

Chromium Tolerant, Highly Active and Stable Electrocatalytic Internal Surface Coating for Cathode of Commercial SOFCs

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Atomic Layer Deposition (ALD) of Solid Oxide Fuel Cells (SOFC)

OVERVIEW

➤ **Project Background & Technical Challenges**

- **Alternative to an interconnect coating: Cr-resistant cathode through infiltration**
- **ALD & technical challenges for the development of an ALD coated Cr-resistant cathode**

➤ **ALD-Enabled, Chromium Tolerant Air Electrode**

LSM/YSZ CATHODE

- **LSM/YSZ: Baseline Performance & nanostructure degradation induced by Cr poisoning**
- **Cell power density & longevity increased by ALD coating**
- **ALD-enabled Significantly increased power density & durability against contamination**

LSCF/SDC CATHODE

- **Baseline nanostructure degradation induced by Cr poisoning**

➤ **Summary**

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Cr-Poisoning SOFC Cathode & Interconnect Coating

Commonly used metallic interconnect materials:

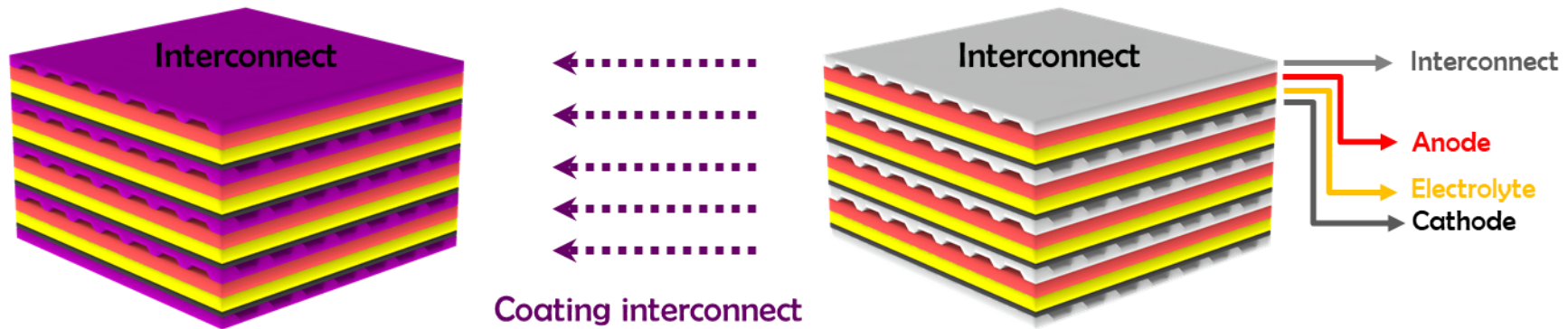
- Ni(Fe)-Cr based heat resistant alloys, Cr-based alloys, and chromia-forming ferric stainless steels.
- Contain Cr to form a protective chromium oxide scale Cr_2O_3 .

Volatilization of Cr species strongly depends on oxygen partial pressure.

Vaporization of Cr species on the SOFC fuel side anode could be neglected.

Cr deposition & poisoning of SOFC cathode:

- Volatile Cr species, such as CrO_3 and $\text{Cr}(\text{OH})_2\text{O}_2$, are generated in oxidizing atmospheres.
- Volatile Cr species react with the cathodes, causing rapid cell degradation.



- Badwal, S.; Deller, R.; Foger, K.; Ramprakash, V.; Zhang, J., Interaction between chromia forming alloy interconnects and air electrode of solid oxide fuel cells. *Solid State Ionics* 1997, 99 (3-4), 297-310.
- Chen, X.; Zhen, Y.; Li, J.; Jiang, S. P., Chromium deposition and poisoning in dry and humidified air at $(\text{LaO}_{0.8}\text{SrO}_{0.2})_{0.9}\text{MnO}_{3+\delta}$ cathodes of solid oxide fuel cells. *International Journal of Hydrogen Energy* 2010, 35 (6), 2477-2485.
- Jiang, S. P.; Chen, X., Chromium deposition and poisoning of cathodes of solid oxide fuel cells—a review. *International Journal of Hydrogen Energy* 2014, 39 (1), 505-531.
- Park, E.; Taniguchi, S.; Daio, T.; Chou, J. T.; Sasaki, K., Influence of cathode polarization on the chromium deposition near the cathode/electrolyte interface of SOFC. *International Journal of Hydrogen Energy* 2014, 39 (3), 1463-1475.

Development of Cr Tolerant Cathode via Infiltration or Coating

For commercial SOFCs with well-developed materials sets, Cathode Cr-resistance can be improved through solution-based infiltration or cathode surface coating.

Structural requirements of Cr-tolerant surface coating layer on SOFC cathode:

- Deeply penetrating into the active layer of the cathode.
- Uniform and conformal on the internal surface of the cathode active layer that possess complex three-dimensional topographies with high aspect ratio, and the TPBs.
- Intimate adhesion and bonding to the cathode surface at atomic scale without spallation.

Coating Film deposition to facilitate the Cr-tolerant cathode through coating:

Magnetron sputtering, sol-gel dip-coating, and electro-deposition techniques used for applying protective coating on ferritic stainless steel interconnects:

- Involving physical vapor deposition or liquid solutions.
- Having the limitation of not deeply penetrating into the cathode active layer or not providing the conformal coating on cathode surface.
- Not ideal for infiltrating and coating the internal surface of porous cathode.

Atomic Layer Deposition: Chemical vapor deposition technique that sequentially applied atomic mono-layers to a substrate, typically alternating compounds to produce a locally balanced atomic distribution of target material.

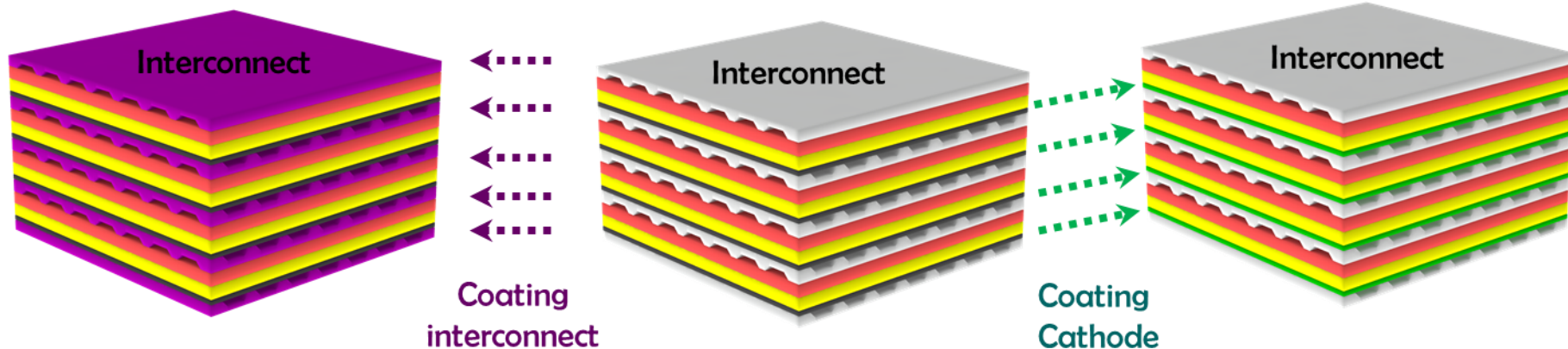
ALD is uniquely suitable for coating uniform and conformal films on complex three-dimensional topographies with high aspect ratio.

Development of Cr Tolerant Cathode Through ALD Coating

For a commercial SOFC with well-developed materials set, Cathode Cr-resistance can be improved through ALD coating.

ALD is uniquely suitable for depositing uniform and conformal films on complex three-dimensional topographies with high aspect ratio.

