

Databasing information for existing pipeline infrastructure / Dual-use infrastructure – technical considerations for dual-use transport of LPG/LNG/H₂/CO₂/Ammonia

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As we see it, there are two primary questions regarding dual purposing:

- Can existing natural gas (NG) pipelines be repurposed for hydrogen, CO₂ and/or ammonia?
- Can liquefied natural gas (LNG), liquid petroleum gas (LPG), and ammonia infrastructure be repurposed for CO₂ and hydrogen (or ammonia)?

Technical considerations based on:

- Compatibility with infrastructure materials and component performance (includes impurities)
- Desired physical state (liquid or gas)



Fluid properties relevant to compatibility

| Gas/fluid type | Structure | Polarity | Solubility in water at 20°C and 1 atm (g/kg _{water}) |
|----------------|--|--------------|---|
| Hydrogen | н—н | Nonpolar | 0.0016 |
| Carbon dioxide | 0=C=0 | Polar | 1.69 |
| Ammonia | н_м—н И | Highly polar | 529.0 |
| Methane | Н Н—С—Н Н | Nonpolar | 0.023 |
| Propane | ңн ^н ,с, ^с ,с,н нннн | Nonpolar | 0.0047 (at 0°C) |

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Quickest, most efficient and cost-effective means of transporting fuels and/or CO₂ is to repurpose existing natural gas (NG) pipeline infrastructure

- Over 300,000 miles of natural gas transmission pipelines and 1800 compressor stations in the continental U. S.
- These pipes can have diameters up to 1.2 m (48 in.) and are composed almost entirely of low-carbon steels, though some cast, ductile, and wrought irons are still in use.
- The gauge pressure in these systems can range from 1.4 to 10.3 MPa (200 psi to 1,500 psi). For most transmission systems the line pressure is relatively constant
- Pipes are designed with safety factors of 1.25 or 1.5 depending on the population density

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Previous and current repurposing compatibility studies have focused solely on the piping sections

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- API steel grades: A25, A, B, X42, X52, X60, XC60, X70, X80 and X100
- Seamless or welded
- Many pipes have an inner epoxy coating to reduce friction and wear (and also corrosion)
- Numerous compatibility studies with hydrogen



- Consists of compressors, odorant inputs, gages, valves, meters, regulators, etc
- Many different components containing a variety of ferrous and nonferrous metals and polymers (plastics and elastomers)
- Very little compatibility studies with hydrogen

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Hydrogen transport in NG pipelines

- Hawai'iGas has been transporting 10-12% H₂ in its pipelines for the past 50 years
- Several field demonstrations are being established by pipeline manufacturers to evaluate H₂ blends with NG
- General assessment that blends less than 10% are suitable for most pipeline steel grades. Lower yield strength steels are mandated:
 - ASME B31.12 recommends maximum grade of X52
 - API Guidelines call for a maximum yield strength of 827 MPa (120ksi)
- Impurities can accelerate stress corrosion cracking
- Impacts of impurities on polymer performances at NG pipeline conditions is not well known
- End use systems may not operate properly with hydrogen blends. This is also true for infrastructure components (e.g. flow meters)



The majority of NG pipelines operate under conditions slightly below the CO₂ critical point

- Typical range is 800-1,000psi, but some systems operate up to 1,500psi (103bar) and down to 200psi (13.8bar)
- Gas or liquid phases fluctuate depending on temperature
- Existing CO₂ pipelines operate above 1,900psi (130bar)
- Not a region that is well-studied, especially for addressing impurity impacts
- Netherlands repurposed a 26 inch, 51-mile NG pipeline for CO₂ transport as a gas at 101-304psi (7-21bar) at a rate of 300Mt/year

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What we know about CO₂ compatibility

For Metals: Extensive body of literature exploring CO₂ corrosion of pipeline steels.

- Corrosion requires the presence of aqueous condensate (carbonic acid is formed)
- Studies have shown that H₂O, H₂S, SOx and O₂ can accelerate corrosion. Less well understood are NOx contributions, which is a primary product of combustion. Note that NOx will form nitric and/or nitrous acid with water.

For Polymers: Much less studied, especially at conditions accompanying NG pipelines.

- Known solvent for some polymers
- High swell observed for some elastomers
- Limited range of polymers studied

Important knowledge gaps include:

- Contribution of impurities (including NOx) to aqueous corrosion
- Corrosivity in liquid and gaseous CO₂ at conditions below the critical point
- Aluminum, stainless steels, zinc, brass, and other steel grades that are used extensively in compressor/regulator stations
- Many compressor station polymers not evaluated, especially at pipeline conditions



NG Pipelines: Ammonia would primarily exist as liquid

- Ammonia is widely and safely transported *in systems specifically designed for handling ammonia*.
- NG pipeline steels are suitable for ammonia
- Oxygen impurities can promote stress corrosion cracking
- Ammonia transport via NG pipeline sections is feasible, but compressor stations are not suitable. Fluorocarbons (FKMs) and some NBRs are not suitable.

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Repurposing LNG or LPG infrastructure for sCO₂, hydrogen, or ammonia.



