

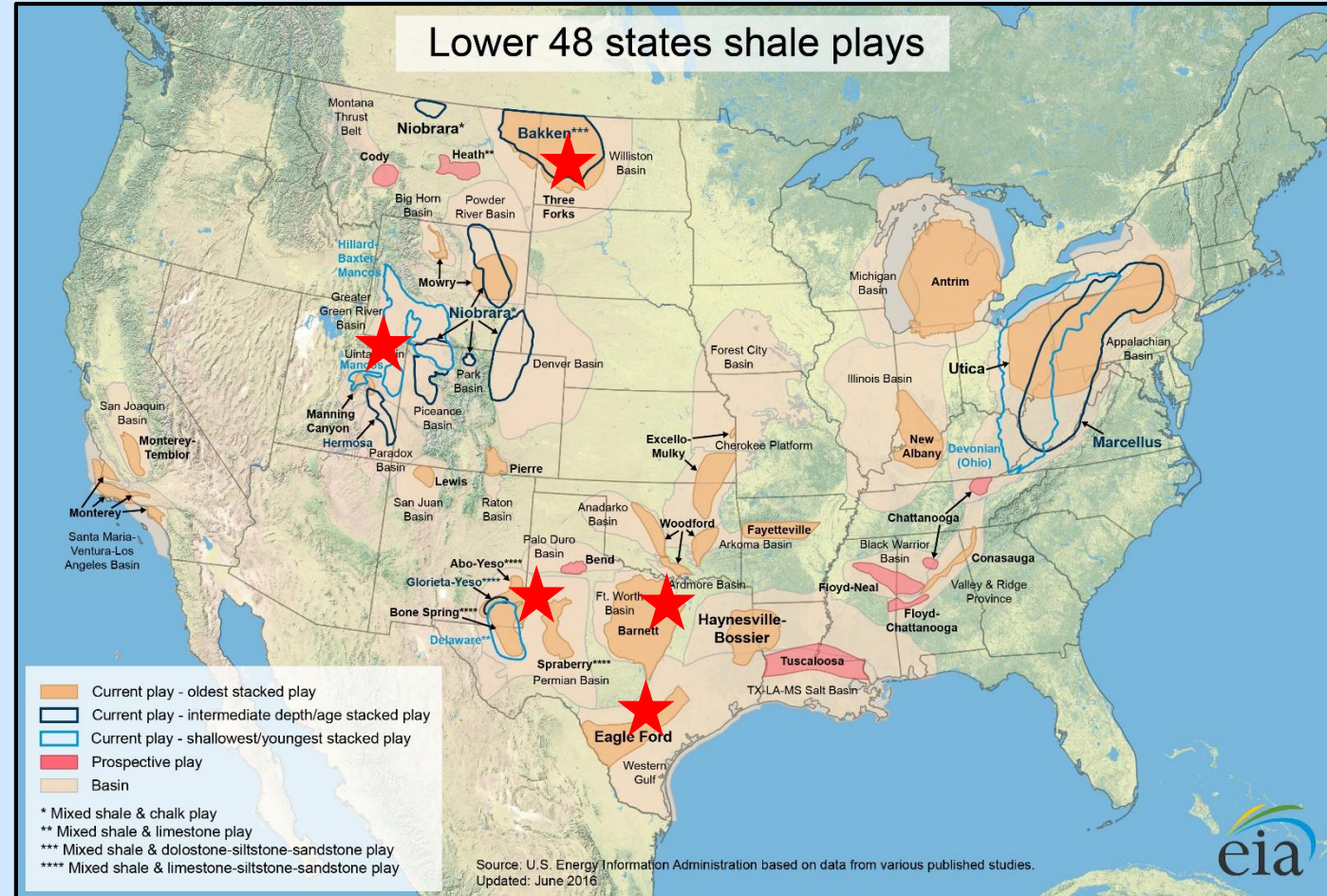
# FUNDAMENTAL RESEARCH PROJECT REVIEW MEETING

Virtual Agenda  
October 16, 2020

## Characterizing CO<sub>2</sub> as a Recovery Agent to Mobilize Hydrocarbons from Shale

Angela Goodman, Foad Haeri, Lauren C. Burrows, Parth G. Shah, Deepak Tapriyal, Robert M. Enick, Sean Sanguinito, Dustin Crandall

U.S. Department of Energy  
National Energy Technology Laboratory  
Oil & Natural Gas  
2020 Integrated Review Webinar



# Characterizing Application of CO<sub>2</sub> as a Recovery Agent to Mobilize Hydrocarbons from Shale

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- **Objective:**
  - Determine viability of CO<sub>2</sub> as an enhanced recovery agent for unconventional oil
- **Challenges:**
  - Primary oil recovery from fractured unconventional formations **is typically less than 10%** - EOR is highly desired by industry
  - However, EOR in **shale is far more challenging** than conventional formations due to their extreme low permeability and mixed wettability
- **Approach:**
  - **Determine how CO<sub>2</sub> and in surfactants dissolved in CO<sub>2</sub> can be used to increase EOR** by simulating subsurface EOR conditions in the laboratory
    - Surfactants – identify CO<sub>2</sub>-soluble surfactants to change wetting properties
    - Contact angle – observe change from oil-wet to water-wet
    - Confined Huff n' Puff core floods – relate to field tests
- **Value:**
  - Successful EOR in shales would lead to tremendous increases in domestic oil production

# Characterizing Application of CO<sub>2</sub> as a Recovery Agent to Mobilize Hydrocarbons from Shale

## Analysis of prior efforts for enhanced oil recovery from shales

- Critical review developed from literature study which defined laboratory R&D needs for EOR

**Laboratory-based** confined huff n' puff tests to relate to the field and are a primary focus of this project moving forward.

**energy&fuels**

"A Literature Review of CO<sub>2</sub>, Natural Gas, and Water-Based Fluids for Enhanced Oil Recovery in Unconventional Reservoirs"

*Energy & Fuels* **2020** 34 (5), 5331-5380

DOI: 10.1021/acs.energyfuels.9b03658

## Findings:

- CO<sub>2</sub> and natural gas are promising fluids for huff 'n puff EOR
- CO<sub>2</sub> EOR shale is a complex process that involves many mechanisms, especially miscibility and diffusion
- High pressure CO<sub>2</sub> and natural gas will recover much more oil than water. However, interest persists in the lower cost, water-based EOR
- CO<sub>2</sub> EOR reduces the carbon intensity of the oil produced by associated CO<sub>2</sub> storage
- Field cores "from depth" and reservoir crude oil (rather than outcrop cores and synthetic crude oil) are needed to improve the reliability of laboratory-scale results

