

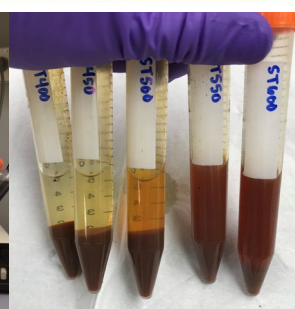
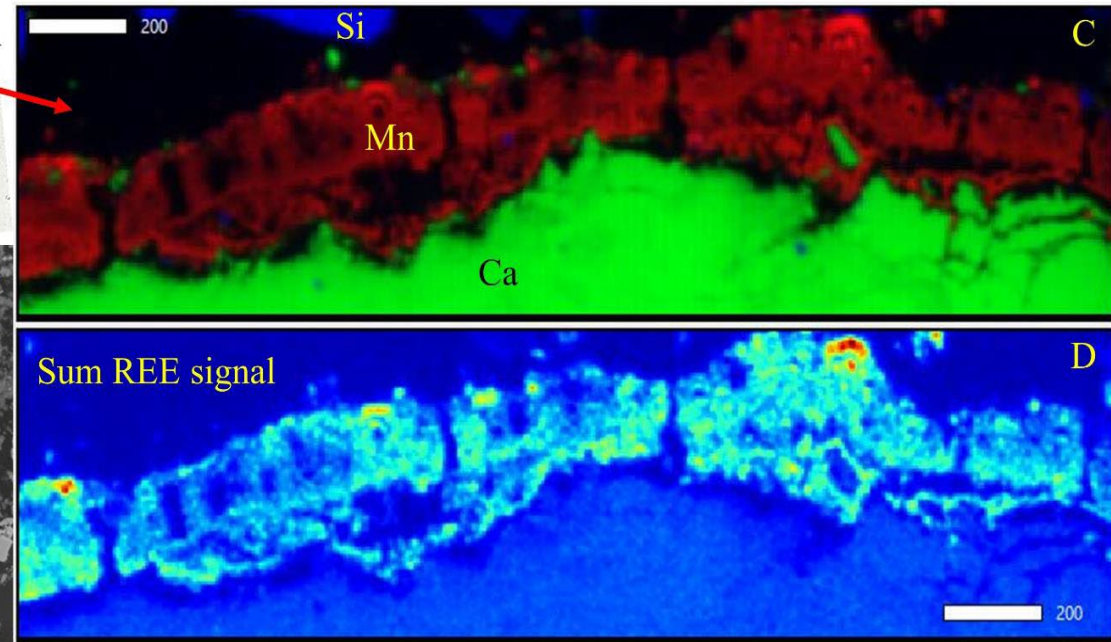
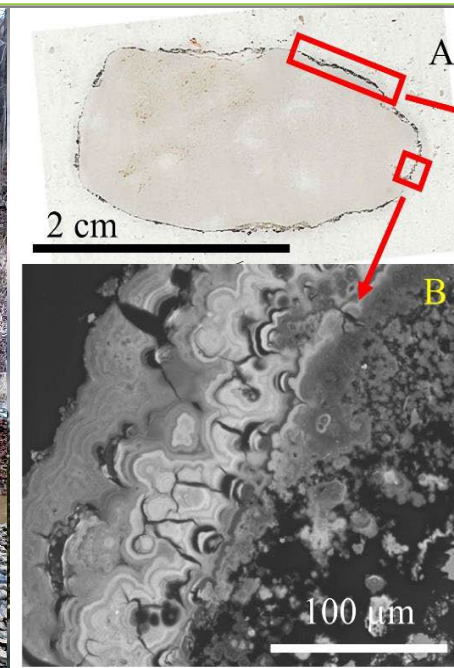
Characterization-Informed Recovery of Critical Minerals from Acid Mine Drainage Treatment Solids

FWP-1022420



Mengling Stuckman, Ph.D.

National Energy Technology Laboratory (NETL), NETL Support Contractor



U.S. Department of Energy
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Resource Sustainability Project Review Meeting
Oct. 25-27, 2022

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Project PIs and Presenter*

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Alison Fritz¹, Bret Howard¹, Tom Tarka¹

¹National Energy Technology Laboratory, 626 Cochran Mill Road, Pittsburgh, PA 15236, USA

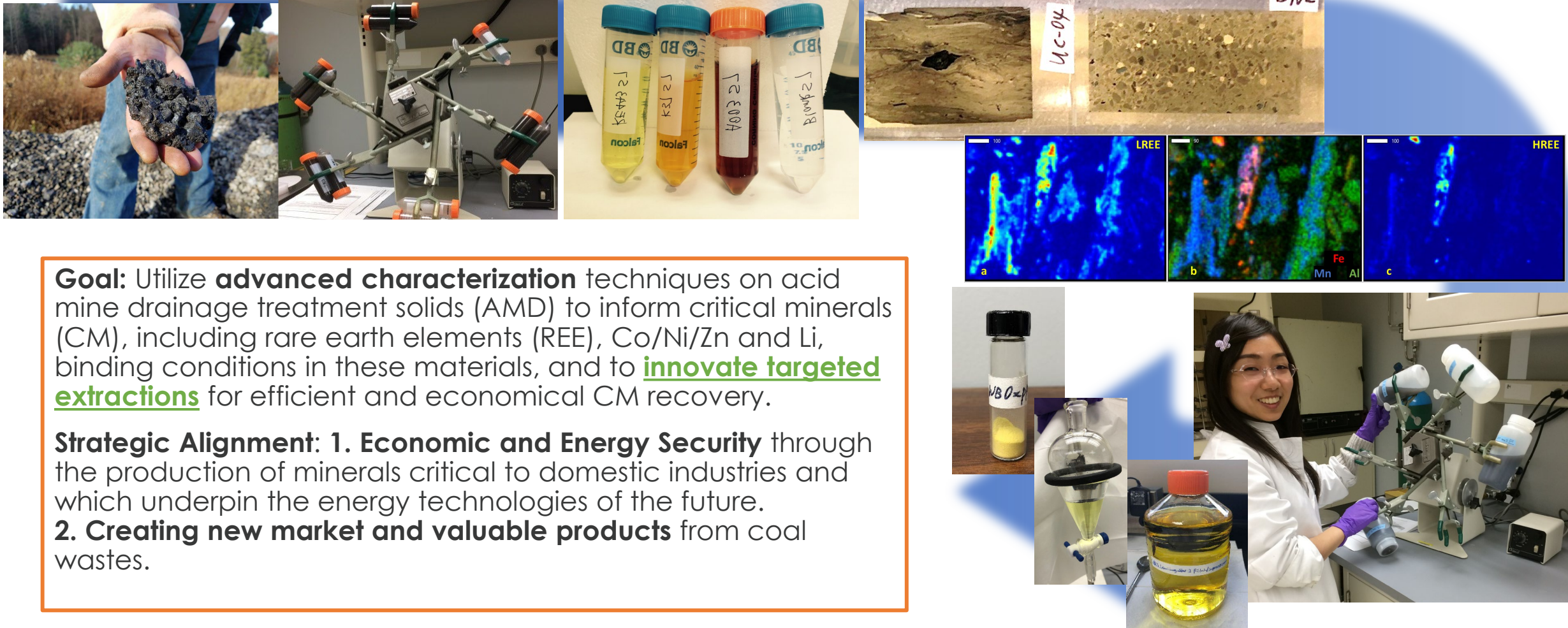
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⁴University of Pittsburgh, 4107 O'Hara Street, Pittsburgh, PA 15260

Project Overview

Funding from NETL Critical Mineral Research Field Work Proposal: \$365k for EY2022, TBD for EY2023 and 2024; Overall Project Performance Dates: EY2022-2024



Goal: Utilize **advanced characterization** techniques on acid mine drainage treatment solids (AMD) to inform critical minerals (CM), including rare earth elements (REE), Co/Ni/Zn and Li, binding conditions in these materials, and to **innovate targeted extractions** for efficient and economical CM recovery.

Strategic Alignment: 1. **Economic and Energy Security** through the production of minerals critical to domestic industries and which underpin the energy technologies of the future.
2. **Creating new market and valuable products** from coal wastes.

Chemically-Informed CM Recovery from AMD Solids

- Technical advances:
 - **Expand CM focus: REE, Co/Ni/Zn, Mn, Al, Fe, Li, etc.**
 - **80% reduction** in reagents, reaction time
 - Reduction in process steps/complexity: **sequential leach of different CMs**
 - Reduction in process intensity: **ambient temperature and pressure**
 - Techno-economic analyses in progress to further explore process performance
- Fundamental and lower technical readiness level (TRL) for innovations:
 - **Work smarter, not harder**
 - Fine-tune extractions to target certain CM
 - Novel reagents and extractions: Different acids, different agents, cocktails, waste streams
 - Modeling to improve design of treatment
- Industry input/validation:
 - Regular communication with industry partners
 - Active discussions on technology licensing and industry needs

