

Dual Function Materials for Direct Air Capture of CO₂

DE-SC0020795

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Susteon, Inc.

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Project Overview



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SBIR Phase II: DE-SC0020795

Funding: \$1,600,000 (DOE share)

Overall Project Performance Dates

BP1: 08/23/2021 – 08/22/2022

BP2: 08/23/2022 – 08/22/2023

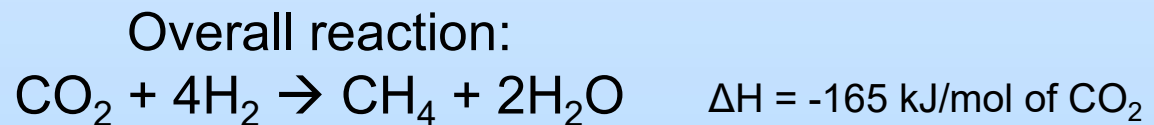
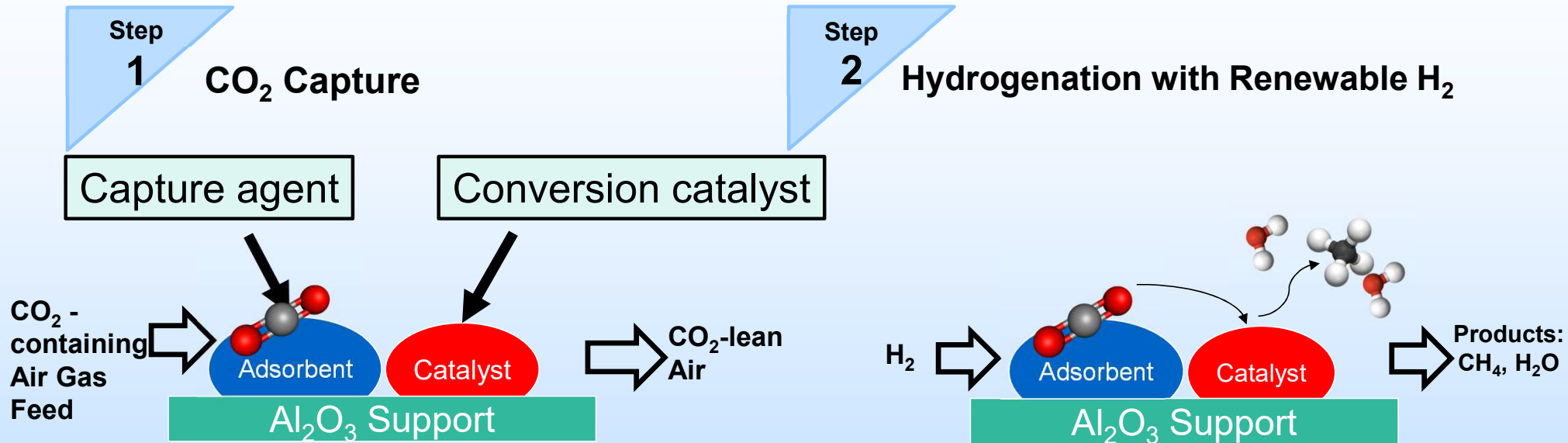
DOE Project Manager: Zachary Roberts



Industrial Partners

Reactive Direct Air Capture (DAC) of CO₂

Dual Functional Material (DFM) captures CO₂ and releases into CH₄ upon conversion



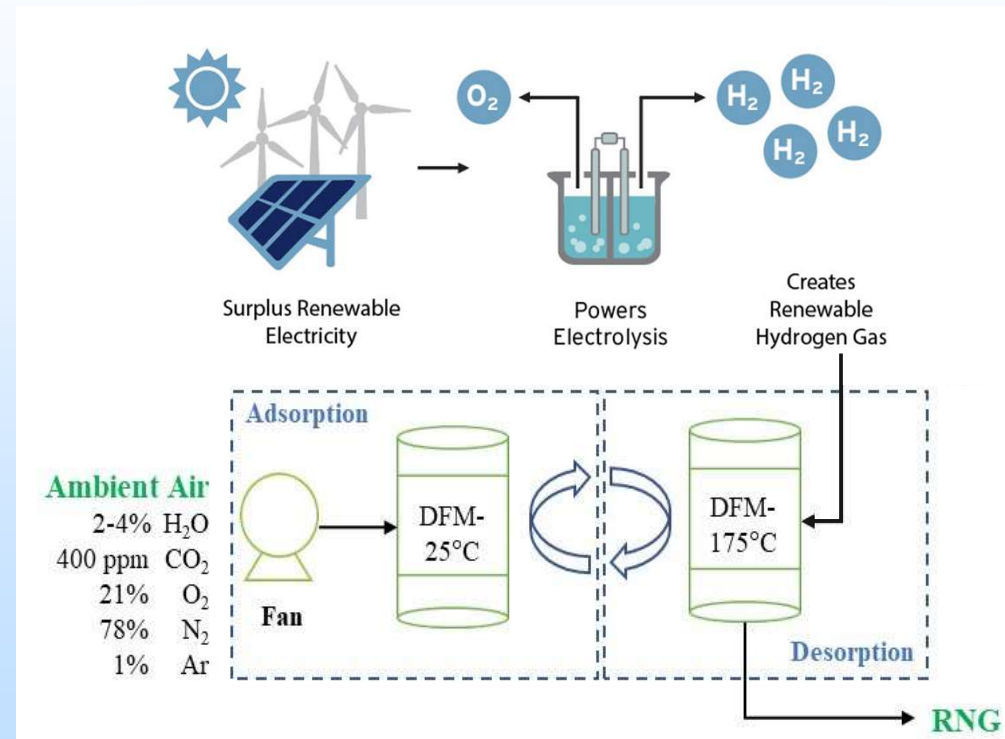
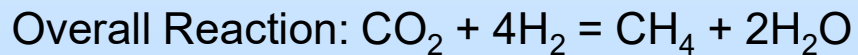
Technology Background

The overall objective of the project is to lower the cost of DAC through development of advanced dual-functional materials (DFM) and production of renewable natural gas (RNG) to offset the cost of DAC. DAC-DFM technology generates RNG from atmospheric CO₂.

Two Step Process Cycle

Step 1: Adsorb CO₂ from air onto DFM at ambient conditions

Step 2: Add renewable H₂ and heat to regenerate the sorbent to produce RNG



This is a **Power-to-Gas** technology using atmospheric CO₂.

