

# Fog+Froth-based Post-combustion CO<sub>2</sub> Capture In Fossil-fuel Power Plants

DE-FE0031733

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<https://caer.uky.edu/power-generation/>

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U.S. Department of Energy

National Energy Technology Laboratory

Carbon Management and Natural Gas & Oil Research Project Review Meeting

Virtual Meetings August 2 through August 31, 2021

# Project Overview

**Title:** Fog+Froth-based Post-combustion CO<sub>2</sub> Capture In Fossil-fuel Power Plants

**Award Number:** DE-FE0031733

	Performance Period	Federal Share	Cost Share	Total
BP1	5/1/2019-1/31/2021	\$1,462,428	\$371,077	\$1,833,505
BP2	2/1/2021-4/30/2022	\$1,484,976	\$366,946	\$1,851,922
Total	5/1/2019-4/30/2022	\$2,947,404	\$738,023	\$3,685,427

**Overall Goal:** Reduce CO<sub>2</sub> capture capital cost by reducing the absorber size

**Project**

**Participants:**



Center for Applied  
Energy Research



**DOE-NETL Team:** Carl Laird, Project Manager; Patrick Mayle, Contract Specialist; and Angela Harshman, Contract Officer

# Outline

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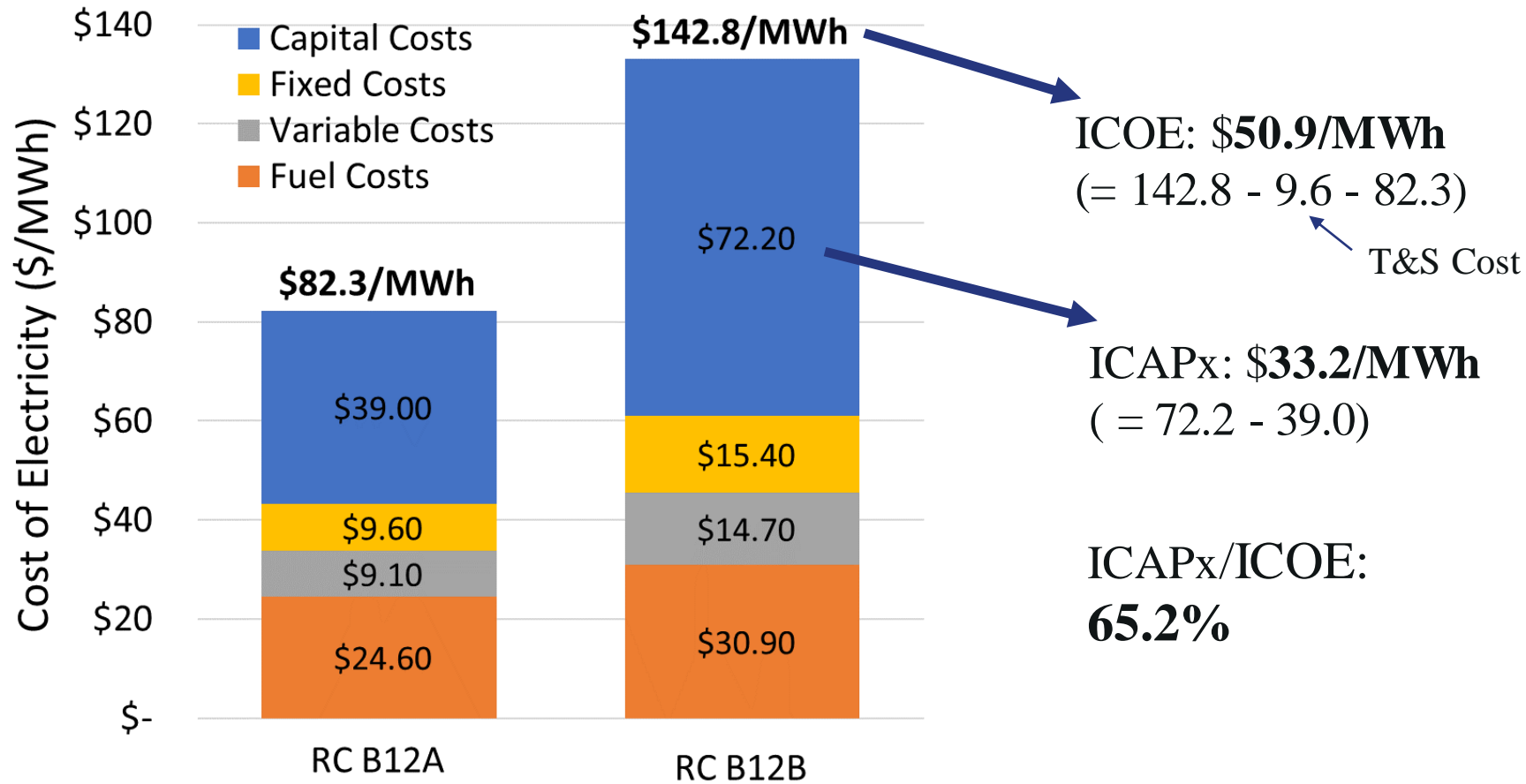
# Executive Summary

- **All BP 1 Deliverables Met and Success Criteria Achieved**
- **On Track to Meet BP 2 Deliverables and Success Criteria**
- Fog and froth sections fabricated and commissioned, 8/30/2019
- Fogging and frothing sections connected with existing 0.1 MWth facility, batch and continuous operations conducted by 1/31/2020
- Parametric Campaign complete on 2/19/2021 after 827 hours of testing with bottled gas
- 60+% CO<sub>2</sub> capture and 0.45 mol C/mol N rich loading demonstrated with 3 ft Fog-n-Froth and 2 ft structured packing, on par with traditional absorber performance with >2X of structured packing
- Long-term Campaign began on 3/3/2021 with ~350 hours of testing with fossil fuel-derived flue gas
- State Point Data Table completed on 4/26/2021 for the UK CAER hindered amine blend solvent



# Technology Background

## CAPEX and OPEX (DOE Baseline)



Cost and Performance Baseline for Fossil Energy Plants Volume 1: Bituminous Coal and Natural Gas to Electricity Revision 3, (DOE/NETL-2015/1723), United States Department of Energy (DOE), National Energy Technology Laboratory (NETL), Pittsburgh, Pennsylvania, July 2015.

# Technology Background

## CCS Direct Costs

<b>Equipment</b>	<b>45%</b>
Columns	50%
Heat Exchangers	25%
Pumps	10%
Instrumentation and Control	5%
Balance of Plant	10%
<b>Civil</b>	20%
<b>Installation</b>	20%
<b>Engineering Fee</b>	10%
<b>Insurance and Others</b>	5%
<b>100%</b>	

# Technology Background

Use the WFGD example to compare and contrast with CCS.

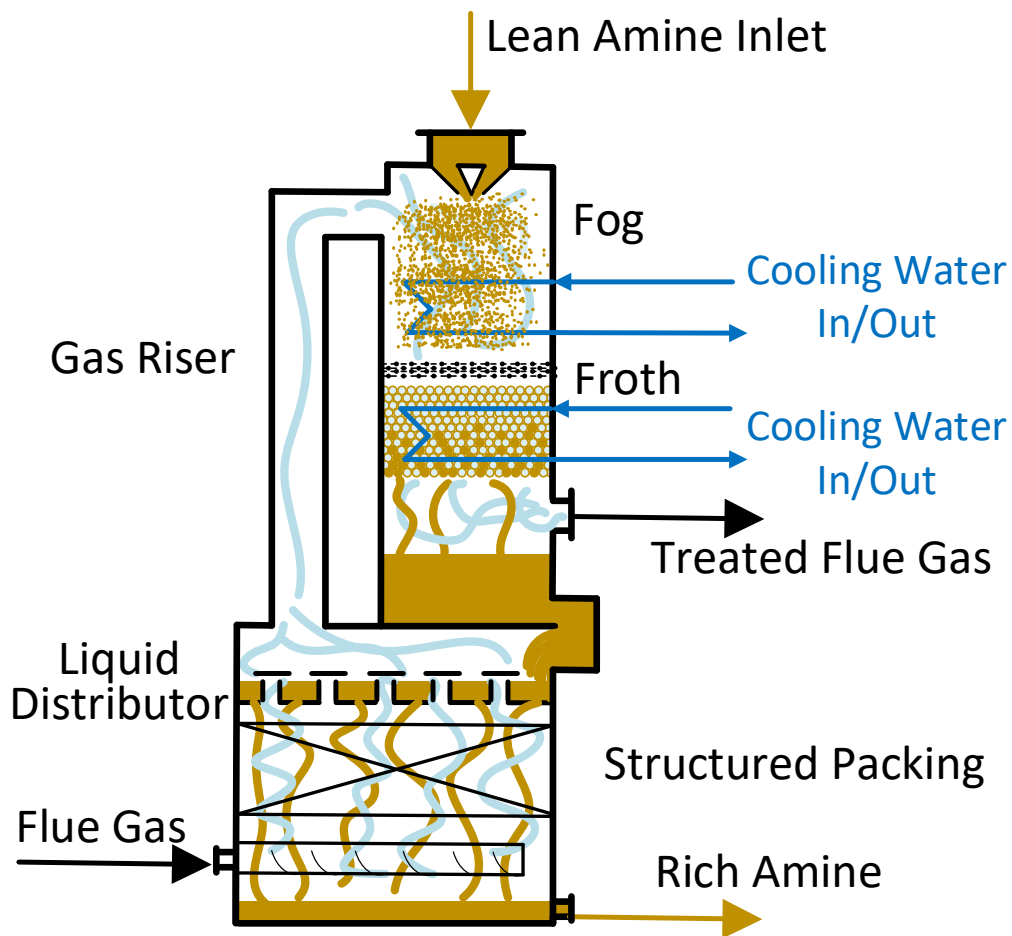
Aspects	Unit	WFGD	CCS
L/G	-	10	3.5
Effective Wet Surface per Unit of Volume	m <sup>2</sup> /m <sup>3</sup>	1523.8	200.0
Reactant Concentration	mol/m <sup>3</sup>	0.03	1262.5
Diffusivity	m <sup>2</sup> /s	1.2E-05	2.0E-09
Reaction Kinetics, K <sub>2</sub>	M <sup>-2</sup>	10000	3000
Inlet Concentration	vol %	0.3	12.5
Outlet Concentration	vol %	0.01	1.6
Log Mean Driving Force	kPa	0.07	4.33

