

Rational Development of Novel Metal-Organic Polyhedra-based Membranes for CO₂ Capture

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Project Overview

- Funding (DOE \$2,857,896 and Cost Share \$975,484)
- Overall Project: 7/1/2019 – 6/30/2024
- Project manager: Krista Hill
- Overall Project Objectives
 - Rationally develop solubility-selective mixed matrix **materials** comprising polar rubbery polymers and metal organic polyhedra (MOPs);
 - Develop thin film composite **membranes** achieving high CO₂ permeance (3000 GPU) and high CO₂/N₂ selectivity (50);
 - Demonstrate separation **performance** and stability with raw flue gas at NCCC; and
 - Perform **techno-economic analysis** on the membrane processes.

BPs and Team Members

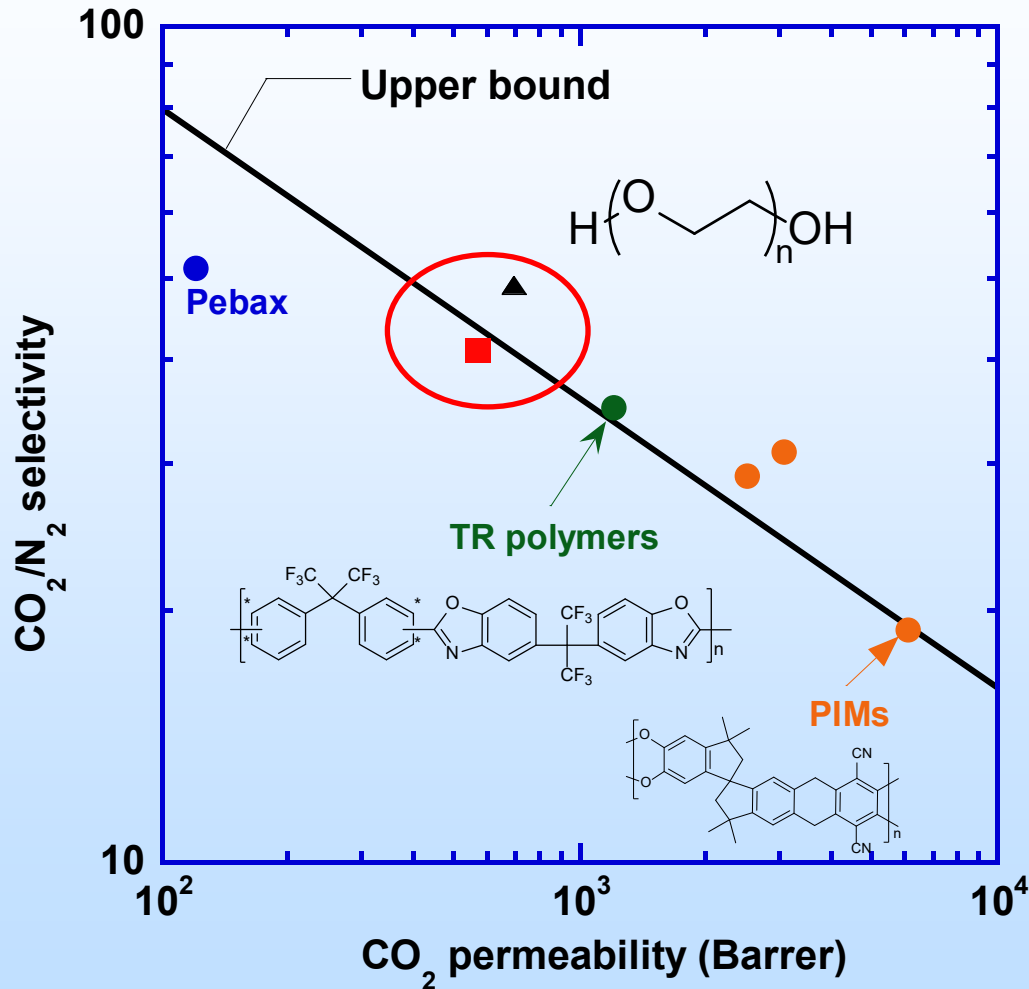
BP 1: Rationally design and prepare freestanding **mixed matrix films** with CO₂ permeability of 1000 Barrer and CO₂/N₂ selectivity of 75 and CO₂/O₂ selectivity of 25 **(7/19 - 6/21)**

BP 2: Prepare and optimize thin film **mixed matrix composite membranes** with CO₂ permeance of 3000 GPU and CO₂/N₂ selectivity of 50 and CO₂/O₂ selectivity of 20 **(7/21 - 12/23)**

BP 3: Prepare bench-scale spiral-wound membrane **modules** and perform **field tests with real flue gas** at NCCC; and complete the **techno-economic analysis (1/24 - 6/24)**

Members	Role
UB	Materials development
Caltech	Computational simulation
RPI	Polymer synthesis scale-up
MTR	Membrane development & field test
Trimeric	TEA
NCCC	Host site