Two Alternative Approaches to Mitigate Cr Poisoning



Approach I
Interconnect Coating

Approach II
ALD Coating of Cathode

Technical challenges for ALD coating of Cr resistance layers

Structure requirement of Cr-tolerant surface coating could be satisfied by ALD:

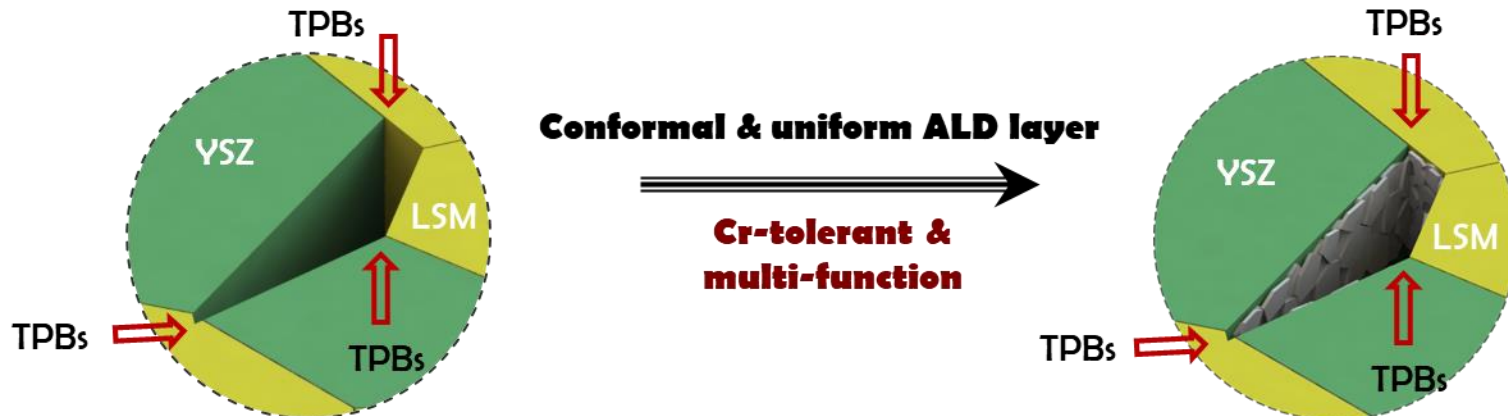
- ✓ Deeply penetrating into the active layer of the cathode.
- ✓ Uniform and conformal on the internal surface of the cathode active layer, that possess complex three-dimensional topographies with high aspect ratio, and the TPBs.
- ✓ Intimate adhesion and bonding to the cathode surface at atomic scale without spallation.

Functional requirement of Cr-tolerant surface coating layer on SOFC cathode:

- Cr-inert ALD coating layer should not impair the electrochemical reactions taking place on the cathode surface.

Ideal Cr-inert ALD coating layer should possess the multiple functions of:

- Conformal on internal surfaces of backbones, prevent Sr out diffusion and Cr inward diffusion.
- Highly active towards electrochemical reactions.
- Super stable upon the long term electrochemical operation at high temperatures.



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LSM/YSZ CATHODE

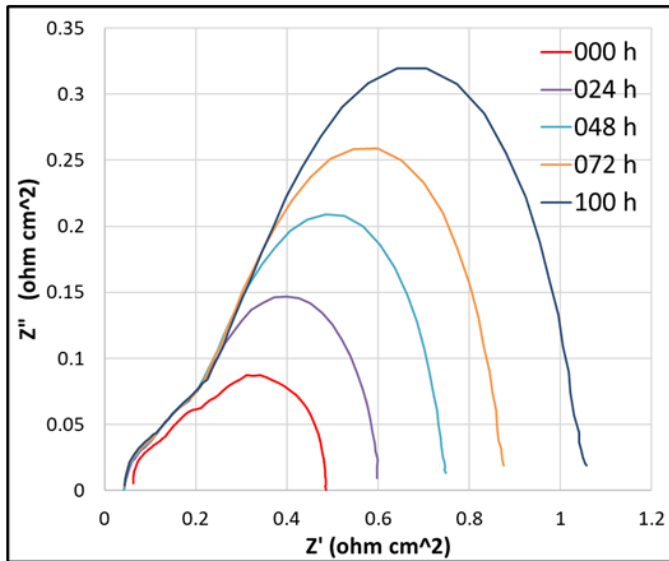
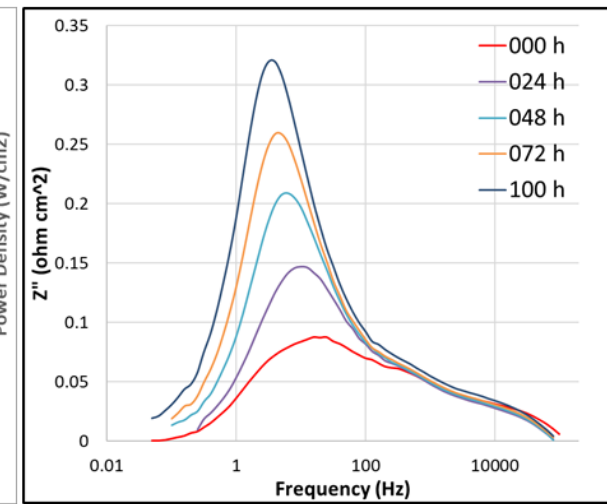
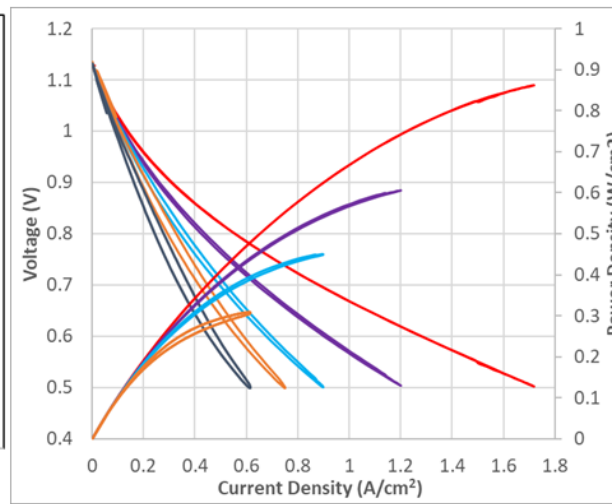
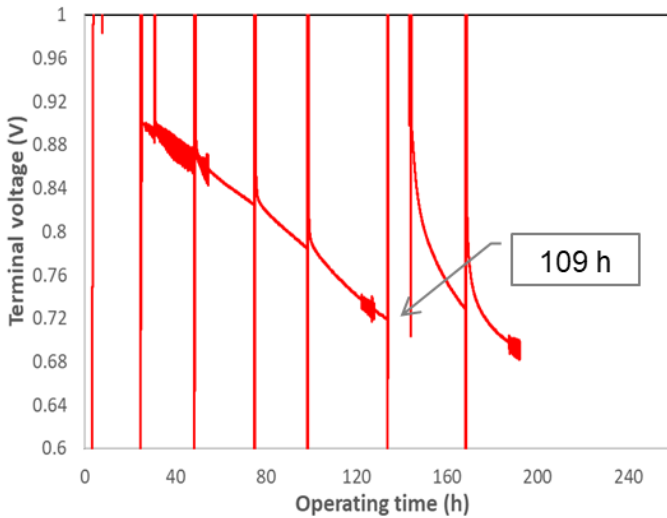
- **LSM/YSZ: Baseline Performance & nanostructure degradation induced by Cr poisoning**
- Cell power density & longevity increase introduced by ALD coating
- ALD-enabled significant increased power density & durability against contamination

LSCF/SDC CATHODE

- Baseline nanostructure degradation induced by Cr poisoning

➤ Summary

Severe Degradation of LSM/SSZ Baseline Cell Upon Cr Contamination



Operating time (h)	Peak power density (W/cm^2)	Percentage degradation (%)
0	0.862	-
24	0.605	29.8 %
50	0.45	47.8 %
74	0.375	56.5 %
109	0.307	64.4 %

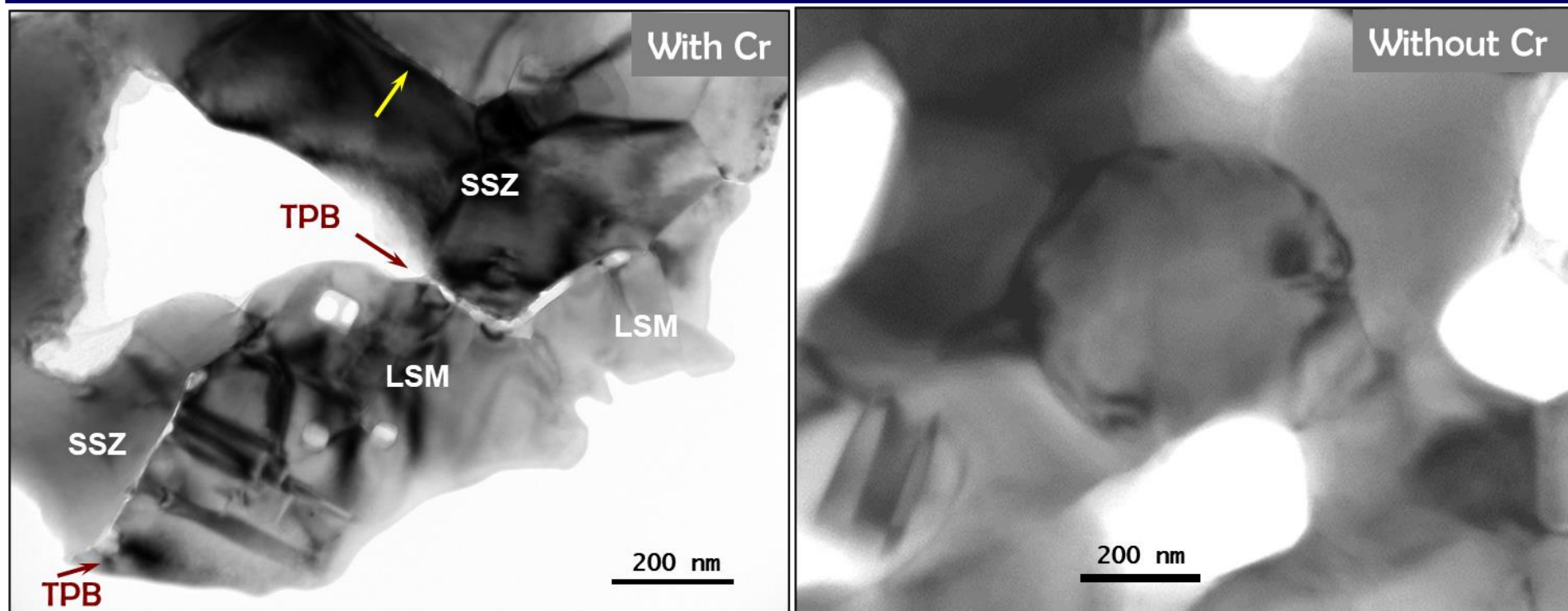
100 h operation, the terminal voltage dropped ~ 190 mV.