$$k'_g = \frac{\sqrt{D_{CO_2} K_2 [Am]}}{H_{CO_2}}$$

$$flux = A \cdot k_G (P^g_{CO_2} - P^*_{CO_2})$$

Assuming Pseudo-first Order	WFGD	Traditional CCS	Compact Absorber
Henry's Law Constant	1	1	
Reaction Kinetics in Absorber	1	0.5	
k <sub>g</sub> '	1	2.0	5.6
Effective Surface Area	1	0.13	0.39
Driving Force	1	62.34	
<b>Overall Mass Transfer per Volume</b>	<b>1</b>	<b>16.6</b>	<b>137</b>
Removal Loading @ 90% CO <sub>2</sub> and 99% SO <sub>2</sub> captured	1	37.88	
<b>The Absorber Size Required</b>	<b>1</b>	<b>2.28</b>	<b>0.28</b>

# Compact Absorber Configuration

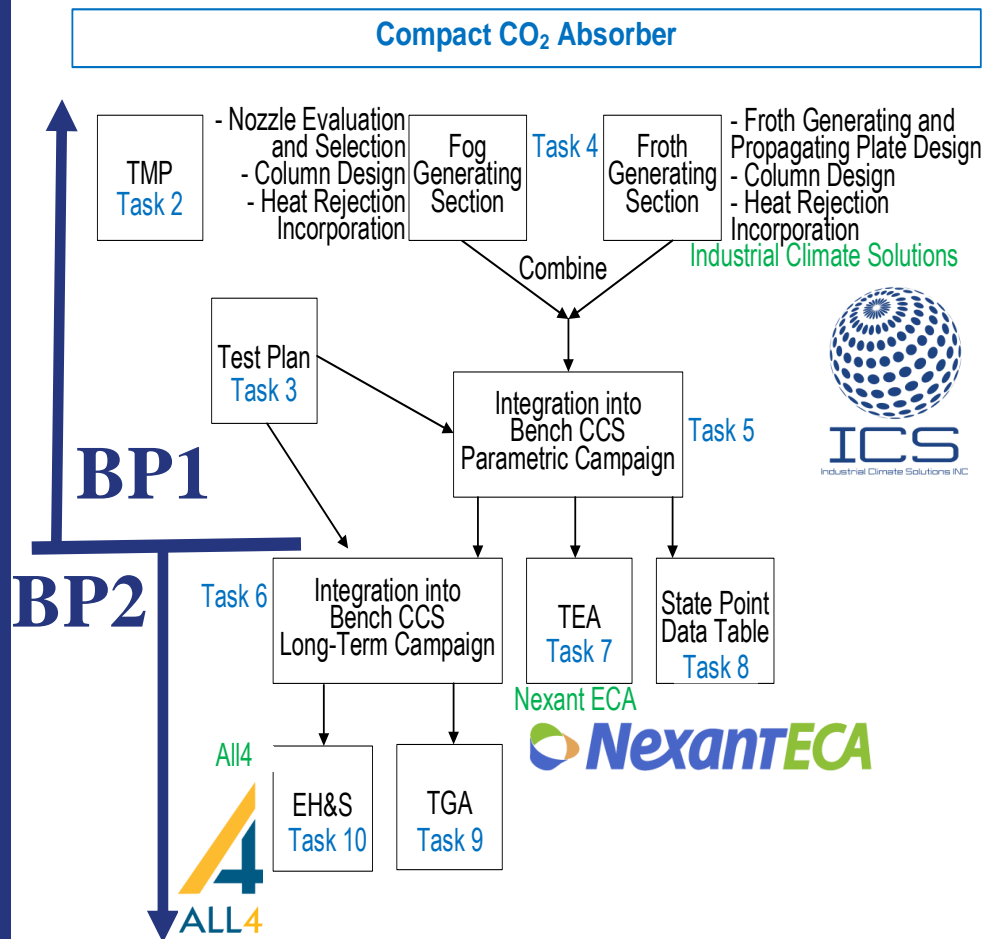




# Technical Tasks and Project Scope



**Reduce CO<sub>2</sub> Capture  
Capital and Operating Costs  
by Reducing the  
Absorber Size**



## • Design, Fabricate and Research a Compact Absorber

- Atomizing Nozzle Selection
- Froth Plate Design
- In-Situ Heat Rejection
- UK CAER's bench post-combustion CO<sub>2</sub> capture facilities

