

Achieving Unprecedented CO₂ Utilization in CO₂Concrete™: System Design, Product Development and Process Demonstration

Project Number: DE-FE0031915

Project Manager: Isaac “Andy” Aurelio, and Kanchan Mondal

Prepared by: Gaurav N. Sant, Ph.D.
Institute for Carbon Management

**Prepared for: 2024 FECM/NETL Carbon Management
Research Project Review Meeting, August 5-9, 2024,
Pittsburgh, Pennsylvania.**

Concrete construction is hostage to cement production's embodied CO₂ intensity: So what can be done now?



Cement production emits nearly 10% of global CO₂

0.9 tons of CO₂ are emitted per ton of cement produced; 4.5 billion tonnes of cement produced annually



Global concrete market ~ \$ 1 trillion / year



Ready mixed concrete has the largest market share

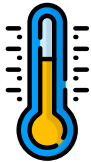


The Increasing use of prefabricated components is a rising industry trend

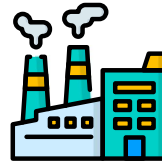


Precast and masonry production allow for controlled conditions, and for better quality control

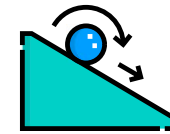
CO₂ utilization (mineralization) implies producing CaCO₃



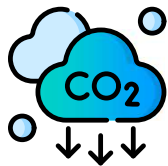
is based on the use of Ca(OH)₂ which can be produced thermally, and electrochemically (without CO₂ emissions)



Carbonation is insensitive to SO_x and NO_x enables direct stack-tap of flue gases – *without a need for a CO₂ capture step*



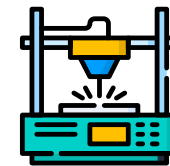
Mineralization is the only “thermodynamically downhill” route for CO₂ utilization



Concrete construction provides a gigaton scale sink for CO₂ emissions annually



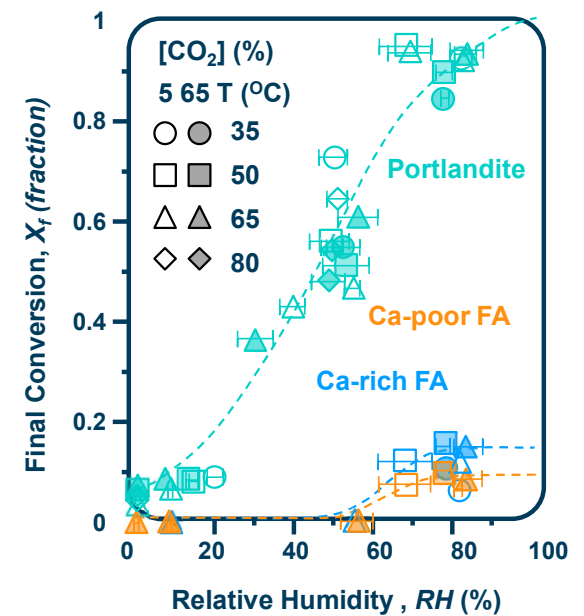
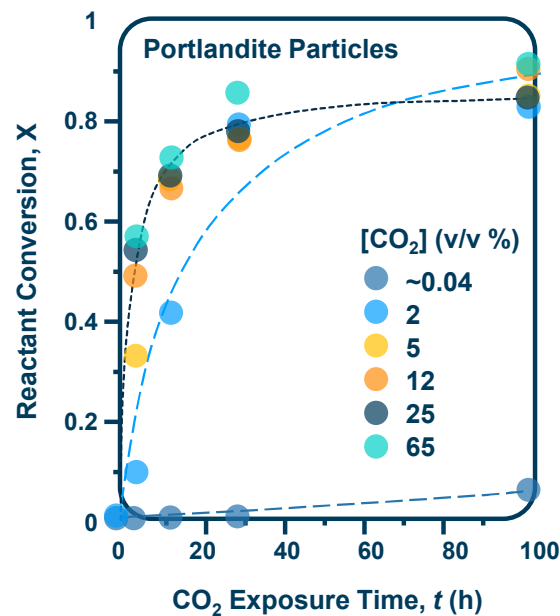
Mobile calcium is available in low-cost reactants, including coal combustion residues (CCRs) and steel slags



Alternate cementation methods may unlock further advantages: fabrication methods (3DP), greater control

Why choose $\text{Ca}(\text{OH})_2$ as a mineralization reactant?

- **Carbonation occurs rapidly** at ambient temperature and pressure without carbon capture step, or gas clean-up (insensitive to SO_x and NO_x)
- **Lower kiln temperature** (800°C) than cement (1500°C) allows reduced energy and CO_2 intensity of production
- **Reaction is downhill:** thermodynamically favorable and requires little, if any, extrinsic energy

