Development of Stable Solid Oxide Electrolysis Cells for Low-cost Hydrogen Production

Contract Number: DE-FE0032105

P.I.: S. Elangovan, **Co-P.I.:** Jenna Pike Tyler Hafen , Dennis Larsen, Joseph Hartvigsen, OxEon Energy Team Prof. Bilge Yildiz Group, MIT Dr. Olga Marina, PNNL

Support: NASA, DOE



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DOE Project Manager: Drew O'Connell



Beyond Current Potential

Company Background



Utah, USA R&D/Manufacturing - 2017

- Office, laboratory, and manufacturing facility (24,000 ft²)
- NASA, DOE, DOD and commercial contracts
- Tape casting, cell and stack production, and testing
- End-to-end power to synfuels pilot plant





Solid Oxide Fuel Cell and Electrolysis Stacks

- Longest running solid oxide fuel cell & electrolysis group in world
- Only flight qualified, TRL 9 SOEC unit with active NASA demonstration on Mars
- 30kW/10kW and 20kW/10kW reversible SOC system test programs

Fuel Reformation and Generation

- Plasma Reformer H₂ or syngas from flare gas; digester gas conversion; clean-up bio-gasification
- Fischer-Tropsch Reactors Modular design for sustainable fuel production from H₂ and syngas



Project Objective



- A solid oxide electrolysis cell (SOEC) stack in a laboratory test bed
 - show improved performance over baseline stacks
 - robustness,
 - reliability,
 - endurance,
 - hydrogen purity, and
 - produce hydrogen at elevated pressure of 2 to 3 bar.

Project Goals



Improved performance over baseline

- Reproducibility and lower polarization by electrode modification
- Long term stability
 - projected lifetime of > 40,000 hours

Robustness

- Capability for thermal cycling of a stack
- Redox cycling of fuel electrode in a stack
- Production of hydrogen at elevated pressure

Robustness

Redox Tolerant Fuel Electrode - Background



Mars OXygen ISRU Experiment aka "The Oxygenator"

CO_2 to $CO + O_2$

First of any kind of demonstration of **In-Situ Resource Utilization (ISRU)** technologies to enable propellant and consumable oxygen production from the Martian atmosphere - Currently onboard Perseverance Rover

MOXIE is a ~0.5% scale prototype of expected final O_2 production rate



Flight Qualification



Baseline Performance

• 21 consecutive stacks built with aerospace quality standards and traceability having a maximum baseline performance of 1.6 ohm-cm² dry CO₂ and 99.9%+ O₂ purity

Cycling Performance

- 3 stacks with 21 cycles of identical test procedure having varying cycle-to-cycle flow rates and final cycle averages of 10.11 g O₂/hr production and 99.8% purity Targets exceeded
- 1 stack to cycle 61 with >99.6% purity at a controlled production rate of 6 g/hr at 55g/hr feed

Structural Stability Testing

- No leak or significant performance change after 10kN crush testing
- Stacks tested to 25kN force with no crossover or external leakage
- Load to failure required 62.2kN (>30 margin of safety from design)

Shock/Vibe Testing

- Stacks vibrated at JPL and post vibe tested at OxEon
- No leak or significant performance change post vibe!
- No leak after shock testing, no significant performance change!

Cryo-Cycling

- Vibe stack cryo-cycled to -40°C (40 cycles), -55°C (3 cycles), -65°C
- Stack performance and purity unchanged in operational cycling post test





Cathode Challenge: Oxidation in dry CO₂





• Early MOXIE Test Stack:

- 15 operational cycles full thermal cycle with 120 min operation on dry CO₂
- Dry $CO_2 \rightarrow O_2$ production ~12% of initial

Dramatic degradation resulted from progressive oxidation front

Oxidation of Ni to NiO causes ~24% vol expansion, and in this case, irreversible damage to the electrode & current collector

MOXIE implemented recycle of produced CO to prevent cathode oxidation

Flight Test Success - First Ever ISRU Demonstration!



First 100 Sols!