At $0.3\ A/cm^2$: Degradation rate is $1.68\ V$ per $1000\ h$

Severe Degradation of LSM-based Baseline Cell

- Little change of series resistance R_s .
- Large increase of the polarization resistance R_p .
- R_p increase could be related to activity loss of oxygen reduction reaction at TPB.

As-Received LSM Baseline Cell Cathode: Typical Nanostructure



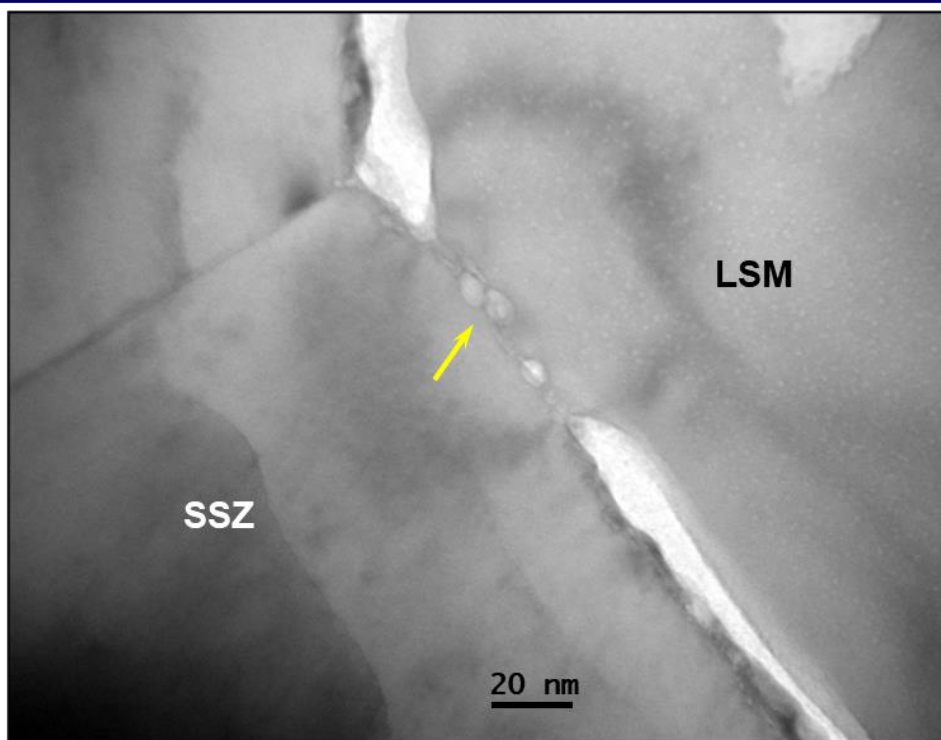
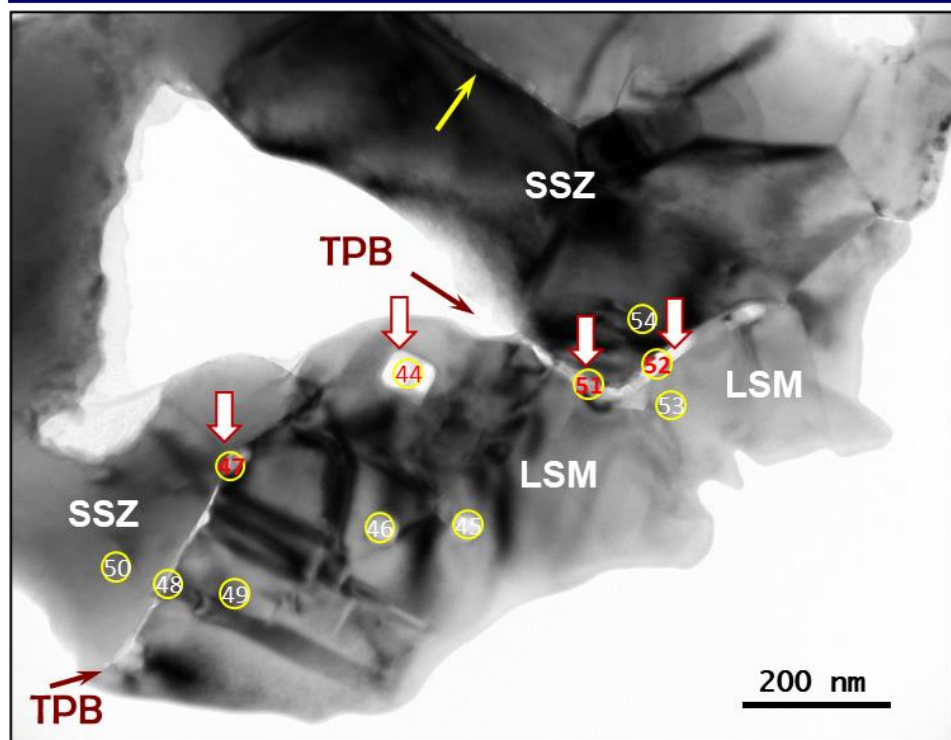
Baseline **without** Cr contamination.

- Intact clean LSM/YSZ interface.
- No formation of the nano-scale MnO_x, SrO_x, or La₂O₃ phases.

Baseline **with** Cr contamination: Original TPB and LSM/SSZ Interface Degradation.

- Internal cracking initiated from the original triple phase boundary (TPB) and propagated along the internal interface between Sc-stabilized ZrO₂ (SSZ) and LSM.
- In the local SSZ/LSM interface region that there is no apparent internal cracking, there are internal nano-pores that are elongated along the SSZ/LSM interface.
- Intragranular structure has spherical nano-pores, while the SSZ grains remained to be intact.