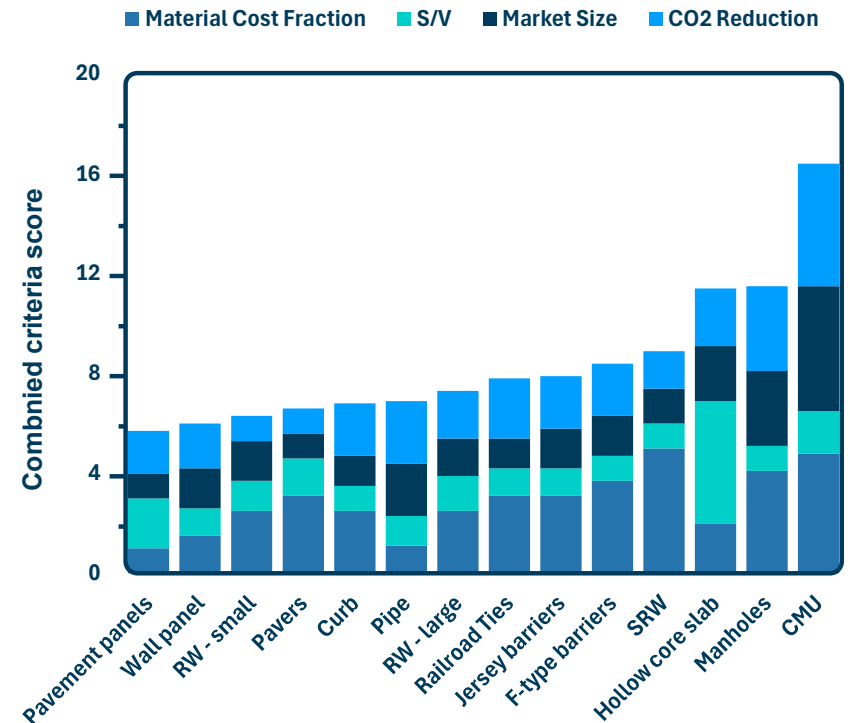


Selecting which product geometries to produce, and why?

Selected three concrete products for mineralization processing:

- **Defined criteria:** Exposed surface-to-volume ratio, estimated CO₂ reduction (reduction in embodied carbon intensity: eCI), market size, cementitious materials cost
- The analysis was informed by data about processing conditions and product performance (e.g., strength)
- Based on combined scoring criteria, we selected: CMU, segmented retaining walls (SRW), and manholes (MAN)

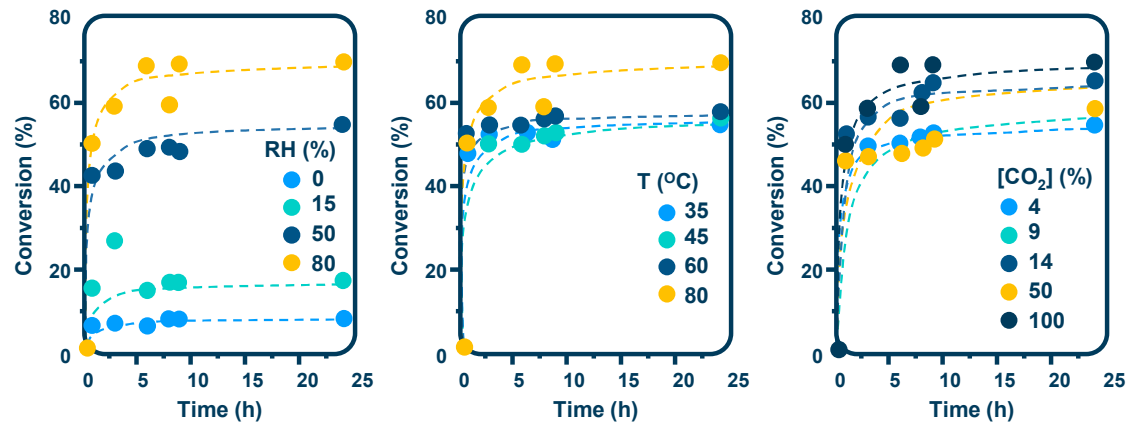
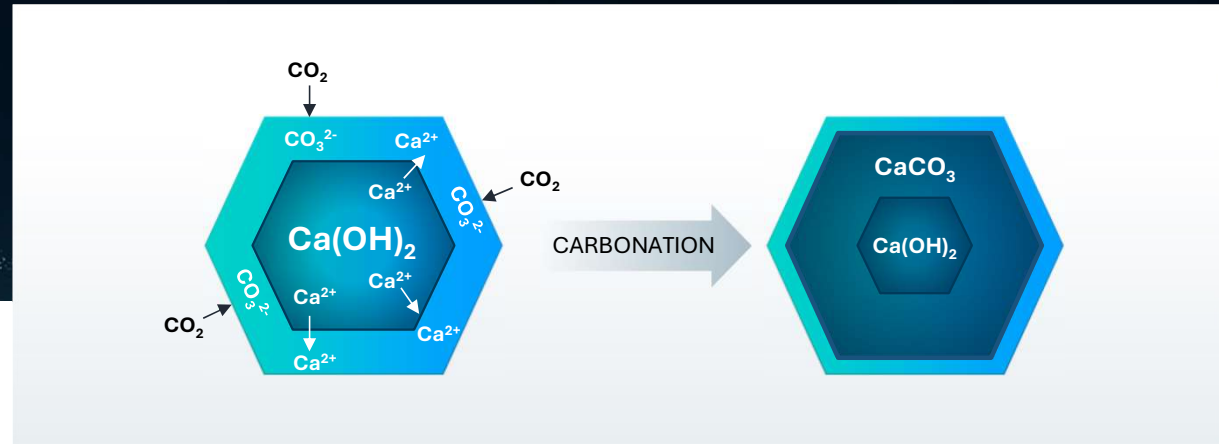
Unit type	Representative S/V (mm ⁻¹)	CO ₂ uptake rate gCO ₂ /gCa(OH) ₂ (K _s - s ⁻¹)
CMU	0.08	9.54E-04
SRW	0.05	6.16E-04