# Experimental approach: CO<sub>2</sub> EOR using shale cores

**Oil-saturated cores**  
Taken from oil-producing shales, at depth. Weigh cores, no cleaning



**Extraction experiments**  
Monitor weight of hydrocarbons extracted



**Grind core to powder**  
Extract oil with methylene chloride/acetone



## Experimental conditions:

- Confined huff n' puff
- Bathing huff 'n puff
- HPHT Contact angle measurements

## Shale samples:

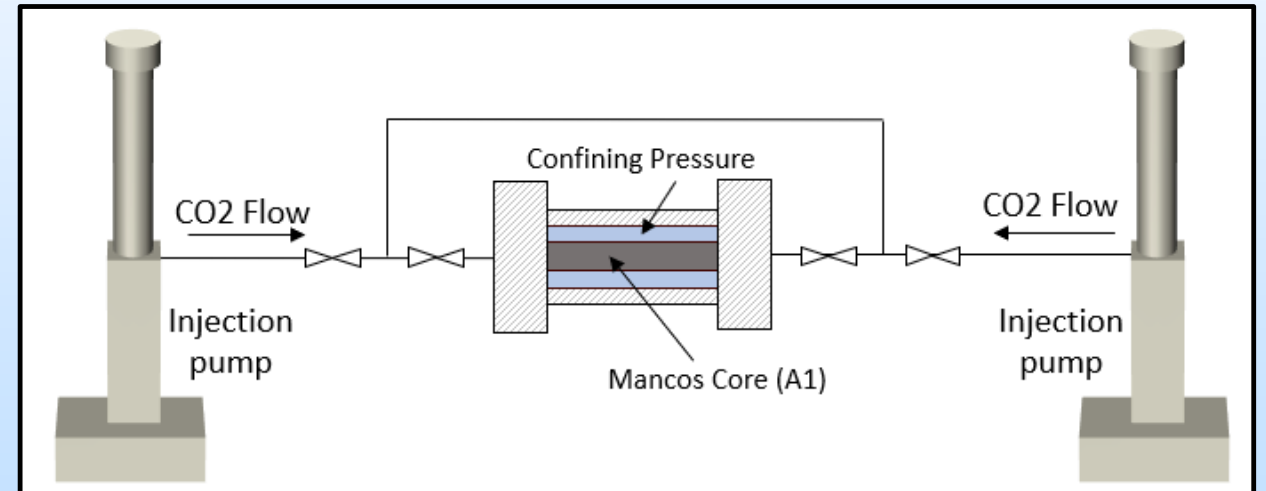
- Eagle Ford,, Mancos, Bakken, Wolfcamp

## Oil:

- Eagle Ford, Bakken, Wolfcamp Live Oil

## Partner for samples:

- HFTS Project (Wolfcamp)



Confined cores to better model field conditions using  
NETL's core flow apparatus

- ✓ Milestone 9D. 06/2019 Obtain shale samples for future CO<sub>2</sub> hydrocarbon extraction tests
- ✓ Milestone 9F. 12/2019 Quantify hydrocarbon oil from shale



# CO<sub>2</sub> huff 'n puff for EOR in unconventional formations

## Oil Recovery Mechanisms

CO<sub>2</sub> extraction of oil

CO<sub>2</sub> diffusion into oil

Oil diffusion into CO<sub>2</sub>

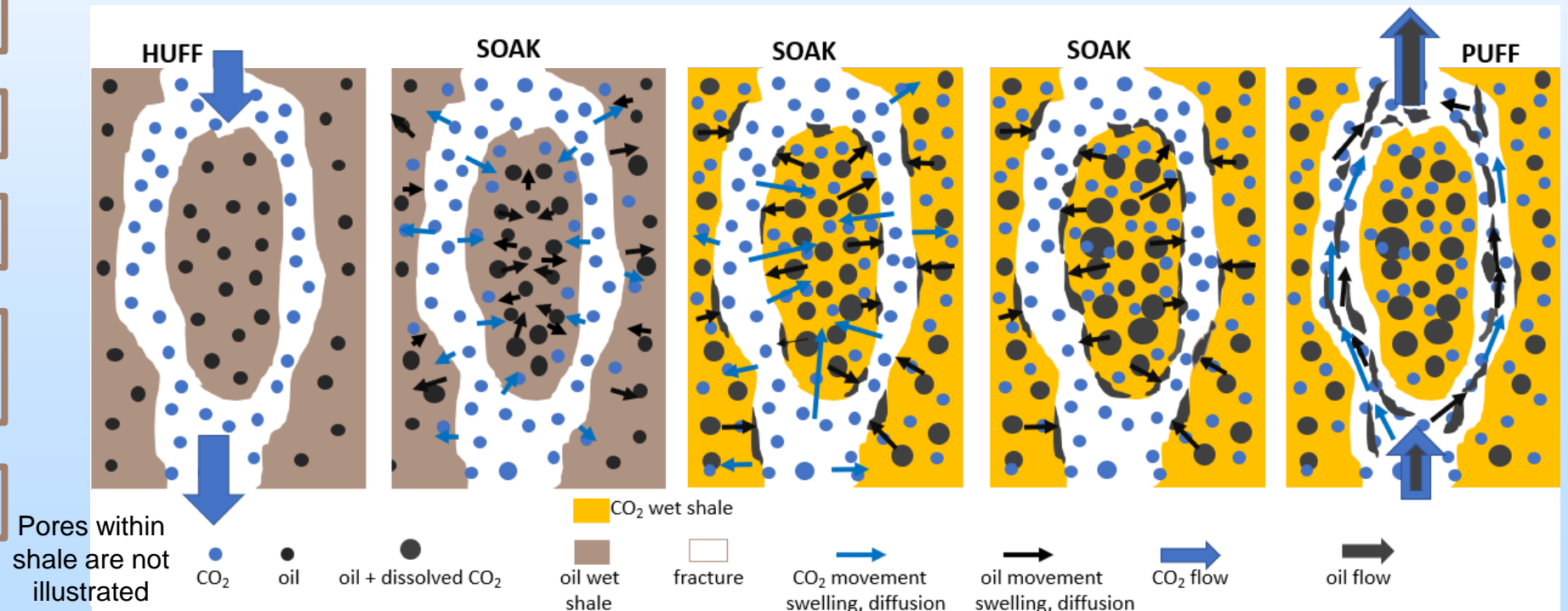
Oil swelling

Oil viscosity reduction

Solution gas drive

## New mechanism

Wettability alteration during soaking  
due to the dissolution of nonionic surfactants in the CO<sub>2</sub>



# Why nonionic surfactants in CO<sub>2</sub>

CO<sub>2</sub> is a **solvent** for nonionic surfactants

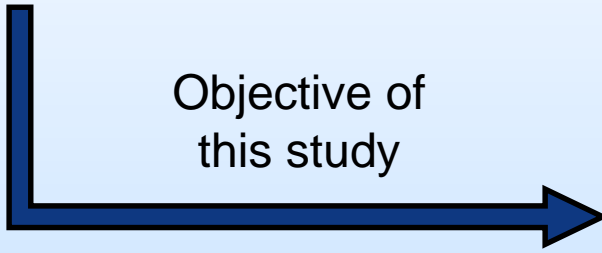
Long-term application in conventional formations