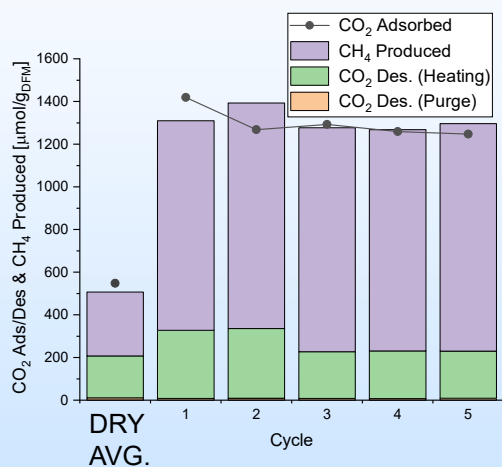
Project Objectives of SBIR Phase II

Technical Objectives

- Optimize DFMs on the selected support structures to achieve maximum CO₂ adsorption capacity, low pressure drop and CO₂ conversion to methane
- Develop an efficient heating method to minimize energy requirement
- Design and build a bench-scale unit (~ 300 g/day CO₂)
- Perform parametric and long-term testing to obtain engineering data needed for a pilot system design
- Develop a process model to accurately represent the DAC-DFM process
- Perform and refine the techno-economic assessment (TEA) to evaluate the commercial potential of the DAC-DFM process

Summary of SBIR Phase I Results

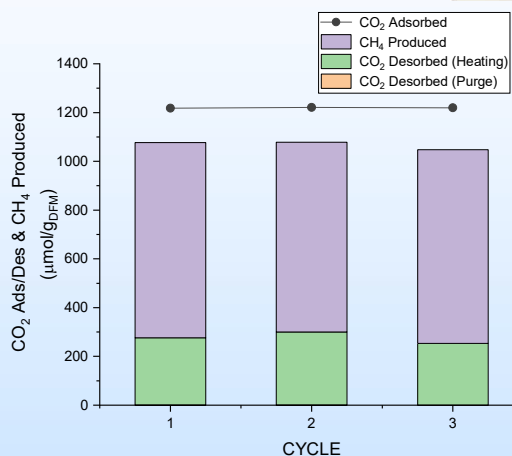
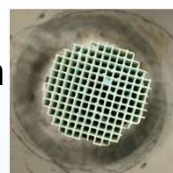
Humid vs. Dry Conditions with Granules



1% Ru, 10% Na₂O/Al₂O₃ granules

>6wt% CO₂ capacity in humid conditions

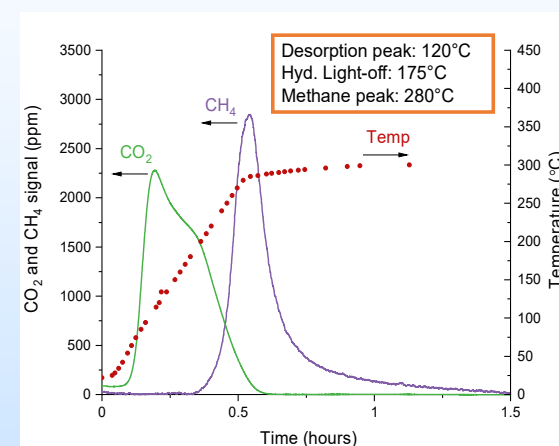
DFM Washcoated Monolith Performance



1% Ru, 10% Na₂O/Al₂O₃//TiO₂ monolith

Demonstrated washcoated monolith performance

TOS Performance With Granules



1% Ru, 10% Na₂O/Al₂O₃ granules

Methanation initiates at 175°C, peak CH₄ production at 280°C

Advantages / Challenges

Advantages

- Sorbent materials are low-cost and widely available.
- DFM performance is stable over 450 hours under various climate conditions and is enhanced by humidity in air.
- Joule-heated monolith provides fast, efficient heating for CO₂ conversion into methane.

Challenges

- Requires low-cost carbon-free hydrogen
- Requires ruthenium catalyst for methanation
- Combining capture and conversion in one unit simplifies the flowsheet, however, it adds some complexity to the reactor design

Technical Approach

Optimize the composition of Ru + Na₂O/Na₂CO₃ on alumina support DFM

- 1) minimize amount of Ru in the DFM
- 2) maximize CO₂ capture capacity
- 3) maximize CO₂ capture kinetics
- 4) maintain high dynamic capture capacity between cycles
- 5) lowest temperature for light-off of the methanation reaction
- 6) minimum degradation over 1000s of adsorption/ methanation cycles

Support the DFM on a low-pressure drop structured support

such as monolith or laminates to minimize the energy consumption due to pressure drop.

- 1) target is **<250 kWh/ton of CO₂ captured** for low pressure-drop
- 2) supporting of DFM will require optimization of the coating process to coat a uniform layer of Na₂CO₃ and Ru/alumina.

Develop a design for heating of DFM layer

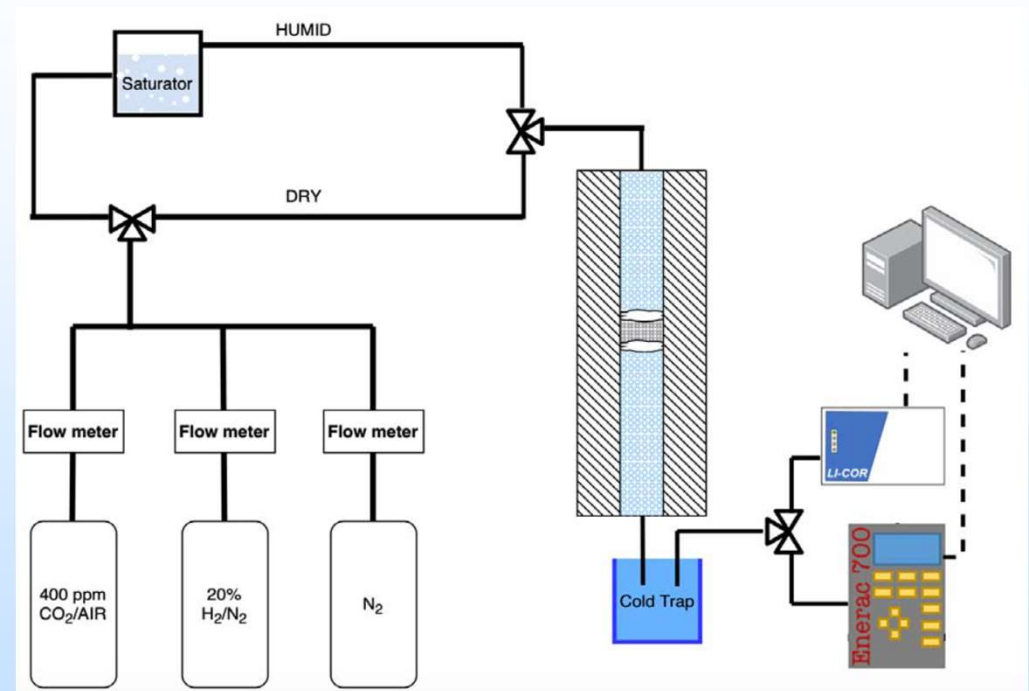
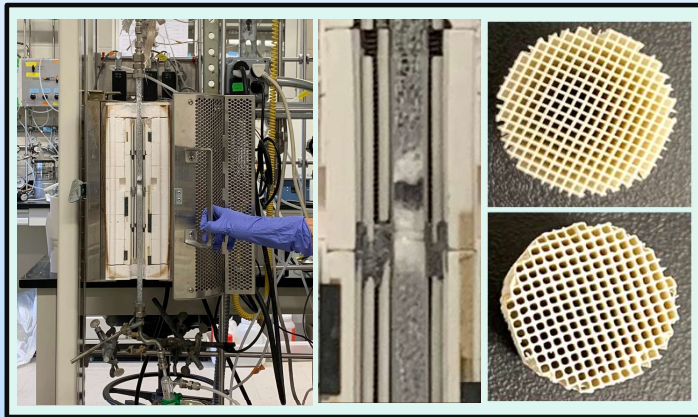
- 1) to initiate the methanation reaction by desorbing CO₂
- 2) to develop a design for a heating layer using Joule heating

