Inorganic Membrane-based Reactive Separation and Reactant Recycle for Direct Synthesis of Dimethyl Carbonate (DMC) DE-SC0019556

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Media and Process Technology Inc.



Project Overview

<u>Program:</u> Phase II Small Business Innovation Research (SBIR)

<u>Funding:</u> Overall project budget: \$1,050,000 (including DCA funding).

Overall Project Performance Dates: March 19, 2020 - March 18, 2022 (24 months)

Project Participants:

- Media and Process Technology...Membrane manufacturer/supplier and technology developer
- TechOpp Consulting, Inc....Discretionary Commercialization Assistance provider (POC: Mary Ann S. Bonadeo)

Overall Phase II Project Objectives:

Demonstrate the proposed inorganic membrane-based DMC Production process at pilot scale at the relevant operating conditions:

- (i) Improved Dimethyl Carbonate (DMC) conversion in a membrane reactor configuration
- (ii) Improved DMC product recovery in a downstream membrane nanofilter.



Motivation: Carbon Capture and <u>Reutilization</u>

Carbon Capture and Reutilization

✓ Effective means to overcome carbon storage problems.
 ✓ Green alternative to petroleum sources of fine chemicals

Focus: Dimethyl Carbonate

- ✓ Green production (CO₂ and biomass MeOH)
- ✓ Low toxicity
- ✓ Low viscosity
- ✓ High solvent power



Markets: DMC Current and Potential Demand



~\$6,000MM/yr

~\$300MM/yr



Synthesis: Dimethyl Carbonate (DMC)

	route	description	reactions
	1	methanol phosgenation	COCl ₂ + 2 CH ₂ OH -> (CH ₂ O) ₂ CO ⁴ + 2HCl
*	Π	oxidative carbonylation of methanol (Eni)	$CO + 2CH_3OH + O_2 \rightarrow (CH_3O)_2CO^a + H_2O$
*	Ш	oxidative carbonylation of methanol via methyl nitrite (Ube)	$N_3O_3 + 2CH_3OH \rightarrow 2CH_3ONO + H_2O$ 2CH ₃ ONO + CO → $(CH_3O)_2CO^a + NO$
*	IV	ethylene carbonate transesterification (Asahi)	$(CH_2)_2O + CO_2 \rightarrow C_2H_4O$ $C_2H_4O + 2CH_3OH \rightarrow (CH_3O)_2CO^a + (CH_2OH)_2$
?	V	ure a transesterification	$2NH_3 + CO_2 + (NH_2)_2CO + H_2O$
			$(NH_2)_2CO + CH_3OH \rightarrow CH_3OCONH_2 + NH_3$
			$CH_3OCONH_2 + CH_3OH \rightarrow (CH_3O)_2CO^a + NH_3$
	٧I	direct synthesis from $\rm CO_2$	$CO_2 + 2CH_3OH \rightarrow (CH_3O)_2CO^a + H_2O$

Technology Background



Direct Synthesis of DMC: Advantages and Challenges



Advantages of Direct Synthesis of DMC

- **1.** <u>Green Process:</u> Produced from CO₂ and biomass derived methanol
- 2. <u>Non-hazardous:</u> Reactants are non-hazardous versus other synthetic pathways.
- **3.** <u>Safety:</u> Considerably safer operating conditions than commercial processes.

Challenges of Direct Synthesis of DMC

- **1.** Equilibrium limitation: Conversions to DMC limited to under 2 to 5%
- 2. <u>Methanol/DMC Azeotrope:</u> Multiple column azeotropic distillation required.
- **3.** <u>Energy intensive:</u> Combination of these problems drives production cost



Solution: Membranes in DMC Synthesis and Recovery

Membrane Reactor: In-situ Dehydration

- 1. <u>Operation Mode:</u> Pervaporation/Vapor permeation.
- 2. <u>Equilibrium Shift:</u> Increase DMC yield via in-situ water removal
- 3. <u>Increased Yield:</u> Reduce downstream separation and recycle requirements.

Membrane Separator: MeOH/DMC

- 1. **Operation Mode:** Nanofiltration
- 2. <u>Azeotrope Break:</u> Remove MeOH (and CO₂) from DMC.
- 3. <u>Reduced Energy and Cost:</u> Eliminate multiple column azeotrope distillation.