## • Evaluation

- Parametric Campaign
- Long Term Campaign
- TEA
- EH&S
- State Point Data Table

## • Develop

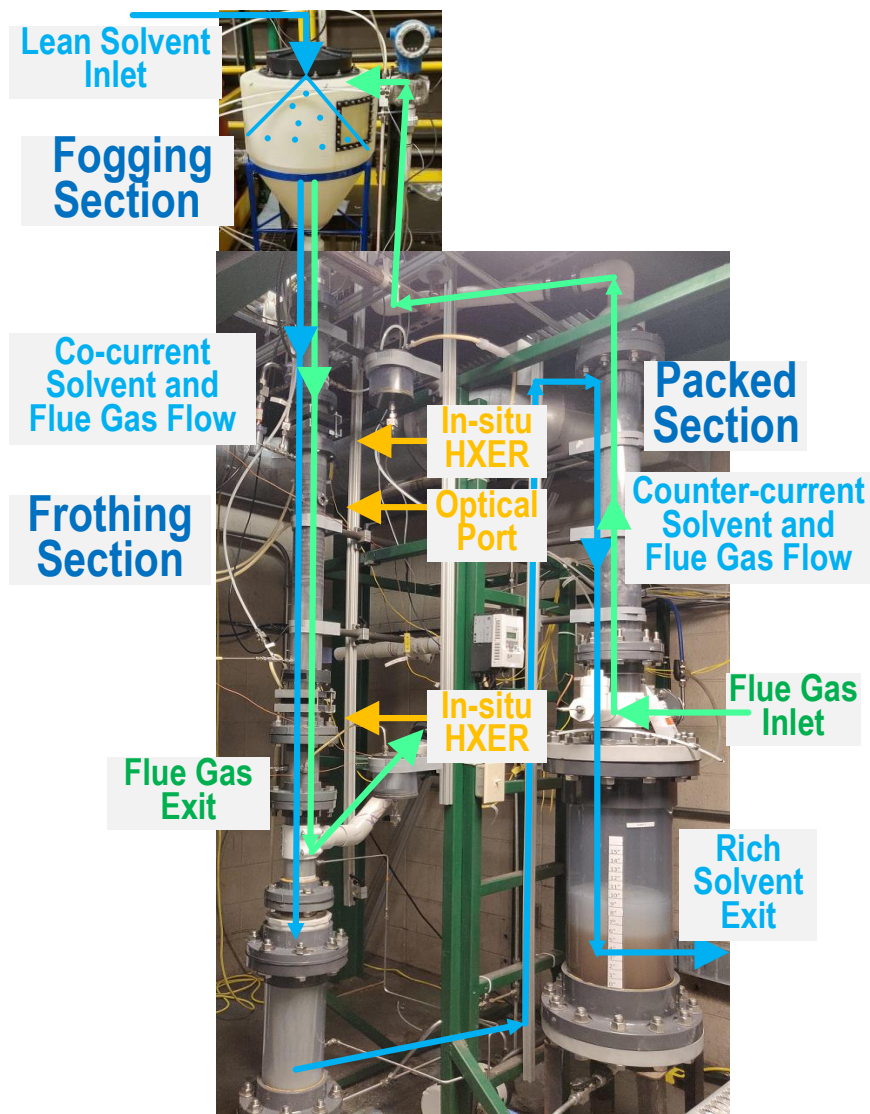
- TMP

# Milestones and Success Criteria

Milestones	Completion Date
1. PMP Updated	5/31/2019
2. Project Kickoff Meeting	5/14/2019
3. TMP Updated	7/25/2019
4. Test Plans Completed	9/15/2019
5. Fog Section of Unit Constructed and Tested	8/2/2019
6. Froth Section of Unit Constructed and Tested	1/31/2020
7. Compact Absorber Integrated into Bench Process	6/30/2020
8. Parametric Test Campaign Complete	1/31/2021
9. Long-term Test Campaign Complete	
10. TEA Complete	
11. State Point Data Table Updated	3/31/2021
12. TGA Complete	
13. EH&S Assessment Complete	

Success Criteria
1. Atomizing nozzles compared, selected and tested ✓ 2. Froth plates compared, selected and tested ✓ 3. Functioning fogging+frothing-based compact absorber with liquid/gas contact area increased by at least 5 times over structured packing ✓ 4. Mass transfer enhancement by at least 4 times ✓ 5. Fog droplet size of 10-50 $\mu\text{m}$ ✓ 6. Froth bubble size of 3-5 mm with liquid film thickness of <10 $\mu\text{m}$ ✓ 7. Open section of hybrid absorber captures 60-70% of the $\text{CO}_2$ and packed section captures 20-30% of the $\text{CO}_2$ ✓
8. Long term verification of fogging+frothing-based compact absorber functionality with solvent degradation, based on ~1000 run hours on the UK CAER bench CCS with at least the same baseline capture efficiency and regeneration energy 9. TEA shows the following: A) capital cost savings of $\geq 10\%$ and cost of $\text{CO}_2$ capture reduction of $\geq 15\%$ compared to DOE RC B12B, B) an absorber column that is ~70% shorter for the same $\text{CO}_2$ removal duty with ~50% electricity savings for the flue gas booster fan due to the shorter column and packing height, C) when the UKy-CAER advanced solvent is used (with a heat of desorption ~20% less than 30 wt% MEA), a specific reboiler duty (energy consumption) of 900 Btu/lb (2.1 GJ/tonne) $\text{CO}_2$ captured can be reached by reducing the primary stripper exhaust $\text{H}_2\text{O}/\text{CO}_2$ ratio to 0.25, and D) ~50% reduction in the CCS capital cost 10. EH&S assessment shows no impediment to technology development

# Bench-scale Unit

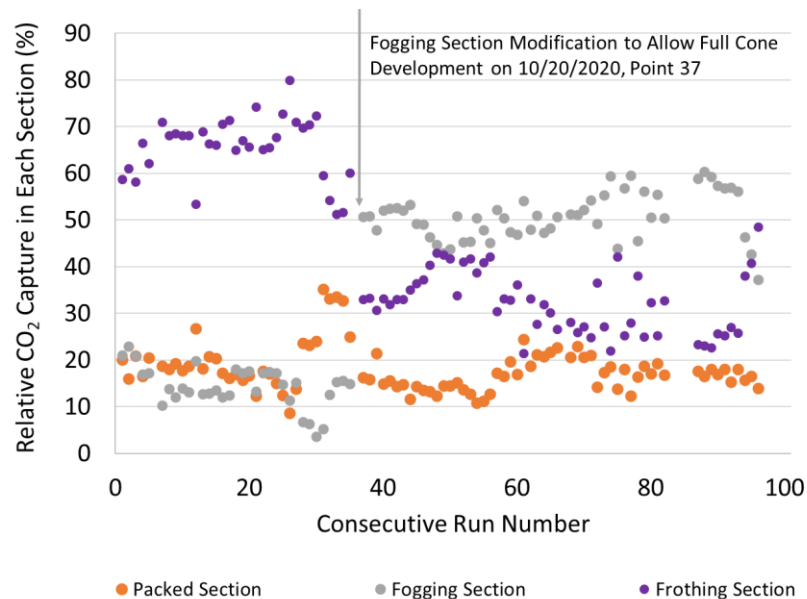
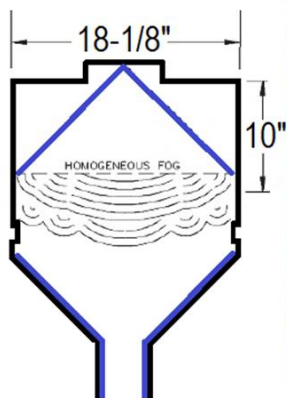
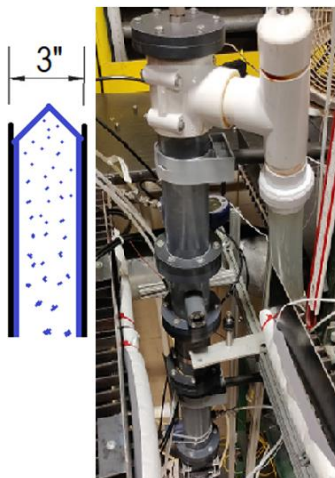
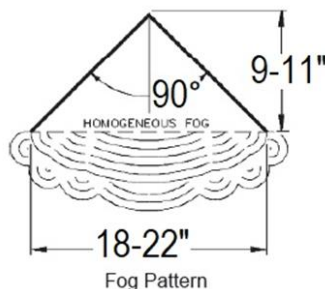


# Progress: The Impact of Fog Formation

Demonstration  
of Fog in 3" ID  
Column



Full Cone  
Development  
Fogging  
Section



# Progress: Fog Droplet Size

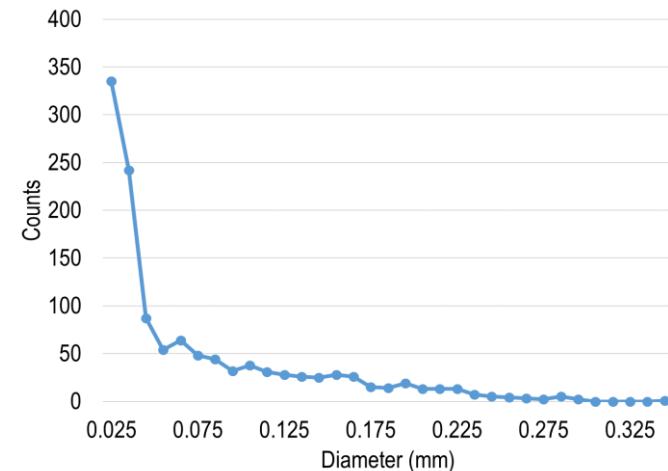
Targeting droplets of  $<100\ \mu\text{m}$  to increase surface area by 3.7X

The BETE® PJ32 hydraulic misting nozzle initially selected and evaluated as the best commercially available option for evaluation.

Subsequently, the PJ24 and PJ20 nozzles have been incorporated.

Comparison of BETE® PJ32, PJ24 and PJ20 Nozzles with Water.			
	<b>BETE® PJ32</b>	<b>BETE® PJ24</b>	<b>BETE® PJ20</b>
Reference Back Pressure (psi)	200	200	200
Spray Angle (degrees)	90	90	90
Spray Pattern	Full Cone of Homogeneous Fog	Full Cone of Homogeneous Fog	Full Cone of Homogeneous Fog
Published Flow Rate per nozzle (gpm)	0.40	0.22	0.15
Published Orifice Diameter (in.)	0.032	0.024	0.020
Published Coverage (in.)	22	16	12
Published Height (in.)	11	8	6
Published Sauter Mean Diameter ( $\mu\text{m}$ )	61	46	41

Evaluation of Water Droplets Produced from BETE® PJ32 Nozzle



# Progress: Fog Droplet Size

The fog droplet Sauter mean diameter (SMD) was estimated by using an equation published by the nozzle manufacturer, BETE. According to this equation, when the compact absorber nozzle pressure exceeds 175 psig, the fog droplet SMD is  $\leq 50 \mu\text{m}$ .

$$D_2 = D_1 \left( \frac{P_2}{200} \right)^{-0.3} (V_2)^{0.2} \left( \frac{ST_2}{73} \right)^{0.5} (SG_2)^{0.3}$$

