

# Scale up Production of Graphite from Carbon Ore and Coal Refuse — FEAA157

## 2022 Resource Sustainability Annual Project Review Meeting

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UK-CAER — Rodney Andrews, Matthew Weisenberger, John Craddock, Vivian Edwards, Robert Pace, David Eaton, Asmund Vego, Kirk Norasak, George Frank, Christina Thompson, Michela Martinel

Penn State — Jonathan Mathews

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# Outline

- Project Objectives
- Background
- Approach
- Results
- Summary

# Project Overview

- **Title:** Scale up Production of Graphite from Carbon Ore and Coal Refuse
- **Objective:** To develop and demonstrate processes for scaling up the production of graphite from carbon ore, coal refuse and waste streams associated with previous coal mining activities
- **Period of Performance:** December 2021 – November 2024
- **Project Managers:** Michael Fasouletos (NETL), Joe Stoffa (NETL), Ana Wendt (DOE-HQ)
- **Project Team:** ORNL and University of Kentucky, Pennsylvania State University.

# Outline

- Project Overview
- Background



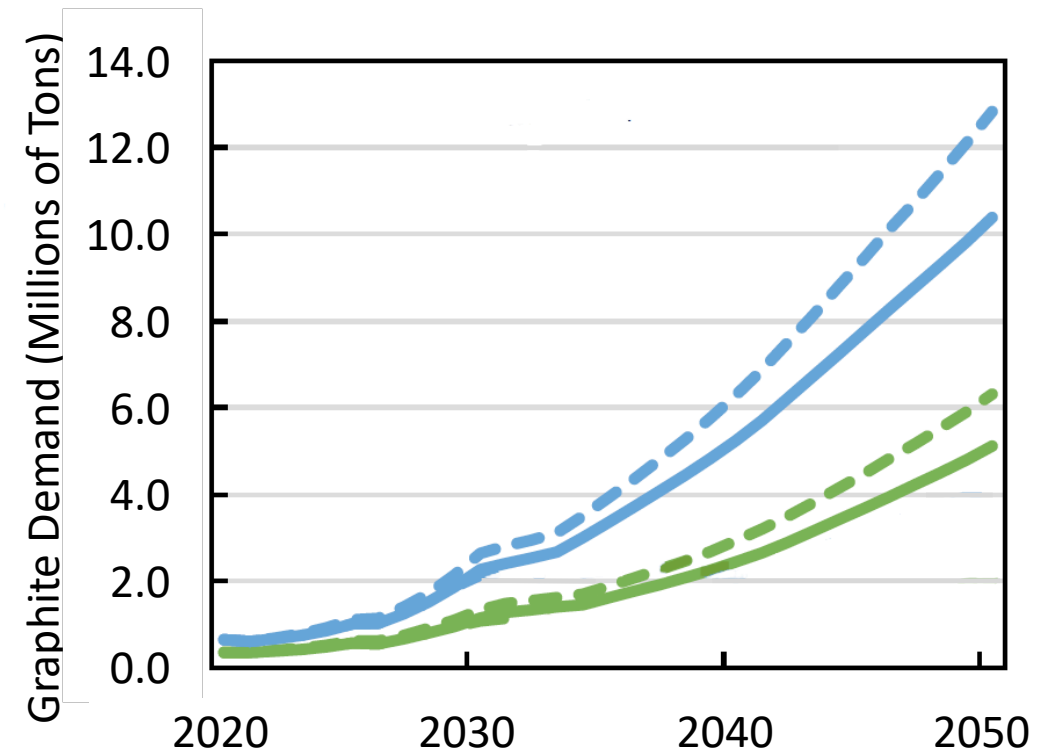
# Background: Demand for Graphite for Li-ion Batteries

The demand for graphite for anodes used in lithium-ion batteries needed for consumer electronics and electric vehicles is expected to grow significantly. Currently there are 88 battery megafactories in the world requiring 165,000 tons of graphite per year. According to various scenarios, more than 2.0 million tons-per-year will be needed by 2030 and more than 10 million tons-per-year by 2050.

International Energy Agency (IEA) scenarios:

- Stated Policies scenario, which incorporates existing government policies.
- Sustainable Development scenario, which is compatible with the climate goals of the Paris agreement and includes the target of reaching a 30% global sales share for EVs by 2030.

———— Nickel-Cobalt-(Manganese , Aluminum) cathodes  
- - - - - LiFePO<sub>4</sub> cathodes



Xu, C., Dai, Q., Gaines, L., Hu, M., Tukker, A. and Steubing, B., 2020. Future material demand for automotive lithium-based batteries. *Communications Materials*, 1(1), pp.1-10.

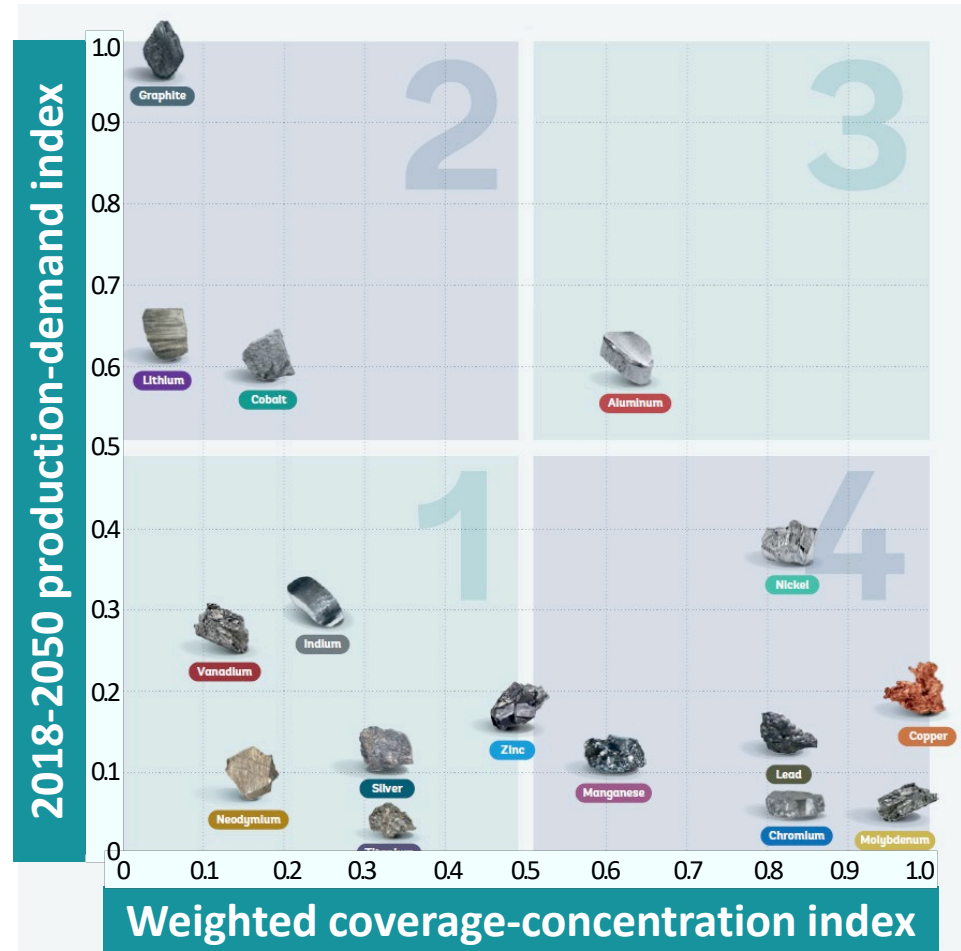
# Background: Demand for Graphite for Li-ion Batteries (cont.)

## 2018-2050 production-demand index:

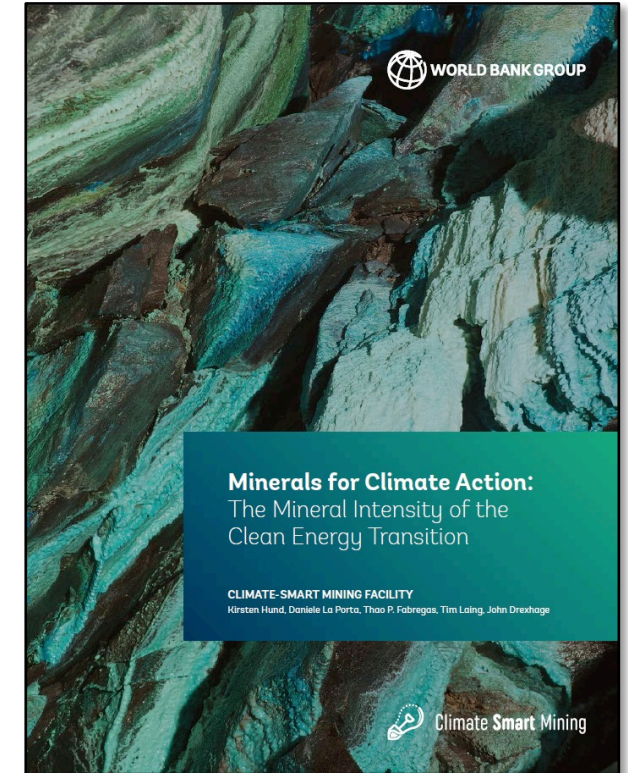
This index captures the scale to which production must scale up to meet demand from energy technologies. The index consists of two parts:

- Relative demand is captured by comparing 2050 demand from energy technologies to 2018 total production of the mineral. An index between 0 and 1 is then computed, with the mineral with highest relative demand, graphite, given a score of 1 and each other mineral given a score relative to graphite.
- Absolute demand is captured through the absolute level of demand in 2050 from energy technologies for each mineral. The mineral with the highest level (aluminum) is given a score of 1 and every other mineral is given a score between 0 and 1 relative to aluminum.

The two parts of the index are given an equal weight to compute an overall production-demand index.



**Weighted coverage-concentration index.** This index captures how cross-cutting or concentrated in a few technologies the minerals are. A value for 1 is given for the most cross-cutting mineral, namely copper, with the scores for all other minerals relative to copper. The index is calculated on an equal weighting of : (1) the number of technologies that require one mineral, and (2) the share of demand for minerals that comes from a single technology.



Under a 2-degree scenario (2DS), production of graphite, lithium, and cobalt will need to be significantly ramped up by more than 450% by 2050—from 2018 levels—to meet demand from energy storage technologies.

# Background: Demand for Graphite for Li-ion Batteries (cont.)

## List of 35 Critical Mineral Commodities

### Earth MRI Phases 1 and 2 Critical Minerals (in red)

- |                      |                              |
|----------------------|------------------------------|
| • Aluminum (bauxite) | • Manganese                  |
| • Antimony           | • Niobium                    |
| • Arsenic            | • Platinum Group Metals      |
| • Barite             | • Potash                     |
| • Beryllium          | • Rare Earth Elements (REEs) |
| • Bismuth            | • Rhenium                    |
| • Cesium & Rubidium  | • Scandium                   |
| • Chromium           | • Strontium                  |
| • Cobalt             | • Tantalum                   |
| • Fluorspar          | • Tellurium                  |
| • Gallium            | • Tin                        |
| • Germanium          | • Titanium                   |
| • Graphite (natural) | • Tungsten                   |
| • Helium             | • Uranium                    |
| • Indium             | • Vanadium                   |
| • Lithium            | • Zirconium & Hafnium        |
| • Magnesium          |                              |

### Why These?

- US has high net import reliance.
- Usage is increasing beyond foreseeable domestic production.
- Focus first on those commodities that, if discovered, may reduce the Nation's net import reliance.
- Lower priority given to those commodities for which improvements in recovery and marketing of current supplies can satisfy domestic markets.

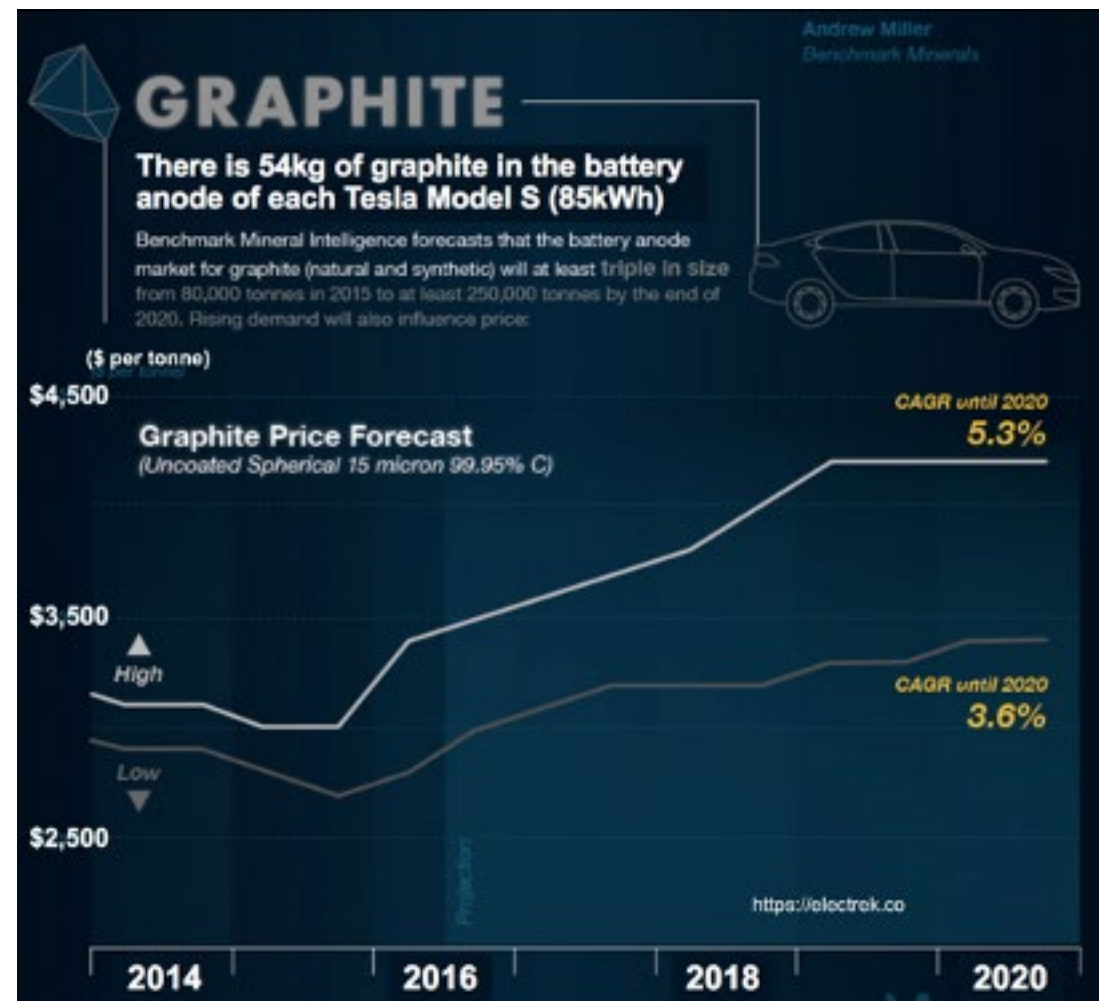


Briefing Materials Only

30

## The USGS has listed graphite as a critical mineral

China is by far the largest producer of natural graphite, accounting for roughly two-thirds of world production. Only 4 percent of the world's natural graphite comes from North America, with no U.S. production in decades. Although natural graphite was not produced in the United States in 2016, about 98 U.S. firms, primarily in the Northeastern and Great Lakes regions, consumed graphite in various forms from imported sources for use in brake linings, foundry operations, lubricants, refractory applications, and steelmaking. Graphite's use in rechargeable batteries, as well as technologies under development (such as large-scale fuel-cell applications), could consume as much graphite as all other uses combined.

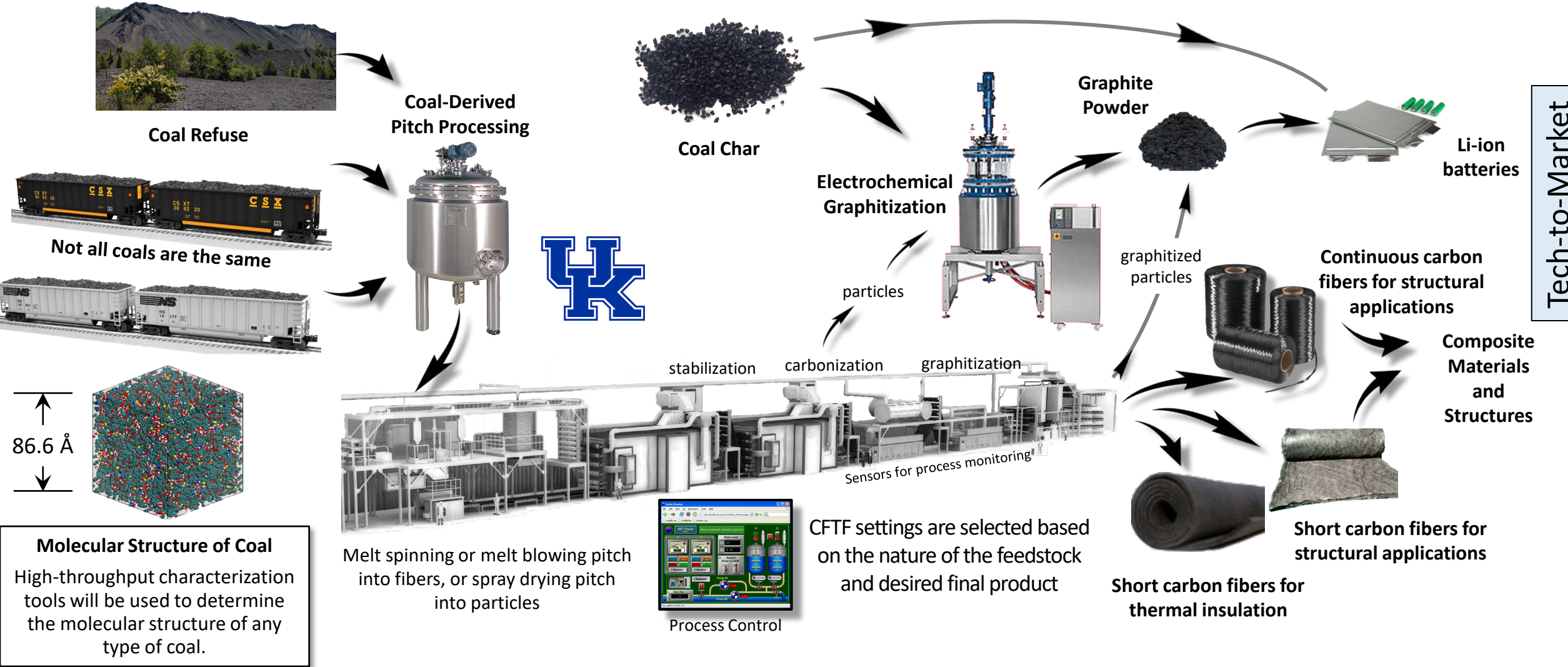


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- Approach



# Carbon Ore and Coal Refuse-to-Products at ORNL's Carbon Fiber Technology Facility

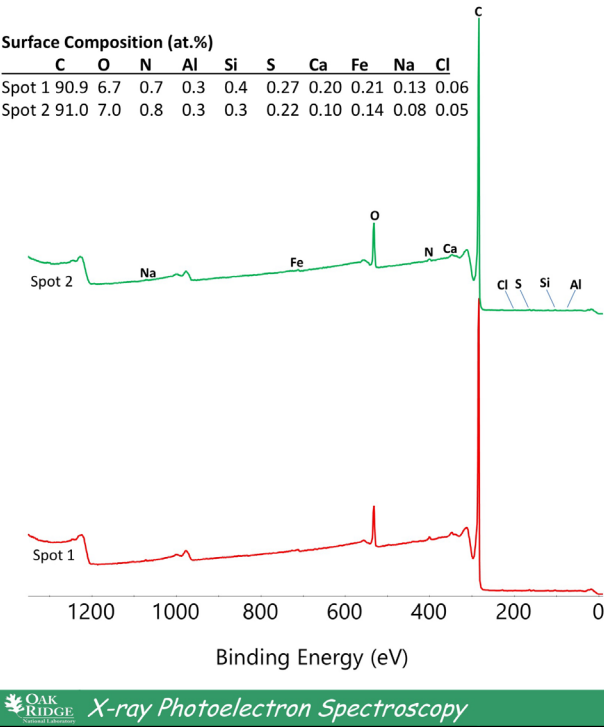
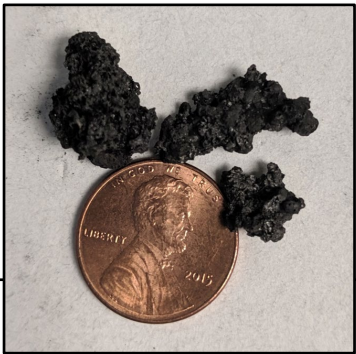
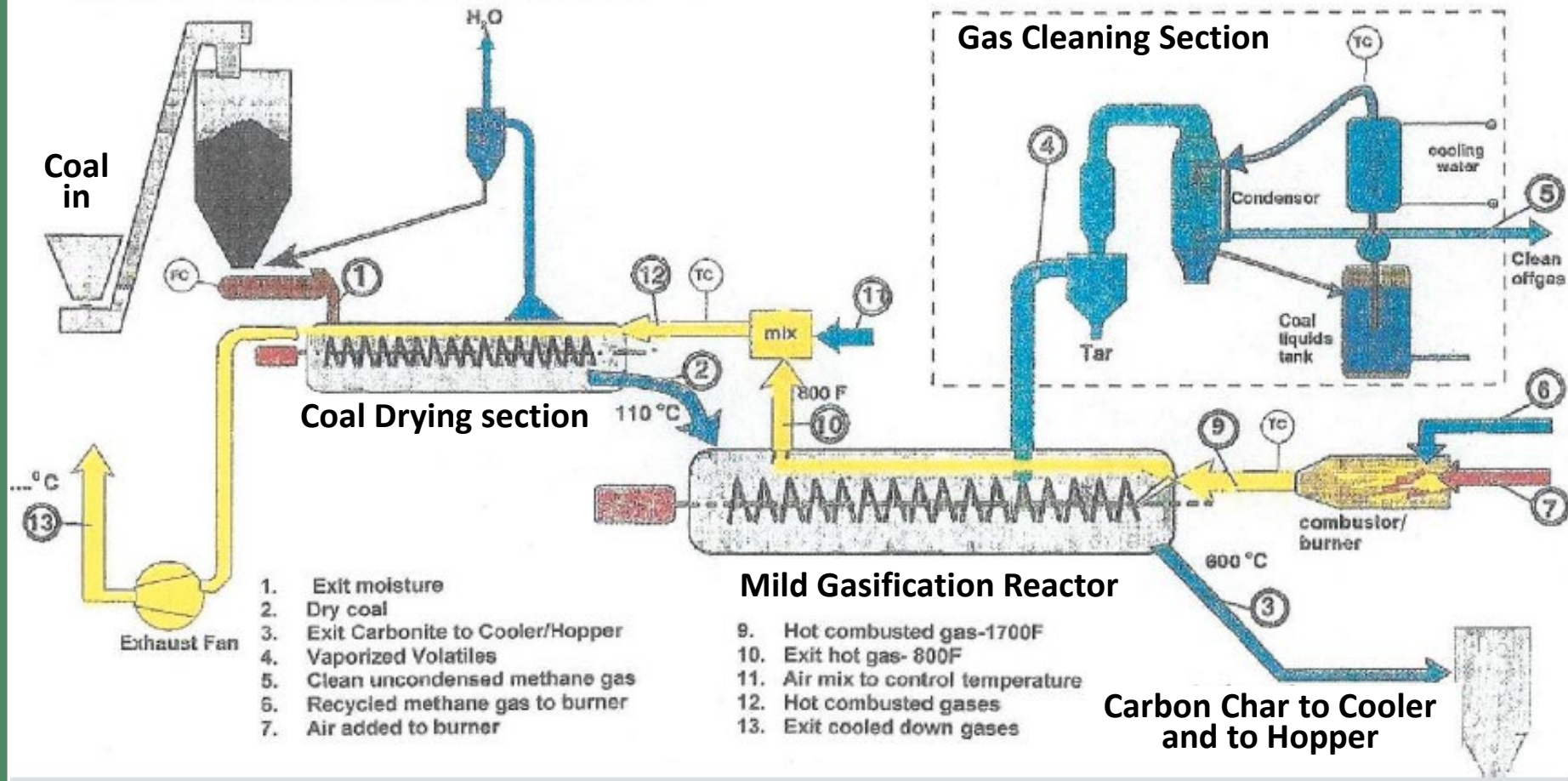