Sol 5 "Aliveness Test" Mon Feb 22, 2021

- Sol 13 First run with Run Control Table (RCT)
- Sol 14 "Health Check" of heaters and compressor
- Sol 59-60 April 20, First Oxygen
 - Produced 5.4 g O₂ pre-dawn, peak rate of 6 g/h (2 A current)
- Sol 81 May 12, 2nd Oxygen
 - Nighttime (early AM) operation
 - Produced 7 g O_2 , 8 g/h peak

Sol 100 May 31 3rd Oxygen

- Mid-day operation with lower atmospheric density
- Extended 8 g/h operation

13 Successful high purity (>99.6%) oxygen production runs on Mars to-date



First Run

NASA Support through JPL

Redox Tolerance for CO₂ Electrolysis (NASA SBIR)

Ni-based electrode

Button Cell: Redox Cycling and Long-Term Stability for CO₂ Electrolysis **()**×Eon



Down-selected Composition Ni-Ceria based (N85)

- Full Redox Cycle = 12 hours off load with dry CO₂ only feed
- Kept at 800 °C to nearly fully oxidize the cathode material (Ni metal → Nickeloxide)
- Load is reapplied
- No external reducing gas

STK-033 Partial and Full Redox Cycles



- Stack: Short (20 min) and long (12 hrs) exposure to CO₂
- Application of voltage full recovery of performance

Redox test #1 Pacific









- 0.35 A/cm² at 1.3V initially with 20%H₂;
- Current increases after each redox cycle, then drops; 3 cycles completed: 1h, 16h, 16h

kW class stack: Redox Cycle in Steam Electrolysis



Stack test at Colorado School of Mines:

Stack in Lunar vacuum

Power supply problem

- Recycle H₂ stopped
- Current to zero
- Voltage dropped
- Restored power supply
- Stack performance recovery

NASA Funded (Tipping Point Project)

Electrode Improvement - DE-FE0032105

Focus: Address known/suspected degradation mechanisms

Integrated Approach to Addressing SOEC Degradation



No.	Degradation mechanism	Effect	SOEC component	Activity	Project Support
1	Cr transport from interconnect	Poisons active electrochemical sites	O ₂ electrode	Poisoning Effect (PNNL) Spinel Coating	DOE/NETL NASA Phase II-E
2	Perovskite composition instability over time	Catalytically inactive and electrically resistive grains/ Non-catalytic secondary phases	O ₂ electrode / current collector	Composition modification	DOE/NETL NASA Phase II E
3	SiO ₂ migration from seal	Contaminates electrodes	O ₂ electrode Fuel electrode	Poisoning Effect (PNNL)	DOE/ NETL NASA Phase II E
4	Cation diffusion	Formation of more resistive phases	Electrolyte CeO ₂ barrier	Process modification	DOE/ NETL NASA Phase II E

Oxygen Electrode Interface Improvement



SDC Barrier layer

- Improve sinterability to lower sintering temperature
- Eliminate interface reaction between ceria and zirconia

Standard SDC

x5,000

5 Mm

15kU





This Project + NASA SBIR

Oxygen Electrode Interface Improvement



SDC Barrier layer

- Sr migration occurs at discontinuous and porous regions in the barrier layer
- "Ideal" sintering temperature balances densification, interface reaction, & manufacturability (co-sintering with electrodes)



Discontinuous >150 °C Reduction barrier layer.



This Project + NASA SBIR



Eliminate Sr in current collector layer by replacing LSCF with Sr-free perovskite mixed with Ag-alloy

- Ohmic resistance close to expected based on electrolyte thickness ۲
- Polarization resistance indicate LSCF layer is not required for water splitting ٠

LCAP10 current collector shows good bonding to electrode layer



	Current	ASR ($\Omega \cdot cm^2$)		
Electrode	collector met Vol% in cermet	Total	Ohmic	Polarizatio n
Sr-free	10%	0.79	0.66	0.13
Sr-free	30%	0.66	0.59	0.08
Sr-free	50%	0.62	0.50	0.12
LSCF	50%	0.71	0.52	0.19

Sr-free electrode

Air electrode barrier

Electrolyte

This Project



Sr-stabilization in oxygen electrode – multiple approaches

- A-site deficiency in the LSCF perovskite (A_{1-x}BO₃)
- Additional dopant in LSCF
- Using LSCF-composite current collector
- LSCF surface treatment

Doped LSCF CC layer cell performance stabilizes after initial degradation, compared with LSCF CC



Fuel Electrode / Current Collector



Reduce current collector cost by reducing Ag-alloy concentration (15%, 30%, 50%)

- Reduce cost but maintain performance and redox tolerance! •
- Button cell performance is comparable with lower Ag-alloy concentration processing ulletmodifications will reduce Ag-agglomeration and improve interconnected metallic network

50% Ag-alloy



This Project + NASA SBIR

30% Ag-alloy



15% Ag-alloy



15% Ag-alloy current collector is not redox tolerant

- Steam redox cycle tests show lower Ag-alloy current collector is not redox tolerant
- Removing H₂ from the feed (N₂ and Steam only) resulted in high degradation rate
- Full redox cycle resulted in large performance drop. Cell did not recover.
- Q: better distribution or more Ag??