Design an efficient process cycle

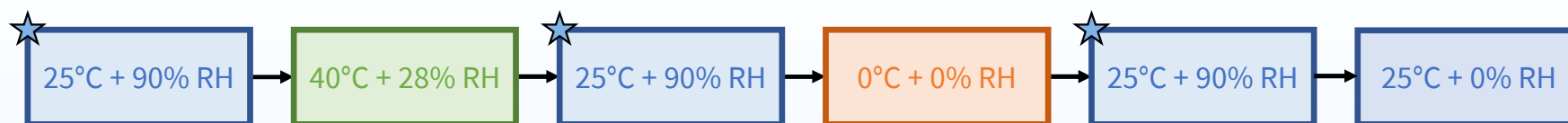
for adsorption, heating, desorption, methanation, and cooling to maximize capital productivity and minimize the overall capex and opex for the technology

Lab Scale DAC-DFM Testing

- Reactor oriented within a tube furnace with single-zone temperature control
- Gas manifold system allows for gas switching between cycle steps
- Measure parts-per-million concentrations of CO_2 and CH_4 at the reactor outlet using NDIR gas analyzers

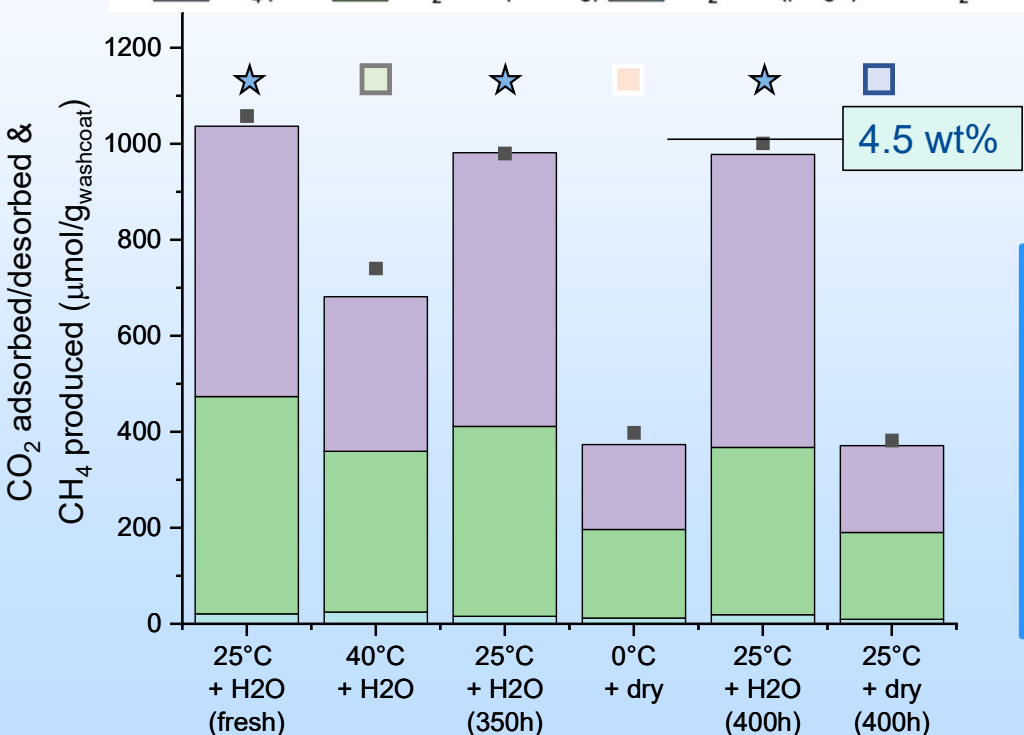
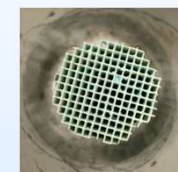


Robust Performance Under Various Climate Conditions



■ CH₄ prod.
 ■ CO₂ des. (heating)
 ■ CO₂ des. (purge)
 ■ CO₂ ads.

Sample — 1%Ru, 10%Na₂O/Al₂O₃//mo (cordierite)
 1.4 g/in³, 600 cpsi, 2 mil wall



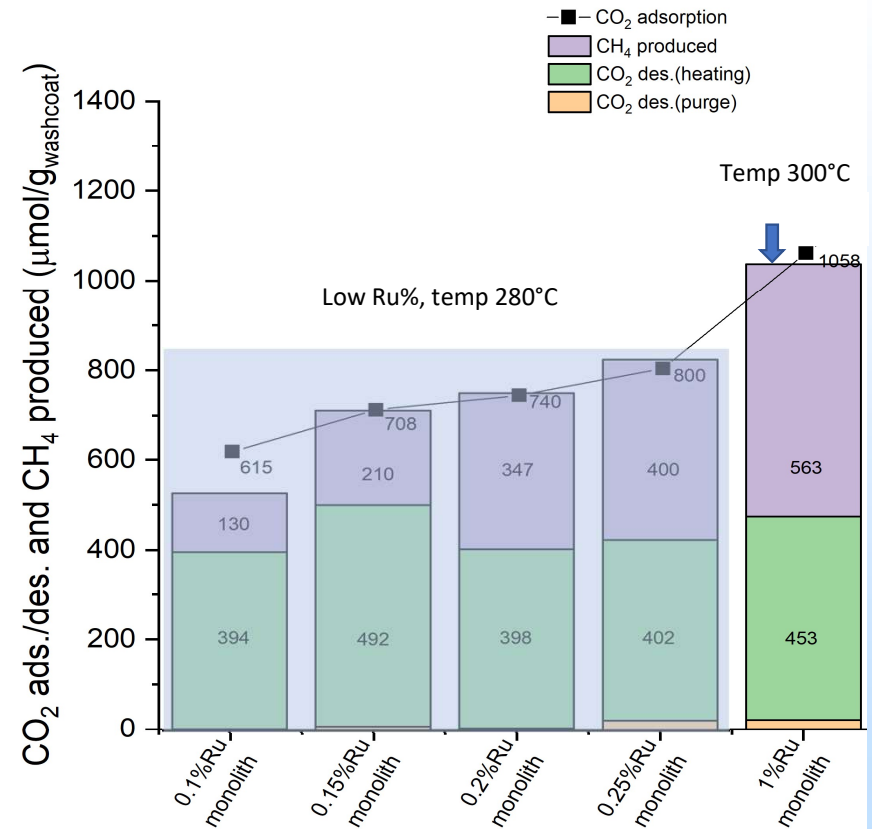
- CO₂ adsorption capacity returns to 4.5 wt% after exposed to extreme climate conditions
- Humidity significantly improves DFM performance

