

# Ammonium Looping with Membrane Absorber and Distributed Stripper for Enhanced Algae Growth

**DE-FE0031921**

**Bradley Irvin, Len Goodpaster, Otto Hoffman, Julia Parker,  
Robert Pace, Heather Nikolic, Mark Crocker and Kunlei Liu**

*Institute for Decarbonization and Energy Advancement  
University of Kentucky  
Lexington, KY*

*<http://uknow.uky.edu/research/unique-public-private-research-consortium-established-caer-co2-capture-pioneers>*

# Project Overview

## Overall Project Performance Dates

Task	Status	Deadline
Budget Period 1	Complete	
Budget Period 2	In progress	12-30-2024
Tasks 8 and 9 Integrated process assembly	Complete	
Task 10 Parametric Campaign	Complete	
Task 11 Long-term Campaign	Complete	
Task 12 TEA	In progress	12-30-2024

## Funding (DOE and Cost Share)

Budget Details (June 2023)	Federal Share	Cost Share (Cooperative Agreements)
Total Project (Award Value)	\$2,999,564	\$751,764
Total Budget Period 2 (planned)	\$1,800,234	\$421,211
Monthly Expenditures (planned)	\$140,397	\$42,776
Total Project (cumulative)	\$1,473,397	\$538,086
Total BP 2 (cumulative)	\$613,661	\$217,544

# Project Participants

## University of Kentucky

Project management

Project execution and communication

Risk identification and mitigation

Process integration and data analysis and reporting

Development and operation of proposed technology and derived facility

## Vanderbilt University (Vanderbilt)

Optimization of the membrane capital and operation cost for scale-up

## Colorado State University (CSU)

Perform the TEA/LCA and provide H&MB tables and equipment sizing for the algae production process

## Trimeric Corporation

Conduct the TEA and LCA analyses

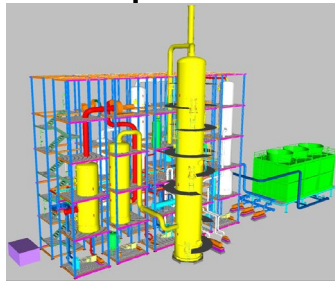
# Technology Background

## Capture then Utilization

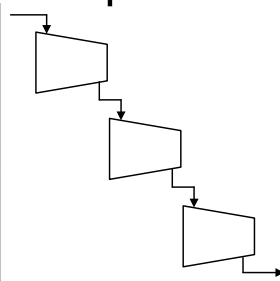
Sources



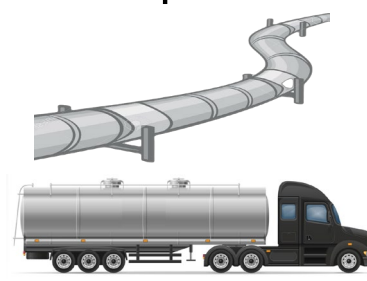
Capture



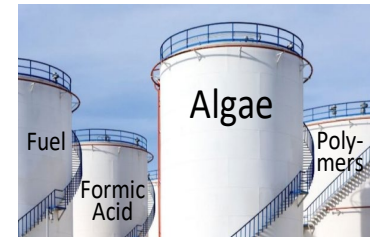
Compression



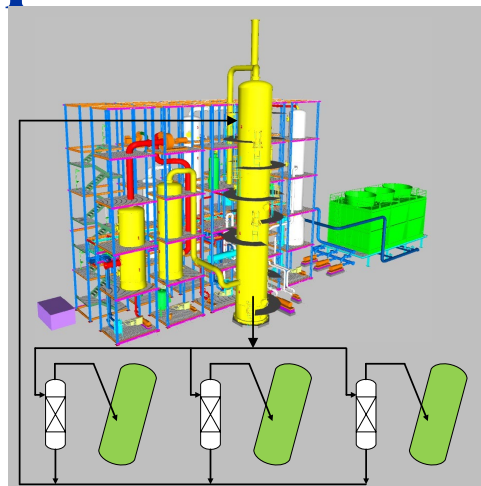
Transportation



Utilization



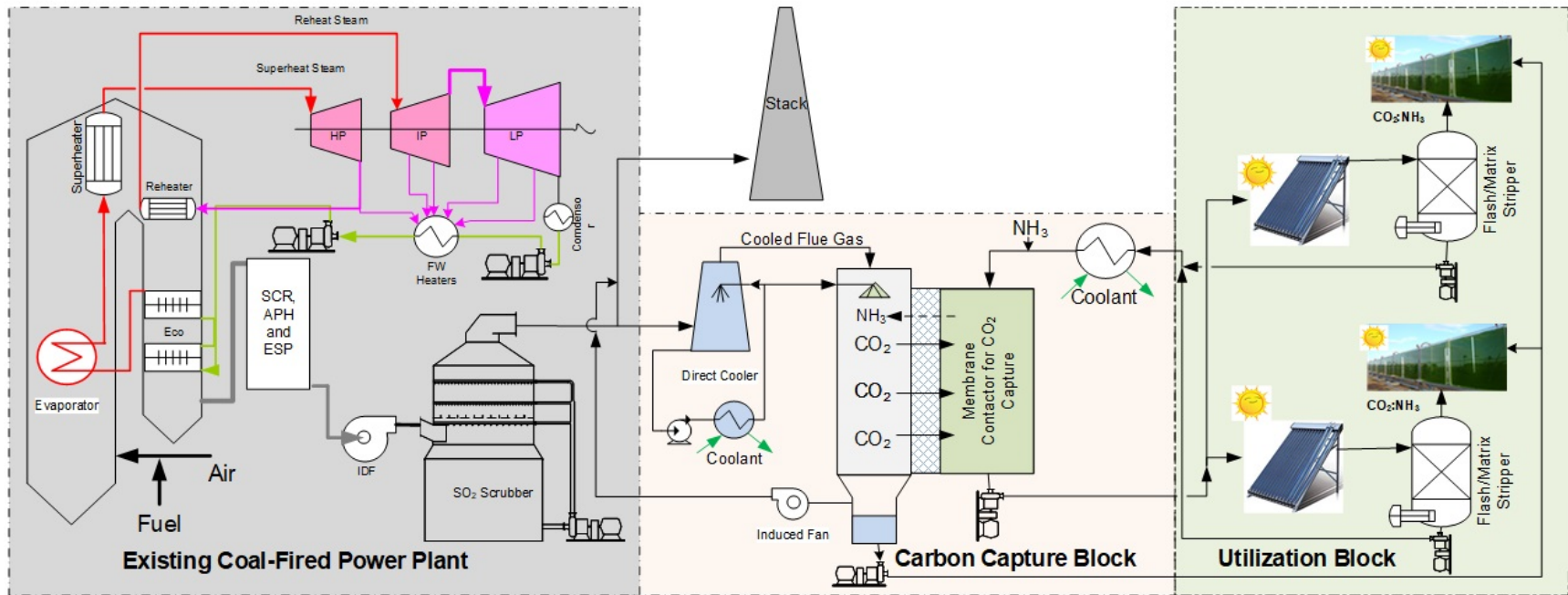
## OR Capture and Utilization



# Process Overview

Unique, integrated CO<sub>2</sub> capture and utilization technology that:

- Reduces the cost of CO<sub>2</sub> capture
- Boosts algae production



Direct to utilization CO<sub>2</sub> capture

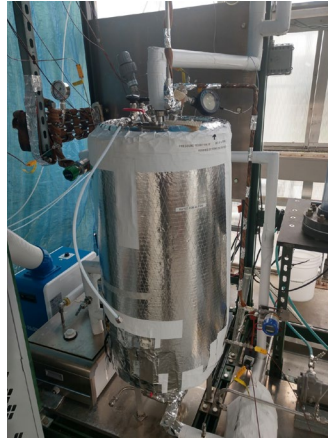


# Pilot-scale Integrated Process

LF Boiler



Stripping Loop



Solar Water Heater



Completed Assembly

# Gas sampling for product CO<sub>2</sub>/NH<sub>3</sub> ratio

Alkalinity:  
0.5 mol/kg

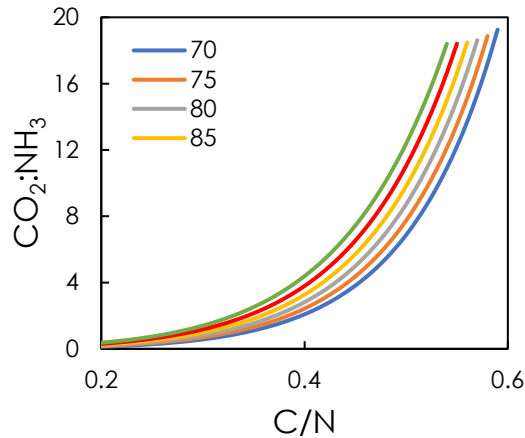
Run	Hot Oil Setpoint (° C)	Stripper bottom temperature (° C)	Stripper top temperature (° C)	Rich feed temperature (° C)	Lean loading (mol C/mol N)	CO <sub>2</sub> :NH <sub>3</sub> (exp mol/mol)
1	140	93	90	89	0.424	2.53
2	0	31	63	90	0.536	13.41
3	0	84	78	79	0.512	11.47
4	90	79	78	98	0.622	21.75

Alkalinity:  
1.8 mol/kg

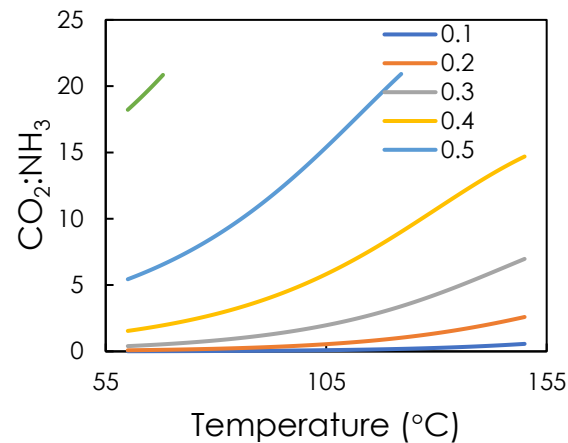
5	0	36	45	68	0.548	13.8
6	0	51	68	91	0.491	4.78
7	0	52	98	92	0.454	4.21
8	0	32	64	91	0.511	7.09
9	0	54	64	92	0.504	5.84
10	100	85	81	95	0.514	1.8

# Determining Nutrient Ratio ( $\text{CO}_2/\text{NH}_4$ )

Aspen simulation  Guideline for operation conditions optimization

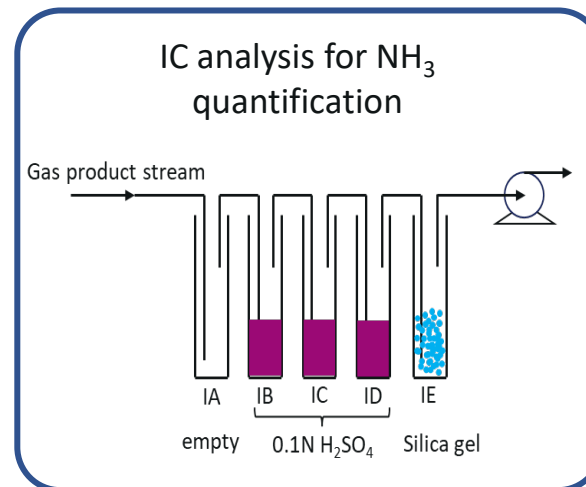
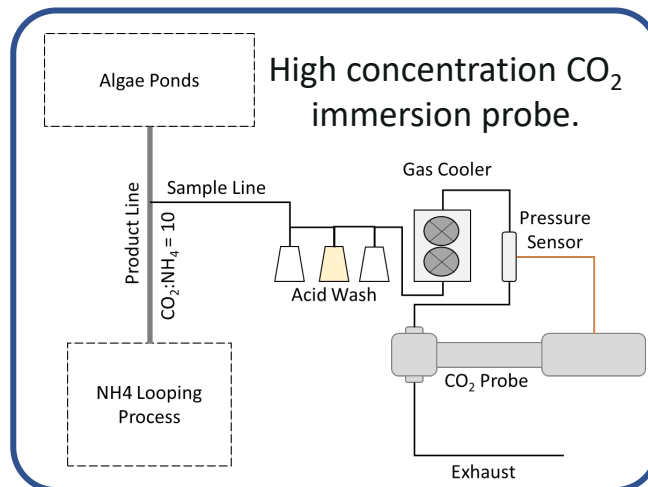