Maximizing reactant consumption and mineralization

Carbonation potentials of reactants – portlandite ($\text{Ca}(\text{OH})_2$)

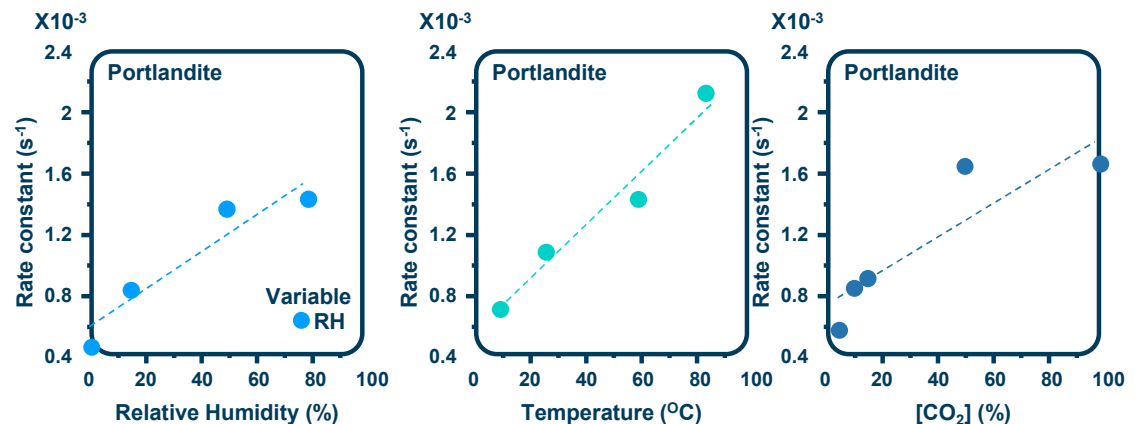
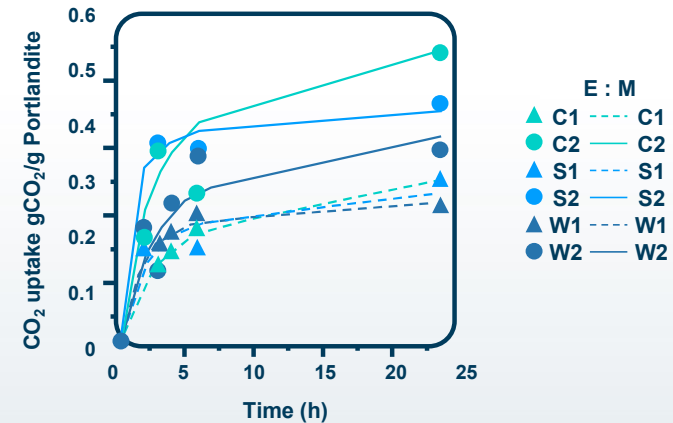
- Portlandite carbonation largely dependent on dissolution and precipitation of $\text{Ca}(\text{OH})_2$ and CaCO_3 , respectively
- T and $[\text{CO}_2]$ have minor affect on extent of reaction which allows ambient conditions
- Carbonation is near-complete within 24 h
- The Significance:** Pressurization, CO_2 enrichment, or significant heating is not required for portlandite carbonation



Microstructural influences on carbonation kinetics

Carbonation kinetics of CO₂ Concrete mixture compositions

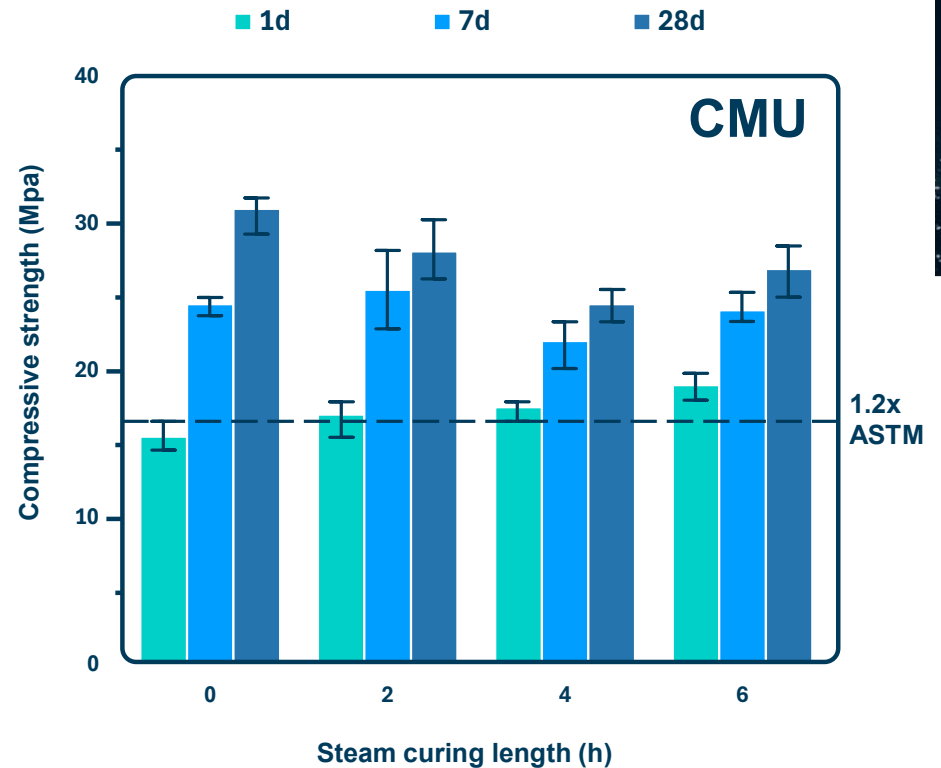
- Kinetics of particulate carbonation depend on relative humidity, T and [CO₂].
- Composite mixture formulation: Cement, portlandite and fly ash that were formed into geometries with suitable surface-to-volume (s/v, m⁻¹) ratios
- Carbonation data was collected across range of operating conditions to simulate flue gas conditions: T: 20-to-65°C, relative humidity (RH): 10-to-80%, and CO₂ concentration: 3-to-15 vol. %
- Important to match the “testing” surface area to volume ratio (s/v) of concrete products/components. In general, the carbonation rates for each component are $K_s > 1E-05 \text{ s}^{-1}$



