

# Ignition, turbulent flame speeds, and emissions from high hydrogen blended fuels

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Performance period: Sept. 2021 – Sept. 2024

UTSR Project: DE-FE0032079 PM: Mark Freeman

UTSR Workshop

# Background and Motivation

- Improve understanding of fundamental phenomena of hydrogen containing fuels for gas turbines.
- Pure hydrogen ( $H_2$ ), carbon free hydrogen carrying fuels (such as ammonia,  $NH_3$ ), mixtures of them and with natural gas
- Lots of data on autoignition delays and flame speeds exist, what is new?
  - Inconsistent trends on experimental data
  - Not much data at practical conditions (e.g., most data are in highly diluted environments)
  - Performance of existing kinetic models diverges significantly

# Objectives

- Objective 1: Development of comprehensive database on autoignition delays for hydrogen containing fuels, including hydrogen/natural gas blends, pure hydrogen and ammonia, and ammonia/hydrogen blends at realistic gas turbine conditions (Task 2).
- Objective 2: Measurement of turbulent flame speeds and emissions of hydrogen containing fuels at different turbulence levels (Task 3).
- Objective 3: Measurement of laminar flame speeds of hydrogen containing fuels at high preheating conditions (Task 4).
- Objective 4: Validation and optimization of existing kinetic models using data obtained from experiments and development of reduced kinetic model specific for hydrogen containing fuels (Task 5).

# Task 1: Project Management and Planning

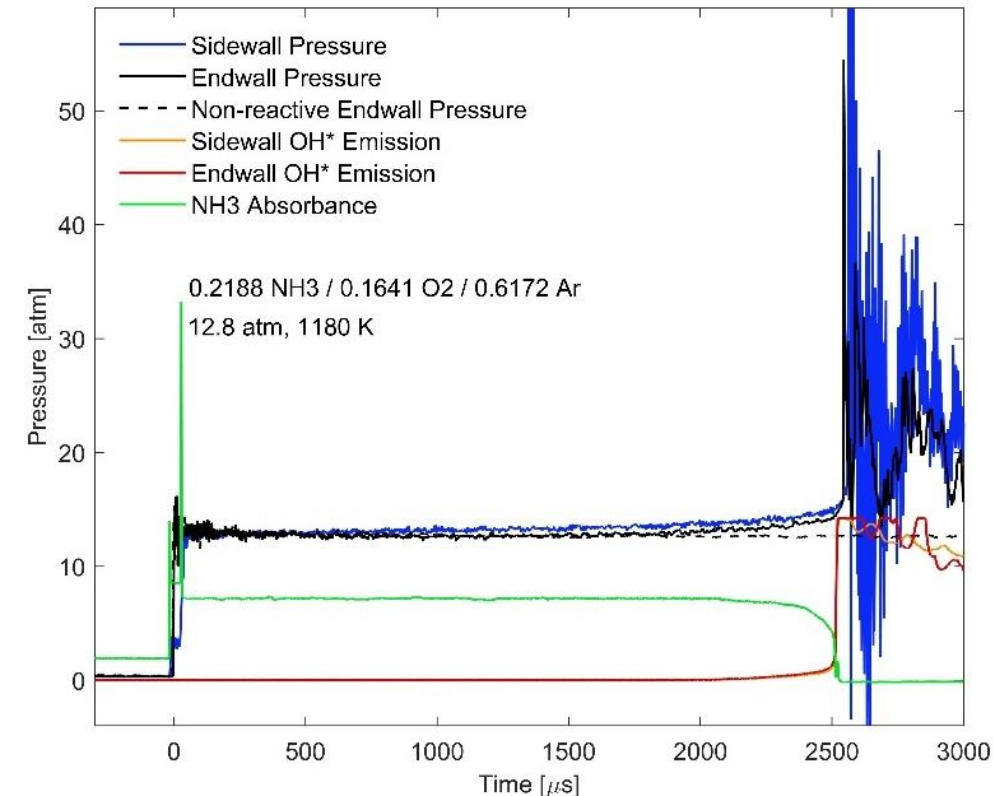
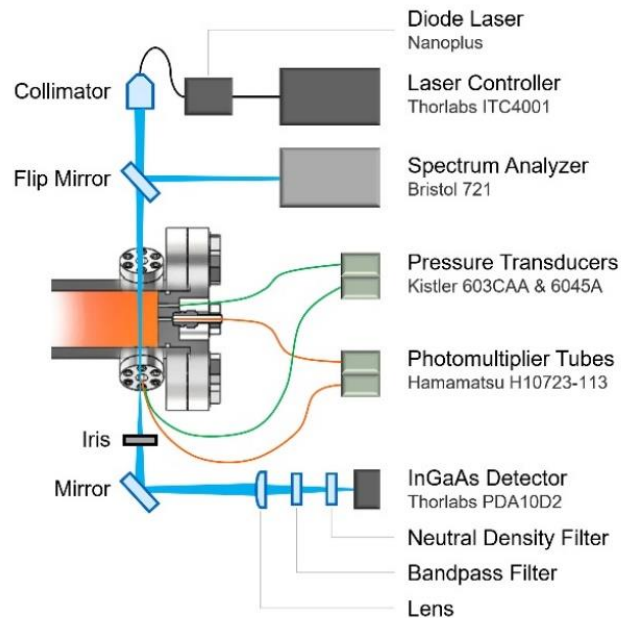
- We will manage and direct the project in accordance with a Project Management Plan to meet all technical, schedule and budget objectives and requirements.
- We will coordinate activities in order to effectively accomplish the work.
- We will update the Project Management Plan 30 days after award and as necessary throughout the project to accurately reflect the current status of the project.
- We will submit quarterly report, attend program review meetings, and arrange regular meetings with program manager.

# Task 2: Investigation of Autoignition of Ammonia/Hydrogen

- Existing data are mostly in highly diluted environments
- Fuels: pure  $\text{NH}_3$  and  $\text{NH}_3/\text{H}_2$  mixture
- Temperature range: 1100 K to 2200 K
- Equivalence ratio: 0.5, 1 and 2
- Pressure:  $\sim 10$ -20 atm
- Facility: high pressure shock tube

