

# **Development & Validation of Low-Cost, Highly-Durable, Spinel-Based Materials for SOFC Cathode-Side Contact**

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Presentation, 2020***

# Outline

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- **Introduction and Project Objectives**
- **Performance Evaluation of the Sintered Spinel Contact Thermally Converted from Pre-alloyed Precursors**
  - Area Specific Resistance (ASR), Chemical Compatibility, etc.
- **Initial Study on Reaction Layer Formation Kinetics/ Mechanism**
- **Reactive Sintering of Dense  $(\text{Mn,Co})_3\text{O}_4$  Coatings**
- **Co-sintering of Spinel-Based Coating/Contact Dual-Layer Structure between Full-sized Ferritic Alloy and LSM Plates**
- **Concluding Remarks**
- **Acknowledgments**

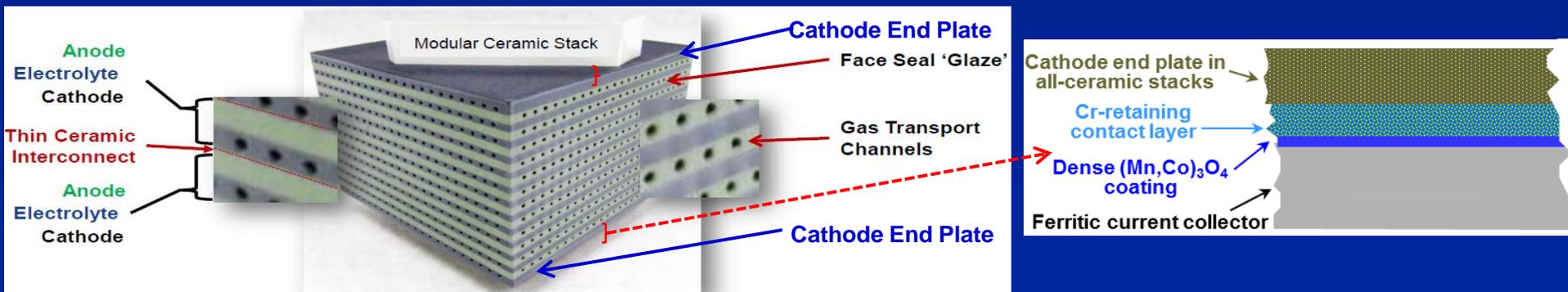
# Need of Contacting for Different SOFC Stacks

- In stacks with anode-supported cells (ASC-SOFC), the contact is required to minimize the cathode-interconnect interfacial resistance.



## Cathode-Interconnect Interface in ASC SOFC Stacks

- In all-ceramic stacks, the contact is required to minimize the interfacial resistance between the current collector plate and cathode end plate.

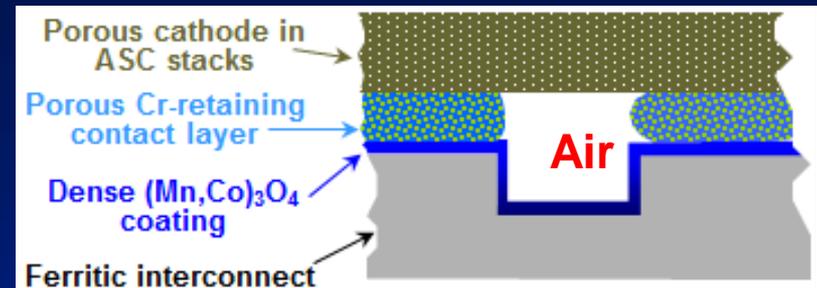


## Cathode-Current Collector Interface in All-Ceramic SOFC Stacks

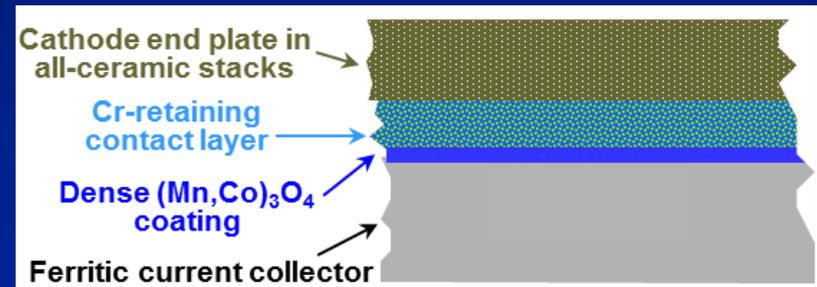
# Contact Material Requirements

- Requirements for contact materials in ASC-SOFC and all-ceramic SOFC stacks are generally similar, including:

- Low material/processing cost
- High electrical conductivity
- Match in coefficient of thermal expansion (CTE)
- Adequate stability and compatibility
- Appropriate sinterability
- Good bonding strength with adjacent stack components
- Absence of volatile species



*Cathode-Interconnect Interface in ASC-SOFC Stacks*



*Cathode-Current Collector Interface in All-Ceramic SOFC Stacks*

- Additionally, a reasonable porosity level in the cathode-side contact is needed in ASC stacks for maximizing the triple phase boundaries for cathodic reaction.

# Different Contact Materials

- While various materials for ferritic alloy-cathode contacting have been studied, most developments have focused on  $(\text{La}, \text{Sr})(\text{Mn}, \text{Co}, \text{Fe}, \text{Ni}, \text{Cu})\text{O}_3$ :
  - Difficulty in balancing the electrical conductivity, CTE, sinterability and chemical compatibility of the perovskites.

