

CREATES: CO₂ REduction for grAphiTE Synthesis

DE-SC0020776



Anthony R. Richard, Ph.D.

Acadian Research & Development LLC

Laramie, WY

U.S. Department of Energy

National Energy Technology Laboratory

Carbon Management and Natural Gas & Oil Research Project Review Meeting

Virtual Meetings August 2 through August 31, 2021

Project Overview

Period of Performance:

Original: June 29, 2020 – June 25, 2021

No Cost Extension (COVID-19): December 31, 2021

Project Funding:

DOE \$239,379

Cost Share \$0

Project Participants:

Acadian Research & Development LLC

Project Overview

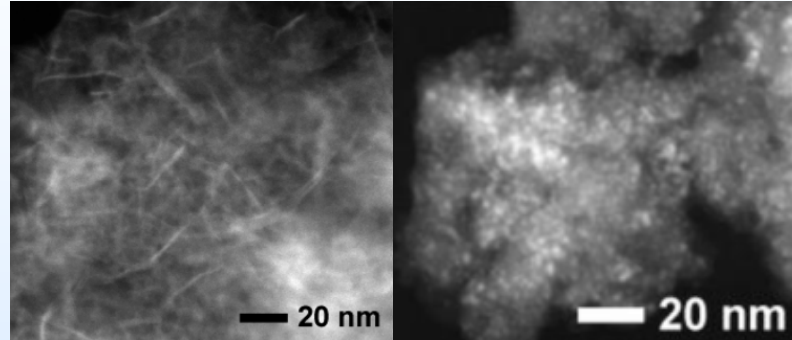
Project objectives are designed to exploit the strengths of the catalysts, and to evaluate catalyst performance and stability.

The five feasibility objectives of this Phase I SBIR project are:

- a) Demonstrate successful synthesis of catalysts
- b) Show that catalyst can reduce CO₂ to produce carbon
- c) Demonstrate catalyst stability of greater than 30h and ability to regenerate
- d) Demonstrate successful operation of the multi-stage reactor
- e) Demonstrate that the dominant allotrope produced is graphite

Technology Background

Previous catalyst development: methanol synthesis from CO₂



Calcined catalyst Reduced catalyst

Metal silicate-based system

- Good CO₂ interaction
- Active at mild temperatures / ambient pressure
- Resistance to coking – carbon rejection
- Resistance to sintering
- Partial reduction yields isolated metal nanoparticles

Promoter addition

- Good promoter incorporation
- Improved methanol synthesis activity
- Stability not decreased by promoter

Richard, A. R.; Fan, M. ACS Catalysis 2017, 7 (9), 5679-5692.

Richard, A. R.; Fan, M. Fuel 2018, 222, 513-522.

Technology Background

Applied technology

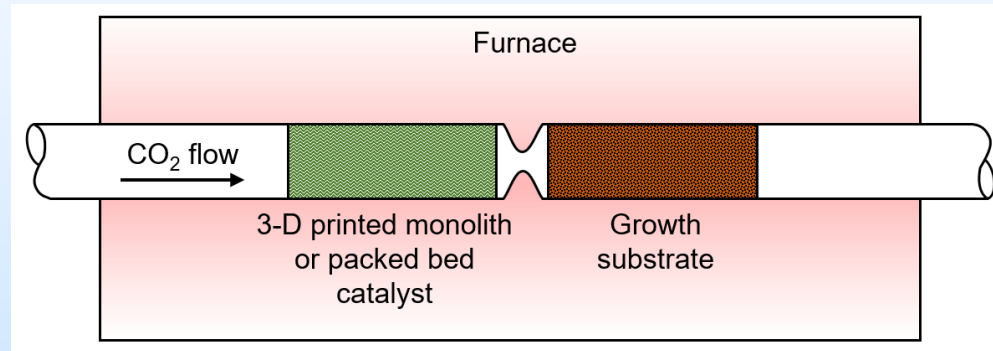
Aspect	Advantages
Catalyst metals	Catalytic activity with CO ₂
Thermal treatment to form catalyst nanoparticles	Sintering resistance
Promoter	Increased activity
3D-printed morphology	High contact surface area and low pressure drop