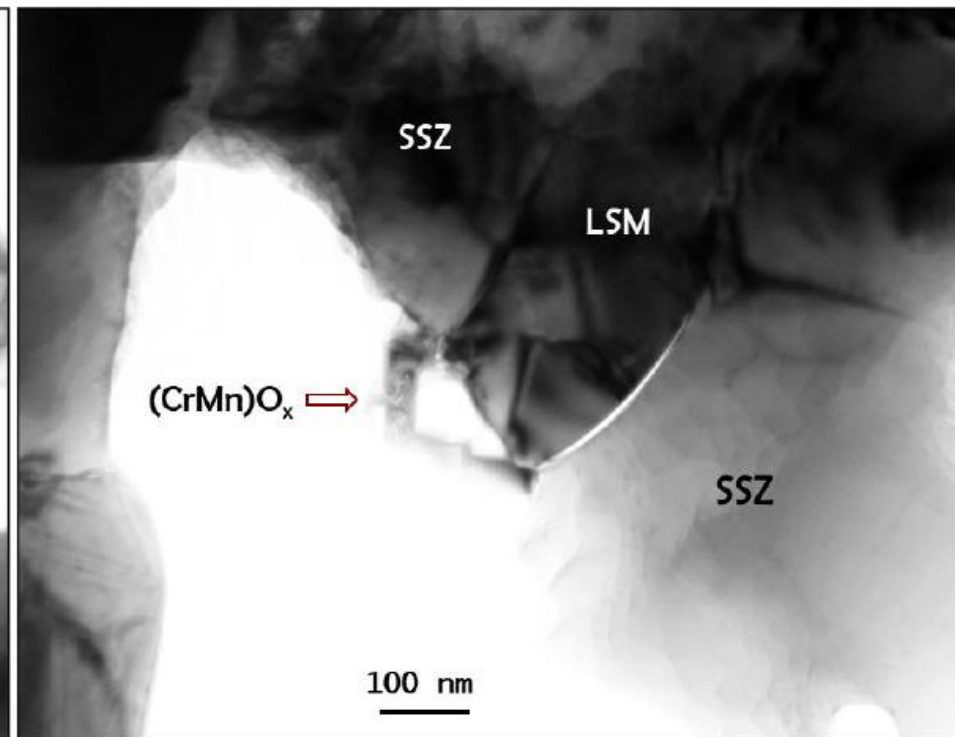
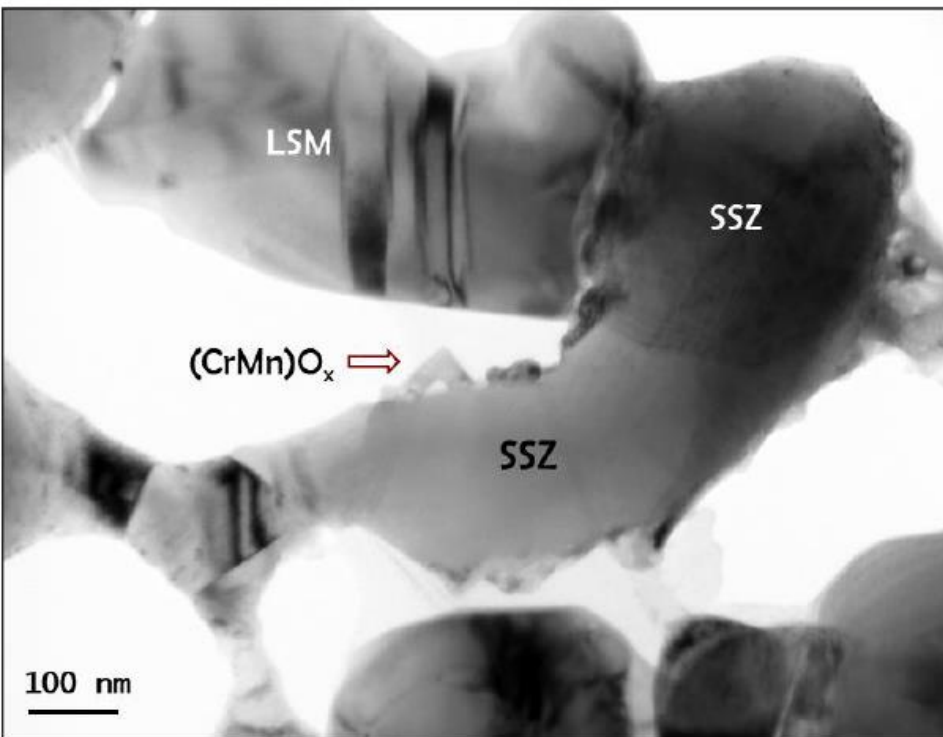
Cr Contamination: Nanostructure Degradation of LSM/SSZ Baseline Cell



Atomic%	O	La	Sr	Mn	Zr	Sc	Cr	Formulation
44	66.37	12	3.1	15.14			3.39	$\text{La}_{0.79}\text{Sr}_{0.21}\text{MnCr}_{0.22}\text{O}_x$
45	71.89	10.37	2.53	14.76			0.46	$\text{La}_{0.8}\text{Sr}_{0.2}\text{Mn}_{1.14}\text{Cr}_{0.04}\text{O}_x$
46	68.01	11.99	3.23	16.65			0.12	$\text{LaSrMn}_{0.97}\text{Cr}_{1.06}\text{O}_x$
47	73.06	7.35		7.15	3.85	0.79	7.8	$\text{LaSrMn}_{0.97}\text{Cr}_{1.06}\text{O}_x + \text{ZrO}_2$ (w 10 % Sc_2O_3)
48	69.81	2.76		3.45	19.29	3.48	1.21	$\text{LaSrMn}_{1.25}\text{Cr}_{0.44}\text{O}_x + \text{ZrO}_2$ (w 9 % Sc_2O_3)
49	71.34	10.66	2.69	15.31				$\text{La}_8\text{Sr}_2\text{Mn}_{15}\text{O}_x$
50	63.35			0.82	29.84	6		ZrO_2 (w 9 % Sc_2O_3) + $\text{Mn}_{0.03}$
51	70.96	2.7		4.14	7.43	1.75	13.04	CrOx enriched
52	76.77	2.73		2.94	12.46		5.11	CrOx enriched
53	73.19	9.96	2.44	12.78			1.63	$\text{La}_{0.8}\text{Sr}_{0.2}\text{Mn}_{1.03}\text{Cr}_{1.13}\text{O}_x$
54	66.96			0.96	27.55	4.54		

Doped ZrO_2 and LSM interface also show nano-pores.

Cr Contamination: Nanostructure Degradation of LSM/SSZ Baseline Cell



Original TPB Degradation & Formation of $(\text{CrMn})\text{O}_x$ on SSZ grain surfaces

- Additional $(\text{CrMn})\text{O}_x$ enriched crystals appear to nucleated at the original TPBs.
- Additional $(\text{CrMn})\text{O}_x$ enriched crystals appear to grown on SSZ grain surfaces.

Applying an ALD coating layer on the internal surface of cathode: Ideal Cr-inert should possess the multi-functionality of:

- Fully conformal to prevent Sr outward diffusion and prevent Cr inward diffusion.
- Highly active towards electrochemical reactions.
- Super stable upon the electrochemical operation at elevated temperatures.

Choice of Cr tolerance ALD Layer: $\text{Mn}_{0.8}\text{Co}_{0.2}\text{O}_x$ is ordinary interconnect coating materials, excellent layer to **prevent Cr penetration.**

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LSM/YSZ CATHODE

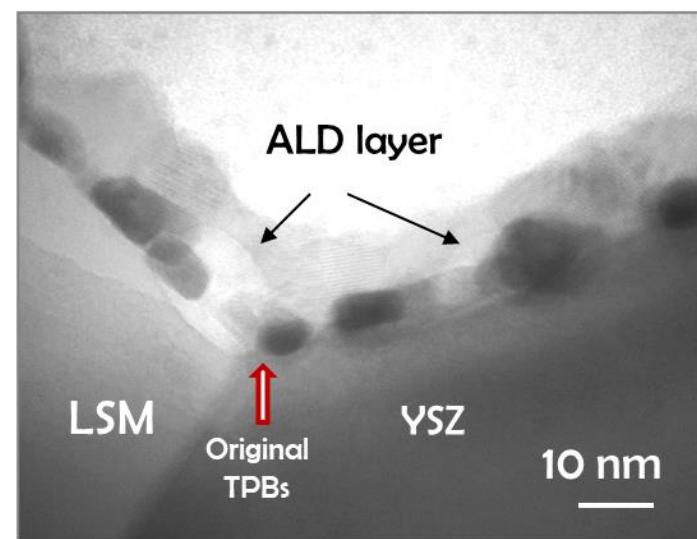
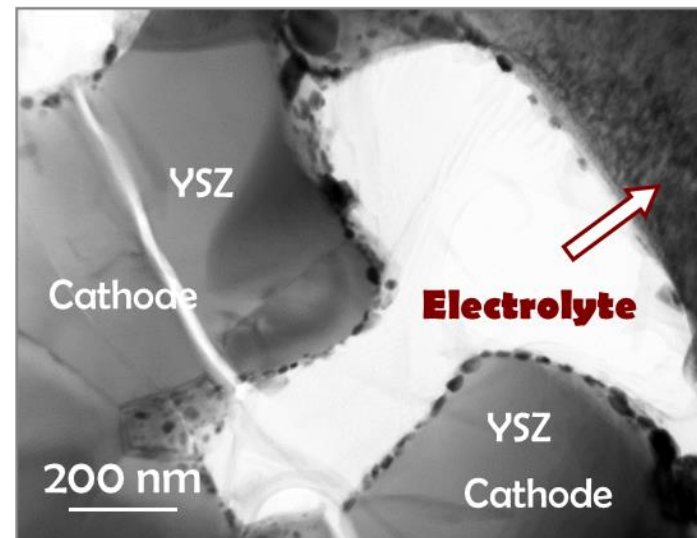
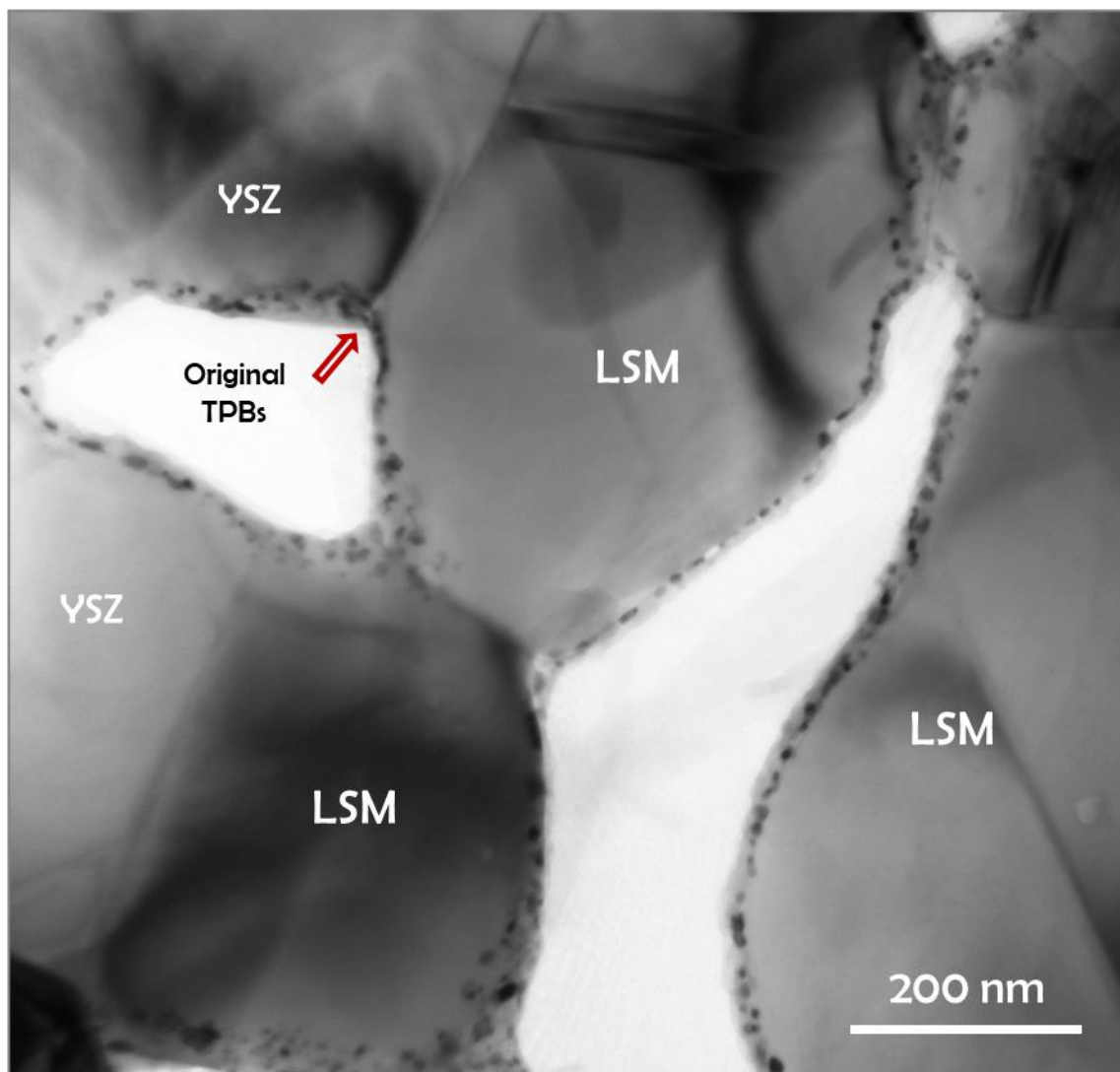
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LSCF/SDC CATHODE

