
- ✓ Flowrate:175 LPM
- **✓ Relative humidity (RH): ~50%**
- ✓ Temperature: 20 °C

#### **Desorption conditions:**

- ✓ Vacuum pressure: -28" Hg
- ✓ Initial air present inside reactor after pulling a vacuum: ~6%
- ✓ Temperature: 90 °C
- ✓ CO₂ recovery and purity: ~95%

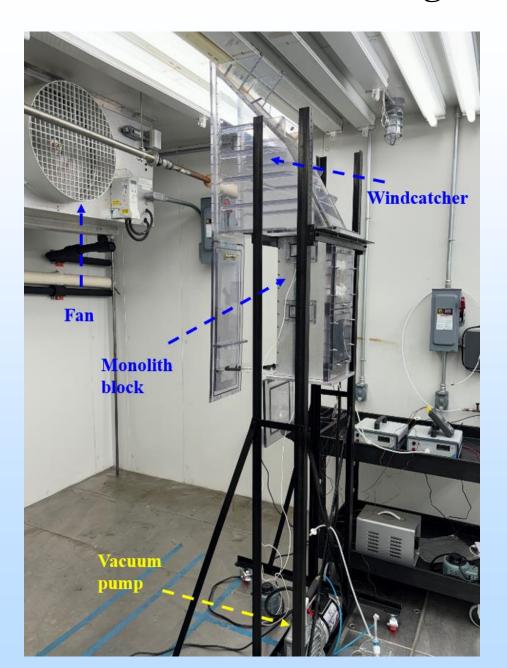


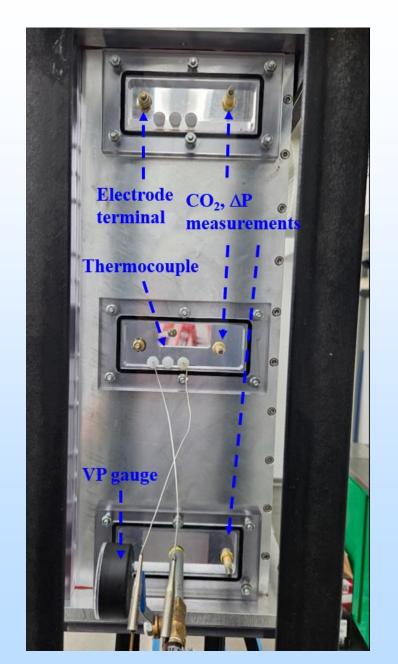
## Working Capacity of Modified PEI-2 on Small Monolith



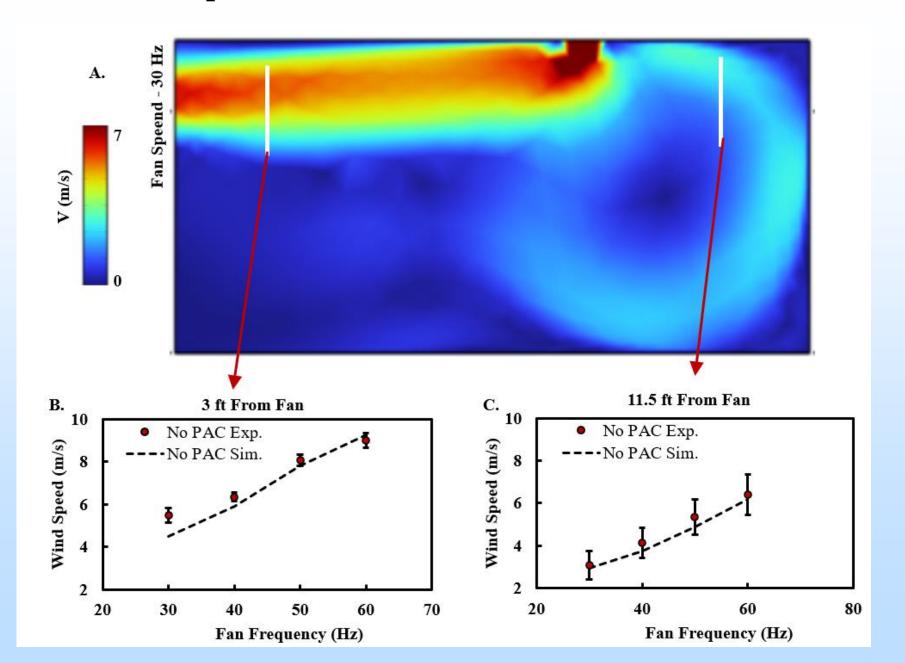
- ✓ **Goal: 100 cycles** for PEI and Modified PEI-2 blocks
- ✓ Flowrate:175 LPM air; SV: 77,778 hr<sup>-1</sup>; RH: ~50%; Adsorp. T: 20 °C; Desorp. T: 90 °C; Switch from desorption to adsorption at 50 °C
- ✓ **Adsorption time for Cycle 1-13:** 90 min for >~95% adsorption capacity
- ✓ **Adsorption time for Cycle 14-20:** 30 min for ~70% adsorption capacity