The influence of carbonation on strength gain?

Establishing processing-property relations

- Measured strength gain of carbonated materials as a function of the duration of steam curing cycle
- In general, a shorter steam curing duration enhances carbonation by reducing the surface coverage of reactant particles – which are otherwise passivated
- Carbonated concrete exceeds the strength requirement for relevant product selections while ensuring CO₂ uptake

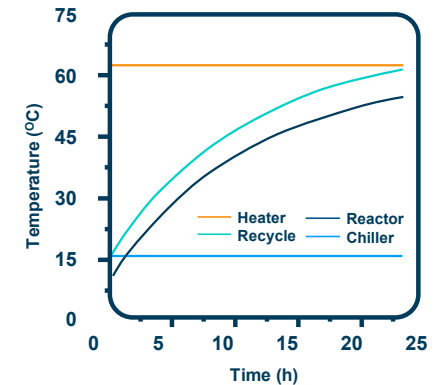
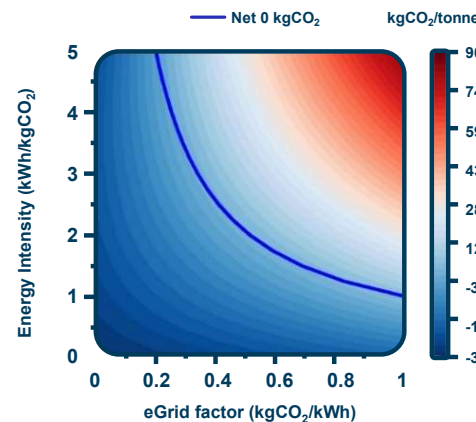
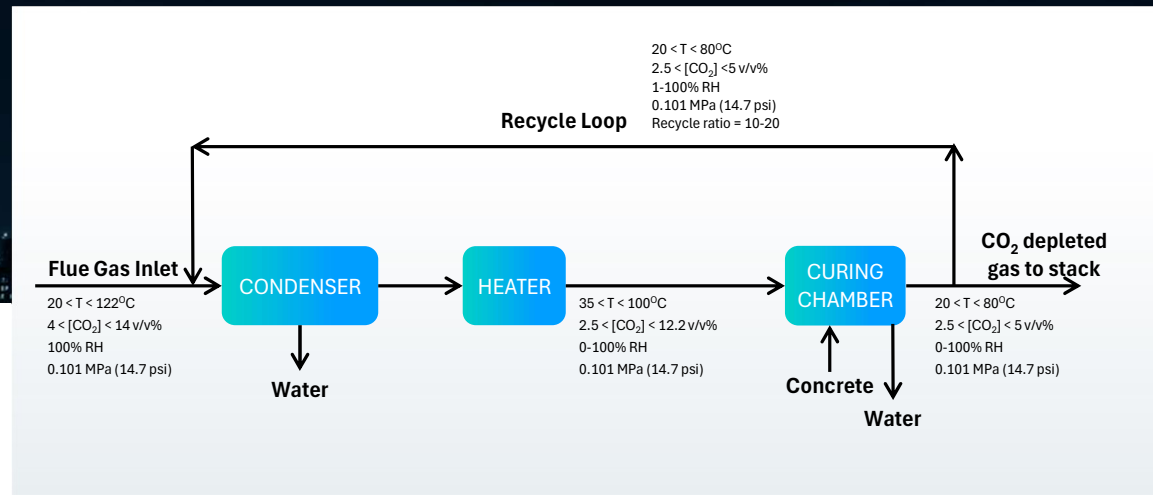


AspenPlus© model of carbonation conditions, in reactor, over time

Development of carbonation/mass transfer model at the bench-scale

Two Aspen Plus© based models were developed:

1. Steady-state model that assesses 24-h average CO₂ uptake based on reactor conditions (RH, [CO₂], T)
2. Dynamic model to simulate the reactor and the process conditions in time over 24-h operating period
3. The dynamic model incorporates a carbonation regression model to determine how time-dependent conditions affect reactant conversion/CO₂ uptake
4. As appropriate, the simulation data was used for developing stream tables for the design basis memorandum (DBM)