State-of-the-Art Polymers for CO₂/N₂ Separation

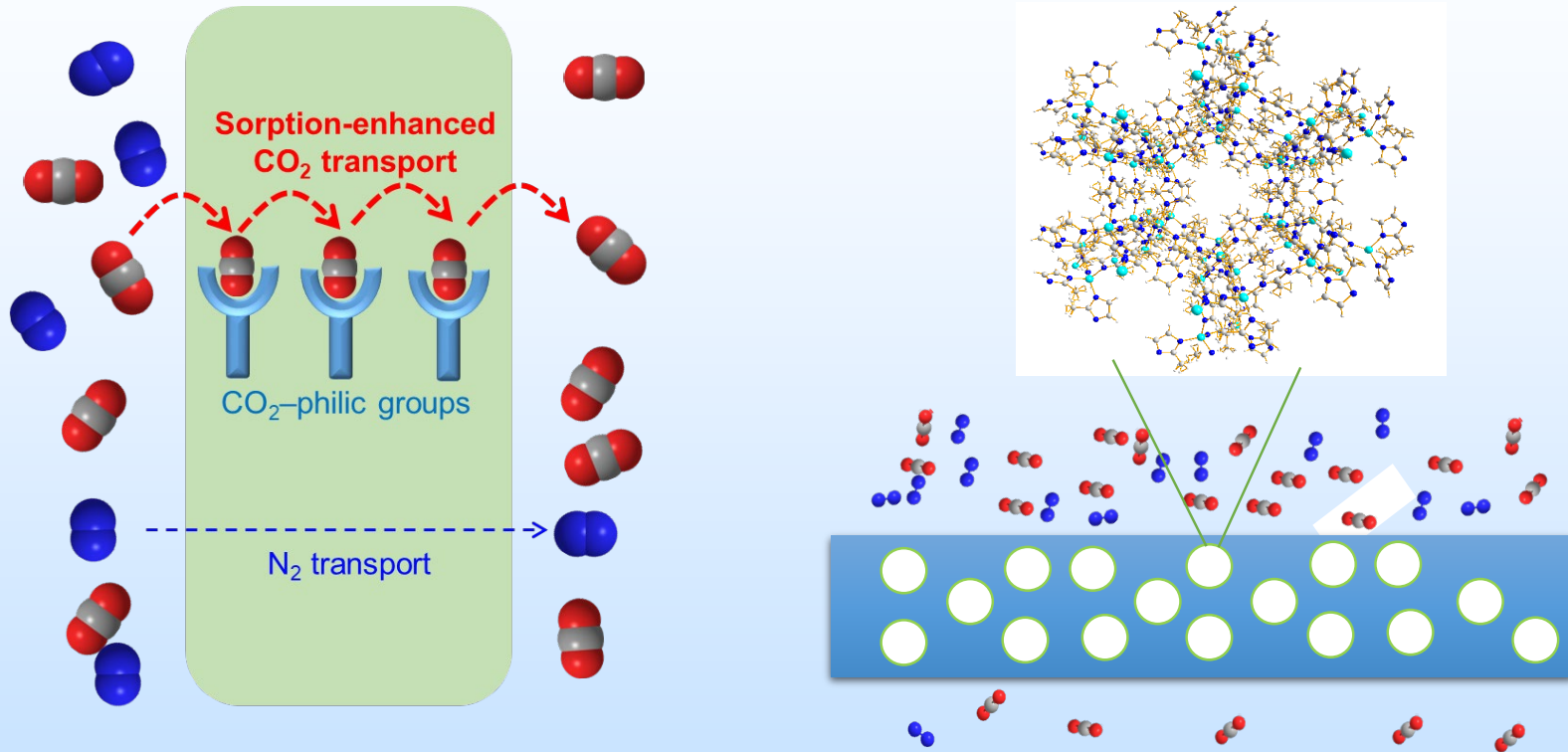


$$\alpha = \frac{P_{\text{CO}_2}}{P_{\text{N}_2}} = \frac{S_{\text{CO}_2}}{S_{\text{N}_2}} \times \frac{D_{\text{CO}_2}}{D_{\text{N}_2}}$$

	T_C (K)	D_A (Å)	V_C (cm ³ /mol)
CO ₂	304	3.3	93.9
N ₂	126	3.64	89.8

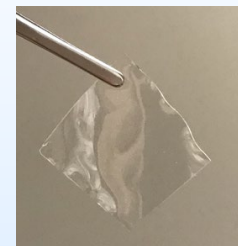
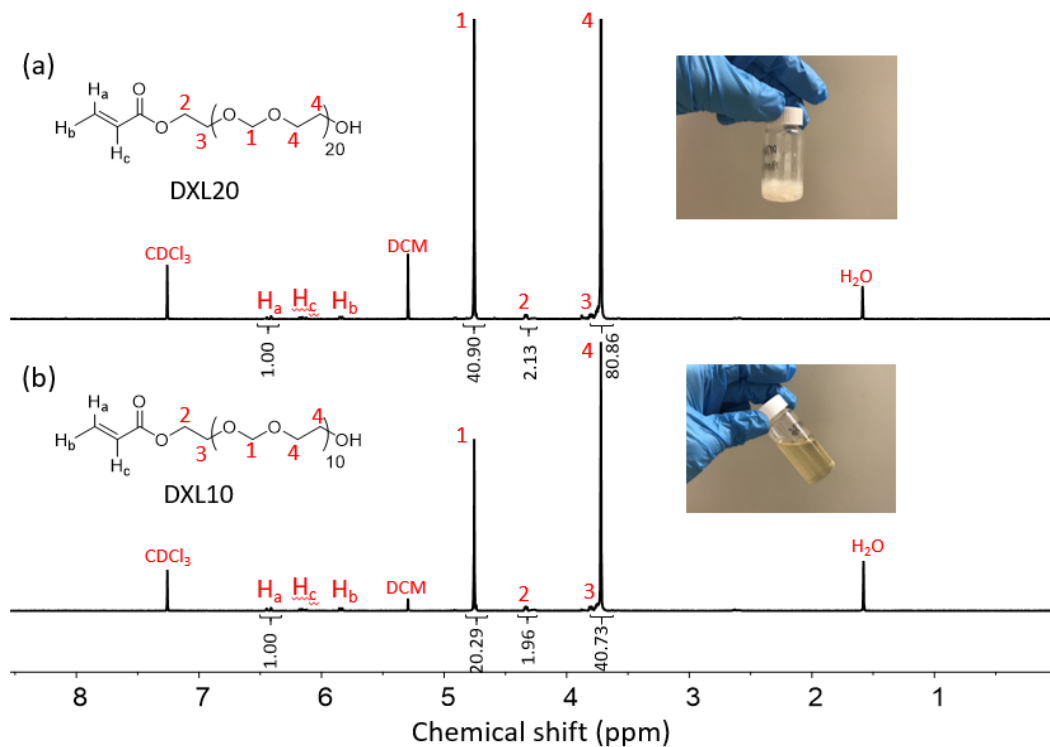
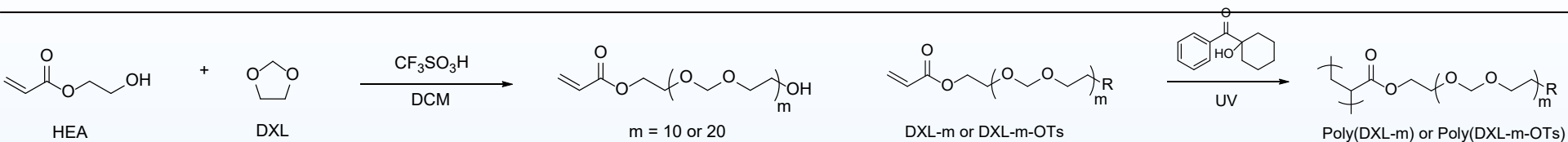
$$\frac{S_{\text{CO}_2}}{S_{\text{N}_2}} \gg 1 \quad \frac{D_{\text{CO}_2}}{D_{\text{N}_2}} \approx 1$$

Our Approach: Sorption-Enhanced Mixed Matrix Membranes

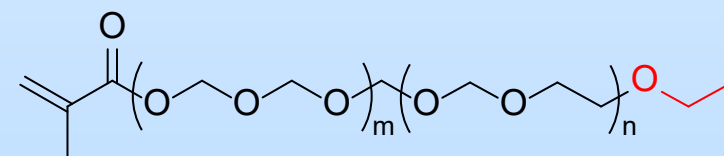


- **CO_2 -philic rubbery polyethers**
- **Porous metal organic polyhedra (MOPs)**

Design and Synthesize Functional Polyethers



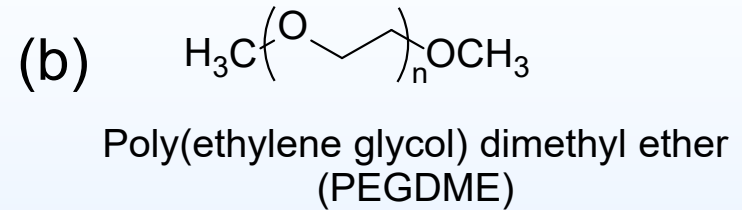
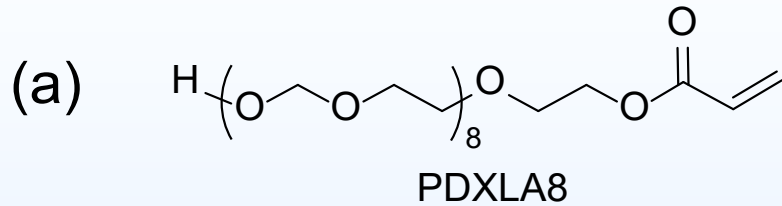
Varying ether oxygen content and chain end groups



