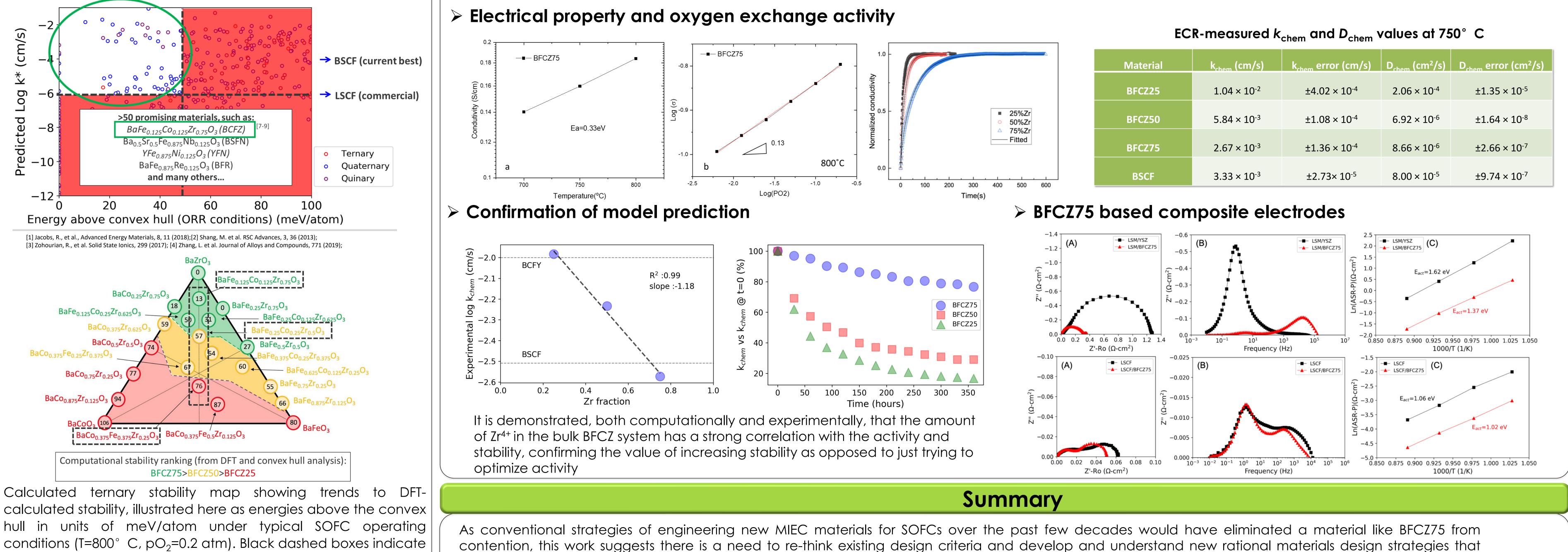
Unconventional Highly Active and Stable Oxygen Reduction Catalysts Informed by **Computational Design Strategies**

Jian Liu¹, Ryan Jacobs², Bo Guan^{1,3}, Tao Yang^{1,3}, Richard Pineault¹, Thomas Kalapos^{1,3}, Gregory Hackett¹, Harry Abernathy¹, Dane Morgan² 1. U.S. DOE, National Energy Technology Laboratory, Morgantown WV; 2. Department of Materials Science and Engineering, University of Wisconsin-Madison, WI; 3. Leidos Research Support Team, Morgantown WV;

Introduction

Discovering and engineering new materials with fast oxygen surface exchange kinetics and robust long-term stability is essential for the large scale, economically viable commercialization of solid oxide fuel cell (SOFC) technology. perovskite catalyst material BaFe_{0.125}Co_{0.125}Zr_{0.75}O₃ (BFCZ75), predicted to be promising from our recent density functional theory calculations and unconventional due to its extremely high Zr content and low electronic conductivity, exhibits oxygen reduction reaction surface exchange rates on par with $Ba_{0.5}Sr_{0.5}Co_{0.8}Fe_{0.2}O_3$ (BSCF) and excellent stability at typical operating temperatures. We engineer new composite electrodes integrating BFCZ75 with commercial electrode materials $La_{1-x}Sr_{x}MnO_{3}$ (LSM) and $La_{1-x}Sr_{x}Co_{y}Fe_{1-y}O_{3}$ (LSCF) and achieve high performance as measured by low area specific resistance (ASR) values.



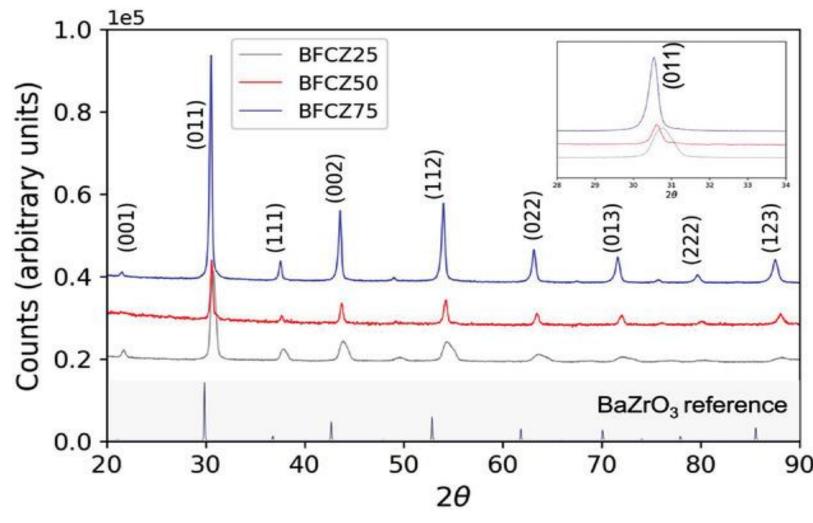


BFCZ material compositions selected for experimental evaluation in this work. For the composition BaCo_{0.375}Fe_{0.375}Zr_{0.25}O₃, the energy above convex hull value of 76 meV/atom was obtained from averaging over 12 different cation configurations, where the standard deviation of these 12 stability values is 6.8 meV/atom.