500



15% Ag-alloy Current Collector Redox Testing 800 C, 1.3 V hold





Historical 10-cell stack tests show incremental improvements in SOFC operation



Stack	STK-37	STK-57	STK-65
Program	INL 30 kW	Stone Edge Farm	NASA SBIR II-E
Fuel Electrode	Ni-SDC	Ni-SDC	Redox tolerant
Fuel Electrode Infiltrant	Yes	Modified	modified
Air Electrode barrier layer	SDC	Mod-SDC	Mod-SDC
Air Electrode I	Sr-free+SDC	Sr-free+SDC	Sr-free+SDC
Air Electrode II	LSCF	LSCF+SDC	LSCF+SDC
Air Electrode Infiltrant	Yes	Surf treat + infilt	Surf treat + infilt

This Project + NASA SBIR + INL 30 kW + Stone Edge Farm

Materials Validation and Performance Mapping

Button cells will be tested in steam electrolysis range of temperature, steam utilization, and cell voltages to assess performance and stability

- Cathodic overpotential modifies the surface of LSCF-SDC electrode and improves the chemical and electrochemical stability
 - Main objective: Suppression of Sr segregation





Effect of Impurities on Fuel Electrode Performance

Effect of Mn impurities on the fuel electrode in SOEC mode PNNL – button cell testing with stainless steel shows no Mn in EDS analysis Tested for 168 hour at 800°C at 0.7V Cell was degrading continuously No Mn detected in Ni by SEM/EDS



Pacific

Tested in 100% H₂ for 168 hour at 800°C at 0.7V: H₂ electrode with SS

Northwest • Electrolyte to H electrode interface – BSE image



IVIn signal	
H electrode	Atomic %
0	34.99
Ni	31.32
Zr	16.71
Ce	7.83
Mg	2.92
AI	2.51
Υ	1.45
Cu	1.39
Sm	0.64
Sc	0.17
Са	0.08
Pr	0.00
Total	100.00





Effect of Impurities on Fuel Electrode Performance



Effect of **Si** impurities on the **fuel electrode** in SOEC mode

- Button cell on steam electrolysis with RO filtered water has measured better stability (BC-104-10).
 - DI water from supplier contained 0.3 mg/L SiO₂
 - No SiO₂ detected after Reverse Osmosis filtration
- Improved stability in SOEC when steam is produced from RO-treated water.
- Distilled water contains trace SiO₂



Effect of Impurities on Oxygen Electrode Performance



Improve Cr-barrier interconnect coating

- Optimized coating dispersant, powder loading, and viscosity to increase barrier layer density
- High-density coating regions near the CFY substrate with porosity further up the coating thickness.
- Next steps characterization of electrodes placed in contact with coated interconnect coupons at 800 C



Two different dispersants selected

NASA SBIR

Oxygen Production (Seal Validation)

High Purity O₂ on Mars

• External to stack Mars ambient ~ 7 millibar

Oxygen production at pressure (steam electrolysis test at CSM in vacuum chamber)

- Stack in vacuum
- H₂ production at 1 bar
- O₂ production up to 3.6 bar via electrochemical compression

This Project

- Short stack testing in pressurized test stand
- Test stand modifications are underway This Project



MOXIE scale stack (left) and demonstration system scale stack (right)





- Cr poisoning tests were performed 3 times, each time using 2 cells
 - Dry air, 800°C, ~ 1 ppb Cr
 - Wet air, 800°C , ~ 1 ppm Cr





Not growing in time in clean air

Cr in getter post-test



Yet no Cr detected in oxygen electrode by SEM/EDS



Test Description



Equilibrium Cr Content in Air







Downstream Filter



Chromia Pellet



Next Task: Pressurized SOEC Operation



Pacific

Northwest

- High-pressure rig was built in 2022 but had to relocate to a different building
- Faced building operation and infrastructure issues and delays preventing H₂ usage; issues have been resolved
- Initiated pressure regulators testing and troubleshooting the system
- Aiming to start cell testing next week

Pressurized cell test rig

Summary



- Multiple projects to provide complementary scope/results
- Redox tolerance validated for steam electrolysis
 - Oxidized Ni electrode recovery without the need for hydrogen in inlet
 - Modifications for improvements will be validated

Electrode materials modification - validation in progress

- Composition to improve thermochemical stability
- Surface modification for improving catalytic property
- Investigation of poisoning effect ongoing
- Pressurized tests: steam electrolysis
 - button cells to begin shortly
 - Stack in Year 2

Thank you

S Elangovan elango@OxEonEnergy.com



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info@oxeonenergy.com +1-801-677-300 oxeonenergy.com



PLASMA

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