Material Type	Example	CTE ( $\times 10^{-6} / \text{K}$ ) (20–800°C)	Conductivity ( $\text{S}\cdot\text{cm}^{-1}$ , 800°C)	Main Concern
Noble Metal	Pt	10.0	Metallic	High Cost
	Pd	12.3	Metallic	High Cost
	Au	16.6	Metallic	High Cost
	Ag	22.0	Metallic	Volatility
Perovskite	$(\text{La}_{0.8}\text{Sr}_{0.2})\text{CoO}_{3-\delta}$	19.2 (20-1000°C)	1400	CTE Mismatch
	$(\text{La}_{0.8}\text{Sr}_{0.2})(\text{Co}_{0.5}\text{Fe}_{0.5})\text{O}_{3-\delta}$	18.3 (20-1000°C)	340	CTE Mismatch
	$(\text{La}_{0.8}\text{Sr}_{0.2})(\text{Co}_{0.5}\text{Mn}_{0.5})\text{O}_{3-\delta}$	15.0 (20-1000°C)	190	CTE Mismatch
	$(\text{La}_{0.8}\text{Sr}_{0.2})\text{MnO}_3$	11.7 (20-1000°C)	170	Sinterability
	$\text{LaMn}_{0.45}\text{Co}_{0.35}\text{Cu}_{0.2}\text{O}_3$	13.9	80	Mn/Cu Migration
Spinel	$\text{MnCo}_2\text{O}_4$	9.7-14.4	24- 89	Sinterability
	$\text{Mn}_{1.5}\text{Co}_{1.5}\text{O}_4$	10.6-11.6	55-68	Sinterability
	$\text{NiCo}_2\text{O}_4$	12.1	0.93	Sinterability
	$\text{NiFe}_2\text{O}_4$	11.8	0.3, 6.8, 17.1	Sinterability
	$\text{Ni}_{0.85}\text{Fe}_{2.15}\text{O}_4$	12.1	15.4	Sinterability

# Why (Ni,Fe)<sub>3</sub>O<sub>4</sub>- and (Mn,Co)<sub>3</sub>O<sub>4</sub>-Based Spinel as Contact Material?

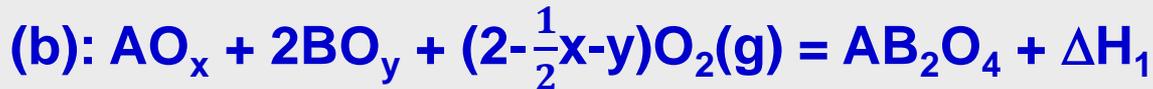
- Conductive spinels based on (Ni,Fe)<sub>3</sub>O<sub>4</sub> and (Mn,Co)<sub>3</sub>O<sub>4</sub>, which have been extensively evaluated as interconnect coating, are also promising for contact application, based on electrical conductivity, CTE, chemical compatibility, etc.

Material Type	Example	CTE ( $\times 10^{-6}$ /K) (20–800°C)	Conductivity (S·cm <sup>-1</sup> , 800°C)	Main Concern
Spinel	MnCo <sub>2</sub> O <sub>4</sub>	9.7-14.4	24- 89	Sinterability
	Mn <sub>1.5</sub> Co <sub>1.5</sub> O <sub>4</sub>	10.6-11.6	55-68	Sinterability
	NiCo <sub>2</sub> O <sub>4</sub>	12.1	0.93	Sinterability
	NiFe <sub>2</sub> O <sub>4</sub>	11.8	0.3, 6.8, 17.1	Sinterability
	Ni <sub>0.85</sub> Fe <sub>2.15</sub> O <sub>4</sub>	12.1	15.4	Sinterability

- Unfortunately, the sinterability of spinels is very poor (typically  $\geq 1000^\circ\text{C}$ ), if metal oxides are used as the starting powders.
- Employment of **metallic powders** (instead of oxide powders) as the starting precursor will lower the sintering temperature via a reactive sintering mechanism called **environmentally-assisted reactive sintering (EARS)**.

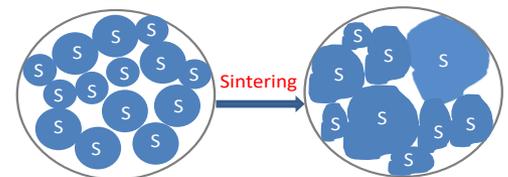
# Utilization of EARS for Reduced-Temperature Sintering of Spinel-Based Contact

- In EARS, with the participation of oxygen from air, the **metallic powder precursor** will be oxidized and reacted to form a well-sintered spinel at a reduced temperature (e.g., 900°C):



$$\Delta H_3 > \Delta H_2 \gg \Delta H_1$$

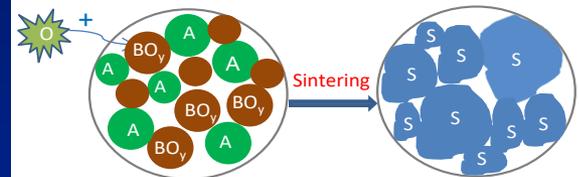
- Enhanced sintering via EARS is likely due to:
  - Heat released during the reaction;
  - Volume expansion upon conversion of metal to metal oxide;
  - Formation of highly-active surface nano-oxides;
  - Shorter diffusion distance when a pre-alloyed powder is employed.



