

# Electrons to Molecules at NREL: Renewable Chemical Technologies

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### **Downward Trends for Renewable Electricity Costs**

#### Unsubsidized Wind LCOE LCOE LCOE (S/MWh) Wind 9-Year Percentage Decrease: (69%) (\$/MWh) Utility-Scale Solar 9-Year Percentage Decrease: (88%) Wind 9-Year CAGR: (12%) Utility-Scale Solar 9-Year CAGR: (21%) \$250 \$450 minu 5- rear critoris (rizia) 400 \$359 200 350 300 150 250 \$135 \$248 \$124 200 100 \$72 \$71 150 \$70 \$125 \$59 \$55 \$98 100 \$47 50 \$45 \$42 50 0 0 2009 2010 2011 2012 2013 2014 2015 2016 2017 2018 2009 2010 2011 2012 2013 2014 2015 2016 2017 2018 LCOE LCOE Version 3.0 4.0 5.0 6.0 7.0 8.0 9.0 10.0 11.0 12.0 Version 3.0 4.0 5.0 6.0 7.0 8.0 9.0 10.0 11.0 12.0 — Wind LCOE Mean Crystalline Utility-Scale Solar LCOE Mean

#### Unsubsidized Solar PV LCOE

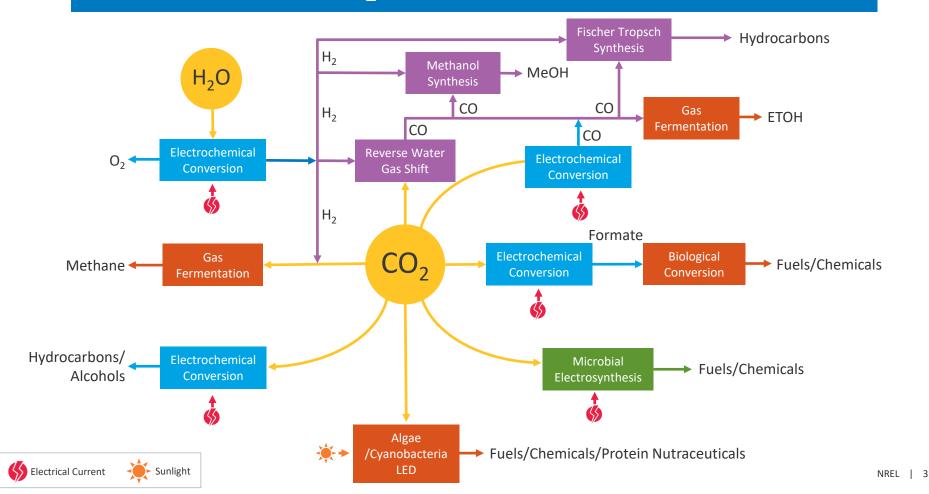
\$79

\$64

\$55 \$50

\$43

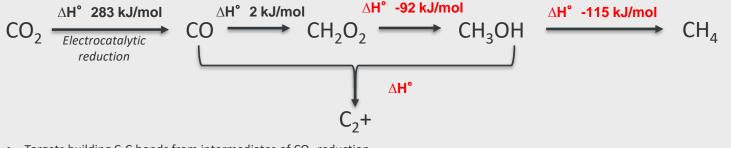
### E2M CO<sub>2</sub> Conversion Pathways



### E2M - CO<sub>2</sub> Reduction and Utilization

$$H_2O \xrightarrow{\Delta H^{\circ} 242 \text{ kJ/mol}}_{Electrolysis} H_2 + 1/2O_2$$

• Leverages Program work from FCTO, BES, etc



- Targets building C-C bonds from intermediates of CO<sub>2</sub> reduction
- Targets novel hybrid approaches (with or without H<sub>2</sub>)

#### Outcome: New concepts, approaches, and

understanding for chemical (carbon-carbon) bond formation using  $CO_2$  and electrons through electro-catalysis, synthetic biology and advanced hybrid processes

### **Potential Impact**

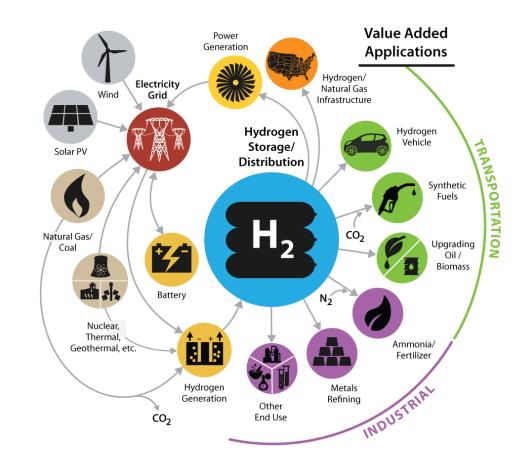
- Promotes CO<sub>2</sub> utilization and valorization
- Provides alternative route to products through low-cost electricity
- Chemical storage option