# Coal Char by Mild Gasification — Carbon Technology Co.

Coal →

Carbon Char

CTC'S Mild Gasification Process Converting Coal into Carbon Char, Coal Oil Liquids and Synthetic Non-Condensable Gas

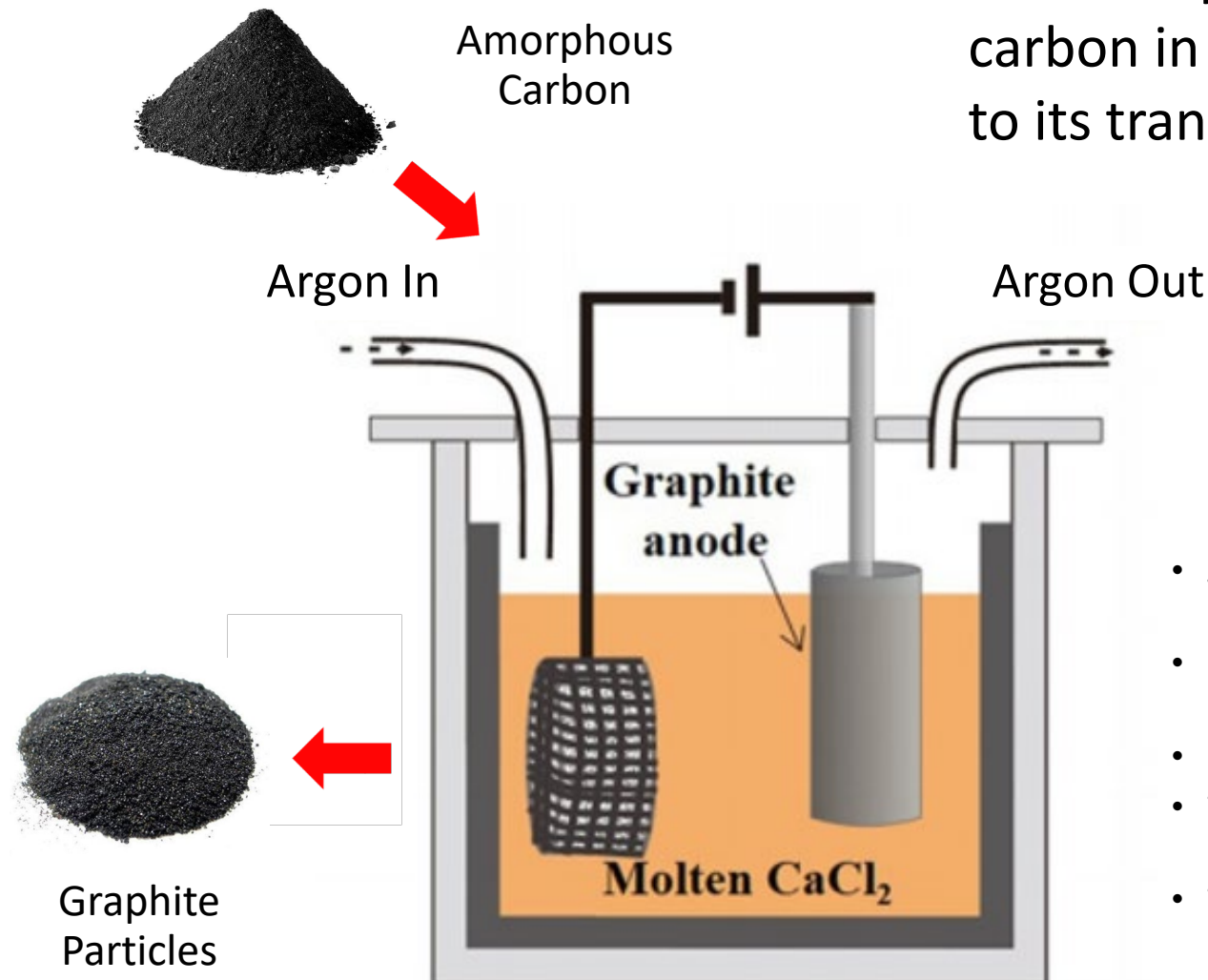


Carbon Technology Co. LLC  
Bristol, Virginia

Blue Gem Coal Char

# Electrochemical Graphitization

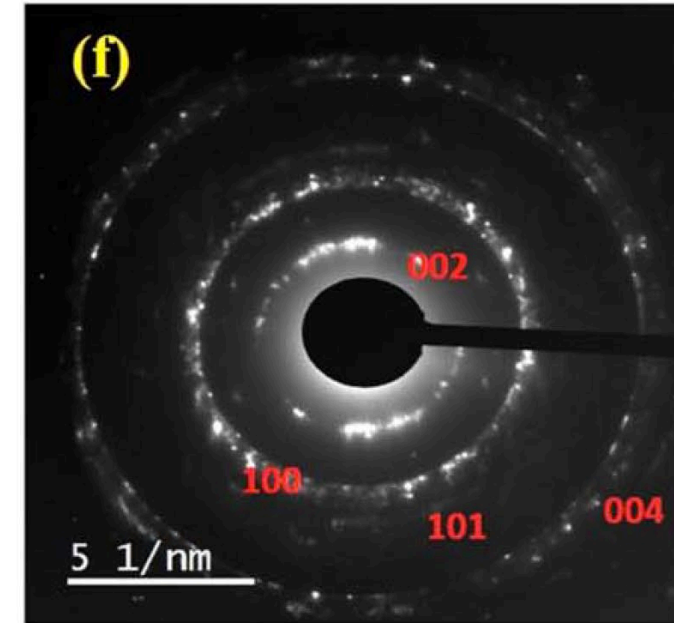
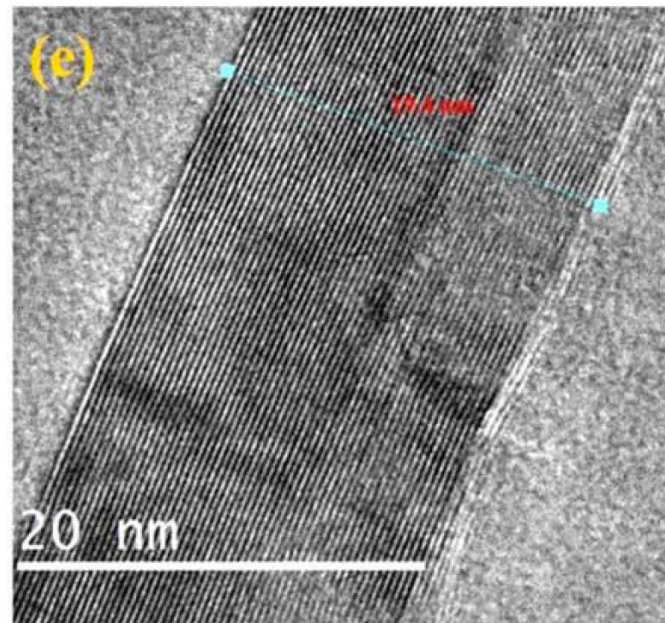
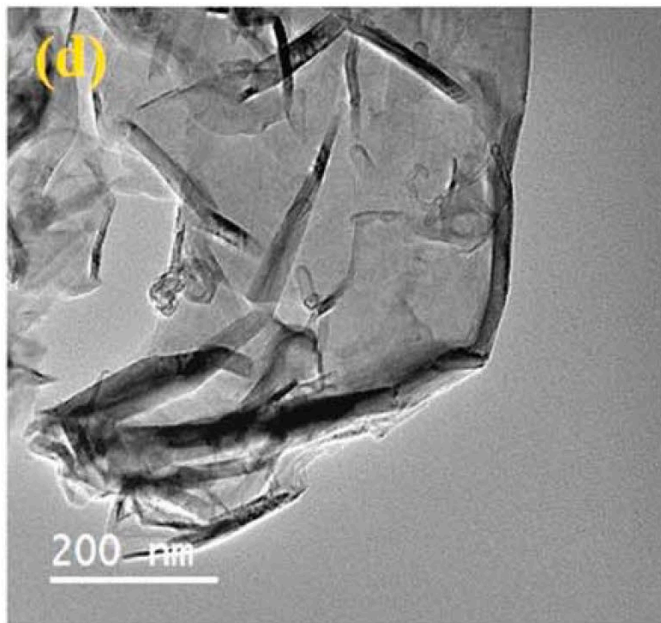
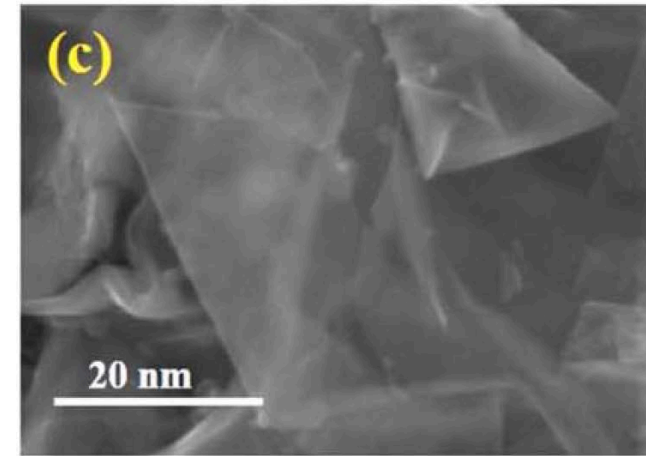
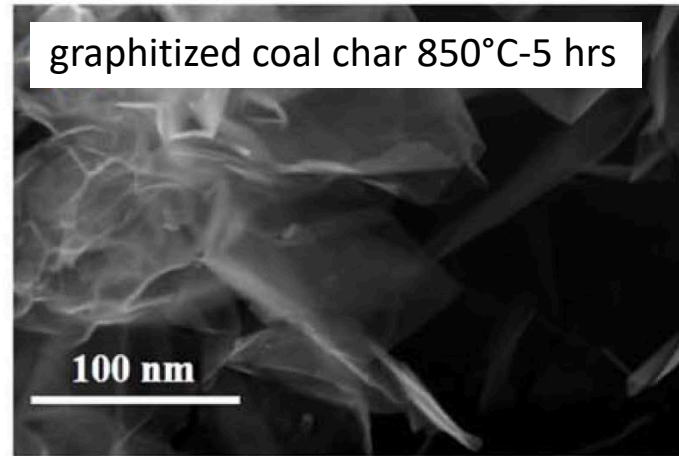
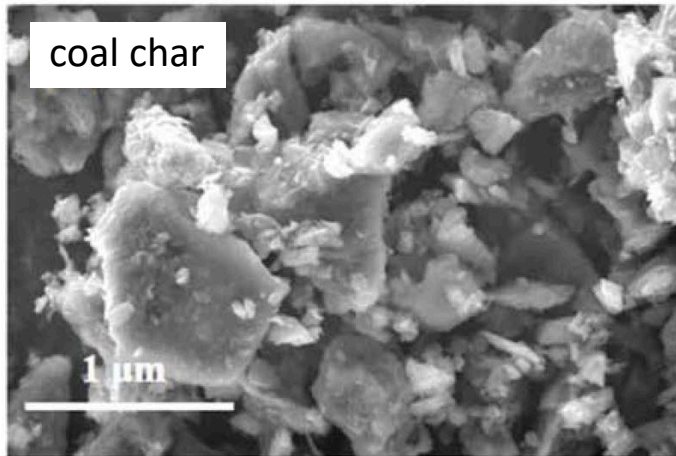
Cathodic polarization of amorphous carbon in molten  $\text{MgCl}_2$  at  $\sim 850^\circ\text{C}$  leads to its transformation into graphite.



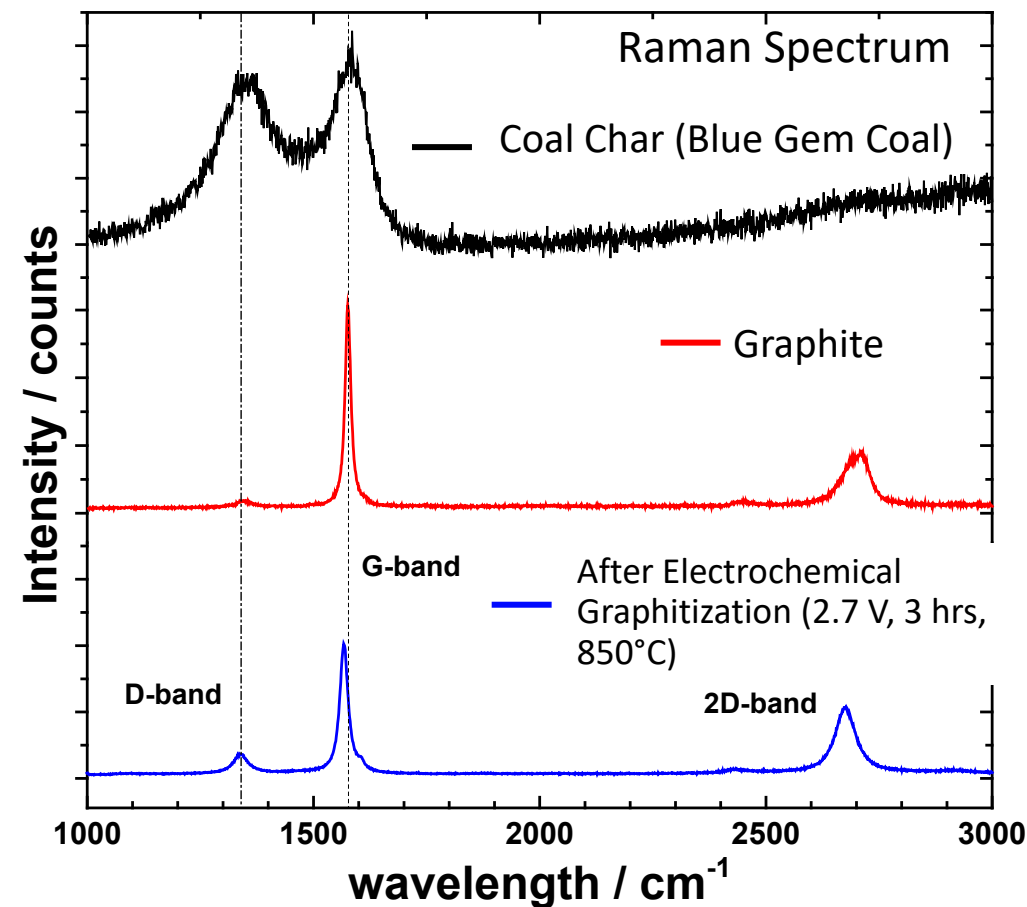
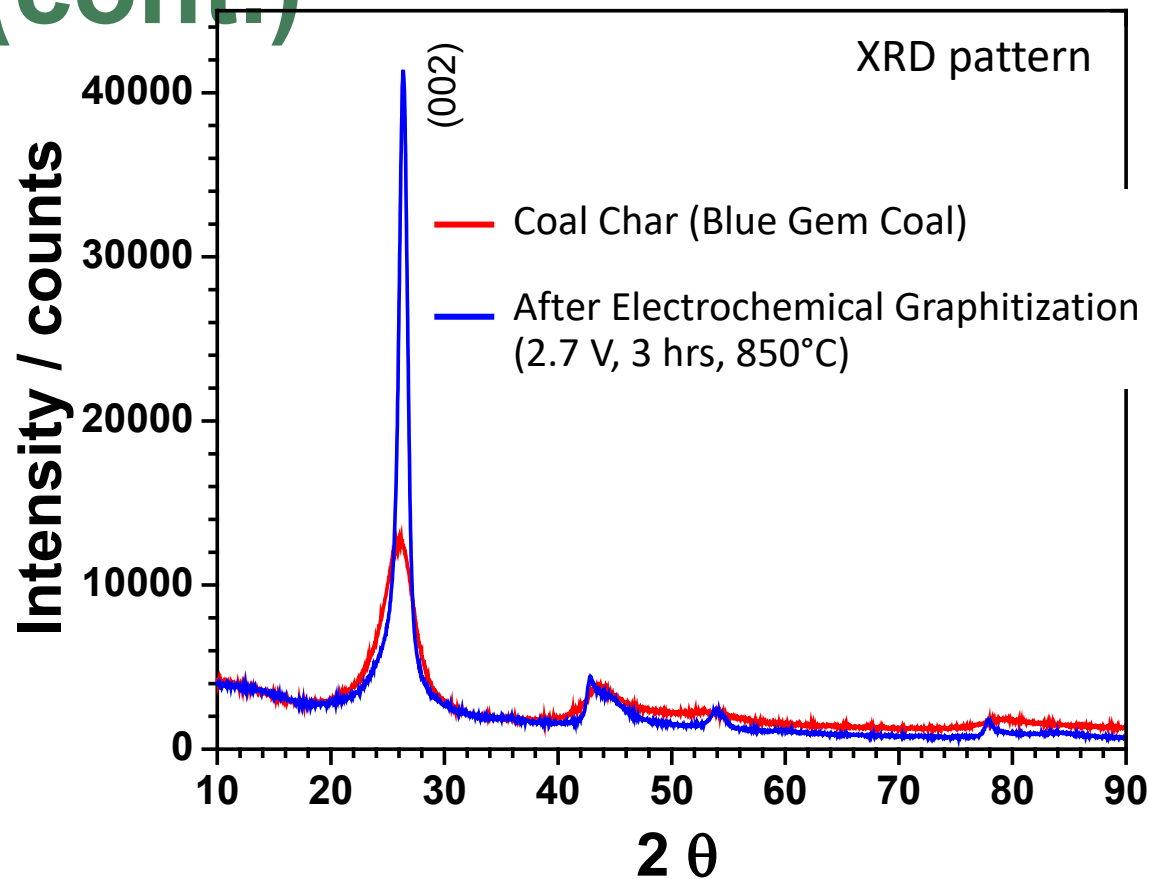
- Jin, X. B.; He, R.; Dai, S. *Chem.-Eur. J.* **2017**, 23, 11455-11459
- Peng et. al., *Angew. Chem. Int. Ed.* **2017**, 56, 1751-1755
- Bagri et. al., *Chem. Commun.* **2020**, 56, 2783-2786
- Thapaliya et. al., *J Electrochem. Soc.* **2021**, 168, 046504
- Thapaliya et. al., *ACS Appl. Mater. Interfaces* **2021**, 13, 4393-4401



# Electrochemical Graphitization of Coal Char

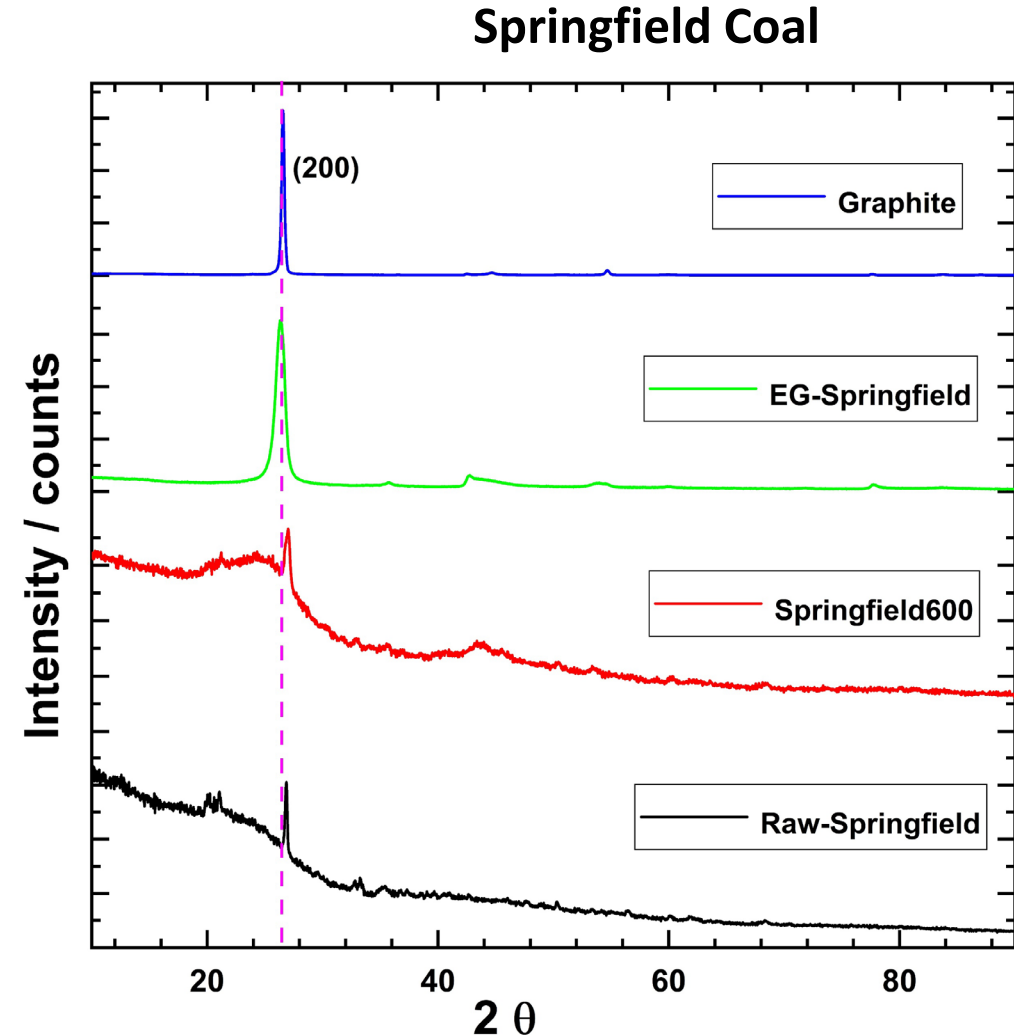
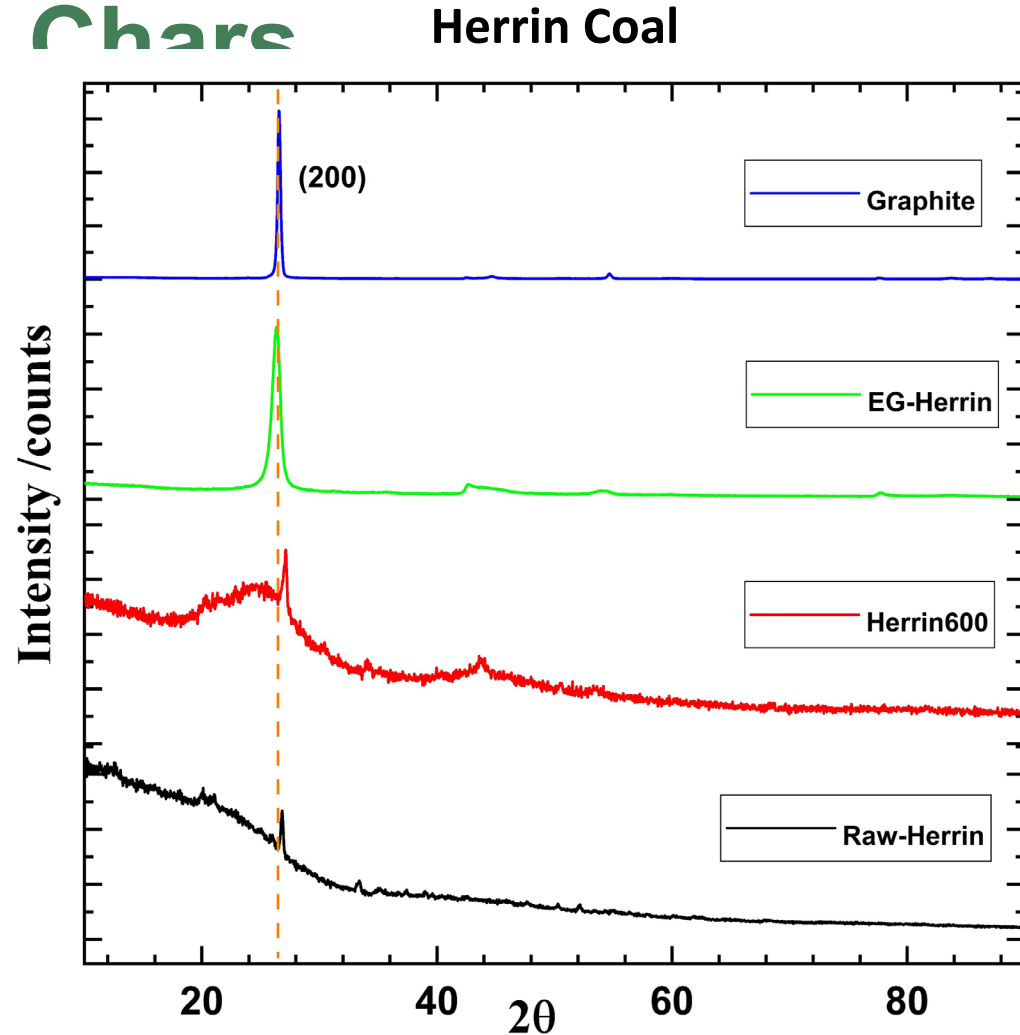