**In-situ generation of CO<sub>2</sub>-in-water mobility control foam as the surfactant partitions into the in situ brine**

to improve sweep efficiency

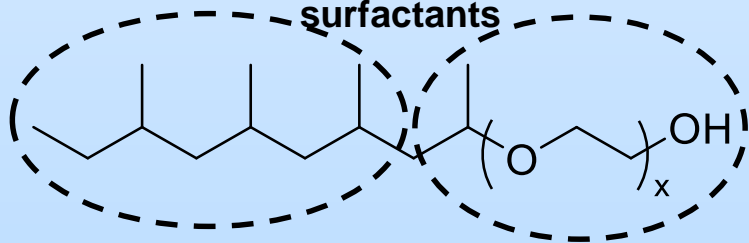
Objective of this study



**Wettability alteration**

toward more oil-phobic and CO<sub>2</sub>-philic

Basic structure of nonionic surfactants



Oleophilic segment

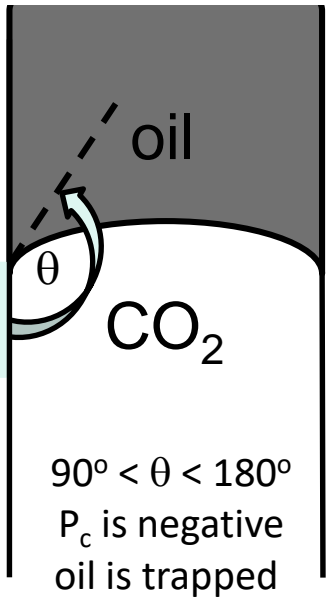
Oleophobic segment

- ✓ To combine the advantages of low viscosity CO<sub>2</sub> with the IFT and wettability-altering capabilities of surfactants in a single phase
- ✓ Inexpensive and commercially available
- ✓ Many options, can be oil-soluble or water-soluble
- ✓ Even **low surfactant** solubility (0.1-1.0 wt.%) in high pressure CO<sub>2</sub> may be more than enough for EOR

# Surfactants added to CO<sub>2</sub>

Potential wettability alteration during CO<sub>2</sub> fracturing and CO<sub>2</sub>-EOR

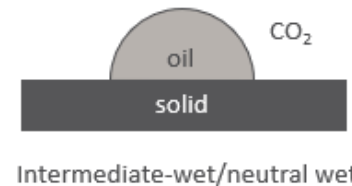
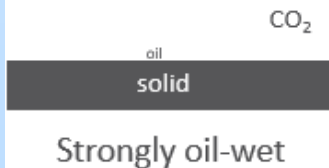
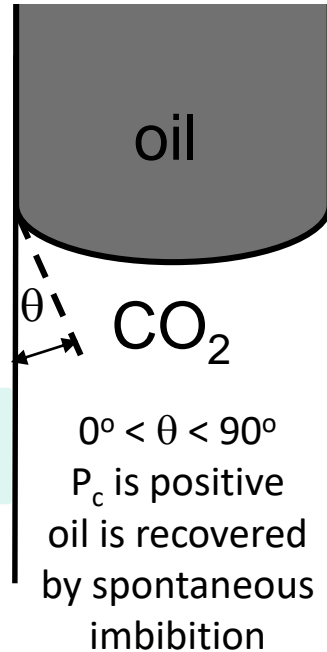
## oil-wet pore



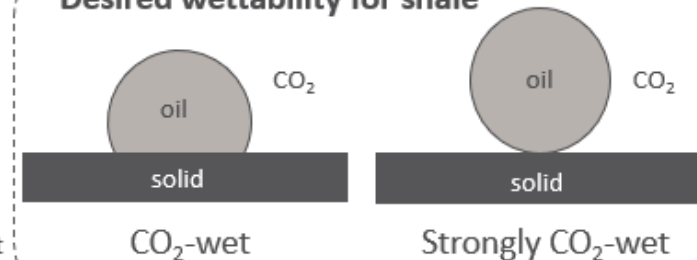
Wettability  
alteration due  
to adsorption  
of CO<sub>2</sub>-soluble  
surfactant



## CO<sub>2</sub>-wet pore



### Desired wettability for shale



## Large positive $P_c$

Surfactant needs to make the surface as CO<sub>2</sub>-wet as possible while reducing the IFT by as little as possible

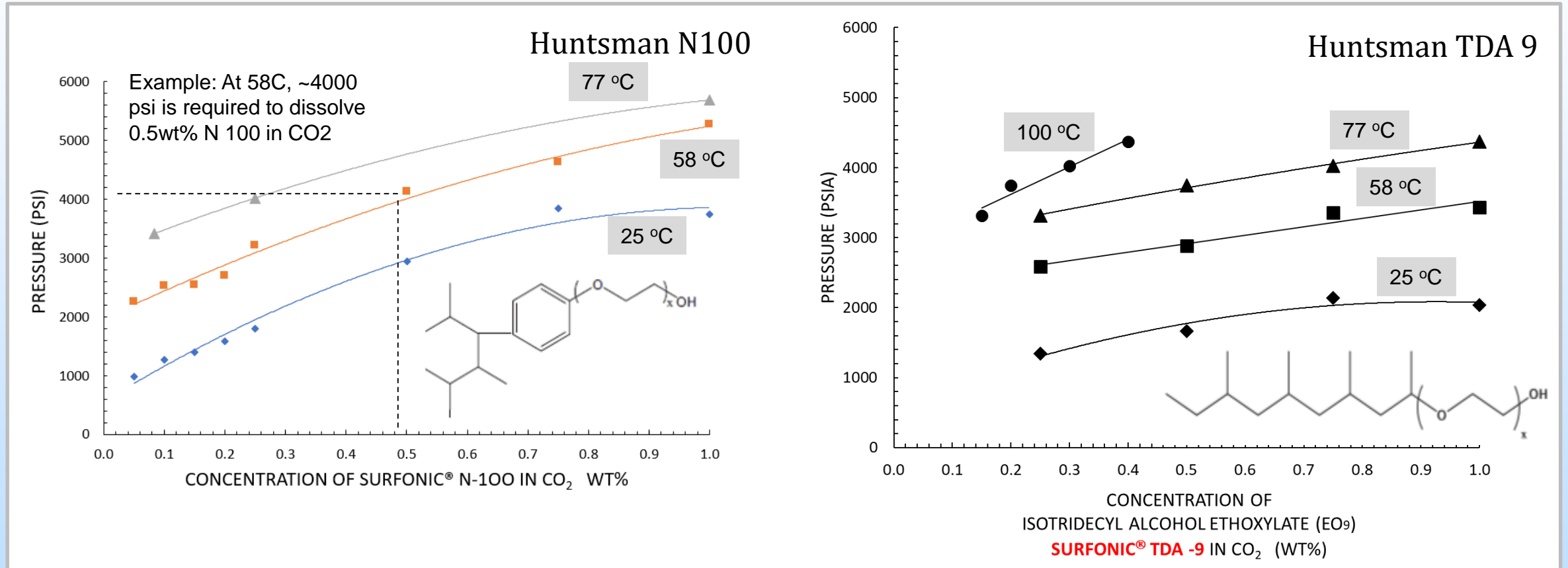
## Risk

With ultralow IFT  
wettability alteration may not have a significant effect on improving the displacement of oil

# Identification of CO<sub>2</sub>-Soluble Surfactants

Two water-soluble, nonionic ethoxylated alcohols were selected for this study.

Huntsman N100, a branched nonylphenol ethoxylate with an average of 10 EO groups (left, average  $x = 10$ ) and Huntsman TDA 9, a branched ethoxylated tridecylalcohol with an average of 9 EO groups (right, average  $x = 9$ ).