### Conceptual H2@Scale Energy System

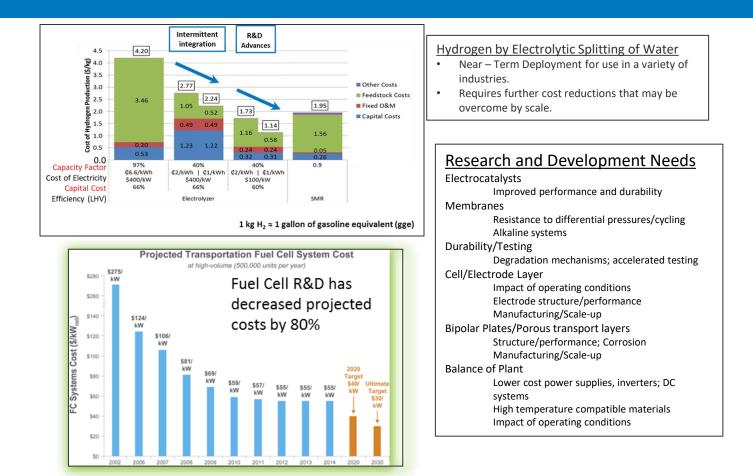
### H2@Scale

- Explores the potential for wide-scale H<sub>2</sub> production
- Enable resiliency of the power generation and transmission sectors
- Aligning with diverse Industries such as metal refining, ammonia, chemicals, and fuels upgrading

H2@Scale website: http://energy.gov/eere/fuelcells/h2-scale



### Pathway to Economical Generation of H<sub>2</sub> by Electrolysis



### **Renewable Methane Production**

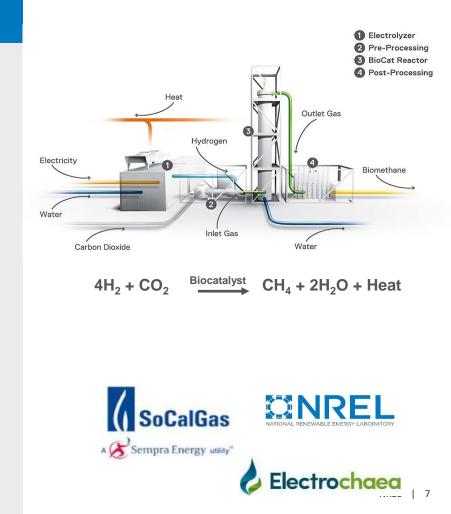
### **Scientific Approach**

- Utilize excess electricity production for the electrolysis of water to H<sub>2</sub> and O<sub>2</sub>
- Optimized strain of methanogenic archaea to perform methanation under industrial conditions
- 98% Carbon efficiency of CO<sub>2</sub> to CH<sub>4</sub>
- Post-processing for pipeline quality natural gas

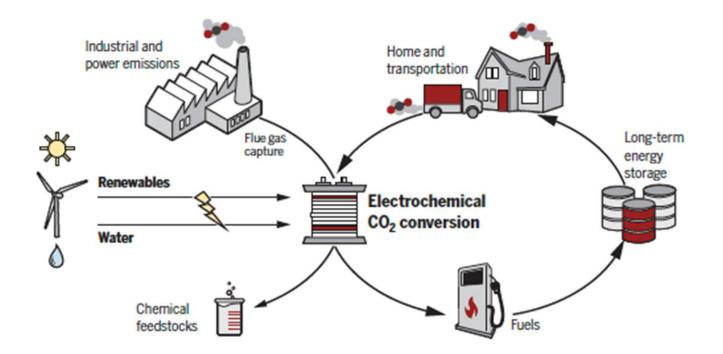
### Significance and Impact

- Potential long term storage strategy via conversion of electricity & CO<sub>2</sub> to CH<sub>4</sub>
- High efficiency CO<sub>2</sub> capture and conversion strategy
- Demonstrated route to renewable methane