# Electrochemical Graphitization of Coal Char (cont.)



- XRD and Raman spectroscopy data confirmed successful graphitization.
  - Increased (002) peak intensity in XRD pattern
  - Low  $I_D/I_G$  ratio in Raman spectrum

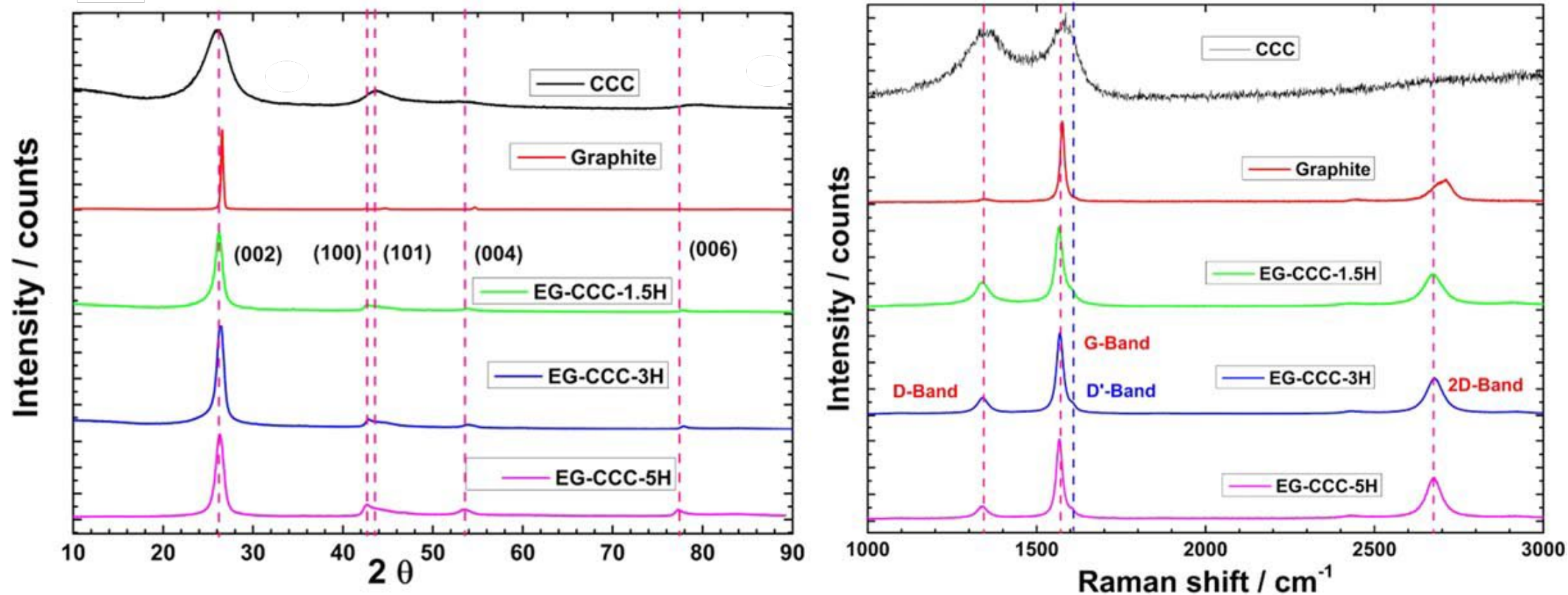
# Electrochemical Graphitization of Different Coal Chars



XRD patterns of raw-coal, of chars obtained by mild gasification (600°C for 6 hours), and electrochemically-graphitized chars at: -2.7 V, 850°C for 4 hours.

# Low-Temperature Electrochemical Graphitization

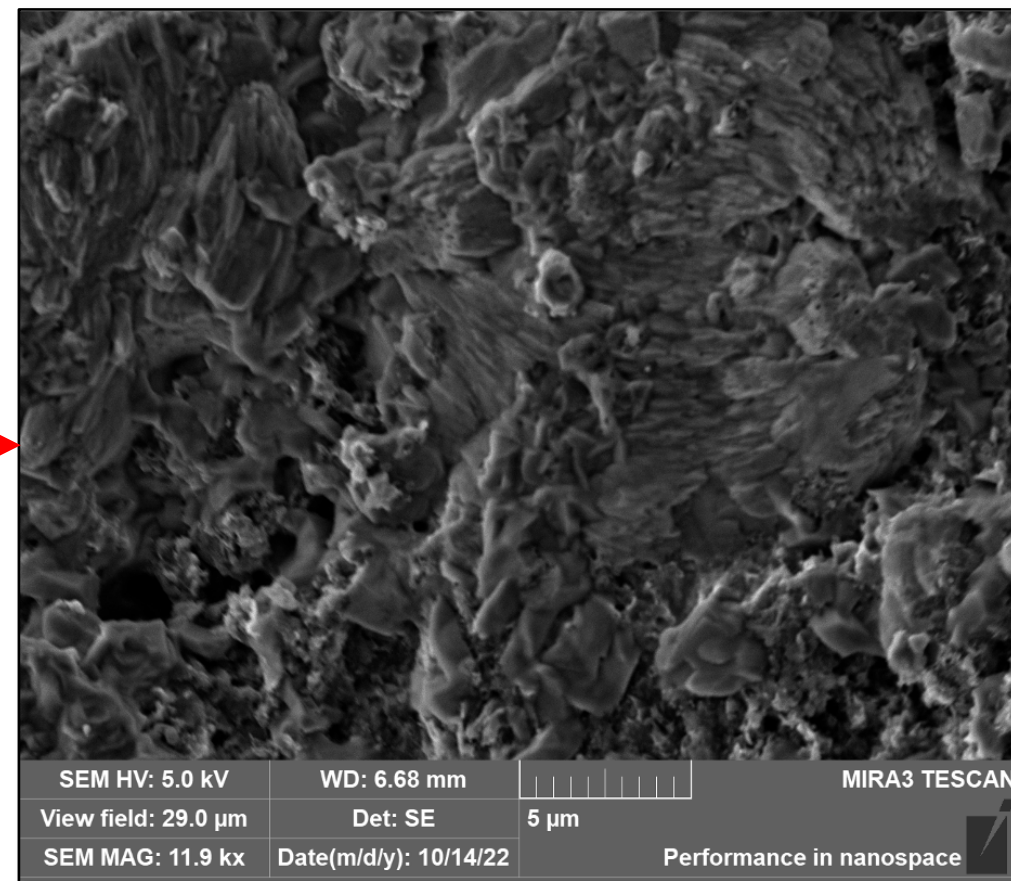
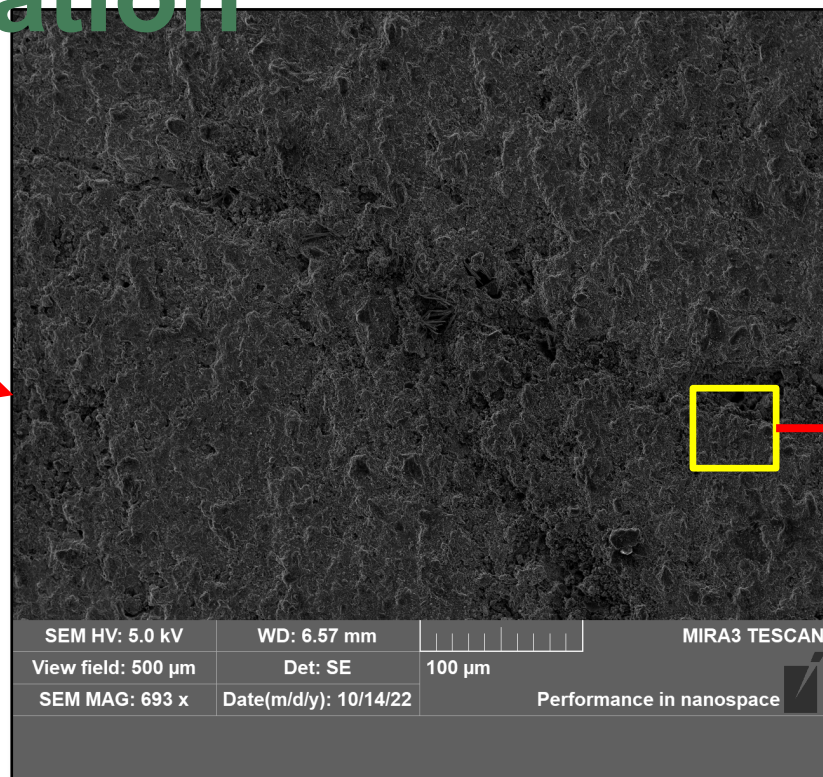
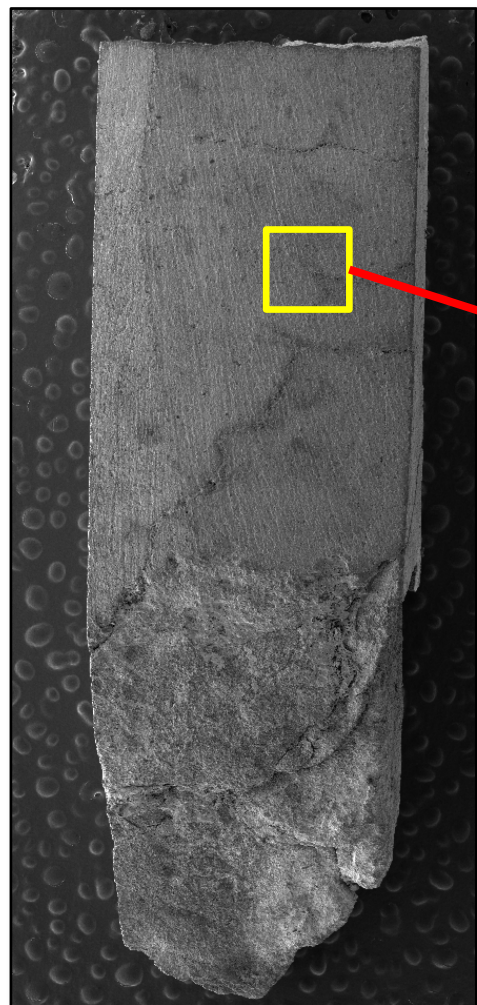
$T = 830^{\circ}\text{C}$ ;  $E = -2.7\text{ V}$



(left) XRD patterns (right) Raman spectra of coal char and electrochemically graphitized coal char (EG-CCC) as a function of time. Blue Gem Coal



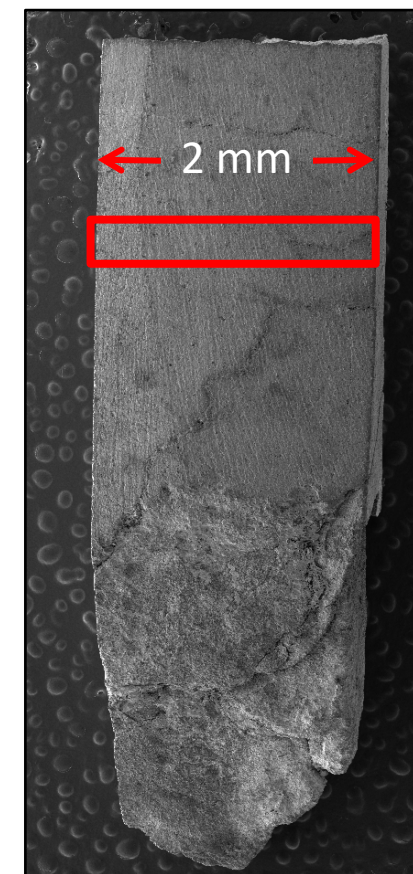
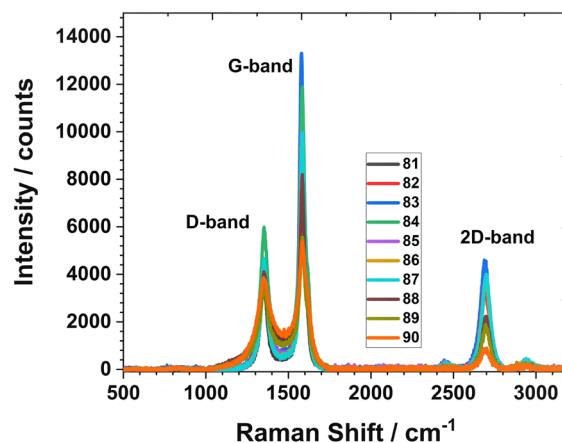
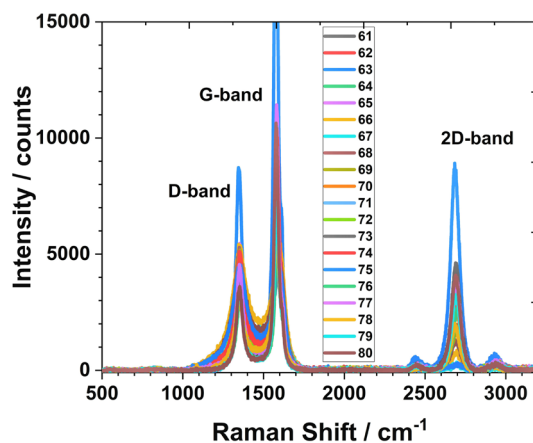
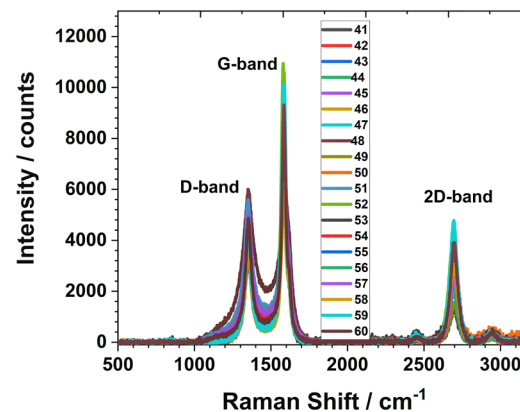
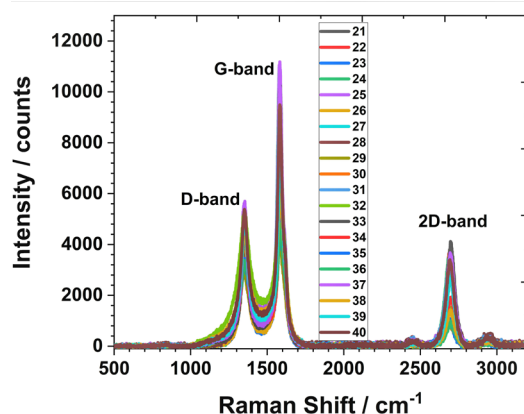
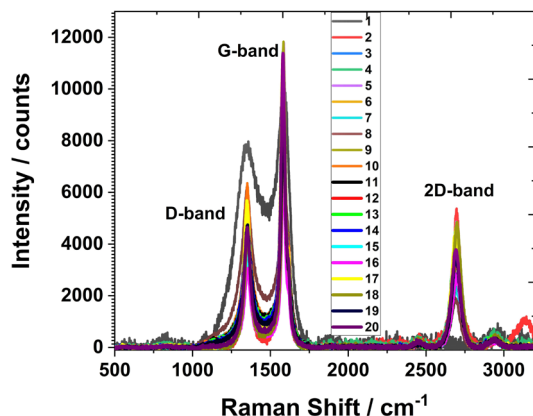
# Low-Temperature Electrochemical Graphitization



Assembly of cold-pressed coal char particles after electrochemical graphitization:  
-2.7 V, 3 hours, 850°C.

# Low-Temperature Electrochemical Graphitization

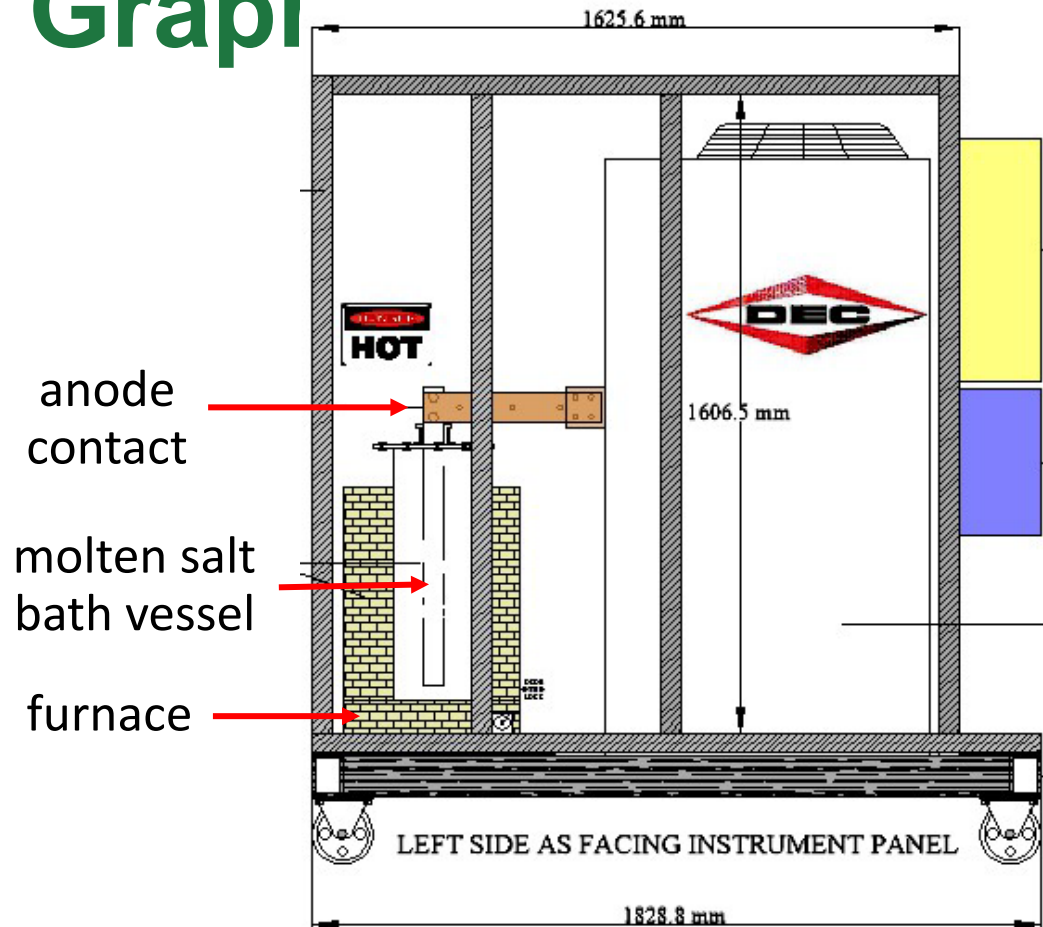
90 Raman spectra were obtained across the width of the sample (assembly of cold-pressed coal char particles)



Results indicate that graphitization occurred across the sample. Analysis is in progress to correlate effect of local microstructure on graphitization.



# Scaling up — Electrochemical Graphite



A system is being assembled to produce kilogram-quantities of graphite per batch.

