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# The problem at hand – CO<sub>2</sub> emissions from cement/concrete

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#### Concrete ~ 8% of global CO<sub>2</sub> emissions

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- Most widely used substance after water
- 90% of emissions from production of cement
- No broadly-applicable alternatives
  - 2x the combined volume of steel, plastics, wood, aluminum
- CO<sub>2</sub> curing appears promising, but...
  - Enabling materials have been expensive or scalelimited
  - Traditionally has required concentrated and purified CO<sub>2</sub>
- Large-scale demonstrations needed
  - Technology feasibility needs to be rigorously proven

Carbon Management Research Project Review Meeting, 15





Global concrete market ~ \$ 1 trillion / year

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 Ready mixed concrete has the largest market share

Laboratory for the Chemistry

of Construction Materials

- Lower construction quality, increased safety risks
- Increasing use of prefabricated concrete is a rising industry trend
- Precast concrete
  - Architectural and structural
- Concrete masonry
  - CMUs (i.e., blocks), Bricks, pavers, etc.



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Ready-mix concrete casting on-site



Concrete Masonry Unit (CMU)



Architectural precast concrete façade

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# Accelerated concrete carbonation curing pathways

Carbonation during mixing



- Utilizes 100% CO<sub>2</sub>
- Requires processing to purify CO<sub>2</sub>
- Additional transport required to obtain CO<sub>2</sub>



Carbonation post-forming

- Pressure reactor CO<sub>2</sub> steadily released into reactor to maintain specified pressure
- Utilizes 100% CO<sub>2</sub>



- Flow through reactor uses CO<sub>2</sub> straight from emitter
- Requires gas processing
- Utilizes 2-100% CO<sub>2</sub>









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# Carbonation of primary feedstock – Portlandite (Ca(OH)<sub>2</sub>)

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 Portlandite carbonation is key reaction in concrete carbonation

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 $CO_2$ +Ca(OH)<sub>2</sub>  $\rightarrow$ CaCO<sub>3</sub>+H<sub>2</sub>O

- Carbonation of portlandite particles is near-complete within 24 h
- Reaction kinetics are largely independent of CO<sub>2</sub> concentration for flue gas concentrations (≥ 2 %)
- The Significance: Pressurization, CO<sub>2</sub> enrichment, or significant heating is not required for portlandite carbonation



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 Identification of three most-preferred geometries compatible with the Reversa<sup>™</sup> process

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- Development of a process model that informs the scale-out of the process to produce diverse precast (structural) components
- Modification and validation of existing prototype
- Completion of TEA and LCA to quantify the market viability and lifecycle impact of the Reversa<sup>™</sup> technology



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# **Concrete component selection and strength targets**

- Concrete units were chosen based on technical and market prospects
- Based on combined scoring criteria: concrete masonry units (CMU), segmented retaining walls (SRW), and wet-cast manholes were selected
- Optimized processing and mixture formulation produced for each product exceeded ASTM strength targets
- Portlandite was used as main feedstock and exceeded conversion targets of 0.2 gCO<sub>2</sub>/gCH







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- Process flow diagram was developed to account for gas processing requirements
- Based on the process flow diagram, two Aspen models were developed:
- Steady-state model that assesses
  24-h average CO<sub>2</sub> uptake based on reactor conditions (RH, [CO<sub>2</sub>] and T)
- Dynamic simulations capable of simulating the reactor and the process conditions across the full 24h operating window



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# CO<sub>2</sub> uptake regression conversion models developed

- Regression models were developed to estimate CO<sub>2</sub> uptake across varying carbonation conditions for each product
- Inputs:
  - Temperature (20-to-65 °C)
  - Inlet gas relative humidity (10-to-80%)
  - $CO_2$  concentration (4-to-14 vol.%)
- Model based on maximum conversion of available hydrates.
- Model parameters:

**X** (conversion) =  $x + a[CO_2] - bRH + cT$ 







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- Integrated regression models uses 24-h conversion of concrete component using [CO<sub>2</sub>], T and RH parameters – fitted well to bench-scale data
- A protocol has been developed for changing the Aspen simulation for different concrete components
- Approach allows for scalable design based on concrete inclusion in reactor and inlet gas flow conditions
- This simulation approach is ideal for high volume sensitivity analysis to determine effect of average operating conditions on net CO<sub>2</sub> uptake