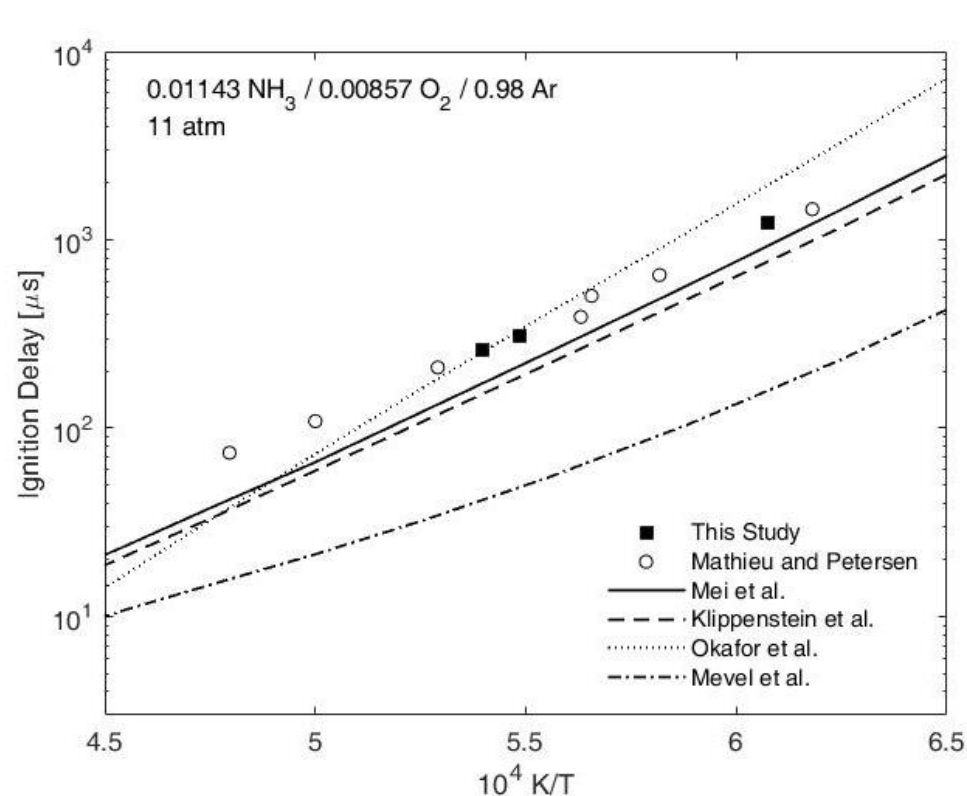
stoichiometric  $\text{NH}_3/\text{O}_2/\text{Ar}$  mixture with 22% fuel at 12.8 atm and 1180 K.

Schematic of experimental setup at the measurement section

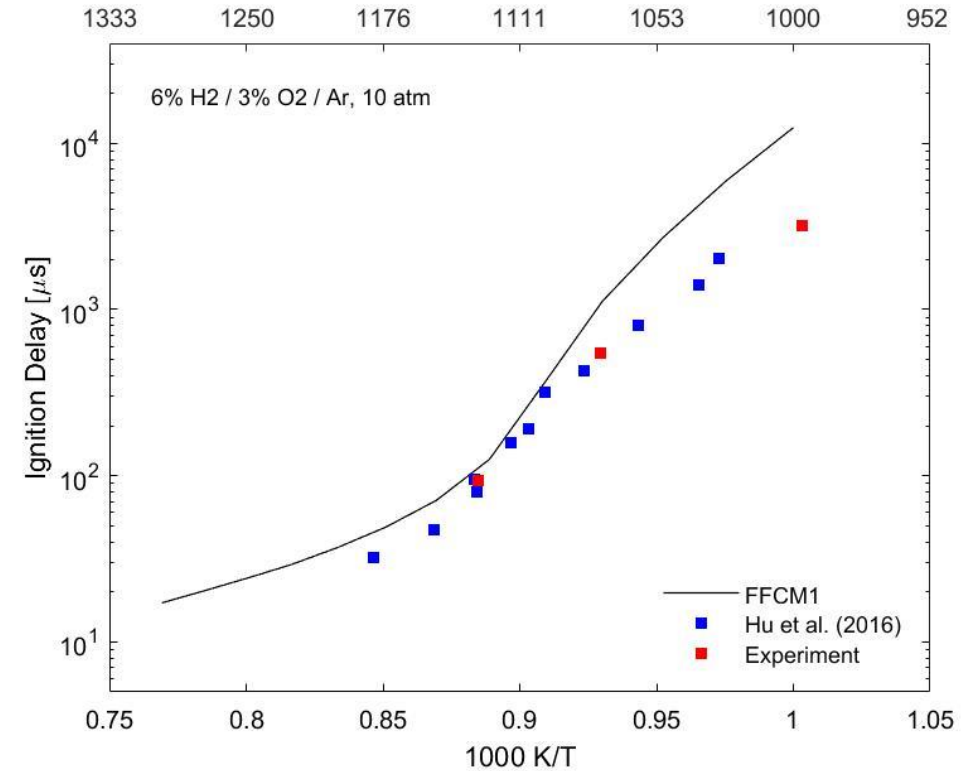


# Benchmark of shock tube IDT measurement

- Repeat experiments reported in literature and compare with simulations



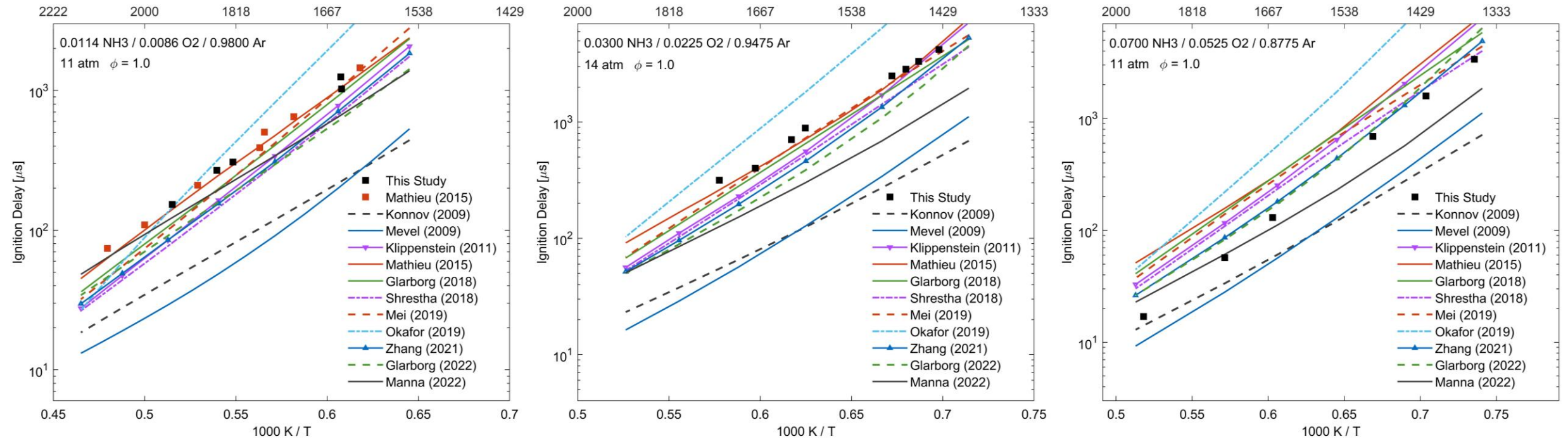
Comparison between experiments and simulation for IDTs of NH<sub>3</sub>/O<sub>2</sub>/Ar mixture at 11 atm



Comparison between experiments and simulation for IDTs of H<sub>2</sub>/O<sub>2</sub>/Ar mixture at 10 atm

# IDTs of NH<sub>3</sub>

- Compared with different kinetic models
  - Good agreement with Mathieu & Petersen (2015) up to 3% ammonia
  - Deviation increases with the increase of fuel concentration
  - No model can reproduce experiments well at all conditions