Effect of Low Ru on DFM Performance

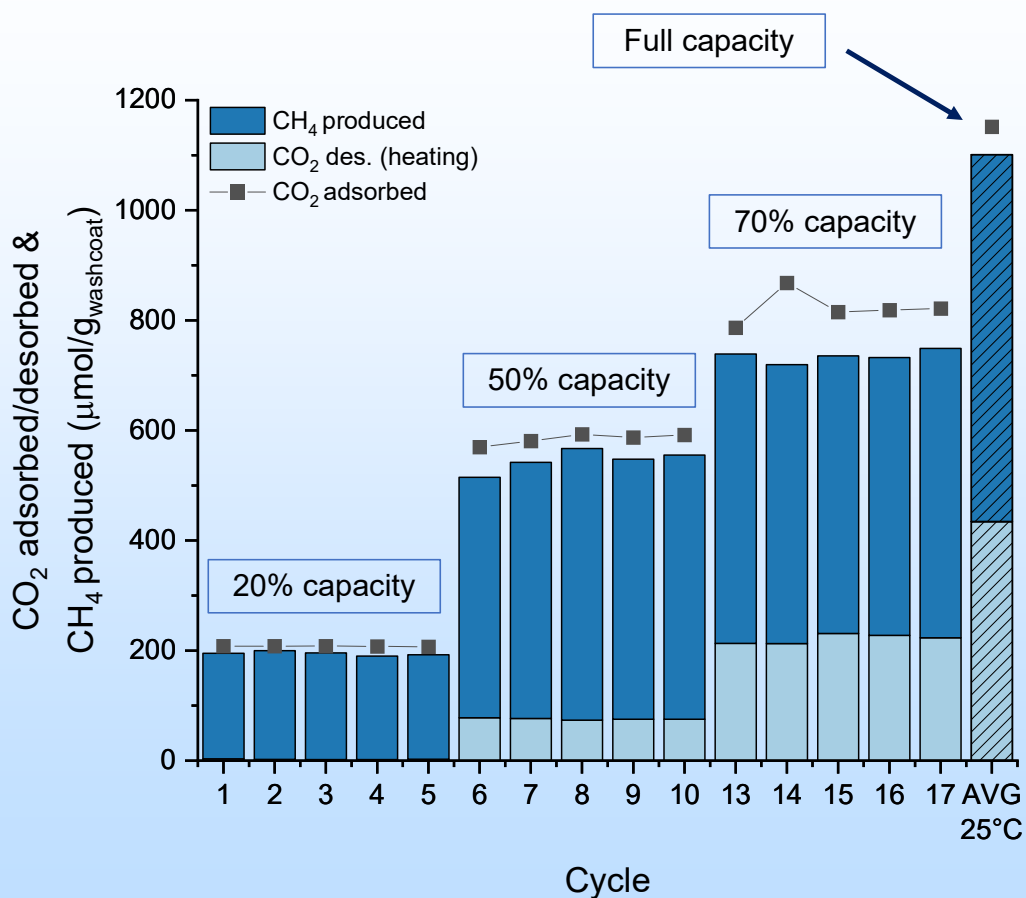
- Ru catalyzes CO₂ methanation
- Increasing Ru has a positive impact on conversion
- Ru helps to free CO₂ adsorption sites during methanation resulting in greater effective capacity

Takeaway:

An economic optimum is near 0.25%Ru for commercial applications.



Optimized Cycle Timing Improved CO₂ to CH₄ Conversion



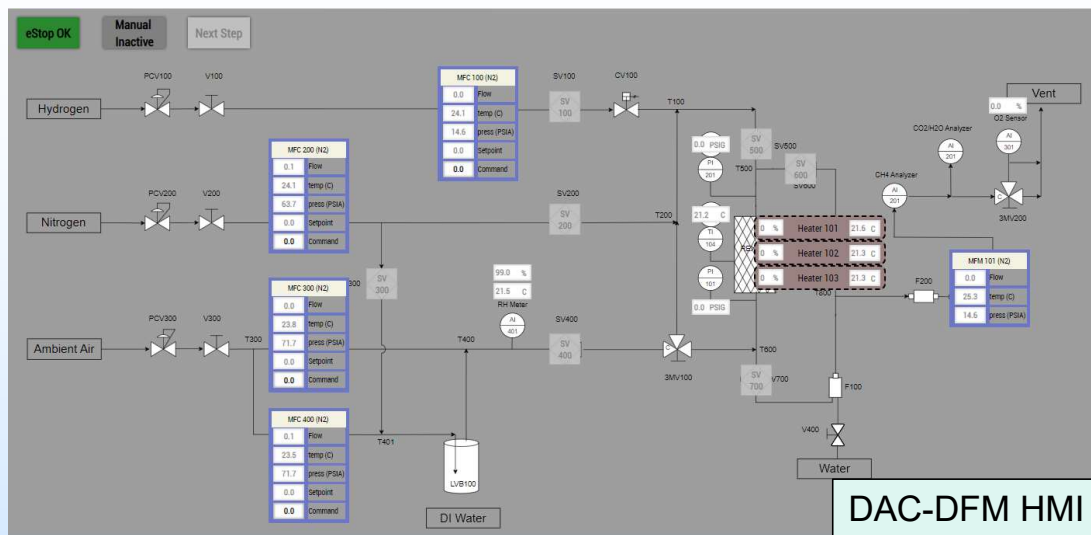
Sample — 1%Ru, 10%Na₂O/Al₂O₃//mo (NGK, cordierite)
1.4 g/in³, 600 cpsi, 2 mil wall

Adsorption Times (min): 15, 50, 90, 270

By operating near 50% capacity

- Adsorption duration is reduced by 75% which results in higher CH₄ productivity in tonnes CH₄/day/m³ DFM reactor volume.
- Near complete conversion of CO₂ into methane