Where:

$D_2$  is the calculated droplet SMD

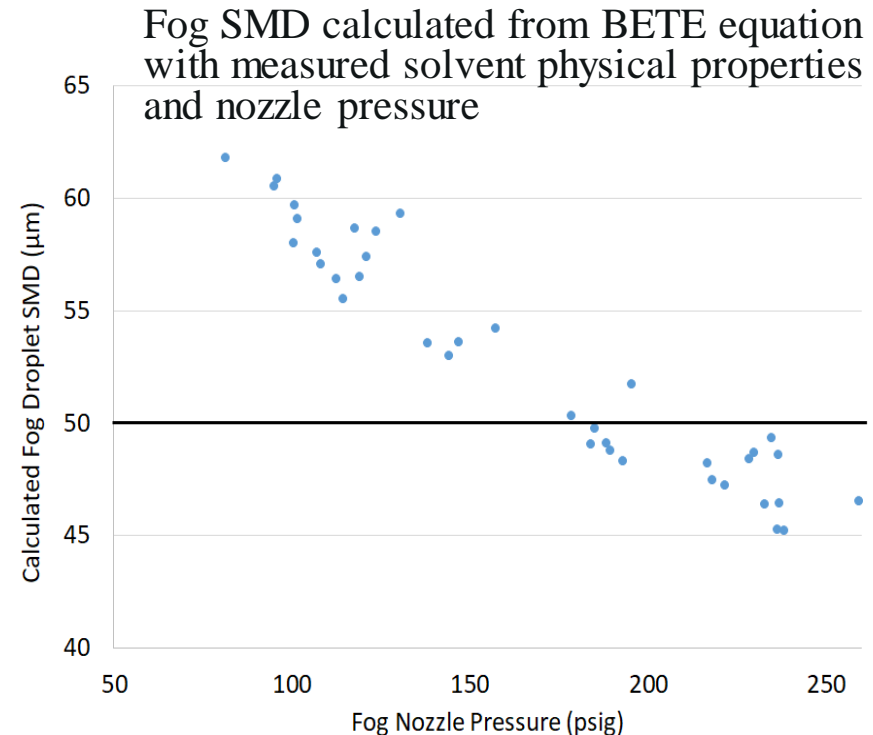
$D_1$  is the BETE measured SMD of water at 200 psi nozzle pressure

$P_2$  is the experimental nozzle pressure in psig

$V_2$  is the experimental solution viscosity in cP

$ST_2$  is the experimental solution surface tension in Dynes/cm at 20 °C

$SG_2$  is the experimental solution specific gravity





# Progress: Froth Bubble Size

Target Froth bubble size of 3-5 mm with liquid film thickness of <10 μm

Based on fogging section mass transfer the bubble size is <3.7 mm.

$$\text{Mass Transfer Rate} = A \cdot k_G (P_{CO_2}^g - P_{CO_2}^*)$$

$$k'_g = \frac{\sqrt{D_{CO_2} K_2 [Am]}}{H_{CO_2}}$$

Specific Absorption in Each Section of the Compact Absorber Parametric Campaign Compared with that in a Traditional Absorber and Corresponding Fog Droplet and Froth Bubble Size.<sup>a</sup>

	<b>Specific Absorption (mol C/hr·ft<sup>3</sup>)</b>	<b>Improvement Over Traditional Packed Absorber</b>	<b>Droplet Size (μm) and Bubble Size (mm) Based on Measured Mass Transfer</b>
<b>Compact Absorber Fogging Section</b>	69-266	2.3-4.6 times	82-165 μm
<b>Compact Absorber Frothing Section</b>	189-782	6.5-13.5 times	<3.7 mm
<b>Compact Absorber Packed Section</b>	22-102		
<b>Traditional Packed Absorber</b>	29-58		

# Progress: Increased Liquid-Gas Surface Area

Target increased by at least 5 times over structured packing.

The surface area is increased by 6.8 times that of 250Y structured packing.

## Fogging Section

Expected Liquid-Gas Contact Surface Area at Target Operation of L/G=3.5 mass/mass									
Uniform Droplet Size ( $\mu\text{m}$ )	10	20	30	40	50	75	100	200	400
Liquid-Gas Contact Surface Area ( $\text{m}^2/\text{m}^3$ )	9356	4678	3119	2339	1871	1247	936	458	234
Improvement over 250Y Structured Packing	37.4X	18.7X	12.5X	9.4X	7.5X	5.0X	3.7X	1.9X	0.9X

## Frothing Section

Expected Liquid-Gas Contact Surface Area at Target Operation of L/G=3.5 mass/mass				
Uniform Bubble Size (mm)	3	4	5	8
Liquid-Gas Contact Surface Area ( $\text{m}^2/\text{m}^3$ )	2000	1500	1200	750
Improvement over 250Y Structured Packing	8X	6X	3X	3X

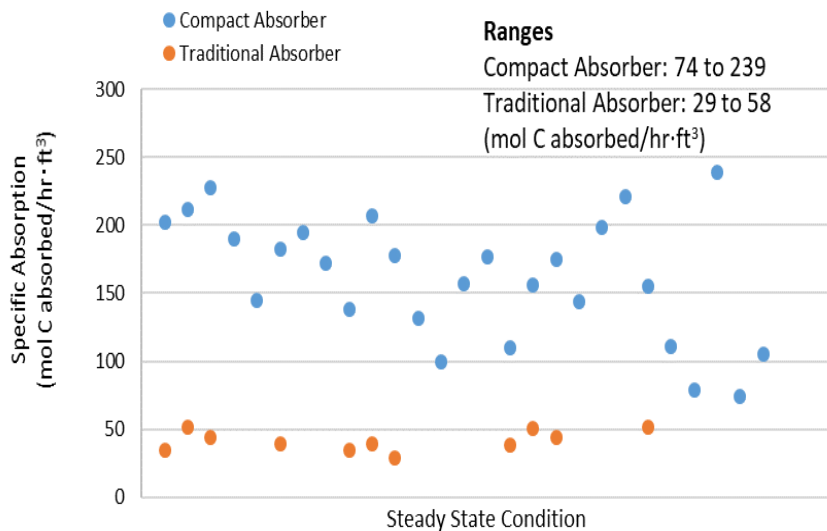


# Progress: Increased Mass Transfer

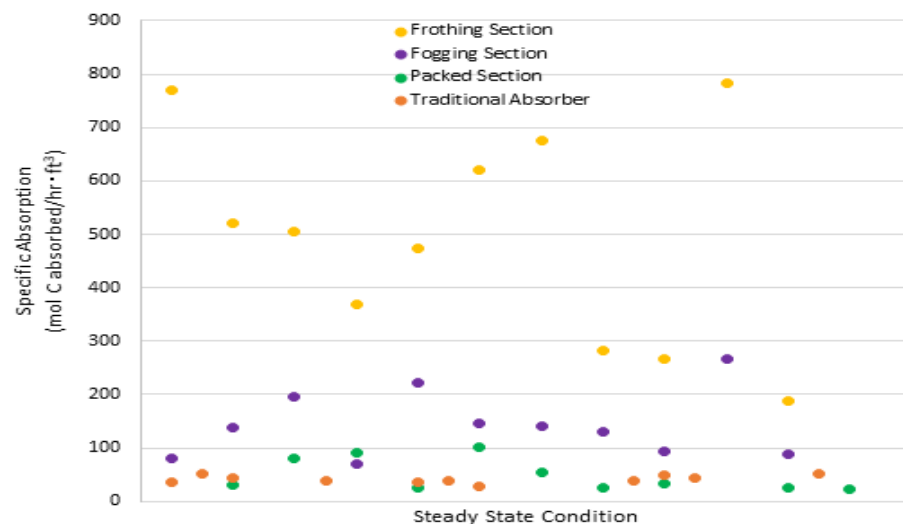
Target Mass Transfer Enhancement by 4X.

Comparing the UK CAER compact absorber performance with a previous campaign using a traditional absorber (structured packing) with the same solvent and varying operating conditions, the specific absorption in the compact absorber is notably higher than in the traditional absorber, 2.6 to 4.1X, depending on the operating parameters.