O. Mathieu, E.L. Petersen, *Combust. Flame* 162 (2015) 554-570.

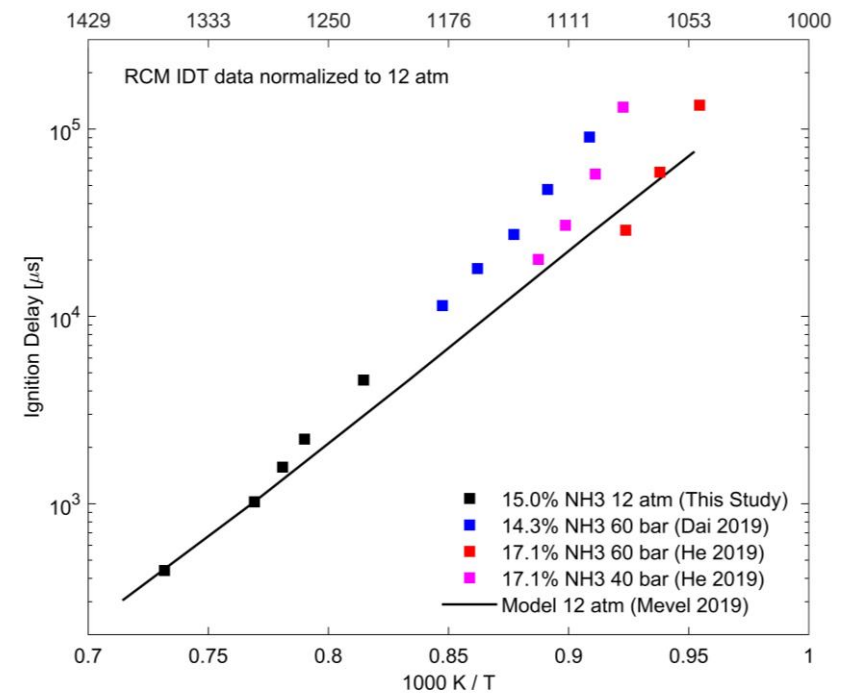
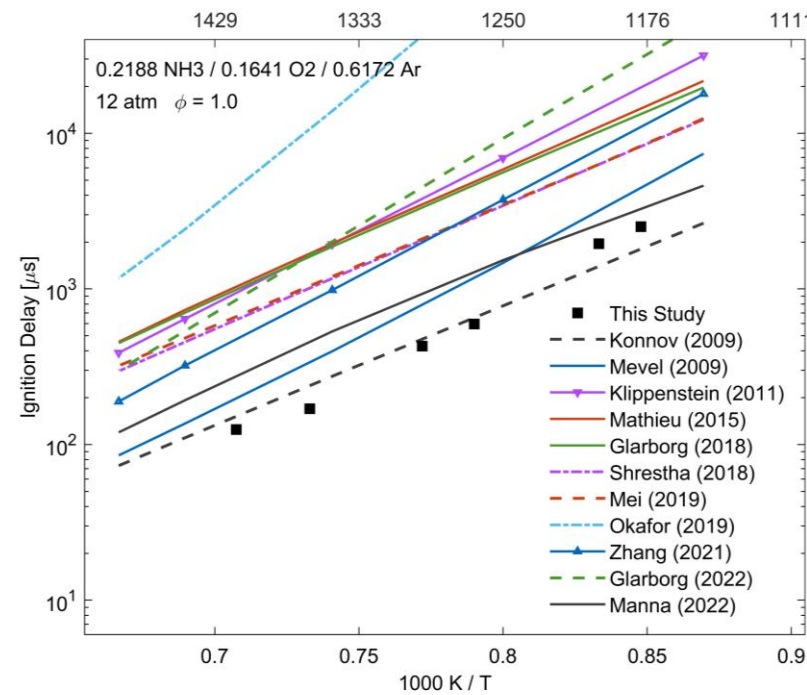
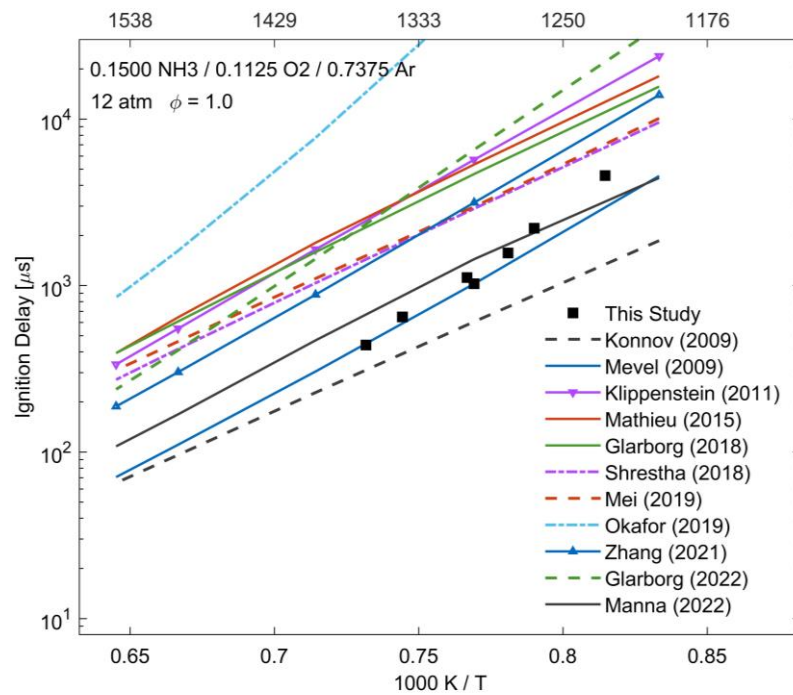
R. Mevel, S. Javoy, F. Lafosse, N. Chaumeix, G. Dupre, C.E. Paillard, *Proc. Combust. Inst.* 32 (2009) 359-366.

A.A. Konnov, J. De Ruyck, *Combust. Sci. Technol.* 168 (2001) 1-46.



# IDTs of NH<sub>3</sub>

- Agreement with Mevel et al. (2009) and Konnov & De Ruyck (2001) at 15-20% ammonia
- Validated with existing RCM (rapid-compression machine) data near 15% ammonia
- Clear trend observed. How to explain the fuel concentration dependence effect? (I don't know!!!)
- Publication: Peng, D. Ranjan, W. Sun, "A shock tube study of fuel concentration effect on high-pressure autoignition delay of ammonia" 2023 Applications in Energy and Combustion Science, 16, 100202