- Baseline nanostructure degradation induced by Cr poisoning

➤ Summary

Structure Requirement for ALD Layer: **Fully Conformal and Uniform**

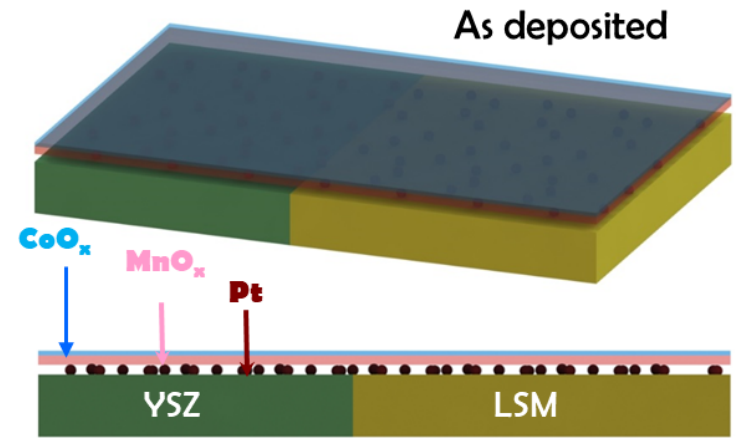
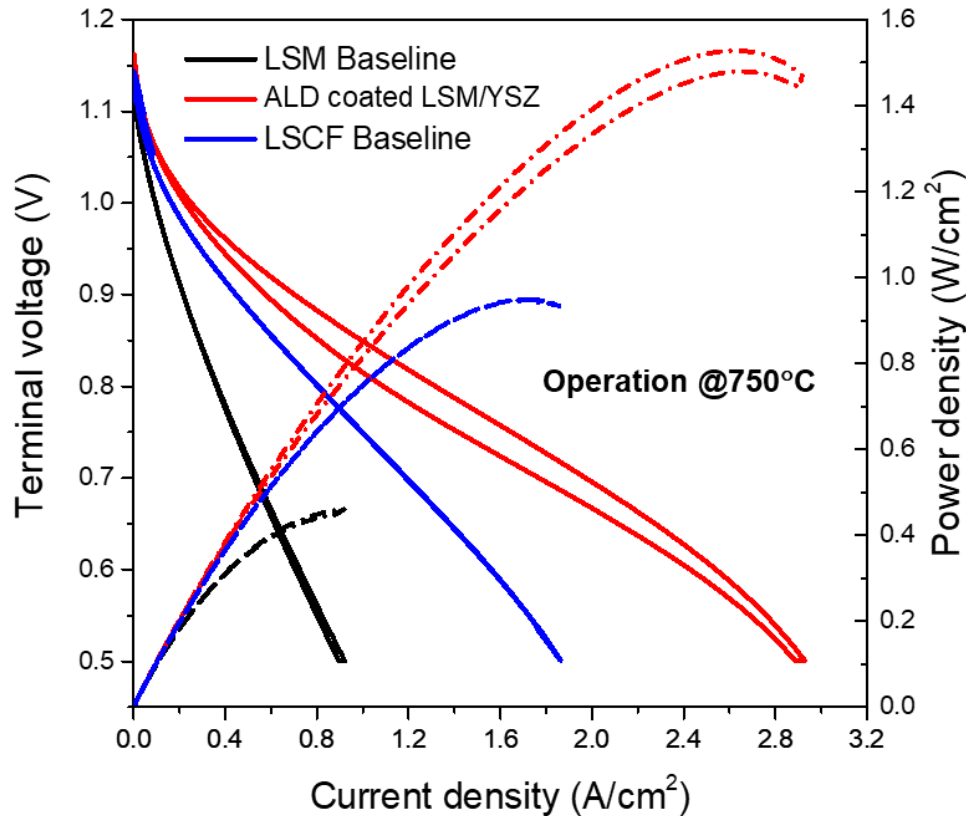


After operation at 750C for 120 hours:

- Stable nanostructure at ALD layer are at **active layer** and at the **cathode-electrolyte interface**.
- Conformal & uniform; ALD layer is with nano-pores for gas penetration; Covering the original TPBs.