Key challenges

Aspect	Challenges
Conversion	Rate must be sufficient for feasibility
Stability	Long catalyst life desired
Regeneration	Ability to repeatedly regenerate
Carbon growth	Growth on substrate / allotrope is graphite

Technology Background

Multi-stage reactor



- CO₂ is decomposed in the first stage
- Carbon is deposited on growth substrate in second stage

Technical Approach/Project Scope

Work plan

- Prepare the catalysts with and without promoter.
- Characterize catalysts via XRD, FTIR, SEM, Raman.
- Complete CO₂ testing in TGA.
- Examine regeneration performance.
- Scale-up catalyst synthesis for 3D printing.
- Testing in multi-stage FBR.

Milestones

1. Successfully synthesize catalysts with and without promoter
2. Decompose CO₂ and produce carbon in FBR

Technical Approach/Project Scope

Project success criteria:

Synthesis: Achieve intended catalyst structure as characterized by XRD and FTIR, and comparable to previous work and literature values.

Performance: Activity for at least 30 hours with final activity $>70\%$ of the initial (first 5 hours) activity.

Regeneration: Activity restored to at least 85% of initial activity.

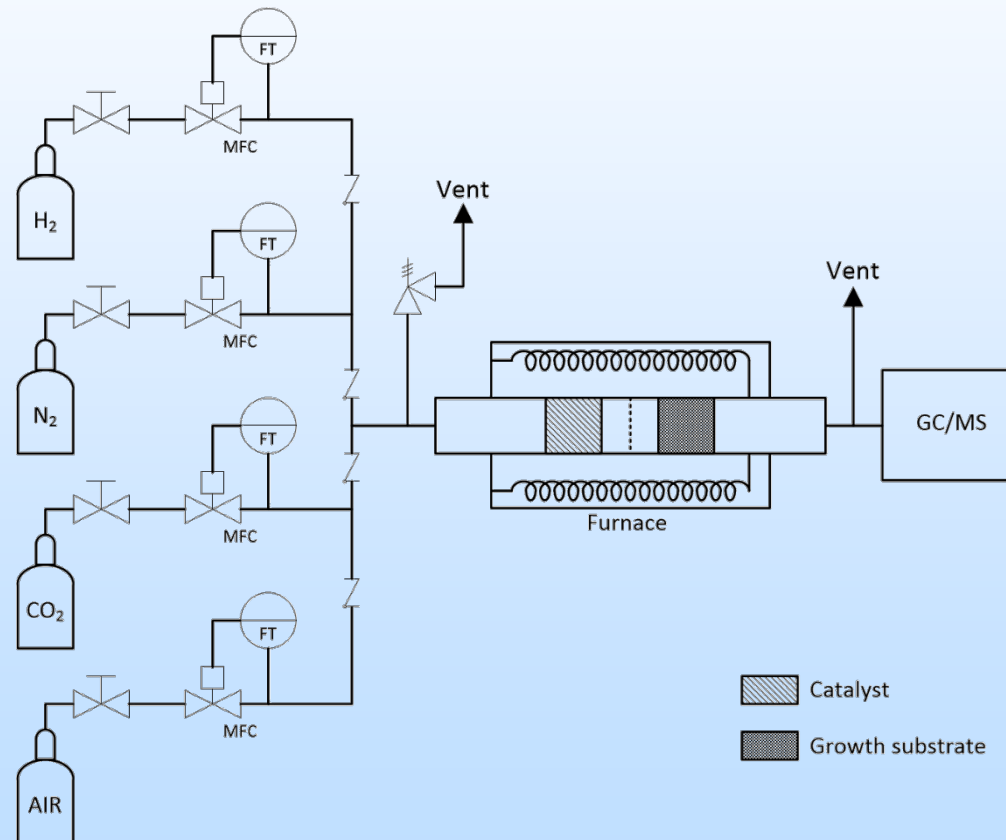
3D-printing: Monolith strong enough for handling, loading into reactor, and testing without significant attrition.

Carbon allotrope: Graphite formation on growth substrate as determined by Raman.

Cycling stability: Maintain $>75\%$ of the initial activity.