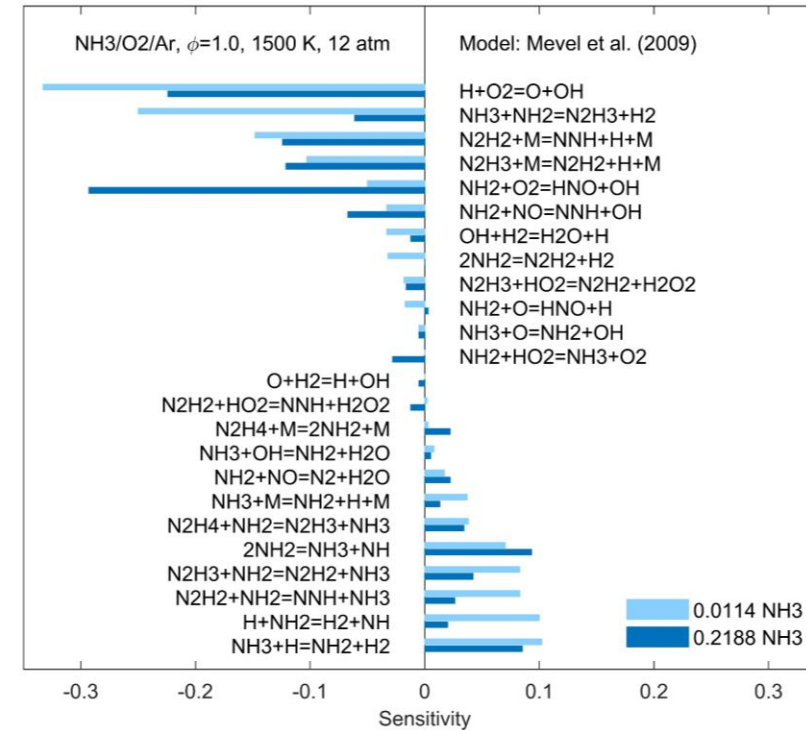
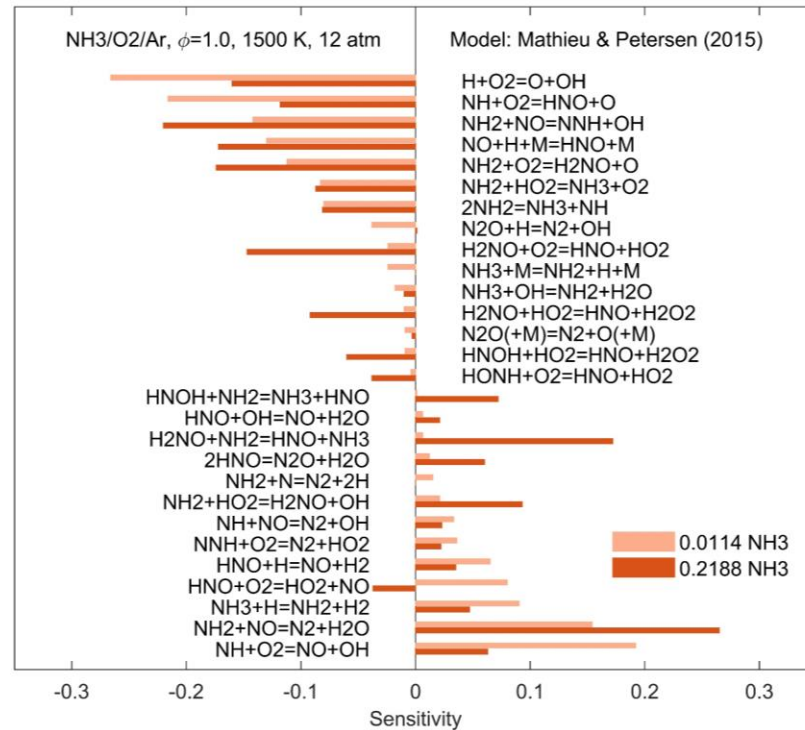
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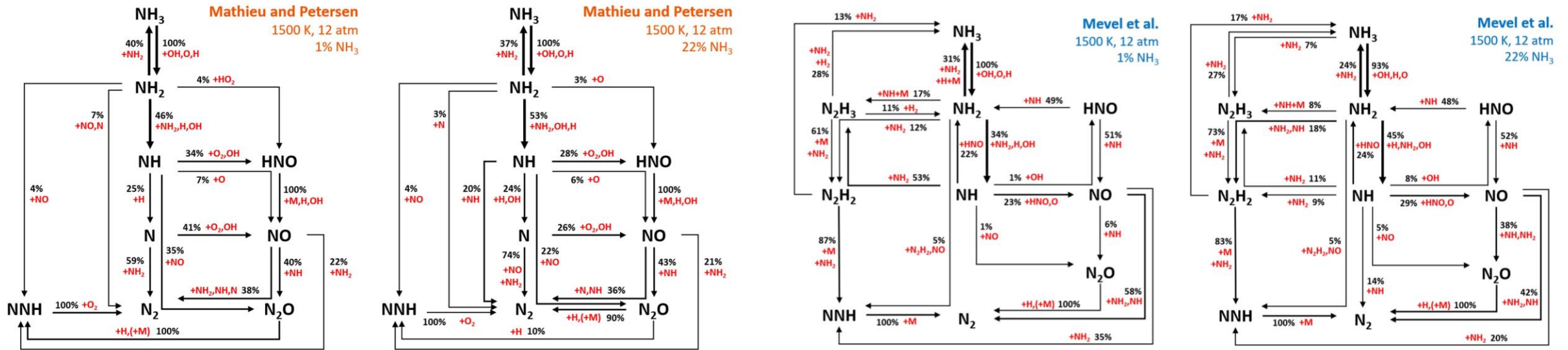


# Sensitivity Analysis



- Compared with kinetic models by Mathieu & Petersen (2015) and by Mevel et al. (2009), each having good agreement with experiments at 1% and 22% ammonia, respectively

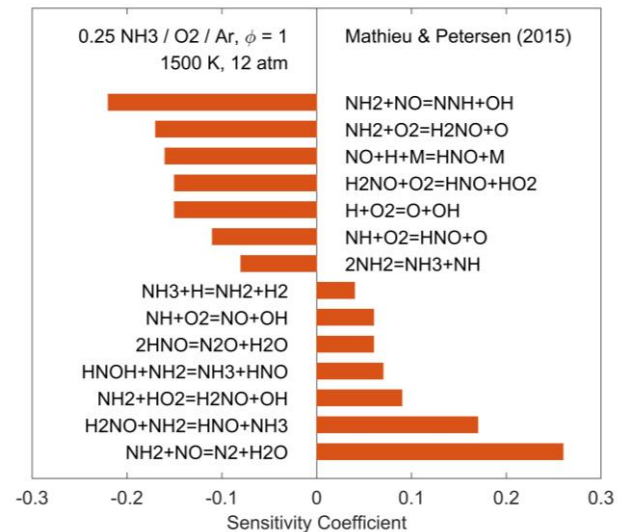
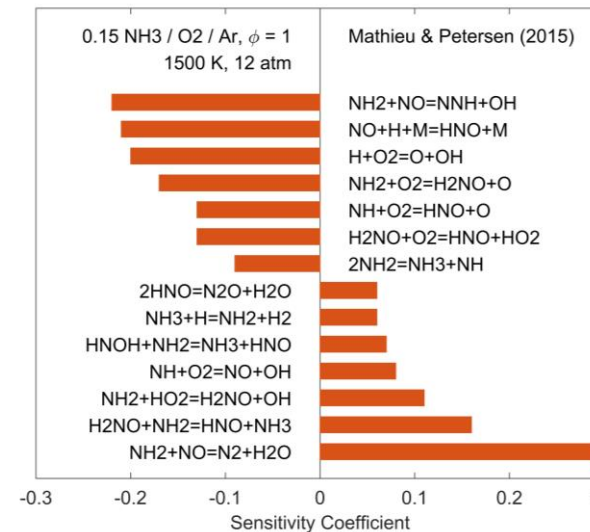
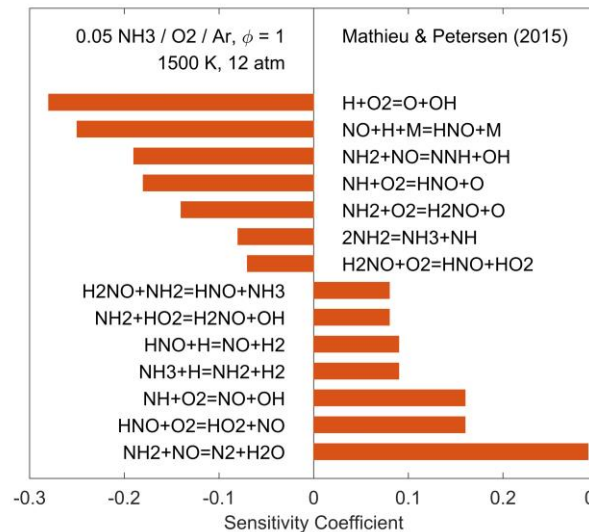
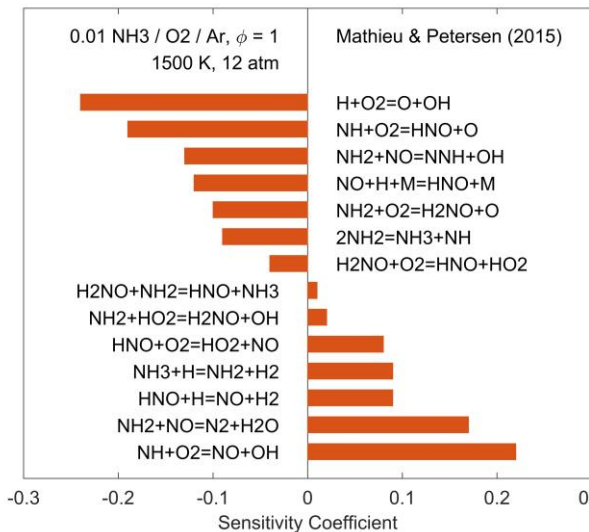
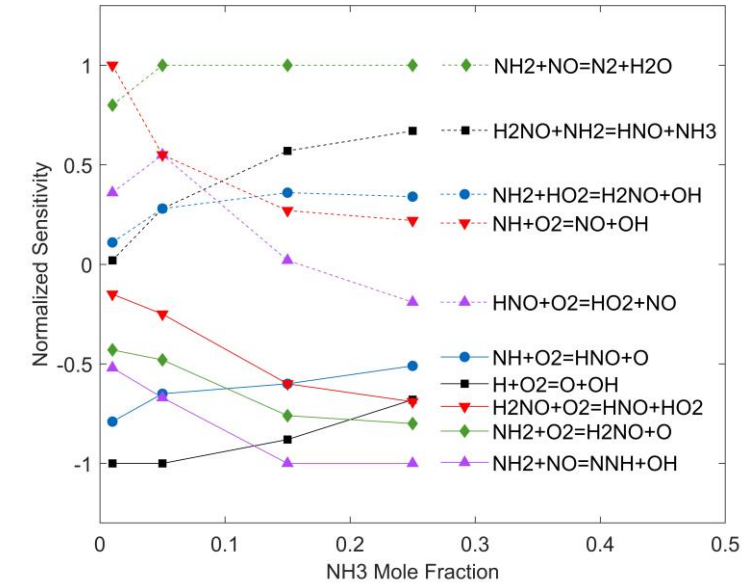
# Reaction Pathways Analysis