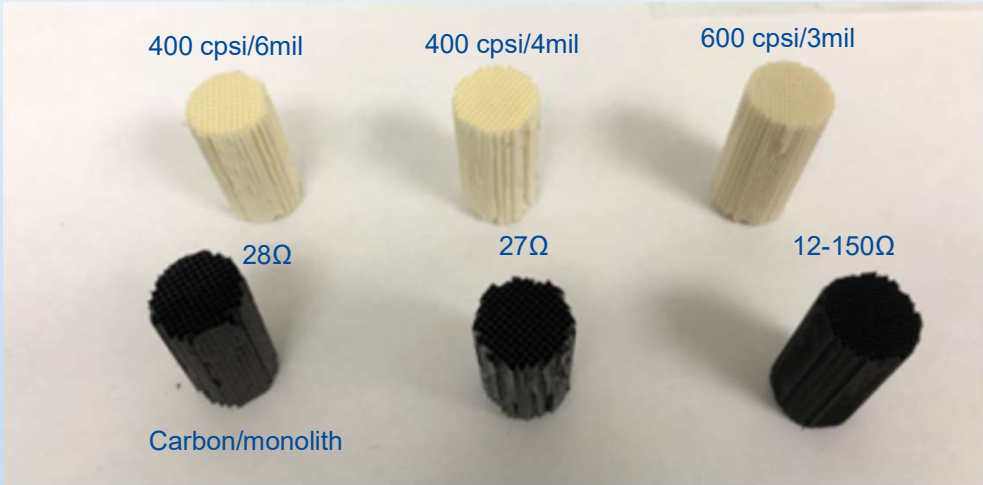
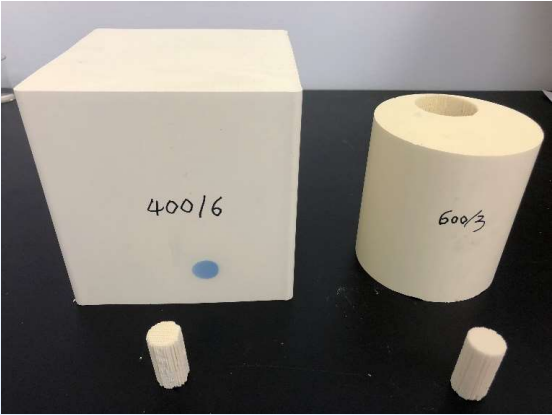
DAC-DFM Bench Scale Unit



- Feed gas manifold
 - Mass flow controllers
 - 0–90% RH control
 - Saturator
- Reactor
 - Isolation valves
 - Joule-heating with 30°C/min heating rate
- Gas Analysis
 - Mass flow meter
 - FTIR for methane
 - NDIR for CO₂ and RH

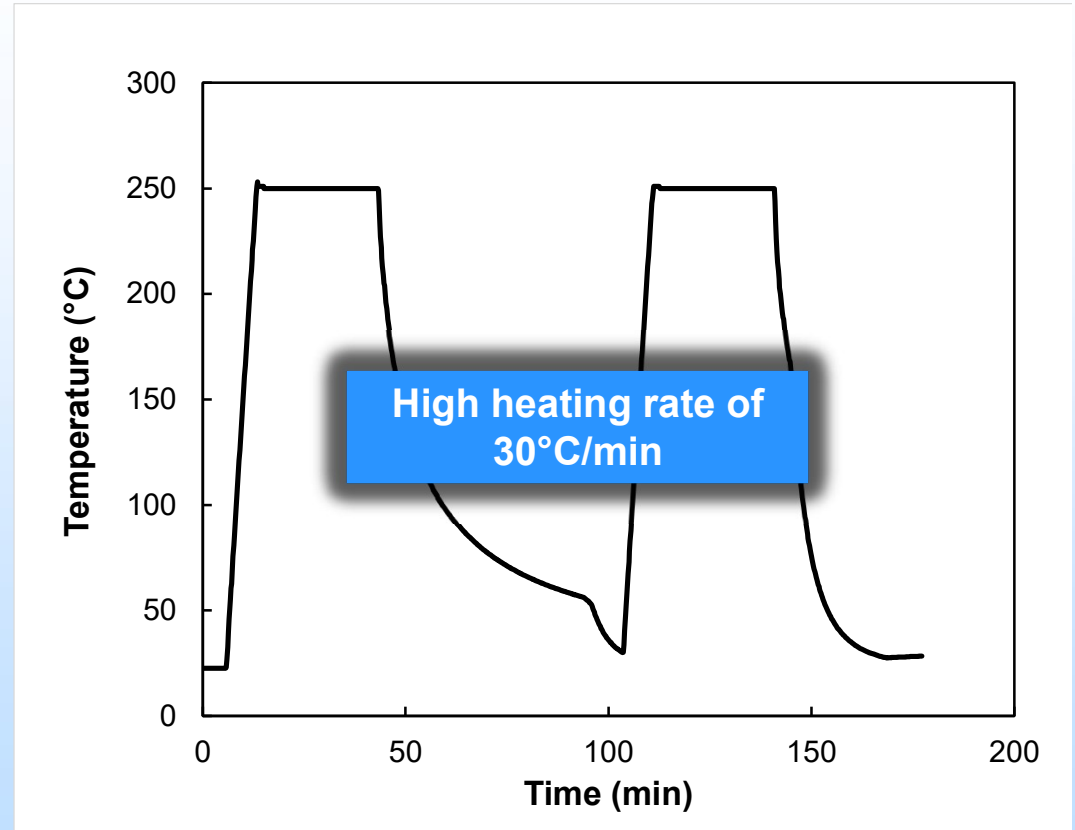
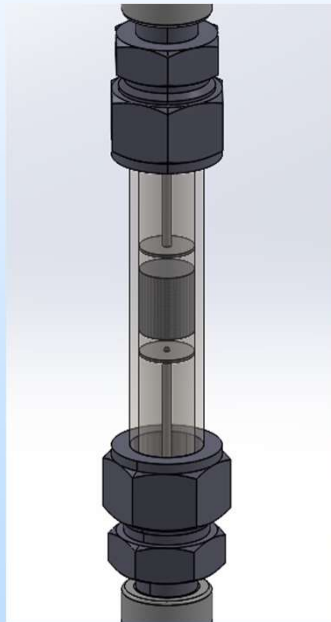


Carbon Coated Monoliths for Joule Heating



Joule Heating

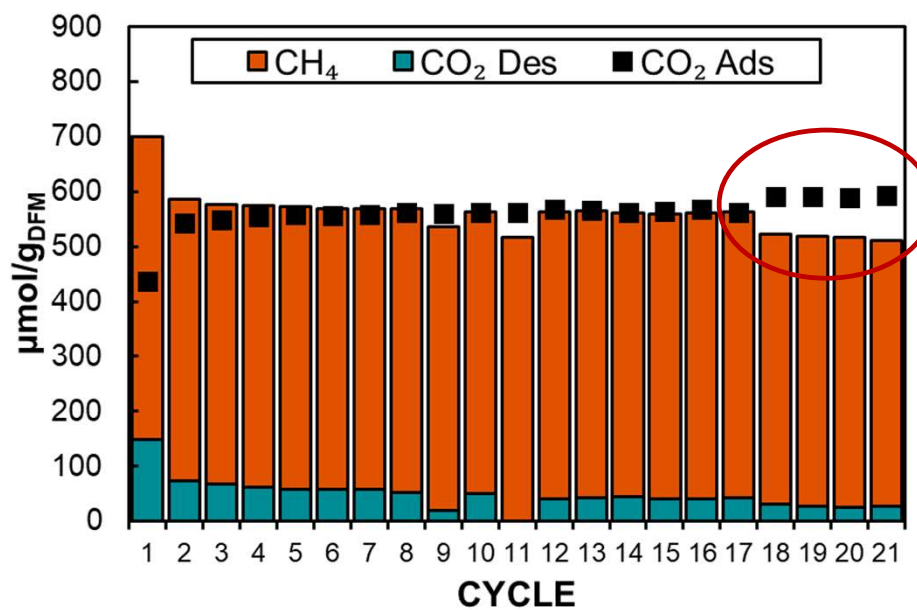
- Carbon heating layer on monolith
- 400 CPSI, 7 mil wall thickness
- Heating cycles under N₂ with PID control
- Heating Temperature: 250°C
- N₂ flowrate: 100 sccm