Background: Acid Mine Drainage (AMD) is Enriched in CM

Domestic Source of CM

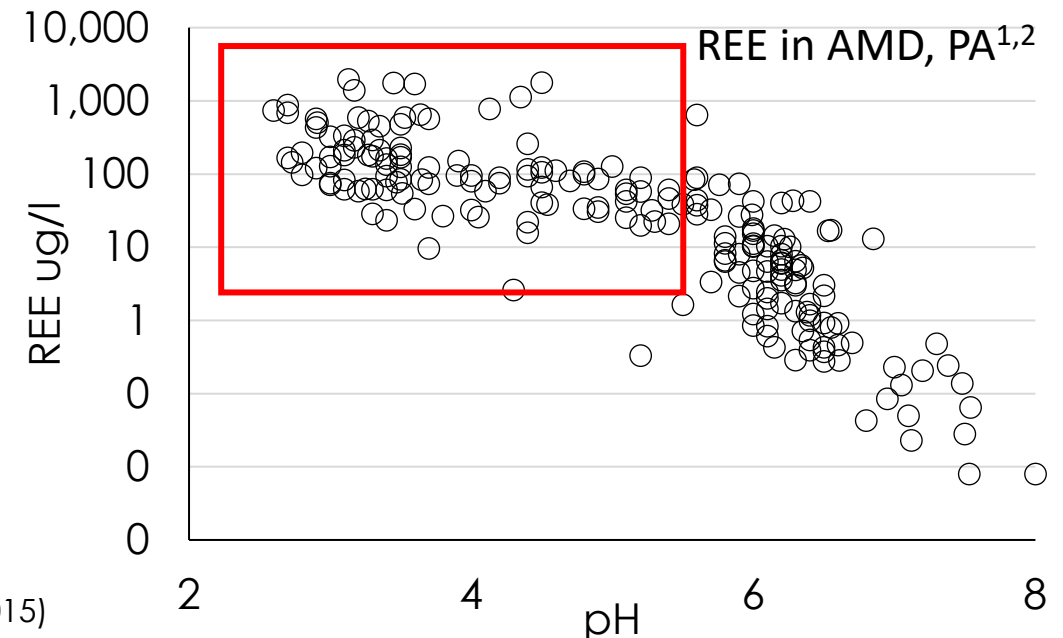
Pyrite (FeS₂) oxidation releases hydrogen ions

Decreases pH, mobilizes metals (e.g., Fe, Mn, Al)

Need to treat toxic levels of metals that negatively effect the water, including REE, Co and Ni under acidic conditions

CMs from 140 discharges across Pennsylvania²

Element	Max conc. (ug/L)	Min conc. (ug/L)	Max loading (kg/year)	Min loading (kg/year)
Mg	210,000	3,600	3,541,140	40
Mn	74,000	19	215,522	4.5
Sr	3,600	27	83,321	0.23
Ni	3,200	2.6	10,428	0.3
REE	1,765	0.4	7,364	<0.01
Co	3,100	0.3	6,952	0.1
Cu	190	0.4	2,086	<0.1
Li	390	11.0	4,513	0.2



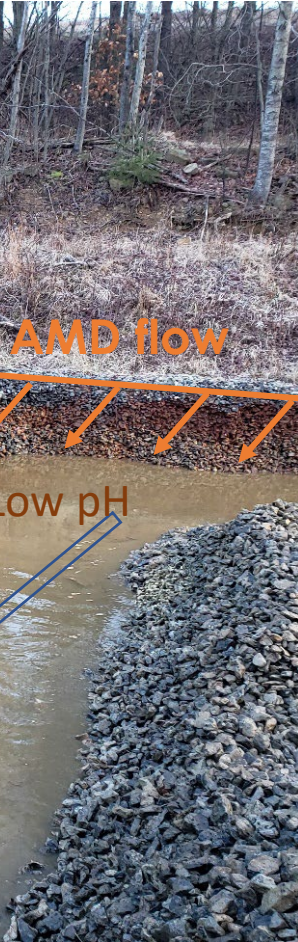
1: Hedin et al. (2019); Stewart et al. (2017); 2: Cravotta (2008); Cravotta and Brady (2015)

AMD Treatment Systems and REEs

Hollywood AMD Treatment Facility
Bennett Branch Watershed
Clearfield County

Active systems PA DEP

\$14.6M construction cost (Cap. Budget + GG + AMD Set-aside)



Passive Remediation Treatment: No chemical added, >200 systems in PA

~85 billion gallons/year AMD treated

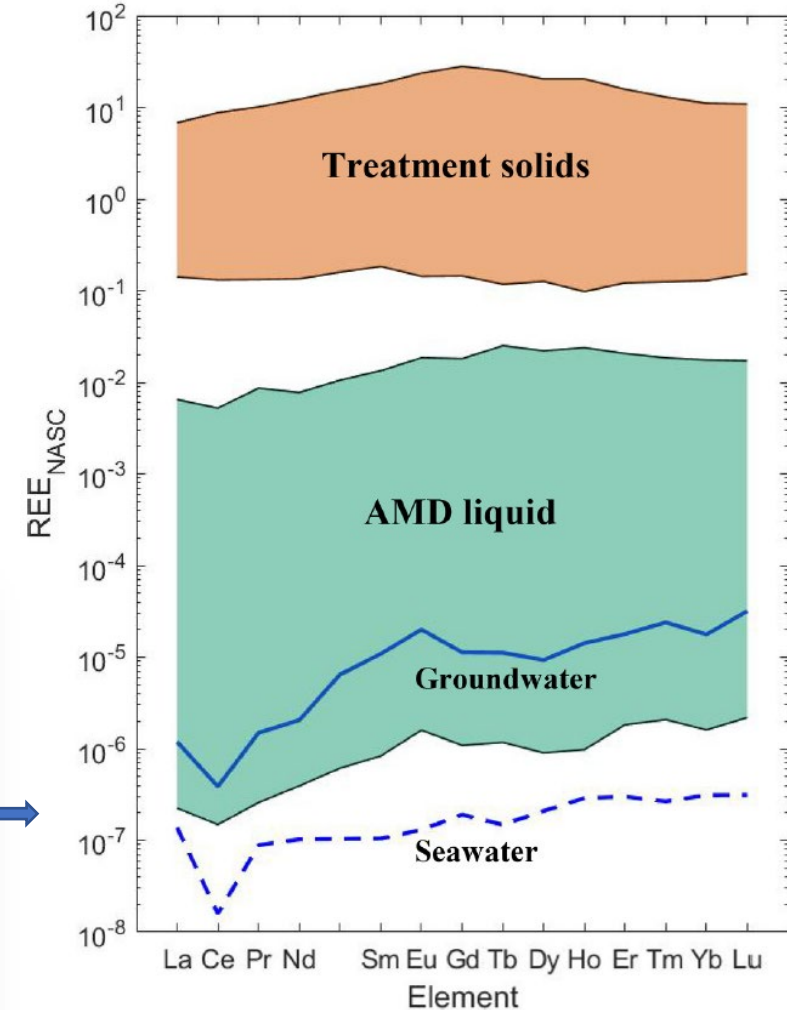
Raises pH of water (Limestone beds)
Precipitate dissolved metals

- 90% REE sequestration REEs precipitate with Fe, Mn, Al

~18,000 tons/year treatment solids produced: AMD solids



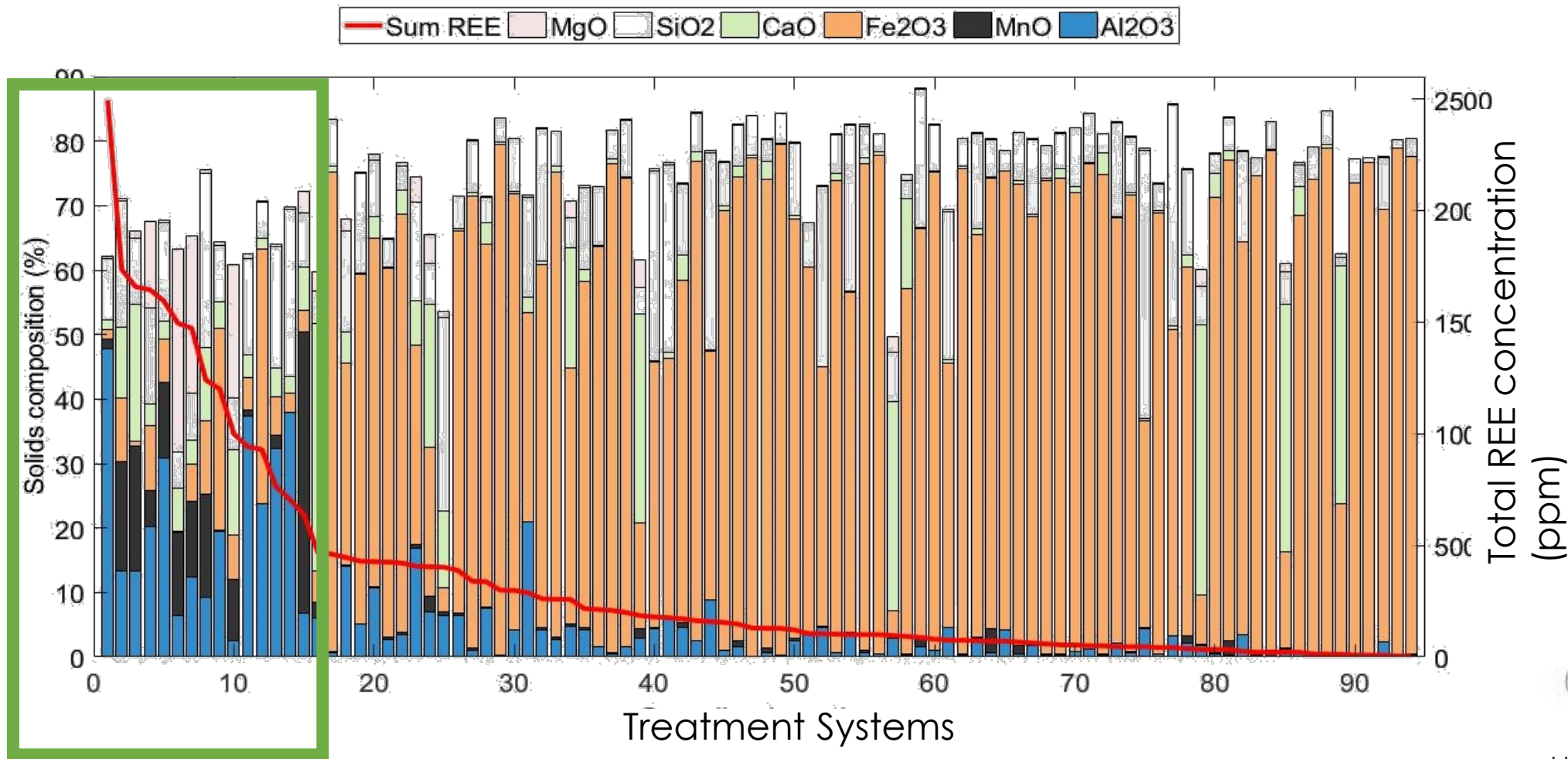
AMD solids
Hedin Env.