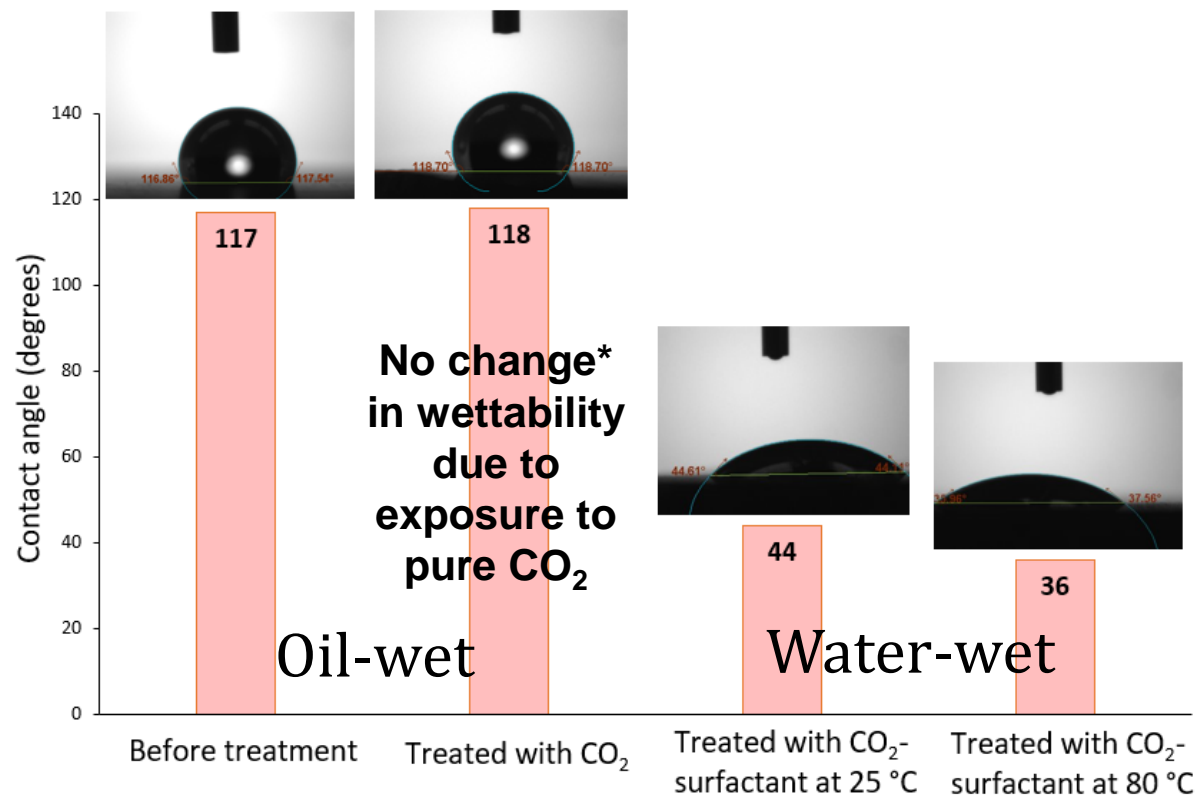
- ✓ Milestone 9I. 03/2020 Generate surfactant solubility in CO<sub>2</sub> data for one surfactant at a low temperature and compare with literature data.



# Contact angle measurements (Wettability)

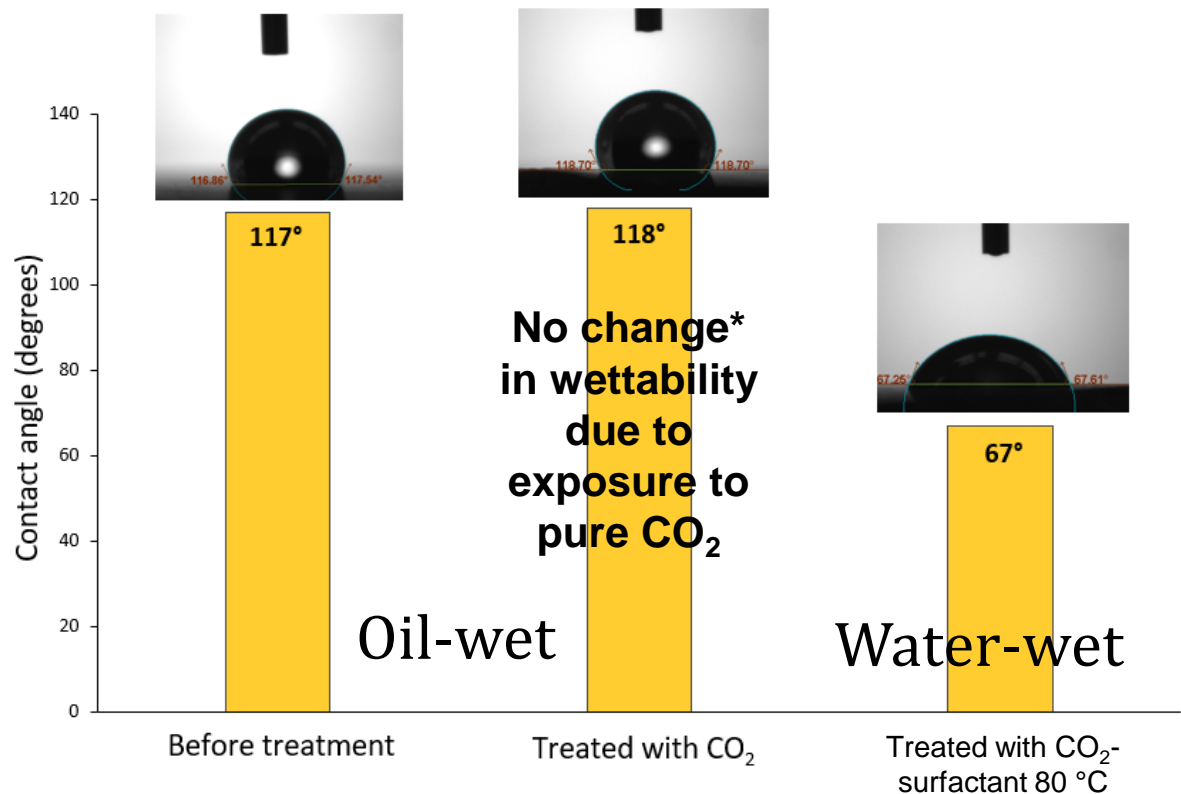
✓ Milestone 9.H 03/2020 Complete shakedown of contact angle apparatus, in preparation for measurement of the wetting properties of shale exposed to CO<sub>2</sub>