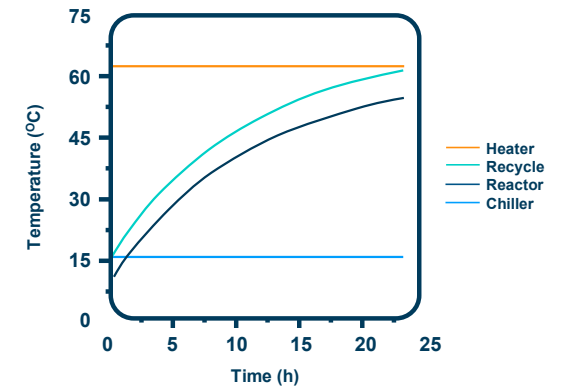
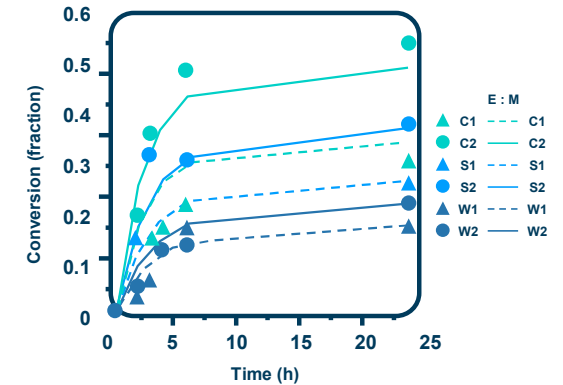
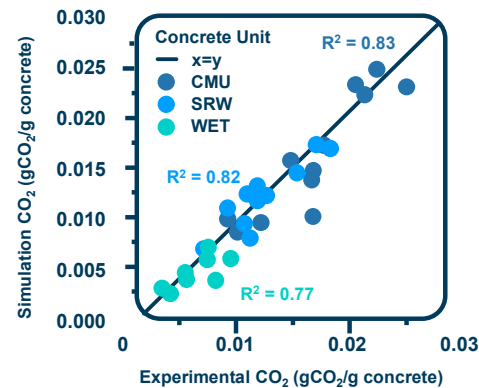
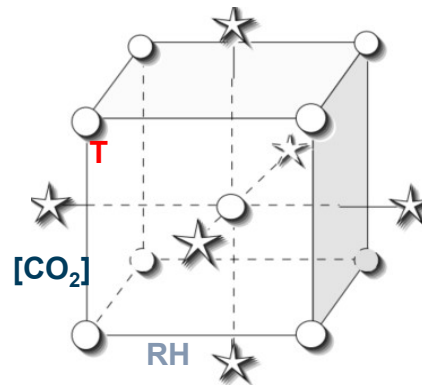


Developing predictive (DOE) model to ascertain CO₂ uptake

Developed predictive mode for the carbonation of CO₂Concrete™ components at the bench-scale

- This model is developed using a “Design-of-Experiments” approach that is used to assess the parametric dependence of T, RH and [CO₂] on CO₂ uptake
- The model estimates CO₂ uptake across varying carbonation conditions for each product geometry across a range of temperatures, RH and CO₂ concentration
- The regression model makes use of Aspen Plus© simulation data to assess how process variables and conditions affect mineralization across diverse conditions

$$X(\text{conversion}) = x + a[\text{CO}_2] - bRH + cT$$

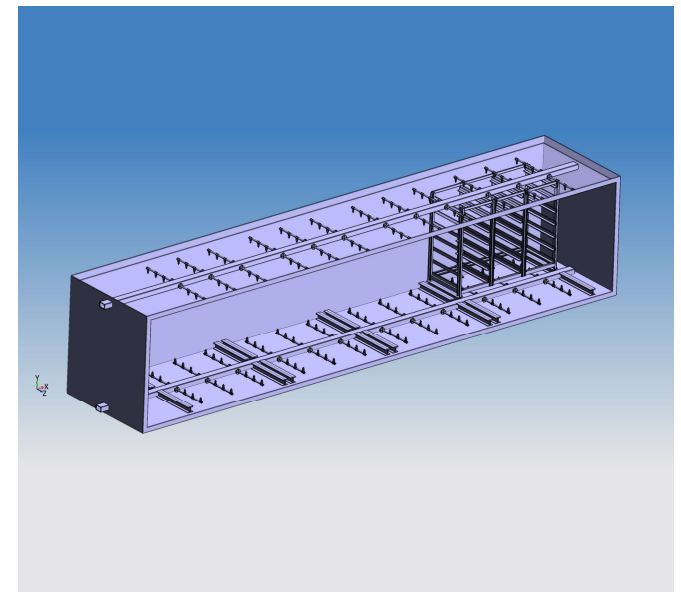
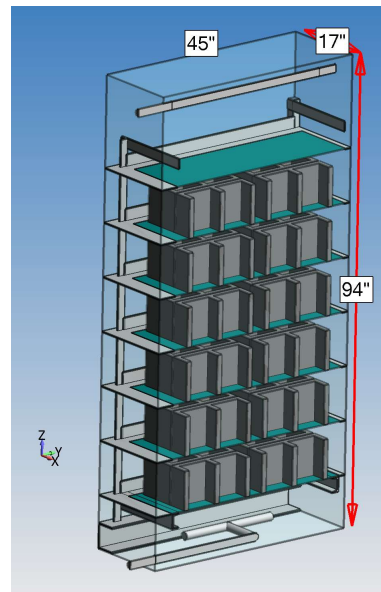