• LNG/LPG conditions

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- LNG tanks are maintained at -161.5°C (-260°F) and 0.34 bar (5 psi)
- LPG is typically stored at pressures up to 1.7 bar (25 psi) and ambient temperature
- H₂: Liquefying and shipping hydrogen requires chilling it to -253 degrees Celsius. Nearly 100 Celsius degrees colder than temps needed to transport LNG
- **sCO₂**: CO₂ cannot liquefy in LNG and LPG tanks (pressure too low)
- Ammonia: will exist as liquid in LNG and LPG systems
- Other considerations include inherent impurities and the high solvency of sCO₂ with many elastomers
- Bottom Line: Dual-purpose infrastructure will need to be designed specifically for LiqH₂ and sCO₂.

CO₂ Phase Diagram Supercritical Critical Di fluid point 72.9 atm Liquid Pressure (not to scale) 5.1 atm Triple Solid point 1 atm Gas -56.7 °C 31 °C -78.5 °C Temperature (not to scale)

Repurposing ammonia infrastructure for sCO₂ or hydrogen

- Ammonia is typically stored/transported as a liquid under pressure to 2bar (29psi) at atmospheric temperature
- Pressures are too low for dense CO₂ and liquid H₂
- Tanks usually composed of carbon and stainless steels. Suitability uncertain for use with hydrogen/NG blends
- Metal components are probably suitable for use with gaseous CO₂, polymer seals need to be evaluated
- Bottom line: more structural operational information needed







Properties of interest include:



- Polymers
 - mass/volume/hardness
 - Compression set (elastomers only)-
 - Spalling/delamination of epoxy coatings
 - DMA
 - GC-MS/FTIR/Raman/X-ray
 - Tensile strength
- Metals

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- Mass loss
- Corrosion characterization
- Surface chemistry
- Yield strength







Temperature



We recently completed a study examining the materials compatibility of repurposing NG infrastructure for H_2 , CO_2 and/or NH_3 transport

- Effort supported by Tim Reinhardt and Evan Frye (Division of Methane Mitigation Technologies)
- Outcomes:

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- Material compatibility assessment based on available literature studies
 - Only study that looked at compressor/regulator stations
 - Identified over 100 components and materials of construction
- GIS analysis of pipelines
- Identified key gaps
- Kass, M. D., Keiser, J. M. et al., "Assessing Compatibility of Natural Gas Pipeline Materials with Hydrogen, CO2, and Ammonia," *J. Pipeline Syst. Eng. Pract.*, 2023, 14(2): 04023007

Subset of tabulated results for polymers (note similar table compiled for metals)

| Component | Sub component | Material | Hydrogen Blend | sCO2 | Ammonia | Comments |
|-------------------------|-----------------------------|---|-------------------|--------------|--------------|---|
| Ball valve | Valve seat | PTFE (Teflon™) | Suitable | Incompatible | Suitable | |
| | Stem o-ring | NBR | Suitable | Incompatible | Questionable | Depends on the NBR grade |
| | Stem packing | PTFE (Teflon™) | Suitable | Incompatible | Suitable | |
| afety shut valve | Diaphragm | NBR with nylon reinforcement | Suitable | Incompatible | Questionable | Depends on the NBR grade |
| | Valve heat/seat | Polyurethane | Suitable | Incompatible | Incompatible | |
| | o-rings | NBR | Suitable | Incompatible | Questionable | |
| | Seal | Graphite | Suitable | Suitable | Suitable | not a polymer |
| Appliance regulators | Diaphragm and valve seat | NBR with and without nylon reinforcement | Suitable | Incompatible | Questionable | Depends on the NBR grade |
| | Valve head | NBR | Suitable | Incompatible | Questionable | Depends on the NBR grade |
| | Cap and assembly | Molded plastic | Suitable | Suitable | Suitable | Plastic wasn't specified, but should be ok based on survey |
| | o-rings | NBR | Suitable | Incompatible | Questionable | Depends on the NBR grade |
| | Valve seat | NBR | Suitable | Incompatible | Questionable | Depends on the NBR grade |
| | | Silicone | Suitable | Suitable | Questionable | Ammonia suitability depends on gas temperature |
| | Lower diaphragm plate | Polyester | Suitable | Suitable | Suitable | |
| | Diaphragm | NBR | Suitable | Incompatible | Questionable | Depends on the NBR grade |
| | | Nylon | Suitable | Suitable | Suitable | |
| | Ventualvalaset | Neoprene | Suitable | Incompatible | Suitable | |
| | vent valve/seat | Delrin (acetal)/NBR | Suitable | Incompatible | Questionable | Depends on the NBR grade |
| | Adjustment Ferrule | Delrin (acetal) | Suitable | Suitable | Suitable | |
| | Seal cap | Polyethylene | Suitable | Suitable | Suitable | |

Key Takeaways

- Existing NG pipelines (excluding compressor stations) likely suitable for low blend levels of hydrogen (in NG), gaseous CO₂, and ammonia
- Existing LNG/LPG infrastructure not currently viable for liquid hydrogen or sCO₂
- Impurities will be important, especially if switching between fluid chemistries
- Current compressor/regulator stations are unsuitable for use with sCO₂ and ammonia
- Need to consider not just material but component performance
- Obtaining precise material information can be challenging



Next Steps

- Obtain more precise operational and material information from industry
- Determine whether existing pipeline safety factors are acceptable
- Improved understanding of impurities and their contributions to hydrogen corrosion, especially at NG pipeline conditions
- Improved understanding of impurities and their contributions to CO₂ corrosion, especially at NG pipeline conditions
- Address key data gaps associated with sCO₂ and polymer compatibility