Eagle Ford Shale



Huntsman N100

Eagle Ford Shale

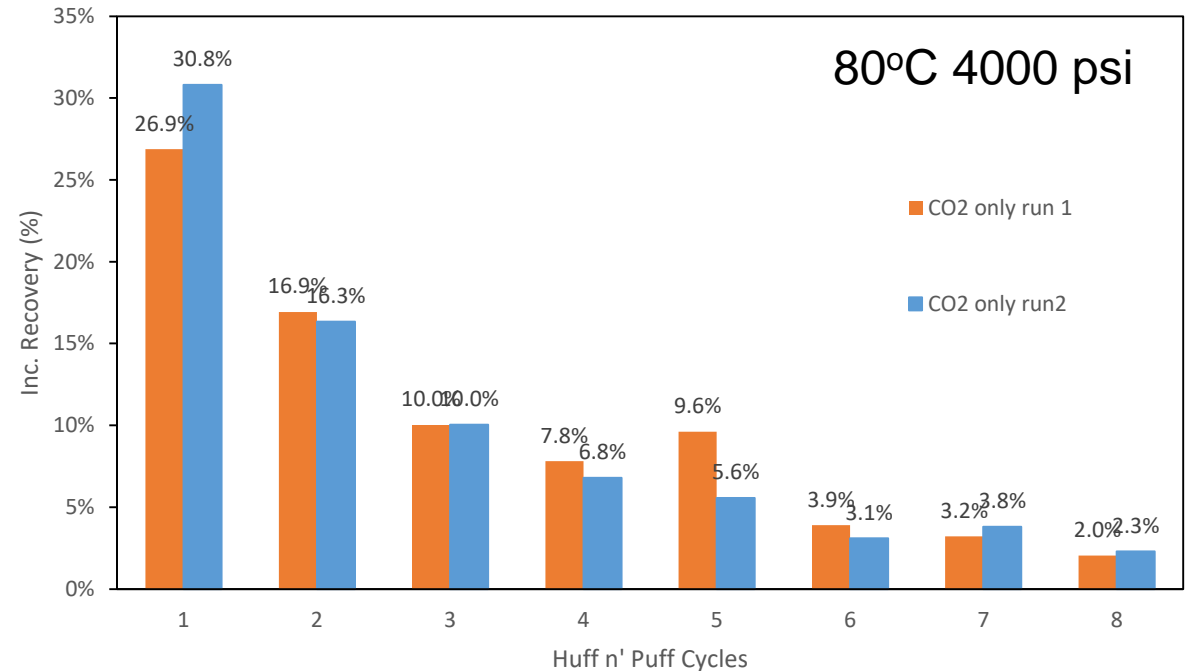
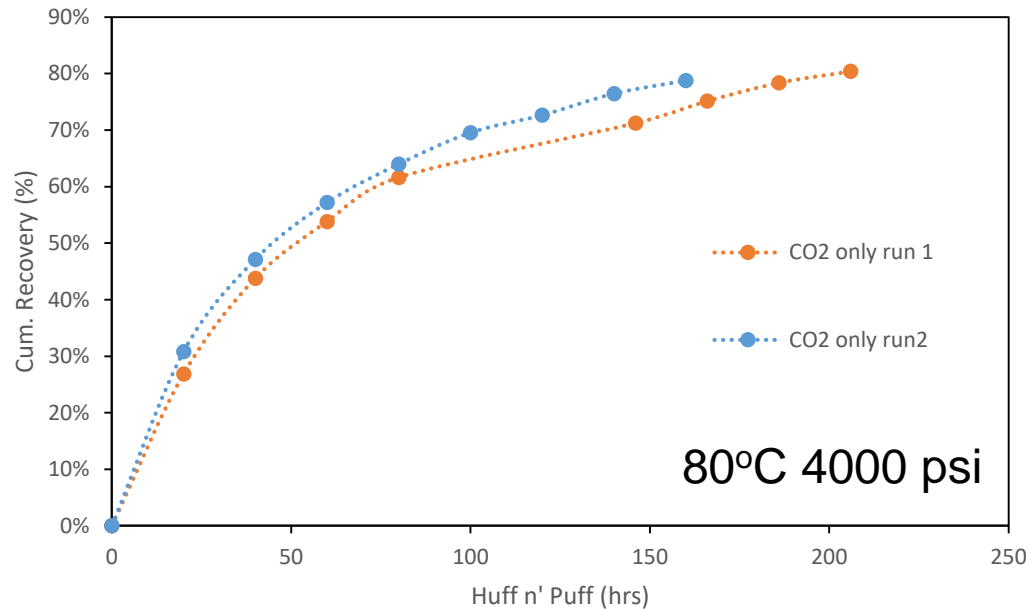


Huntsman TDA 9

\* Note: a prior study did observe a shift toward water-wet for samples exposed to pure CO<sub>2</sub>. Alharthy, N., Teklu, T., Kazemi, H. et al. 2015. Enhanced Oil Recovery in Liquid-Rich Shale Reservoirs: Laboratory to Field. Presented at the SPE Annual Technical Conference and Exhibition, Houston, Texas, 28 – 30 September. SPE-175034-MS.

# Huff n' Puff Experiments with CO<sub>2</sub>

8 Huff n' Puff Cycles: 79% recovery with pure CO<sub>2</sub>

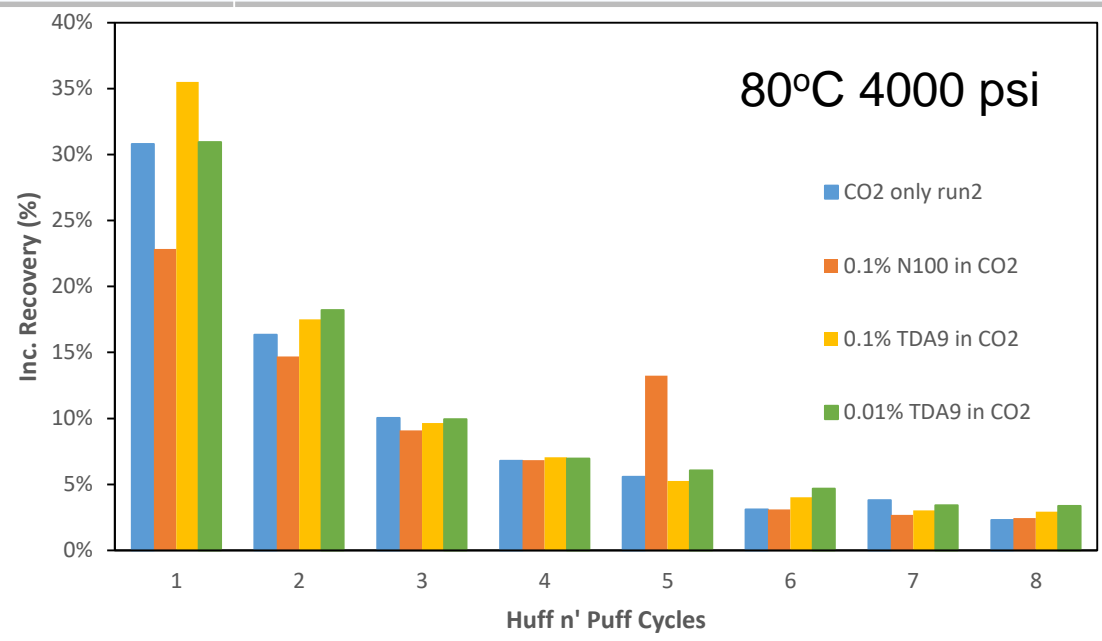
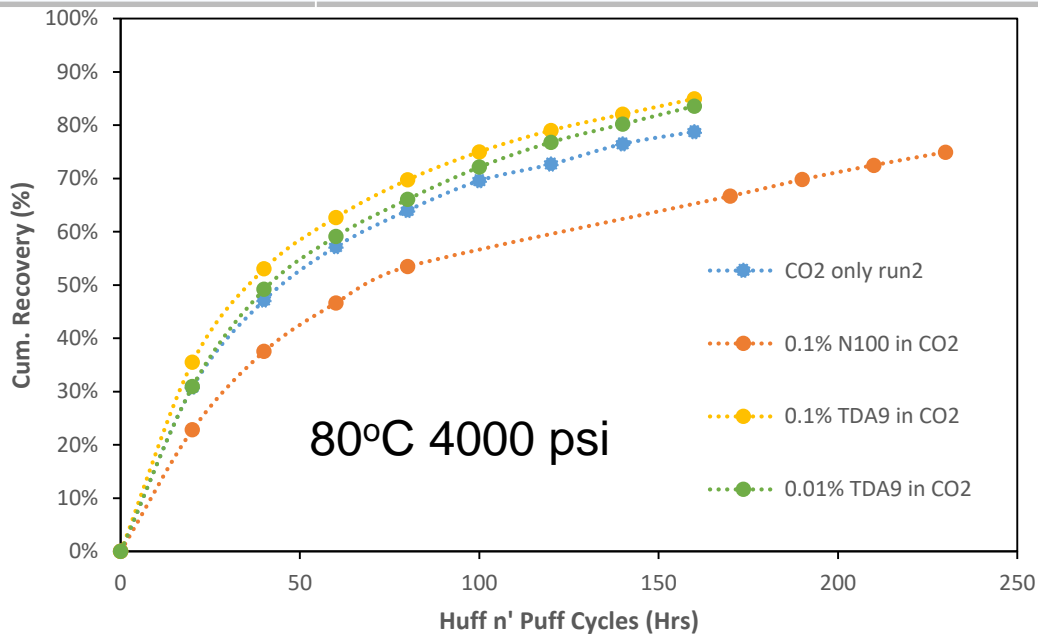