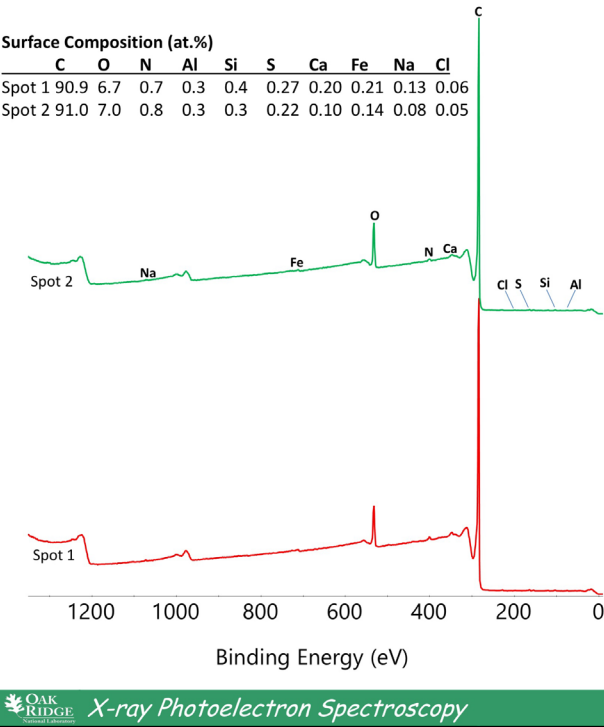
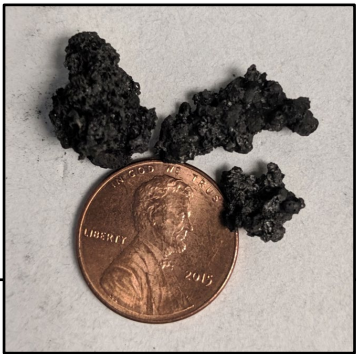
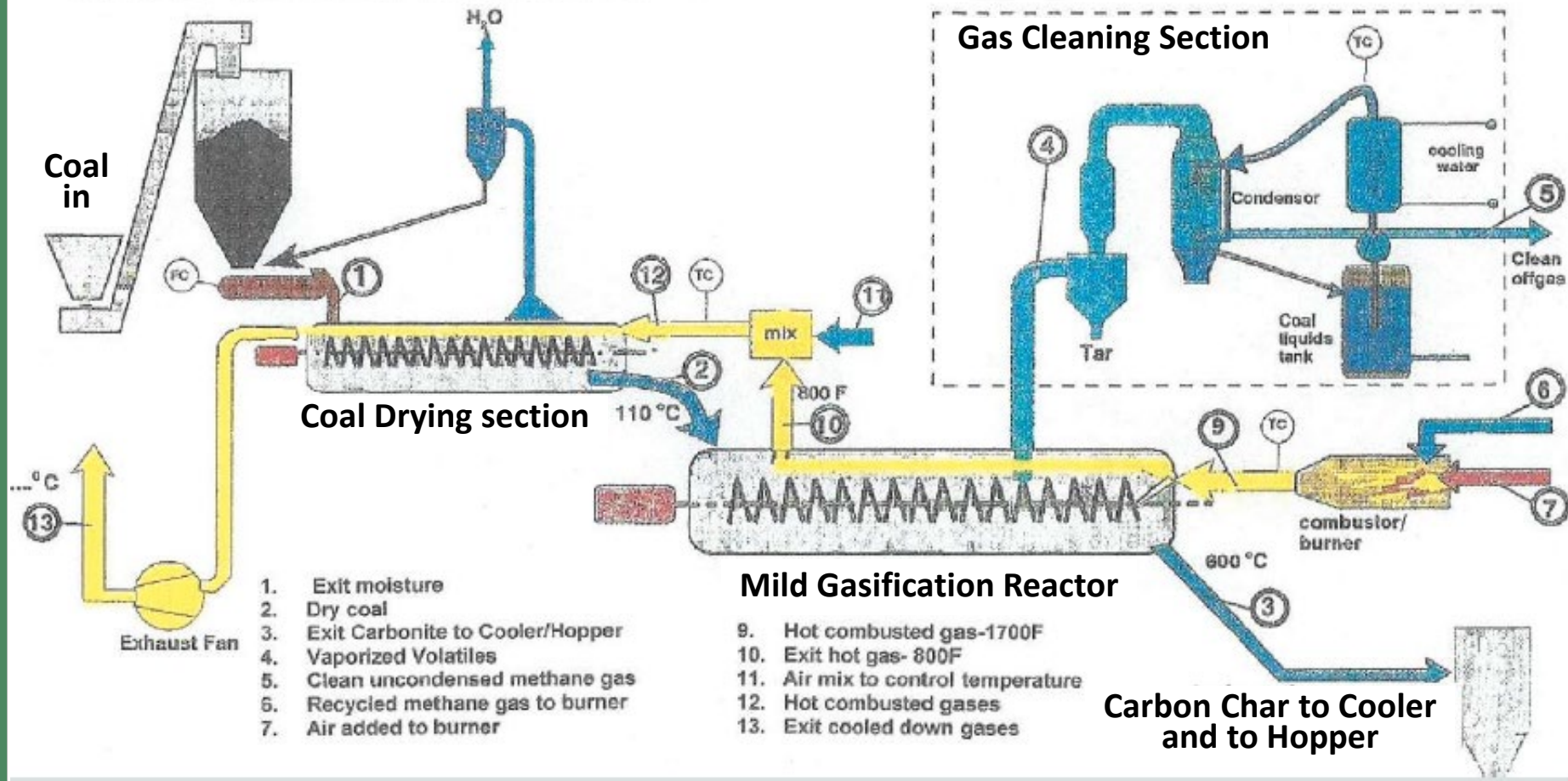


# Coal Char by Mild Gasification — Carbon Technology Co.

Coal →

Carbon Char

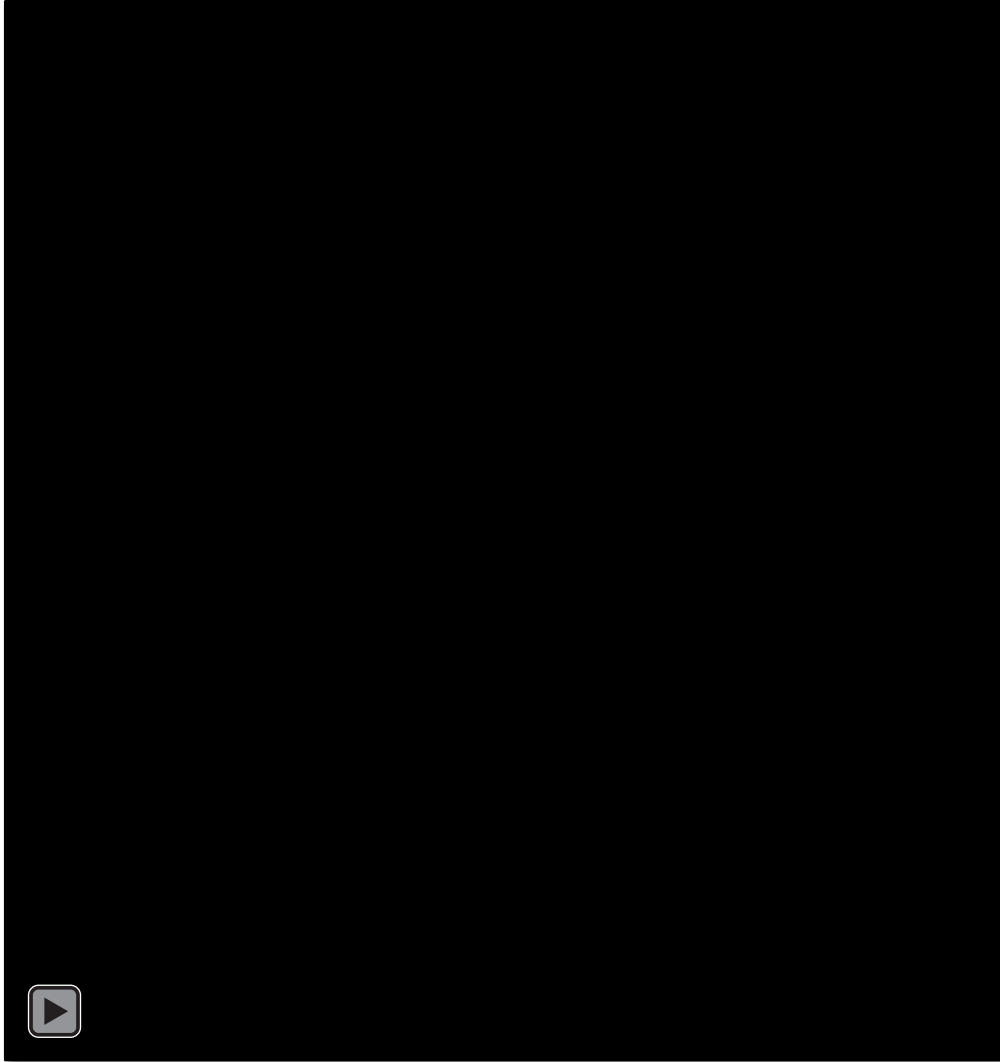
CTC'S Mild Gasification Process Converting Coal into Carbon Char, Coal Oil Liquids and Synthetic Non-Condensable Gas



Carbon Technology Co. LLC  
Bristol, Virginia

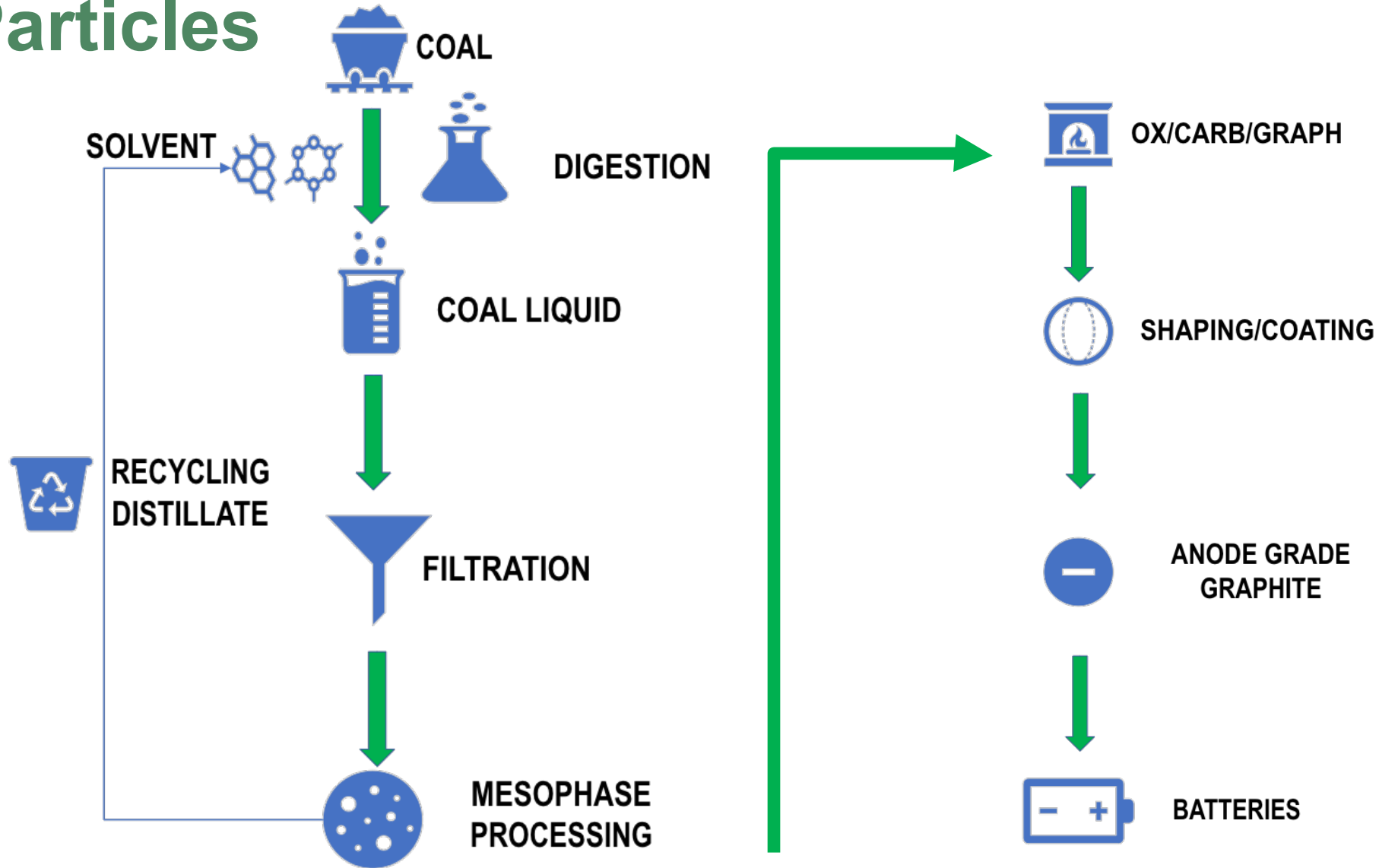
Blue Gem Coal Char

# Production of spherical carbon particles by Spray Drying



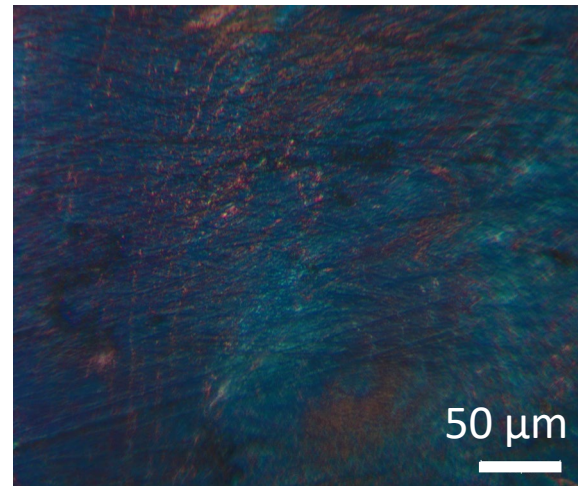
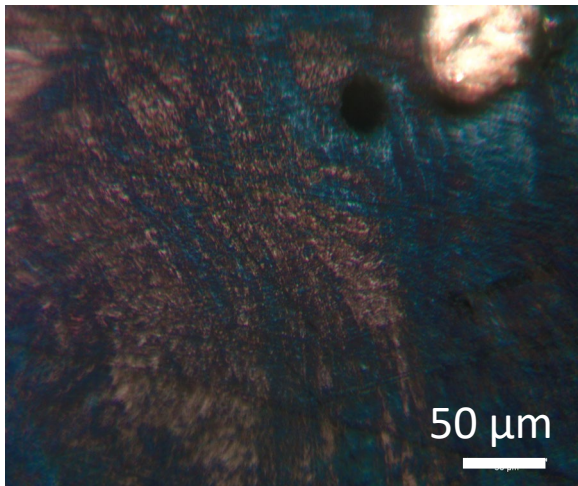
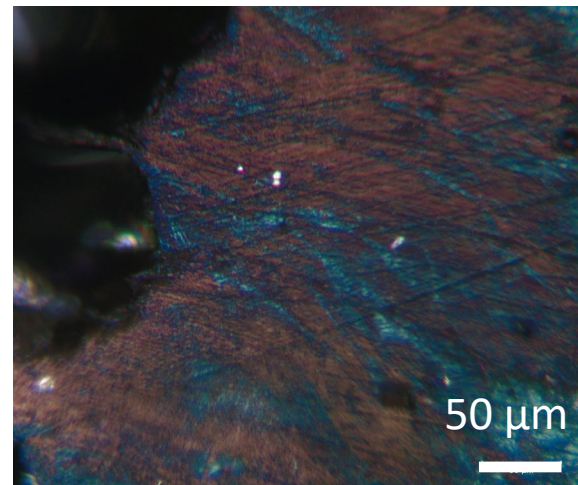
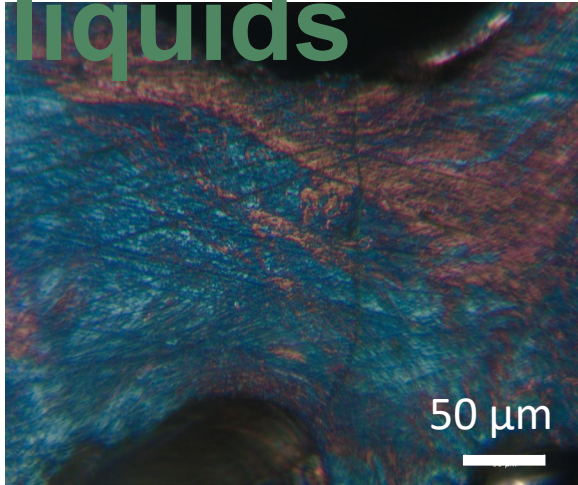
Solid carbon particles are used as feedstock for electrochemical graphitization. Process enables control of particle size distribution

# Alternative Route for Manufacturing of Graphite Particles





# Mesophase (Graphite Precursor) from coal liquids



Springfield Coal/Decant Oil  
410°C, ~240 min

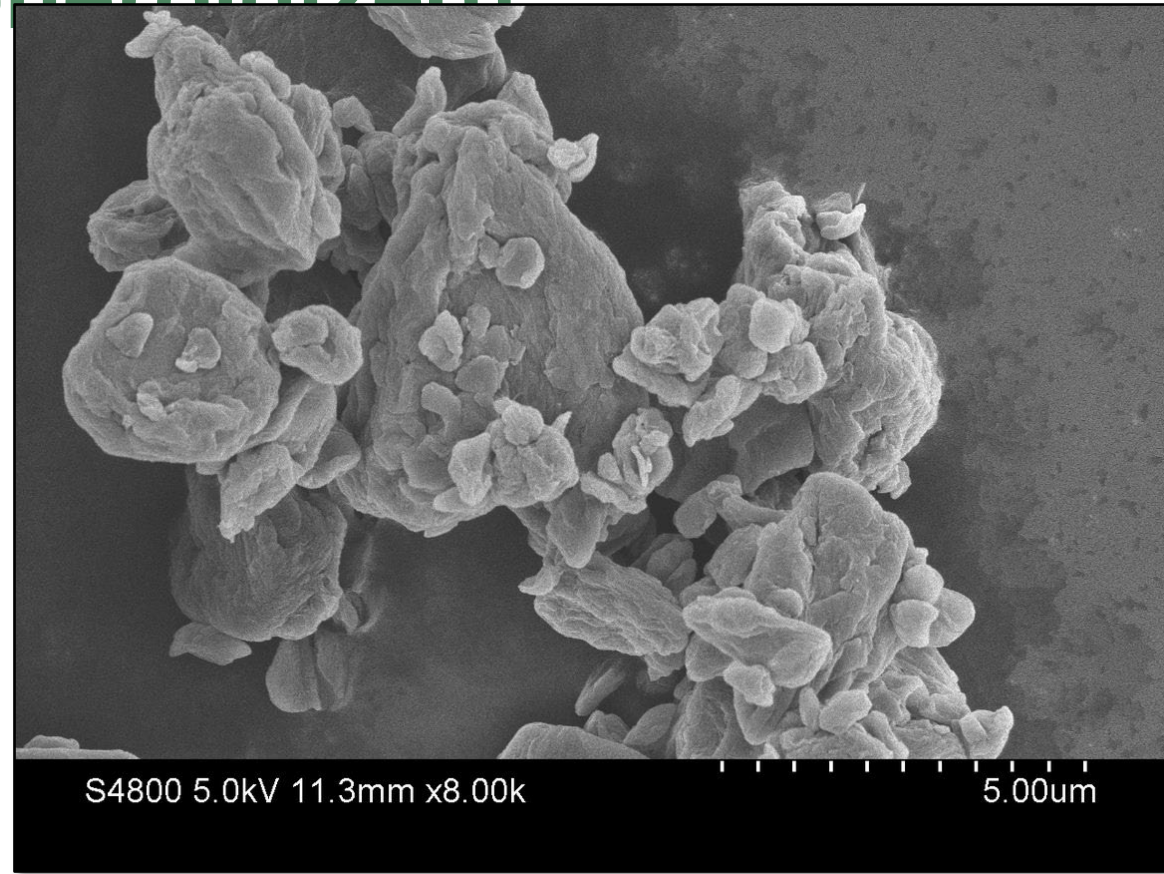
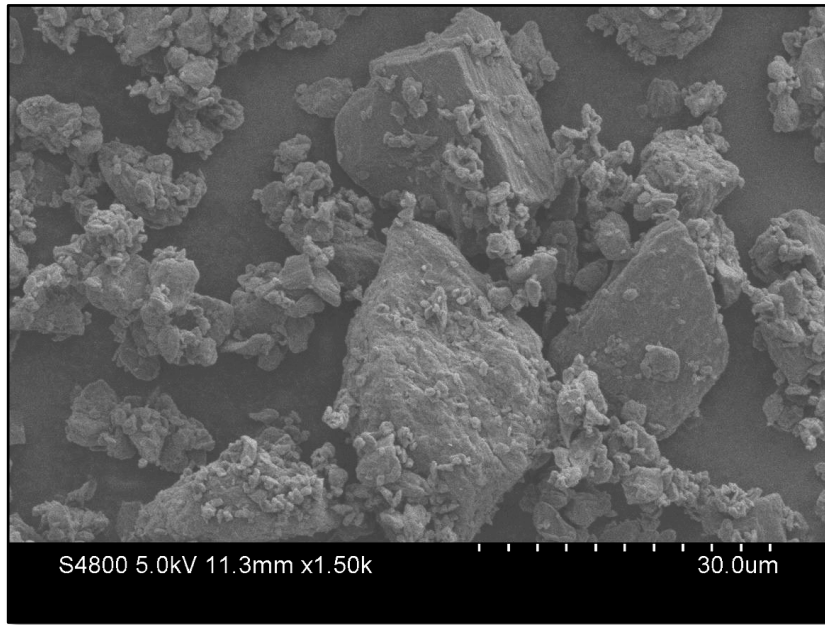
- $T_{sp}$ : 388 °C
- Mesophase count: 100%
- M173 was powdered via ball mill and sieved to < 200 mesh before oxidation.
- **Oxidation Profile:** 120 °C to 310 °C
- **Carbonization/Graphitization Profile:** 900 °C then 2400 °C