Structural Requirement for ALD Layer: Enhancement Cell Performance

Large 380% power density enhancement of commercial SOFCs



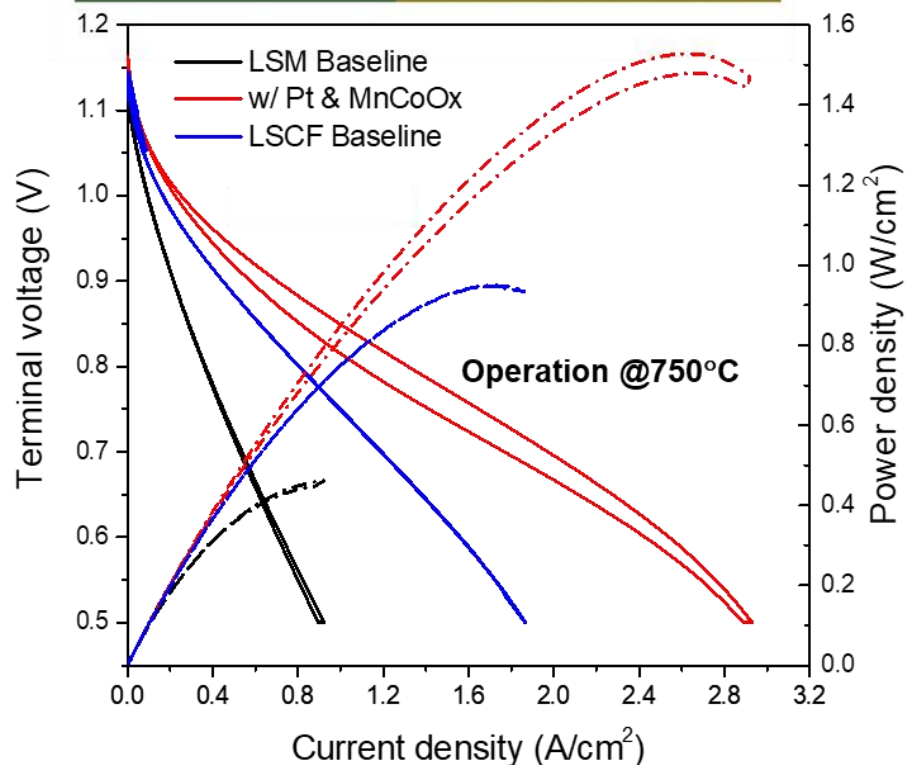
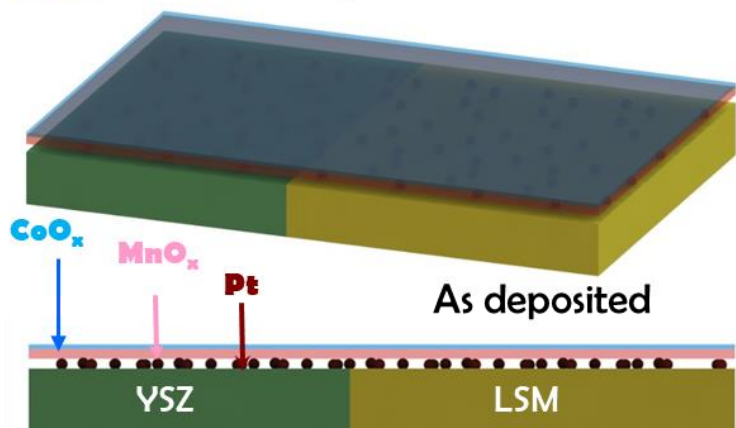
ALD coated LSM/YSZ cell:

- power density is **380%** of that from the LSM/YSZ baseline at 750°C.
- power density is **170%** of that from the LSCF/SDC baseline at 750°C.

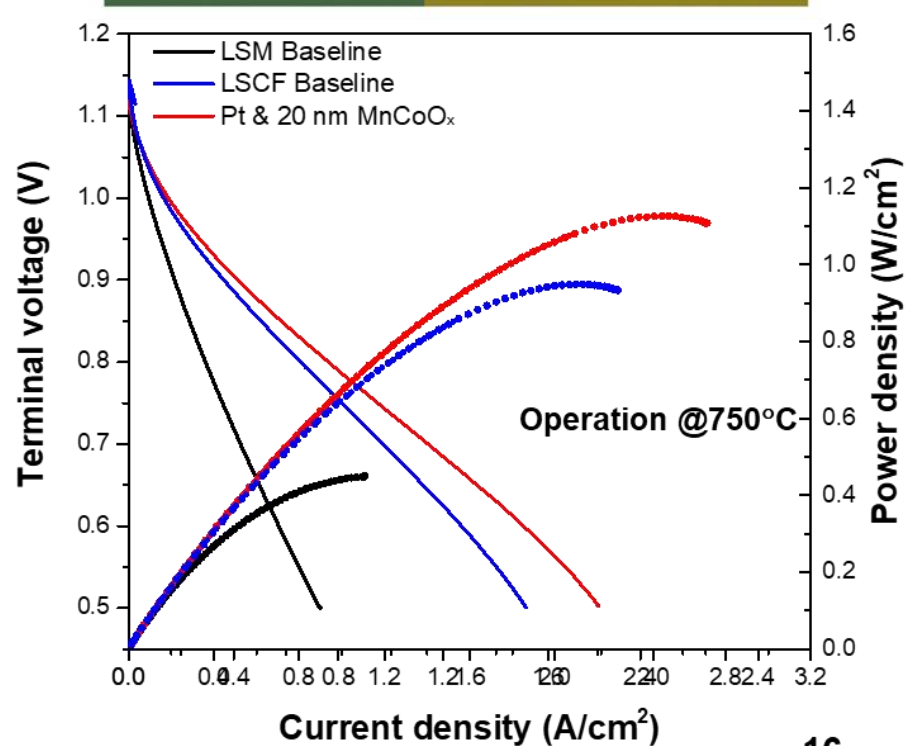
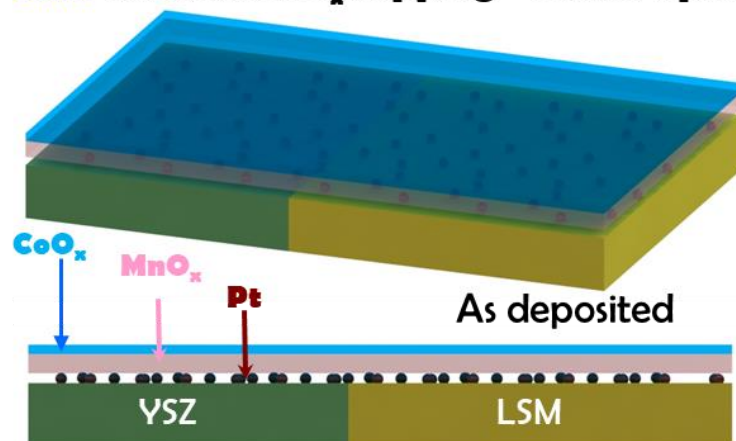
Surface architecture	R_{total} (Ω/cm^2)	R_s (Ω/cm^2)	R_p (Ω/cm^2)	i at 0.8 V (A/cm^2)	P at 0.8 V (W/cm^2)	Factor enhancement to LSM/YSZ	Factor enhancement to LSCF/SDC
LSM/YSZ Baseline	0.619	0.045	0.574	0.350	0.28	1	/
ALD LSM/YSZ	0.324	0.051	0.273	1.320	1.06	380%	170%
LSCF/SDC Baseline	0.350	0.110	0.239	0.790	0.63	/	1

Tuning the Cell Performance by Adjusting ALD Layer Thickness (no Cr)

10 nm thick MnCoO_x capping ~ 3 nm Pt particles

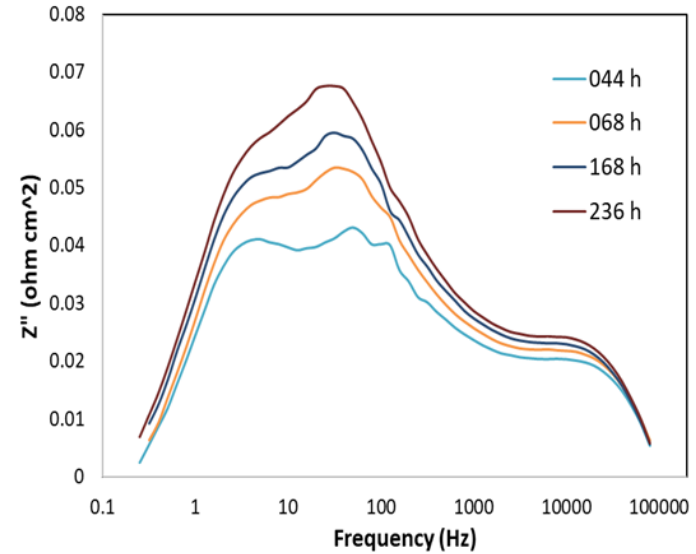
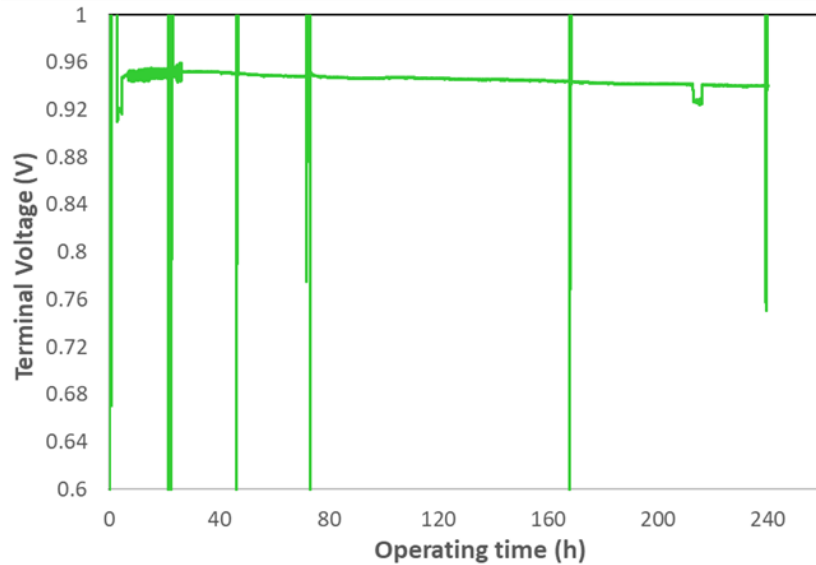