Hedin et al. (2019 & 2020), IJCG

Approximately 100 AMD Solids from Passive Systems in PA

Hydrated oxides/hydroxides (e.g., $\text{Al}(\text{OH})_3$, $\text{FeO}(\text{OH})$)



- ~50% critical REE (Eu, Nd, Tb, Dy, and Y)
- <50 ppm U, Th
- \$3 to \$400 REE value and \$0.04 to \$217 Co value/metric ton dry solid

Challenge of the Project: Chemical Diversity

Unit: wt% for major elements and mg/kg for trace elements

	C	S	Al	Si	Fe	Mn	Mg	Ca	K	Ti	REE	Li	Co	Ni	Cu	Zn
Al-rich solid	2%	2%	18.0%	19.3%	2.1%	0.1%	0.2%	1.2%	0.7%	0.3%	1113	38	22	50	106	315
MnCa-rich solid	4%	ND	3.5%	6.1%	0.5%	18.1%	0.6%	16.8%	0.4%	0.1%	1590	108	6026	8889	89	13585
AlMnFe-rich solid	1%	1%	15.4%	9.7%	5.2%	8.5%	0.2%	2.8%	0.3%	0.1%	1900	440	2059	3002	518	5812

The **transition metal** contents are sometimes higher than REE; **Lithium** content is also reasonably high MnCa-rich solid has higher accumulation of Co, Ni and Zn



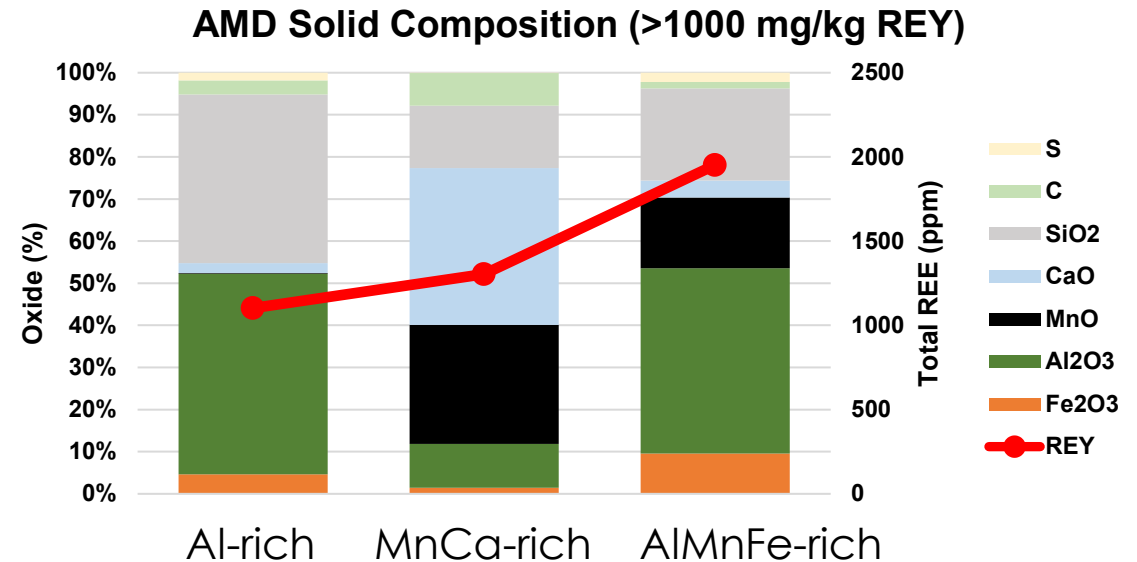
Al-rich solids



MnCa-rich solid



Al-, Mn-, Fe-rich solid



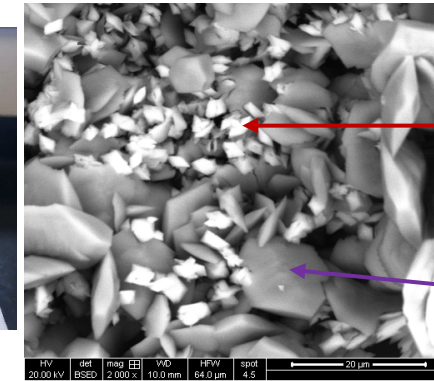
Stuckman, PLOGA Tech seminar, 2021

Hedin et al., IJCG, 2019

Research Success Metric in Previous Years

Yearly Progression Demonstrated

- AMD waste solids
- Est. 1102 metric tons REO/yr
- 12% of U.S. annual demand

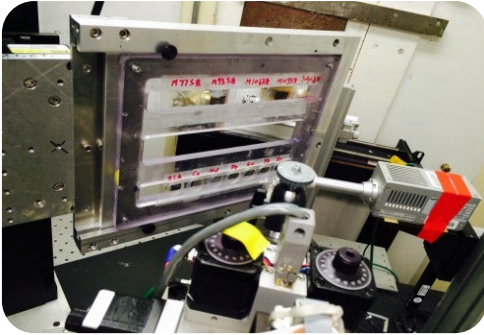


REY
oxalates

(Mn, Ni, Zn,
Co)O

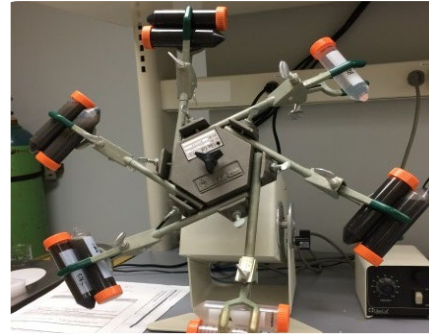
EY	Wt% REOs	TRL Level	%Recovery	Notes
2018	9%	1 - 2	60%	Mn and Fe impurities, with <u>10%Co, 26%Ni and 21%Zn</u>
2019	4%	2	80%	Fe and Ca impurities were further addressed, leachate underwent L:L separation for comparison
2019	13%	2 - 3	> 80%	Reduced impurities <u>with 48% Zn, 11%Ni, 2%Co</u> Additional AMD solid field sampling (early 2020)
2020	95% (mg scale)	2 – 3	>80%	ROIs in prep ; By-product recovery in development; Upscaling and optimization of lab scale processes
2022- 2024	Co, Ni, Zn	3 - 4	>80%	Collaboration with Hedin Environmental, Inc.; Explore biotechnologies to sequester, recover and purify REEs and critical elements in planning

Characterization Informed Recovery



EY22 Characterization of CM Binding

- CM distribution:
In-house SEM, XRD, Raman, **synchrotron-based spectroscopy, seven-step sequential extraction**
- Geochemical modeling (PHREEQ-N-AMDTreat from USGS)



EY23 Optimized Targeted Extraction

- Sequential leach
- Innovative leaching strategies

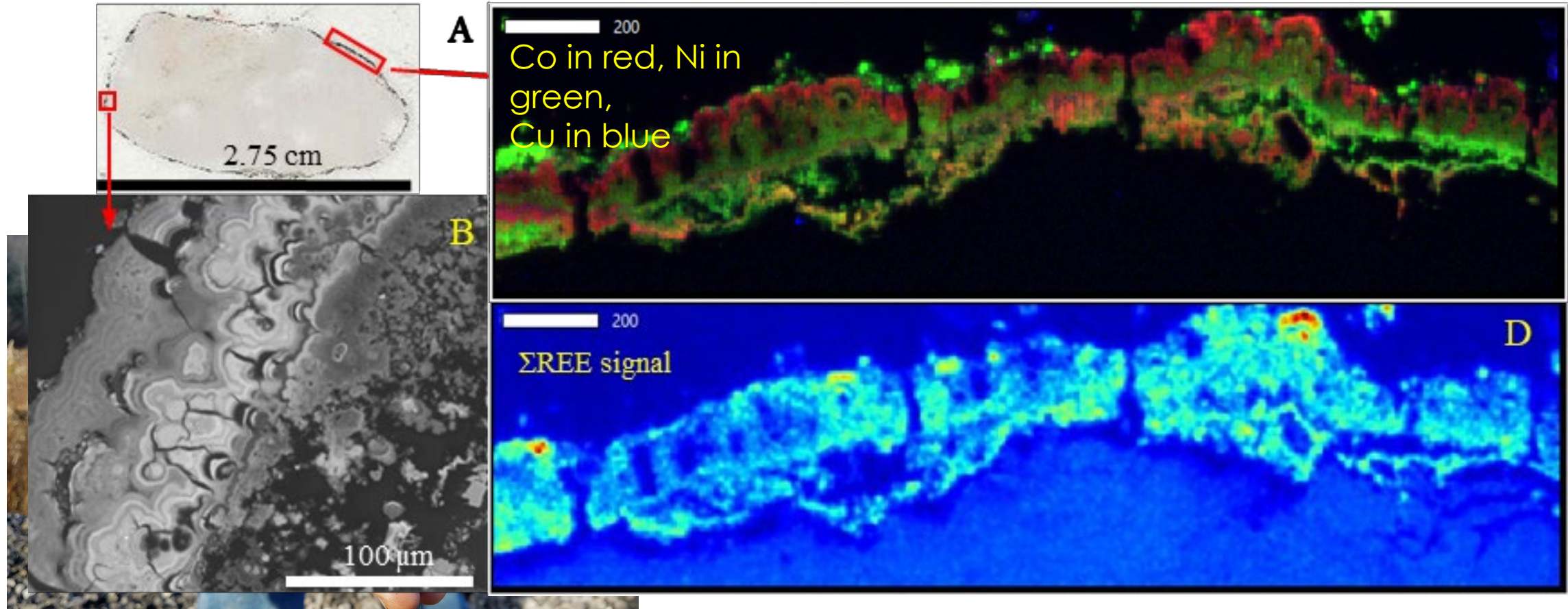


EY24 Downstream Optimization and Operation Optimization

- Optimization of the extraction
- Explore to bypass costly traditional solvent purifications further downstream