Correlation of pH measurements with  $\text{NH}_4$  concentration



Simple and fast  $\text{NH}_4$  concentration estimation

Gas phase sampling

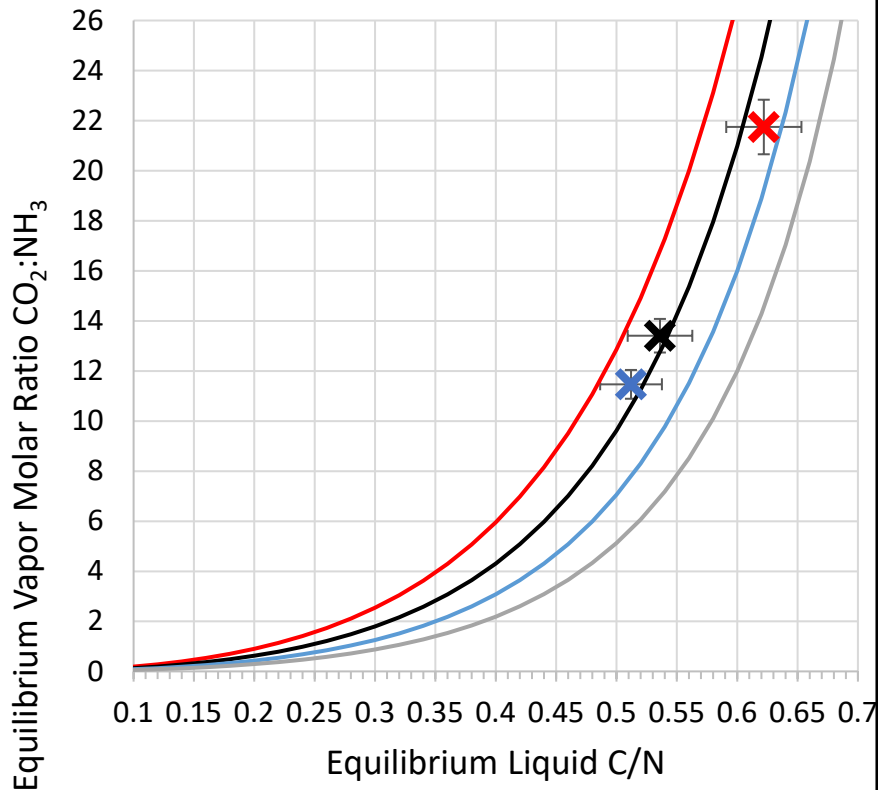




# Product Ratio: Experimental vs. Model

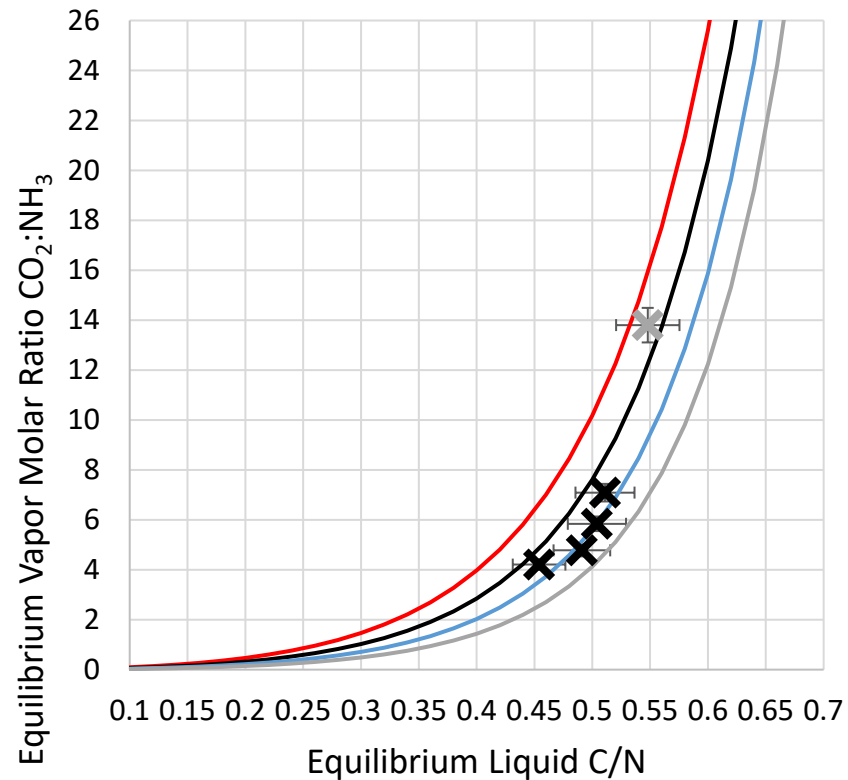
0.5  
Mol/kg

— T = 100°C    — T = 90°C    — T = 80°C    — T = 70°C  
 ✘ Run 2 90°C    ✘ Run 3 80°C    ✘ 2

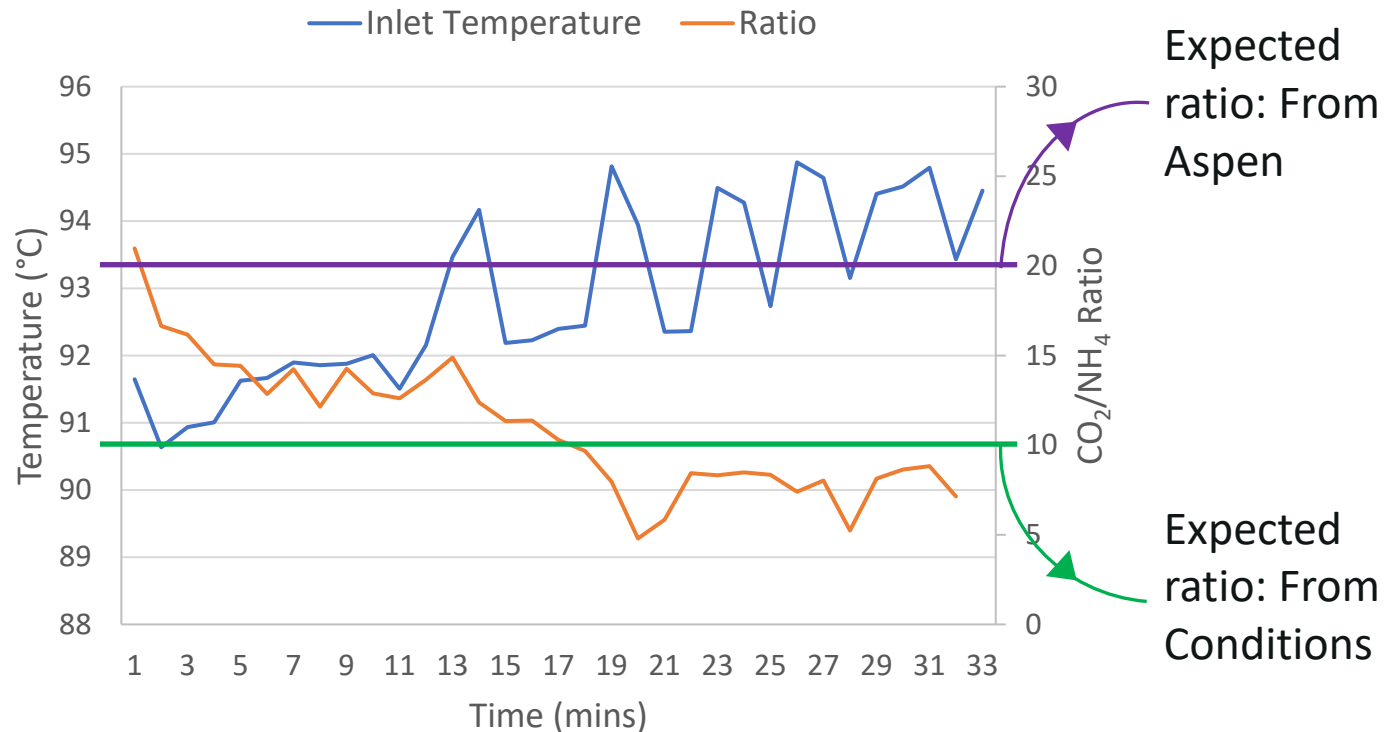


2.0  
Mol/kg

— T = 100°C    — T = 90°C    — T = 80°C  
 — T = 70°C    ✘ Run 4 90°C    ✘ Run 5 90°C  
 ✘ Run 6 90°C    ✘ Run 7 90°C    ✘ Run 8 70°C



# Product Ratio: Determined from High concentration CO<sub>2</sub> immersion probe.



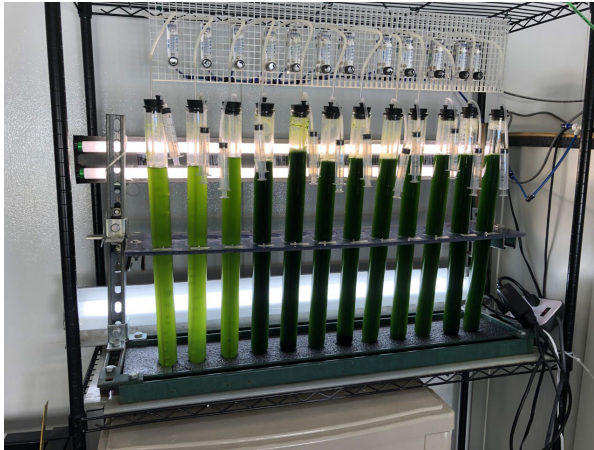
- Expected ratio is about 20, per ASPEN model.
- Ratio is stable around 10, as desired.
- Ratio decreases with increasing solvent inlet temperature. This is controllable by adjusting the cold rich flow to the stripper.