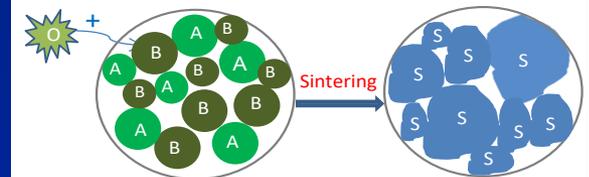
(a) with a spinel (S) powder



(b) with a mixture of metal oxides



(c) with metal and oxide powders



(d) With two metal powders

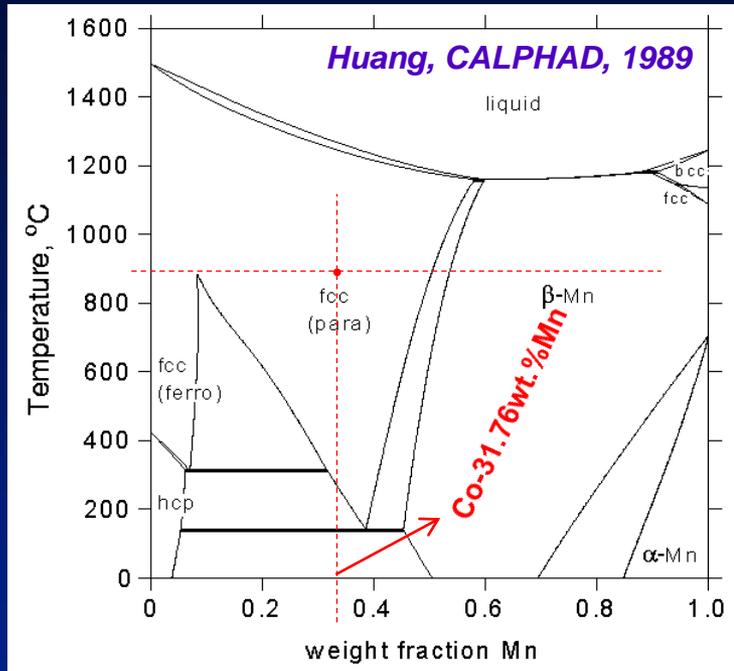


(e) With a pre-alloyed powder

# Project Objectives

- **Optimization of the multi-component alloy precursor composition as contact material.** The alloy compositions will be optimized via composition screening in the  $(\text{Ni,Fe,Co,X})_3\text{O}_4$  and  $(\text{Mn,Co,X})_3\text{O}_4$  system, alloy design using physical metallurgy principles, and cost considerations. The desired alloy powders will be manufactured & characterized in detail.
- **Demonstration/validation of the contact layer performance in relevant SOFC stack environments.** Long-term ASR behavior and in-stack performance of the contact layer in relevant stack operating environments, its microstructure, chemical compatibility & Cr-retaining capability will be evaluated.
- **Further cost reduction and commercialization assessment.** Approaches to further reducing the stack cost will be explored, such as co-sintering of the interconnect coating and contact layer. Cost analysis and scale-up assessment will be conducted for potential commercialization.

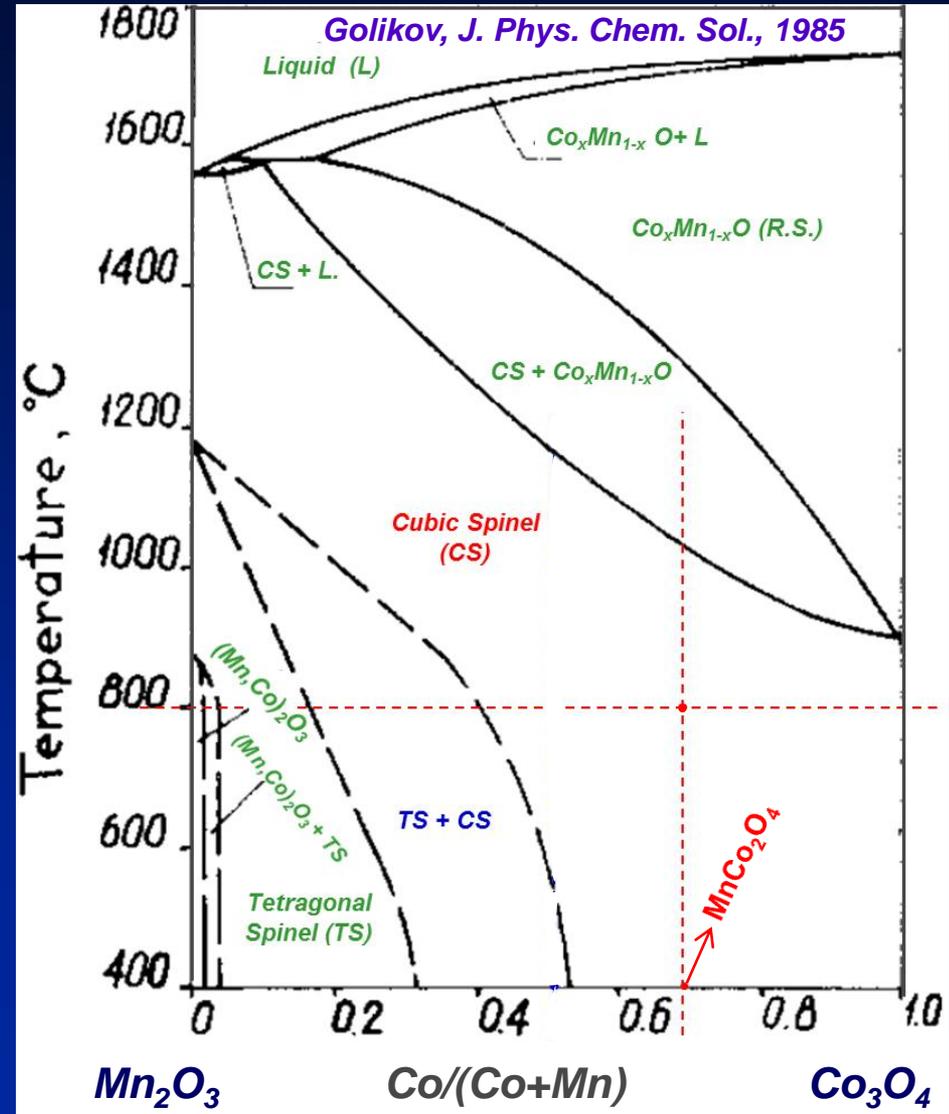
# Compositional Design for Co-Mn Based Alloy Precursor



**Phase Diagram of the Co-Mn System**

Co-31.76 wt.%Mn was selected as the baseline composition, which after thermal conversion would lead to the  $\text{MnCo}_2\text{O}_4$  spinel formation:

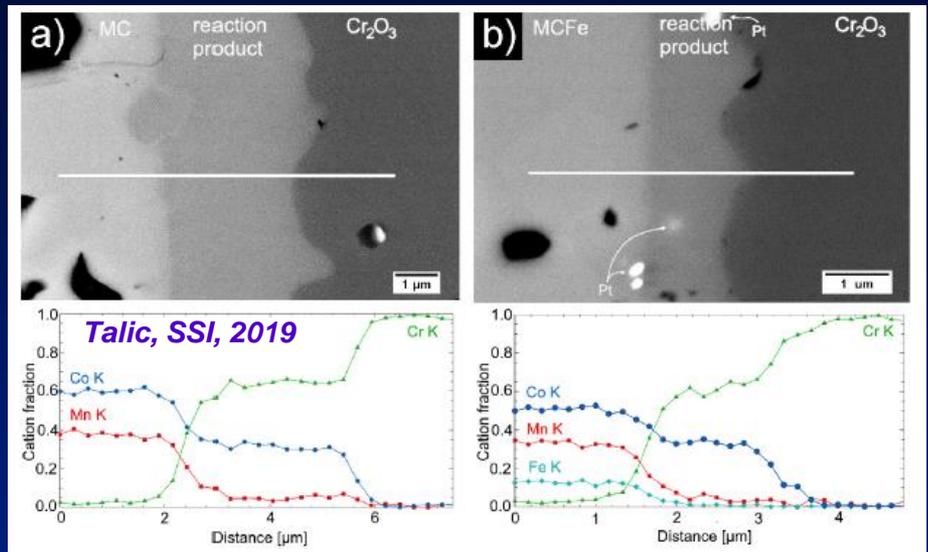
- ✓ The Co-31.76 wt.%Mn alloy has the single-phase fcc structure at 900°C;
- ✓  $\text{MnCo}_2\text{O}_4$  has a cubic structure at both 800 and 20°C



**Phase Diagram of the Co-Mn-O System in Air**

# Compositional Design for Co-Mn Based Alloy Precursor

- Fe addition to the binary Co-Mn alloy for reducing the reaction layer growth;
- Ce doping for slowing down the  $\text{Cr}_2\text{O}_3$  scale growth on the alloy and improving the scale adhesion via “reactive-element effect”.



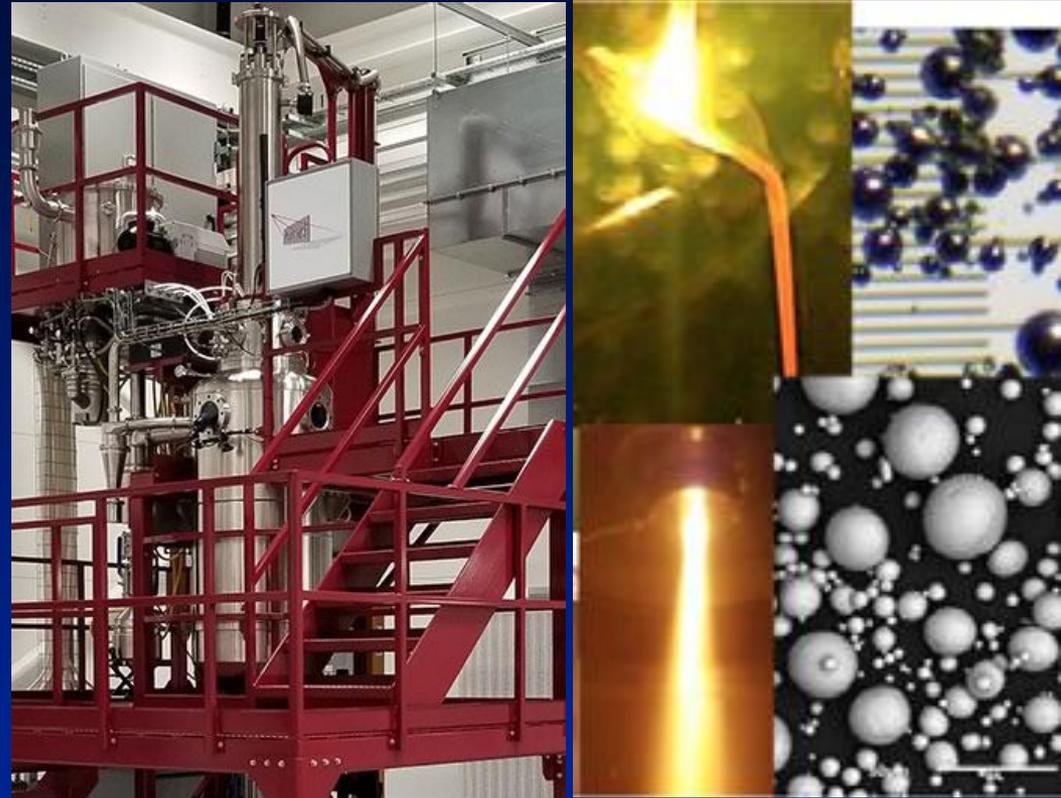
The interface of (a)  $\text{MnCo}_2\text{O}_4$  (MC)/ $\text{Cr}_2\text{O}_3$  & (b)  $\text{MnCo}_{1.7}\text{Fe}_{0.3}\text{O}_4$  (MCF)/ $\text{Cr}_2\text{O}_3$  diffusion couple after 900°C for 300 h

## Alloy Compositions & Corresponding Spinel Composition after Thermal Conversion

(in wt.%)	Co	Mn	Fe	Ce	Spinel Composition
Alloy 1	68.21	31.79	—	—	$\text{MnCo}_2\text{O}_4$
Alloy 2	64.72	31.80	3.23	—	$\text{MnCo}_{1.9}\text{Fe}_{0.1}\text{O}_4$
Alloy 3	64.72	31.77	3.23	0.41	$\text{MnCo}_{1.895}\text{Fe}_{0.1}\text{Ce}_{0.005}\text{O}_4$

# Gas Atomization of Selected Alloys

- A number of Co-Mn based compositions were selected for the powder preparation.
- The powders with the desired composition and particle size were manufactured using a semi-industrial gas atomizer.