POM-ran-PDXLEA

50 - 100 g/batch production demonstrated

Advanced Materials for CO₂/N₂ Separation

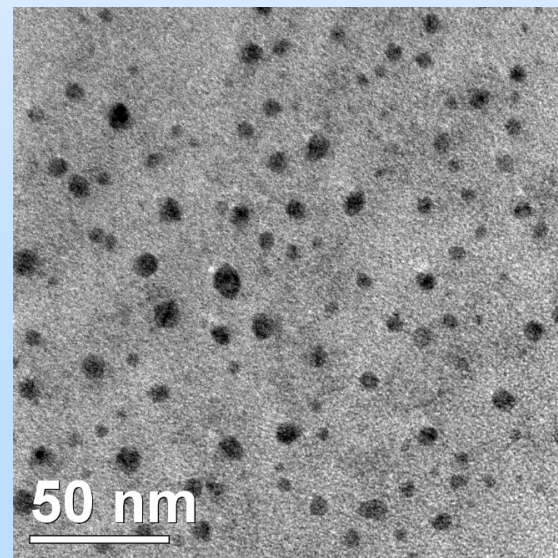
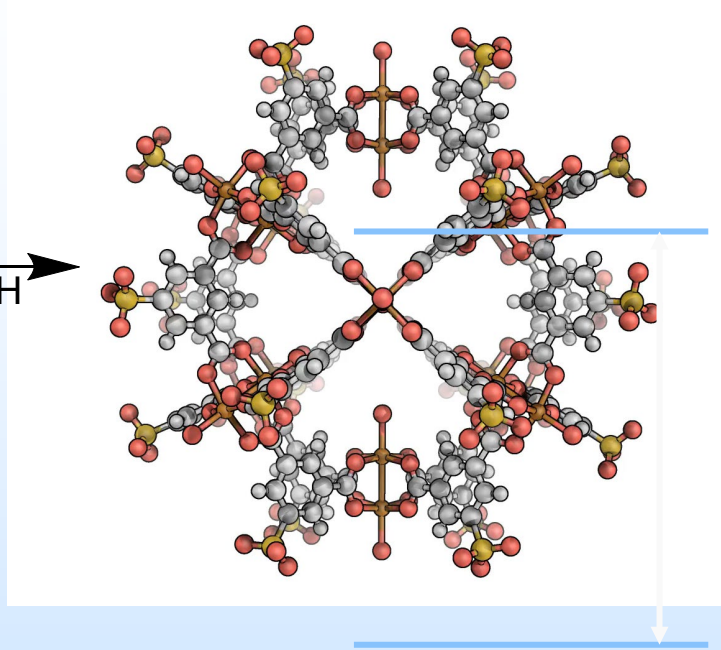
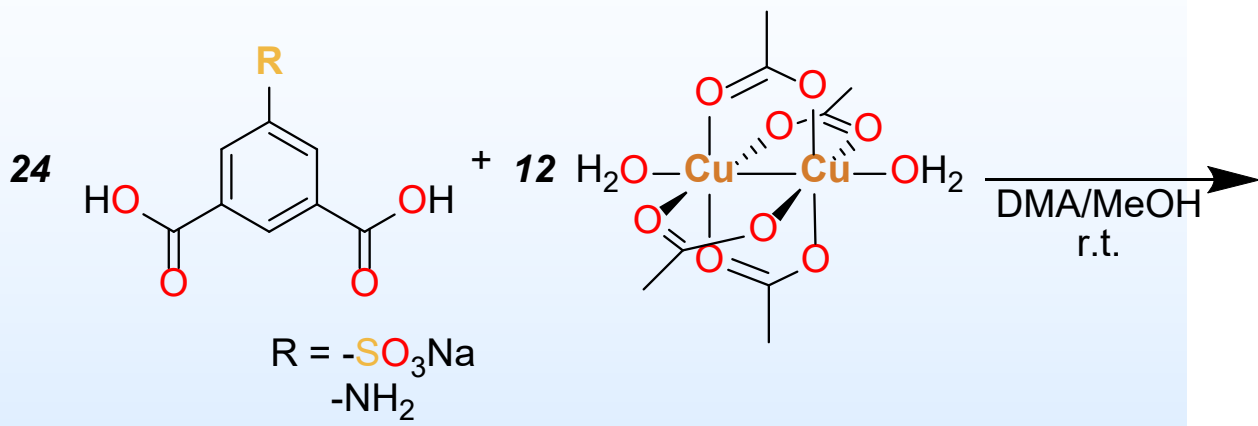


PEGDME (wt%)	T _g (°C)	CO ₂ permeability (Barrer)	CO ₂ /N ₂ selectivity
0	-59	223	56
10	-68	380	51
30	-81	830	48
45	N/A	1406	45
50	N/A	1681	42
60	N/A	1671	46

Increasing PEGDME content increases chain flexibility and gas permeability

Pure-gas
at 35°C

MOP Synthesis: 10 g/batch



Unexpected Improvement by Low-loading MOP-3

PDXLA8; test at 35°C

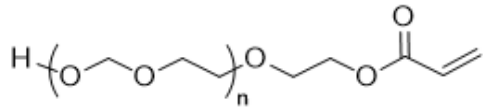
MOP-3 content (wt.%)	P_{CO_2} (Barrer)	CO_2/N_2
0	223	56
1	293	56
2	344	59
3	362	57
10	263	55
20	233	57

PDXLA8/PEGDME (50/50)

MOP-3 content (wt.%)	T (°C)	P_{CO_2} (Barrer)	CO_2/N_2	CO_2/O_2
0	25	1005	56	22
1	25	662	48	25
3	25	1343	62	24
3	21	1106	74	31
5	25	703	59	-

Meet the project target
 CO_2 : 1000 Barrer; CO_2/N_2 : 75

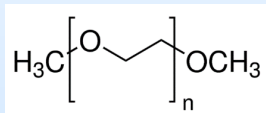
Mixed-gas Tests



PDXLA8/PEGDME

Mixed gas: CO₂/N₂=15/85, 100-160 psig

PDXLA8, n = 8



PEGDME
240 g/mol

PEGDME (wt%)	T (°C)	Mixed-gas CO ₂ (Barrer)	Mixed-gas CO ₂ /N ₂ selectivity
45	35	1200	60
50	35	1450	62
50	25	1290	79