Progress: The REE and CM Binding in MnCa Solid

Synchrotron μ XRF

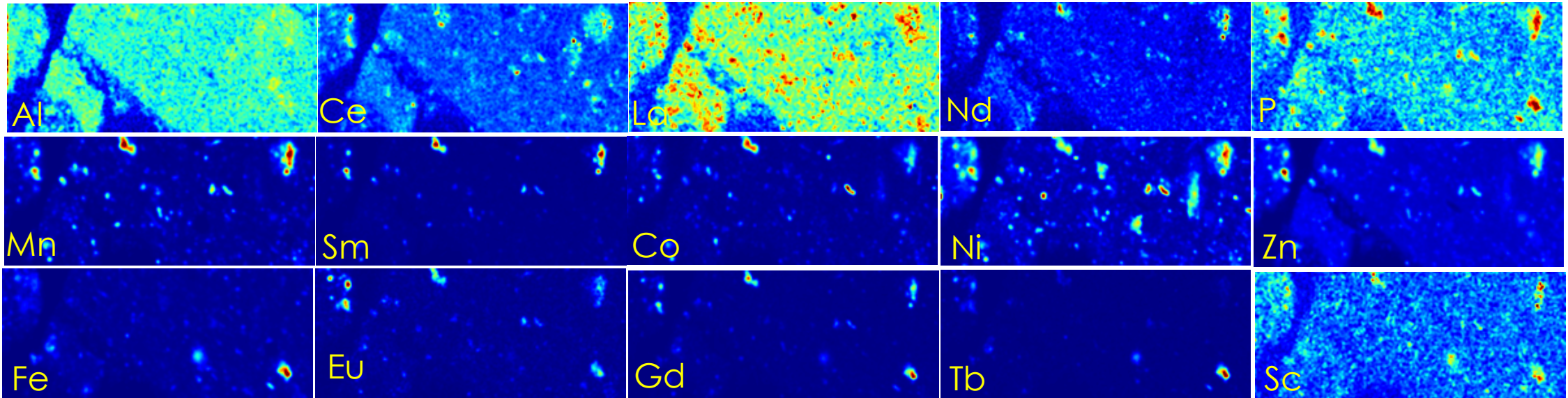
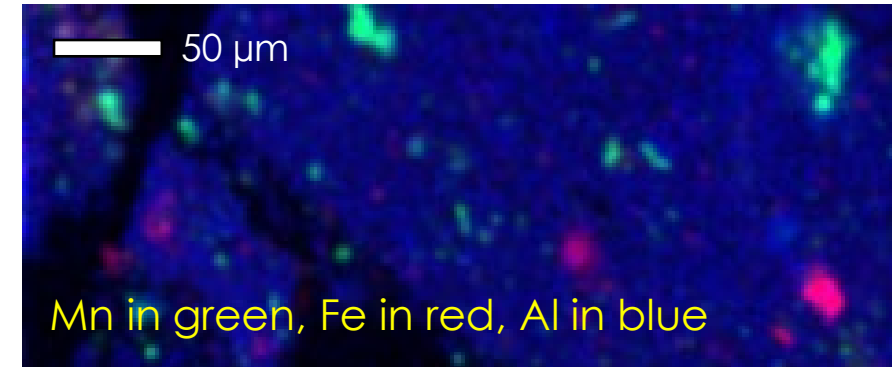


- REE associated with Mn layer in CaMn-rich solid

Stuckman, PLOGA Tech seminar, 2021



Progress: The REE and CM Binding in Al-Rich Solid



- REEs Co-localized with Al and Mn, selected heavy REEs (Gd, Tb) co-localized with Fe
- Co, Ni, Zn co-localized with Mn

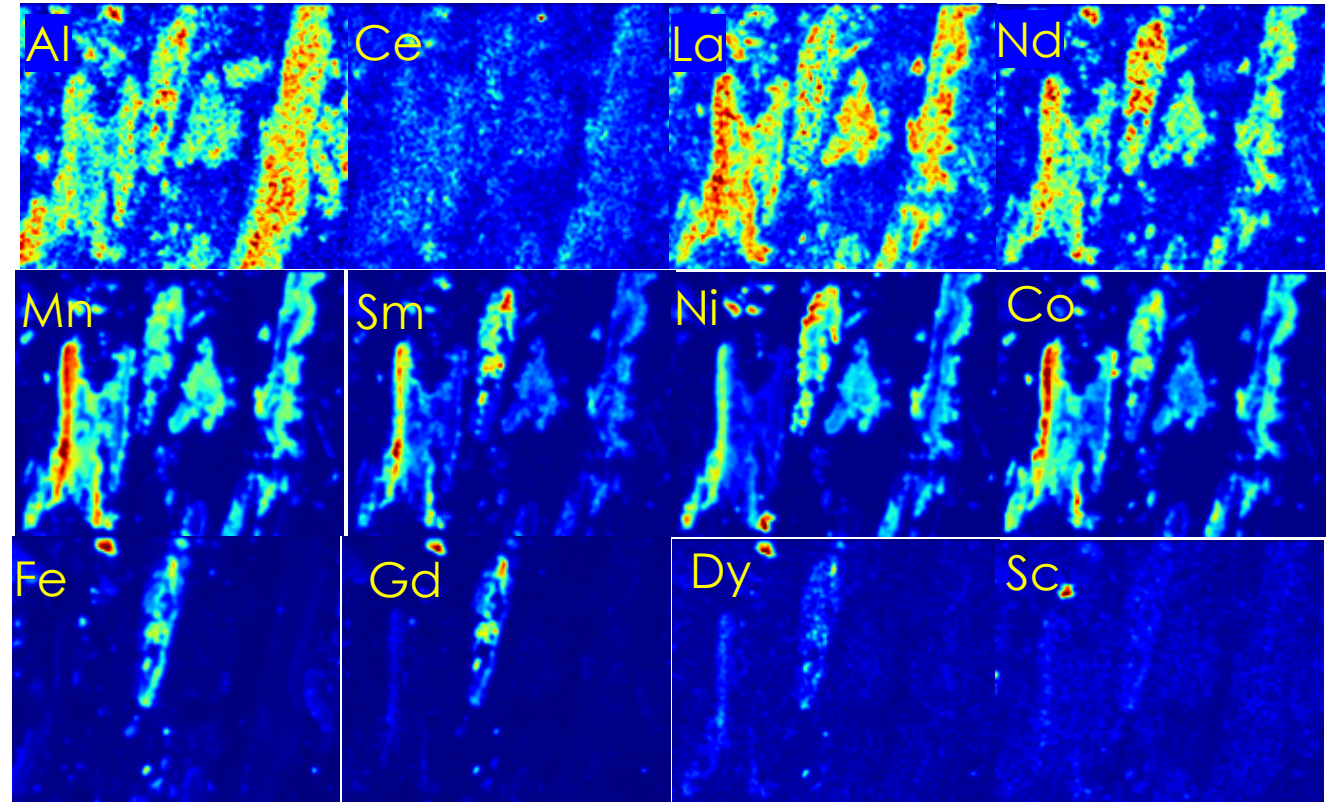
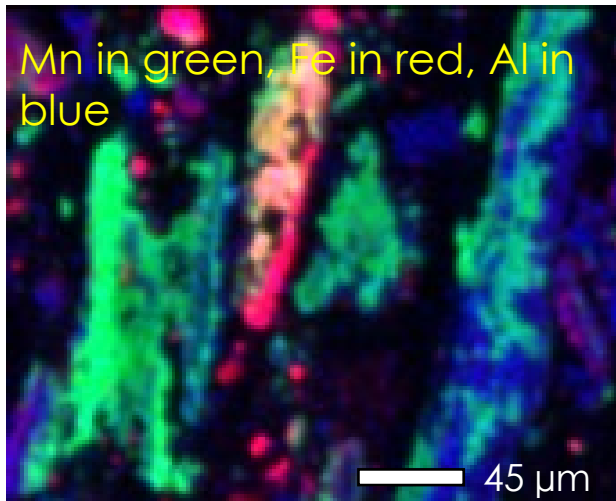
Stuckman, PIOGA Tech seminar, 2021

Progress: The REE and CM Binding in AlMnFe-Rich Solid

- REEs Co-localized with Al and Mn, selected heavy REEs (Gd,Dy) co-localized with Fe
- Co, Ni, Zn co-localized with Mn