Specific Absorption in the Compact Absorber Parametric Campaign Compared with that in a Traditional Absorber

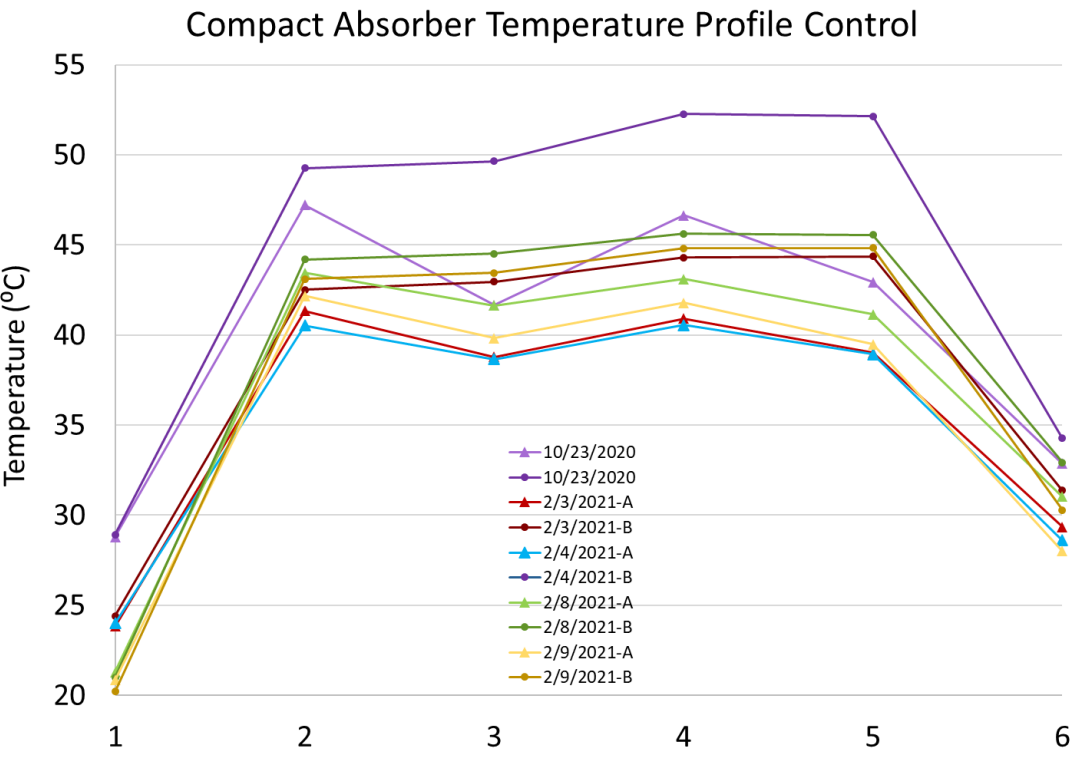
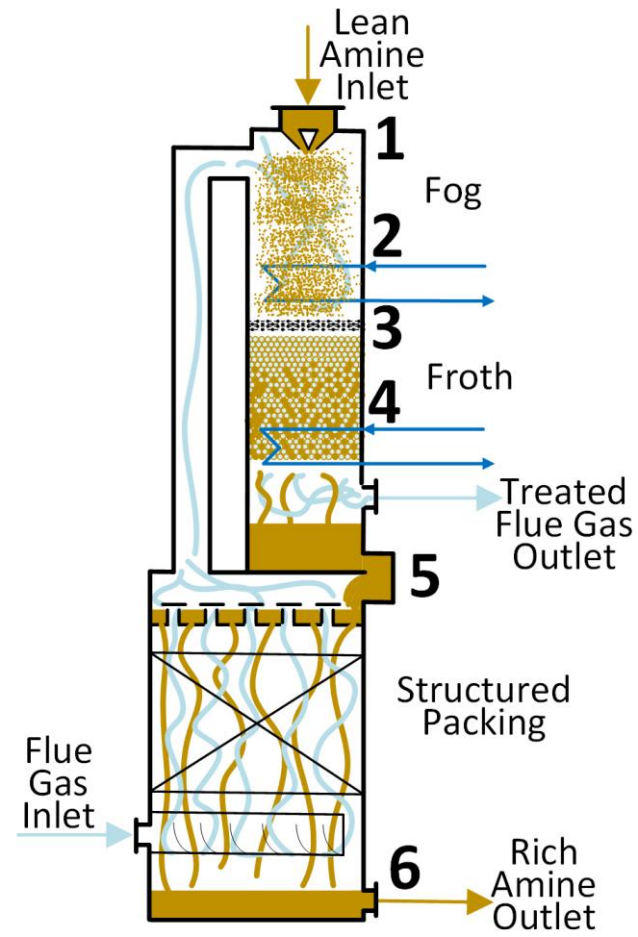


Specific Absorption in Each Section of the Compact Absorber Parametric Campaign Compared with that in a Traditional Absorber

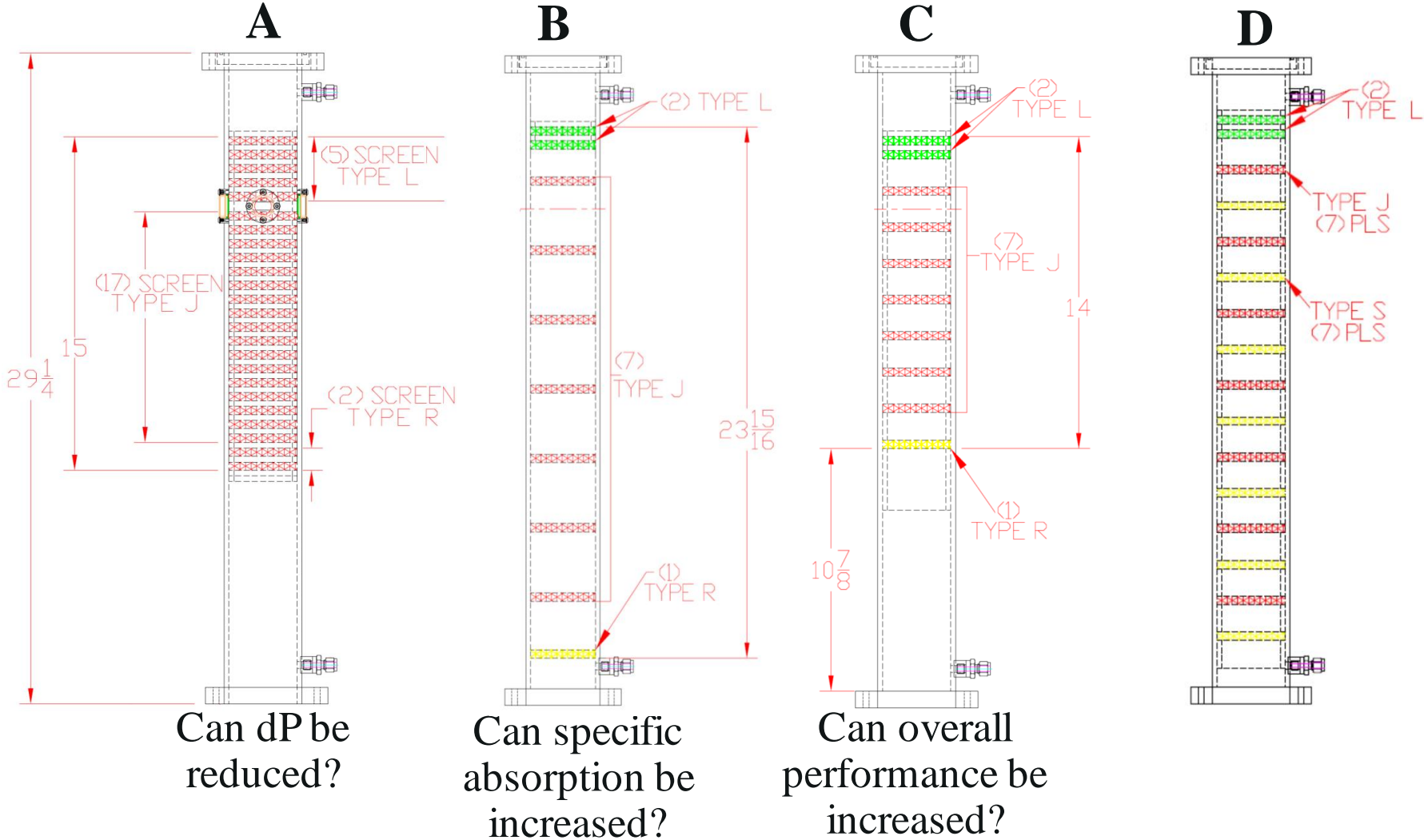


# Progress: Column Temperature Control

Target flatter absorber temperature profile obtained with in situ heat exchangers.



