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An Experimental and Computational Approach to Investigating CO₂ Uptake of Cellulose-Producing Algae from Cellulosic Ethanol Product (FE0032207)

Wafa Maftuhin, Bryan Wong, Charles M. Cai University of California - Riverside



Outline

- Short introduction
- Why use *Nannochloropsis* algae for CO₂ uptake
- Large-scale simulations for cellulose formation
- Preliminary work with density functional tight binding calculations
- What's next?



Introduction



Cai Group (Experimental)

Center for Environmental Research and Technology (CE-CERT)



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Wong Group (Computational)

Materials Science & Engineering (MSE) Building



Cellulosic ethanol





Nannochloropsis for CO₂ Uptake

- Nannochloropsis: cellulose-producing algae
- Can capture waste CO₂ from ethanol fermentation to produce
 - Algal lipids used for biodiesel production microcrystalline cellulose as high-value coproduct



Nannochloropsis for CO₂ Uptake

Economic & environmental benefits

- Capturing CO₂ from ethanol fermentations supports algal production and enables potential greenhouse gas savings
- Lignin generated as byproduct can provide heating to support cellulosic ethanol production & algal cellulose production
- Lignin is biogenic: its combustion is carbon neutral

Pretreatment Approaches

Co-Solvent Enhanced Lignincellulosic Fractionation (CELF) developed by experimental collaborator (Dr. Cai) to pretreat lignincellulosic biomass

CELF nondestructively fractionates lignin, oleic, & protein in any plantbased

Could be extended to algae fractionating & extract *nannochloropsis*



Lignin Precipitation, and Acid Neutralization



Process diagram flow

co-locating cellulosic ethanol fermentation with algal lipid and cellulose production





Life cycle analysis

Techno-economic & life cycle analysis will support quantitative outcomes



AspenOne software using modified GREET model for cellulosic ethanol will be used



Preliminary work with CELF

- Performed CELF pretreatment of industrial hemp and corn stover to prepare glucan-rich material for digestion by SSF (simultaneous saccharification and fermentation).
- CELF pretreated corn stover achieved >80% glucan, however, CELF pretreated industrial hemp only reached 75% (target 80%).
- We have begun to optimize CELF conditions to further improve glucan composition for industrial hemp. Once we
 have obtained similar glucan concentrations in both feedstock types, we can begin to measure CO₂ emission from
 fermentation.
- Our modeling team has successfully integrated CELF pretreatment, solids filtration, solvent recovery, and neutralization processes in AspenOne. We are currently optimizing heat recovery to calculate initial energy balance Our goal is to target total heat utilization of 4.5kwh/tonne biomass input.



Preliminary work with CELF



Figure 1. Knife-milled industrial hemp stalk subjected to mild CELF pretreatment

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Figure 2. SSF of CELF-pretreated industrial hemp at 100 g/L initial loading. Observed rapid solubilization of solids over first four days. Ethanol analysis under way.

Molecular modelling

- DFT good for small systems
- Classical molecular dynamics can handle large systems but are missing the quantum part
- DFTB merges reliability of DFT with computational efficiency of tight binding



Molecular modelling

Large-scale Density Functional Tight Binding (DFTB) calculations will probe cellulose formation





Cover of May 2019 issue (NETL Project DE-FE0030582)

S. I. Allec, Y. Sun, C.-en A. Chang, B.M. Wong *J. Chem. Theory Comput.* **15**, 2807 (2019)



DFTB: coarse-grained, parameterized DFT with atomic-centered basis functions





pre-tabulated as function of R_{AB}

simplifies integrals in SCF procedure



 $E_{DFTB} = \underbrace{E_{BS}} E_{Coul} + E_{rep}$



 $E_{BS} = \sum_{i}^{\nu cc} \sum_{\mu \nu} c_{\mu}^{a*} c_{\nu}^{a} H_{\mu \nu}^{0}$ **0CC**

 $H^0_{\mu\nu} = \left\langle \phi_\mu | H^0 | \phi_\nu \right\rangle$

$$S_{\mu\nu} = \int \phi_{\mu}(r)\phi_{\nu}(r)d^3r$$

parametrized beforehand from DFT calculations







Computational Materials Science 47 (2009) 237-253

3ob-3-1 skf parameter

param	e _d	e _p	e _s	U _d	U _p	U _s	f _d	f _p	f _s
С	0.0	-0.194	-0.505	0.365	0.365	0.365	0.0	2.0	2.0
н	0.0	0.0	-0.239	0.419	0.419	0.419	0.0	0.0	1.0
0	0.0	-0.332	-0.879	0.236	0.236	0.236	0.0	4.0	2.0
S	0.321	-0.258	-0.630	0.328	0.328	0.328	0.0	4.0	2.0

e = onsite energy, U = Hubbard U, f = occupation number

parametrized beforehand from DFT calculations





Pre-parameterization allows fast calculations of large systems

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Simulated CELF Pretreatment

Lignin subunits:



Model of lignin dimer β -O-4:





DFTB



- DFTB3 (3ob-3-1) + DFT-D3 disp effect
- PBC (mimic periodicity)
- DFTB relaxation (Force < 0.07 eV/A)
- Co-solvent:
 - 1:1 THF:Water v/v ratio
 - Acid (H_2SO_4)

GG dimer β -O-4 (2560 atoms)



Preliminary work with DFTB-MD



- GG dimers
- NVT (423 K): Nose-Hoover Thermostat
- @NERSC (National Energy Research Scientific Computing)



Preliminary work with DFTB-MD





Preliminary work with DFTB-MD

• SG dimer





Summary



CELF



DFTB enables fast predictions of cellulose formation



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Outlook

CELF Pretreatment

- Improve glucan composition for industrial hemp
- Measure CO₂ emission from fermentation
- AspenOne: optimizing heat recovery

DFTB-MD calculations for cellulose formations:

- Longer time scale
- Vary temperature

• H-H • S-S

• G-H • H-S



- Add more acid (SO_4H_2)

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Web: <u>http://bmwong-group.com</u>

E-mail: bryan.wong@ucr.edu

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Thank You