Step	Yield
Oxidation	108.7%
Carb/Graph	51.1%
Final	55.5%



SFDO2(13)-M173

# SEM images of Graphitized Mesophase from Coal Liquid (not spheroidized)



2400°C

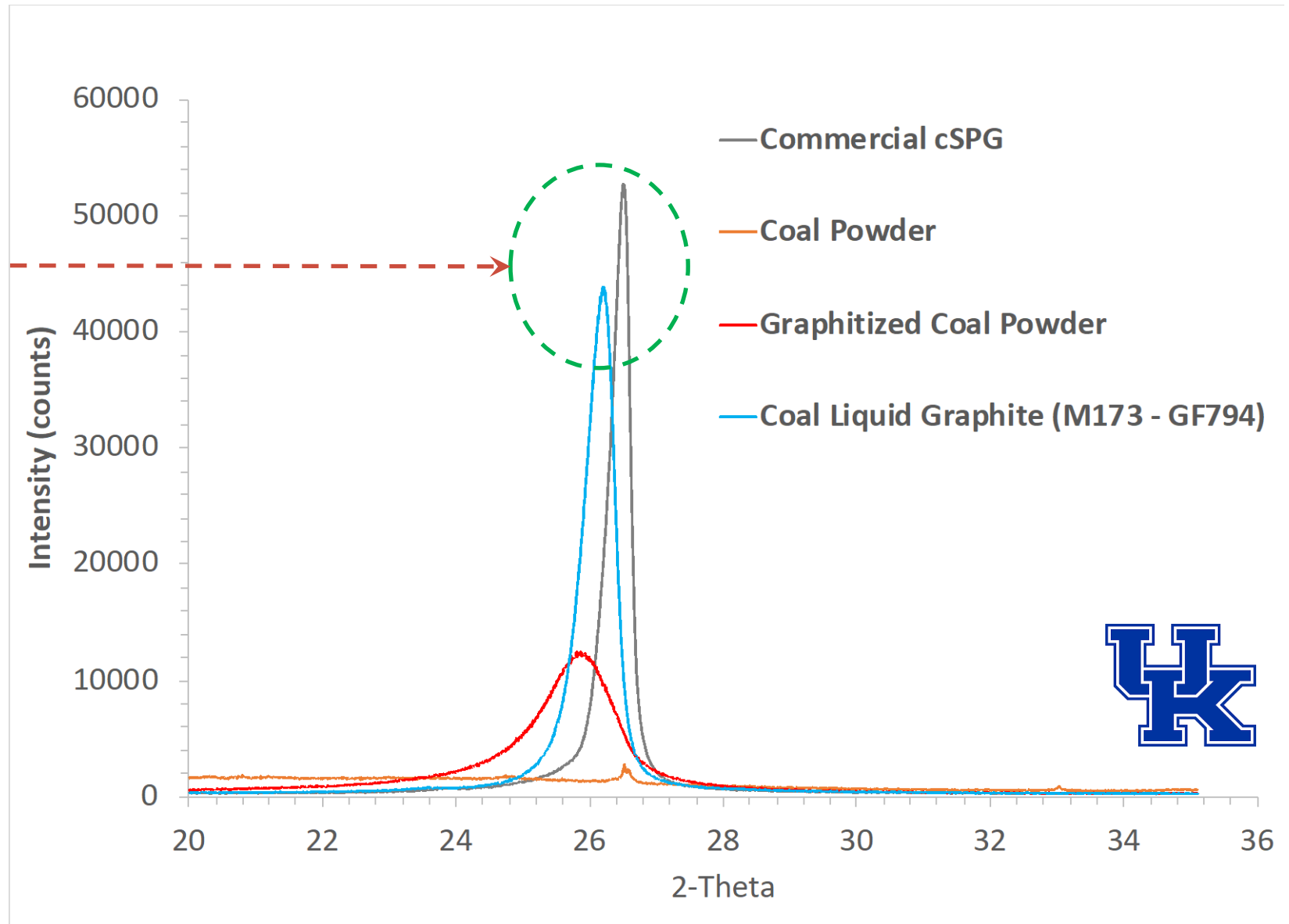
M173



# Graphitized Mesophase from Coal Liquid — XRD Results

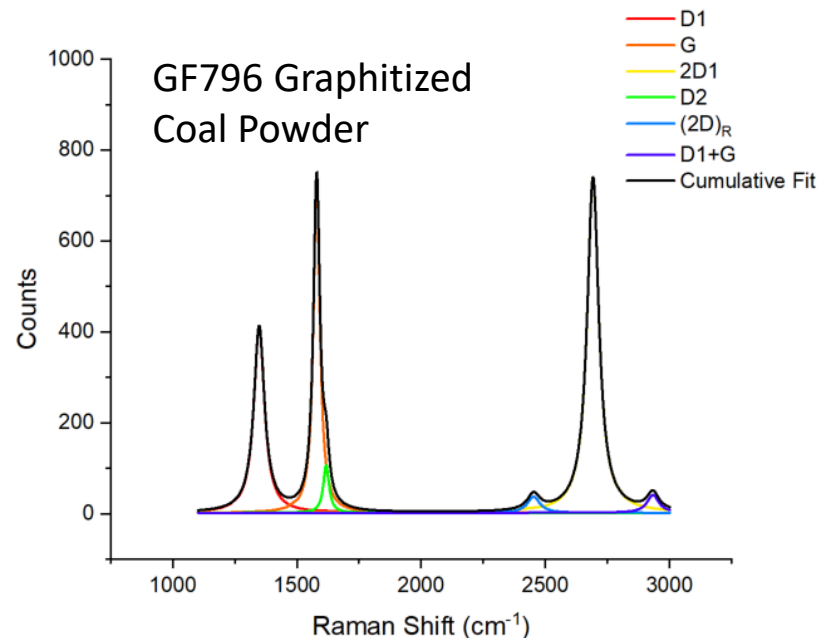
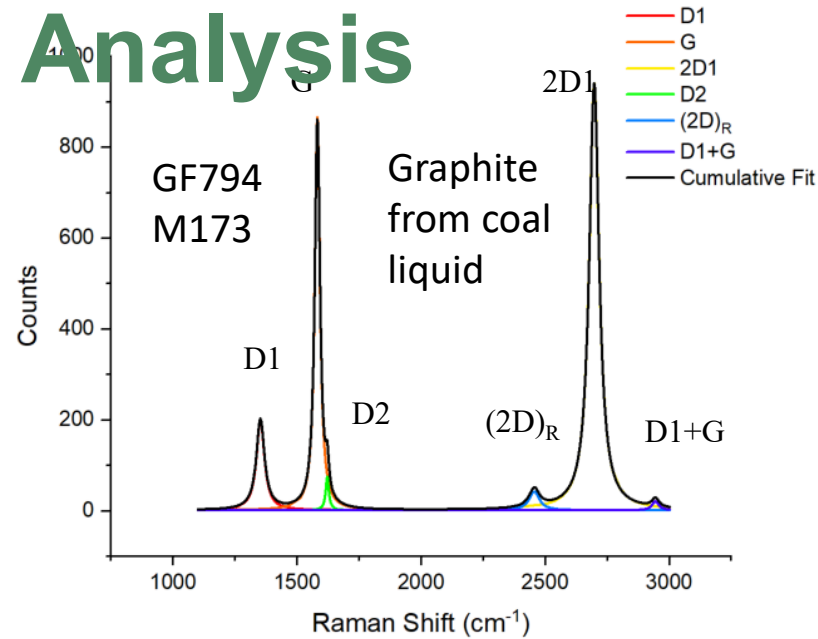
Graphitization of coal powders does not generate a well-formed graphite crystal

Graphite stemming from the coal-liquid-derived mesophase compares favorably with a commercial (coated) spherical graphite, after a heat treatment at 2400 °C.





# Graphitized Mesophase from Coal Liquid — Raman Analysis



	$I_{D1}/I_G$	$A_{D1}/A_G$	G-FWHM
GF794 M173 Graphite-mesophase	0.23	0.36	25
GF796 Graphite - Coal Powder	0.55	0.91	31

Graphitized mesophase pitch (GF794 M173) compared to graphitized coal (GF796)

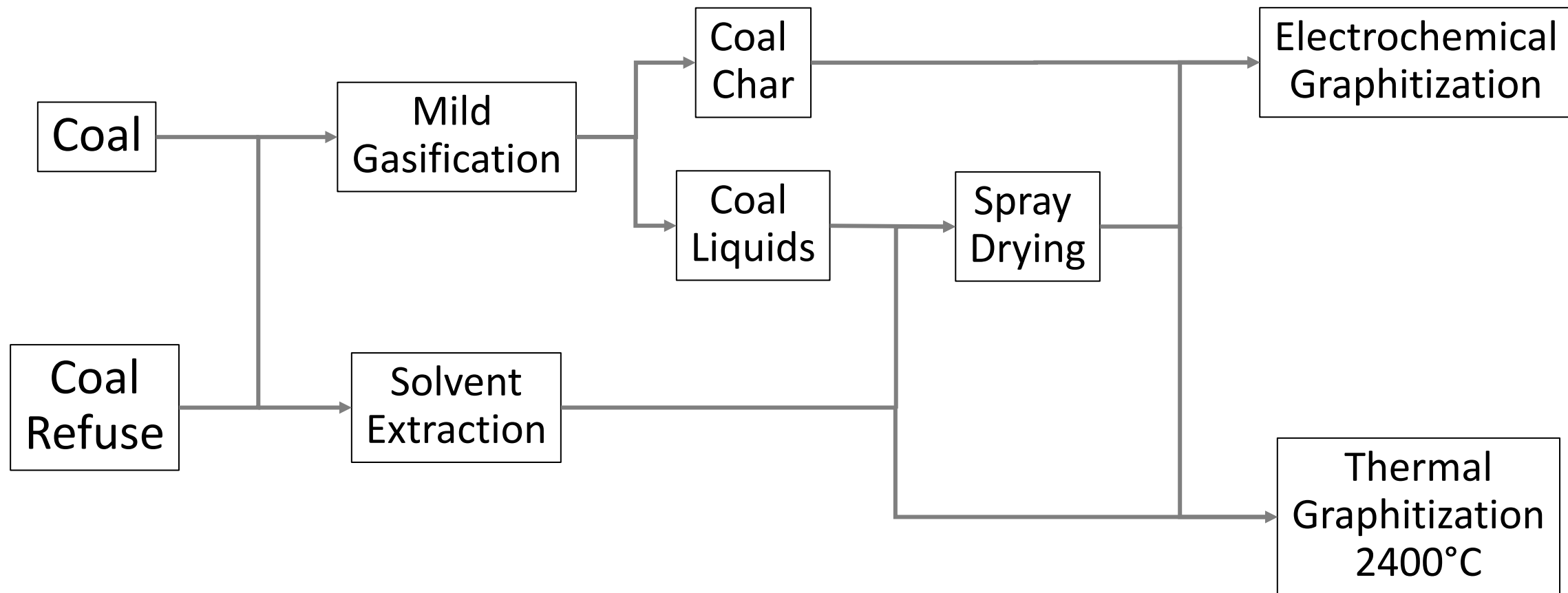
- Increased order evident with reduction in  $I_{D1}/I_G$
- $A_{D1}/A_G$  and G-FWHM decrease with metamorphism, which is commonly used in characterizing the increasing structural order
- For most coal derived graphites, the  $A_{D1}/A_G$  values are between 0.85 and 1.70<sup>1</sup>
- G-FWHM less than 30.3 cm<sup>-1</sup> is classified as a graphite, while a G-FWHM greater than 30.3 cm<sup>-1</sup> is classified as a semi-graphite<sup>2</sup>
- Emergence of 2D1 and D1+G peaks indicate stacking along the c-axis
- The (2D)<sub>R</sub> peaks indicates the formation of a large aromatic ring system

<sup>1</sup>Yuan, L., et al. *Energy Fuels* **2021**, 35, (3), 2335-2346.

<sup>2</sup>Zhang, S., et al. *Carbon* **2020**, 157, 714-723.

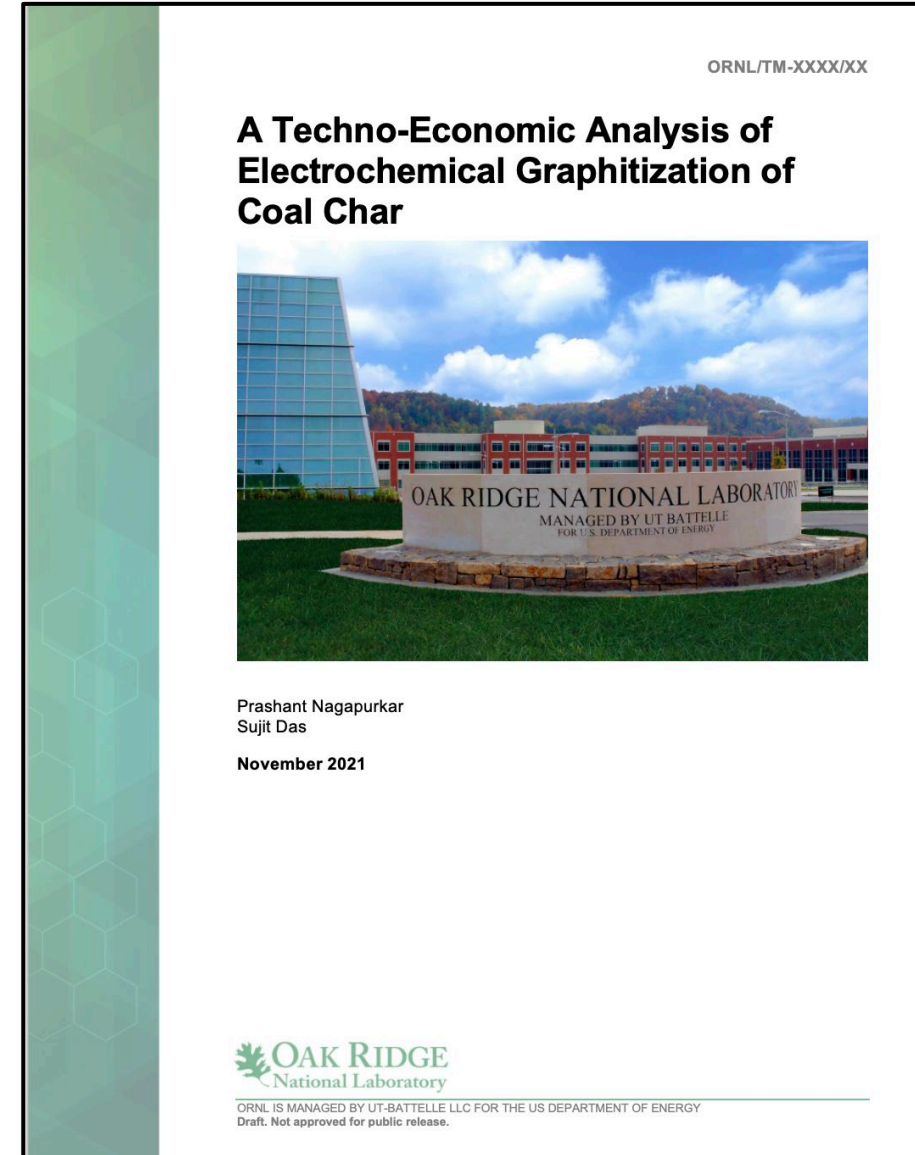


# Different routes to graphite



# Techno-Economic and Life-Cycle Analyses

Technical, economic, and environmental analyses are being formulated to assess the market viability of graphite derived from carbon ore and coal refuse for lithium-ion batteries.



# Technology-to-Market

- One objective of this project is to advance the commercial viability of graphite obtained from carbon ore and coal refuse, including the development of strategies to maximize the chances of commercialization.
- Workshops will be organized to disseminate the results from this project and to engage stakeholders, including Lithium-ion battery manufacturers and OEMs to explore technology transfer and commercialization opportunities.

# Summary

- A process is being developed to scale up the electrochemical graphitization of carbons from different sources.
- Mild gasification of coal and coal refuse can be used to produce high-carbon-content char and coal liquids.
- Spray drying is being used to obtain carbon particles with controllable particle size distribution from coal liquids and mesophase.
- Graphite particles have been obtained by heat treating mesophase obtained by solvent extraction at 2400°C.
- A technoeconomic analysis is being formulated to identify the most cost-effective route for producing graphite for lithium-ion batteries

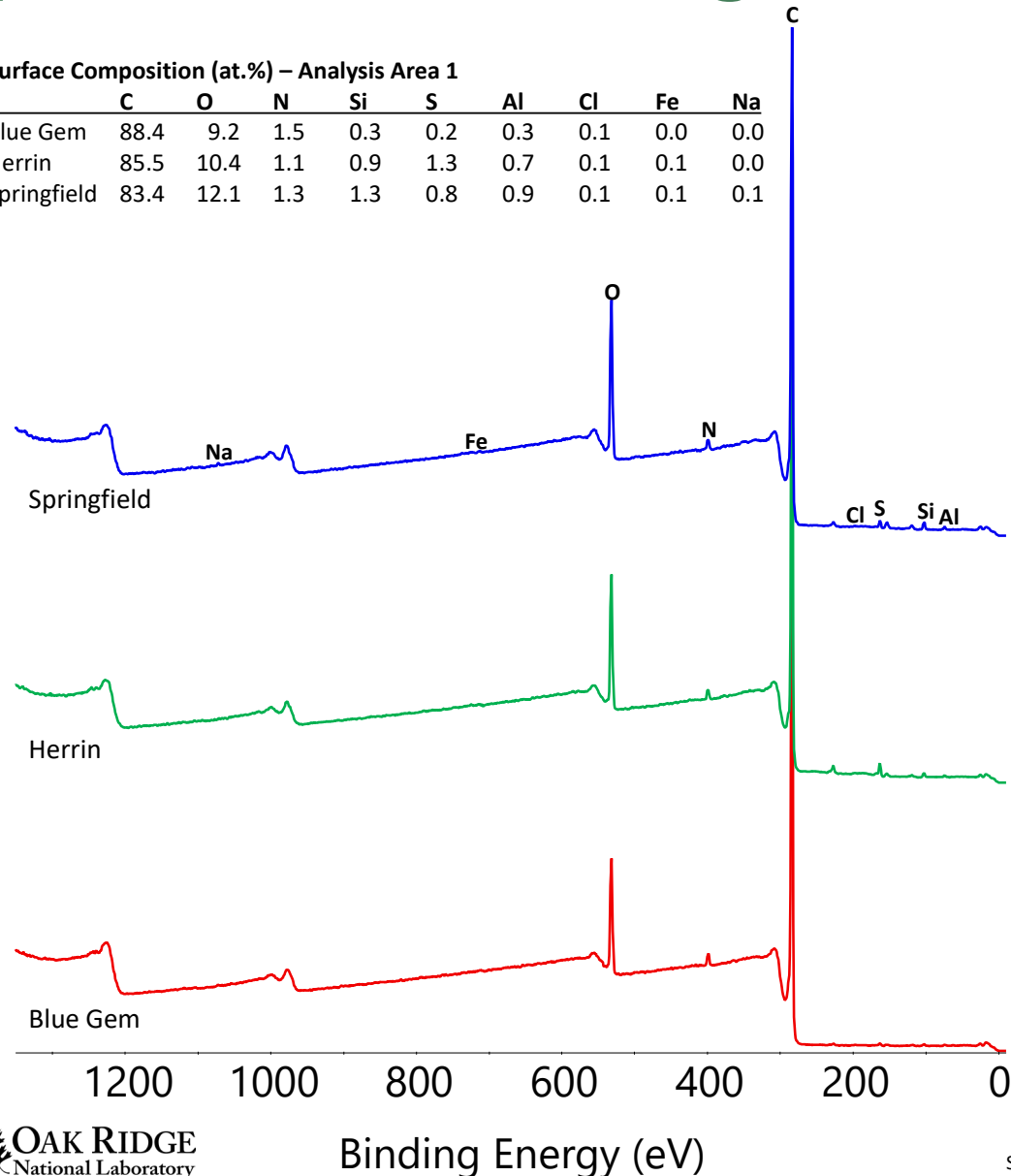
# Questions?

# Additional Slides

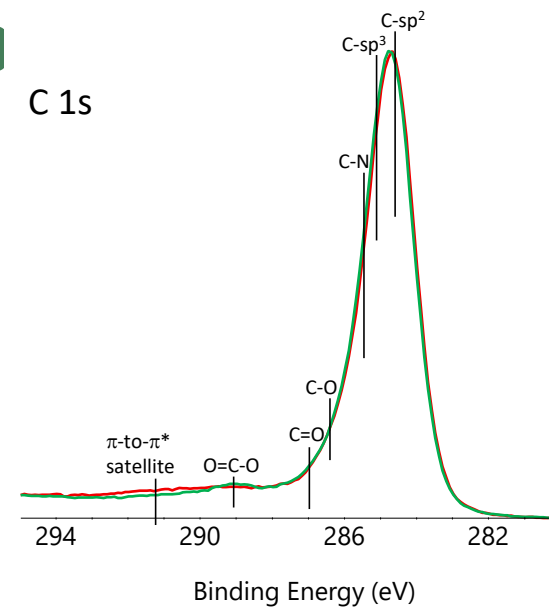
# XPS data provides elemental quantification as well as quantification of organically b

Surface Composition (at.%) – Analysis Area 1

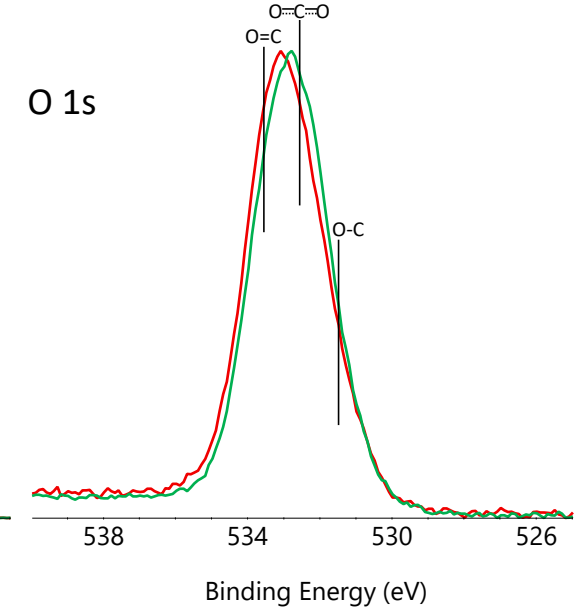
	C	O	N	Si	S	Al	Cl	Fe	Na
Blue Gem	88.4	9.2	1.5	0.3	0.2	0.3	0.1	0.0	0.0
Herrin	85.5	10.4	1.1	0.9	1.3	0.7	0.1	0.1	0.0
Springfield	83.4	12.1	1.3	1.3	0.8	0.9	0.1	0.1	0.1