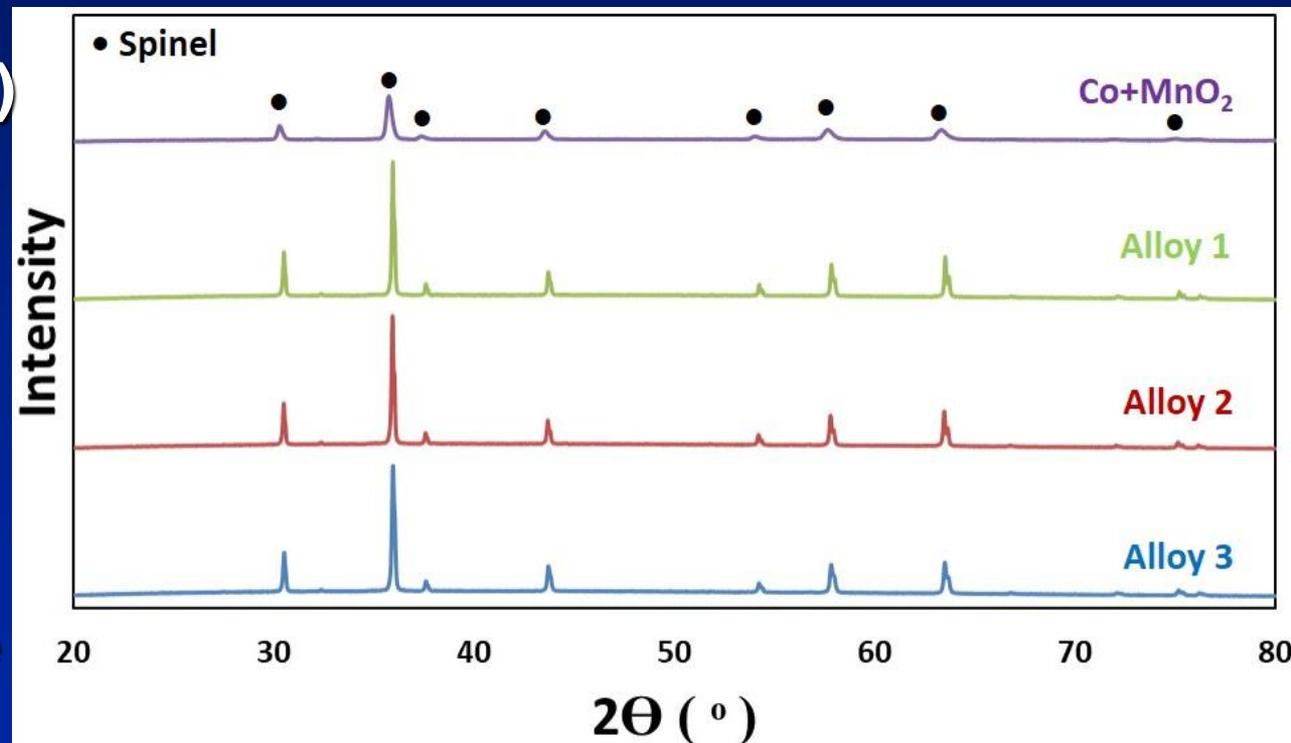
**Gas Atomization System**

- Chemical analysis of the collected powders was conducted at Dirats Lab – a close match with the targeted compositions was achieved.

A lab-scale gas atomizer is currently being installed at TTU under a DoD DURIP project.

# Thermal Conversion of Alloy Precursors

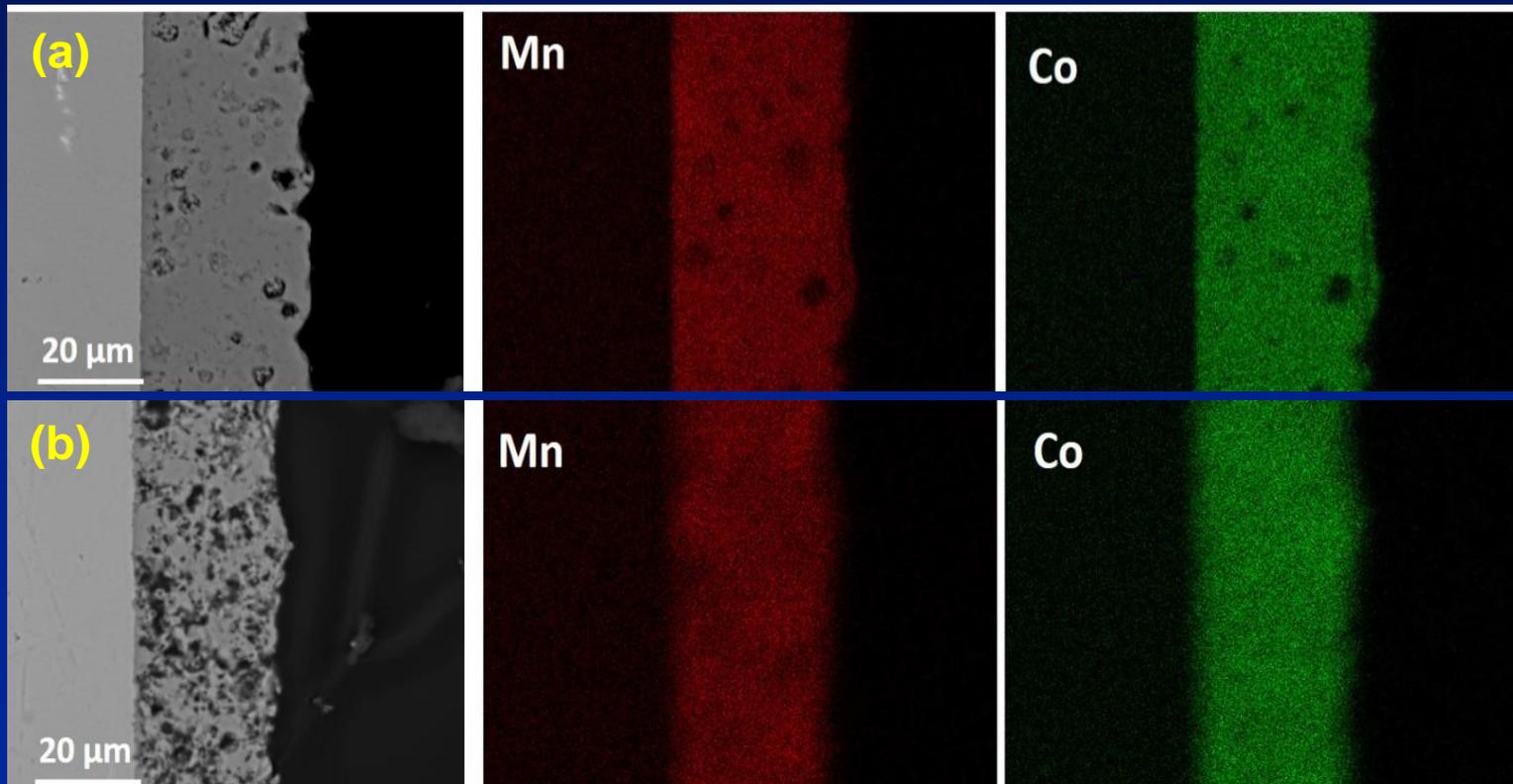
- Three alloy powders and a Co+MnO<sub>2</sub> mixture were used as the precursor for synthesis of the desired spinel layer.
- After screen printing, the precursor layer was sintered at 900°C for 2 h in air to convert it to the spinel layer.
- A single-phase (Mn,Co)<sub>3</sub>O<sub>4</sub> (MCO) spinel layer was formed for all the precursors.
- The alloy precursors led to narrower peaks, indicating a more homogeneous composition.