- Compared with kinetic models by Mathieu & Petersen (2015) and by Mevel et al. (2009), each having good agreement with experiments at 1% and 22% ammonia, respectively

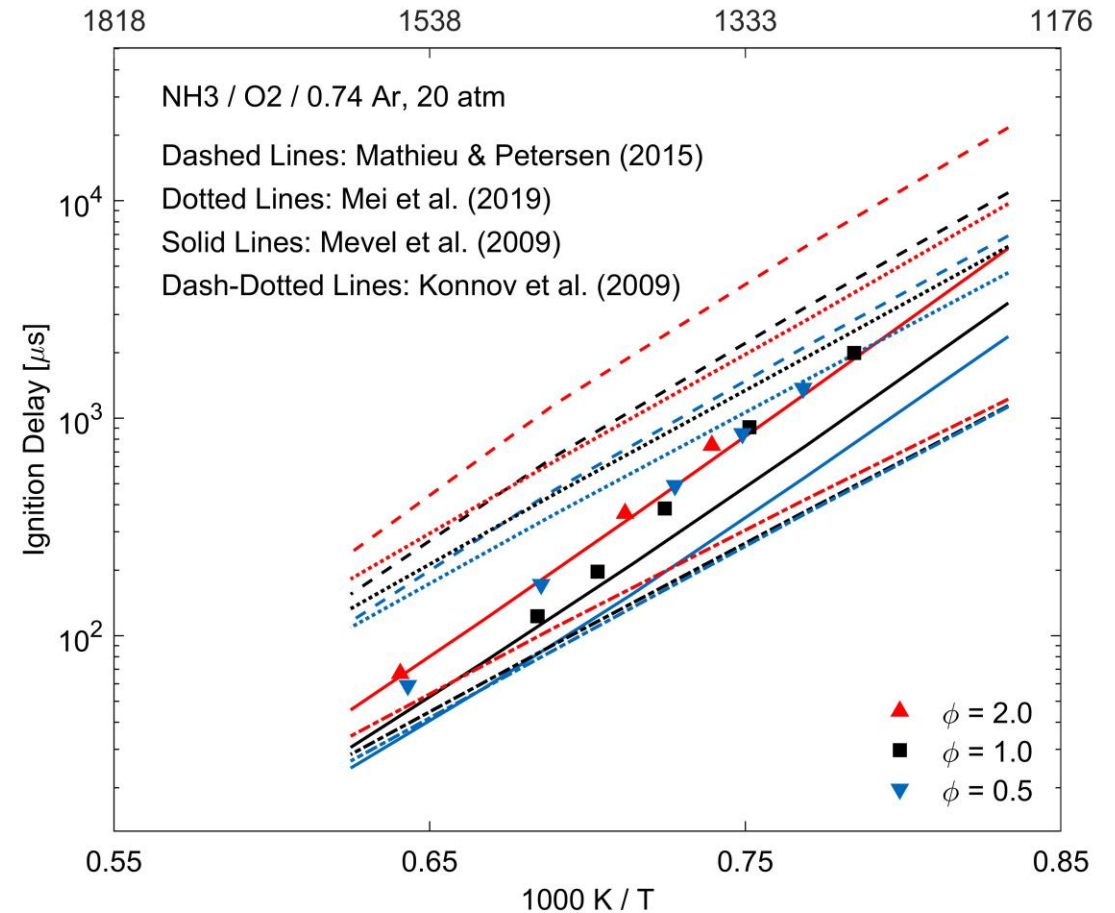
# IDTs of NH<sub>3</sub> – Sensitivity on Fuel Concentration

- As fuel concentration increases from 1% to 25% (model by Mathieu and Petersen, 2015)
  - The strongest ignition promoter changes from  $\text{H}+\text{O}_2=\text{O}+\text{OH}$  to  $\text{NH}_2+\text{NO}=\text{NNH}+\text{OH}$
  - The strongest ignition inhibitor changes from  $\text{NH}+\text{O}_2=\text{NO}+\text{OH}$  to  $\text{NH}_2+\text{NO}=\text{N}_2+\text{H}_2\text{O}$
  - $\text{H}_2\text{NO}+\text{O}_2=\text{HNO}+\text{HO}_2$  vs.  $\text{H}_2\text{NO}+\text{NH}_2=\text{HNO}+\text{NH}_3$  increasingly promote and inhibit ignition, respectively



# IDTs of NH<sub>3</sub> – Equivalence Ratio Effect

- Compared with existing kinetic model at equivalence ratios of 0.5, 1.0, and 2.0
- IDTs not very sensitive to equivalence ratio
- Most models predict increased IDT with increased equivalence ratio
- Models by Mathieu and Petersen (2015) and Mevel et al. (2009) show the strongest dependence on equivalence ratio
- Model by Konnov et al. (2001) predicts the least significant equivalence ratio effect