	Core	Length cm	Diameter cm	Bulk Volume cc	Pore Volume cc	Porosity %	Permeability μD	Dry Weight g	Soaked Weight g	oil in place g
CO <sub>2</sub> run 1	Eagleford	5.076	2.552	25.95	1.96	7.55	5-15	56.45	58.78	2.33
CO <sub>2</sub> run 2	Eagleford	5.022	2.555	25.74	1.69	6.56	5-15	56.12	58.26	2.15

✓ Milestone 9.C 06/2020 Complete shakedown of continuous core flooding apparatus, in preparation for hydrocarbon extraction from tight and shale cores using supercritical CO<sub>2</sub>

# Huff n' Puff Experiments with CO<sub>2</sub> and Surfactant

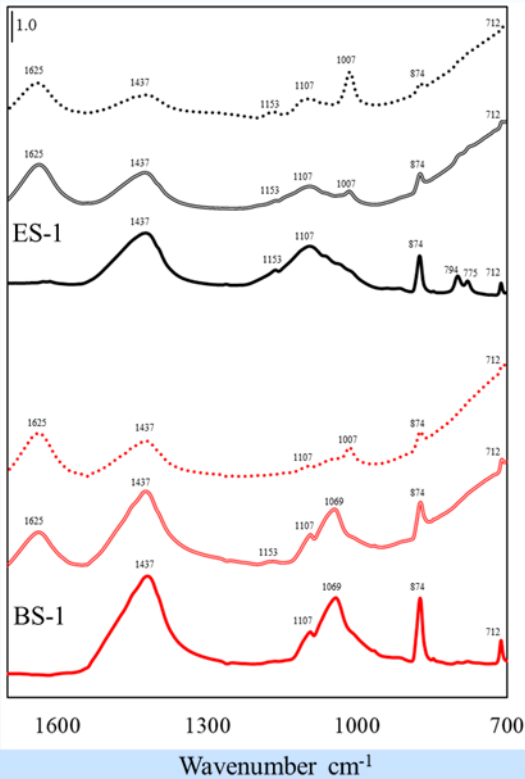
- 8 Huff n' Puff Cycles:
- 79% recovery with pure CO<sub>2</sub>
  - 85% recovery with surfactant (TDA9) dissolved in CO<sub>2</sub>
  - 75% recovery with surfactant (N100) dissolved in CO<sub>2</sub>



	Core	Length cm	Diameter cm	Bulk Volume cc	Pore Volume cc	Porosity %	Permeability μD	Dry Weight g	Soaked Weight g	oil in place g
CO2 run 2	Eagleford	5.022	2.555	25.74	1.69	6.56	5-15	56.12	58.26	2.15
0.1% TDA9 in CO <sub>2</sub>	Eagleford	4.523	2.556	23.20	1.80	7.78	5-15	50.33	52.30	1.97
0.01% TDA9 in CO <sub>2</sub>	Eagleford	4.719	2.556	24.20	1.81	7.48	5-15	52.49	54.56	2.07
0.1% N100 in CO <sub>2</sub>	Eagleford	5.032	2.553	25.75	1.86	7.22	5-15	55.99	58.24	2.24

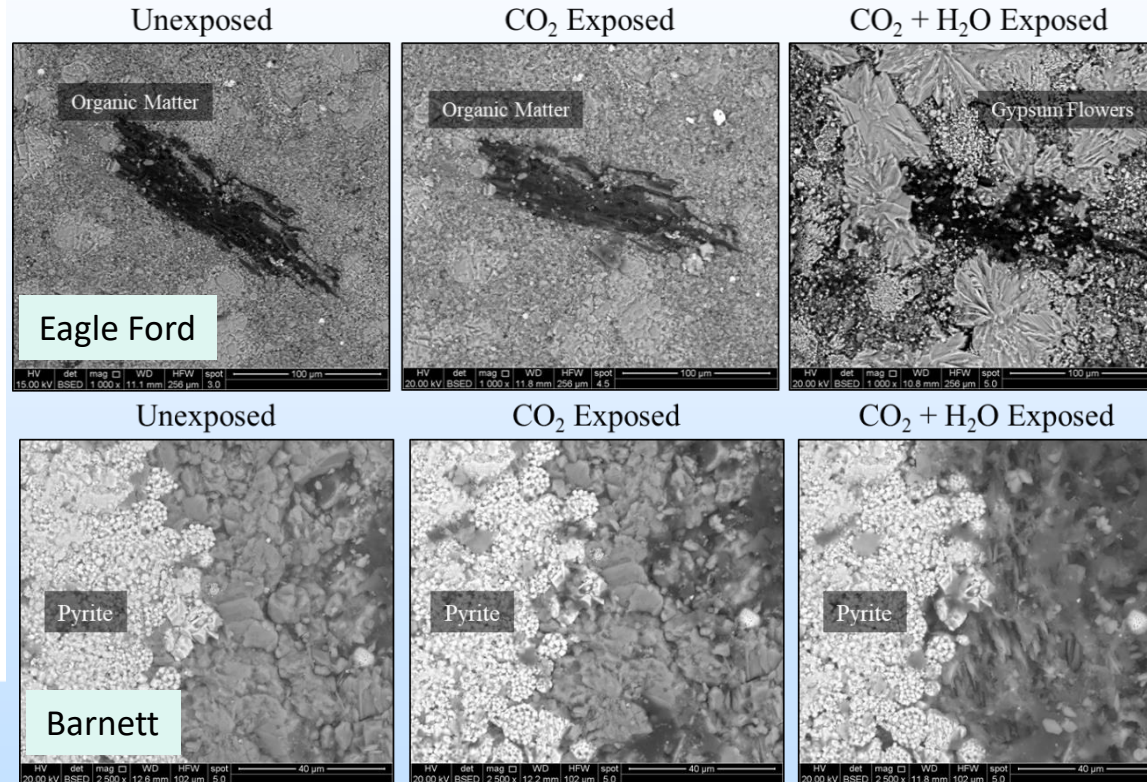
# Physical and chemical alterations of Eagle Ford and Barnett Shale after hydrocarbon extraction with CO<sub>2</sub>

- ✓ Milestone 9G. 03/2020 Identify key physical and chemical alterations for Eagle Ford and Barnett Shales after hydrocarbon extraction with CO<sub>2</sub>.