Joule-Heated DFM Cycle Testing Results - Capacity

- 0.8 wt% Ru/ 38 wt% Na₂O/carbon//monolith
- Improved carbon heating layer (low resistance 40.8 ohms)
- 1 hr adsorption time
- 550 $\mu\text{mol CO}_2/\text{gDFM}$ working capacity (1 hr)
- >90% Conversion of CO₂ to CH₄

Step	Duration	Gas	Flowrate
Adsorption (35°C)	1 hr	400 ppm CO ₂ / humid air	40 L/h/g (1000 mL/min)
Purge (30°C)	10 min	N ₂	4 L/h/g (100 mL/min)
Methanation (300°C)	1 hr (30°C/min heat rate)	100% H ₂	4 L/h/g (100 mL/min)



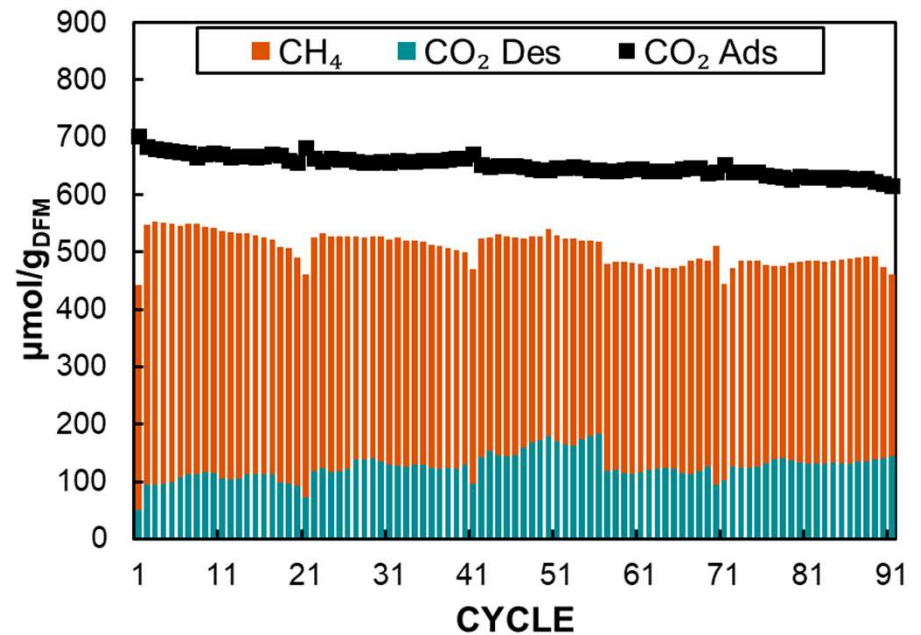
Electrode connectivity and PLC controls issues

>90% Conversion of CO₂ to CH₄

Joule-Heated DFM Cycle Testing Results - Capacity

- 0.5wt% Ru/ 30wt% Na₂O/alumina/carbon//monolith
- Improved carbon heating layer (higher resistance 228 ohms)
- 1 hr adsorption time
- ~650 $\mu\text{mol CO}_2/\text{g}_{\text{DFM}}$ working capacity (1 hr)
- 72-78% conversion of CO₂ to CH₄

Step	Duration	Gas	Flowrate
Adsorption (35°C)	1 hr	400 ppm CO ₂ / humid air	40 L/h/g (1000 mL/min)
Purge (30°C)	10 min	N ₂	4 L/h/g (100 mL/min)
Methanation (300°C) Cycles 1- 56	1 hr (30°C/min heat rate)	100% H ₂	4 L/h/g (100 mL/min)
Methanation (300°C) Cycles 57-92	1 hr (15°C/min heat rate)	100% H ₂	4 L/h/g (100 mL/min)