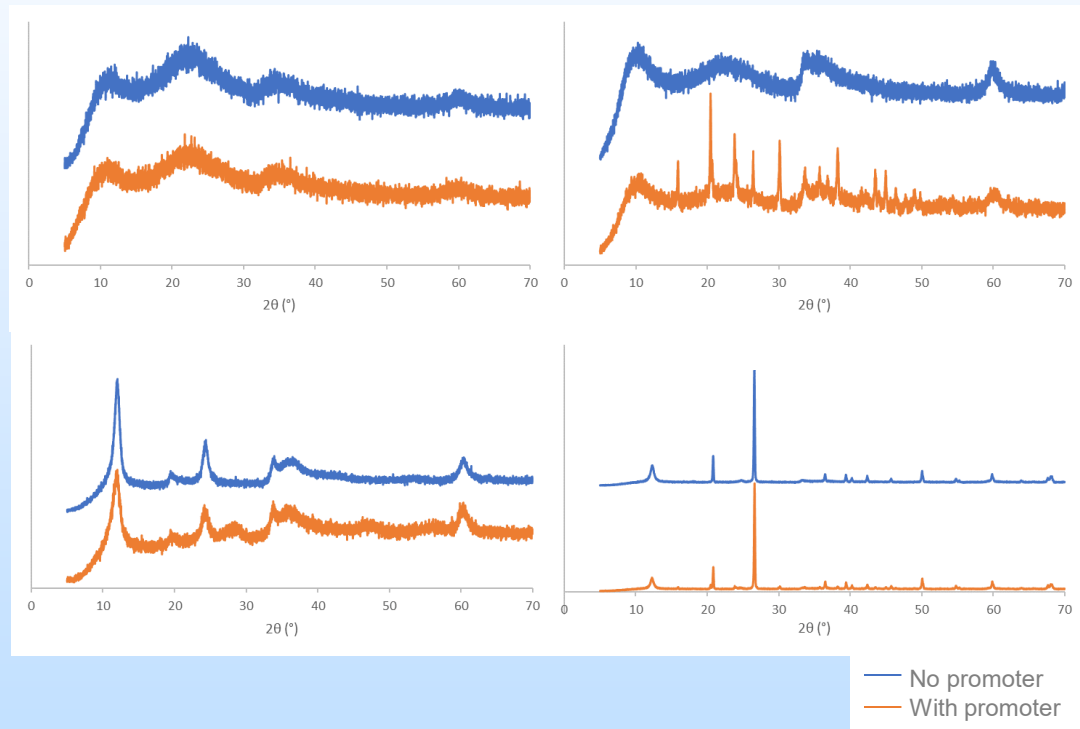
Progress and Current Status of Project

FBR testing setup



Progress and Current Status of Project

Synthesis



Progress and Current Status of Project

Performance

FBR testing with catalyst powder and sand

Catalyst reduced in situ

Reaction gas: Only CO₂

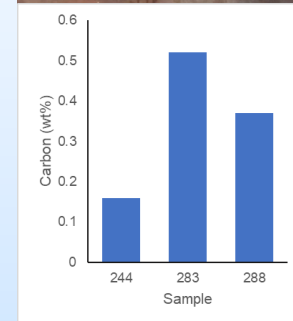
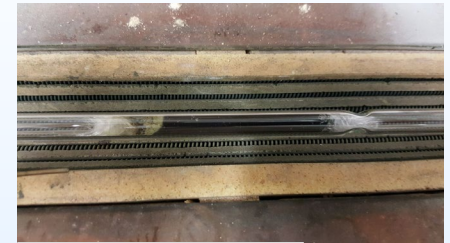
Temperature: 350–650 °C