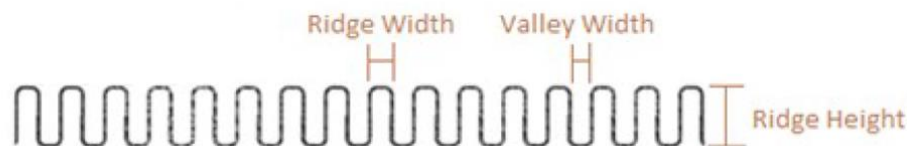
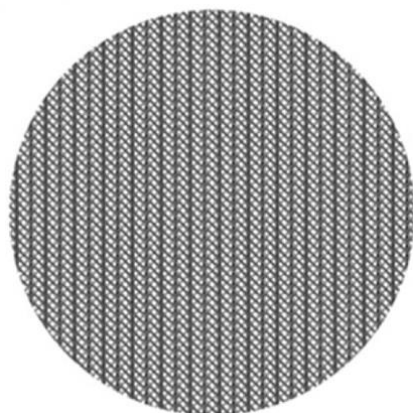
# Progress: Frothing Section Configuration



# Progress: Frothing Section dP

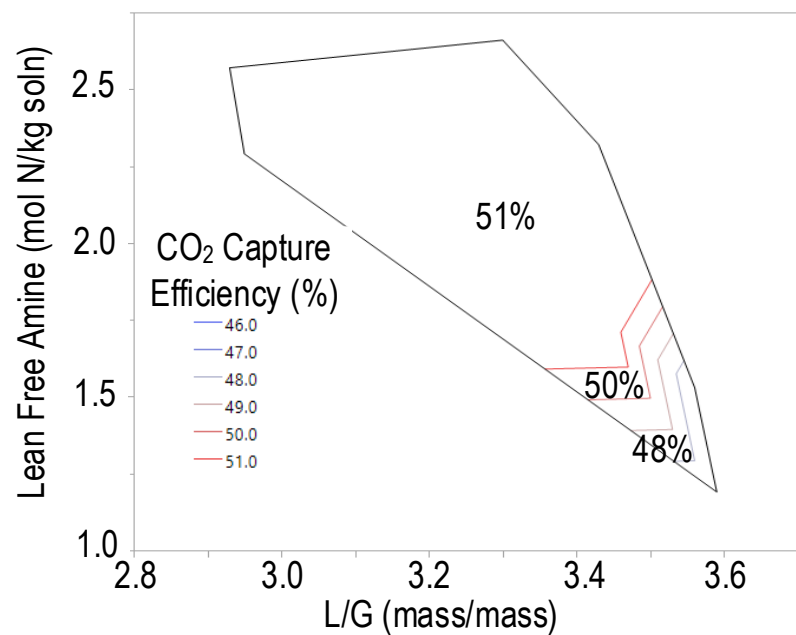
Target <5 in H<sub>2</sub>O

	L/G (mass/mass)	Gas Velocity (m/s)	Pressure Drop (in. H <sub>2</sub> O)
Evaluation with Water and Gas	4.62	1.7	4.5
Operation with UK CAER Solvent			
A	2.9-3.6	1.5	6.2-7.8
B	2.8-3.5	1.5	2.4-2.7
C	2.9-3.6	1.4	2.2-2.9
D	2.9-3.6	1.4	1.9-3.4

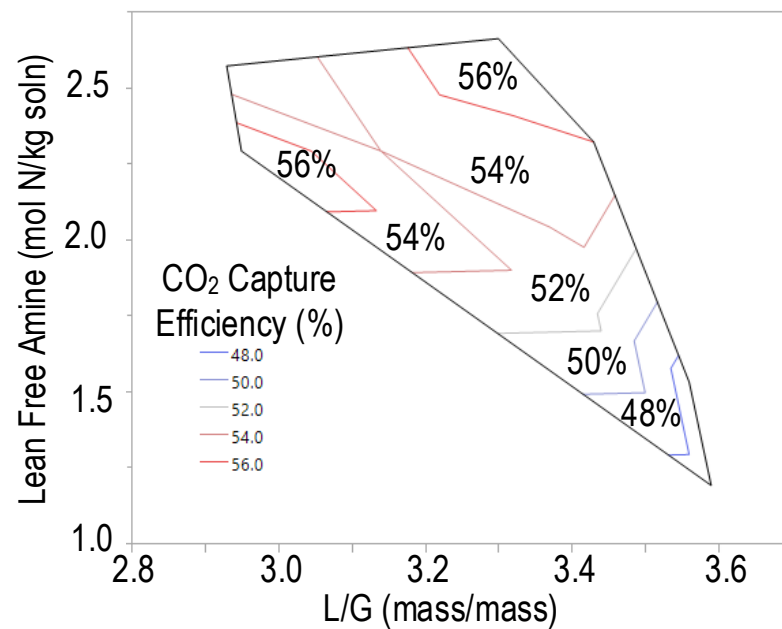


# Current Status

Frothing Configuration C

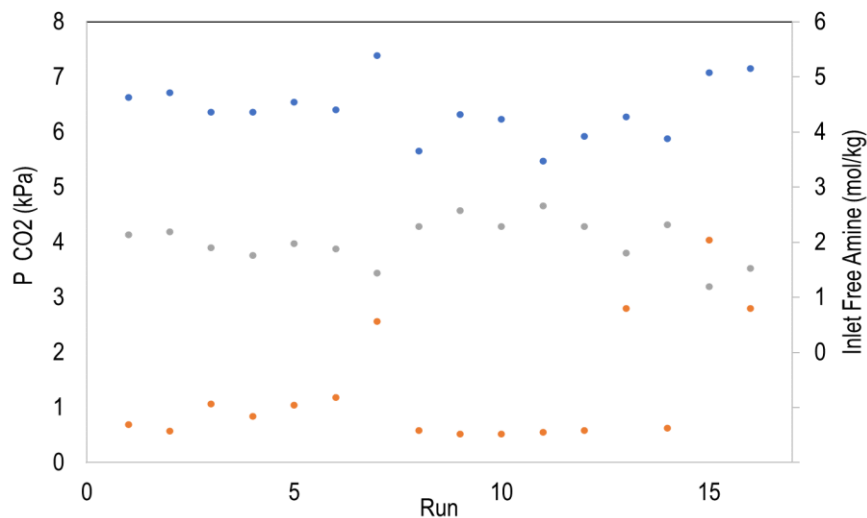


Frothing Configuration D

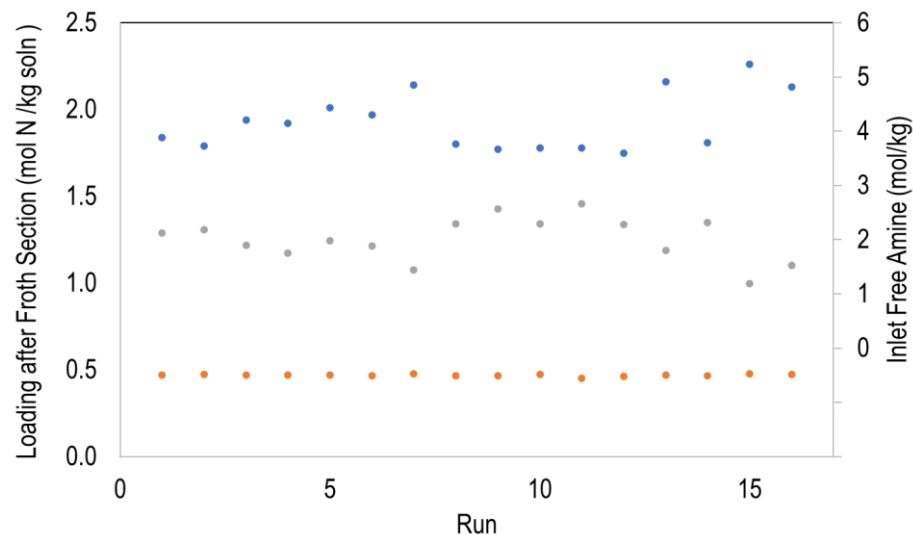


# Current Status

• Experimentally Determined PCO<sub>2</sub> • Predicted Equilibrium PCO<sub>2</sub> • Lean Free Amine