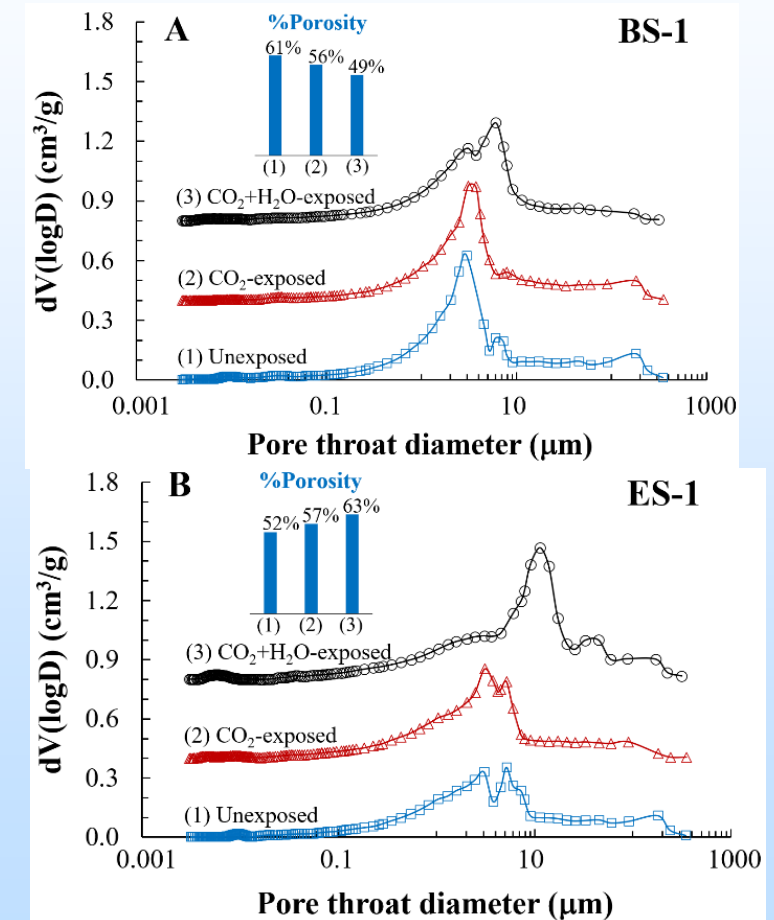


FTIR:

In situ characterization



SEM: Visualization of shale matrix alterations



BET: Pore size distribution changes



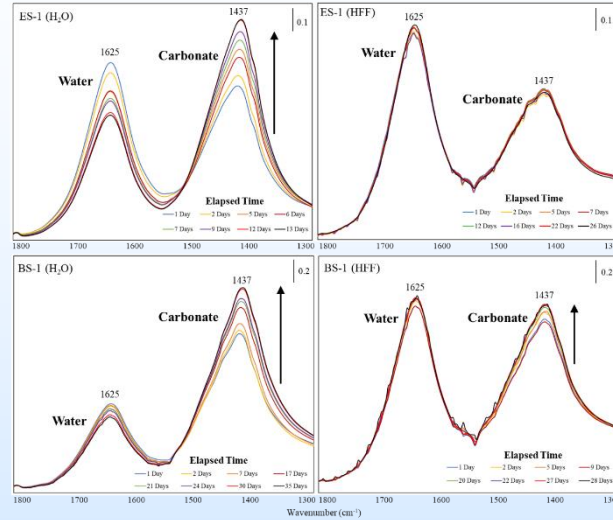
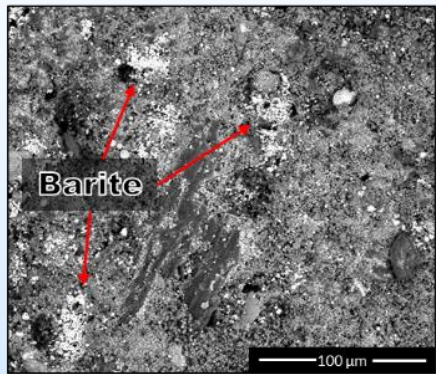
# Physical and chemical alterations of Eagle Ford and Barnett Shale after hydrocarbon extraction with CO<sub>2</sub>

## FTIR: In situ characterization

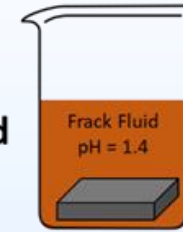
✓ Milestone 9G. 03/2020 Identify key physical and chemical alterations for Eagle Ford and Barnett Shales after hydrocarbon extraction with CO<sub>2</sub>.

### Barite Formation

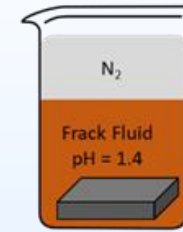
HFF Exposed



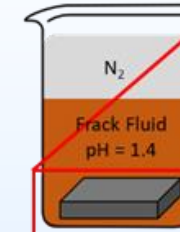
### HFF-Exposed



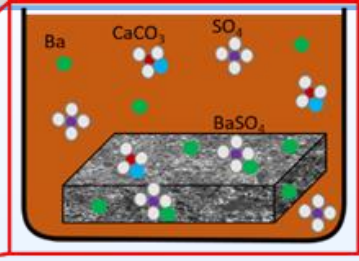
Sample submerged in synthetic frack fluid



N<sub>2</sub> introduced to the system (10.3 MPa)



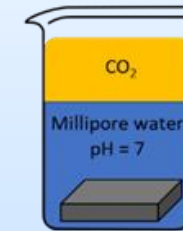
Frack fluid reacts with shale matrix



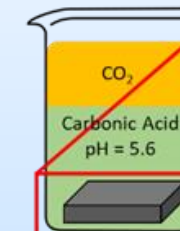
### HFF-Unexposed



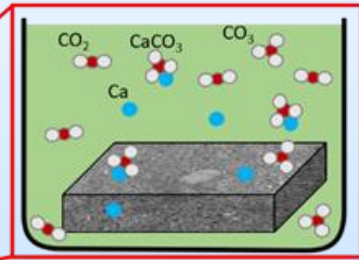
Sample submerged in Millipore water



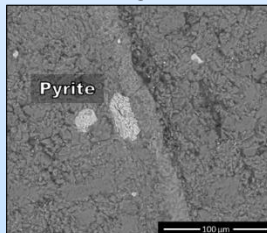
CO<sub>2</sub> introduced to the system (10.3 MPa)