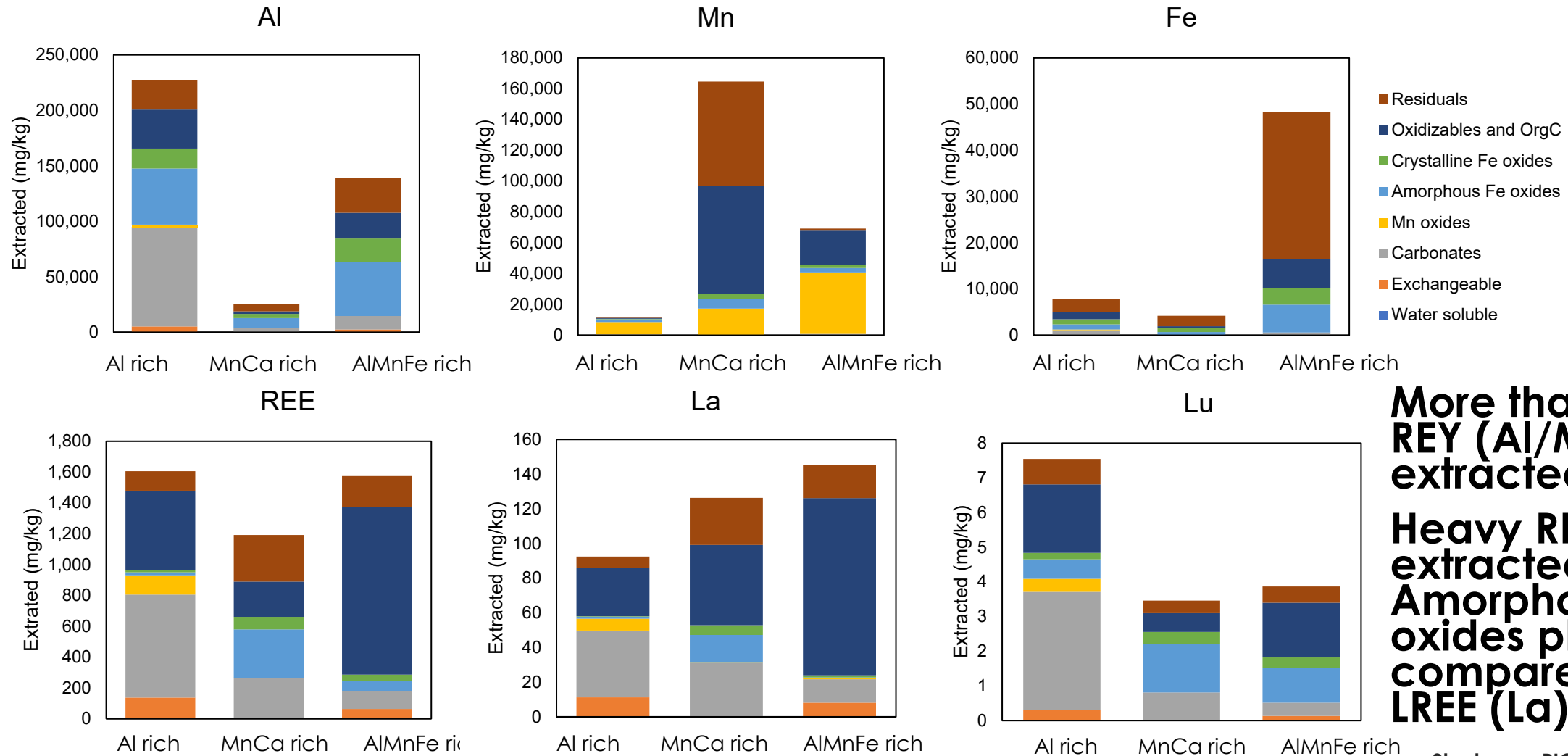
Al-, Mn-, Fe-rich solid



Stuckman, PIOGA Tech seminar, 2021

90 μm

Progress: REE Sequential Extractions (Total Exacted vs. Residuals)

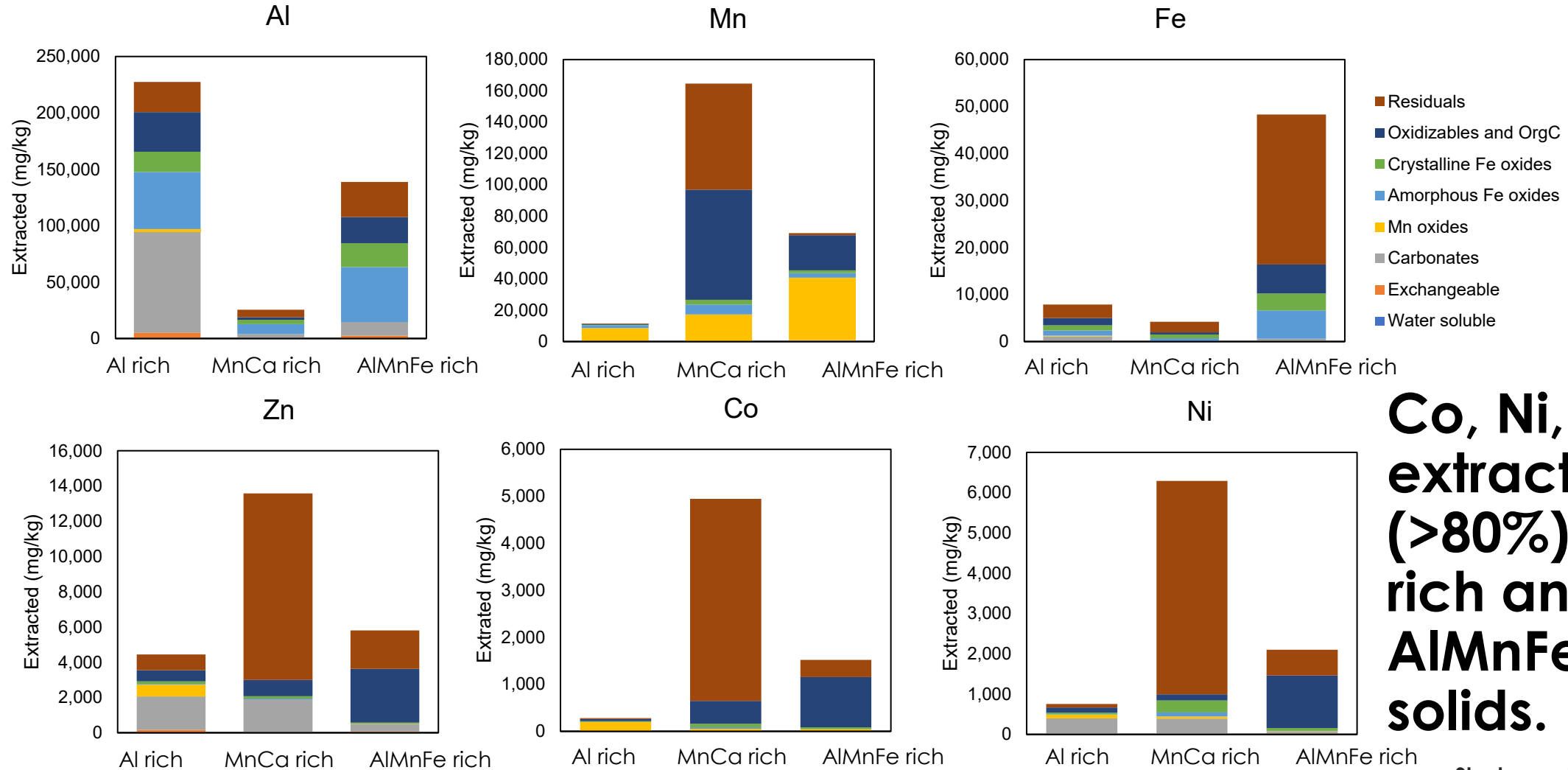


**More than 80%
REY (Al/Mn)
extracted**

**Heavy REE (Lu)
extracted in
Amorphous Fe
oxides phases
compared to
LREE (La)**

Stuckman, PIOGA Tech seminar, 2021

CM Sequential Extractions (Total Exacted vs. Residuals)



pH from 6 to 2

**Co, Ni, Zn
extractable
(>80%) in Al
rich and
AlMnFe rich
solids.**

Stuckman, PIOGA Tech seminar, 2021

Progress: Geochemical Modeling

PHREEQ-N-AMDTreat developed by C. Cravotta (2022)

- Current focus on REE prediction for one type of treatment system
- Future efforts will incorporate other CMs in existing modeling

Select Workspace: C:\Users\hedin\Documents\AMDTitrationREYs_watex

	Soln#A	Soln#B	Soln#A	Soln#B
Design flow (gpm)	49.4	0	As (ug/L)	2.47 1E-08
Mix fraction	1	0	Ba (ug/L)	8.9 1E-08
Temp (C)	13.5	0.01	Cd (ug/L)	34.4 1E-08
SC (uS/cm)	5550	0	Co (ug/L)	4770 1E-08
DO (mg/L)	5.9	0.01	Cr (ug/L)	21.2 1E-08
pH	3	3	Cu (ug/L)	358 1E-08
Acidity (mg/L)	982	0	Ni (ug/L)	5110 1E-08
<input type="checkbox"/> Estimate NetAcidity	1080.3	0	Pb (ug/L)	9.8 1E-08
Alk (mg/L)	0	0	Sc (ug/L)	149 1E-08
TIC (mg/L as C)	19.2	0	Se (ug/L)	19.3 1E-08
<input type="checkbox"/> Estimate TIC	1.2	0	Sr (ug/L)	520 1E-08
Fe (mg/L)	40.7	1E-08	U (ug/L)	35.4 1E-08
Fe2 (mg/L)	29.6	0	Zn (ug/L)	18800 1E-08
<input type="checkbox"/> Estimate Fe2	0	0	La (ug/L)	201 1E-08
Al (mg/L)	128	1E-08	Ce (ug/L)	350 1E-08
Mn (mg/L)	129	1E-08	Pr (ug/L)	66.4 1E-08
SO4 (mg/L)	5000	1E-06	Nd (ug/L)	235 1E-08
Cl (mg/L)	1.9	0	Sm (ug/L)	79.7 1E-08
Ca (mg/L)	422	1E-06	Eu (ug/L)	23.1 1E-08
Mg (mg/L)	652	1E-06	Gd (ug/L)	99.3 1E-08
Na (mg/L)	17.8	0	Tb (ug/L)	21.3 1E-08
K (mg/L)	3.46	0	Dy (ug/L)	122 1E-08
Si (mg/L)	30.8	0	Ho (ug/L)	24.3 1E-08
NO3N (mg/L)	0.25	0	Er (ug/L)	67.4 1E-08
PO4P (mg/L)	0.01	1E-11	Tm (ug/L)	8.85 1E-08
F (mg/L)	0.5	0	Yb (ug/L)	54.4 1E-08
DOC (mg/L as C)	2	0	Lu (ug/L)	7.82 1E-08
Oxalate (mg/L as C)	0.1	1E-11	Y (ug/L)	600 1E-08