<u>Challenge</u>

<u>Operating Conditions</u> 80 to 150°C; 300 to 1,500psig <u>Operating Environment</u> MeOH, DMC, CO₂, water

Solution

<u>Inert Membrane</u> Inorganic Nanofilter



Reaction. In-situ Dehydration in Direct Synthesis

MR Configuration

Dehydration Approach	DMC Yield Impact	Advantages	Barriers/Problems
Organic Dehydrating Agent	10 to >95%	 a. Highest DMC yields observed. b. Continuous processing capable and demonstrated on bench. c. Very high reaction rates demonstrated in continuous versus "batch" reactors. d. High selectivity to DMC for some selected systems. 	 The dehydrant yields a. Byproduct formation due to dehydrant degradation. b. Catalyst deactivation. c. Limited upper operating temperature (<140°C). d. Additional separation and regeneration equipment. e. Dehydrant loss; hence makeup chemicals required.
Inorganic Adsorbent	<10 to ~40%	 a. Commercial examples of cyclic "dehydration" are installed/operational (corn ethanol production, for instance). b. No byproduct formation. >99% selectivity c. No catalyst deactivation. 	 a. Low H₂O adsorbent capacity at reaction temperatures. b. Discontinuous highly unfavorable cyclic operation due to mismatch of the reaction and adsorption/regeneration temperatures. c. Long reactor residence times.
Membrane Reactor	Up to ~10%.	 a. No byproduct formation. >99% selectivity b. No operating temperature limit in the expected range. c. Continuous processing capable. d. No chemical makeup required. e. No catalyst deactivation 	 a. Yields above 10% have not been laboratory demonstrated. <u>Very low-quality membranes have been tested.</u> b. Limited range of membrane candidates; CO₂ and MeOH are good plasticizers; T > 120C; P>>300psig



Separation: Nanofiltration for MeOH/DMC separation



Equipment Impact

Equipment	Code	Key specifi- cations	Height [m]	Diameter [m]
Distillation columns	Dist-01 Dist-02 Dist-03 Dist-04	50 stages 79 stages 67 stages 49 stages	27.5 42.0 36.0 27.0	3.16 2.92 2.00 1.58
Flash drum	-		10.6	3,32
Compressors	C-01 C-02	1530 kW 30 kW	-	-
Pumps	P-01 P-02 P-03 P-04	4 kW 133 kW 28 kW 8 kW	-	-
Heat	P-05 Reboiler Dist-01	13 kW 471 m ² 270 m ²	-	-
exchangers	Reboiler Dist-02 Reboiler Dist-03 Reboiler Dist-04	104 m ² 142 m ²	-	-
	Condenser Dist-01 Condenser Dist-02	1861 m ² 175 m ²	-	-
	Condenser Dist-03 Condenser Dist-04 He-01	326 m ²	-	-
	He-02	10 m ²	-	-
	He-04	99 m ²	-	-
	He-05 He-06	9 m ² 41 m ²	-	-
	He-07	15 m ²	-	-
	He-08 He-08	5 m ² 5 m ²	-	-

MPT Membrane Technology





Advanced Inorganic Molecular Sieving Membranes



MPT 57-tube Bundle (Carbon Molecular Sieve Membrane)

Package into Multiple Tube Bundle



CMS Membranes in Direct Synthesis of DMC. Proof of Concept

Phase I Results MR Water Separation 160°C; 350 to 450psig



Vapor	Kinetic Diameter [Å]	Permeance [GPU]	Selectivity [H ₂ O/x]
Water	2.6	230 to 280	-
MeOH	3.8 to 4.1	0.55 to 1.0	>290
DMC	>4.6 to 6.3	0.26 to 0.54	>530

Phase I Results NF MeOH/DMC Separation 120 to 200°C; 100 to 300psig



DMC Rejection [%]

Cumulative test duration [hours]

Project Technical Scope and Approach





M&P TECHNOLOGY Phase II Program

Phase II Project: Multiple Tube Bundle Scaleup at Target Conditions

Primary Technical Goal

Process Scaleup at Target Operating Conditions

Demonstrate multiple tube membrane bundles in MR and NF operation at the target operating conditions.