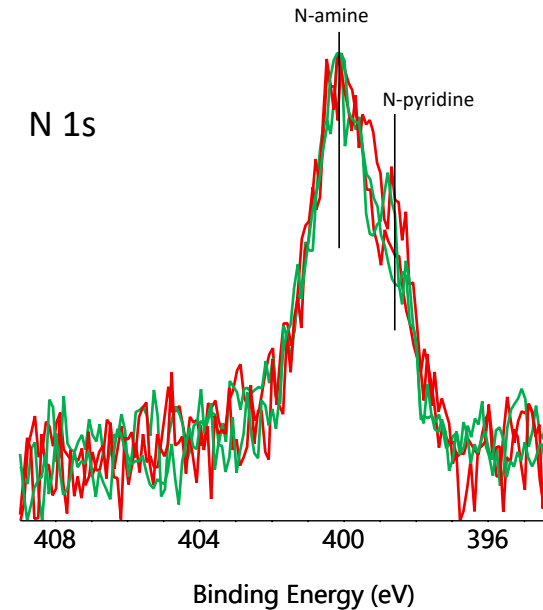
C 1s



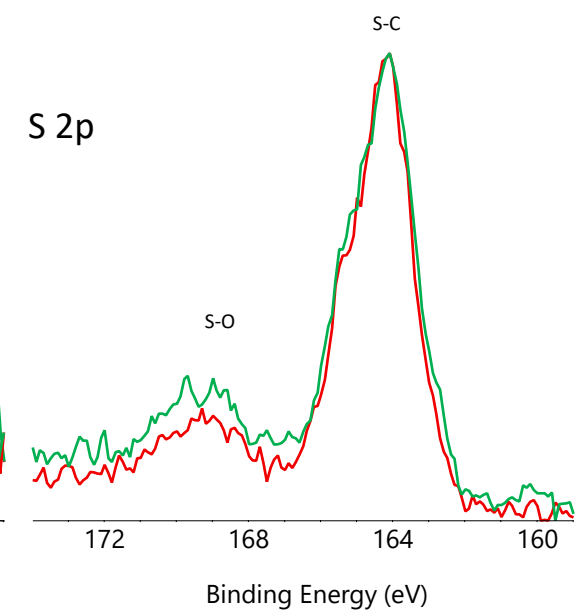
O 1s



N 1s

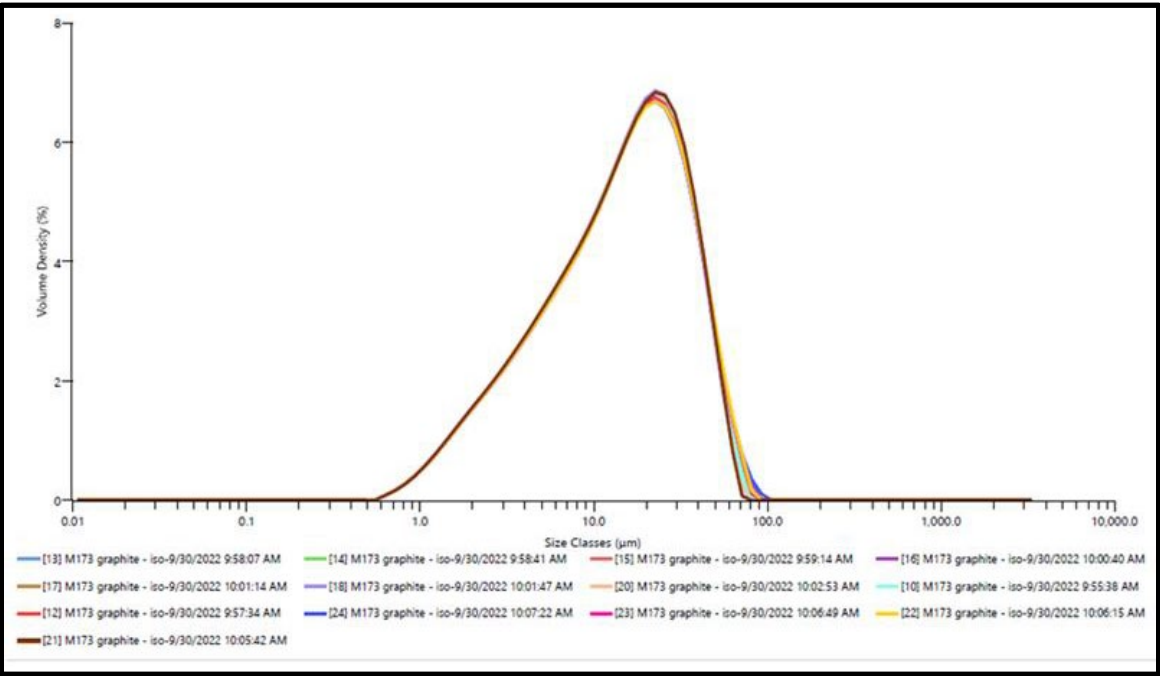


S 2p



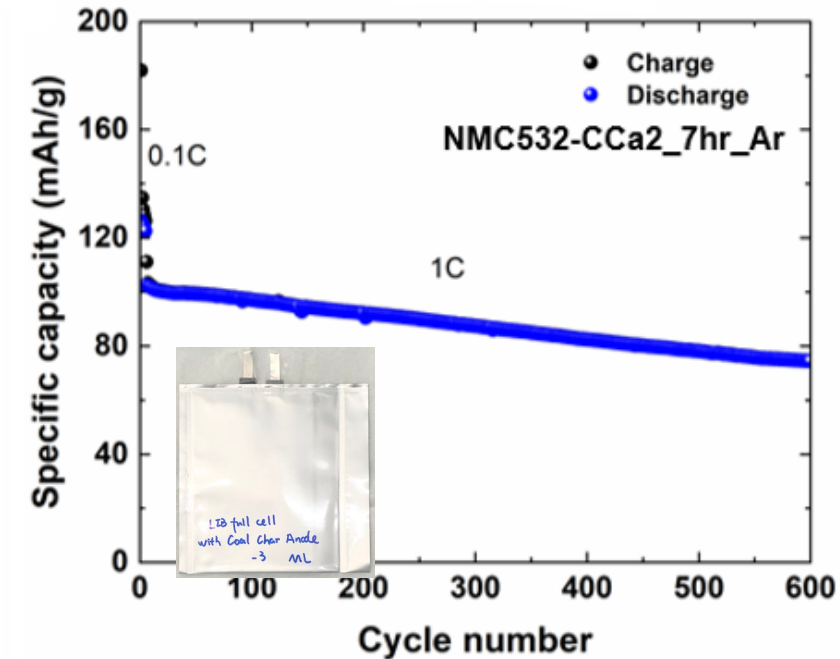
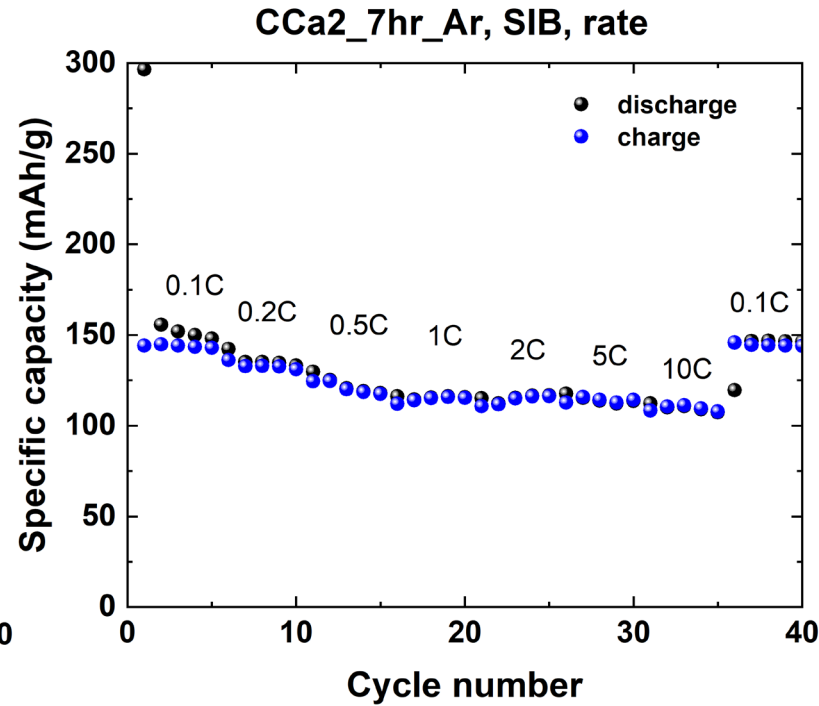
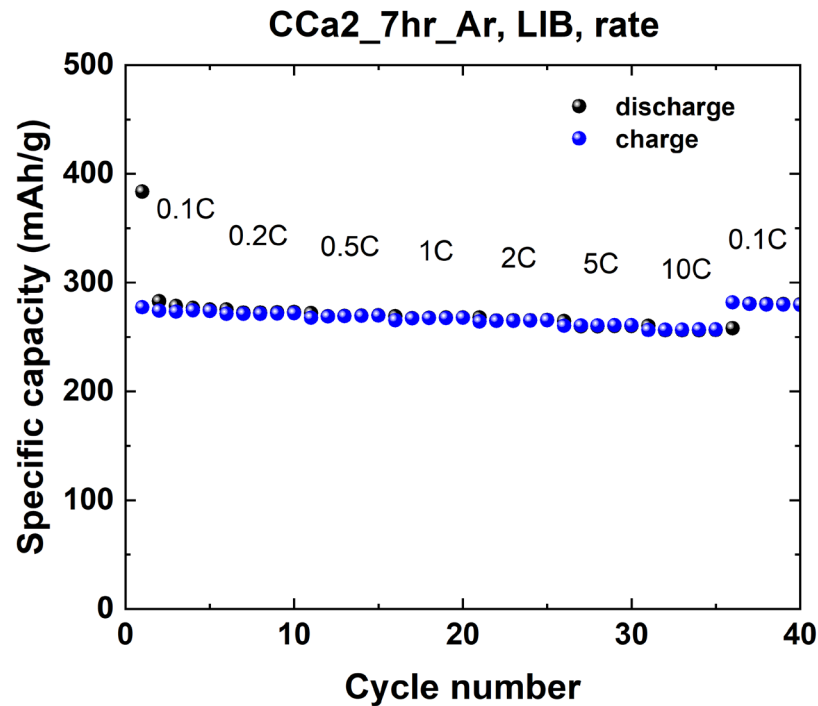


# Size Distribution of Graphitized Mesophase Particles from Coal Liquid



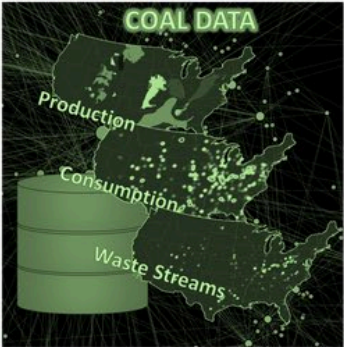


Record Number	Sample Name	Dx (10) (µm)	Dx (50) (µm)	Dx (90) (µm)
13	M173 graphite - iso	2.95	14.8	39.6
14	M173 graphite - iso	2.95	14.7	39.2
15	M173 graphite - iso	2.93	14.6	38.5
16	M173 graphite - iso	2.94	14.7	39.6
17	M173 graphite - iso	2.92	14.5	37.8
18	M173 graphite - iso	2.92	14.5	37.9
20	M173 graphite - iso	2.93	14.5	38.4
10	M173 graphite - iso	2.94	14.6	38.9
12	M173 graphite - iso	2.96	14.8	39.5
24	M173 graphite - iso	2.95	14.8	40.0
23	M173 graphite - iso	2.95	14.8	39.7
22	M173 graphite - iso	2.95	14.8	39.9
21	M173 graphite - iso	2.93	14.5	38.0
Mean		2.94	14.7	39.0
1xStd Dev		0.0129	0.115	0.784
1xRSD (%)		0.438	0.785	2.01

# Electrochemical Performance of Acid-Treated



- Ar-annealed,  $\text{HNO}_3$  treated coal char exhibited excellent rate performance in lithium-ion batteries, and improved capacities at high rate for sodium-ion batteries.
- Full cell powered by Ar-annealed,  $\text{HNO}_3$  treated coal char anode and NMC532 cathode exhibited good cycling @ 1C with 73.2% capacity retention over 500 cycles.

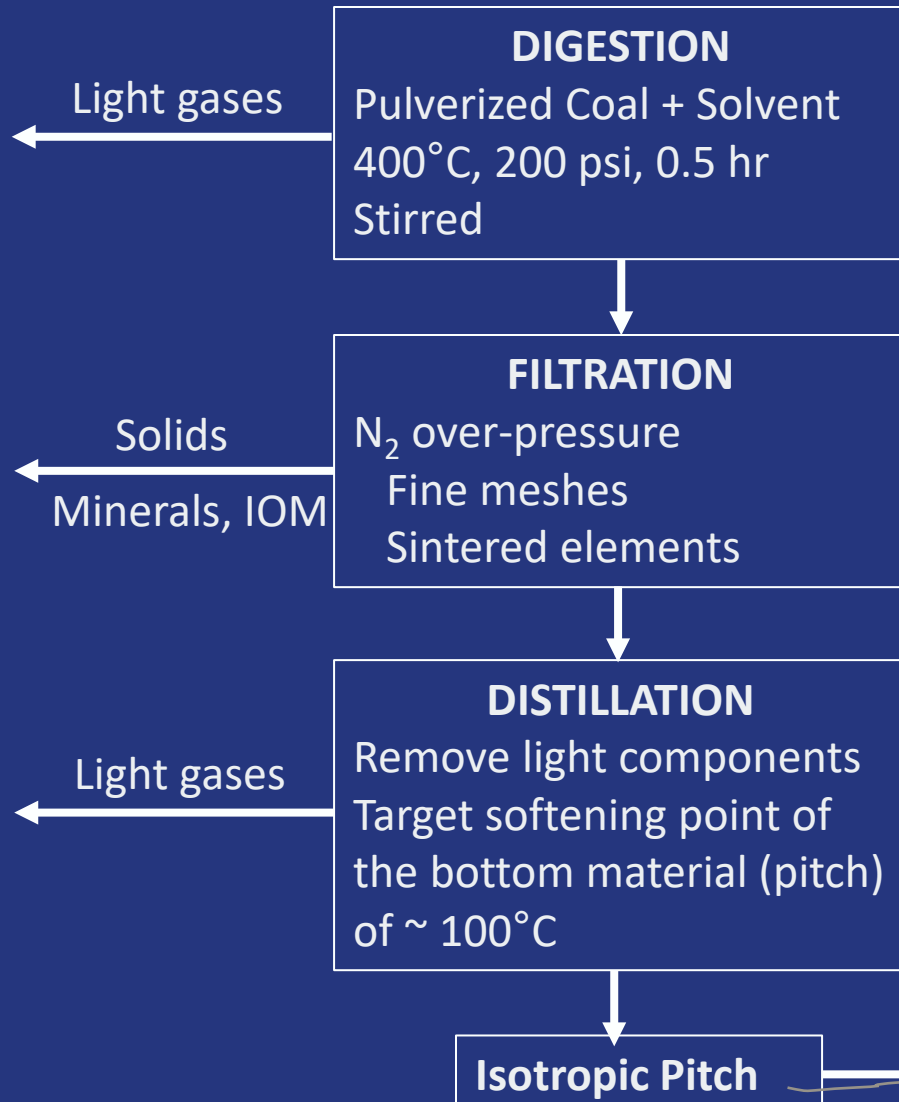
# Data Management

 <p><b>ACD Coal Materials Resources</b> American Coal Database (ACD) is a database under dev at NETL to support coal...</p> <p>♡ Favorite</p>	 <p><b>eXtremeMAT Consortium - IP Council</b> The objective of the eXtremeMAT consortium (XMAT) is to develop...</p> <p>♡ Favorite</p>	 <p><b>NETL-ORNL Coal data share</b> Workspace to share coal data between NETL and ORNL</p> <p>♡ Favorite</p>
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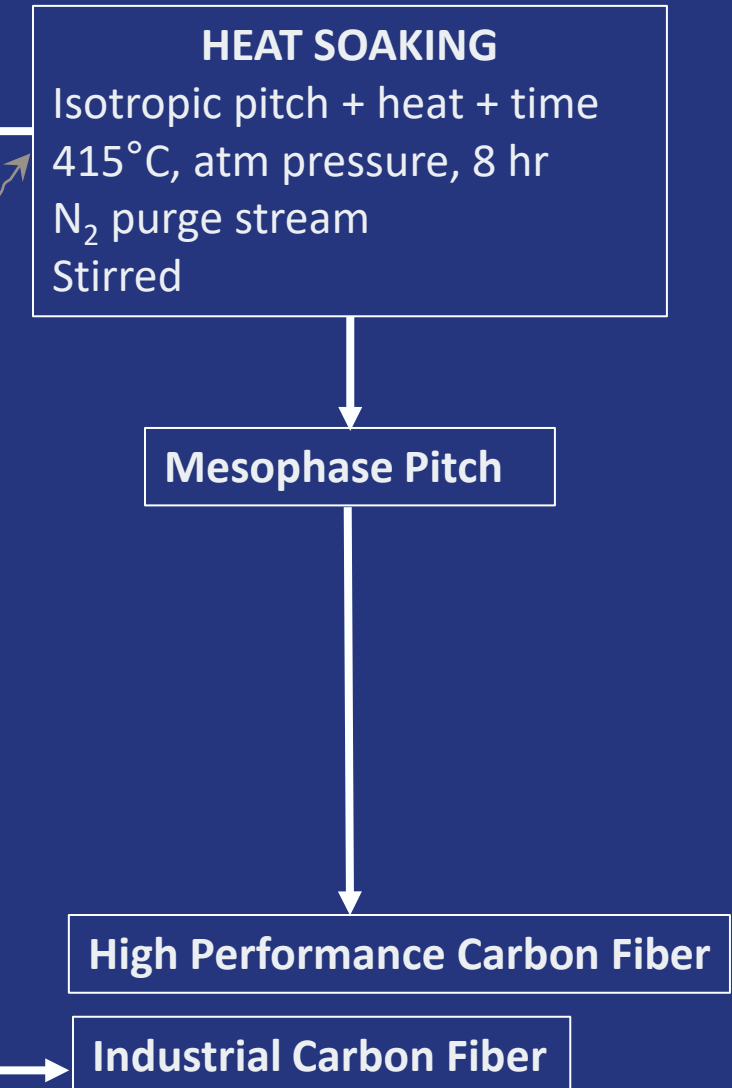
**EDX will be the repository for all data generated in this project**

# Overview – COAL & PITCH Processing

## SOLVENT EXTRACTION PROCESS



## MESOPHASE PROCESS



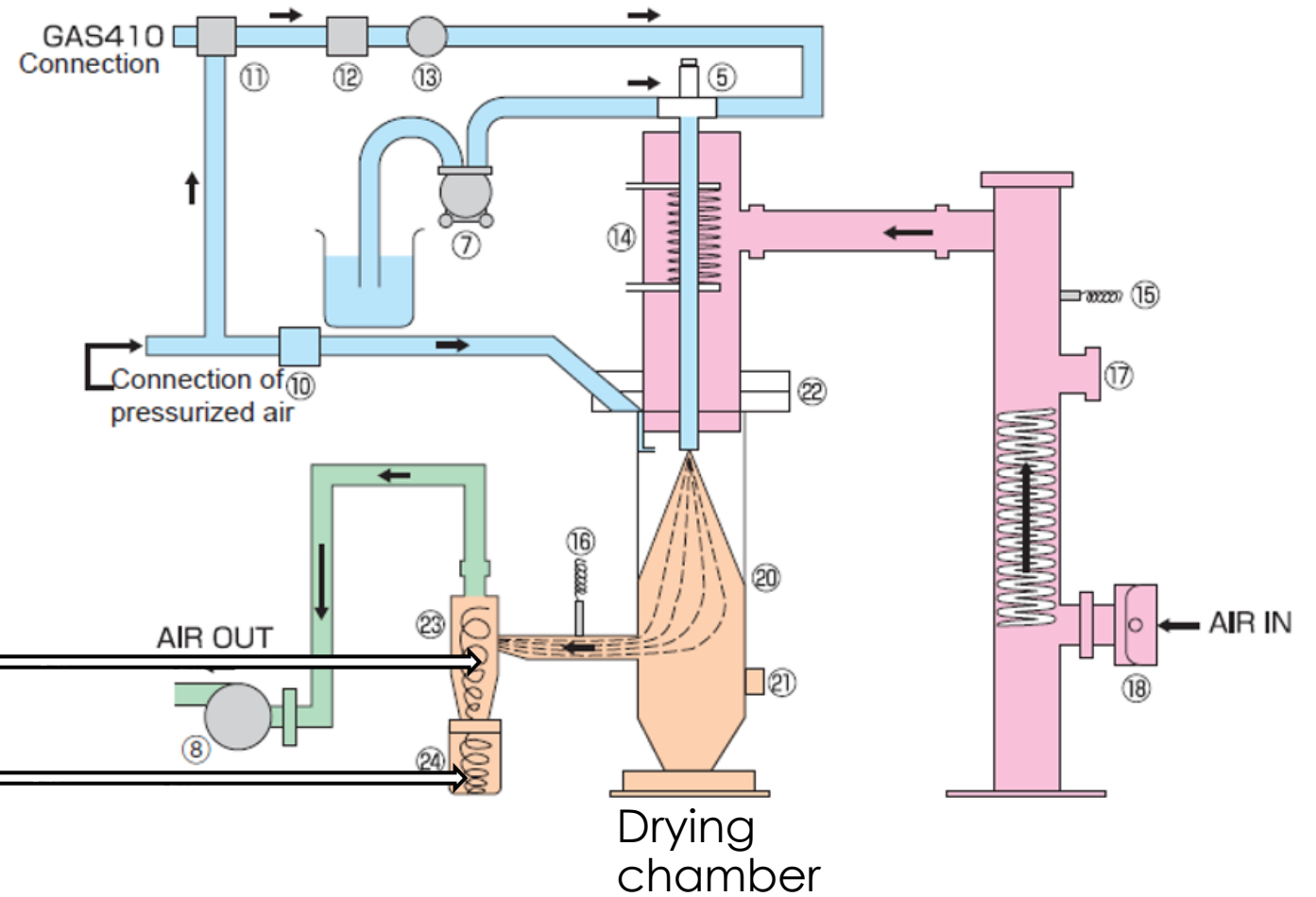
# Spray drying unit GAS410



Drying chamber

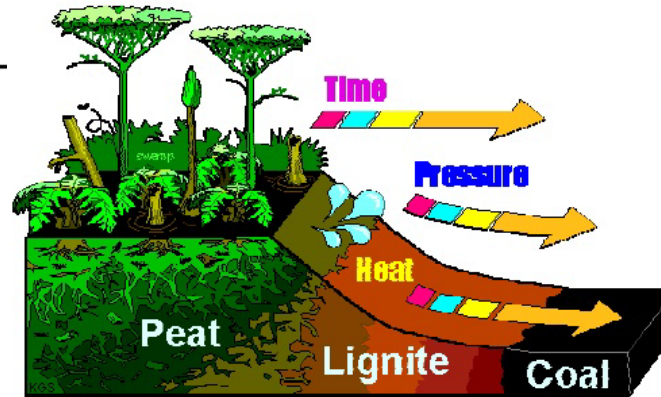
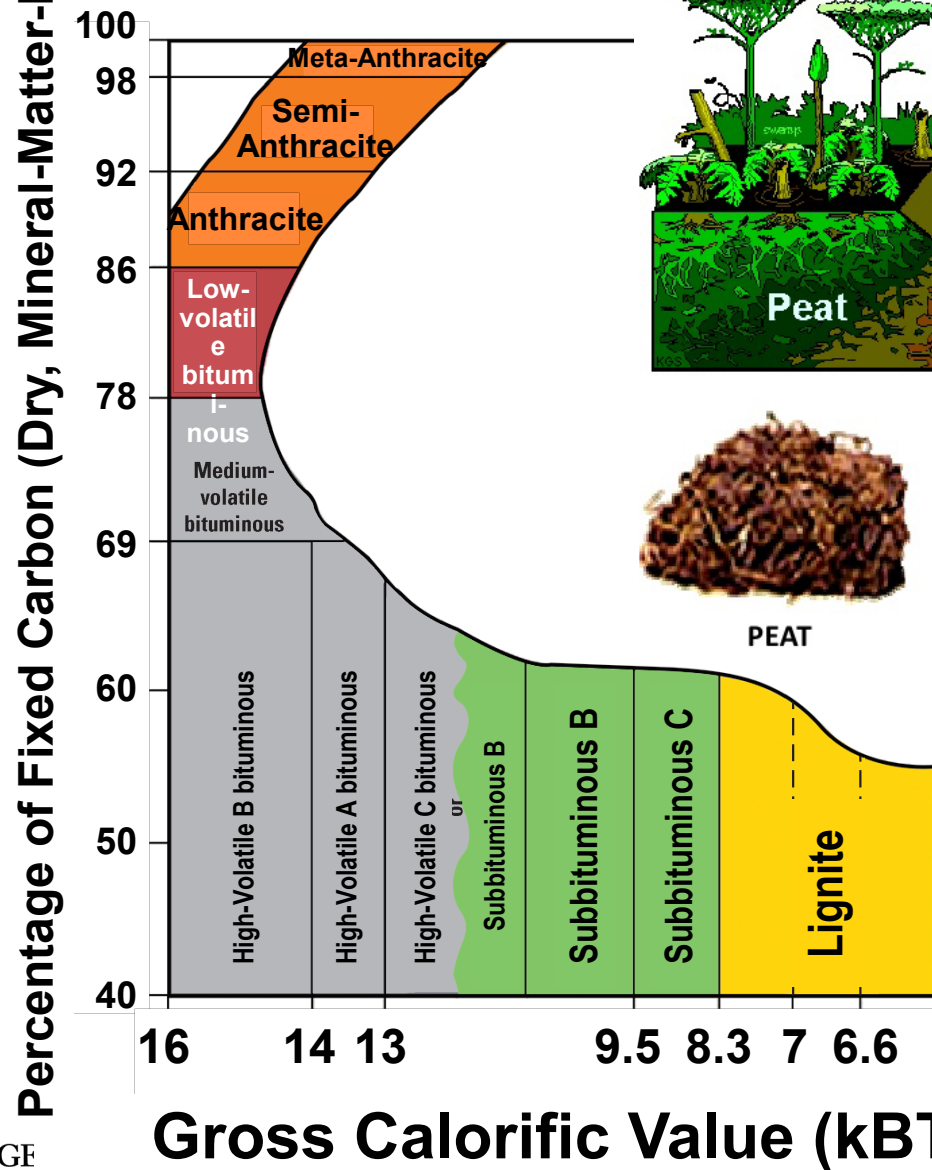
Cyclone