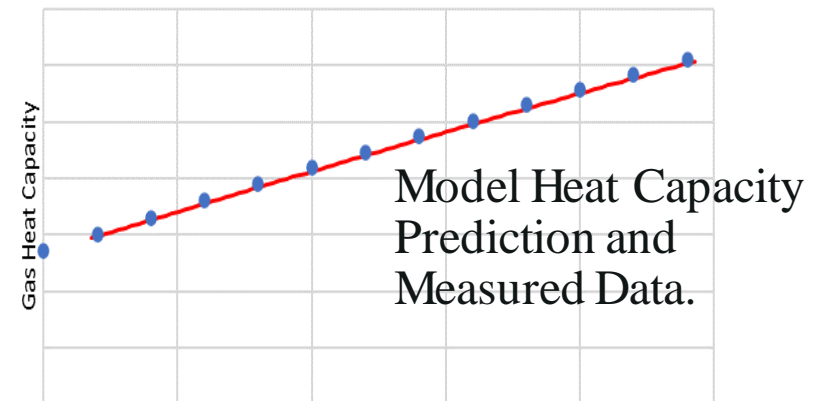
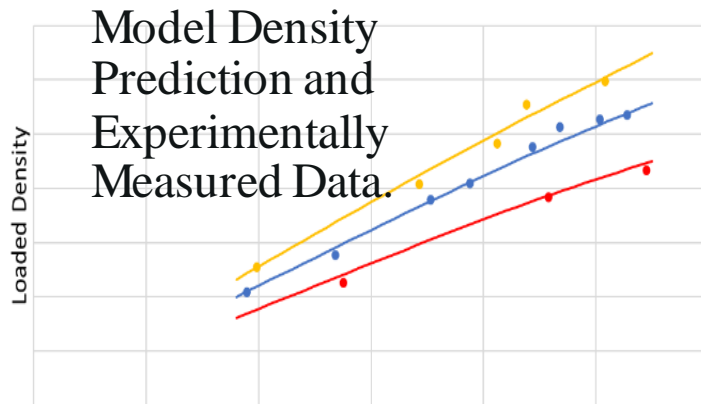
• Experimentally Measured C/N • Predicted Equilibrium C/N • Inlet Free Amine



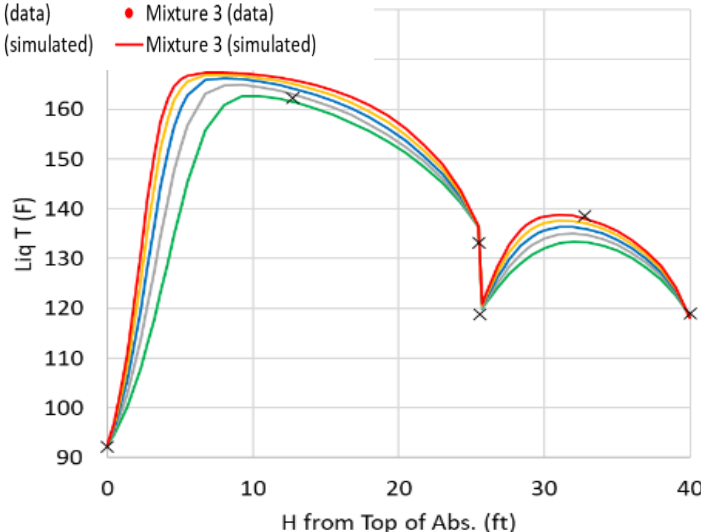
# Current Status

Rate based solvent model complete with good physical property and absorber profile prediction ability.

Process model underway for H&MB stream table generation.



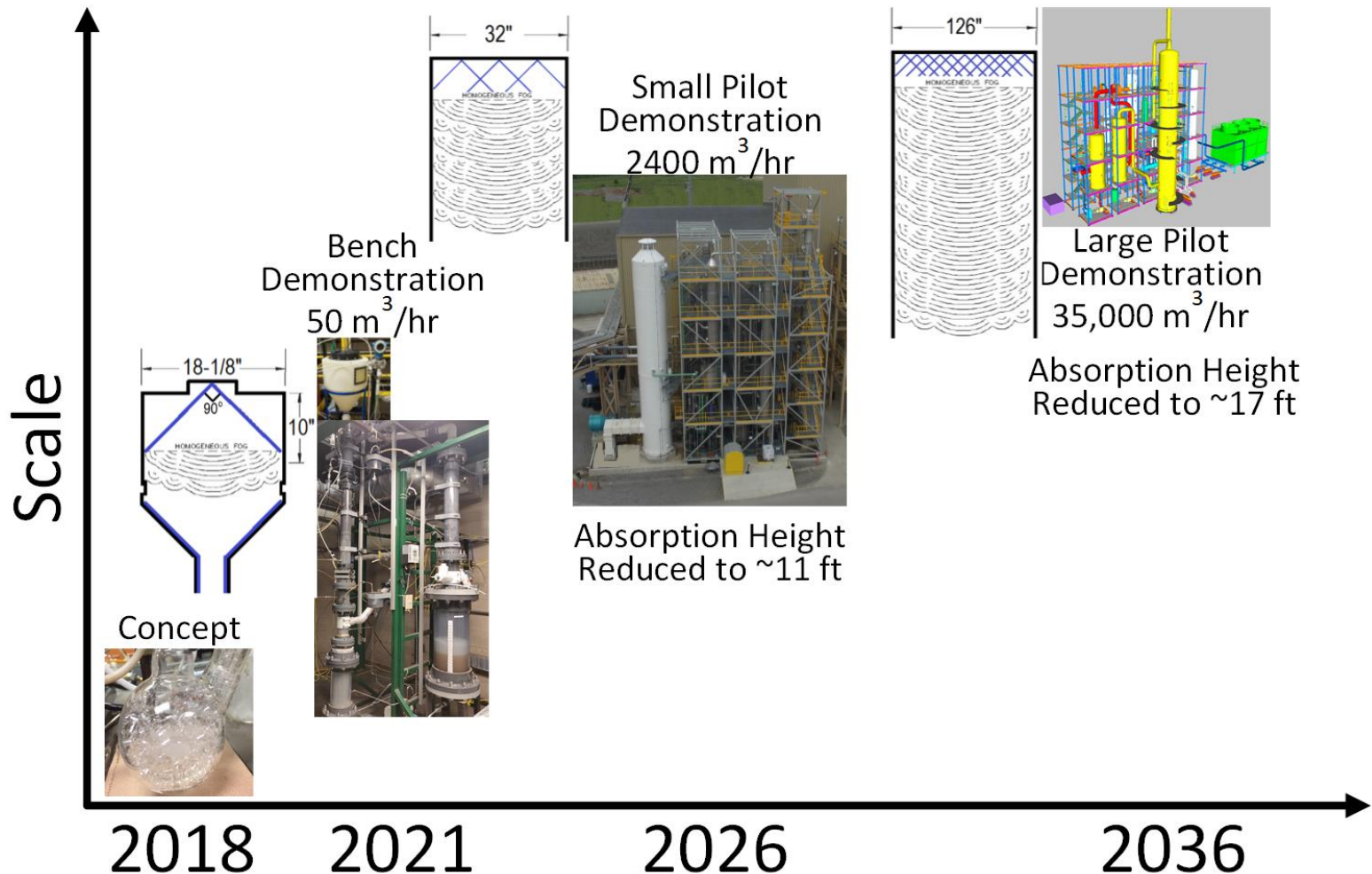
● Mixture 1 (data) ● Mixture 2 (data) ● Mixture 3 (data)  
— Mixture 1 (simulated) — Mixture 2 (simulated) — Mixture 3 (simulated)



Model Predicted Absorber Temperature Profile Interfacial Area Sensitivity Study (solid lines) and UK CAER 0.7 MWe Small Pilot Experimental Data (X).



# Future Testing and Development





# Summary

- All BP1 Deliverables and Milestones Met; All BP1 Success Criteria Achieved
- On Schedule to meet and achieve BP2 Deliverables, Milestones and Success Criteria
- 60+% CO<sub>2</sub> capture and 0.45 mol C/mol N rich loading demonstrated, on par with traditional absorber performance with >2X of structured packing
- Fog droplet SMD  $\leq 50$   $\mu\text{m}$  in diameter
- Froth bubble size <3.7 mm in diameter
- Up to 4.1X increased absorption than traditional column
- 6.8X greater surface area in fogging and frothing sections than 250Y structured packing
- 65-90% absorption occurs in the fogging and frothing sections
- Multimedia i-poster presentation at 2020 Virtual AIChE Meeting, 11/16/2020-11/20/2020

