



HIGH TEMPERATURE GAS SENSOR FOR COAL COMBUSTION SYSTEM

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OUTLINE

- Background & Scientific Approaches
- Project Objectives
- Project Team
- Planed Tasks & Milestones
- Research & Development Progress
- Summary







Background

Structure of a coal-fired power plant



The variation of key combustion parameters with air/fuel ratio







Background







GC/MS, Infrared spectroscopy, Chemiluminescent etc.

SiC-base (Schottky diode) sensors - Silide formation

Physical properties based sensing (mass, dielectric constant, temp, surface stress etc.)

Electro-Chemical Sensors Potentiometric Amperometric





Nicholas F. Szabo, Prabir K. Dutta: Correlation of sensing behavior of mixed potential sensors with chemical and electrochemical properties of electrodes, Solid State Ionics 171 (2004) 183–190



PROJECT OBJECTIVES



- To develop an accurate, robust, high temperature oxygen sensor based on refractory, reliable, catalytically inactive materials capable of monitoring combustion in a coal-fired plant in real time to improve combustion performance;
- (2) To investigate the feasibility and sensitivity of a new catalytic/non-catalytic sensor design to detect "oxidizable" target gases at high temperatures where other electrochemical sensors have failed;
- (3) To test the basic components of the proposed sensor in a commercial, 700 MW power plant.







Project Team Member – Longview Power

Monongalia County,

near Maidsville, WV





• Officially a "zero discharge" power plant in WV

Location

- Includes a new air pollution control system that results in emissions that are Among the lowest in the nation for coal plants.
- Emits less CO2 than most other coal plants because of its <u>fuel efficiency</u>





PLANNED TASKS & MILESTONES

I.D.	Task								
		Q1	Q2	Q3	Q4	Q5	Q6	Q7	Q8
1.0	Project Management								
2.0	Sensor Development								
3.0	Sensor Packaging								
4.0	Lab-scale Sensor Testing								
5.0	Post-mortem Characterization								
6.0	Electrochemical Mechanisms Investigation								
7.0	Sensor Testing in Utility Boiler								

- Task 1.0 Quarterly, annual, and final reports
- Task 2.0 High temperature gas sensor with the temperature capability up to 1300°C
- Task 3.0 Packaging for the sensor developed by LANL in Task 2
- Task 4.0 Library of performance matrix for the sensor in lab-scale power plant simulator
- Task 5.0 Microstructures of high temperature gas sensor after lab-scale testing
- **Task 6.0** Verification of the electrochemical mechanisms of high temperature gas sensing on maximum reading and temperature-proportional signal.
- **Task 7.0** Library of performance matrix for the sensor in utility boiler & microstructures of high temperature gas sensor after testing

DECISION POINTS:

- 1.Q1 Finish PMP
- 2.Q3 Sensing ability (lab) <=800ppm CO concentration in a Po_2 range of 0.5-2% @ 1000C 3.Q7 Sensing ability (lab) <=400ppm CO concentration in a Po_2 range of 1-3% @ 1000C



Current Mixed Potential Sensors



CO electrochemical oxidation kinetics $i_{CO} = 2FAD_{CO} \frac{C_{CO}}{d}$

Oxygen reduction kinetics

$$\dot{i}_{O_2} = \dot{i}_{O_2}^o \frac{4F}{RT} (Em - E_{O_2})$$

Heterogeneous catalysis decrease CO available for electrochemical oxidation



Fernando H. Garzon, Rangachary Mukundan, <u>Eric L. Brosha</u>: Solid-state mixed potential gas sensors: theory, experiments and challenges, Solid State Ionics 136–137 (2000) 633–638



Sensor Development







Determine reducing gas composition in a background of oxygen Working principle

(1)

(2)

(3)

 $O'_{2} + 4e^{-} \Leftrightarrow 2O^{=}$ $CO' + O^{=} \Leftrightarrow CO'_{2} + 2e^{-}$ $CO' + \frac{1}{2}O_{2} \Leftrightarrow CO'_{2}$

Current Mixed-potential sensors

- Mixed potential sensors
- $T_{op} < 600 \text{ °C}$
- High sensitivity to CO/HCs/NOx
- High durability
- Dense electrodes/Porous electrolyte



- CO electrochemical oxidation
- CO heterogeneous oxidation

Proposed HT sensors

- Oxygen (Free vs Equilibrium)
- T_{op} up to 1500 °C
- Higher sensitivity as T↑ and P₀₂
- High durability
 - One dense and one porous electrode



Sensor Development



Schematic illustration of the lab-scale testing for the sensor sample, two-chamber testing





Sensor Testing – Lab-Scale

<u>3rd decision point – Demonstrate</u> Sensing ability (lab) <=400ppm CO concentration in a Po₂ range of 1-3% @ 1000C



Sensor Testing – Lab-Scale



Selectivity tests for $CO_2 CH_4$ and steam

- Insensitive to CO₂
 - CH₄ increases the signal oscillation, however doesn't change the medium value in each CO content.
- H_2O increases the sensitivity to CO in the low concentration range



Effect of Gas Transport and Electrode Area



- Increased gas transport increases the sensitivity to CO
- Enlarged electrode area is beneficial to a increased signal. However, it's more related to the distribution of gas relative to the sensor electrode surface.















- Temperature: 950-975 °C
- Although testing 20 hours, only 1st hour data is valid, because of the connection problem
- FluePt-RefPt signal is -47 mV means 2.5%-3.5% O₂
- Sensing sensor-RefPt signal is 20 mV means CO content is 2000 ppm according to polynomial fitting





Images of the sensor after 20 h operation in the Longview's utility boiler.

- Dusts and ashes already covered the sensor top after 20-h field test
- The packaging method is good because the sensor is not broken and the sealing is good.



2nd trial for 25 days



- 2-week data was lost because of the accidental shutdown of the power at that floor in Longview
- After 25-Day test, the connection of electrodes were broken. Durability of sensor itself is good





- Dusts and ashes fully covered the sensor top after 25-day field test. That layer was very stiff and hard.
- The dusts are very aggressive because the Pt wires were already corroded.
- The packaging method is good because the sensor pellet is not broken and the sealing is good.



- 3rd trial with a modified packaging.
 - Add porous refractory bricks to prevent the dusts and ashes contaminating the sensor.
 - Inner chamber is made of stainless steel as well, for a better robustness.

		Observation port
<image/>	stainless steel tube	Stainless steel

2 weeks 930-950 C



Sensing electrode vs. RefPt

Sensing electrode vs. RefPt



• The huge oscillation is due to the violent atmosphere change in the sensor chamber as the flue ash started to build up on the sensor shield.

We suspect that the ashes and dusts blocked the porous structure of the porous refractory bricks eventually, preventing the gas exchange between the boiler and the sensor chamber.



SUMMARY & FUTURE WORK

Major Progress To-Date

- Developed/Identified a promising sensing material
- Clearly met the 3rd Go/No-Go Target
- TRL-4
- Successful installation of the sensor test station in Longview boiler and obtained a good preliminary data and proved the durability

Future Work

- Mechanisms Investigations
- Packaging Development for a better performance in field test in Longview boiler



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