## **Evaluation of Large Monolith Block in PAC**

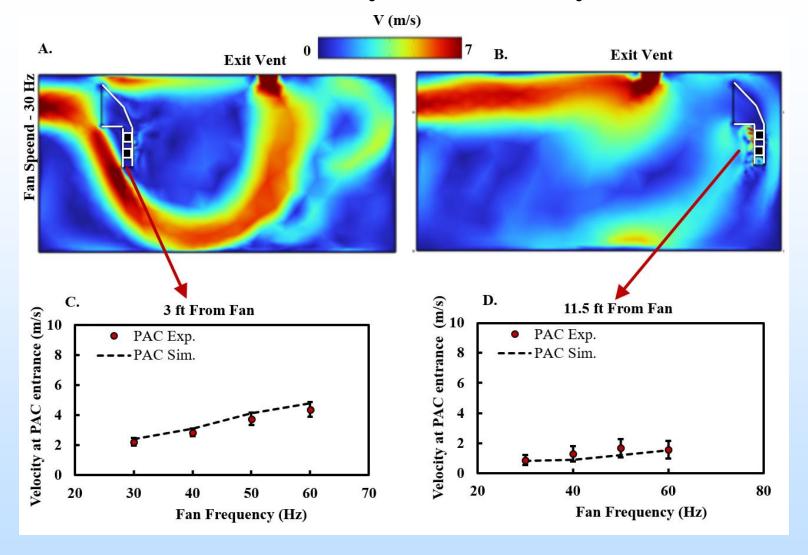




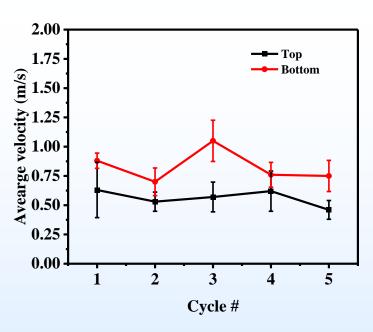
## Wind Speeds without Passive Air Contactor (PAC)

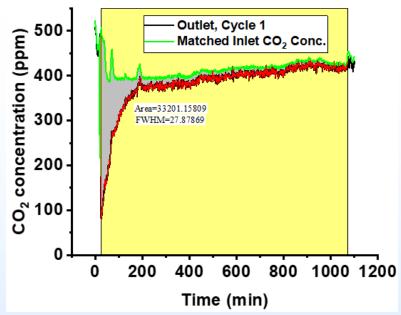


## **Linear Velocity inside DAC System**



- ✓ T: 0-40 C; RH: 20-80%
- ✓ Wind speeds: ~3-8 m/s
- ✓ LVs at monolith outlet: ~0.6-4 m/s