Meet the project target
CO₂: 1000 Barrer; CO₂/N₂: 75

Simulate Gas Transport Properties

Solubility

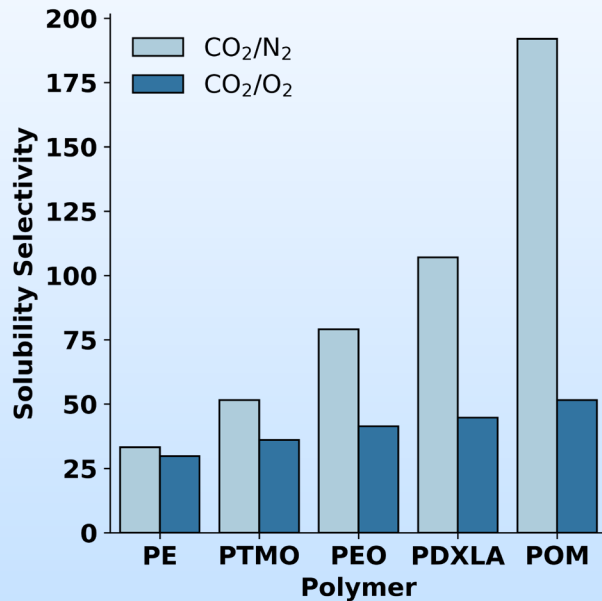


Diffusivity

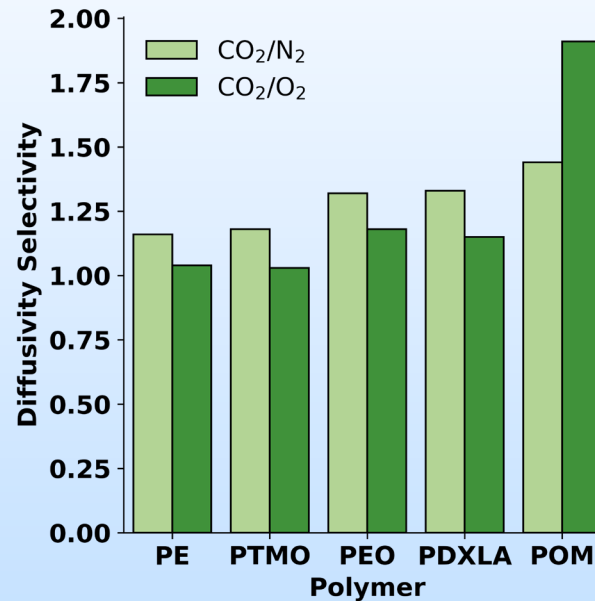


Permeability

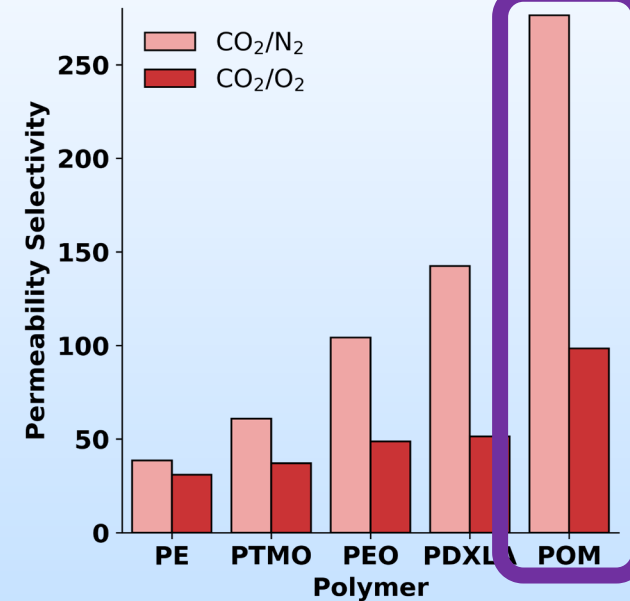
a)



b)



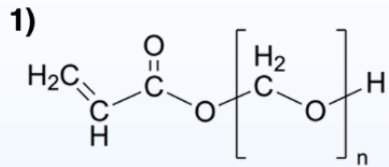
c)



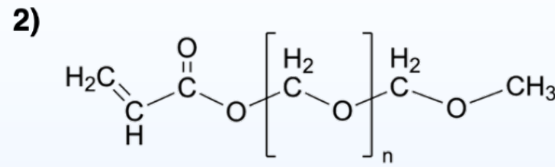
Increasing ether oxygen content increases separation properties

Effect of Chain End Groups and Ether Oxygen Content

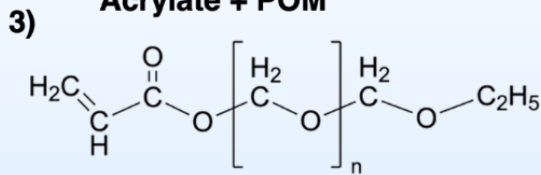
Polymer 9 shows the most promise with better CO_2/N_2 and CO_2/O_2 separation performance.