### Electrochemical CO2 Reduction: Where are we now?



### NREL's Multi-Scale Electrochemical Capabilities

#### Rotating Disk Electrode



Flow – Through Single Cell



Watts

### Flow – Through Stacked System



KiloWatts

MilliWatts

### World Class Test Stations



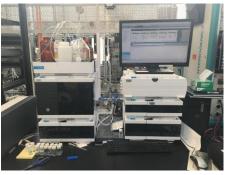
### **Current Capabilities**

- Anode and cathode can flow liquid (0-100 mL/min) or gas (0-4 SLPM)
- In-line automated gas sampling (two Agilent 490 MicroGCs)
  - $H_2$  , CO, CH<sub>4</sub>, C<sub>2</sub>H<sub>4</sub>, CO<sub>2</sub>
- HPLC with autosampler for liquid product analysis
- PEEK-PTFE backpressure (0-60 psig) regulators optimized for dual-phase flow
- Ambient to 85 °C operation
- Safety N<sub>2</sub> purge
- Flammable gas leak detection
- Enclosure ventilation exceeds NREL standard for chemical fume-hoods

### GC with sampling manifold



HPLC with autosampler

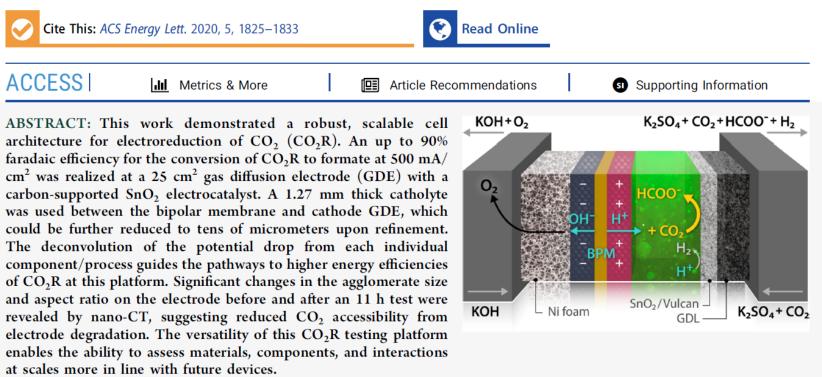


### **Future Plans**

- Four additional stations designed for maximum modularity and ease of and maintenance/modifications
- Time-of-Flight Mass-Spectrometer
  - Real time product analysis

### Formate: Identifying Pathways to Higher Energy Efficiencies

Yingying Chen, Ashlee Vise, W. Ellis Klein, Firat C. Cetinbas, Deborah J. Myers, Wilson A. Smith, Todd G. Deutsch, and K. C. Neyerlin\*



#### Scientific Approach

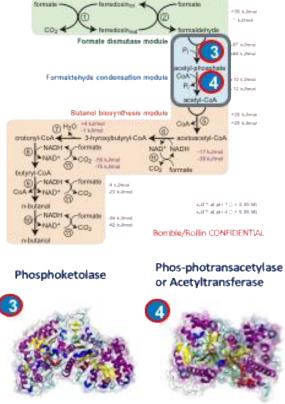
- Cell free approaches can be directly coupled to electrocatalytic production of formic acid from CO<sub>2</sub>. The same electrolyser can be used to recycle CO<sub>2</sub> formed in enzymatic reactions.
- We can leverage enzyme promiscuity to convert formaldehyde to essential intermediates such as acetyl-phosphate and acetyl-CoA
- Focus on Formaldehyde condensation and Butanol biosynthesis modules and use rational design to engineer enzymes for increased stability, enhanced selectivity, and formaldehyde tolerance.

#### Significance and Impact

- An estimated 2.85 Gt CO<sub>2</sub> and 300 TWh (\$0.02/kWh) are available for utilization each year.
- Developing approaches to convert these feedstocks to liquid fuels or biochemicals using excess electrons will promote efficient energy storage.

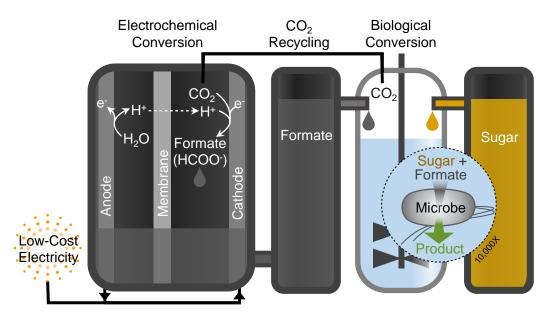
#### Partners

This project will benefit from a collaboration with Global Bioenergies S.A. and Philippe Marliere from Scientist of Fortune.



## Electrochemical reduction of CO<sub>2</sub> to improve sugar conversion

• We envision a system in which CO<sub>2</sub> emitted during biological conversion is recycled by electrochemically reducing it to generate formate.



• Formate can be stored and used as an auxiliary energy source to improve biological conversion for a wide variety of hosts, products and processes.

## Thank you

### www.nrel.gov

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NREL is a national laboratory of the U.S. Department of Energy, Office of Energy Efficiency and Renewable Energy, operated by the Alliance for Sustainable Energy, LLC.