Product collecting container





# Not all coals are the same!



## Coal – Stages of Formation

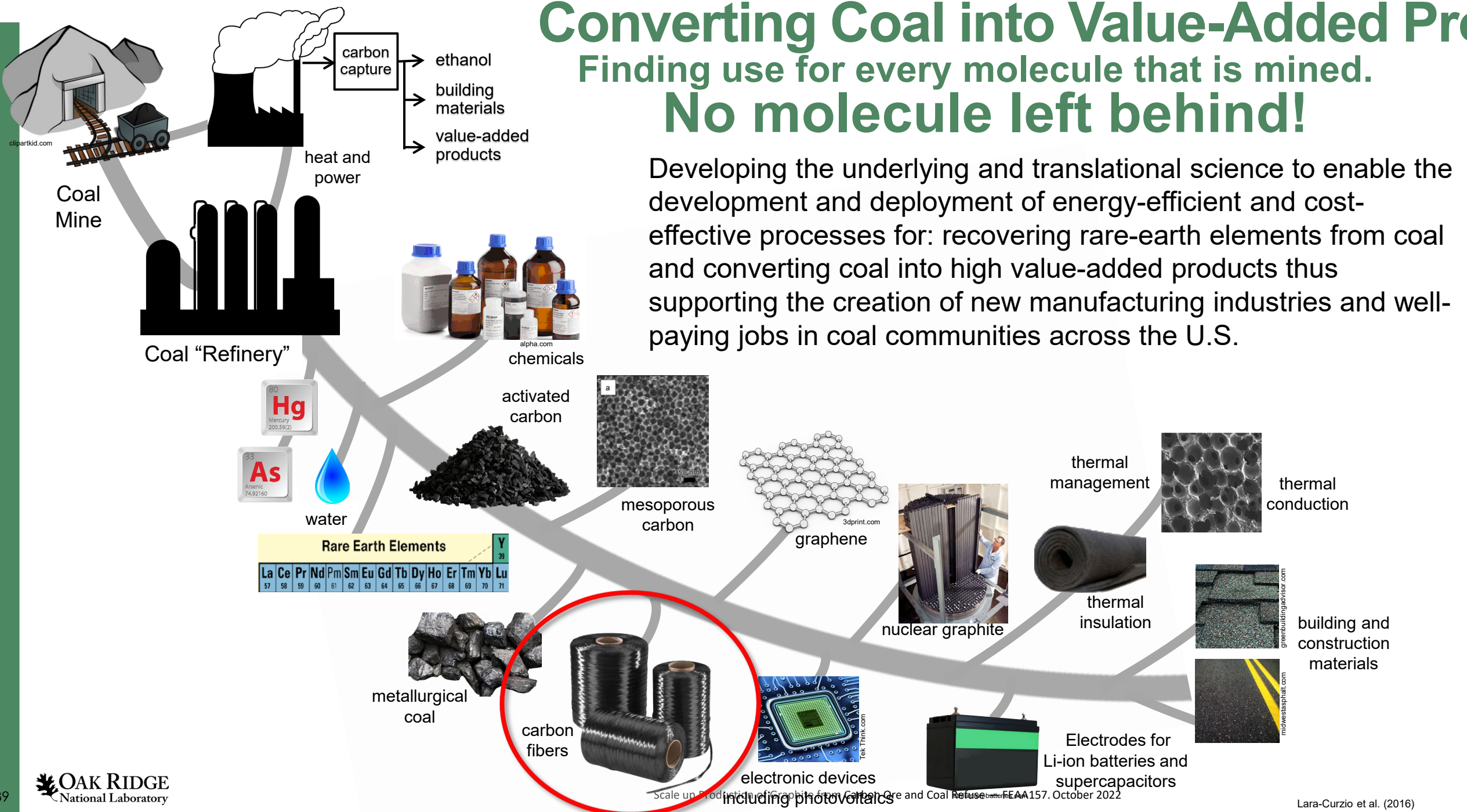


Fossil Fuels. Energy Information Association

# Converting Coal into Value-Added Products

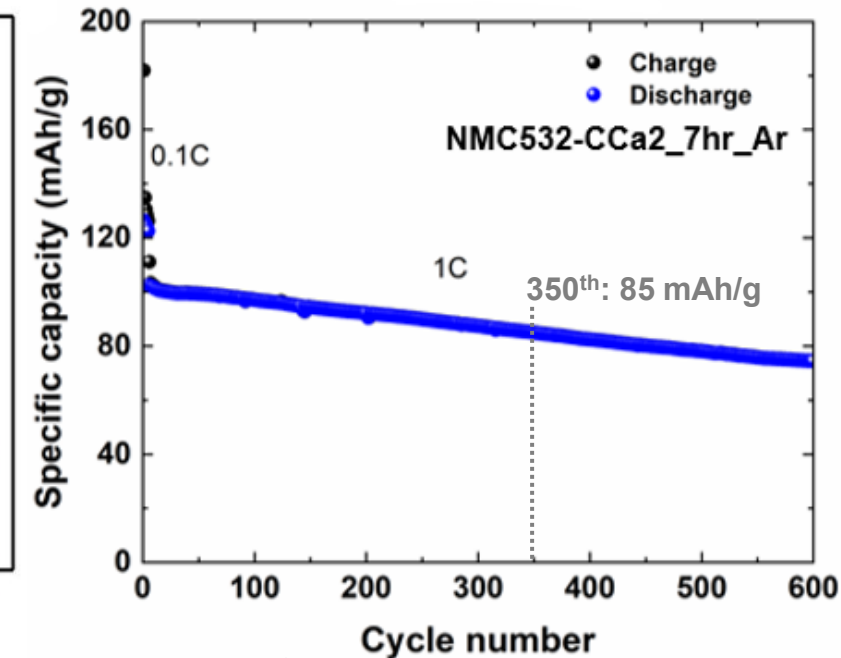
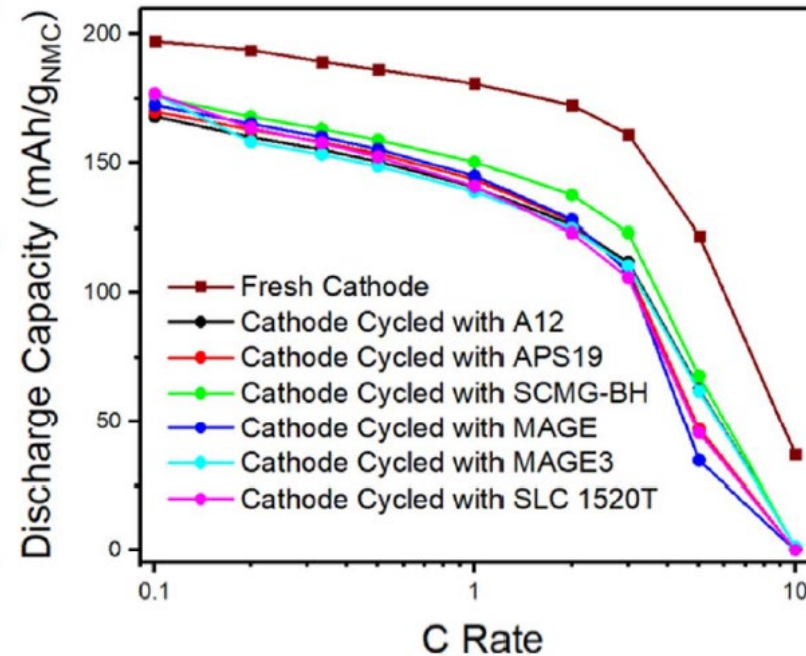
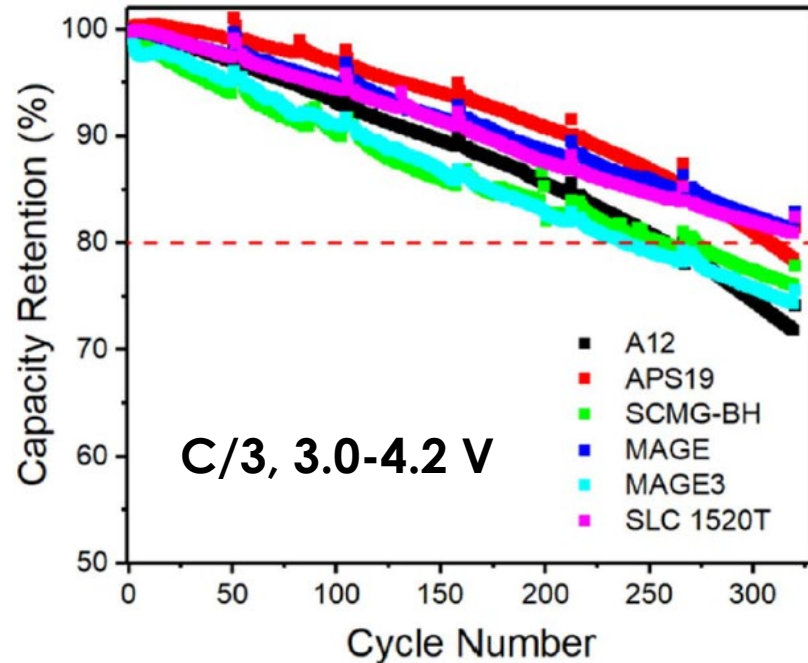
## Finding use for every molecule that is mined. No molecule left behind!

Developing the underlying and translational science to enable the development and deployment of energy-efficient and cost-effective processes for: recovering rare-earth elements from coal and converting coal into high value-added products thus supporting the creation of new manufacturing industries and well-paying jobs in coal communities across the U.S.



# Conventional Graphite Anode in LIBs as a Comparison

- Different types of graphite against NMC811 in LIB full cells (capacity normalized to cathode mass):



J. Electrochem. Soc., 2018, 165 (9) A1837-A1845.

★ Coal anode (our work)

- Initial capacity (graphite anode) at 1C charge/discharge rate: ~ 140 mAh/g.
- Assuming 80% capacity retention over 350 cycles, estimated discharge capacity (graphite anode) at 350<sup>th</sup> cycle: ~ 112 mAh/g.
- Notice: the specific capacity of NMC532 is lower than NMC811.



# Demand for Graphite for Li-ion (ont.)

Press Release

## Electric Vehicle Companies Could Have a Massive Graphite Supply Problem

Published: March 25, 2021 at 9:30 a.m. ET



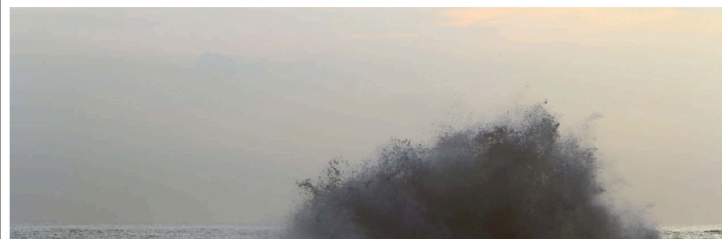
The MarketWatch News Department was not involved in the creation of this content.

Mar 25, 2021 (Bays electric vehicles from Motors (NYSE:GM) increasing the need as Ceylon Graphite (OTC:GPHOF), and graphite is an essential batteries using in e



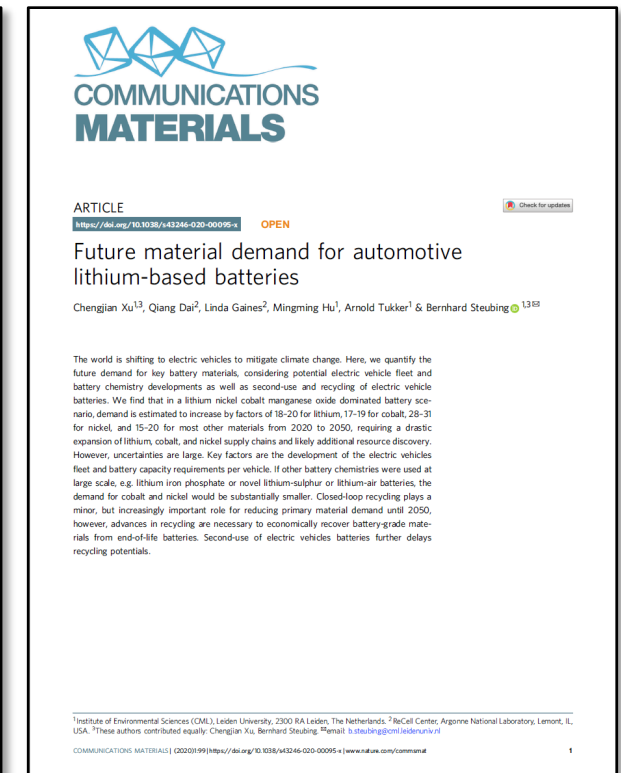
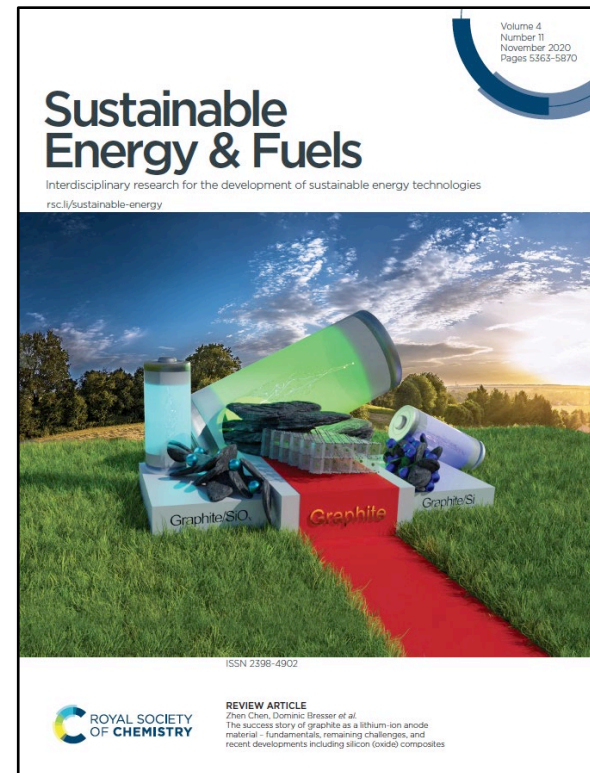
HOME » MARKET OPINION » MARKET ANALYSIS INTEL

## The coming tsunami of electric car demand will need many more anode megafactories and graphite



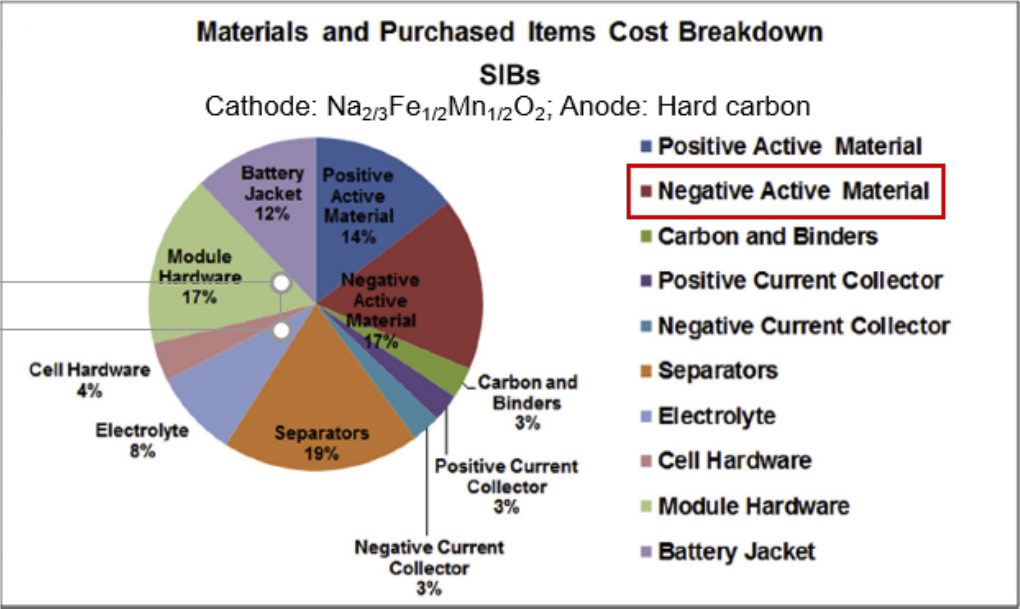
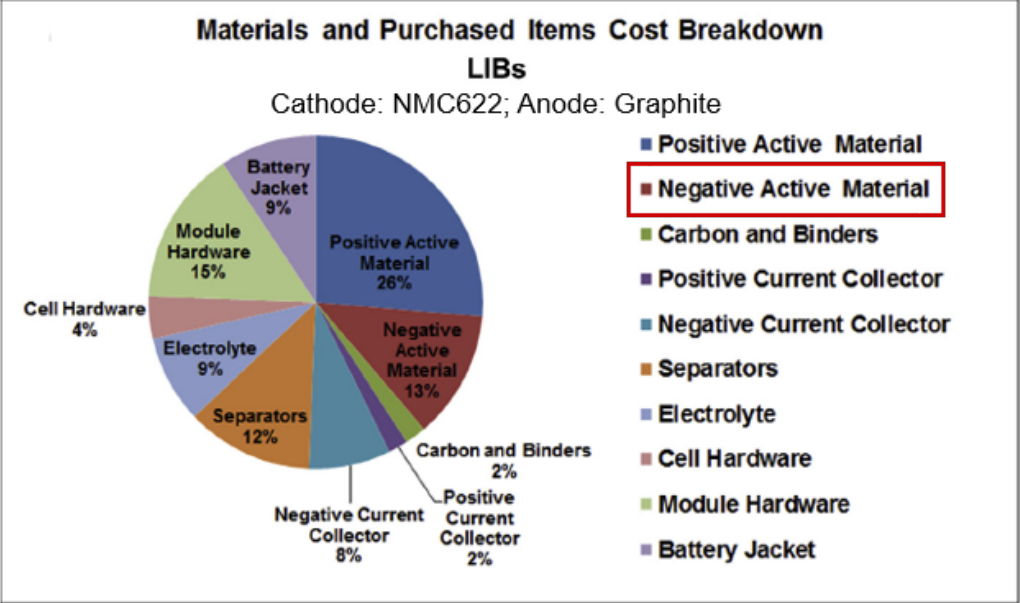
MATTHEW BOHLSSEN | APRIL 29, 2019 | 2 COMMENTS

Most investors don't yet understand the tsunami of electric car demand that is just around the corner. Bloomberg New Energy Finance forecasts that by 2020 there will be over 289 different models of electric cars. Just recently Bloomberg has revised their targets now saying the same as I have said for the past 3 years. Bloomberg now says by 2022 electric cars will become price competitive with conventional cars. Previously they said by 2025. Even Volkswagen predicts that EVs will go mainstream in 2022.

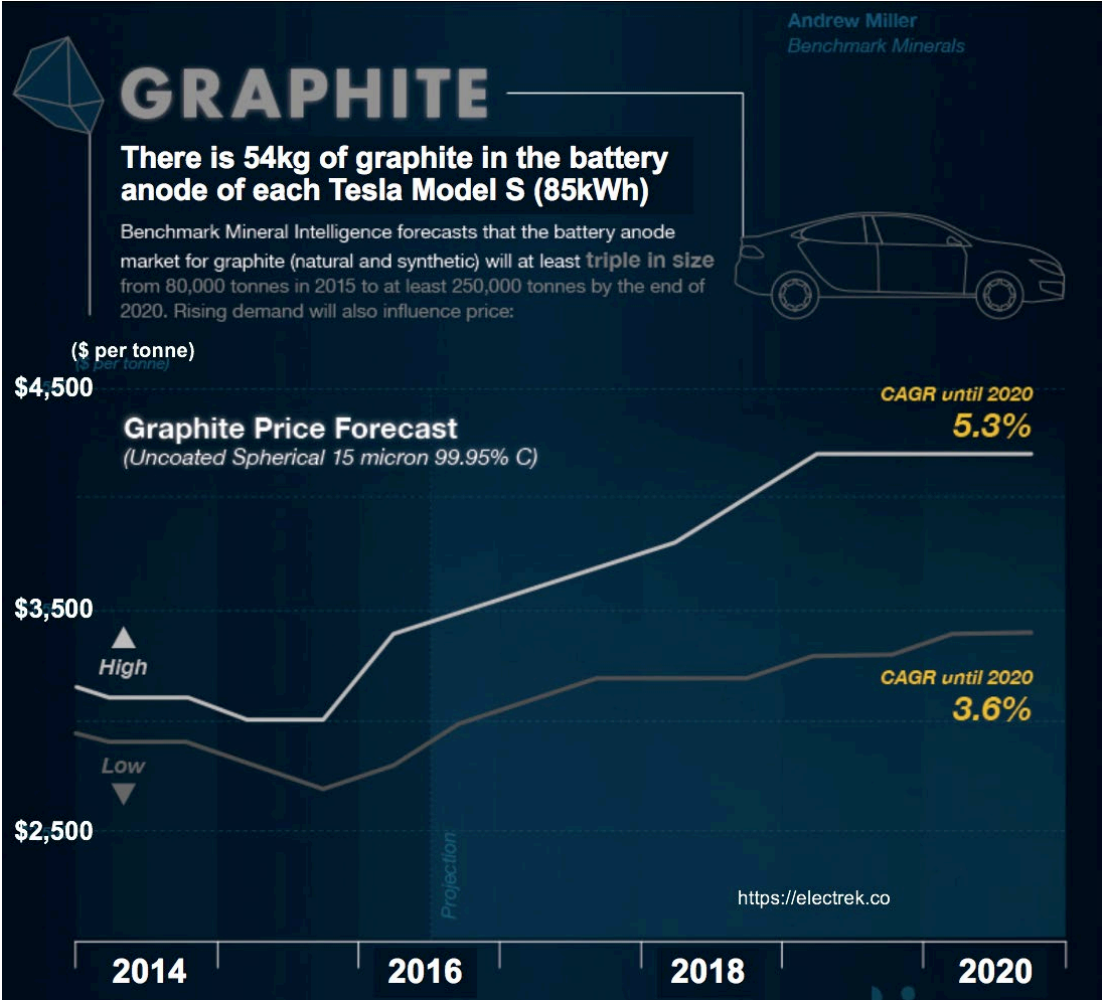




# Cost Perspective in Batteries



	Negative active materials cost percentage
Li-ion Batteries (LIBs)	13%
Na-ion Batteries (SIBs)	17%

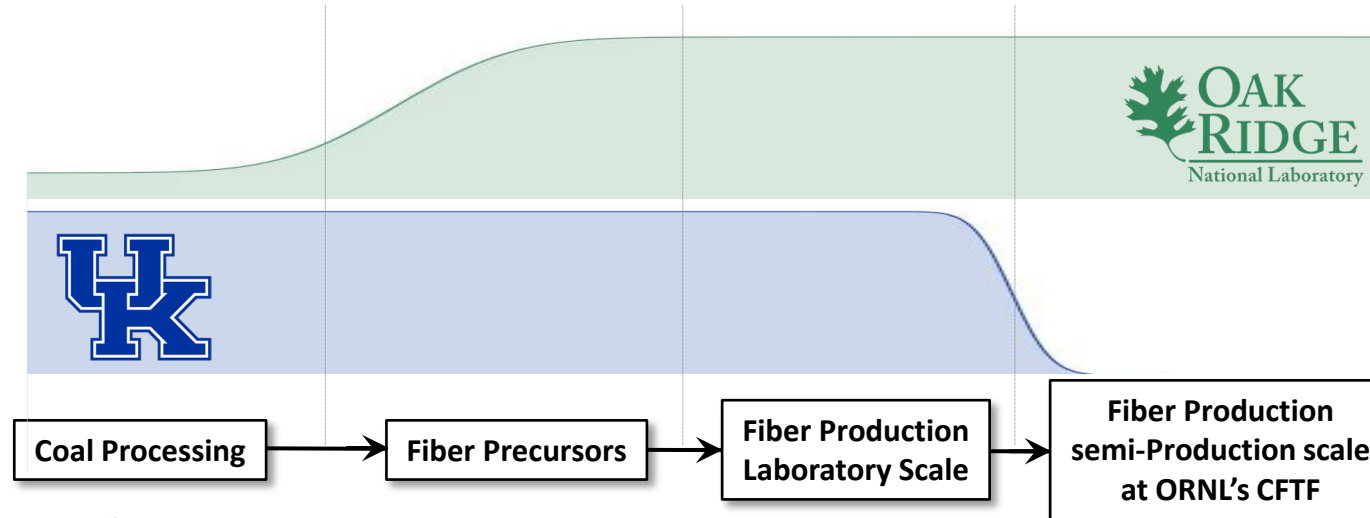


# The ORNL-UK Partnership to Develop Coal-Derived Carbon Fibers

## A Perfect Match

Together, ORNL and UK bring **complementary and unparalleled capabilities** in fundamental science and translational research and development, with particular expertise in:

- coal processing
- separation science and technology
- carbon science & technology
- computational chemistry and high-performance computing
- advanced characterization
- advanced manufacturing



Continuous carbon fibers for structural applications



Composite Materials and Structures



Short carbon fibers for structural applications



Short carbon fibers for thermal insulation



**Objective:** To develop scalable, efficient, cost-effective, and environmentally sustainable processes for manufacturing **low-cost** coal-derived carbon fibers with **tunable physical and mechanical properties**. A key element of this project is **scaling** fiber production from the laboratory benchtop level up to semi-production scale at ORNL's Carbon Fiber Technology Facility. This project will demonstrate a clear path for competitive industrialization of coal-derived carbon fibers and composites for a wide range of applications.