20 nm thick MnCoO_x capping ~ 3 nm Pt particles



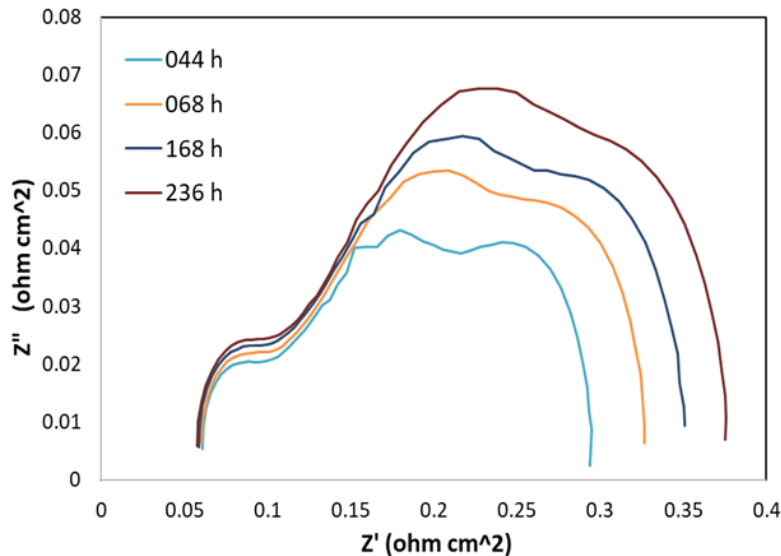
10 nm ALD layer, peak power density is 380% of baseline. 20 nm ALD layer, peak power density is 280% of baseline.

Stable Performance of ALD Coated Cell Upon Operation With Cr



ALD coated LSM based cell at 0.3 A/cm²
Degradation rate is 0.054 V per 1000 h

R_s and R_p (taken at 0.3 A/cm²) changes

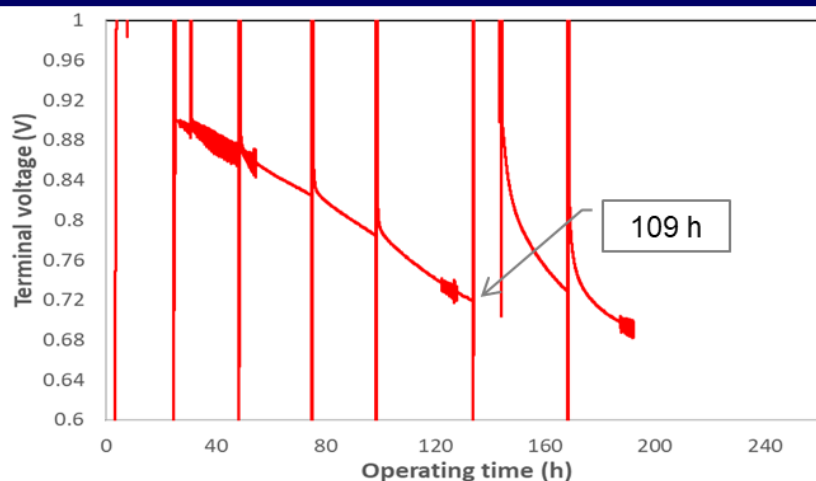


Operating time (h)	R_s (Ω cm ²)	R_p (Ω cm ²)
44	0.06	0.304
68	0.058	0.328
168	0.058	0.354
236	0.055	0.381

LSM/YSZ with ALD coating of 20 nm Mn_{0.8}Co_{0.2}O_x layer, operation with Cr contamination.

- Zero hour: peak power density is 280 % of baseline @ zero hour;
- 168 hours: peak power density is 200 % of baseline @ zero hour.

Comparison of Baseline & ALD Coated Cells Upon Cr Contamination

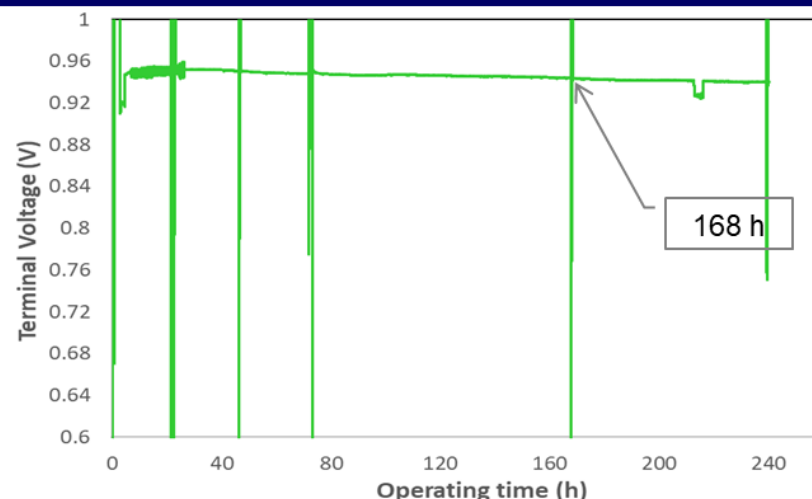


LSM based Baseline Cell:

baseline at 0.3 A/cm²

Degradation rate is 1.68 V per 1000 h

Operating time (h)	Peak power density (W/cm ²)	Percentage degradation (%)
0	0.862	-
24	0.605	29.8 %
50	0.45	47.8 %
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ALD Coated Cell LSM Based Cells

ALD coated cell at 0.3 A/cm²

Degradation rate is 0.054 V per 1000 h

Operating time (h)	Peak power density (W/cm ²)	Percentage degradation (%)
23	1.254	-
44	1.127	0.101276
68	1.073	0.144338
168	1.015	0.19059

Baseline Cell: Peak power density remained to be **36%** of baseline @ zero hour after 109 hours operation.

ALD coated one: Peak power density remained to be **200%** of uncoated baseline @ zero hour after 168 hours operation. **ALD coated cell power density is ~ 600%** of that baseline cell upon operation with Cr contamination.

Cr tolerance: Large performance enhancement (> 200% power density) by ALD coating of Cr-tolerant Mn_{0.8}Co_{0.2}O_x and remained to be stable even operation with Cr contamination.