# Summary: Capture

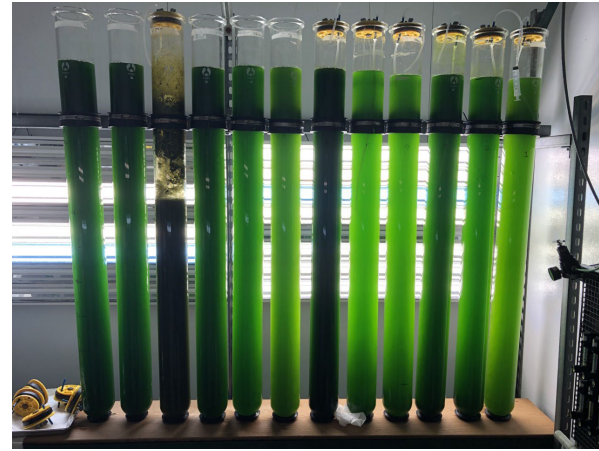
- The ammonia looping process can function and produce desired product ratios at varying solvent alkalinities, even as low as 0.5 mol/kg.
- Product ratios of 10 or more are easily achievable with higher rich carbon loadings. Preferable greater than 0.45 mol/kg.
- Can capture CO<sub>2</sub> at any inlet concentration. However, if the desired product ratio of CO<sub>2</sub>:NH<sub>4</sub> is 10 or more, then the minimum inlet CO<sub>2</sub> concentration needs to be 2% or more.

# Algae Culturing

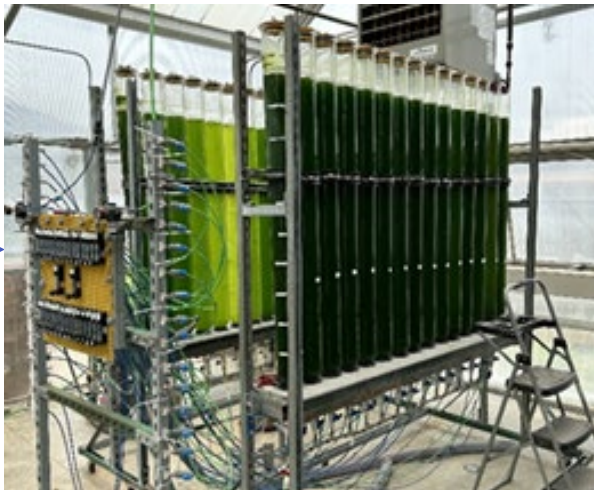
Start



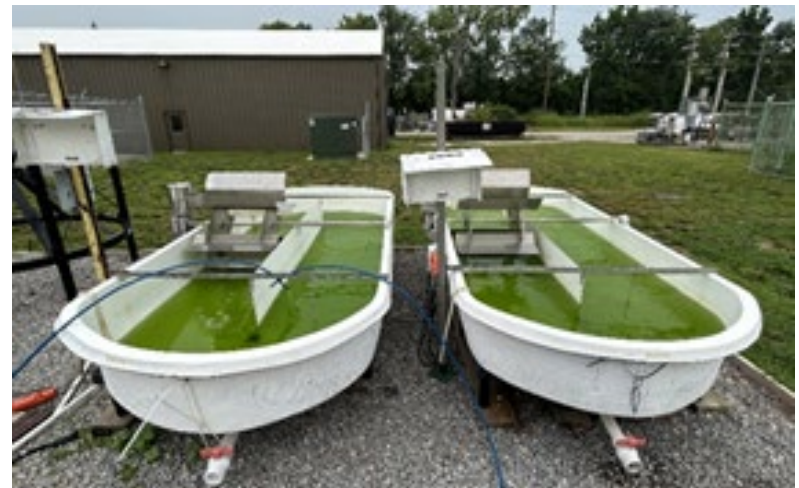
12 x 800 mL tubes



11 x 10 L tubes



30 x 10 L tubes

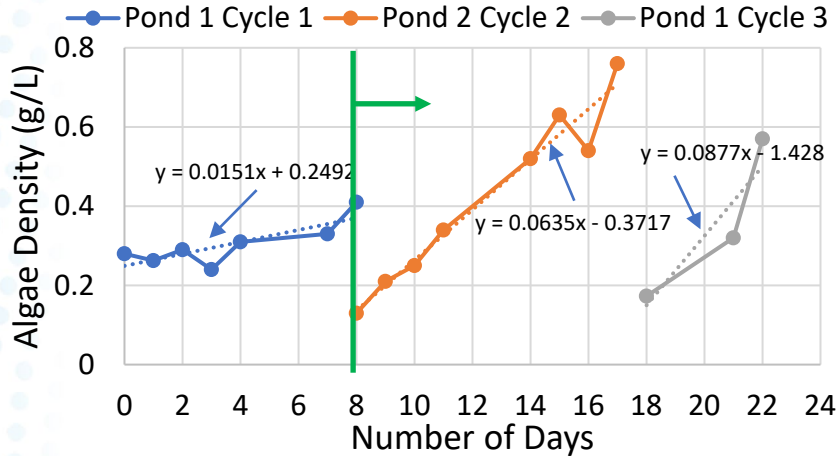


2 x 900 L ORPs

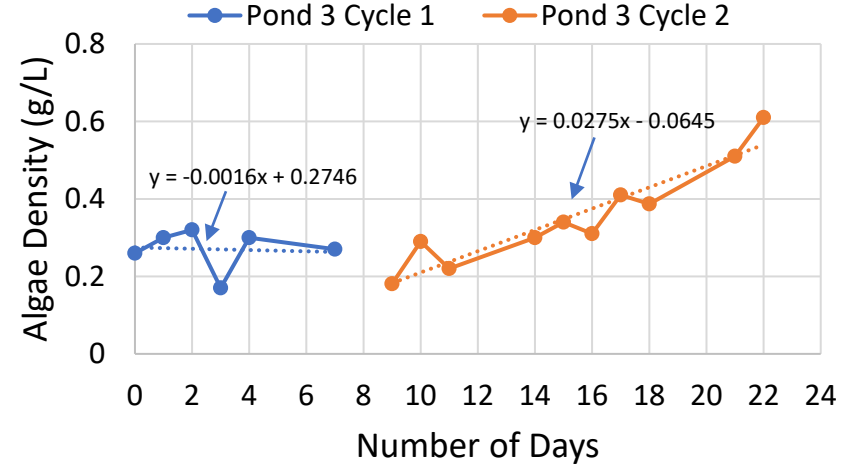
End

# Initial Algae and ORP Data

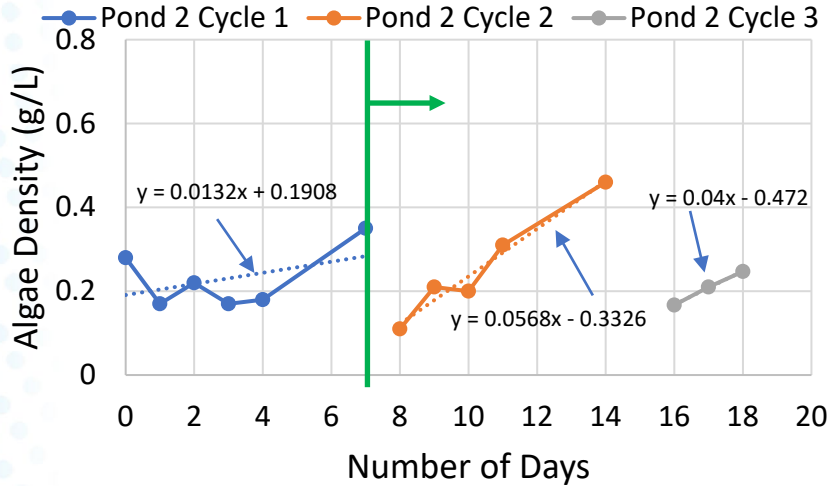
### Pond 1 (Test)