O. Mathieu, E.L. Petersen, *Combust. Flame* 162 (2015) 554-570.

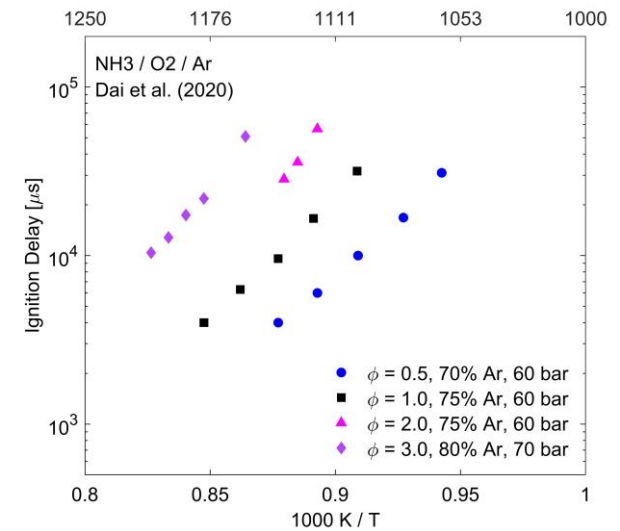
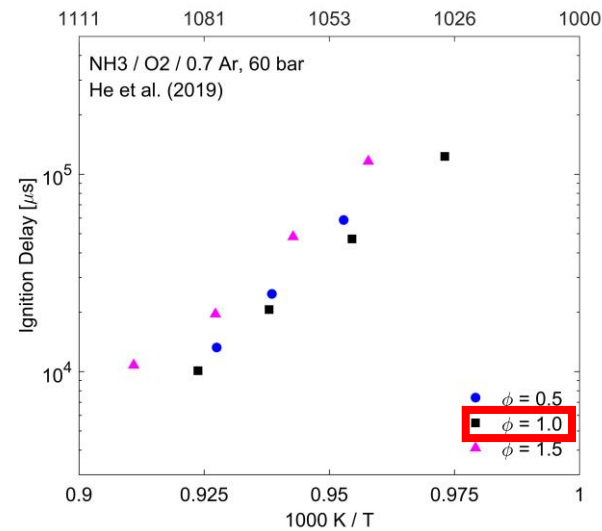
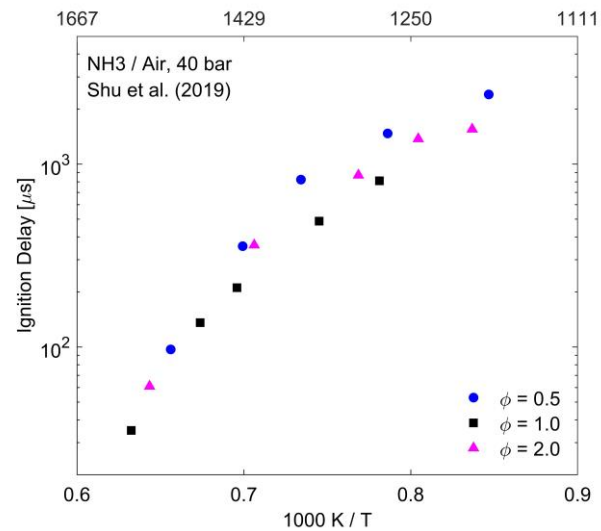
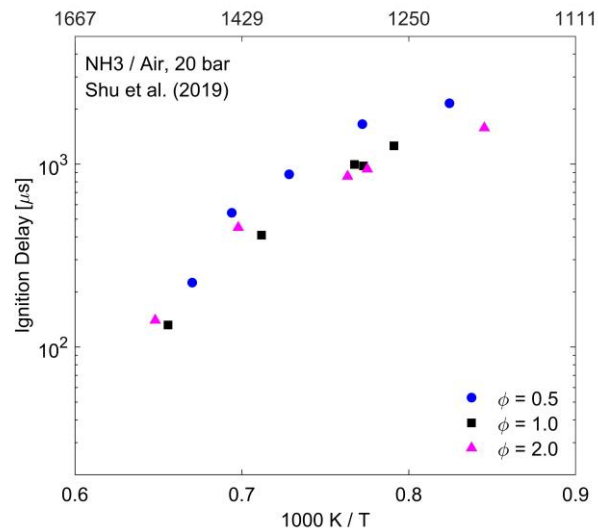
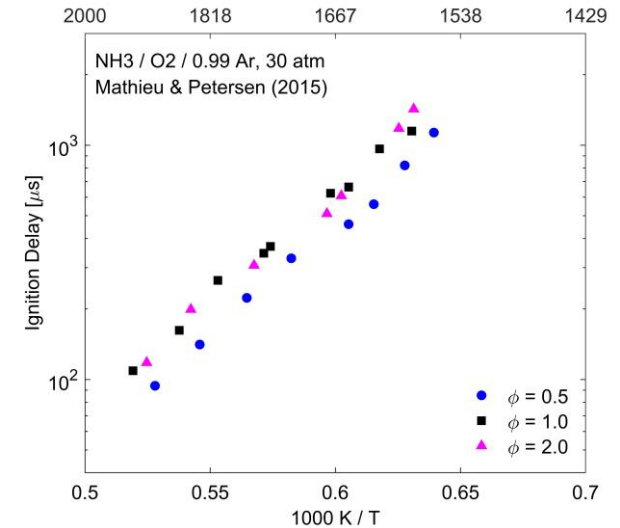
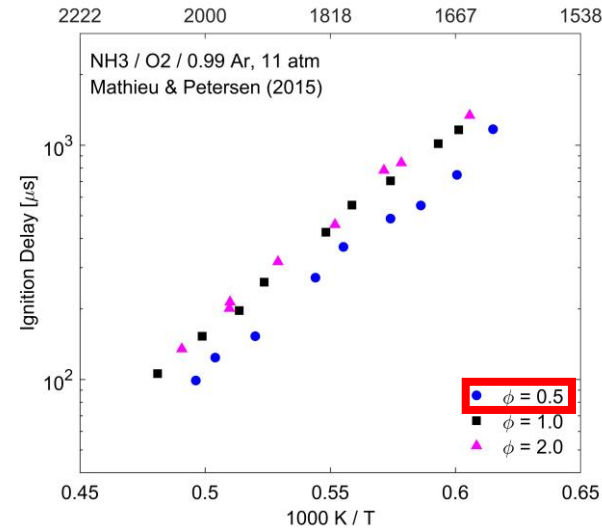
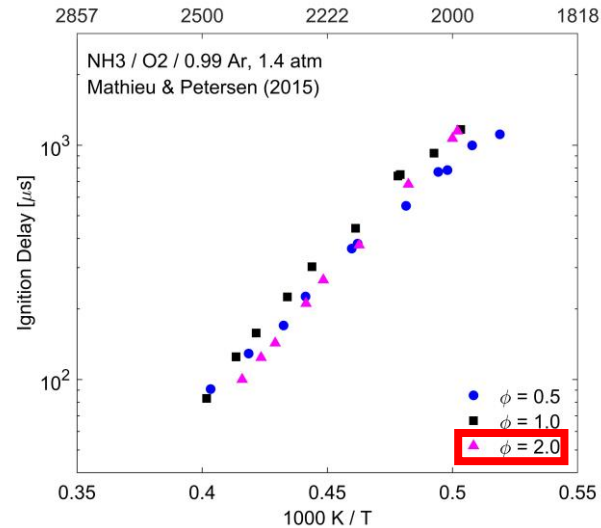
R. Mevel, S. Javoy, F. Lafosse, N. Chaumeix, G. Dupre, C.E. Paillard, *Proc. Combust. Inst.* 32 (2009) 359-366.