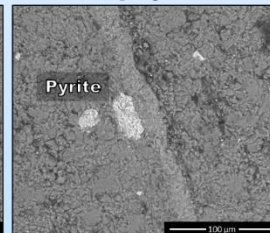
CO<sub>2</sub> dissolves into water to form carbonic acid and reacts with shale matrix



### Unexposed



### CO<sub>2</sub> Exposed

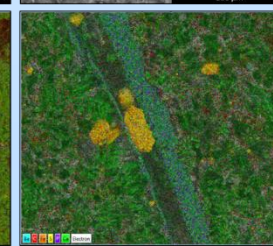
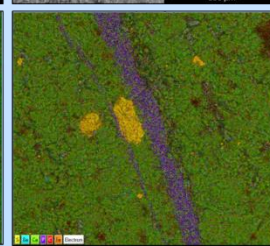
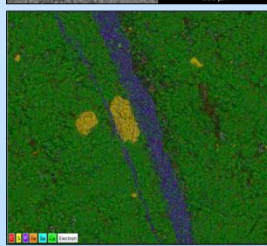
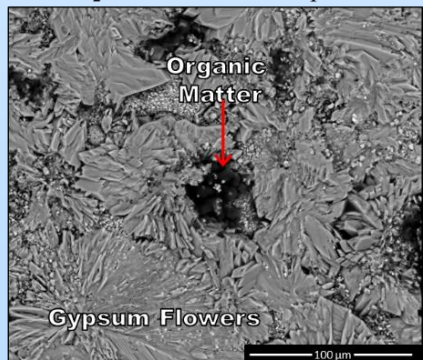


### CO<sub>2</sub>-Saturated Fluid Exposed

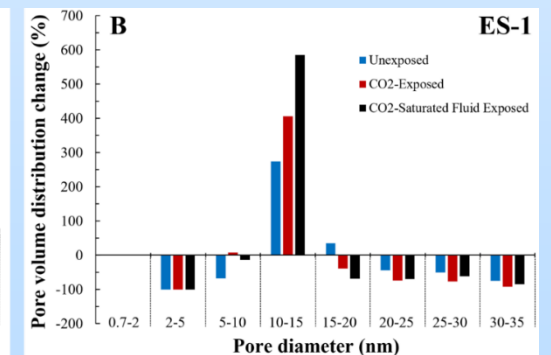
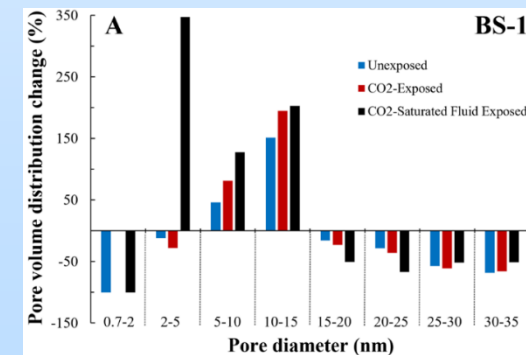


### Gypsum Formation

CO<sub>2</sub>-Saturated Fluid Exposed



SEM: Visualization of shale matrix alterations



BET: Pore size distribution changes



# Technology Transfer

## Published Papers



### A Literature Review of CO<sub>2</sub>, Natural Gas, and Water-Based Fluids for Enhanced Oil Recovery in Unconventional Reservoirs

Lauren C. Burrows, Foad Haeri, Patricia Cvetic, Sean Sanguinito, Fan Shi, Deepak Tapriyal, Angela Goodman, and Robert M. Enick  
*Energy & Fuels* **2020** 34 (5), 5331-5380  
DOI: 10.1021/acs.energyfuels.9b03658

**2019:** Filed **patent application** 62/931,653 “Method of Oil Recovery Using Compositions of Carbon Dioxide and Compounds to Increase Water Wettability of Formations.” Developed and submitted critical literature review to Energy and Fuels.

URTeC: 2774

### Improving CO<sub>2</sub>-EOR In Shale Reservoirs using Dilute Concentrations of Wettability-Altering CO<sub>2</sub>-Soluble Nonionic Surfactants

Foad Haeri<sup>1,2</sup>, Lauren C. Burrows<sup>1,3</sup>, Peter Lemaire<sup>4</sup>, Parth G. Shah<sup>4</sup>, Deepak Tapriyal<sup>1,2</sup>, Robert M. Enick<sup>4</sup>, Dustin M. Crandall<sup>1</sup>, Angela Goodman<sup>1</sup>, 1. National Energy Technology Laboratory, 2. Leidos Research Support Team, 3. Oak Ridge Institute of Science and Education, 4. Dept. of Chemical and Petroleum Eng. University of Pittsburgh.

## Accepted abstracts



- ✓ Milestone 9.E 09/2019 Submit the article, “A Critical Review of Enhanced Oil Recovery in Unconventional Liquid Reservoirs” in a peer-reviewed journal.

# Summary

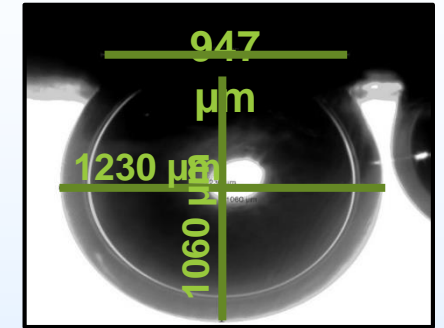
- We are determining how  $\text{CO}_2$  and  $\text{CO}_2$ /surfactant can be used to increase EOR by simulating subsurface EOR conditions in the laboratory by changing wetting
- Successful EOR in shales would lead to tremendous increases in domestic oil production
- Examples of simulated laboratory EOR techniques we are performing include:
  - Confined huff n' puff and Bathing huff n' puff

In progress:

- Currently soaking Wolfcamp in live oil
- Preparing for Huff n' Puff (confined and bathing)
- Comparing oil recovery with  $\text{CO}_2$  and  $\text{CO}_2$  and surfactants (URTEC)
- Soaking cores in fracture fluid or brine prior to oil recovery
- Trying a new surfactant - Surfonic L12-6
- High pressure contact angle experiments with  $\text{CO}_2$  and oil in contact with oil-wet shale.
- High pressure IFT experiments to determine the degree of IFT reduction



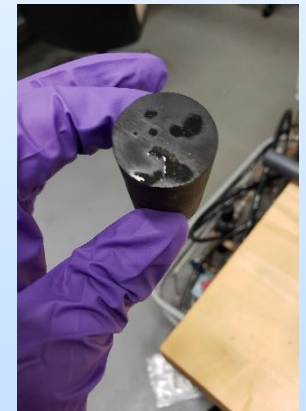
High pressure cell for  
**Bathing Huff n' Puff**



Contact angle



Coreflood setup for confined  
**Huff n' Puff**



Extracted oil

# Appendix

# Organization Chart

- NETL: Angela Goodman, Foad Haeri, Lauren C. Burrows, Deepak Tapriyal, Sean Sanguinito, Dustin Crandall
- University of Pittsburgh: Robert Enick, Parth Shah

# Gantt Chart Task 9 Project Timeline Overview

## Characterizing Application of CO<sub>2</sub> as a Recovery Agent to Mobilize Hydrocarbons from Shale