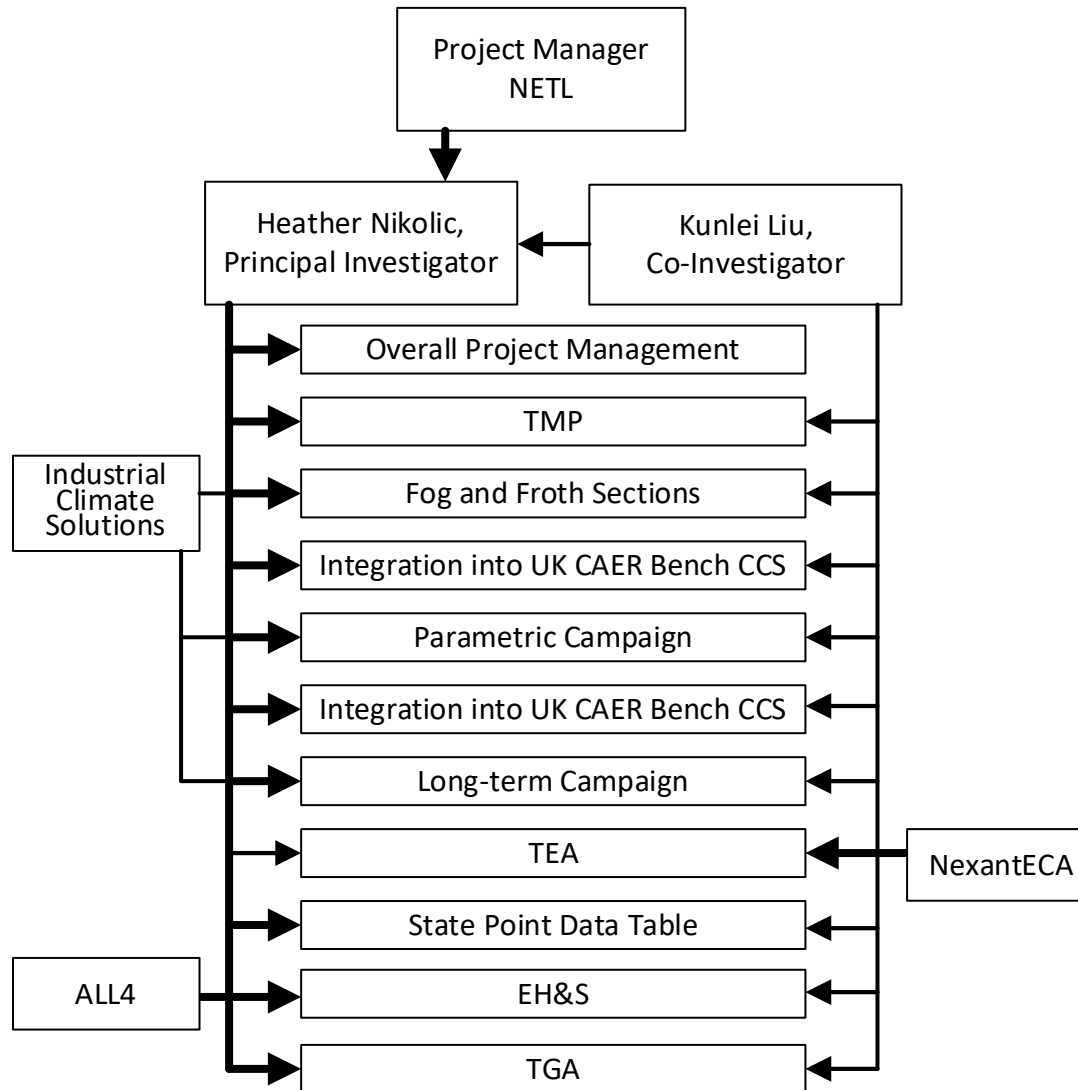
# Acknowledgements

Thank you to:

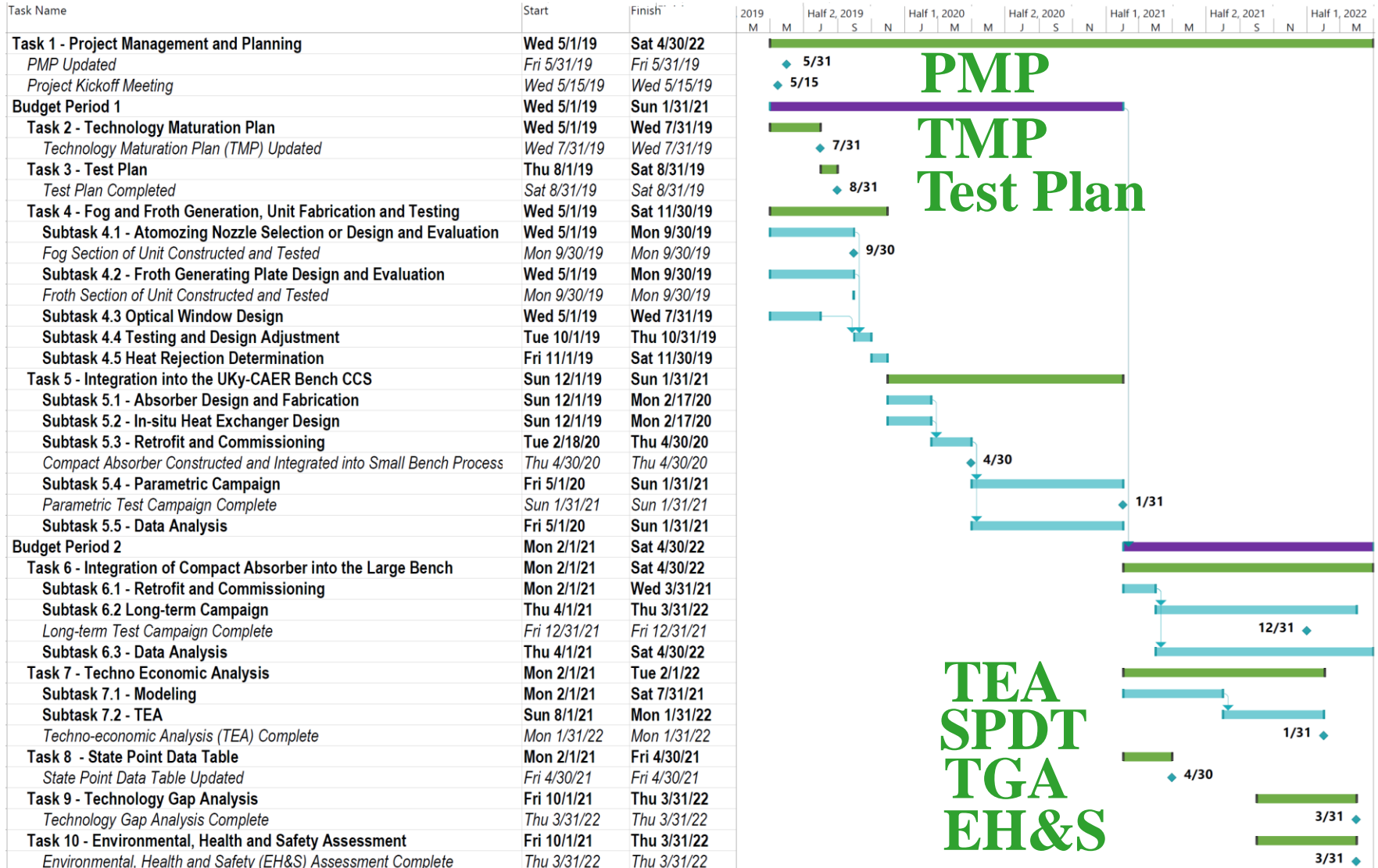
- **Carl Laird, Patrick Mayle and Angela Harshman, DOE NETL**
- **Bill Hargrove and Richard Adamson, ICSI**
- **Clay Whitney, ALL4**
- **Haoren Lu, NexantECA**
- **Shanice Edwards, Reynolds Frimpong, Len Goodpaster, Roger Perrone, Aaron Smith, Xiaoshuai Yuan and Kunlei Liu, UK CAER**



# Organization Chart



# Gantt Chart



## State Point Data Table for CAER Hindered Primary Amine Blend Solvent.

	Units	Measured/Estimated Performance During BP2
<b>Pure Solvent</b>		
<b>Molecular Weight</b>	g mol <sup>-1</sup>	<90
<b>Normal Boiling Point</b>	°C	160
<b>Normal Freezing Point</b>	°C	2
<b>Vapor Pressure @ 15 °C</b>	bar	0.0007\$
<b>Working Solution</b>		
<b>Concentration</b>	kg/kg	<45*
<b>Specific Gravity (15 °C/15 °C)</b>	g/mL	1.01*
<b>Specific Heat Capacity @ 30 °C and 1 bar</b>	kJ/kg·K	3.5*
<b>Viscosity @ 40 °C</b>	cP	2.8 (fresh condition, C/N=0)*
<b>Surface Tension @ STP</b>	dyn/cm	51*
<b>Absorption</b>		
<b>Pressure</b>	bar	1
<b>Temperature</b>	°C	35-55*
<b>Equilibrium CO<sub>2</sub> Loading</b>	gmol CO <sub>2</sub> /kg	2.3*
<b>Heat of Absorption</b>	kJ/mol CO <sub>2</sub>	55-60 <sup>#</sup>
<b>Solution Viscosity @ 40 °C</b>	cP	4.3 (rich condition)*
<b>Desorption</b>		
<b>Pressure</b>	bar	1.6 – 2.2*
<b>Temperature</b>	°C	117-120*
<b>Equilibrium CO<sub>2</sub> Loading</b>	gmol CO <sub>2</sub> /kg	1.1-1.4*
<b>Heat of Desorption</b>	kJ/mol CO <sub>2</sub>	55-60 <sup>#</sup>

\*Experimental data, # Estimated, \$ Calculated based on VLE