Design modifications to existing mineralization system

OBJECTIVES

Identify modifications required to the existing CO₂ mineralization system or design and fabricate new system for selected precast concrete components

- **Subtask A** – CFD model development
- **Subtask B** – System design modifications/fabrication
- **Subtask C** – HAZOP



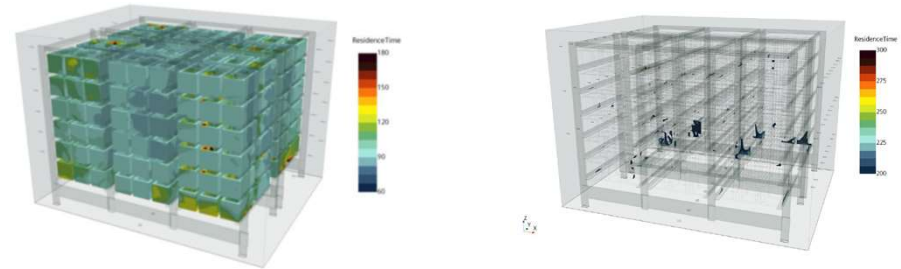
Design modifications to existing mineralization system

SUBTASK 6.1

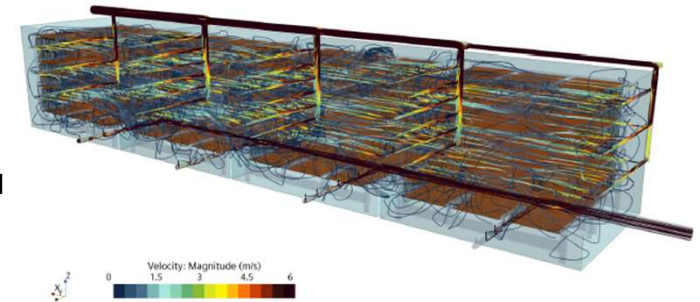
CFD Model Development

- Computational fluid dynamics (CFD) used to optimize gas flow and internal reactor geometry for concrete elements
- CFD helped guide the required concrete component flowrate coverage to meet conditions indicated by the process model
- System gas control skid was modified and a redesign of the carbonation reactor was completed based on the CFD simulation results
- An updated design basis memorandum (DBM) was submitted for fabrication

TRANSIENT HEAT AND MASS TRANSFER ANALYSIS



GAS FLOW DISTRIBUTION OPTIMIZATION



System fabrication, installation and commissioning

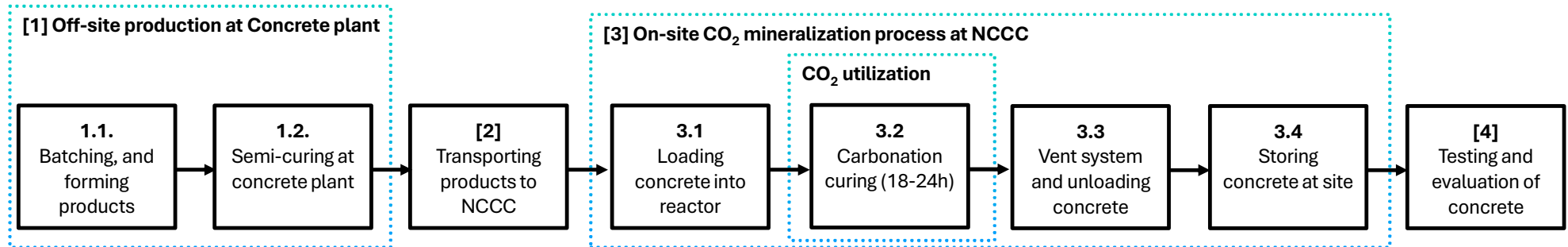
- The system was fabricated, and post-FAT (for functionality, and quality) was shipped to the NCCC for installation
- At the NCCC, internal reactor instrumentation was installed and tested including:
 - RH-T sensors
 - Flow and pressure sensors
 - CO2 concentration sensors
 - Electric power meters
 - Water flow meters
- System functionality and heating rate of the reactor were evaluated following commissioning
- Following system commissioning, multiple test runs were carried out to produce concrete products with support of local precast/masonry producers



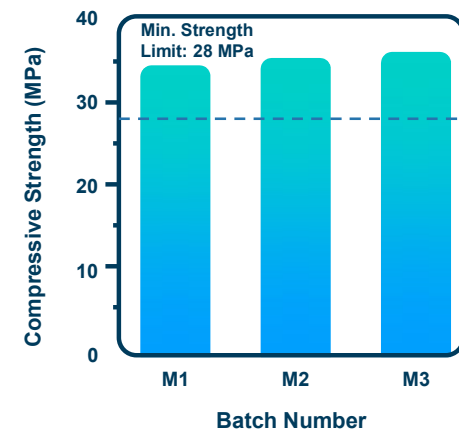
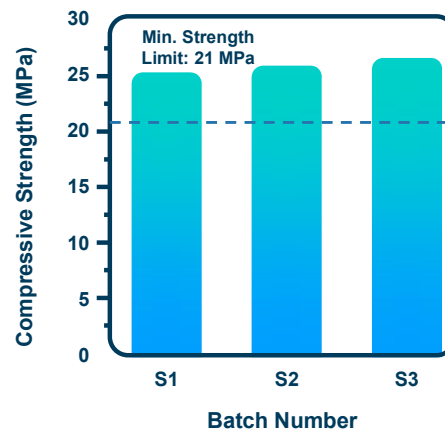
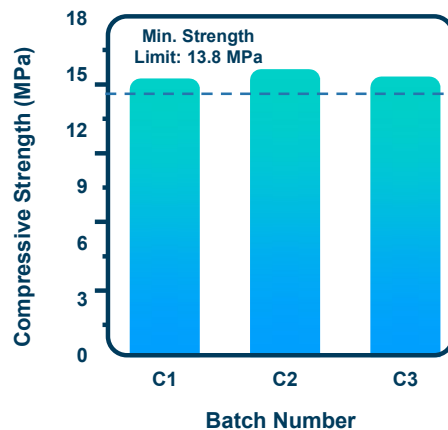
National Carbon Capture Center (NCCC), AL, 2024