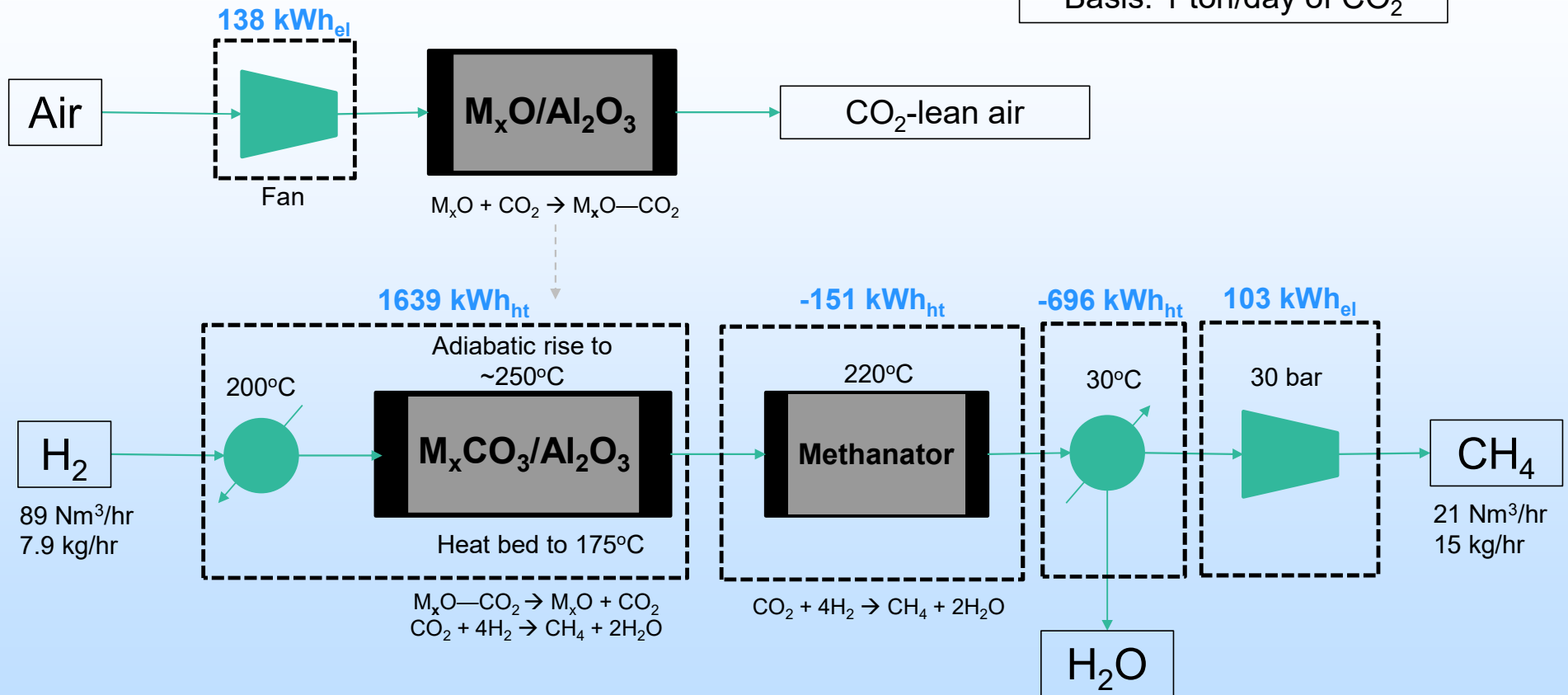


72-78% Conversion of CO₂ to CH₄

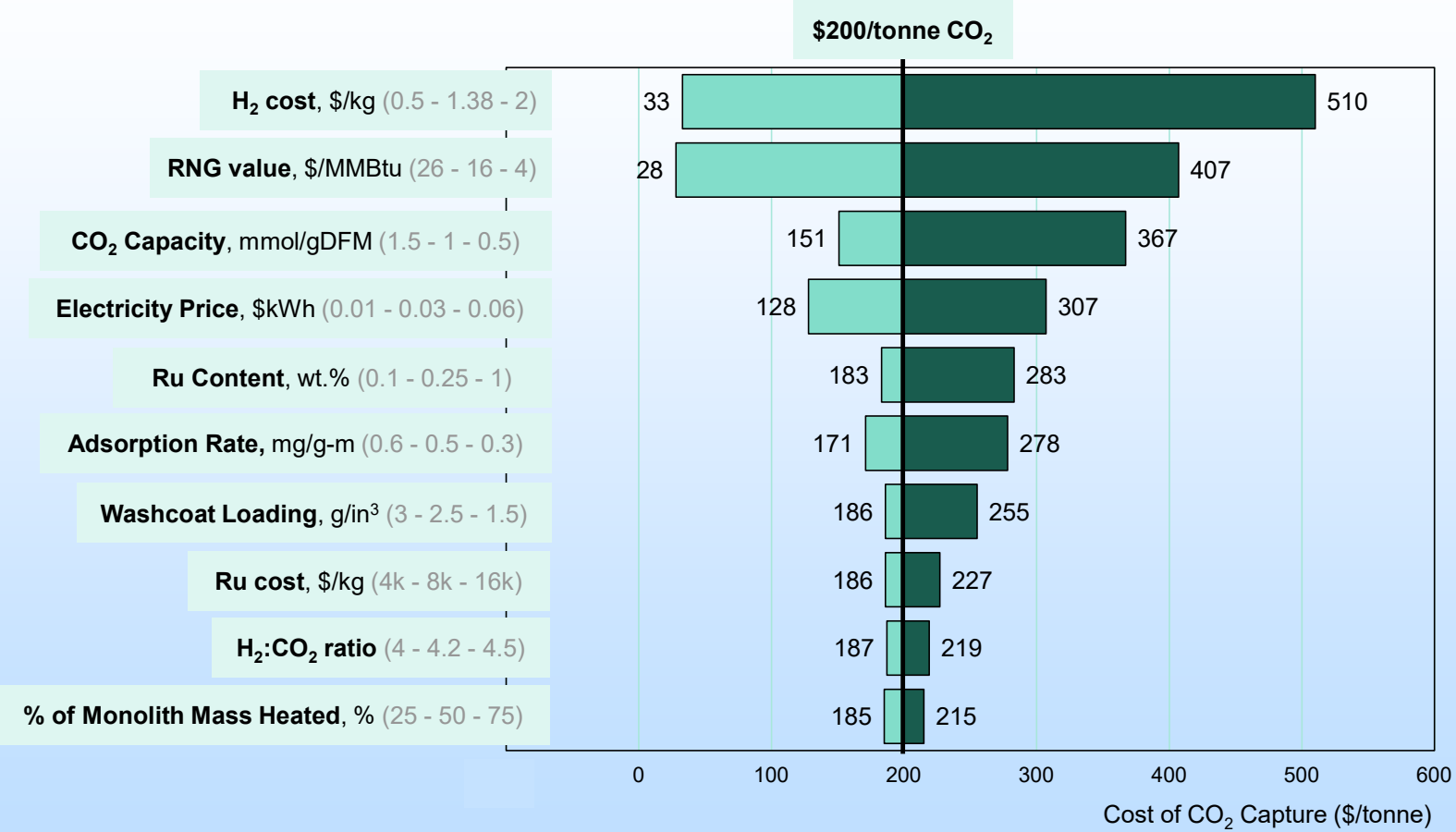
>300 Hours of continuous testing with joule heating

Process Design

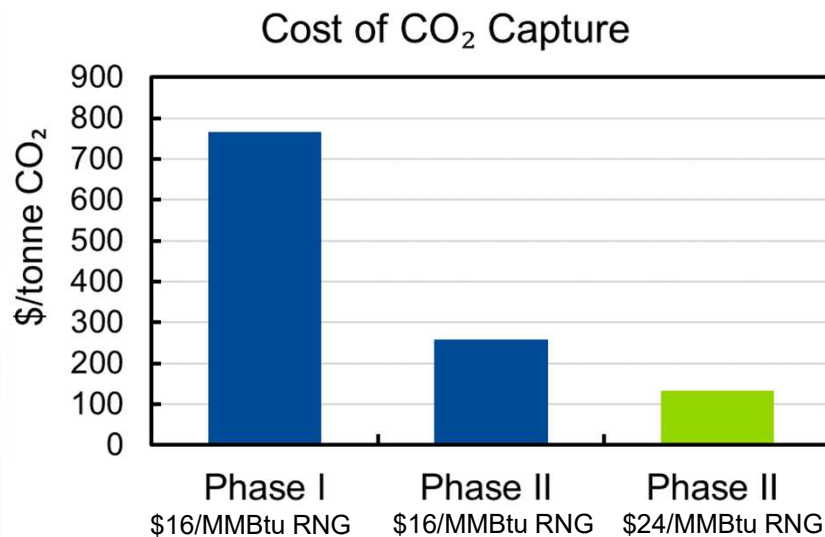
Basis: 1 ton/day of CO₂



Sensitivity Analysis on Key Parameters



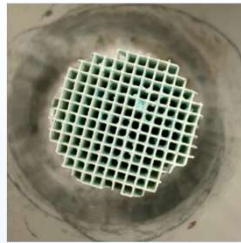
Cycle Design Impact on Process Economics