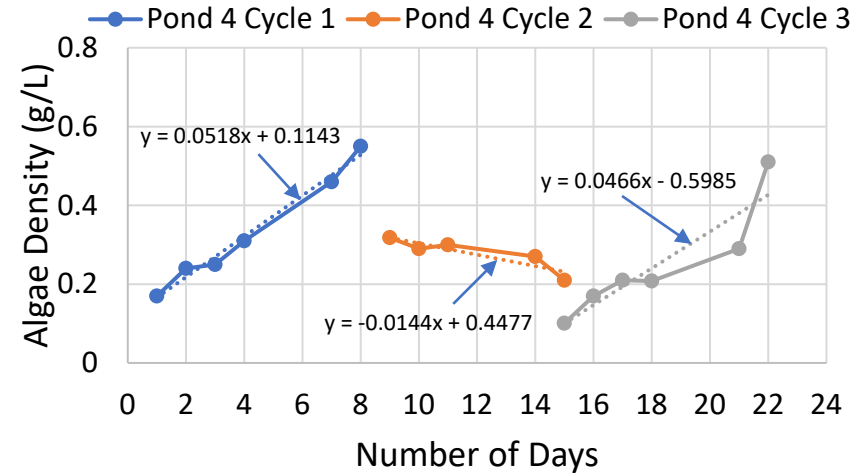
### Pond 3 (Reference)