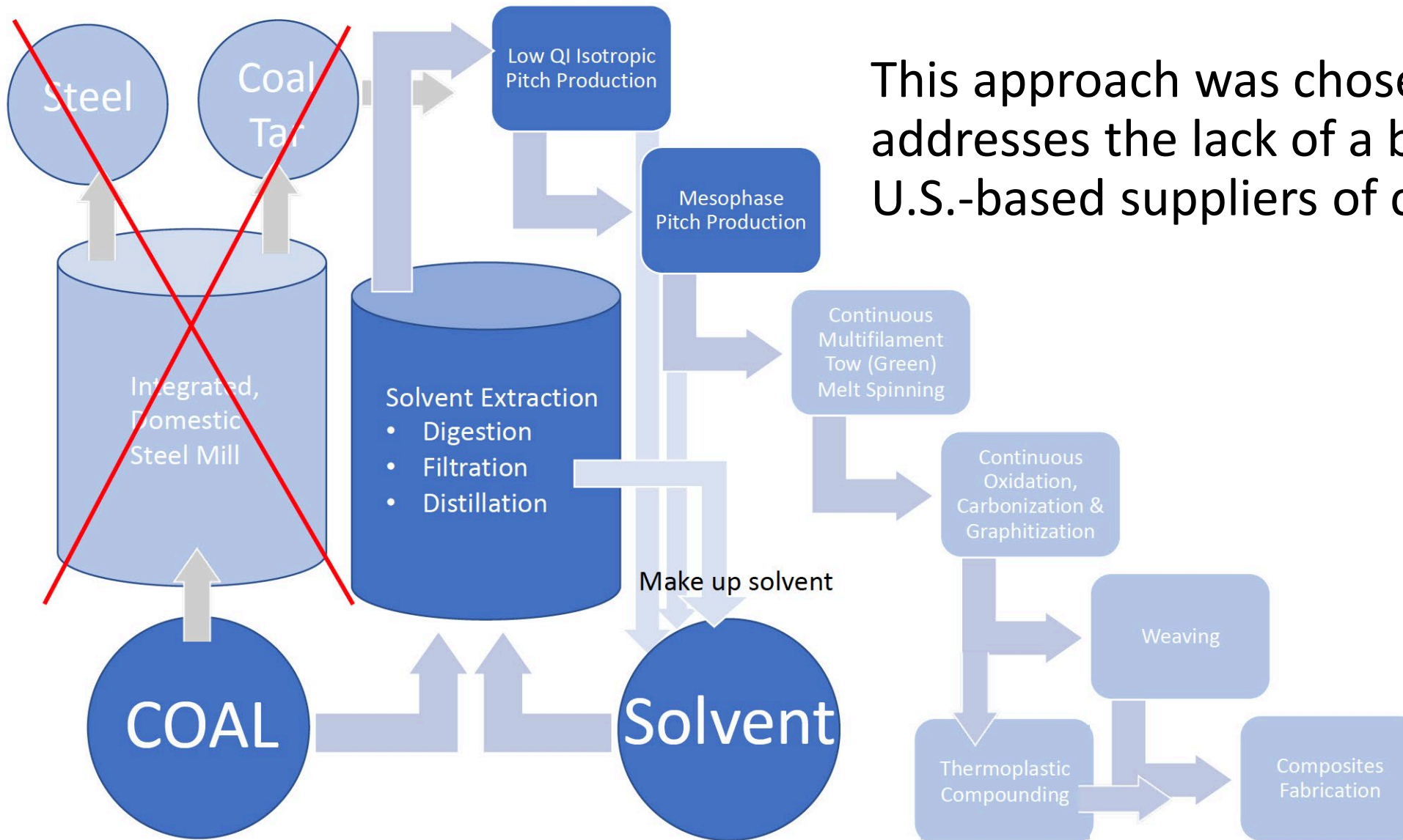
# ORNL's Carbon Fiber Technology Facility (CFTF)



Established in 2013, the CFTF is the Department of Energy's only designated user facility for carbon fiber innovation.

- 42,000 sq. ft. facility
- 390 ft. long processing line, capable of custom unit operation configuration
- Up to 25 tons per year

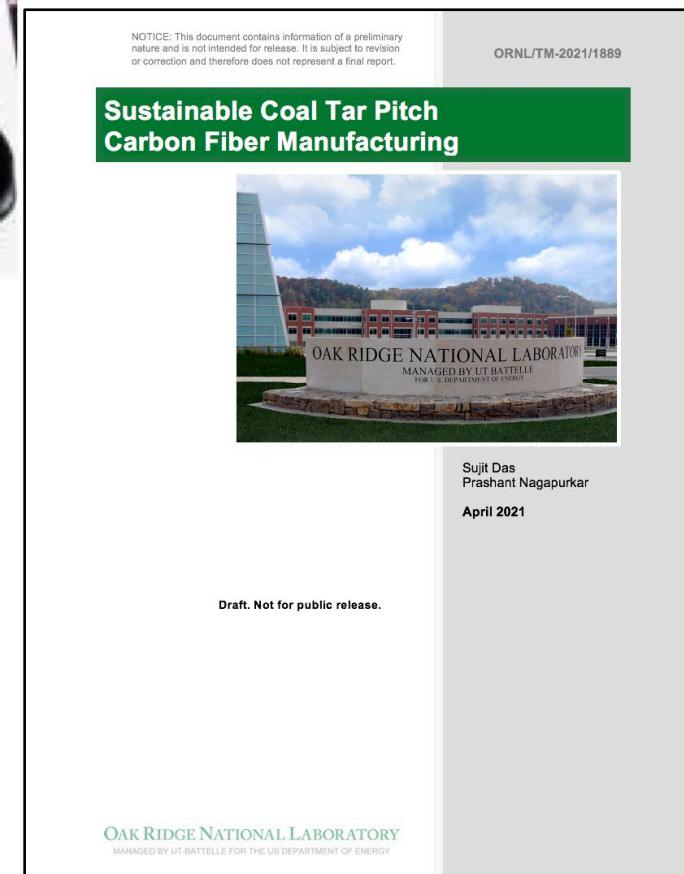
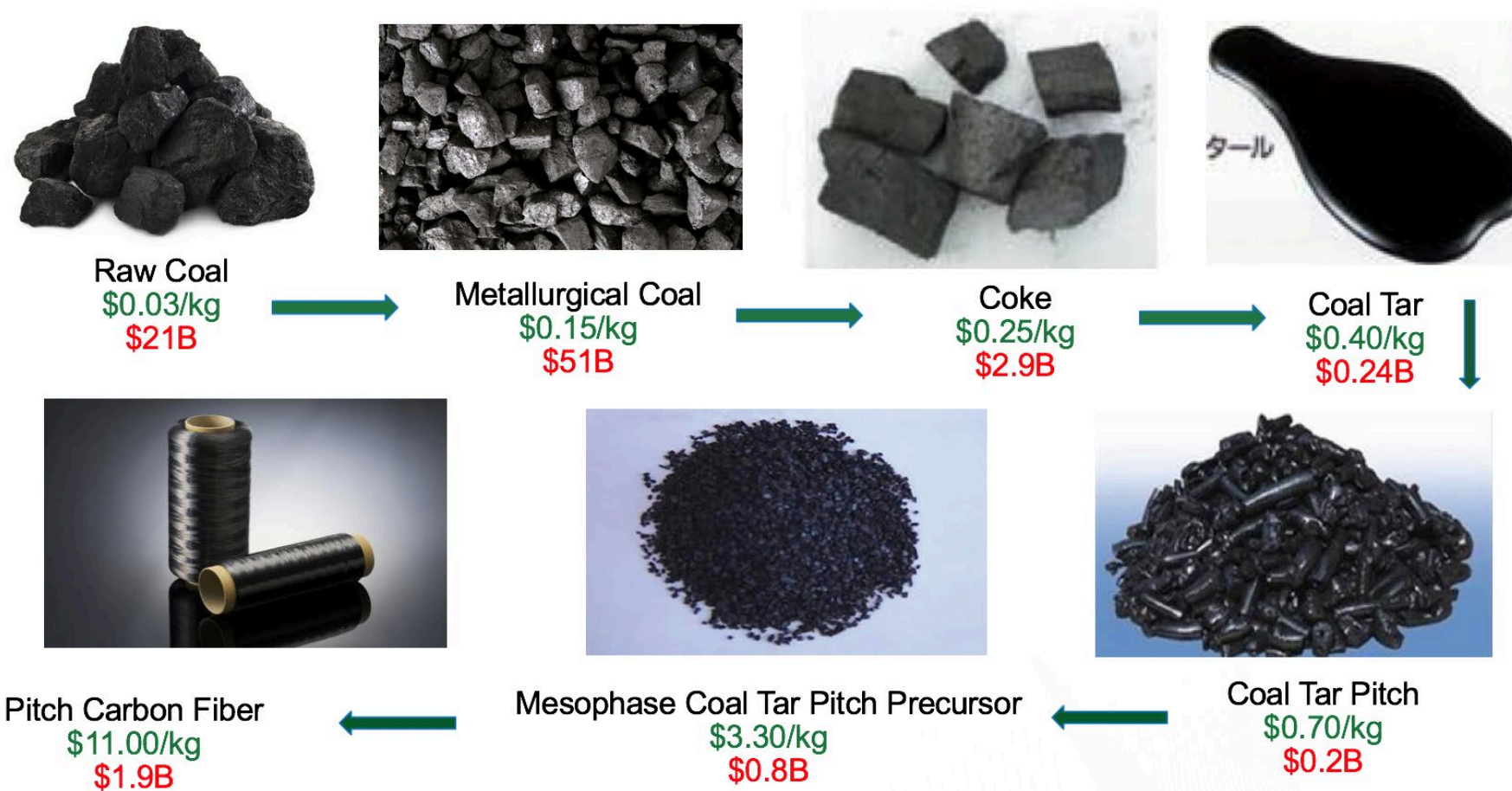
# Advanced Coal Processing at the University of K



This approach was chosen because it addresses the lack of a broad base of U.S.-based suppliers of coal tar pitch.



# Life Cycle and Technoeconomic Analyses of Coal-Derived Carbon Fiber Manufacturing



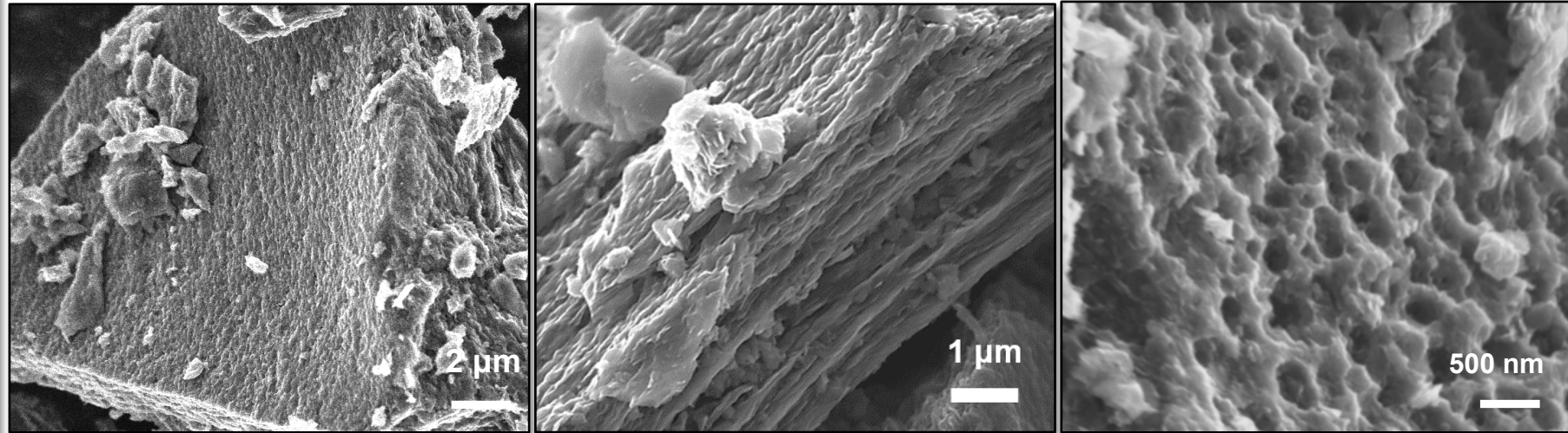
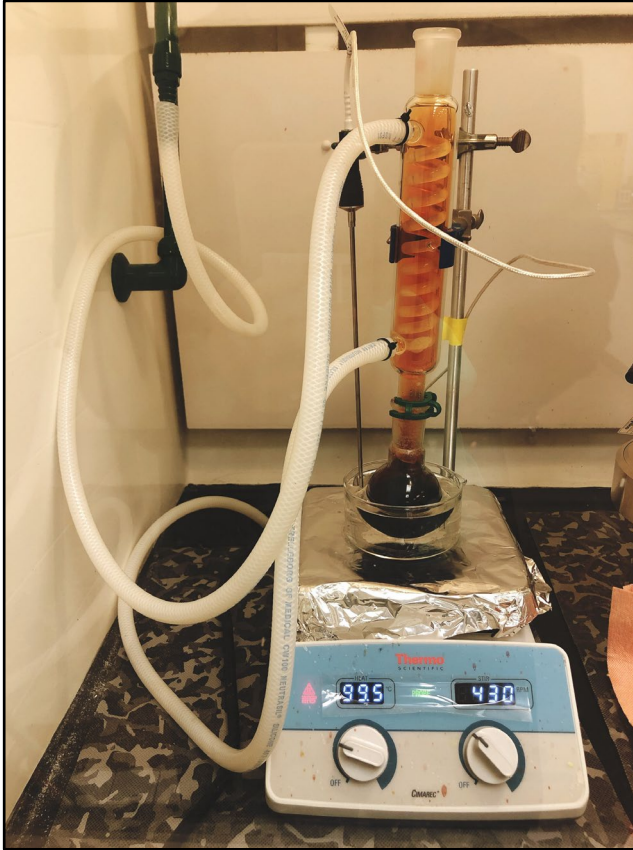
If there is no pre oxidation, the powder coalesces into a single piece during carbonization and graphitization





# $\text{NH}_4\text{NO}_3$ Treatment of Coal Char

Acid-reflux system (100°C)

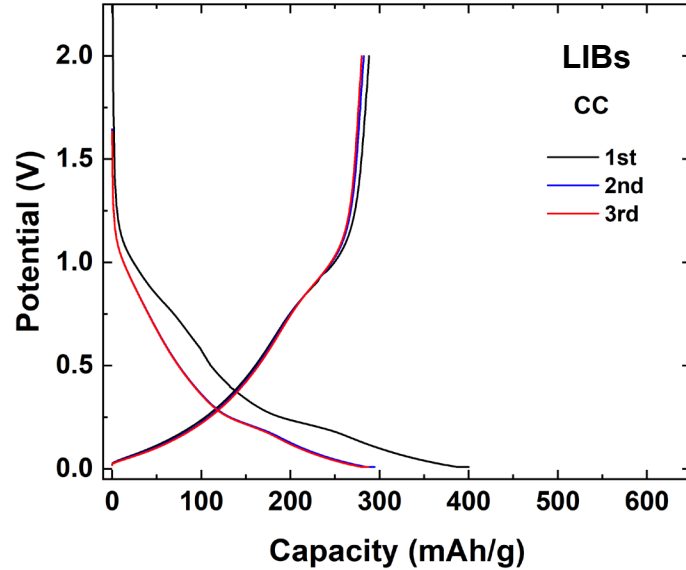


SEM of nitric acid-treated coal char:

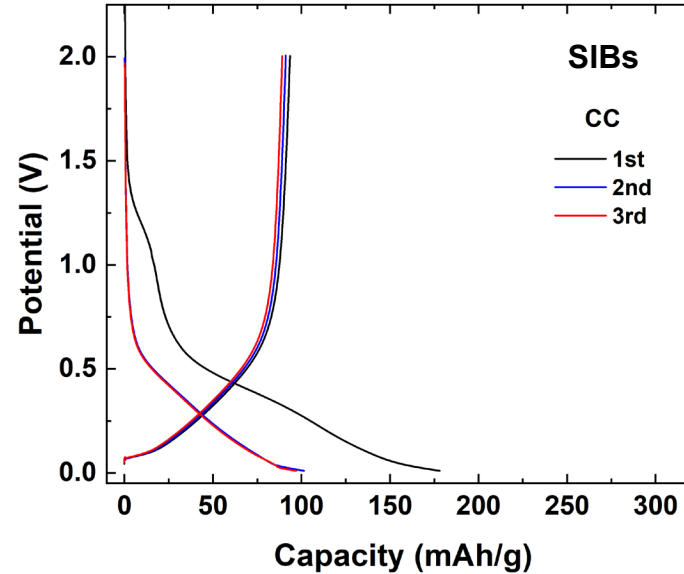
- Porous surface with uniformly distributed nanopores
- Layered structure

# Electrochemical Performance of as-Received Coal

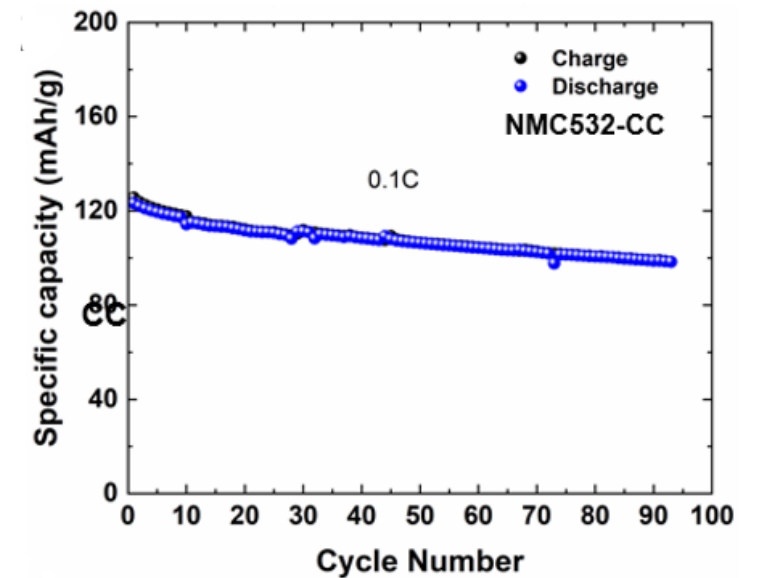
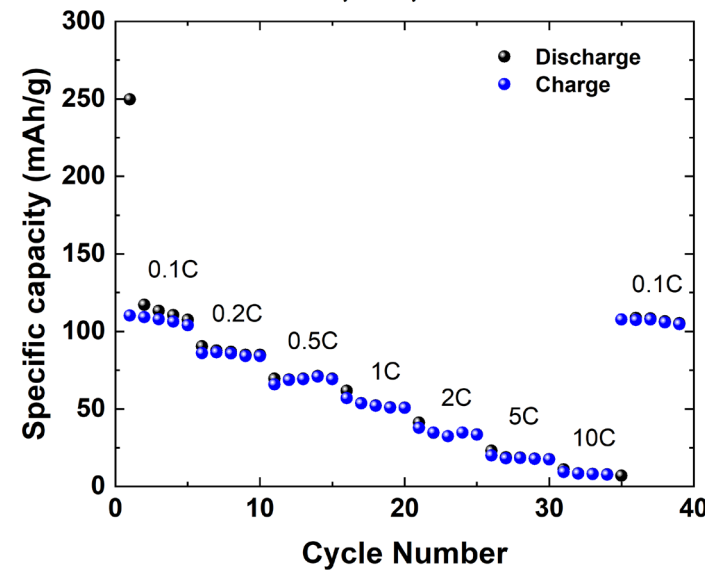
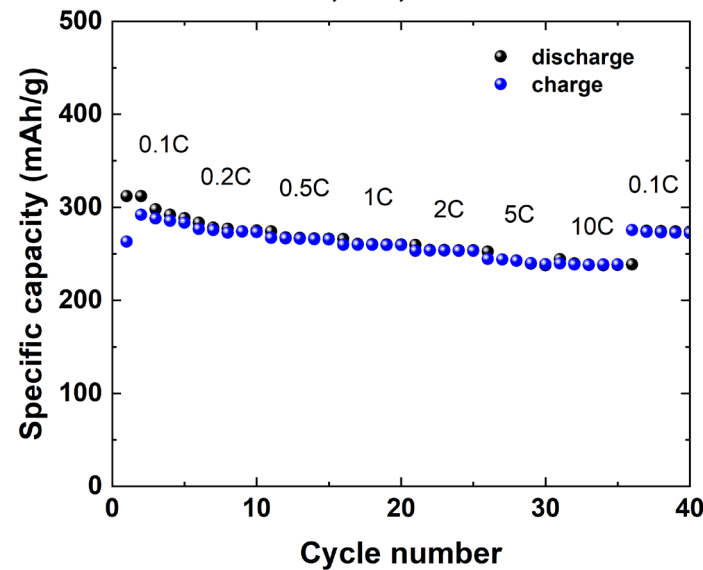
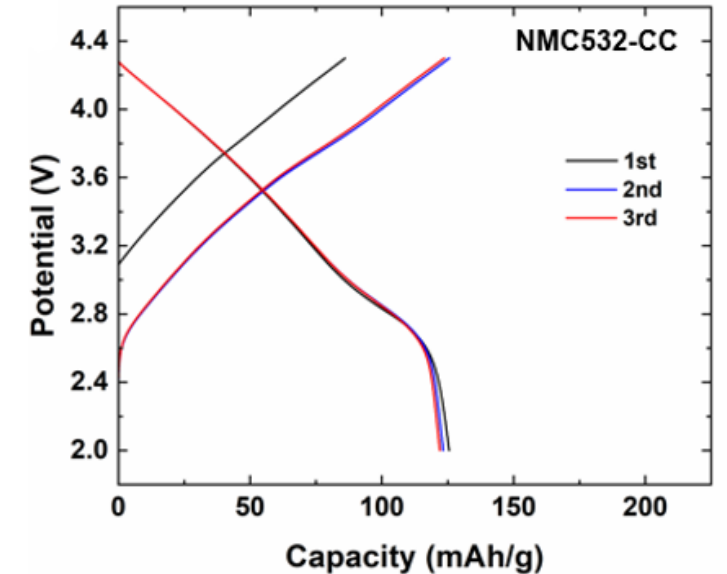
Li-ion battery half cell



Na-ion battery half cell



Li-ion battery full cell (pouch)





# Manufacture of spherical carbon particles by Spray Drying



Coal liquids, a byproduct from mild gasification of coals and coal refuse, are being used as feedstock for manufacturing carbonaceous spheres

Inlet temperature: 170 to 200°C  
Outlet temperature: 70 to 80°C

Control of particle size distribution

# Lower Chinese graphite production could mean higher batte

China has steadily been forcing producers of graphite — a material commonly used in smartphone and laptop batteries — to close in response to rising pollution, as mining companies fail to improve the conditions of local land and water resources.

**Why it matters:** Batteries account for roughly 30% of the global demand for graphite, and China produces 70% of the world's graphite supply.

As demand for the material continues to grow, battery prices may rise, which would impede progress toward the \$100 per kilowatt hour lithium-ion target that many companies, including Tesla and BMW, are looking to achieve.

**Background:** The expanding electric vehicle market has been a major contributor to the global rise in graphite demand.

Axios. March 1, 2019