Concrete product manufacturing, and performance validation of carbonated products

- Concrete was semi-cured at concrete facilities (Blair Block and Alcrete)
- This method of offsite production required substantial “green strength” for the products to be transported which resulted in more cement use
- The formed products were shipped to NCCC and loaded into carbonation reactor for 18-to-24 hours of processing
- Once completed, the concrete was unloaded and stored for ASTM testing (carried out by 3rd-party testing entity)
- **Total mass produced:** CMU: 31 tonnes, SRW 51 tonnes, Manholes: 40 tonnes; with a production rate of >10 tonnes of production per day



Performance validation, and product compliance (with existing ASTM standards) with emphasis on strength properties



- Carried out standardized testing for each product based on existing standards (e.g., by NCMA's testing laboratories)
- Each concrete product passed ASTM strength requirements. And, each test run exceeded target $0.2 \text{ gCO}_2/\text{g}$ reactant
- Despite variation in CO_2 uptake – minimal variations in strength observed in concrete samples

Technoeconomic analysis, lifecycle assessment, and overall commercialization strategy

- Key success of demonstration was to produce a concrete product with reduced embodied carbon intensity (ECI)
- Net CO₂ reduction via replacement of raw materials and CO₂ uptake
- Lifecycle analysis identified net CO₂ reduction across all products:
 - CMU: 30.1%
 - SRW: 31.3%
 - MH: 19.8%
- These values are below project target of 45% but this is a function of the green strength requirement for transport

BLOCK MAKING ≈ STANDARD BLOCKS

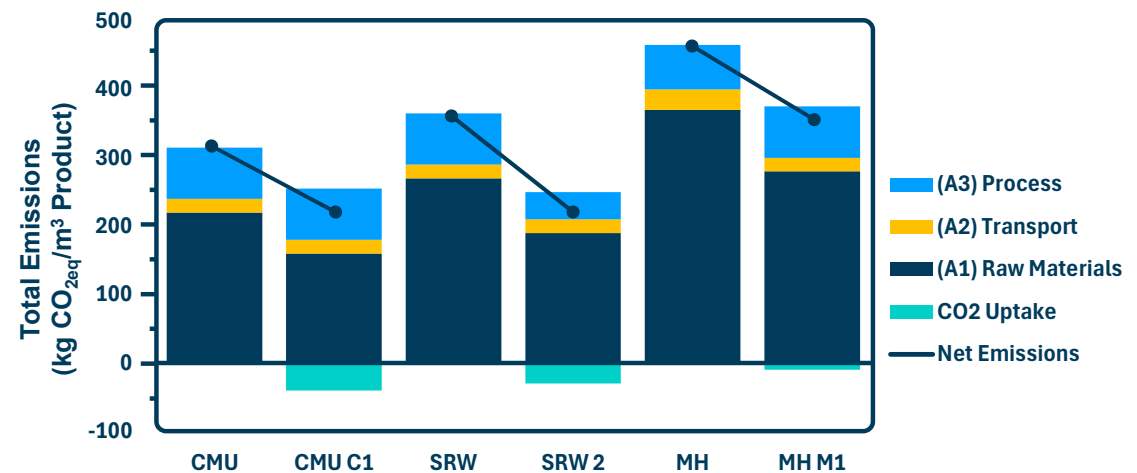
TRANSPORTATION ≈ STANDARD BLOCKS

CARBONATION PROCESS > STANDARD BLOCKS

RAW MATERIALS << STANDARD BLOCKS

CO₂ UPTAKE >> STANDARD BLOCKS

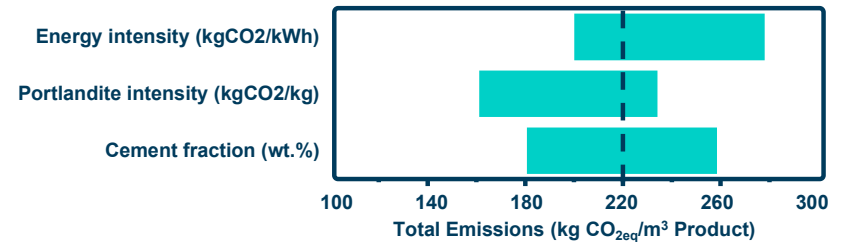
NET CO₂ EMISSIONS << STANDARD BLOCKS



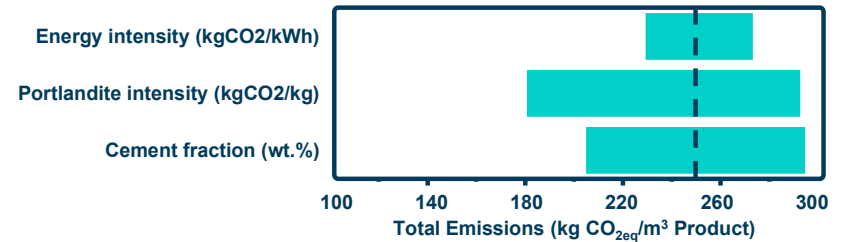
Technoeconomic analysis, lifecycle assessment, and overall commercialization strategy