Time: 1–55 h

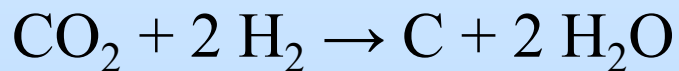
Carbon formation in bed

Formation rate: 0.03 wt%/h

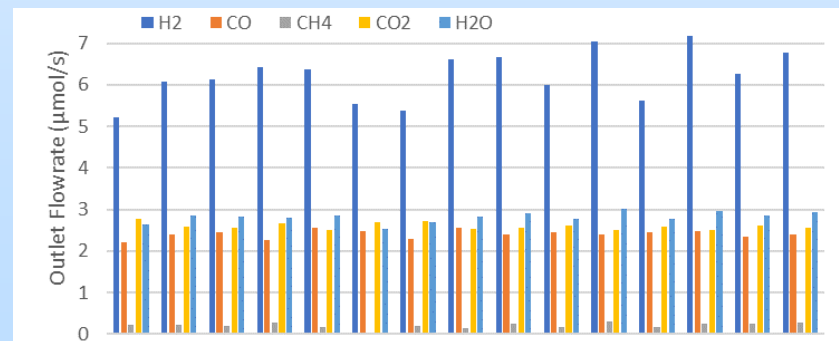
Surface content: 0.52 wt%



Bosch Reaction:



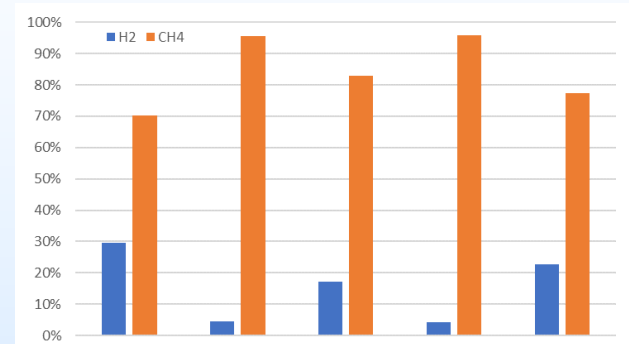
Moderate
conversion and
CO production



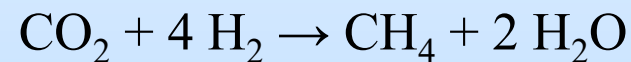
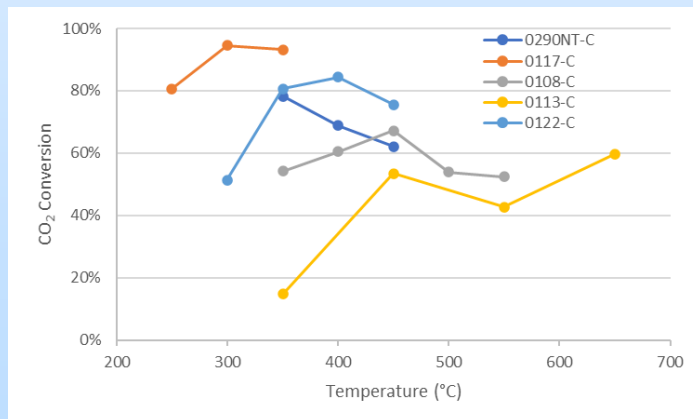
Progress and Current Status of Project

Performance

CH₄ decomposition: to confirm catalyst activity and system performance, methane decomposition was performed



Good performance led to testing CO₂ decomposition via the Sabatier reaction

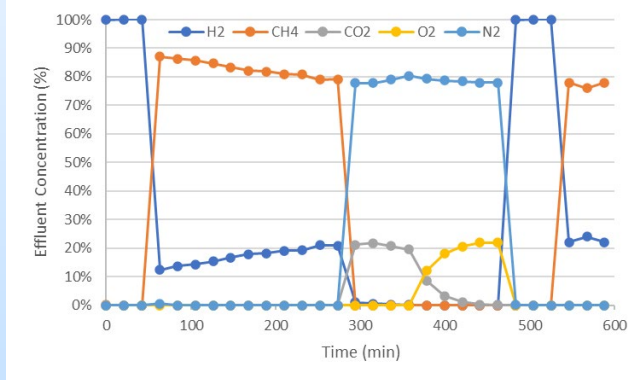


- CH₄ favored at low temperature
- CO appears above 500 °C

Progress and Current Status of Project

Regeneration

Before regeneration testing, catalyst was loaded with carbon via methane decomposition.



Catalyst regeneration in air at 450 °C showed equivalent or better performance after regeneration.