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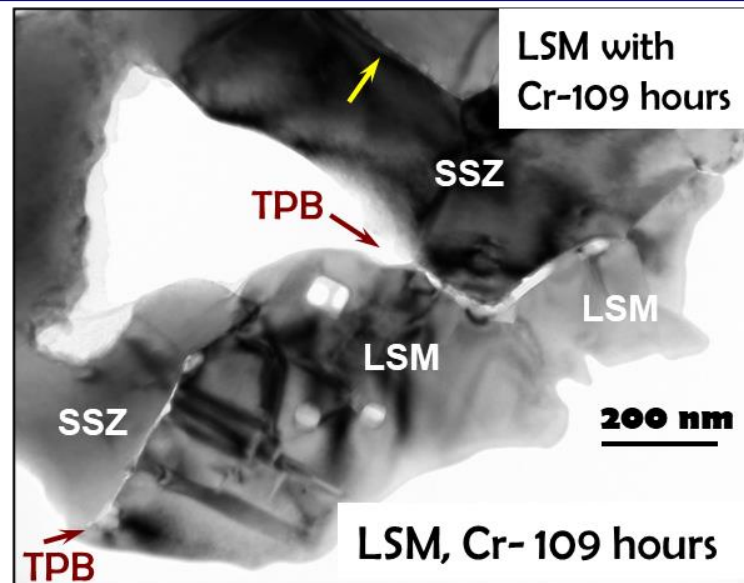
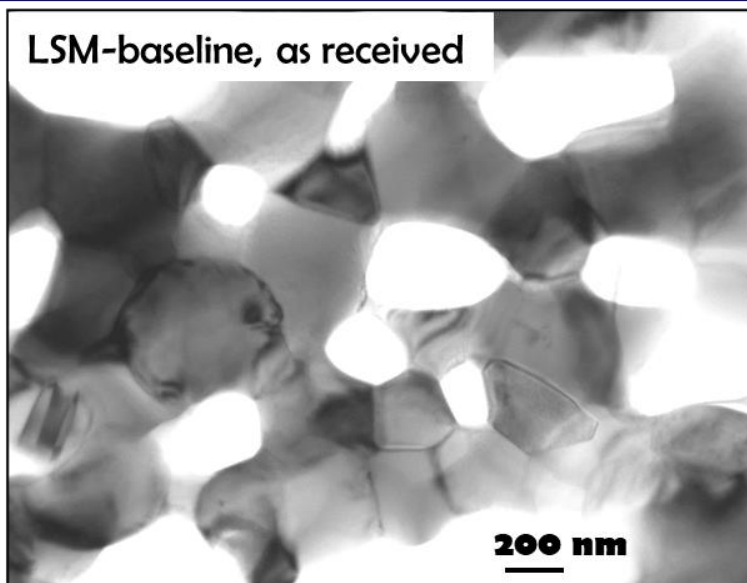
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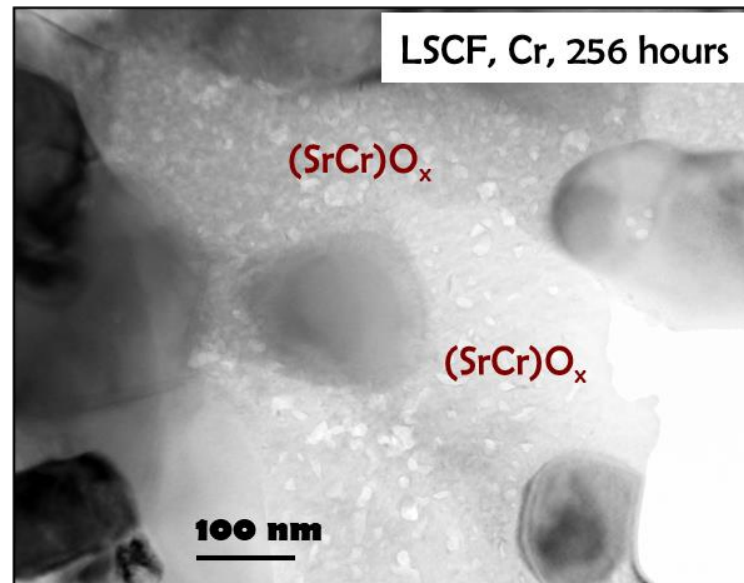
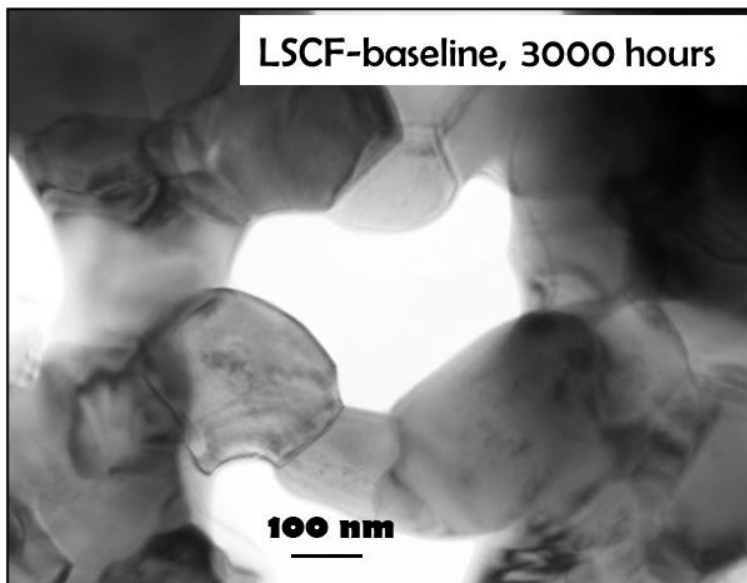
- **Baseline nanostructure degradation induced by Cr poisoning**

➤ Summary

Cr contamination: Different Degradation between LSM and LSCF based Cells

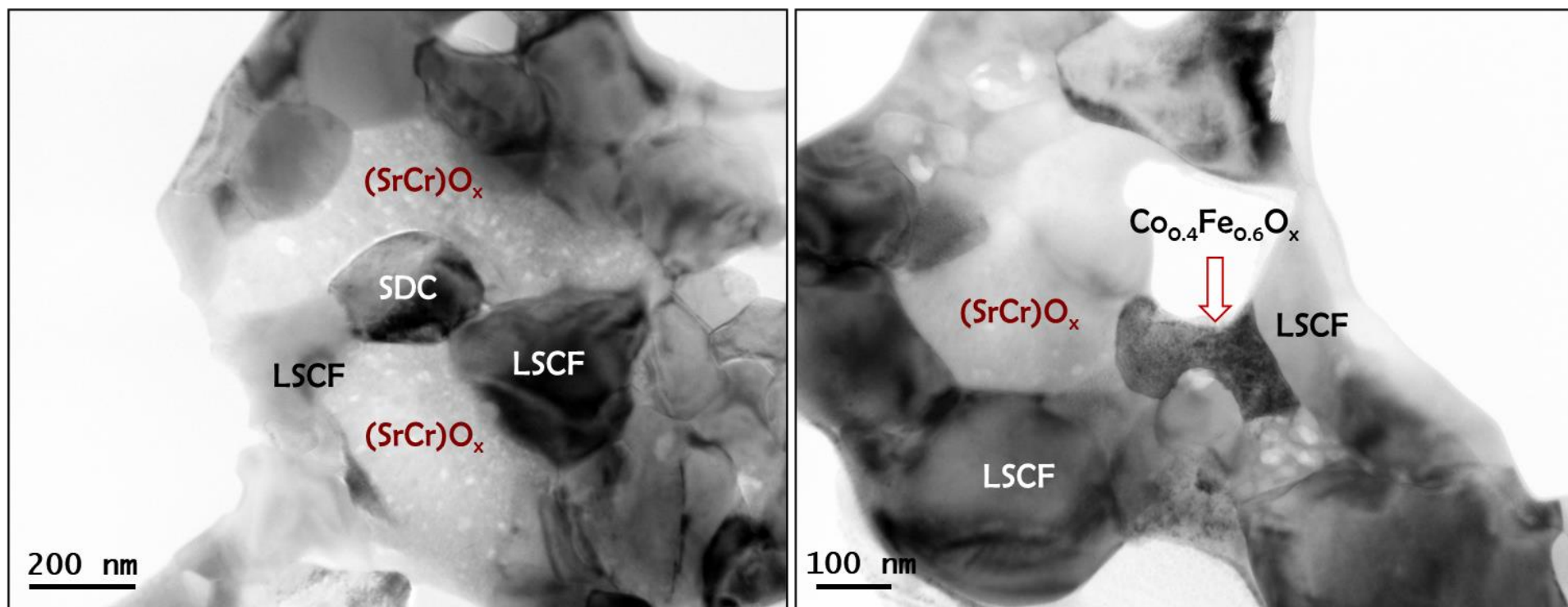


LSM: Cr is attacking the TPB. No accumulation at the original pore region.



Accelerated Sr segregation induced by Cr contamination

Cr contamination: Different Degradation Mechanisms from LSM & LSCF Cells



Accelerated Sr segregation induced by Cr contamination

LSCF-SDC cathode nanostructure degradation caused by **Cr-contamination**.

- $(\text{SrCr})\text{O}_x$ are filling the original pores of the LSCF/SDC after 256 hours operation at 750°C .
- Formation of $(\text{FeCo})\text{O}_x$ nano-grains at LSCF/LSCF surface grain boundaries.

ALD layer on LSCF/SDC backbone ideally needs to be conformal to prevent the Sr evaporation/Segregation and their further interaction with Cr vapors.

SUMMARY

LSM/YSZ or LSM/SSZ based cells:

- **LSM/SSZ baseline performance & nanostructure degradation by Cr**
 - **Peak power density loss of 64% after 109 hours operation. Dramatic increase in R_p .**
 - **Cracking at LSM/SSZ interface, LSM grains. SSZ remains intact, but with $(CrMn)O_x$.**
- **ALD coating $(MnCo)O_x/Pt$ dramatically improves the Cr resistance**
 - **Power density is 280-380% of the baseline cell depending on the ALD layer thickness.**
 - **For cell with a 20 nm thick ALD layer, there is a large performance enhancement (> 200% power density) induced by ALD coating of Cr-tolerant $Mn_{0.8}Co_{0.2}O_x$.**
 - **For cell with 20 nm thick ALD layer, after 168 hours at 750°C power density of ALD-coated cell is ~ 600% of that baseline cell upon operation with Cr contamination for 109 hours.**

LSCF/SDC cells:

- **LSCF/SDC baseline performance & nanostructure degradation by Cr**
 - **With the Cr source, there is no apparent Sr surface segregation phase even for the baseline cell operated for 3000 hours at 750°C.**
 - **With the Cr source, there is significant amorphous $(SrCr)O_x$ phase accumulated in the original pore region.**
- ❖ **There is completely different nanostructure degradation mechanisms between LSM and LSCF cells induced by Cr contamination.**
 - ❖ **LSCF based cells ideally need the conformal ALD layer to prevent Sr migration and Cr contamination.**

ACKNOWLEDGEMENTS

DOE-SOFC Program

National Energy Technology Laboratory

Program Manager: Dr. Joseph Stoffa

Program Manager: Dr. Debalina Dasgupta

Program Manager: Sarah Michalik

Technology Manager: Dr. Shailesh Vora