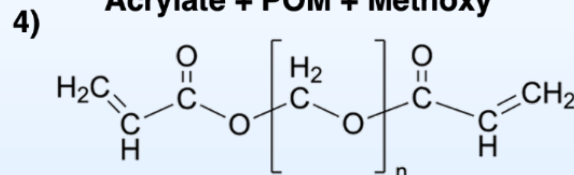
Acrylate + POM



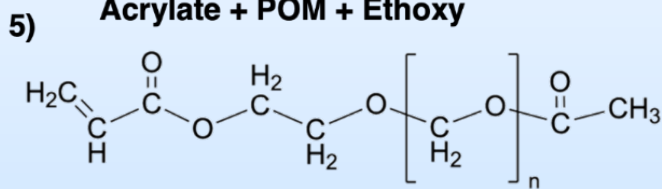
Acrylate + POM + Methoxy



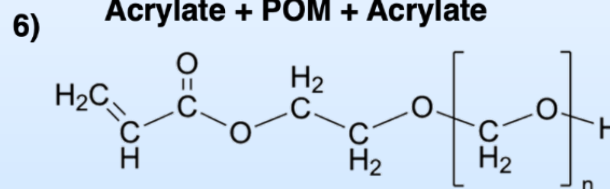
Acrylate + POM + Ethoxy



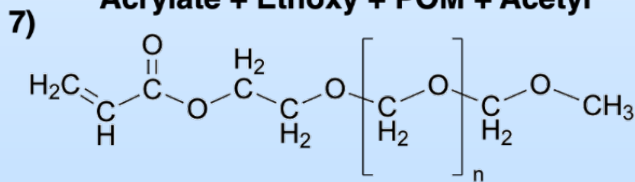
Acrylate + POM + Acrylate



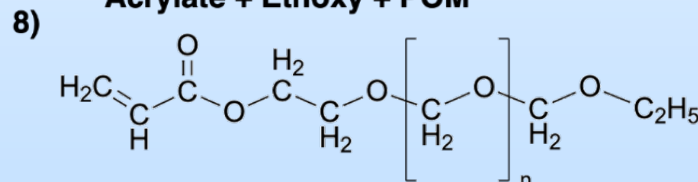
Acrylate + Ethoxy + POM + Acetyl



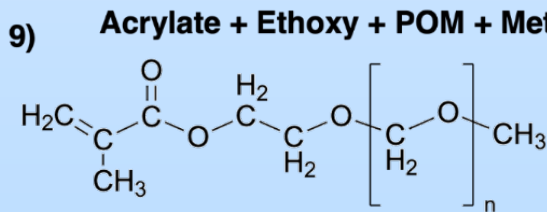
Acrylate + Ethoxy + POM



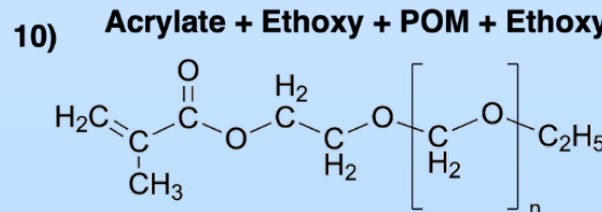
Acrylate + Ethoxy + POM + Methoxy



Acrylate + Ethoxy + POM + Ethoxy

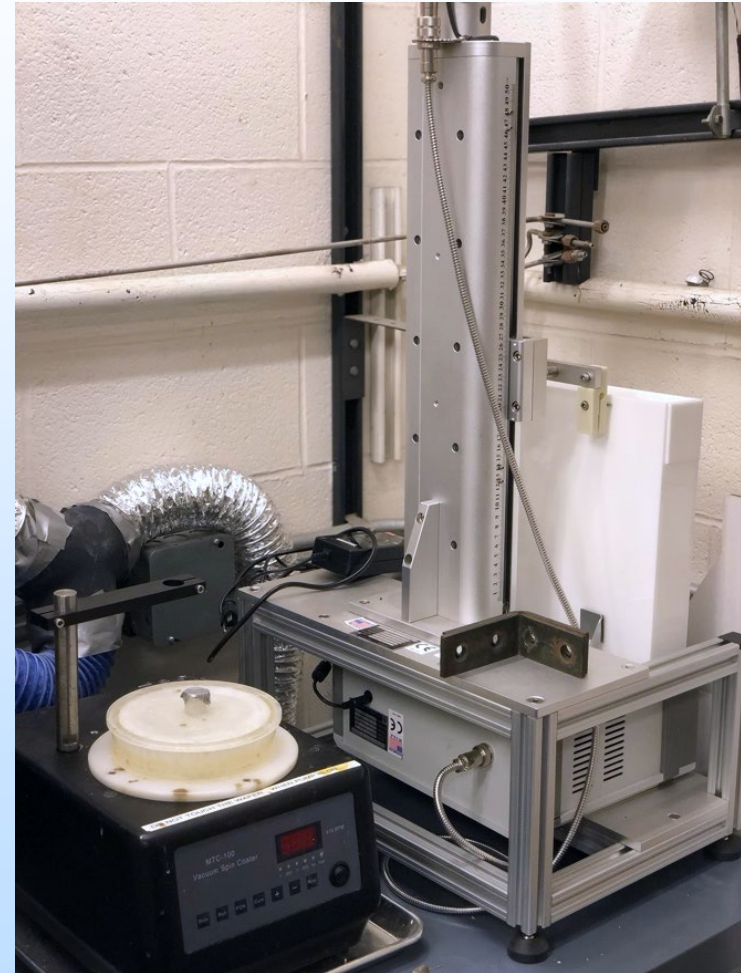
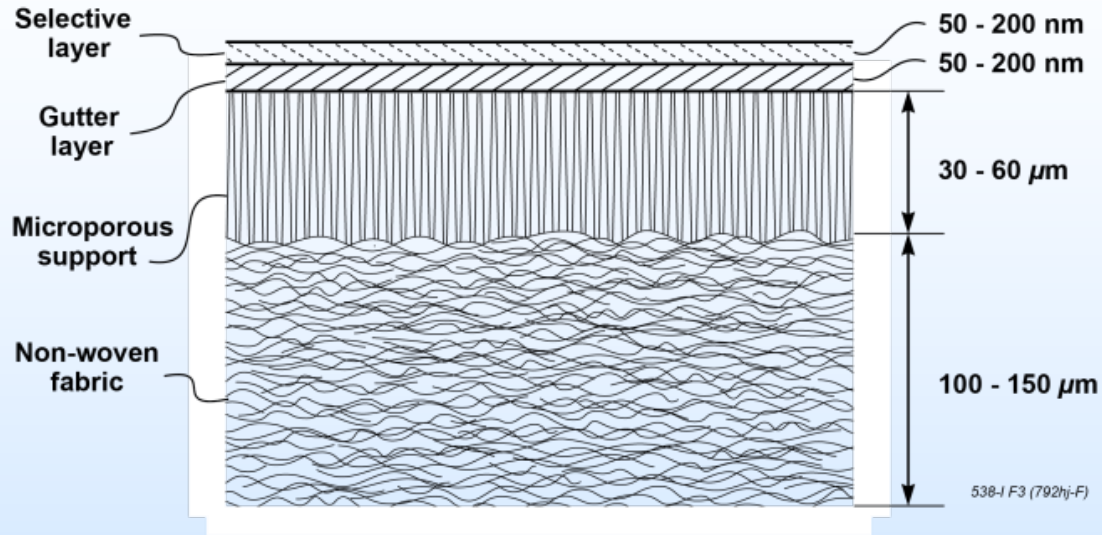


Acrylate + Ethoxy + POM + Methyl

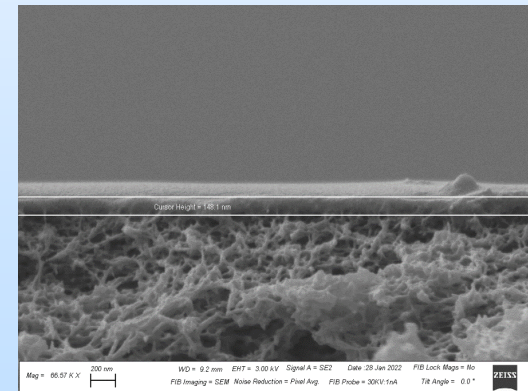
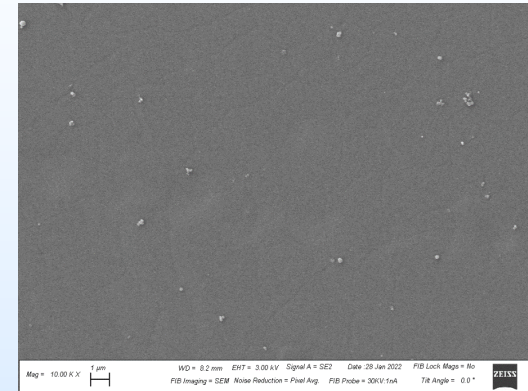
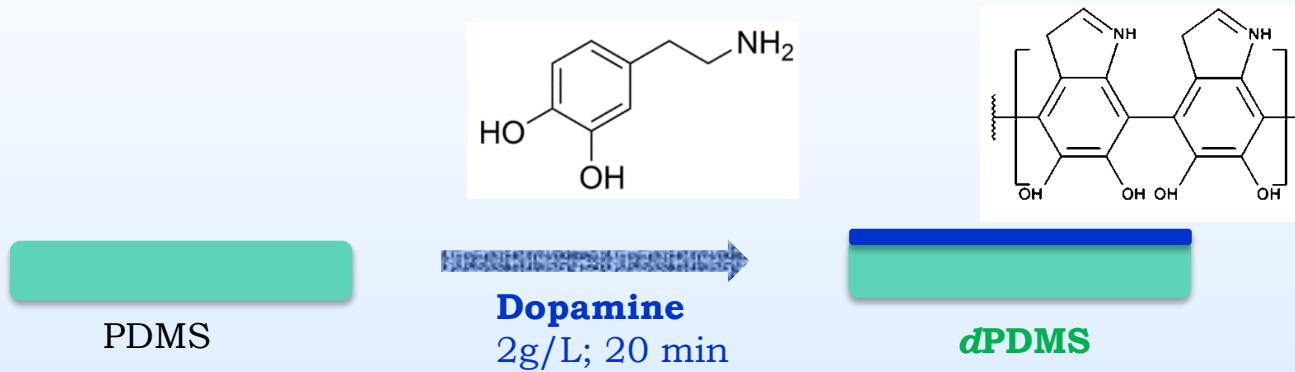