### Pond 2 (Test)



### Pond 4 (Reference)



Test

$0.062 \pm 0.014$  g/L per day

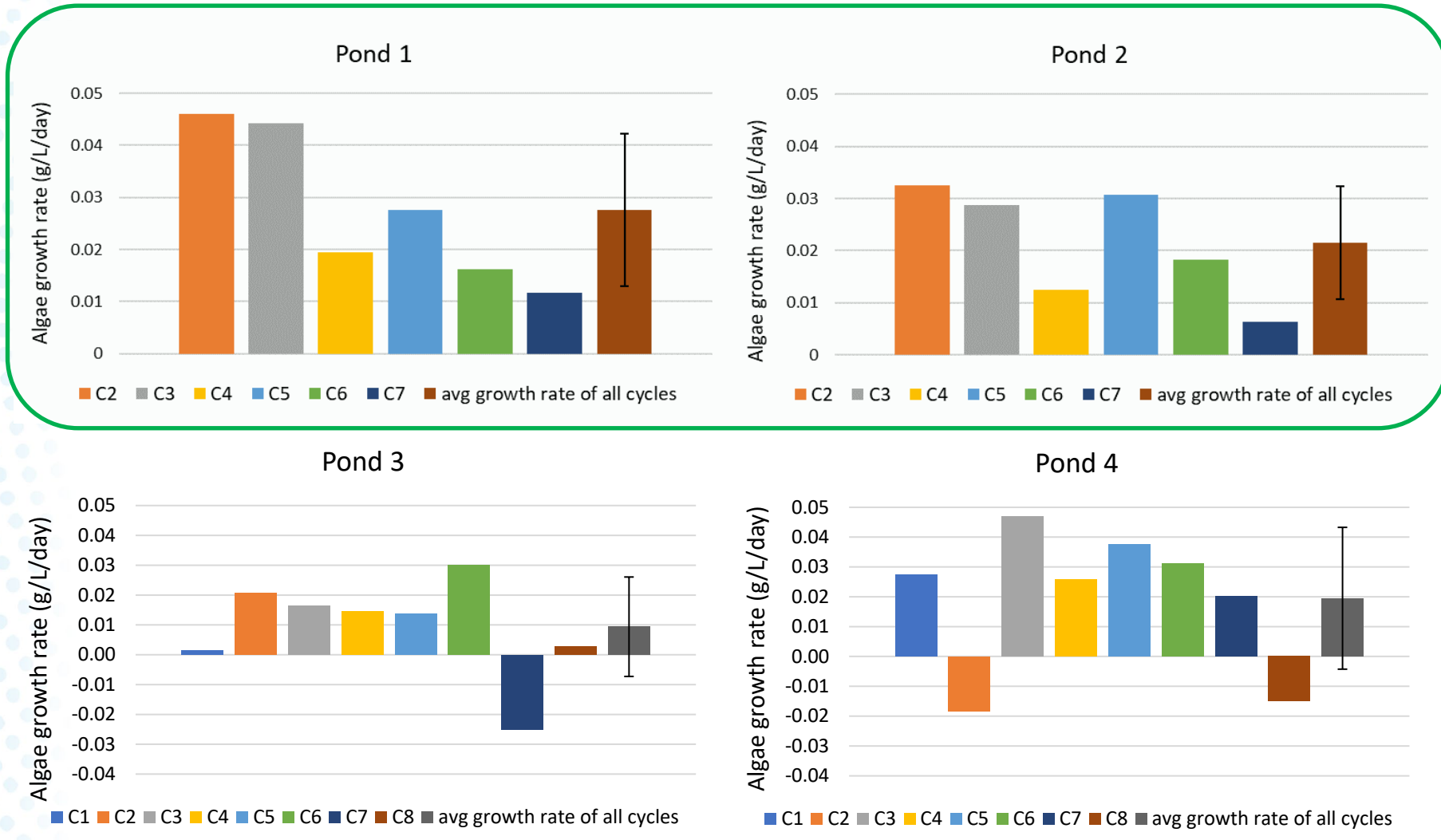
3X increase ←

Reference

$0.021 \pm 0.007$  g/L per day

# Algae Culturing: Comparison of Algae Productivity (Volumetric)

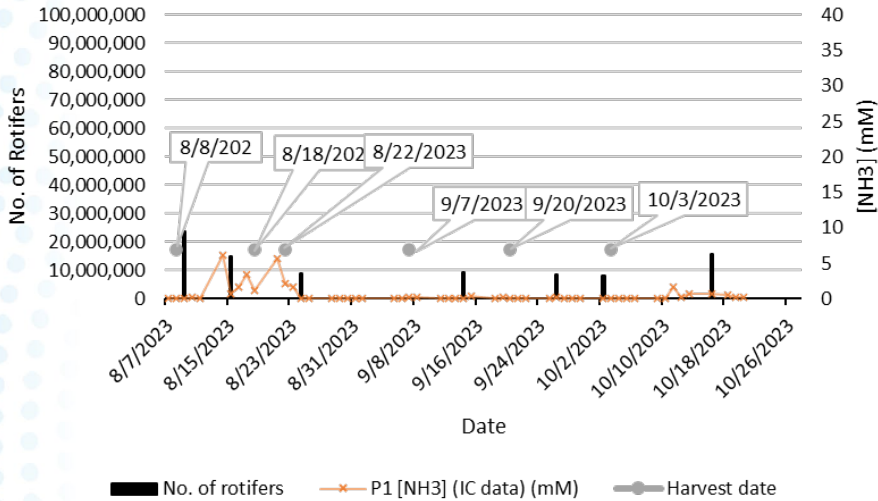
## Test Ponds



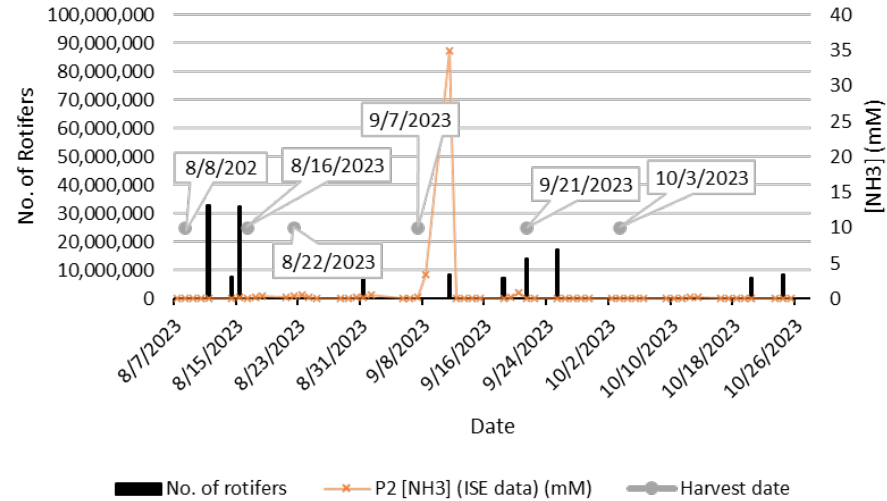


# Algae Culturing: Rotifer Population Data

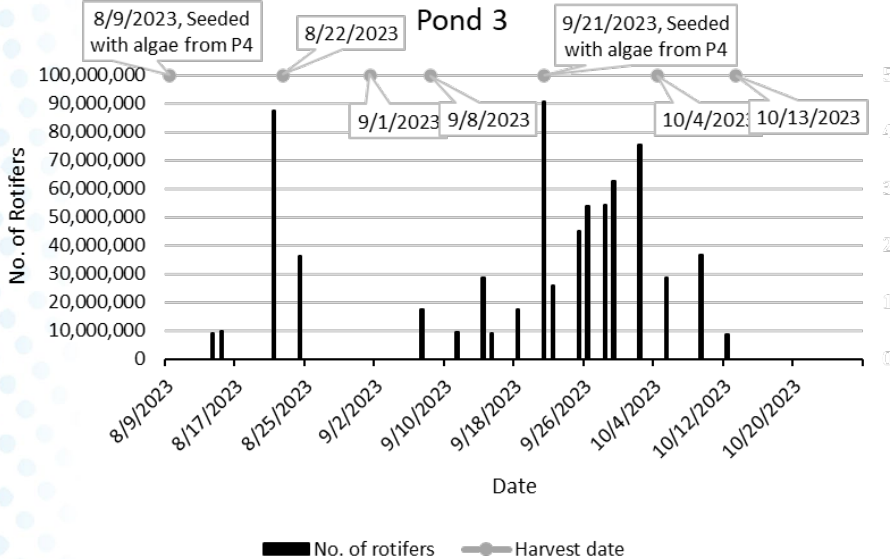
Pond 1



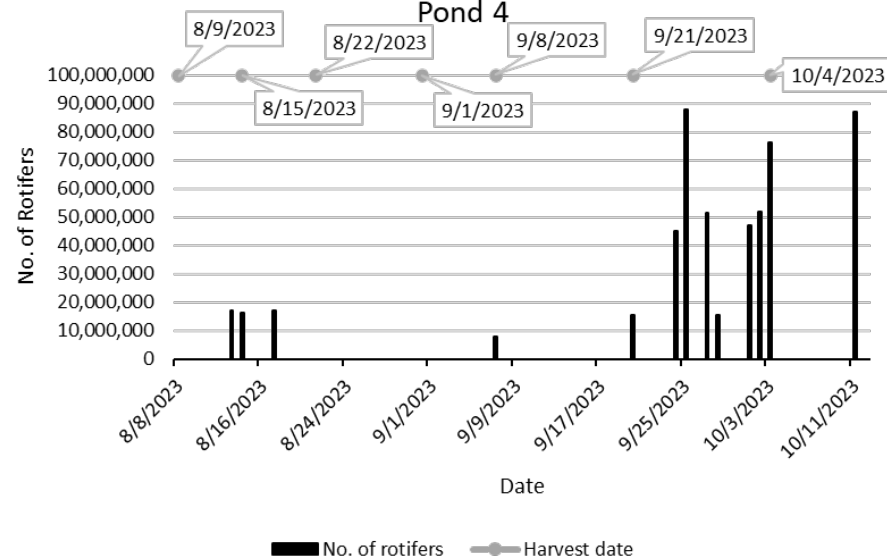
Pond 2



Pond 3

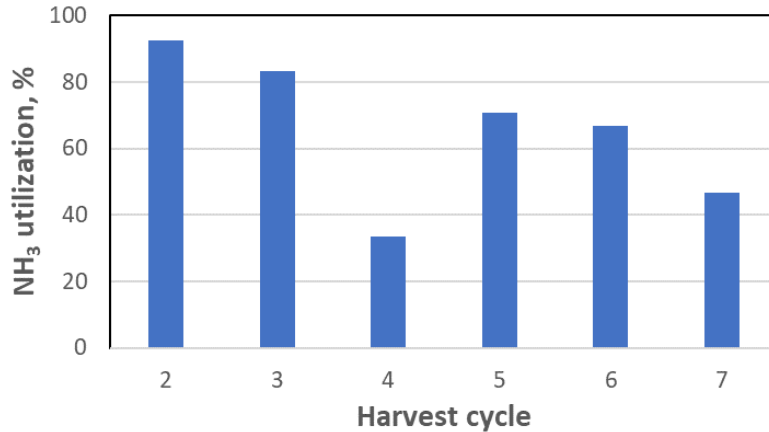


Pond 4

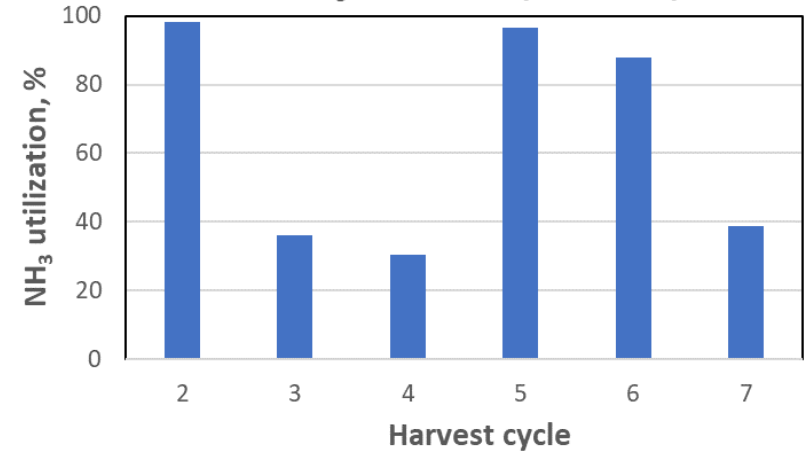


# Algae Culturing: $\text{NH}_3$ Utilization ( $\eta_{\text{NH}_3}$ )

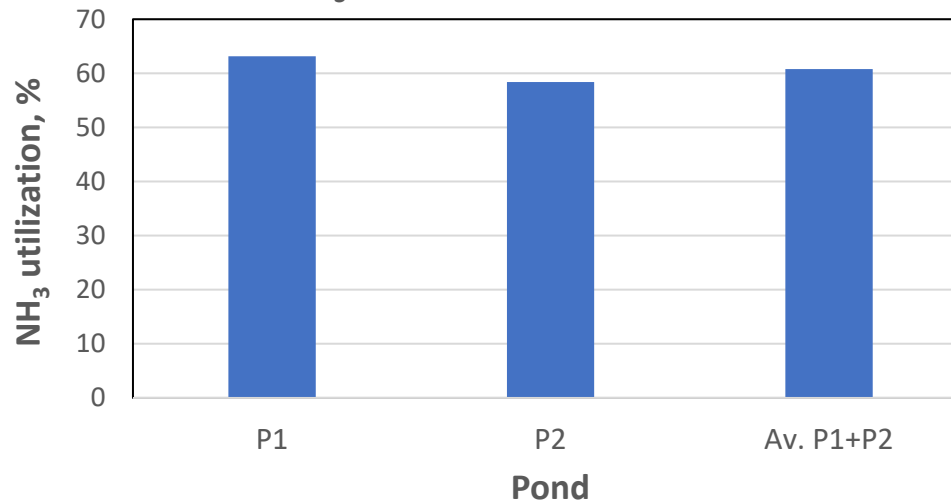
Pond 1:  $\text{NH}_3$  utilization by harvest cycle