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# **Dynamic Aspen simulations**

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- Rate kinetics data included with regression models for time dependent conversion
- Coupled heat mass transfer model from inlet gas conditions to concrete within reactor determines reactor and unit conditions
- Based on calculated reactor conditions conversion of concrete altered
- Estimated CO<sub>2</sub> uptake based on concrete type and reactor conditions is possible



Reactor conditions determine CO<sub>2</sub> uptake from concrete components based on RH, T and [CO<sub>2</sub>]

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# **Presentation takeaways**

- CO<sub>2</sub> mineralization enables an approach to produce construction products while utilizing CO<sub>2</sub> emissions, with strong market potential
  - CO<sub>2</sub> utilization in concrete without a CO<sub>2</sub> capture step
  - Impactful potential due to the large market size of concrete
- The UCLA team has successfully identified suitable concrete products for the Reversa process
- Appropriate mixture designs and operating conditions have been identified for optimum CO<sub>2</sub> uptake for each selected concrete component
- A process model has been developed to assist with scaling system design and assisting design work for system fabrication



## **Future Work**

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 System design underway to incorporate carbonation gas processing requirements

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- Computational fluid dynamics will be employed to ensure equal gas flow distribution across reactor
- System build will aim to process 10-to-30 tonnes of concrete per day
- The process will be deployed at the National Carbon Capture Center (NCCC) in Wilsonville, Alabama in 2023
- Completion of TEA and LCA to quantify quality of system design post-demonstration



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### Thank you for listening

## **Questions??**

**Contact information** 

- Dr Dale Prentice:
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Useful links: Carbon built website: https://www.carbonbuilt.com/

Institute for Carbon Management: https://icm.ucla.edu/ Production of Inorganic Materials – Concrete and Cement

#### Technology Performance Data

	Units	Measured/Current	Projected/Target
		Performance	Performance
Reaction			
Thermodynamics <sup>1,2</sup>			
Chemical Equation	mol <sup>-1</sup>	Ca(OH) <sub>2</sub> +CO <sub>2</sub>	$\rightarrow$ CaCO <sub>3</sub> +H <sub>2</sub> O
ΔH° <sub>rxn</sub>	kJ/mol	-115.102	
ΔG° <sub>rxn</sub>	kJ/mol	-74.953	
Reaction Conditions			
CO <sub>2</sub> Source <sup>3</sup>		Synthetic coal-fired flue gas	Coal-fired flue gas
Pressure	bar	1.01325	1.01325
CO <sub>2</sub> Partial Pressure	bar	0.02-0.14	0.02-0.14
Temperature	C°	25-75	25-75
Nominal Residence Time – batch reactor <sup>4</sup>	h	18-24	18-24
Alkaline Reactant Source <sup>5</sup>		Ca(OH) <sub>2</sub>	Ca(OH) <sub>2</sub>
Process Route <sup>6</sup>	(direct/	Direct	Direct
	indirect)		
Once-Through			
Performance <sup>7</sup>			
CO <sub>2</sub> Conversion <sup>8</sup>	(%)	NA	NA
CO <sub>2</sub> Uptake Potential <sup>9</sup>	(g-CO <sub>2</sub> /g material)	0.59	0.59
CO <sub>2</sub> Uptake Actual <sup>10</sup>	(g-CO <sub>2</sub> /g material)	0.35-0.53	0.2
Product Properties <sup>11</sup>			
Desired Product			
Compressive Strength <sup>12</sup>	(MPa)	20-49	13.8-42
Density	(kg/m <sup>3</sup> )	2000	2000
Product Production	(kg/h)		
Commercial Product		Current	
Properties <sup>13</sup>			
Commercial Product		Concrete Masonry Units, Segmented	
		retaining walls, and concrete manholes	
Compressive Strength <sup>12</sup>	(MPa)	13.8-42	
Density	(kg/m <sup>3</sup> )	2000	
U.S. Market Size	(Tonnes/yr)	500 M	
Global Market Size	(Tonnes/yr)	>10 B	
Market Price	(\$/kg)	0.06-0.40	