- Limitations of performing demonstration at host-site influenced final eCI reductions:
 - **Cement mass fraction** – minimum required to achieve transport strength
 - **Hydrated lime eCI** – future scenarios could obtain from renewable sources
 - **Site energy supply** – limited to local eGrid values. Plan to retrofit sites with biomass generators -> CO₂ supply and energy
- Sensitivity analysis was conducted for each product to assess these key areas

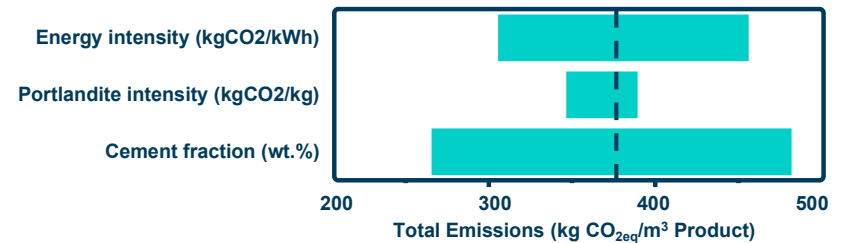
CMU



SRW

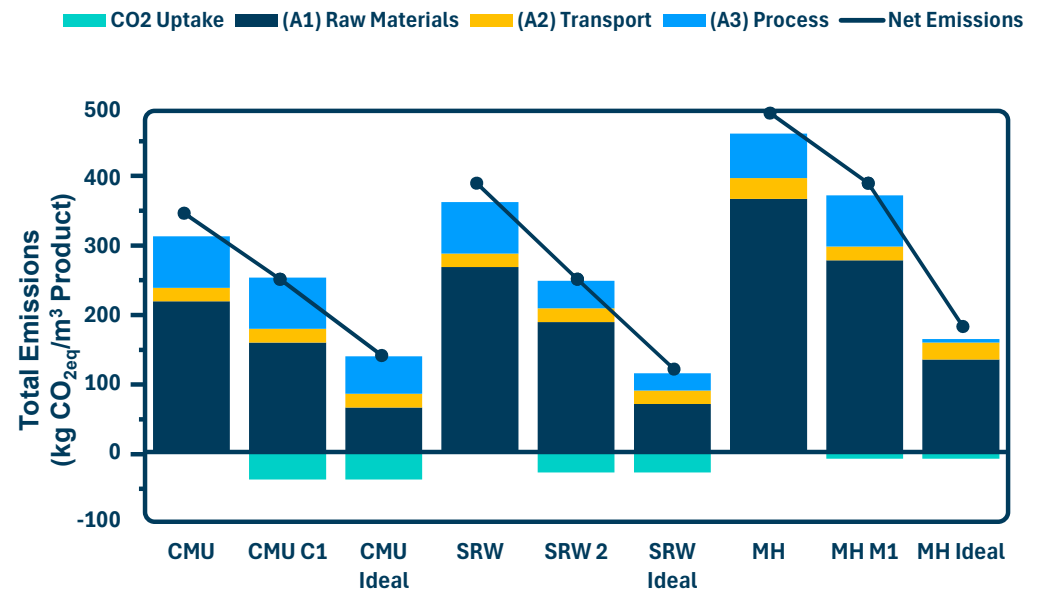


Manhole



Technoeconomic analysis, and lifecycle assessment: What we've accomplished, and what more can be done?

- Demonstrated an approach to reduce eCI of precast and concrete masonry units by up to 30% by: reducing cement use, and enabling CO₂ mineralization
- The reduction was lower than targeted due to the need to use more cement to ensure “green-strength” during transport prior to carbonation
- There are three clear pathways to reduce eCI further: **Reduce cement use, use low(er) eCI lime and lime containing residues, use green(er) electricity**
- With these reductions, for typical onsite production, eCI reductions of up to 70% can be realized quickly
- CarbonBuilt is delivering on this approach as is in commercial operations today (in Alabama, with other projects in the pipeline).



Summary and Conclusions

- CO₂ mineralization is a simple, straightforward, and ready to deploy pathway for CO₂ utilization and low-carbon concrete production
- CO₂ mineralization in concrete can be achieved using dilute CO₂ streams, and without a carbon capture step – unlike other pathways
- The large size of the concrete market is an opportunity, but requires spatially delocalized access to (dilute) CO₂ for mineralization
- The UCLA team has demonstrated the feasibility of this technology at pilot-scale and exceeded project milestones
- This project has advanced the technology from TRL 3+ to TRL 6+
- This technology is already being commercialized by CarbonBuilt at concrete masonry plants across the United States