Pond 2:  $\text{NH}_3$  utilization by harvest cycle



Overall  $\text{NH}_3$  utilization (based on entire dataset)



$\text{NH}_3$  utilization ranged from 30% - 98% by harvest cycle

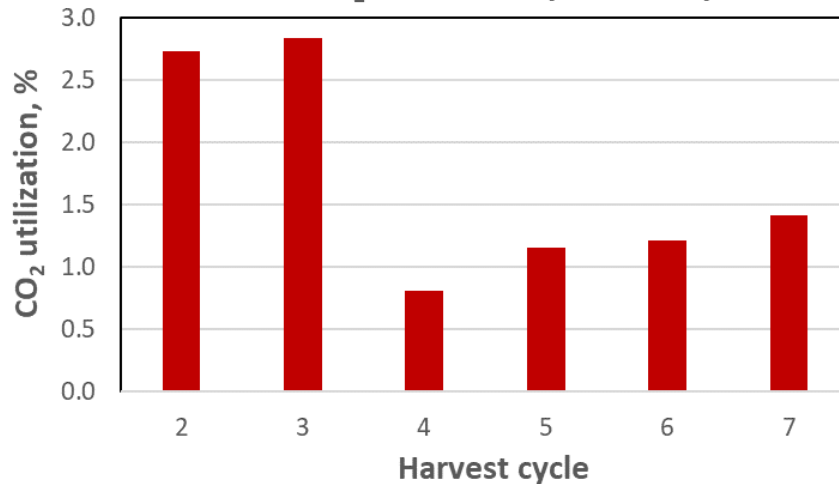
Periods of high  $\text{NH}_3$  utilization correspond to harvest cycles when algae growth was strongest

$$\eta_{\text{NH}_3} = \frac{100 * m_{\text{SA}} * X_{\text{N}}}{m_{\text{NH}_3}}$$

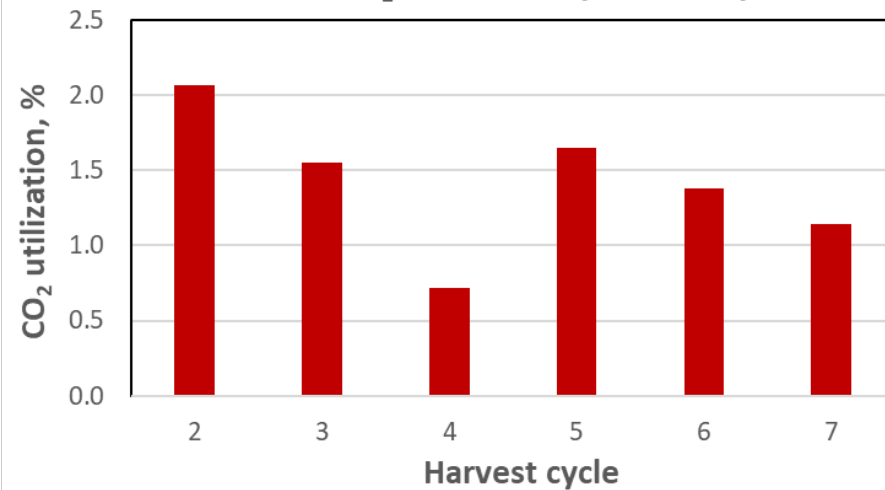
$m_{\text{SA}}$  = mass of algae produced  
 $X_{\text{N}}$  = N fraction in the algae (from elemental analysis)  
 $m_{\text{NH}_3}$  = mass of  $\text{NH}_3$  fed

# Algae Culturing: CO<sub>2</sub> Utilization ( $\eta_{CO_2}$ )

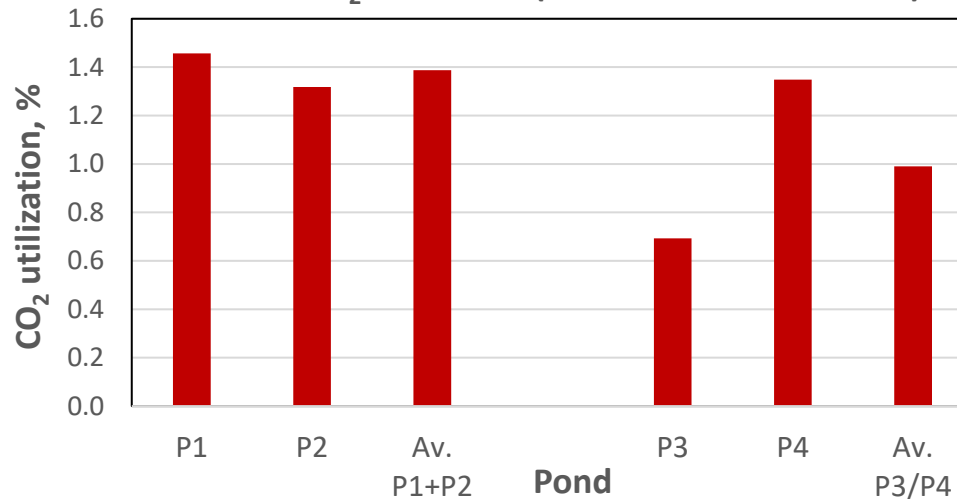
Pond 1: CO<sub>2</sub> utilization by harvest cycle