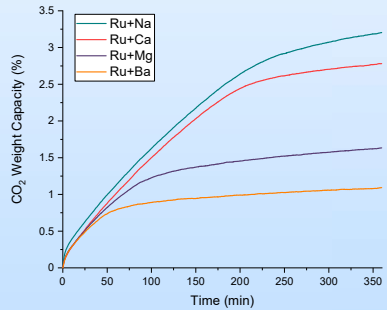
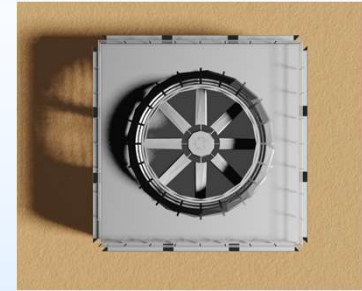
Parameter	Units	Phase I	Phase II
Ru Content	wt%	1	0.25
Adsorption Time	hrs	4	1
Capacity	mmol/gDFM	1	0.6
CO ₂ Conv.	%	50	90
Product Gas Composition	mol%	CH ₄ – 50% CO ₂ – 50%	CH ₄ – 90% CO ₂ – 10%

Current lab results are now in-line with our TEA process assumptions

DAC-DFM Process Scale-Up

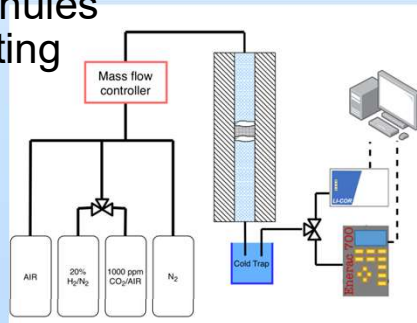


We are here

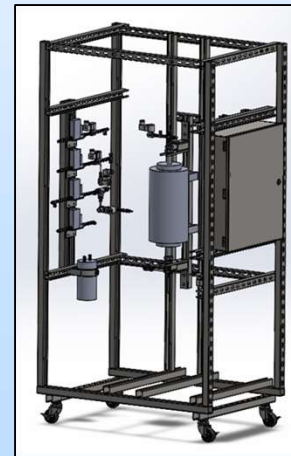


Proof of concept

DAC-DFM granules testing

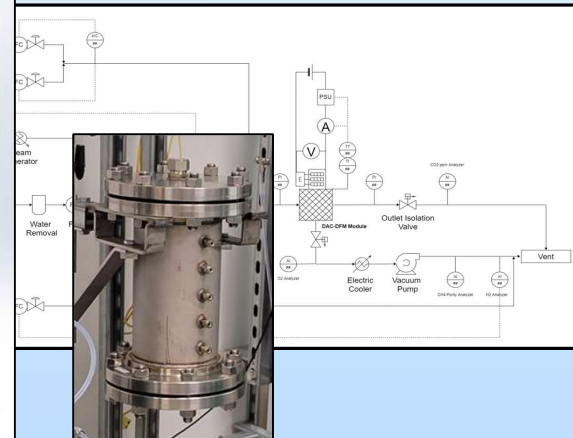


DAC-DFM washcoated monolith testing



Bench-scale unit testing

Pilot-scale demo unit



Summary and Key Findings

- ✓ Demonstrated a robust DFM washcoated monolith formulation with stable performance (>450 hours of testing) that is enhanced with humidity achieving up to 1.2 mmol CO₂/g_{DFM} capacity.
- ✓ Lowered the Ru content in the DFM from 1 wt% to <0.25 wt%, thus reducing the Capex and Opex of the process and reduced the cost of RNG production.
- ✓ Learned that CO₂ preferentially adsorbs on strong sites on DFM in the beginning of adsorption step followed by weaker sites towards the end of adsorption step. This led to operating near 50% capacity via 75% shorter adsorption step resulting in >90% conversion of CO₂ into CH₄.
- ✓ Demonstrated continuous Joule-heated DFM cycle testing for over 300 hours with heating rates up to 30°C/min.
- ✓ Fully integrated monolith supported DFM and reactor scale-up by 600X from 0.5 to 300 g/day
- ✓ TEA and LCA demonstrate commercial viability while maintaining a net negative carbon emissions on a cradle-to-gate basis.

Acknowledgements

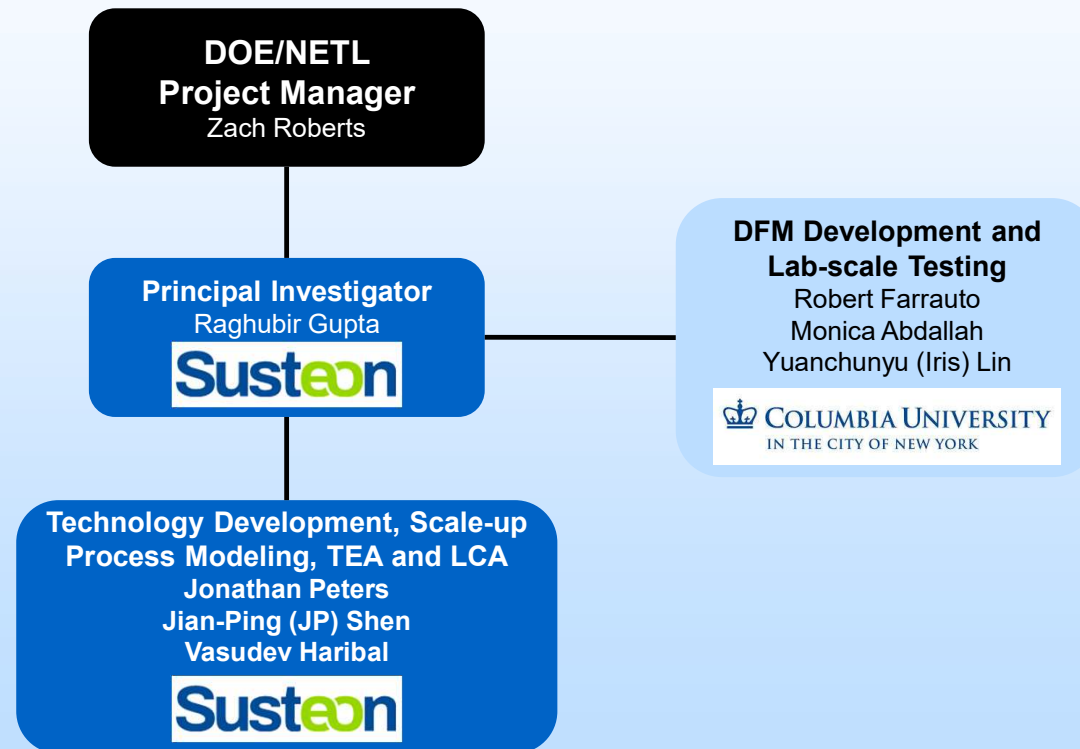
- Project Team
 - Susteon
 - Columbia University
- DOE/NETL
 - Zach Roberts
 - DE-SC0020795
- Industry Partners



Appendix

- These slides will not be discussed during the presentation **but are mandatory.**

Organization Chart



Gantt Chart