HMeO:mg 0 Fe% 14 Mn% 43 Al% 43
 SPECIFIED CONSTANT SORBENT (EXISTING)
 HMeO (mg/L Fe+Mn+Al, not oxides); existing, added to fresh HMeO ppt from soln
 <-Surface area, m2/g 64122.6 79047.7 5304.4 Surface area, m2/mol, comp.
 <-Site density, sites/rm2 1.925 1.91 4.6
 0.2 0.0903 0.0405 <-Site density (weak or y), mol/mol, computed
 0.005 0.1605 <-Site density (strong or x), mol/mol, computed
 HFO HMO HAO
 FRESHLY PRECIPITATED SORBENT (ADDITIONAL)
 600 746 68 <-Surface area, m2/g 64122.6 79047.7 5304.4 Surface area, m2/mol, comp.
 1.925 1.91 4.6 <-Site density, sites/rm2
 0.2 0.0903 0.0405 <-Site density (weak or y), mol/mol, computed
 0.005 0.1605 <-Site density (strong or x), mol/mol, computed

Specified Saturation Index Value at Which Precipitation Will Occur--ADDED TO SORBENT

SI_Fe(OH)3 0.0 SI_Al(OH)3 0.0 SI_MnOOH 0.0
 SI_Schwertmannite 99 SI_Basakuminite 0.0 SI_Mn(OH)2 0.0
 SI_Fe(OH)2 0.0 SI_CaCO3 2.5 SI_FeCO3,MnCO3 99

Specified Saturation Index Value at Which Precipitation of REE Will Occur--COMPETES WITH SORPTION

SI_REE(OH)3 0.0 SI_REE(CO3)1.5 0.0 SI_REE(PO4) 99 SI_REE(C2O4)1.5 99

Select titrant:
☒ NaOH 6 wt% soln ☐ Ca(OH)2 ☐ CaO ☐ Na2CO3 ☐ CaCO3 Maximum pH (<=11): 11

RUN MODEL

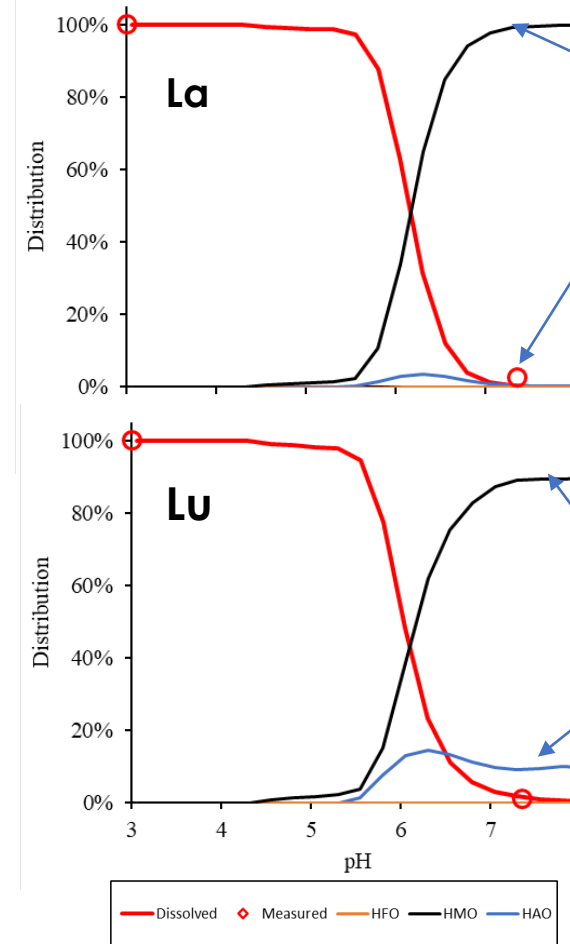
Select output matrix to be saved and graphs to display

☐ Short Output File ☒ Long Output File ☐ Print PHREEQC Output Report

☒ Plot REYs_HMeO ☒ Plot REYs_ppt
☐ Plot Sc ☐ Plot Y ☒ Plot La ☐ Plot Ce ☐ Plot Pr ☐ Plot Nd ☐ Plot Sm ☐ Plot Eu
☒ Plot Gd ☐ Plot Tb ☐ Plot Dy ☐ Plot Ho ☐ Plot Er ☐ Plot Tm ☐ Plot Yb ☒ Plot Lu
☐ Plot Cations_HMeO ☐ Plot Anions_HMeO ☐ Plot Alkalinity ☐ Plot Al ☐ Plot Fe ☐ Plot Mn
☐ Plot Ca ☐ Plot Mg ☐ Plot Ba ☐ Plot Sr ☐ Plot Cd ☐ Plot Co ☐ Plot Cr ☐ Plot Cu
☐ Plot Ni ☐ Plot Pb ☐ Plot Zn ☐ Plot U ☐ Plot As ☐ Plot Se ☐ Plot PO4 ☐ Plot SO4

CausticTitrationMix2REYs.exe created by C.A. Cravotta III, U.S. Geological Survey. Release version 1.0.0. June 2022

Different sorption for La and Lu



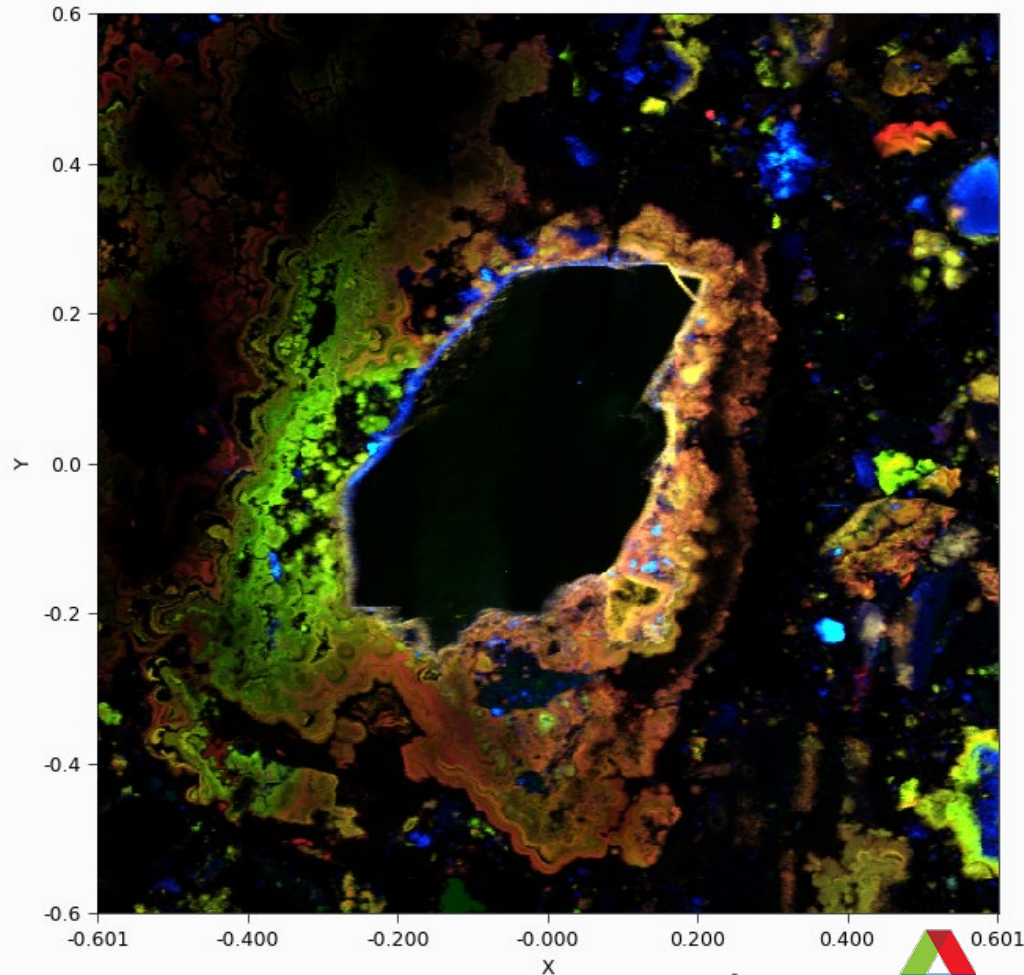
<https://code.usgs.gov/water/phreeq-n-amdtreat>

Hedin et al., ICARD, 2022

Characterization Summary

Mn in red, Ni in green, Co/Fe in blue

Image 1.2 mm x 1.2 mm, 2 mm spot (APS GSECARS, Sector 13)

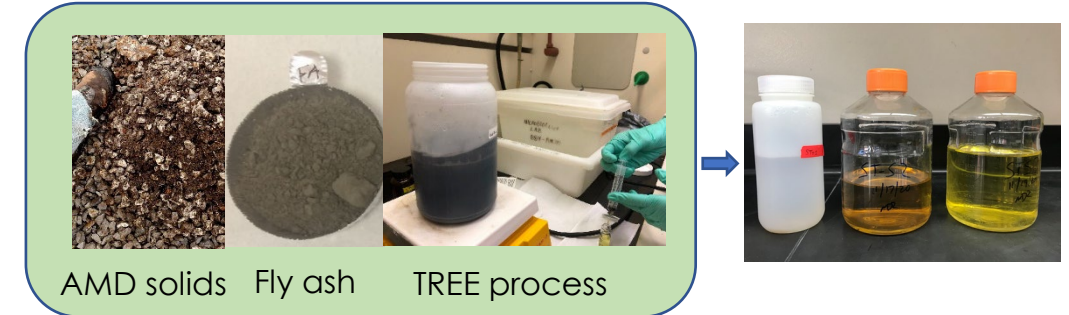


- **AMD solids have diverse chemical composition, so extractions may need to be tailored toward different chemical compositions.**
 - REEs mostly co-localized with Al and Mn, selected HREEs (Gd/Dy) co-localized with Fe
 - Regardless of composition, Co, Ni, Zn mainly co-localized with Mn (hydr)oxides in AMD solids
- **Disseminate results:** 4 peer review journals published, 3 in prep, report(s) of invention in progress
- Hedin, B., Stuckman, M., Lopano, C., Cravotta, C., and Capo, R. "Characterization and modeling of REE binding behaviors in passive acid mine drainage remediation solids" In preparation for submission to *Applied Geochemistry*
- Miller, J.D., Stuckman, M.Y., Means, N., Lopano, C.L., and Hakala, J.A. (2022) "Determination of transition metal ions in fossil fuel associated wastewaters using chelation ion chromatography" *Journal of Chromatography A*, 1668, 462924. <https://doi.org/10.1016/j.chroma.2022.462924>
- Stuckman, M., Lopano, C.L., and Tarka, T. (2022) "Step Leaching Process of Rare Earth Elements from Ash Materials Using Mild Inorganic Acids at Ambient Conditions", U.S. Patent: US 2022/0017992 A1. <https://patents.google.com/patent/US20220017992A1/en>
- Fritz, A. G., et al. (2021). "Technoeconomic Assessment of a Sequential Step-Leaching Process for Rare Earth Element Extraction from Acid Mine Drainage Precipitates." *ACS Sustainable Chemistry & Engineering* 9(28): 9308-9316.
- Hedin, B. C., et al. (2020). "Critical metal recovery potential of Appalachian acid mine drainage treatment solids." *International Journal of Coal Geology* 231: 103610.
- Hedin, B. C., et al. (2019). "The evaluation of critical rare earth element (REE) enriched treatment solids from coal mine drainage passive treatment systems." *International Journal of Coal Geology* 208: 54-64.