XRD Patterns of Different Precursors after Conversion

# Microstructure of Converted Spinel Layer

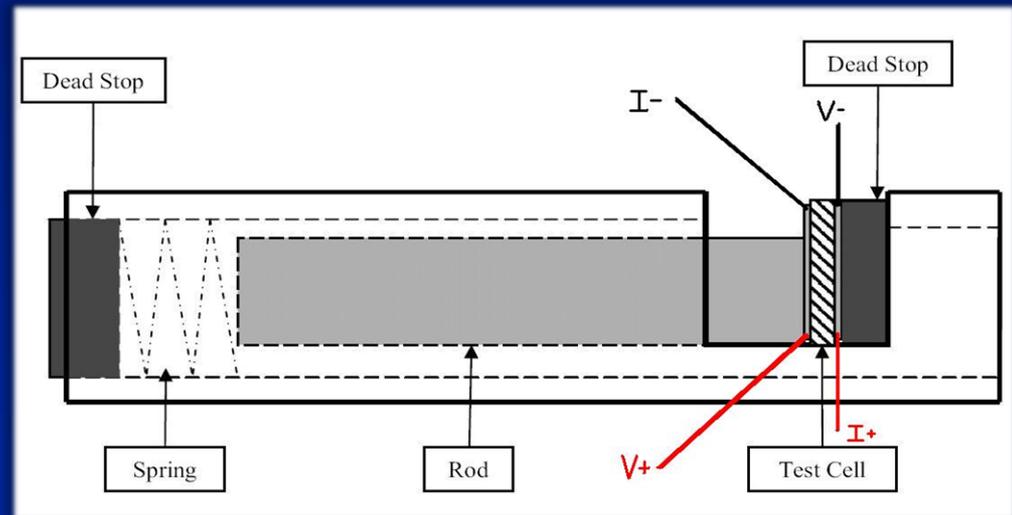
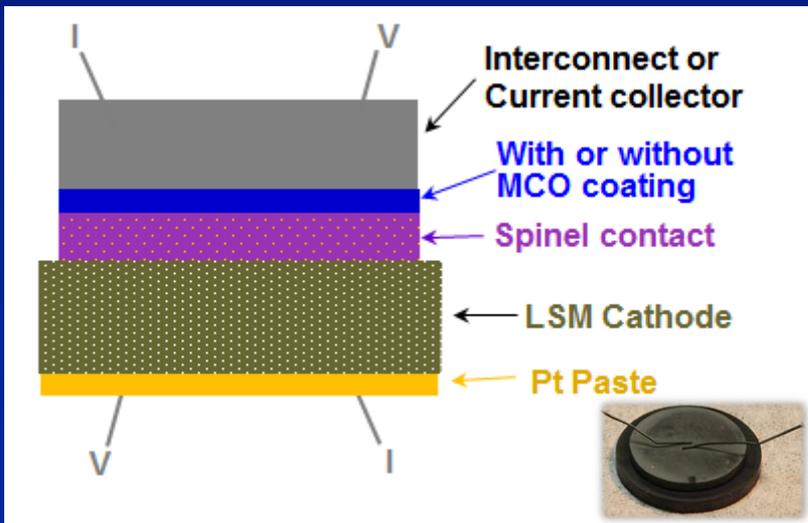
- The MCO layer formed with the alloy powders was more dense, compared to that with  $\text{Co}+\text{MnO}_2$ . This is a result of more volume expansion with the alloy precursor.



**Cross-sectional View of the Spinel Layer Converted from Different Precursors: (a) Alloy 1 and (b)  $\text{Co}+\text{MnO}_2$**

# Area-Specific Resistance (ASR) Measurement

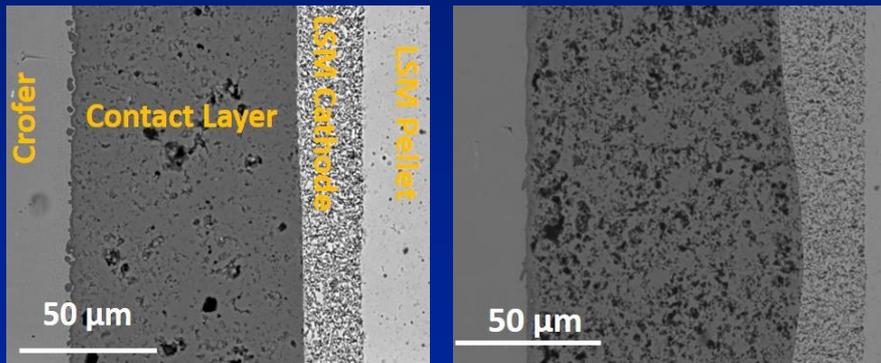
- A number of test cells were constructed, with the spinel-forming contact precursor layer sandwiched between the ferritic alloy Crofer 22 APU and the LSM cathode.
- The test cells were spring-loaded and heated to and held at  $900^{\circ}\text{C}$  for 2 h; after cooling down to  $800^{\circ}\text{C}$ , the cell ASR change during isothermal exposure at  $800^{\circ}\text{C}$  in air was monitored using a special 6-cell test rig.