Pond 2: CO<sub>2</sub> utilization by harvest cycle



Overall CO<sub>2</sub> utilization (based on entire dataset)



$$\eta_{CO_2} = \frac{100 * m_{SA} * X_C}{m_{CO_2}}$$

$m_{SA}$  = mass of algae produced

$X_C$  = C fraction in the algae (from elemental analysis)

$m_{CO_2}$  = mass of CO<sub>2</sub> fed

Low CO<sub>2</sub> utilization efficiencies due to high supplemental CO<sub>2</sub> flow and discontinuous nature of membrane absorber operation (i.e., algae productivity was nitrogen-limited during weekends)

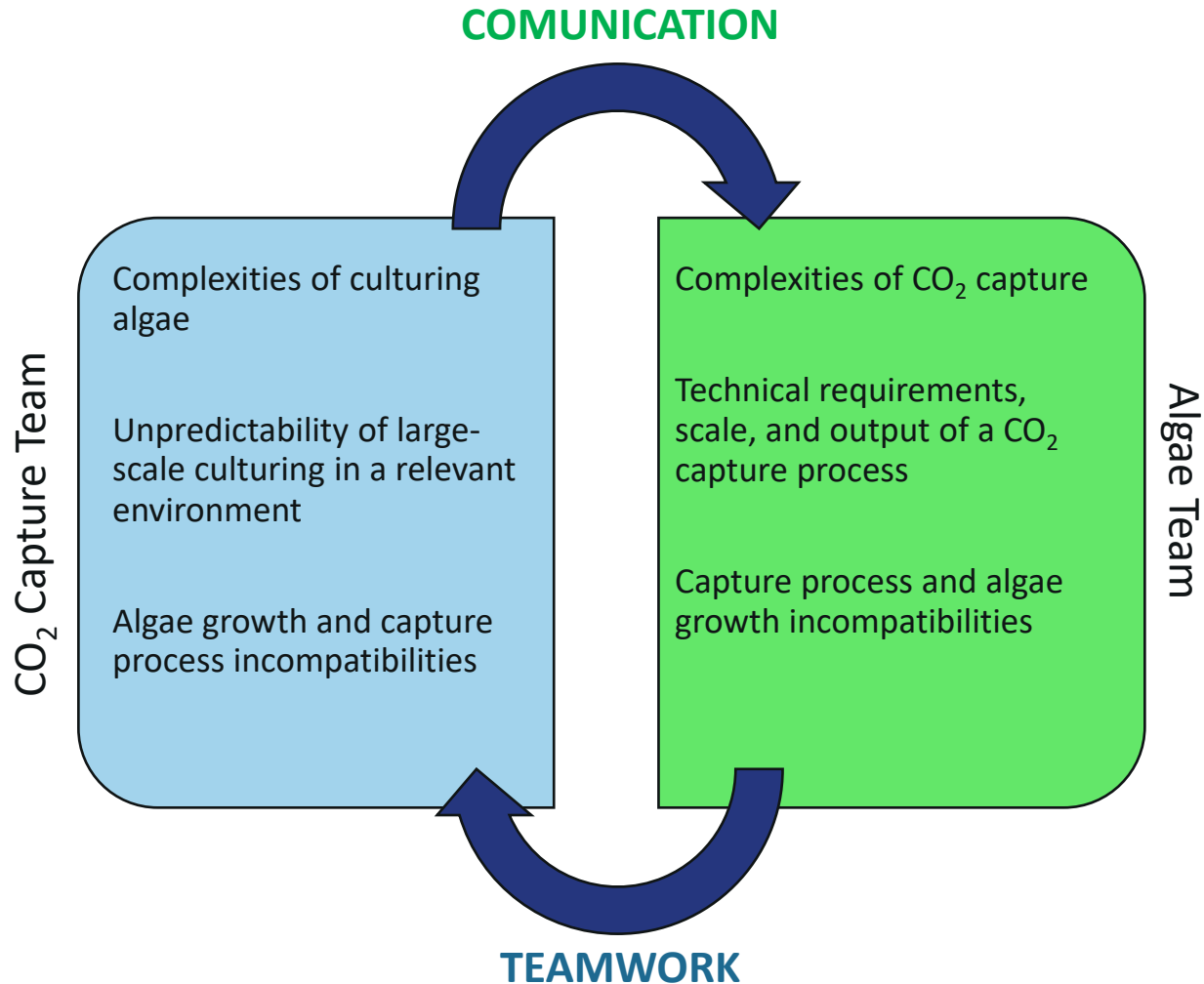
Highest CO<sub>2</sub> utilization corresponds to periods of strong algae growth

# Algae Culturing: Compositional Analysis

	C (%)	N (%)	Ash (%)	Protein (%)
<b>2nd harvest cycle</b>				
<b>ORP 2</b>	48.1 ± 0.4	9.5 ± 0.1	6.0 ± 0.1	45.2 ± 0.1
<b>ORP 4</b>	48.6 ± 0.6	8.5 ± 0.1	n.d.	40.7 ± 0.1
<b>3rd harvest cycle</b>				
<b>ORP 1</b>	50.2 ± 0.5	9.7 ± 0.1	5.9 ± 0.1	46.6 ± 0.1
<b>ORP 2</b>	48.4 ± 0.5	9.2 ± 0.1	7.1 ± 0.2	43.9 ± 0.1
<b>ORP 3</b>	50.9 ± 0.4	8.6 ± 0.2	5.5 ± 0.1	41.1 ± 0.2
<b>4th harvest cycle</b>				
<b>ORP 1</b>	50.5 ± 0.1	9.2 ± 0.0	5.0 ± 0.1	43.9 ± 0.0
<b>ORP 2</b>	47.9 ± 0.2	9.0 ± 0.1	6.3 ± 0.1	42.9 ± 0.1
<b>ORP 4</b>	51.8 ± 0.1	9.1 ± 0.0	6.0 ± 0.1	43.6 ± 0.0
<b>5th harvest cycle</b>				
<b>ORP 1</b>	50.2 ± 0.3	9.5 ± 0.0	6.5 ± 0.1	45.2 ± 0.0
<b>ORP 2</b>	50.4 ± 0.3	9.9 ± 0.2	6.9 ± 0.1	47.3 ± 0.2

- Ash content low in all cases (albeit higher than for indoor experiments)
- No significant differences in protein or ash content between test and reference ponds

# Lessons Learned



# Plans for Future Testing & Development

## Future testing

Scale up algae growth: Currently we are only using 1% of the total capacity of the capture unit, feeding 1800L of algae



X 100 → Total capacity, 180,000L  
(which about the size of a large swimming pool)

## Development

- Streamline the process, experimental → commercial
- Could be used as a polishing step for point source capture to achieve net negative emissions.
- Direct air capture



# Acknowledgements

---

- **DOE-NETL:** Isaac Aurelio, Gregory Imler, Patricia Rawls, Joseph Stoffa
- **UKy:** Lisa Richburg, Xiaoshuai Yuan, Fritz Vorisek, Yaying Ji, Steve Summers, and Reynolds Frimpong
- **Colorado State University:** Jason Quinn
- **Vanderbilt University:** Shihong Lin
- **Trimeric Corporation:** Andrew Sexton