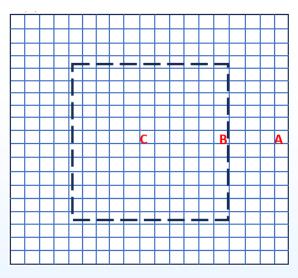


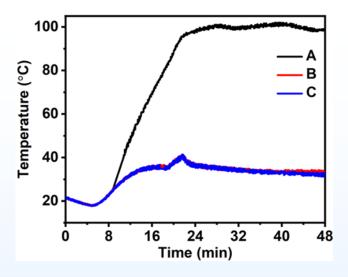
✓ Single block with 64 cpsi

✓ RH: ~50%; Wind speed: ~7 m/s; Initial vacuum pressure: -28" Hg

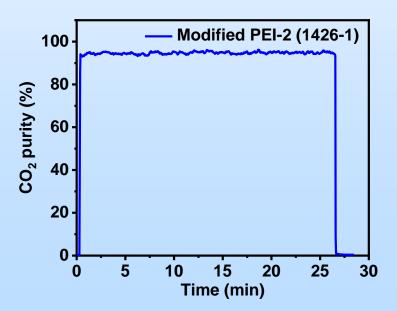
✓ CO<sub>2</sub> purity:  $94 \pm 1\%$ 

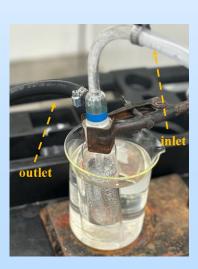
Cycle #	Avg. LV (m/s)	SV (h-1)	Pressure drop (Pa)	Adsorption T (°C)	Desorption capacity (mmol CO <sub>2</sub> /g)
1	0.88	28,746	28	21	0.69
2	0.70	22,862	28	21	0.55
3	1.05	34,311	27	21	0.55
4	0.76	24,836	29	30	0.62
5	0.75	24,498	27	30	0.51





- ✓ Poor heat-transfer property of monolith is an issue for Joule heating.
- ✓ Total 2,025 cells (=  $45 \times 45$ ); # of cells inside dashed line:  $529 = 23 \times 23$
- ✓ Maximum recoverable  $CO_2$  from outside dashed line = <74%
- ✓ Plan for desorption: vacuum convection oven







## **Lessons Learned**

- 1. Modified PEI sorbent technology has potential for separating high-purity CO<sub>2</sub> at high working capacity with high CO<sub>2</sub> recovery from desorption at 85-90 °C.
- 2. Modified PEI sorbent technology has potential for switching from desorption to adsorption at ~50 °C without significant degradations.
- 3. Coating is critical when it needs to be used for structured system.
- 4. Poor heat-transfer property of cordierite is a hurdle for separation when non-steam direct injection-based desorption is used.

## Plans for future testing/development

- Task 6: Performance evaluations
  - ✓ 100 cycles of small blocks with PEI and modified PEI-2 (weight, purity, degradation (MS, IR))
  - ✓ Large blocks

Parameters	Values		
Block samples	Two modified PEI vs. PEI		
T (°C)	0, 15, 30		
RH (%)	20, 50, 80		
Wind speed (m/s)	~3, 5, 7		

- Task 7: TEA and LCA (UC + Trimeric)
  - Development of TCM and LCI
  - ✓ Assessment of DAC technology
- Future Plan
  - ✓ Hybrid monolith
  - ✓ Different structured DAC system with improved heat transfer properties
  - Partnerships for commercialization

## **Summary Slide**

- Advanced sorbent technology from powdered form to structured form showing high CO<sub>2</sub> recovery and purity at 80-90 °C with resistance to degradations
- Successful development of sorbent-coated monolith for scalable DAC system
- Systematic and rational PAC design using CFD confirmed with experimental data
- Need for development of structured system with improved heattransfer properties for scaled DAC system
- Open to partnerships for different structured system, scale-up, and commercialization