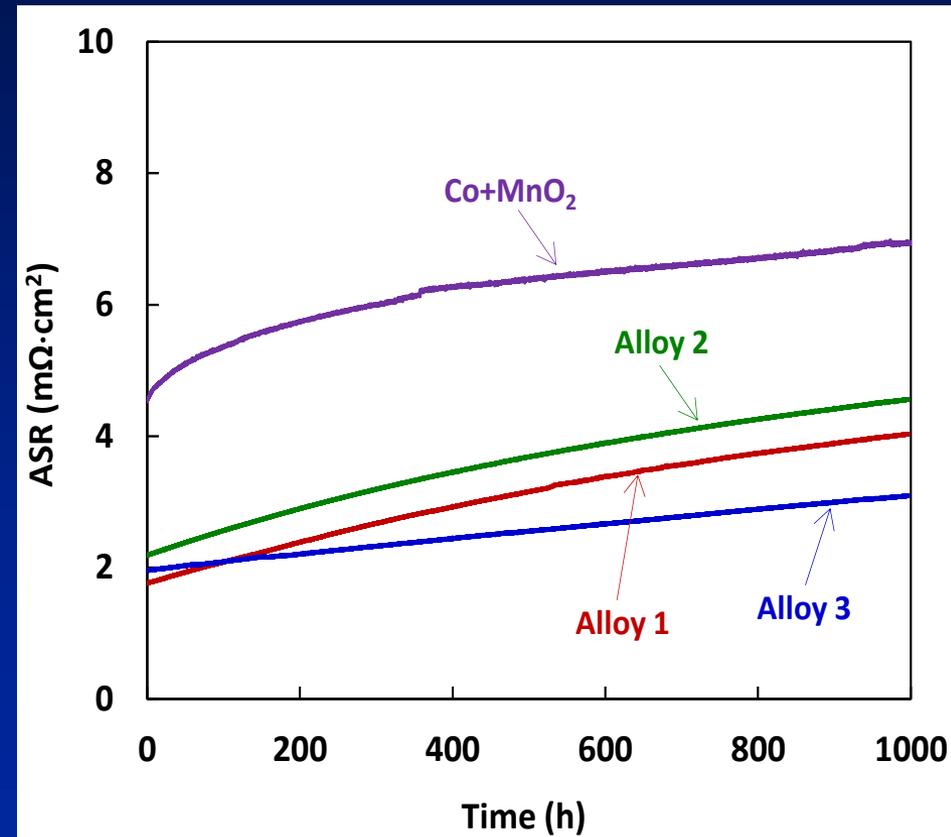
Schematic of the ASR Test Cell and Test Configuration

# Cell ASR with Different Metallic Contact Precursors

- The ASR for Crofer/contact /LSM cells with the alloy precursors was much lower than that with the  $\text{Co}+\text{MnO}_2$  precursor over the duration of the test.
- The higher cell ASR with the  $\text{Co}+\text{MnO}_2$  precursor is partially due to the higher porosities in the contact layer.



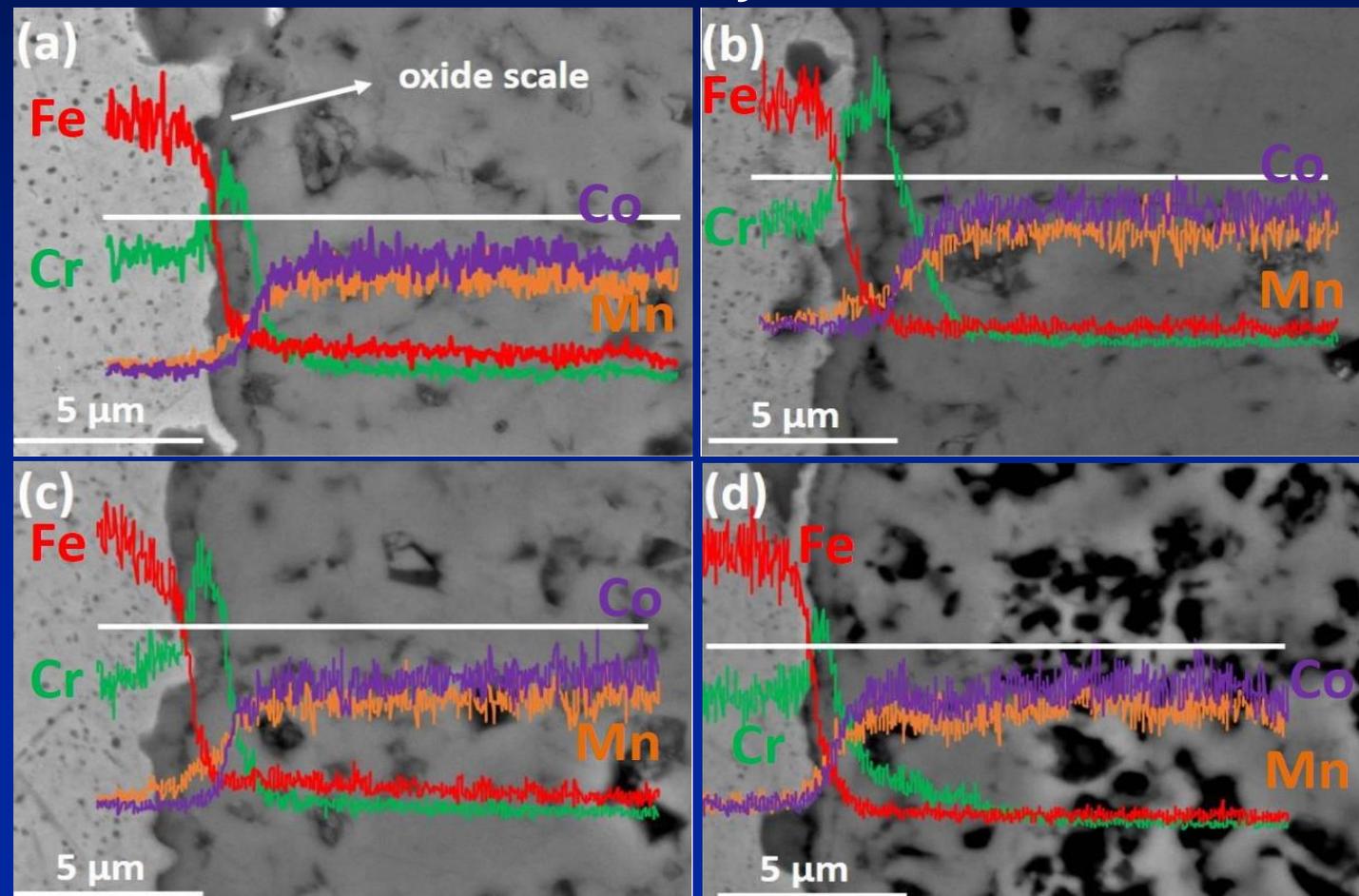
Cross Sections of Crofer/Contact/LSM cells with Different Contact Precursors