Acrylate + Ethoxy + POM + Ethyl

Thin Film Composite (TFC) Membranes

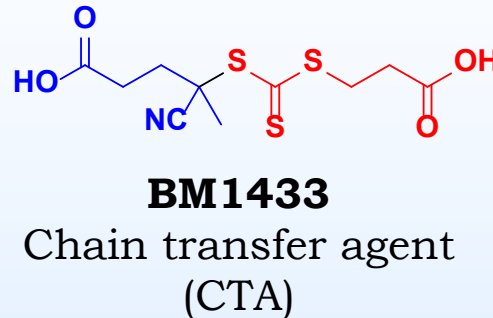
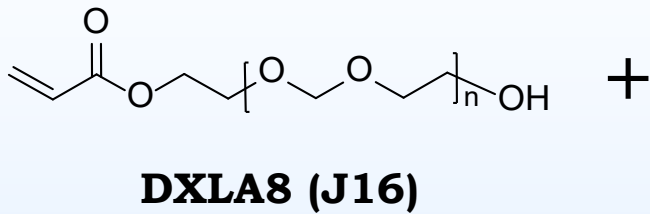


Improve PDMS Surface Compatibility via Dopamine Modification

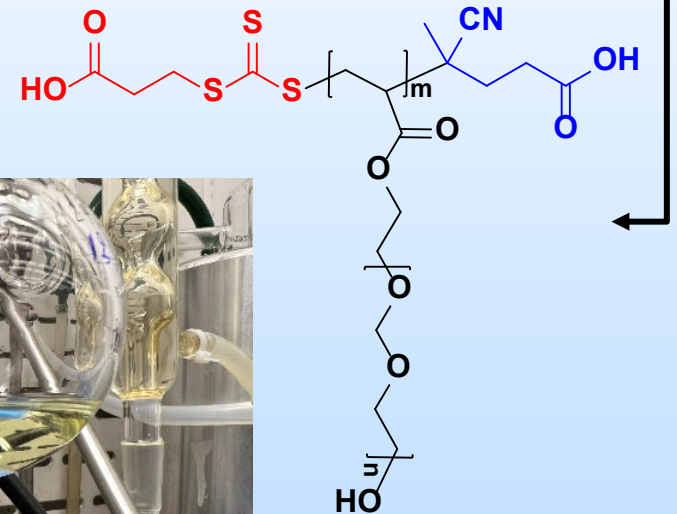


Gutter layer	Water contact angle (°)	CO ₂ permeance (GPU)	CO ₂ /N ₂ selectivity
PDMS/PSF	110	9,000	8.0
dPDMS/PSF	95	6,800	8.7

Atom Transfer Radical Polymerization (ATRP)

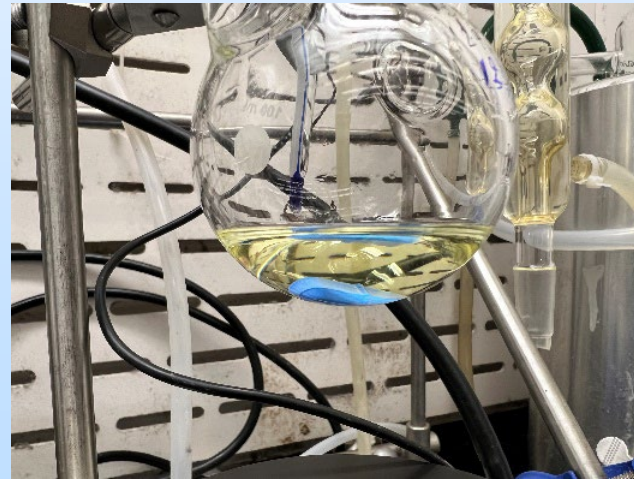


AIBN
75°C, DMF



- [DXLA]:[BM1433]:[AIBN] = 100:1:0.1
- [DXLA] = 0.5 M (25 wt.% DXLA in DMF)