Objective	Target/Goal	Yr
#1	Membrane Reactor (MR) Testing with Full length single tube membrane testing at up to 1,500 psig and 200°C	1
#2	Nanofiltration (NF) Testing with Full length single tube membrane testing at up to 1,500 psig and 200°C	1
#3	Multiple tube bundle fabrication (MR and NF membranes) for testing under target operating conditions.	1
#4	Multiple Tube Bundle Demonstration Unit Construction Target is ~0.1 to 1 lb/hr DMC production.	2
#5	Operation of the Multiple Tube Bundle Demonstration Unit in MR & NF modes	2
#6, #7, #8	Optimize and update process flow diagram and TEA. Continue commercialization development.	1,2



M&P TECHNOLOGY Phase II Program

Phase II Project Task Plan

				1		Yea	ar I					Year II								
M	onth	1 2	3	4	5	6	7	8	9 1	0 1	12	13	14 1	5 16	17 1	18 19	9 20	21 2	2 23 2	4
Task 1. Fabrication of CMS Membranes for								MII-	1											
Phase II Program.						V		IVI II-	2											
Task 2. Conduct Single Tube Testing at the																				
Relevant Operating Conditions																				
Subtask 2a. System Shakedown Testing at Targe	et						N	4II-3			Τ					Τ				1
Operating Conditions								111-7												_
Subtask 2b. Single Tube Membrane Reactor				MII	-4	♦			N	111-5	l o							N	III-6	
Testing									IV.	III-5										
Subtask 2c. Single Tube Nanofiltration			-	MII-	8	•				мп_0	¢					Τ		M	П 10	
Membrane Testing														<u>ا</u>				IVI		
Task 3. CMS Membrane Bundle Fabrication							мπ_	13				l N	ИП-1 ИП-1	1 2						
Task 4. Multiple Tube Bundle Testing Unit							.v111				Ø			Τ						
Construction and Shakedown									N	4II-1-	4									
Task 5. Scaleup with Multiple Tube Bundle MR	2																	MI	1-15	
and NF with Testing														┛				M	I-16	
Task 6. Update Process Flow Diagram and												M	(II-17				MII-	18		
Process Economics; Prepare a Commercialization	on																			
Package for Potential Industrial Partners																		M.	ш-19	

Project Progress and Status





Task 1. CMS Membrane Preparation

Phase II Milestones MII-01. MR Membrane Fabrication

Phase II Milestones

MII-02. NF Membrane Fabrication

Membrane	Permeano	Selectivity						
ID	He	\mathbf{N}_2	He/N ₂					
[-]	[GPU]	[GPU]	[-]					
CMS-30Rg-2001	426	1.96	217					
CMS-30Rg-2002	303	1.04	292					
CMS-30Rg-2014	414	2.44	170					
CMS-30Rg-2018	346	2.19	158					
CMS-30Rg-2019	422	1.48	285					
CMS-30Rg-2020	112	0.52	216					
CMS-30Rg-2021	281	1.48	189					
CMS-30Rg-2022	388	1.57	248					
CMS-30Rg-2027	232	0.65	355					
CMS-30Rg-2030	342	2.11	162					
CMS-30Rg-2031	423	1.87	227					
CMS-30Rg-2034	403	2.62	153					
CMS-30Rg-2036	443	1.62	273					
CMS-30Rg-2037	399	1.23	325					
CMS-30Rg-2038	261	0.77	338					
CMS-30Rg-2040	361	1.90	190					
CMS-30Rg-2041	299	1.17	257					
CMS-30Rg-2043	564	1.53	369					
CMS-30Rg-2044	303	1.08	280					
CMS-30Rg-2045	542	1.26	431					
CMS-30Rg-2046	231	0.41	564					
CMS-30Rg-2047	524	2.87	183					

Membrane	Gas P	ermeance	[GPU]	Selec	trvity
ID	He	N ₂	SF ₆	He/N ₂	N ₂ /SF ₆
182NF	695	8.8	0.11	78.6	80.9
183NF	892	11.3	0.13	78.9	84.8
186NF	1024	31.7	0.09	32.3	340.9
191NF	1120	47.2	0.59	23.7	79.5
199NF	950	6.9	0.04	137.0	155.7
203NF	955	11.7	0.06	81.8	180.0
205NF	1197	12.6	0.34	94.8	37.6
206NF	832	27.8	0.17	29.9	161.4
207NF	1060	20.4	0.13	51.9	155.2
208NF	1037	12.4	0.20	83.5	62.8
209NF	911	11.3	0.33	80.6	34.6
210NF	1002	6.4	0.12	156.2	52.8
213NF	1123	15.5	0.08	72.4	204.3
215NF	1019	10.2	0.27	99.6	37.8
216NF	1036	10.1	0.05	102.5	192.2
217NF	706	18.6	0.15	38.0	124.0