A.A. Konnov, J. De Ruyck, *Combust. Sci. Technol.* 168 (2001) 1-46.

B. Mei, X. Zhang, S. Ma, M. Cui, H. Guo, Z. Cao, Y. Li, *Combust. Flame* 210 (2019) 236-246.

# IDTs of NH<sub>3</sub> – Equivalence Ratio Effect from literature

- IDTs in literature do not show consistent trend
- Effect of equivalence ratio appears to depend on mixture and conditions
- No clear trend observed



O. Mathieu, E.L. Petersen, *Combust. Flame* 162 (2015) 554-570.

B. Shu, S.K. Vallabhuni, X. He, G. Issayev, K. Moshhammer, A. Farooq, R.X. Fernandes, *Proc. Combust. Inst.* 37 (2019) 205-211.

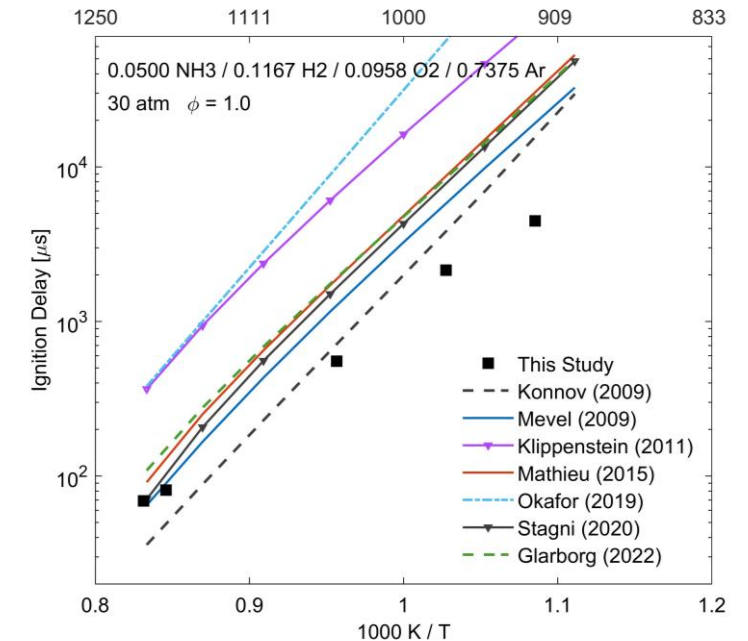
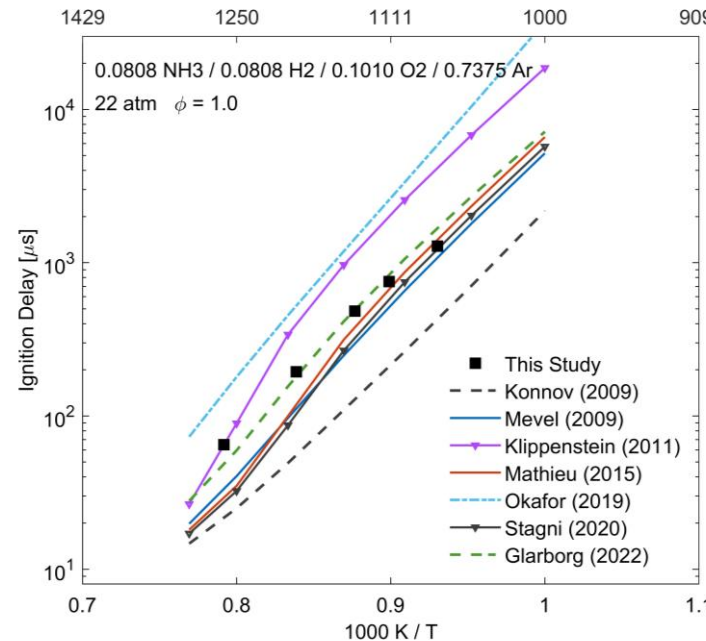
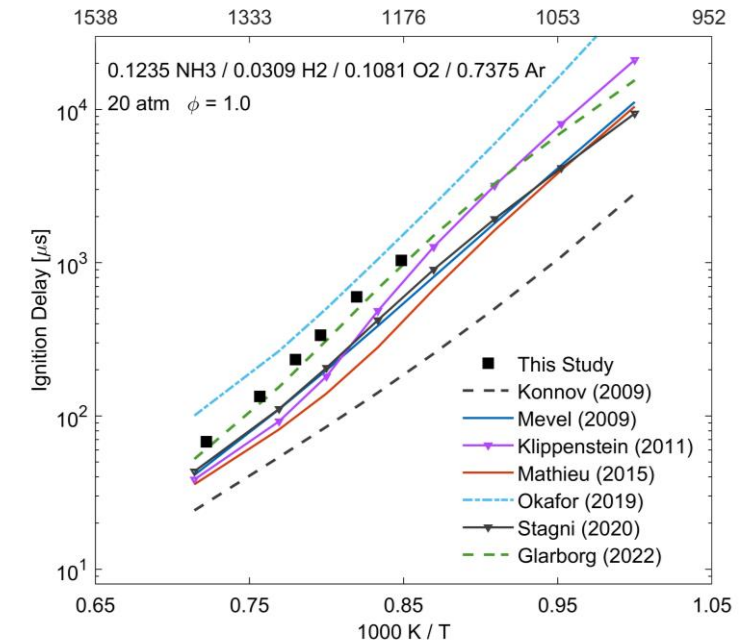
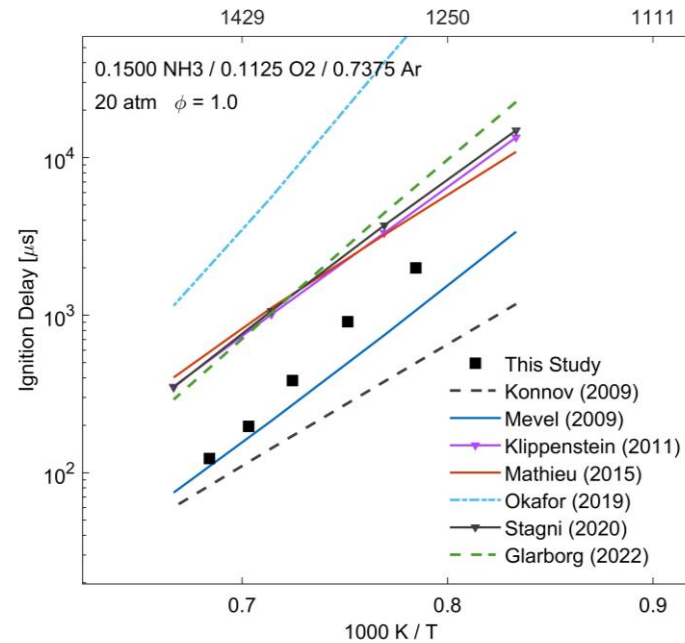
X. He, B. Shu, D. Nascimento, K. Moshhammer, M. Costa, R.X. Fernandes, *Combust. Flame* 206 (2019) 189-200.