Synthesis of BFCZ phases and characterization





Powder XRD patterns of BFCZ materials showing that synthesis of BFCZ25 (black), BFCZ50 (red) and BFCZ75 (blue) all exhibit the cubic perovskite structure.

Synthesis: Sol-gel combustion

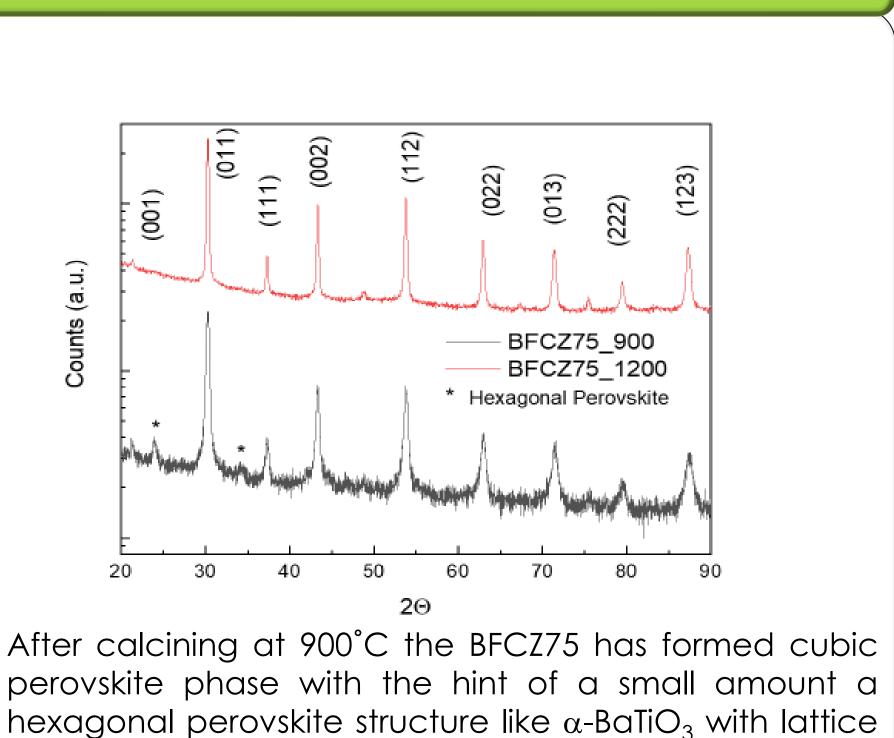
contention, this work suggests there is a need to re-think existing design criteria and develop and understand new rational materials design strategies that challenge conventional wisdom and chemical intuition. The ultimate goal of cathode design of SOFCs is to achieve a balance among electrical conductivity, ionic conductivity, oxygen exchange activity and long-term stability. Instead of searching for a replacement for unstable MIEC like LSCF, which works well in a ceria based composite electrode, some stable composites contain a highly conductive material mixing with a less conductive but highly ORR active and stable MIEC such as BFCZ75 or even triple-phase composites could be an alternative way to advance the current SOFC technology. (This work is published in May 2022 issue of Advanced Energy Materials. https://doi.org/10.1002/aenm.202201203 -->)

• This project was funded by the Department of Energy, National Energy Technology Laboratory an agency of the United States Government, through a support contract. Neither the United States Government nor any agency thereof, nor any of its employees, nor the support contractor, nor any of their employees, makes any warranty expressor implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by trade name, trademark, manufacturer, or otherwise does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof.

Research & Innovation Center



Results



parameters a \sim 5.22A and c \sim 12.82A.

Material	k _{chem} (cm/s)	k _{chem} error (cm/s)	D _{chem} (cm²/s)	D _{chen}
BFCZ25	1.04×10^{-2}	$\pm 4.02 \times 10^{-4}$	2.06 × 10 ⁻⁴	±
BFCZ50	5.84 × 10 ⁻³	$\pm 1.08 \times 10^{-4}$	6.92 × 10 ⁻⁶	±
BFCZ75	2.67 × 10 ⁻³	±1.36 × 10 ⁻⁴	8.66 × 10 ⁻⁶	±
BSCF	3.33 × 10 ⁻³	±2.73× 10 ⁻⁵	8.00 × 10 ⁻⁵	±
	BFCZ25 BFCZ50 BFCZ75	BFCZ25 1.04 × 10 ⁻² BFCZ50 5.84 × 10 ⁻³ BFCZ75 2.67 × 10 ⁻³	BFCZ25 1.04×10^{-2} $\pm 4.02 \times 10^{-4}$ BFCZ50 5.84×10^{-3} $\pm 1.08 \times 10^{-4}$ BFCZ75 2.67×10^{-3} $\pm 1.36 \times 10^{-4}$	BFCZ25 1.04×10^{-2} $\pm 4.02 \times 10^{-4}$ 2.06×10^{-4} BFCZ50 5.84×10^{-3} $\pm 1.08 \times 10^{-4}$ 6.92×10^{-6} BFCZ75 2.67×10^{-3} $\pm 1.36 \times 10^{-4}$ 8.66×10^{-6}

Science & Engineering To Power Our Future