NF Target: N₂/SF₆ >30 to 50

MR Target: He/N₂>150



Task 2. MR Membrane Performance Testing

<u>Phase II Milestones</u>

MII-03 to 6. Membrane Performance Verification at Target Conditions; Long Term Stability





Task 2. NF Membrane Performance Testing

Phase II Milestones

MII-07 to 10. Membrane Performance Verification at Target Conditions; Long Term Stability





Task 4. System Construction and Testing





Phase II Milestones MII-11 to 14. System upgrade/testing. 200°C; 1,500psig 7-tube Bundles



Nanofiltration System





Next Step

Task 3. Multiple Tube Bundle FabricationTask 5. Multiple Tube Bundle Performance Testing

CMS Bundle Targets

- ➤ Full Length (>30") CMS Tubes
- Full Ceramic Potting
- ➤ 3 and 7-Tube Full Ceramic Bundles
- ➢ Rating: <500°C; >1,500 psig

Phase II Milestones

MII-11. MR Bundle(s) Fabricated MII-12. NF Bundle(s) Fabricated MII-15. Complete MR Bundle Testing MII-16. Complete NF Bundle Testing

MPT Pilot/Field Testing Modules



MPT 57-tube Bundle (Carbon Molecular Sieve Membrane)



Rating: 1,500 psig; up to 450°C





M&P TECHNOLOGY Summary

Summary of MPT Process Impacts and Advantages

MPT Ceramic Membrane Enhanced DMC Production Process Impacts, Advantages, Key Takeaway

MPT Process Features	Impacts and Advantages							
Membrane Reactor subsystem onerates	 Enhanced methanol conversion 							
at high tomporature and prossure	 Enhanced DMC concentration 							
at ingli temperature and pressure.	Reduced "loads" on downstream unit operations							
NF subsystem operates at the MR	Direct integration with the MR without the need							
conditions (T and P)	for thermal or pressure cycling							
	Reagent recycle at reactor operating T and P							
	Further "load" reduction on downstream units							
NF subsystem shows excellent DMC	Improved reactant and product separation							
rejection	 Azeotropic distillation eliminated 							
NF subsystem permeate operated at	Substantially reduced CO ₂ recycle							
elevated pressures	recompression costs.							
Azeotropic Distillation eliminated	 Significantly reduce capital cost and energy 							
	consumption.							

Questions?



Appendix





Poly-generation: Advanced Inorganic Membranes

Project Team

Team Member	CV/Capabilities	Responsibilities
	MPT has nearly 30 years of experience in inorganic membrane materials and process development in high temperature, high pressure, aggressvive chemicals gas and liquid separations. Since 2003 MPT has been a commercial manufacturer and supplier of ceramic membrane technology into a wide range of ultrafiltration applications.	Project management; membrane bundle and housing development; performance testing; TEA update
	TechOpp Consulting has over 10 years of experience in transitioning new technologies from ideas to products by targeting the right opportunities, securing funding, establishing partnerships, and engaging with the customer to understand the application requirements.	Discretionary Commercialization Assistance provider



Poly-generation: Advanced Inorganic Membranes

Project Timeline and Milestones

						Ye	ar I						Year II									
Ma	onth	1	2	3 4	5	6	7	8	9	10	11	12	13	14 1	5 16	5 17	18	19 20	0 21	22 23	24	
Task 1. Fabrication of CMS Membranes for Phase II Program.						\diamond		MI MI	I-1 I-2													
Task 2. Conduct Single Tube Testing at theRelevant Operating Conditions																						
Subtask 2a. System Shakedown Testing at Targe Operating Conditions	et					♦	N	ИП- ИП-	3 7													
Subtask 2b. Single Tube Membrane Reactor Testing				MI	I-4					MII	-5								L	MII-6	•	
Subtask 2c. Single Tube Nanofiltration Membrane Testing				MI	-8	♦				M	I-9				l				1	MII-10		
Task 3. CMS Membrane Bundle Fabrication							мп	13					N N	1II-1 1II-1	1 2							
Task 4. Multiple Tube Bundle Testing Unit Construction and Shakedown							VI 11-	15		MI	[-14											
Task 5. Scaleup with Multiple Tube Bundle MR and NF with Testing	2														l				N N	4II-15 4II-16		
Task 6. Update Process Flow Diagram and Process Economics; Prepare a Commercialization Package for Potential Industrial Partners	on												М	II-17				MI	I-18 N	↓ ∕Ш-19	•	

End