L. Dai, S. Gersen, P. Glarborg, H. Levinsky, A. Mokhov, *Combust. Flame* 215 (2020) 134-144.



# IDTs of H<sub>2</sub>/NH<sub>3</sub>

- IDTs with 0 to 70% hydrogen in fuel
- Compared with different kinetic models
  - Good agreement with Glarborg (2022) at 20% and 50% hydrogen in fuel
  - Most models over-predict IDT at 70% hydrogen in fuel; over-prediction increases towards lower temperatures (well-known H<sub>2</sub> IDT issue at low T)
- Need 2D optical access to examine ignition quality with H<sub>2</sub> addition

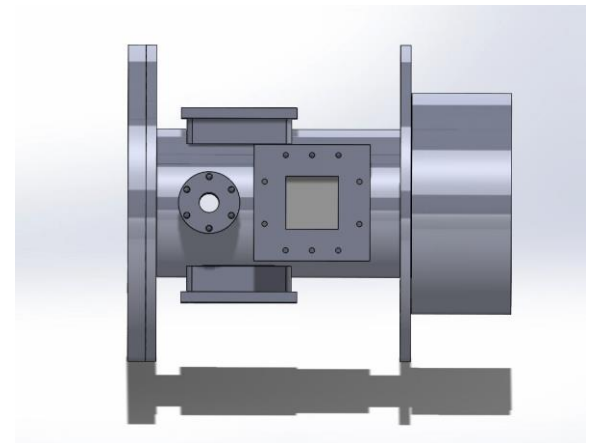




# Task 4: Measurement of laminar flame speeds of hydrogen containing fuels at high preheating conditions



- 6-in-diameter optical access from end-wall and effective 3-in-by-7-in from side-wall
- 2D optical access allows flame speed measurements and examination of ignition uniformity for H<sub>2</sub>
- Currently validating performance with IDT experiments and setting up instrumentation for optical measurements



Complete Side View

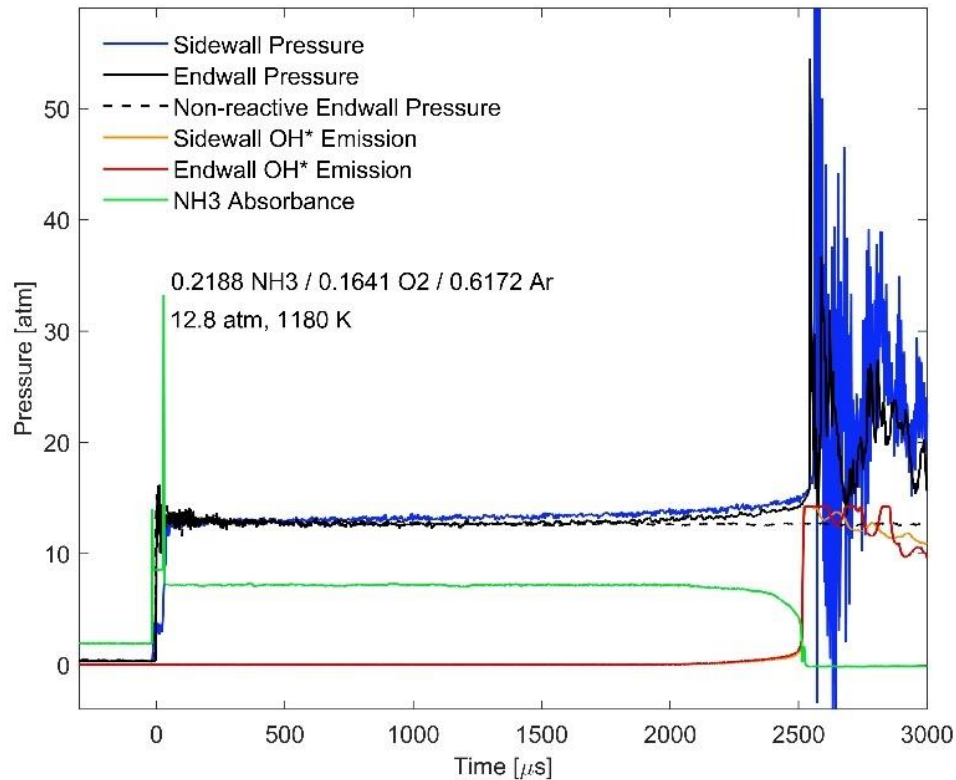
# Conclusions from Ignition Study

- $\text{NH}_3$  IDTs obtained at different fuel concentrations. No kinetic model can predict the IDTs at all conditions.
- $\text{NH}_3$  IDTs show no clear trend on equivalence ratio dependence
- Discrepancy on  $\text{NH}_3/\text{H}_2$  IDTs is observed and being investigated

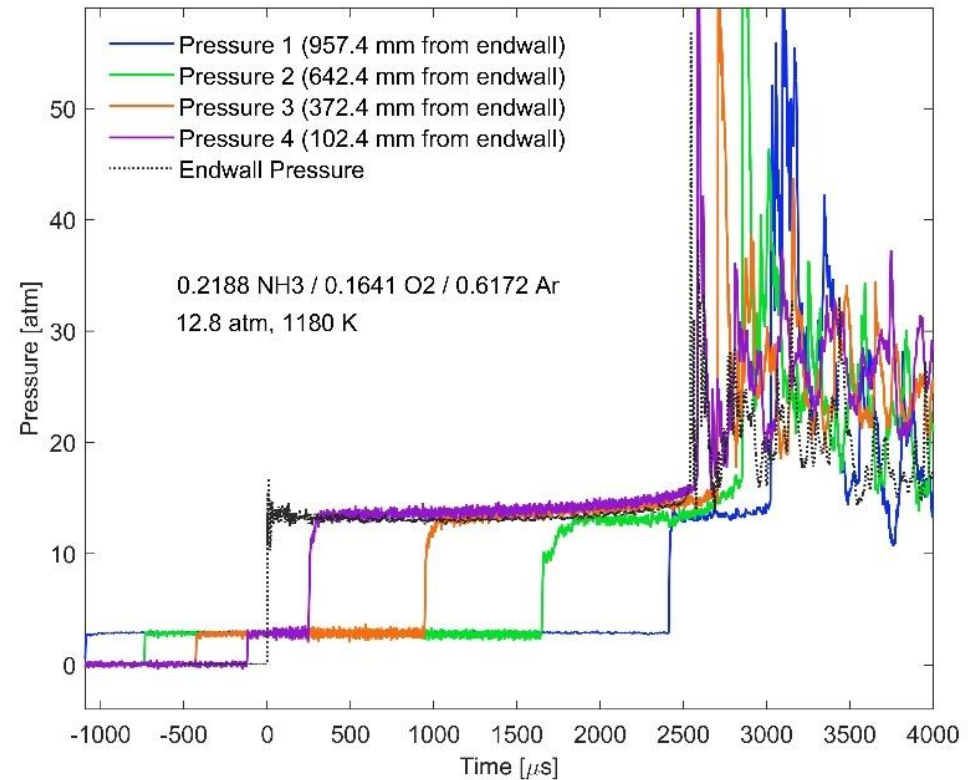
Thank you & Questions?

# Pressure Traces during IDT Measurement

- Clean Pressure traces, no sign of inhomogeneous ignitions



Signals at the measurement section from a typical experiment in this study, for a stoichiometric NH<sub>3</sub>/O<sub>2</sub>/Ar mixture with 22% fuel concentration at 12.8 atm and 1180 K. OH\* emission and NH<sub>3</sub> absorbance signals are of arbitrary units.



Pressure signals near the measurement section from a typical experiment in this study, for a stoichiometric NH<sub>3</sub>/O<sub>2</sub>/Ar mixture with 22% fuel concentration at 12.8 atm and 1180 K