Reaction time:
3 h, 75°C

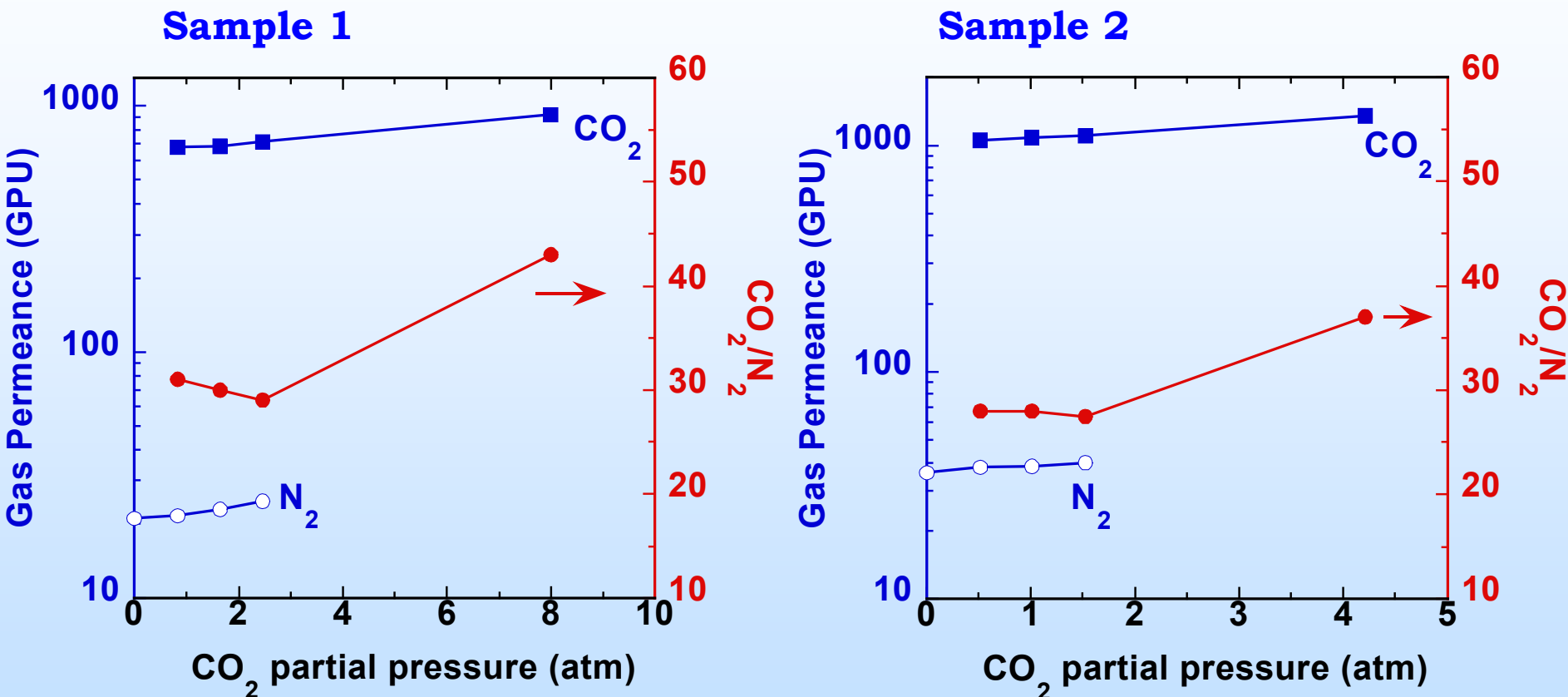


$M_{n,theo.} =$
71.1 kDa

Representative TFC Membranes

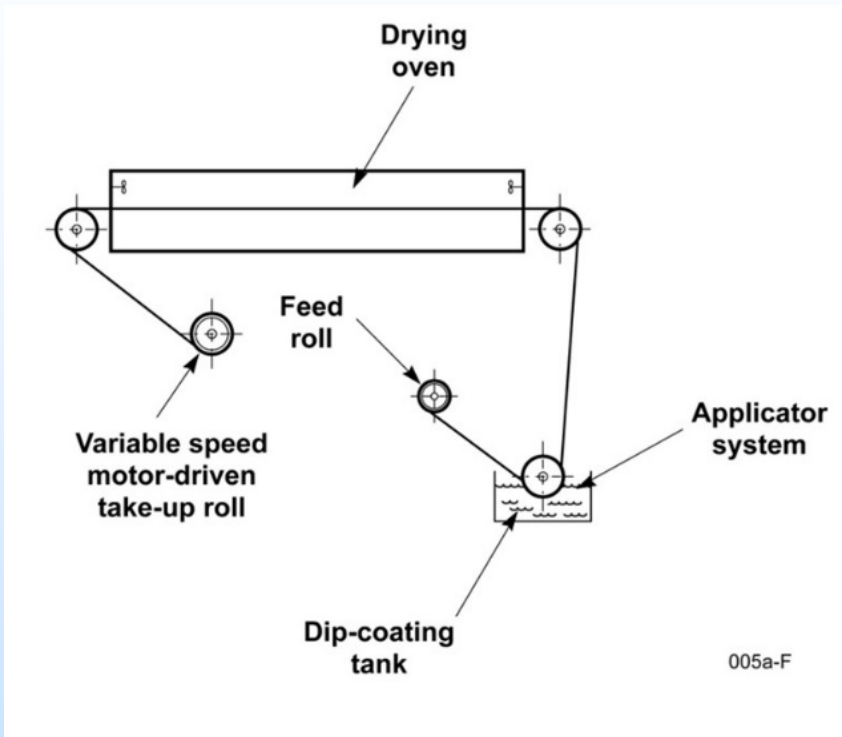
Membranes	Polymer content (wt.%)	Temp. (°C)	CO ₂ (GPU)	CO ₂ /N ₂ selectivity
pDXLAc8	2	22	1563	47
pDXLAc12 (B2) (#1)	2	22	3690	42
pDXLAc12 (B2) (#2)	2	22	2684	57
		35	3722	38
pDXLAc12 (B2) (#3)	1.5	22	3040	48
		35	4270	39
pDXLAc12 (B2) (#3)	1.5	35	922	51
pDXLAc12 (B3) (#5)	1.5	35	1344	37

Mixed-gas performance of pDXLAc12 at 35°C



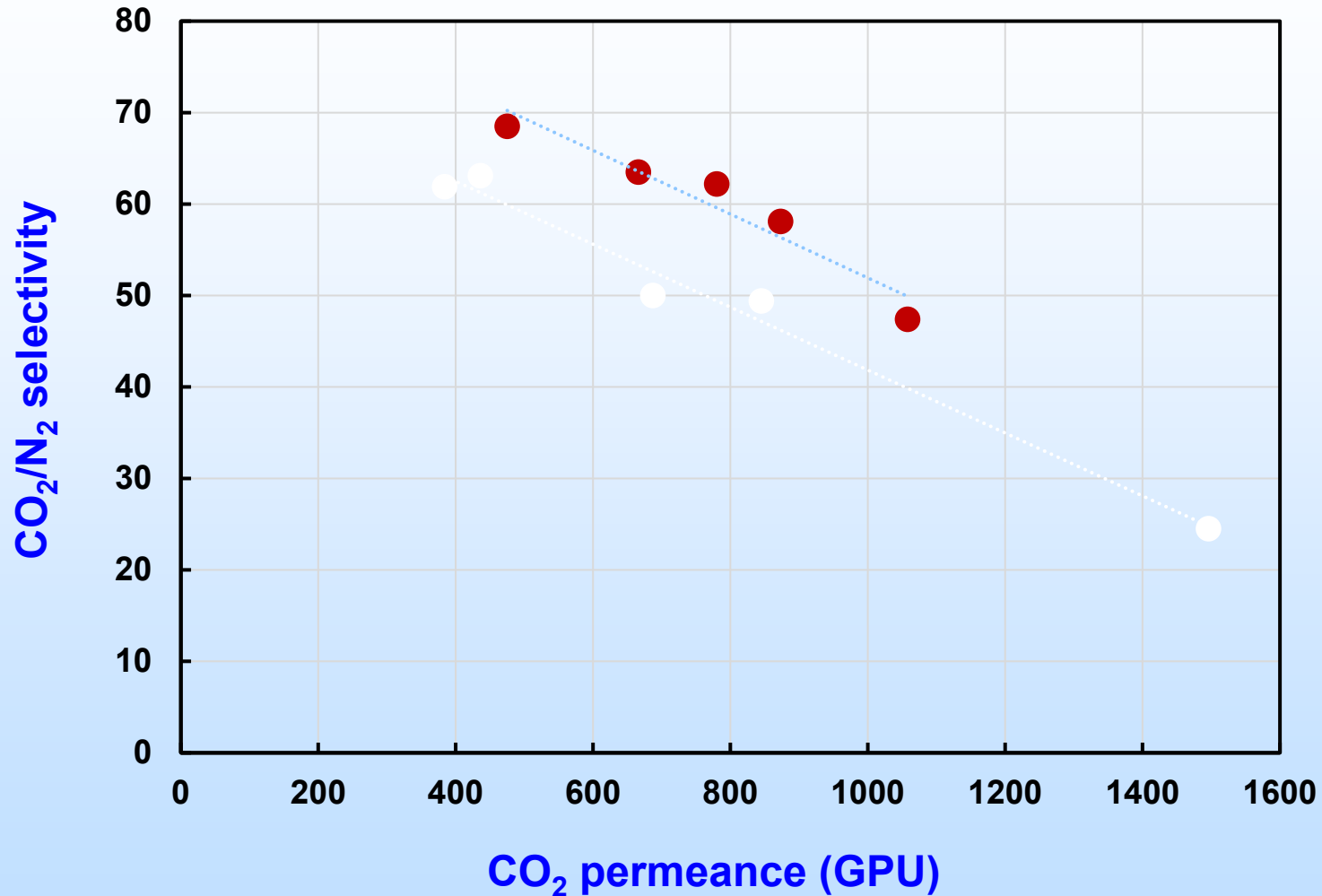
CO₂ permeance and selectivity increase with increasing CO₂ partial pressure