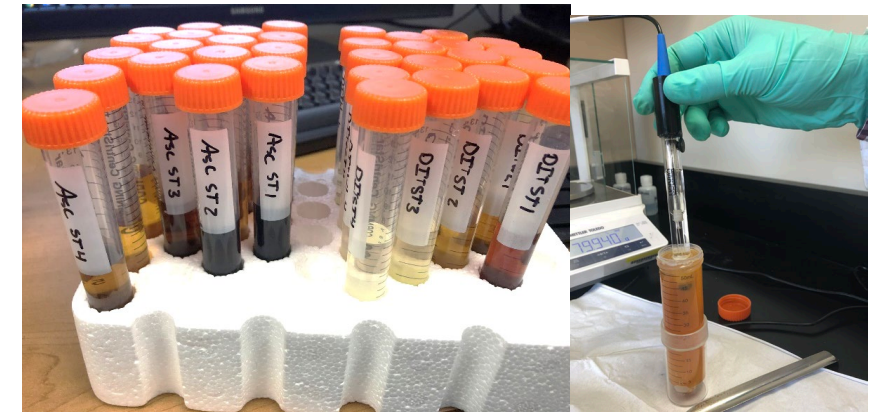
Progress: Optimized Targeted Extraction

Update: Innovative Extractions

- Optimized sequential leach
 - Targeted Rare Earth Extraction (TREE)*
 - 90% REE and up to 70% Co recovered in separate, targeted steps
- Extraction utilizing REDOX
 - Preliminary results showed up to 80% CM recovery in one extraction using neutral pH extractant, report of invention in preparation
 - Optimization of conditions needed



*Stuckman, M.Y., Lopano, C.L. and Tarka, T. (2021) U.S. Patent: US 2022/0017992 A1, <https://patents.google.com/patent/US20220017992A1/en>

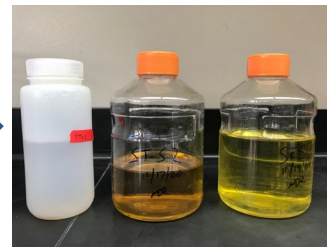
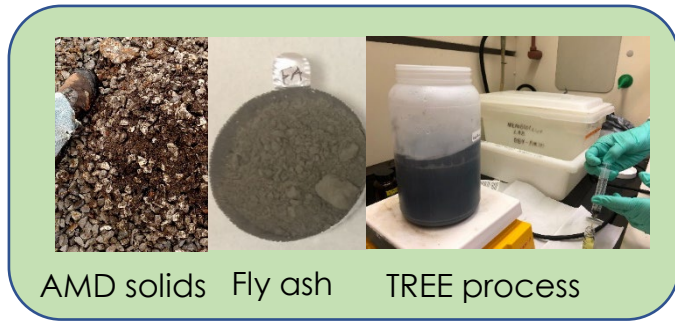


Achievement: TREE

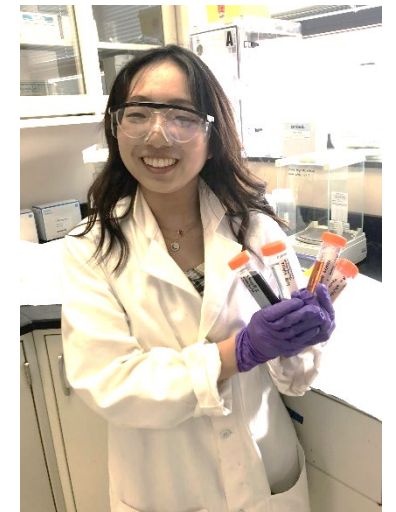
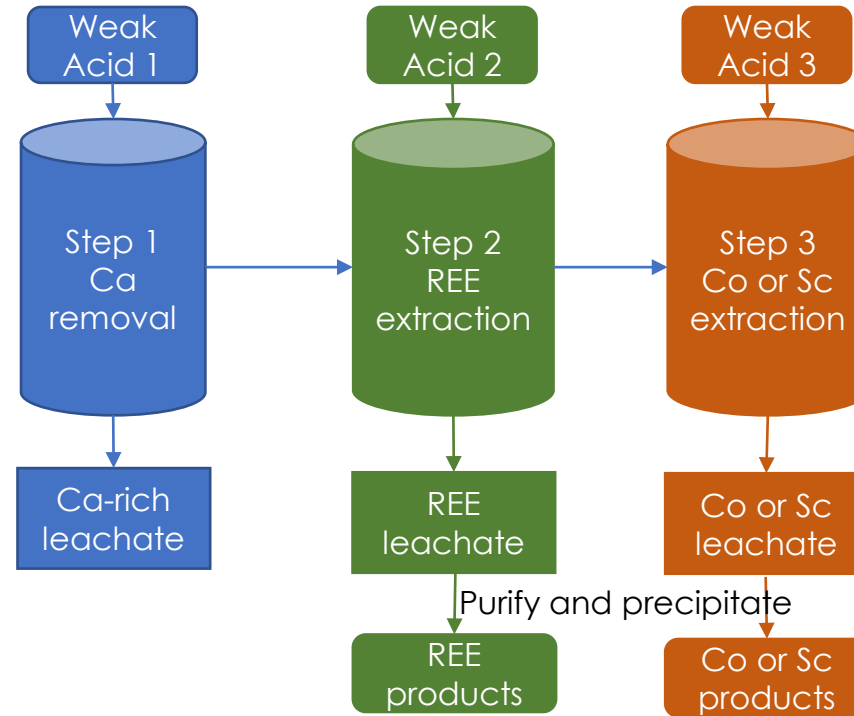
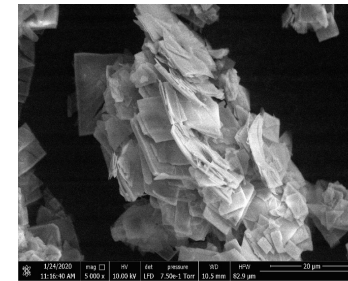
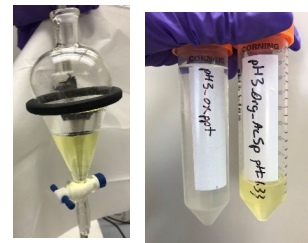
Reduce Extraction Steps and Conditions

TREE Advantages over REE mining:

1. Domestic/local waste feedstocks
2. No-pretreatment
3. Up to 90% acid reduction
4. No heating/no pressure
5. No solvent use or reduced solvent use
6. Less waste management cost
7. Additional value streams such as cobalt, nickel or scandium



3 step leachates



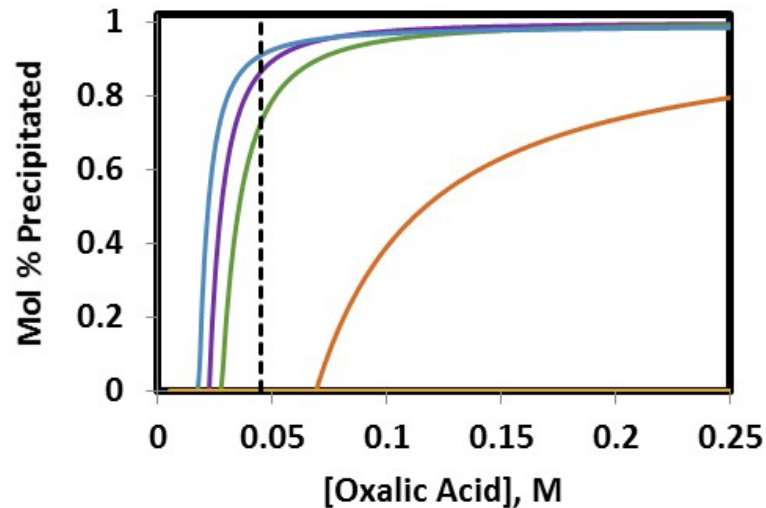
*Stuckman, M.Y., Lopano, C.L. and Tarka, T. (2021) U.S. Patent: US 2022/0017992 A1, <https://netl.doe.gov/node/10318>