Progress and Current Status of Project

Catalyst geometry



Extruded pellet support with catalyst coating



Extruded mixed support/catalyst

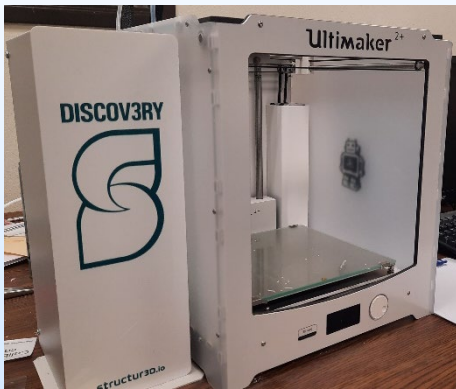


3D-printed:

- Printed substrate, coated
- Printed mixed support/catalyst

Progress and Current Status of Project

3D-printing



3D-printer paste extruder system

Ink development (paste for extrusion)

Components:

- Binder
- Liquid
- Lubricant
- (Catalyst)

Requirements:

- Appropriate viscosity
- Layer adhesion
- Reasonable curing time/temperature



Plans for future testing/ development/commercialization

This project:

Complete scheduled tasks

Identify best catalyst candidates

Next:

Scale-up production of catalysts

Scale-up of testing

Summary

- Pure CO₂ decomposition is possible at these reaction conditions, but the rate is slow.
- Bosch reaction shows moderate conversion.
- High methane decomposition performance inspired CO₂ conversion via Sabatier reaction.
- Graphite formation elusive.
- Successfully developed ink formulations for 3D-printing catalysts.
- Successfully grown catalyst on support surface.

Thank You

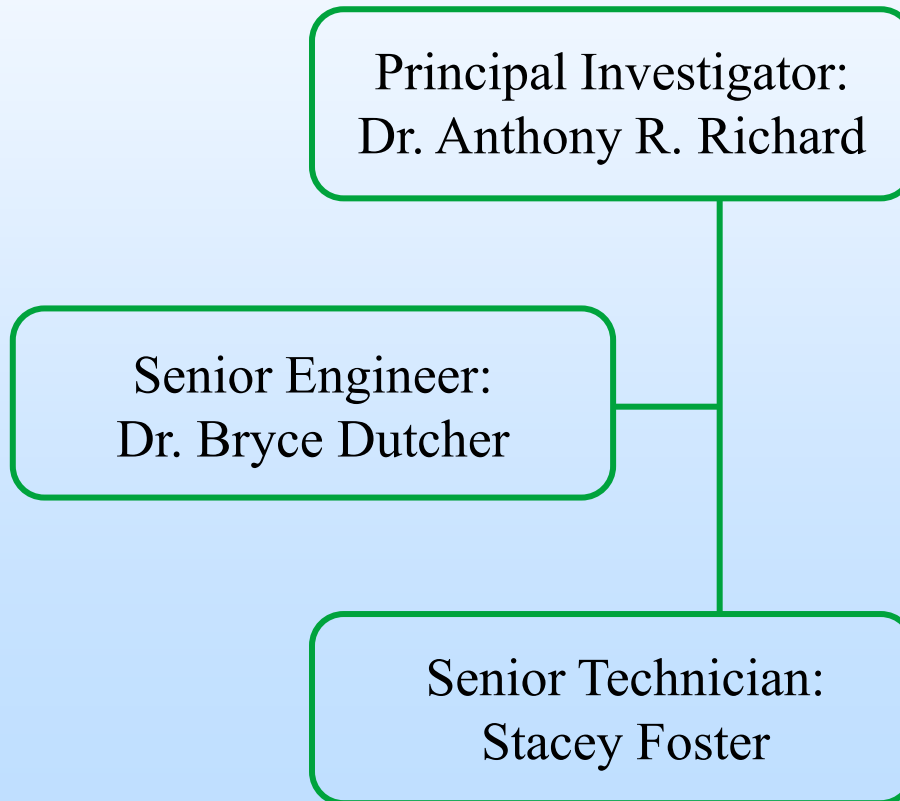
Questions?

Appendix

- These slides will not be discussed during the presentation **but are mandatory.**

Organization Chart

Project Team - Acadian Research & Development



Gantt Chart