Development of Industrial Membranes



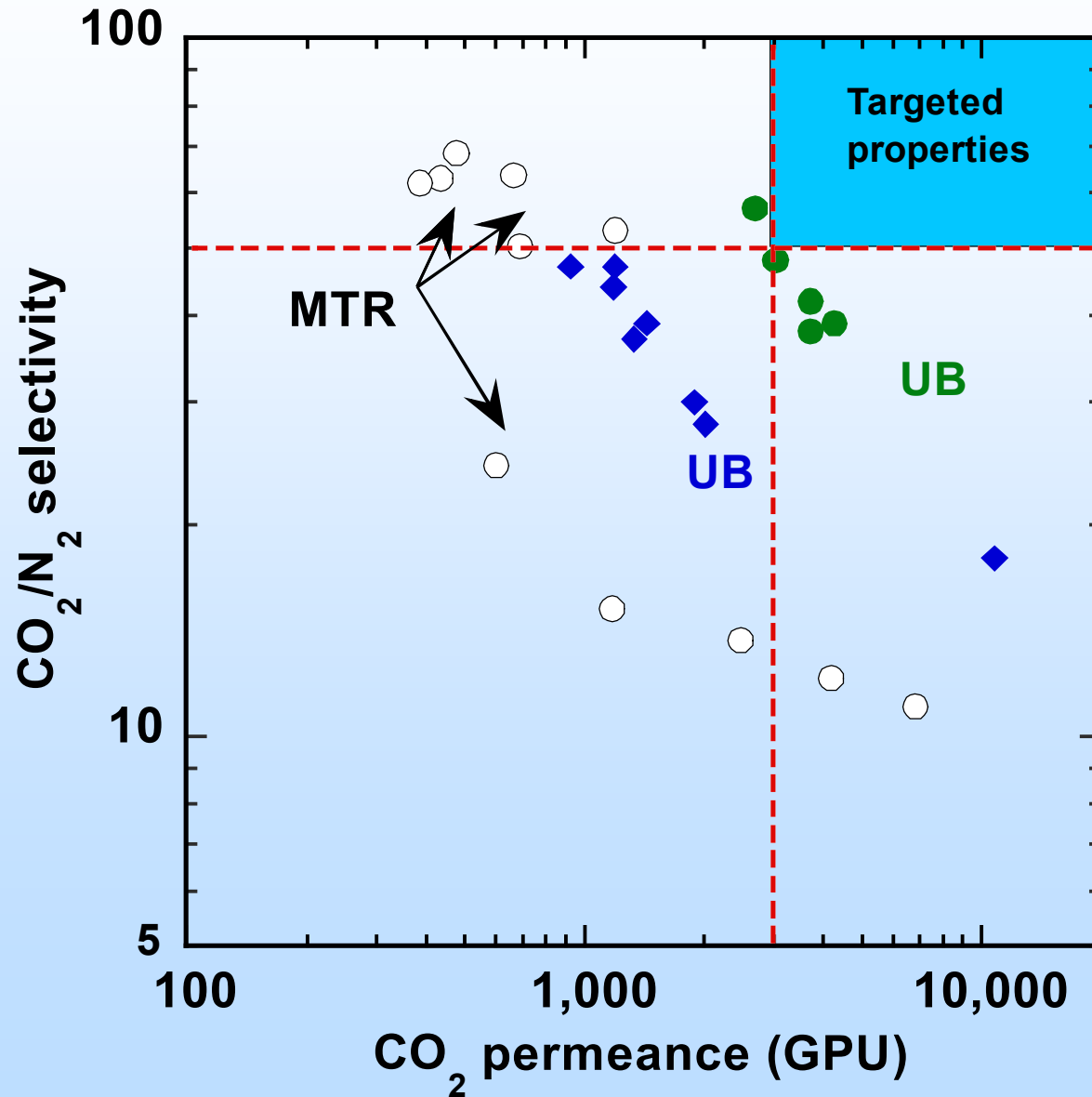
- MTR leads TFC membrane scale up activities
- Research-scale (12-inch width) roll-to-roll coating equipment has been used

Pure-gas Results of TFC Membranes



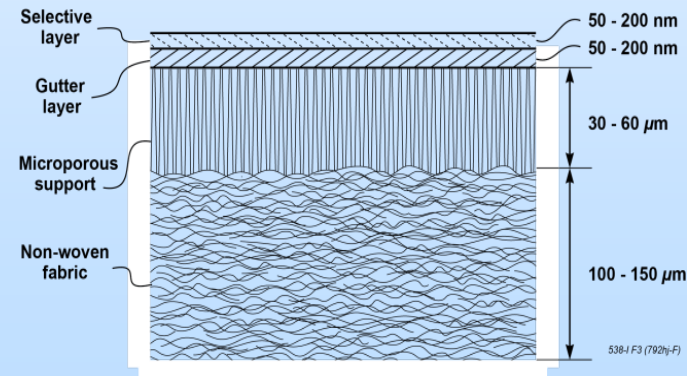
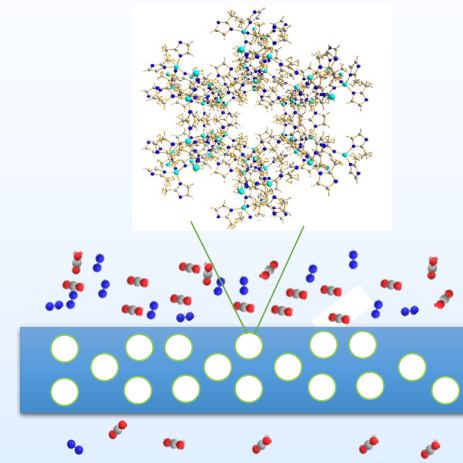
Selectivity higher than the project target of 50

Summary and Challenge of the Membrane Development



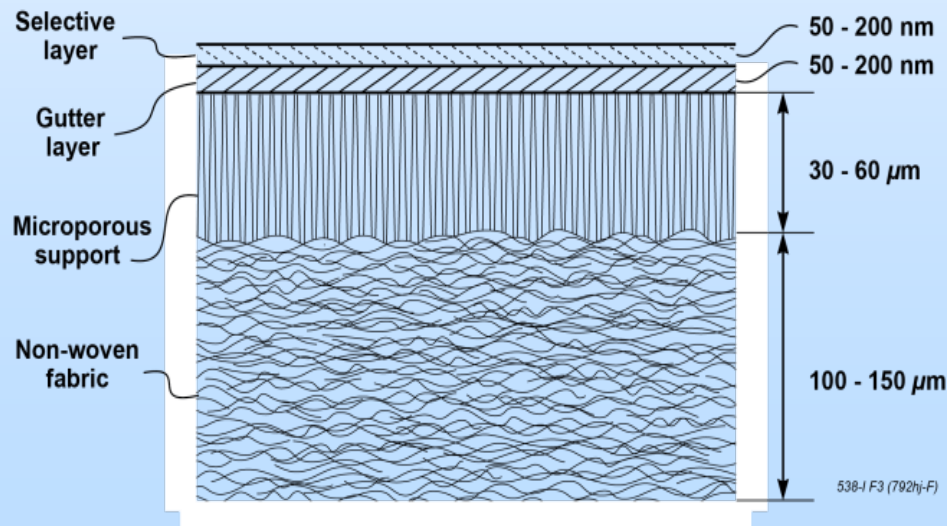
Summary

- We developed novel polyethers and MOP-3 achieving CO_2 permeability of 1000 Barrer and CO_2/N_2 selectivity of 75
- We demonstrated the feasibility of fabricating TFC membranes using a roll-to-roll process.
- Future work will focus on the optimization of polymer synthesis and membrane fabrication to improve reproducibility



Summary: Lessons Learned

- It is not trivial to convert freestanding films to industrial thin film composite membranes, particularly for mixed matrix materials
- Reproducibility is critical to developing new polymers and membranes with various nanostructures



Acknowledgement

- Project manager: Krista Hill
- Funding from the NETL
- Costshare from NYSTAR, Caltech, RPI, and MTR