Progress: Downstream Optimization-Bypass Solvent Extraction

Direct Oxalic Precipitation from Select TREE Leachate

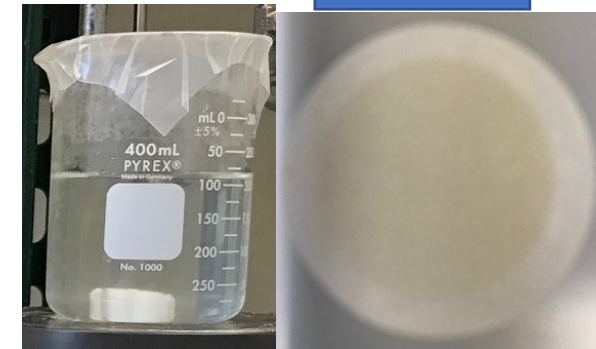
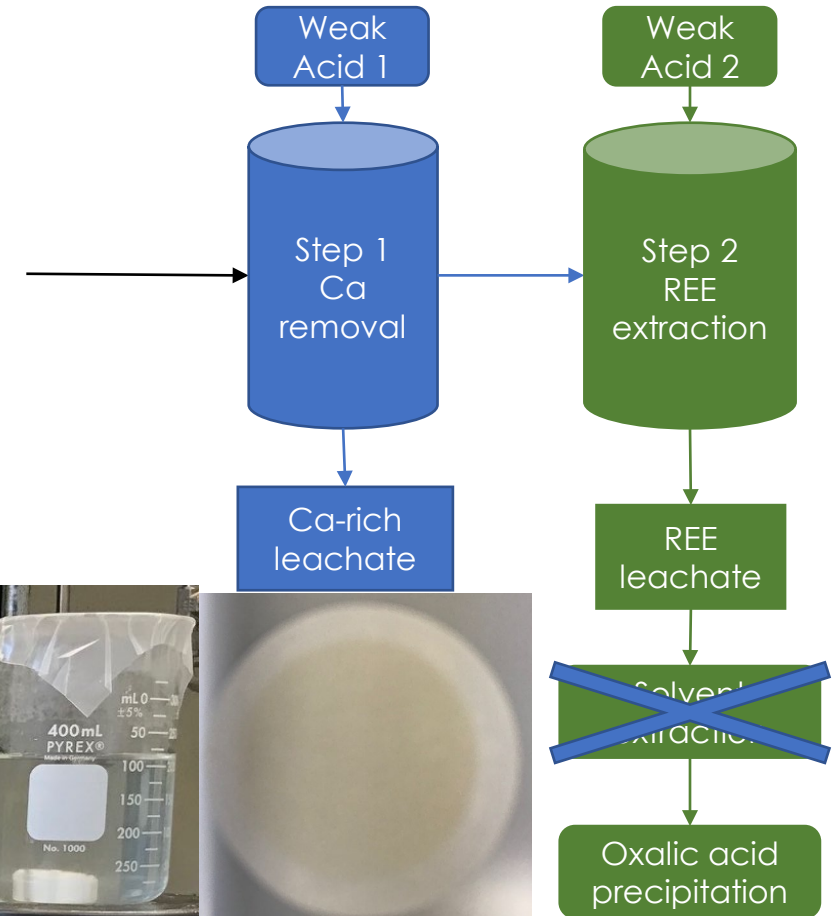
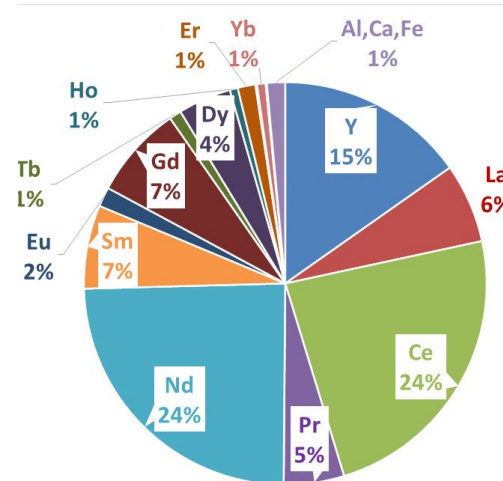
After TREE process, precipitation modeling guided oxalic acid precipitation

- Minimized oxalic acid use
- Successful rejects of Ca
- 99%wt REE oxalate recovered



— Nd
— Dy
— Y
— Ca
— Fe(II)
— Mn
--- [Ox]0

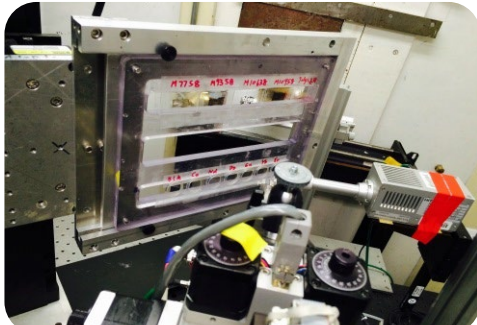
AMD solid



REE Oxalate, 99%wt

- **Outside Organizations/Funds Engaged in Research**
- Critical Minerals Task 4: Biological Recovery of Critical Materials from Acid Mine Treatment Solids
- Critical Mineral Task 16: Sorbent for Li Recovery Derived from Acid Mine Drainage (AMD) Treatment Solids
- Hedin Environmental: Previous collaboration with Hedin Environmental will be leveraged to access AMD treatment solids to characterize CM recovery potential and develop deployable technology.
- University of Pittsburgh: Collaboration with the isotope geochemistry group at Pitt and their students to better understand AMD geochemistry.
- Synchrotron sources: University of Chicago & Argonne National Lab; Stanford University and Stanford Linear Accelerator Lab.
- **Outreach**: Pennsylvania Independent Oil and Gas Association (PIOGA) Tech seminar (2021 and 2022), Stony Brook University, University of Pittsburgh, Leidos headquarter, and Brazilian Coal Association
- **Workforce Development**: 3 ORISE and Mickey Leland summer interns, Ph.D. students or graduates from University Pittsburgh and Duke University

• Characterization-Informed Recovery



EY22 Characterization of CM Binding (On track)

- REE complete
- Co, Ni, Zn in progress



EY23 Optimized Targeted Extraction (In progress)

- Demonstrate sequential leach of REE and Co/Ni from multiple AMD solids
- Demonstrate 80-100 % CM recovery via different acids, and/or reducing agents



EY24 Downstream optimization and operation optimization (In progress)

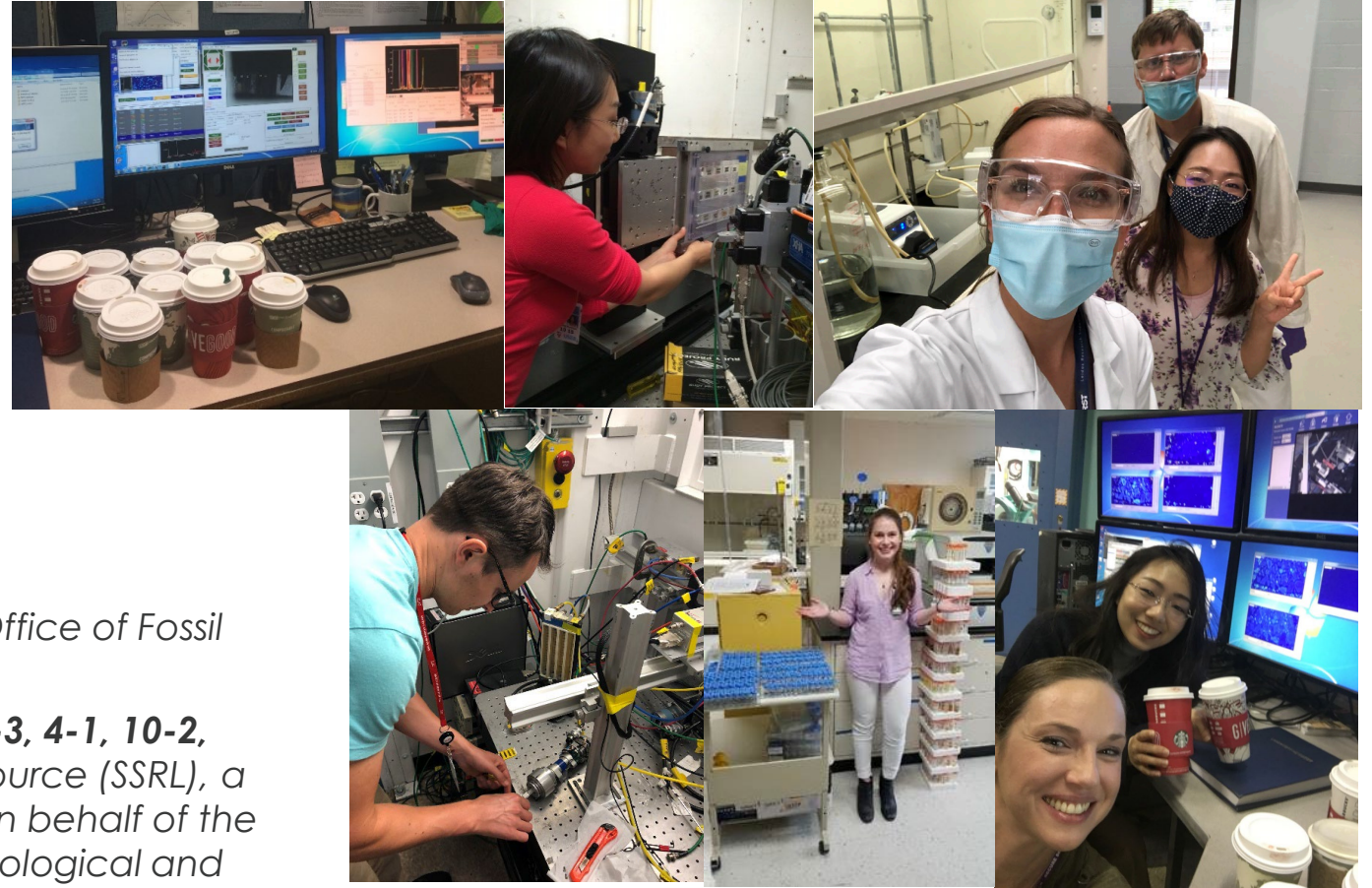
- Direct oxalic acid precipitation modeling established
- Optimized of the extraction, such as liquid: solid ratio, kinetics and temperature

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