Cell ASR vs. Time at 800°C in Air

# Crofer-Contact Interface: Formation of $\text{Cr}_2\text{O}_3$ Scale and $(\text{Mn,Co,Cr})_3\text{O}_4$ Reaction Layer

- A  $\text{Cr}_2\text{O}_3$  scale on Crofer 22 APU surface and a reaction layer (RL) between the  $\text{Cr}_2\text{O}_3$  scale and the MCO contact were formed after testing.
- Alloy contact precursor composition had a significant effect on the thicknesses of these two layers.



Cross-sectional views of the ASR-tested cells with different contact precursors: (a) Alloy #1; (b) Alloy #2; (c) Alloy #3; and (d)  $\text{Co}+\text{MnO}_2$ . Images were taken near the Crofer-contact interface with the superimposed EDS line scans.

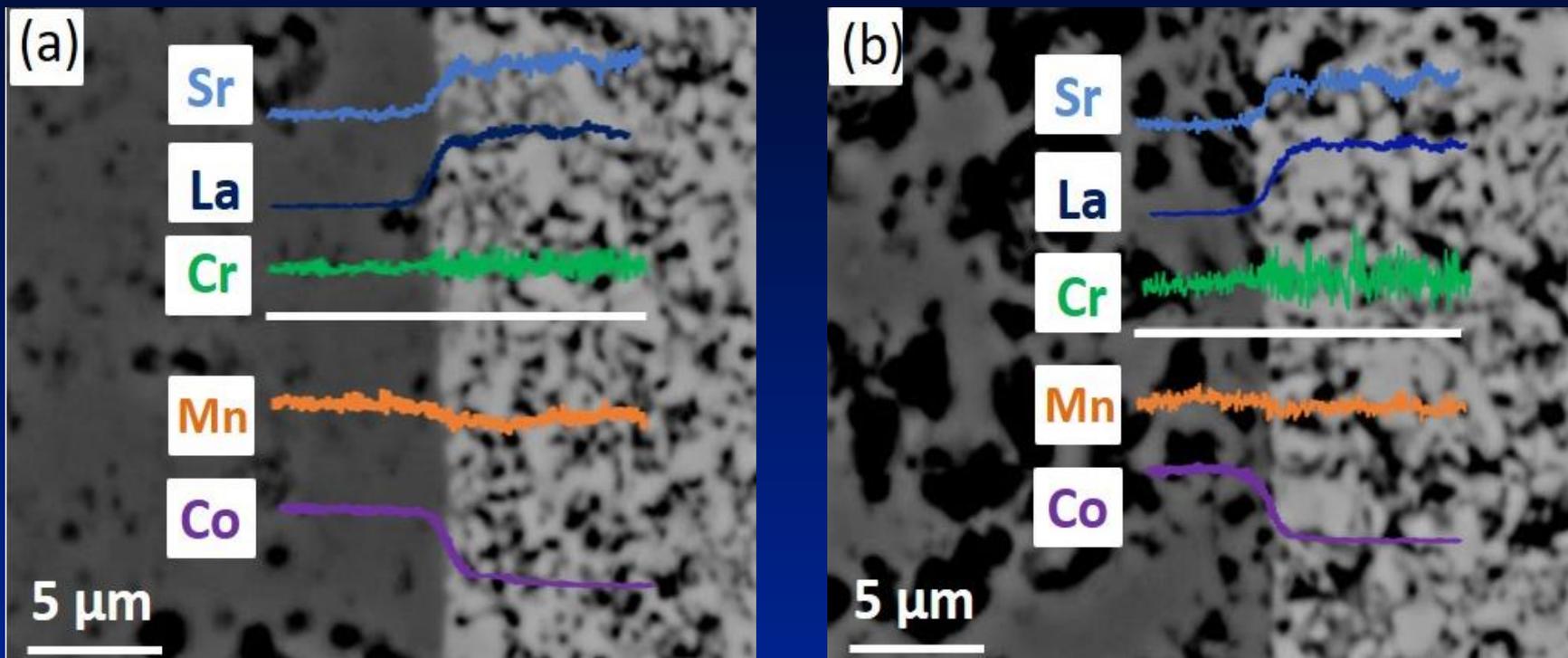
# Summary of ASR and Cr<sub>2</sub>O<sub>3</sub>/Reaction Layer Thickness

- While Fe addition into the Co-Mn alloy powder had no beneficial effect on the cell electrical performance, Ce doping (in Alloy #3) led to the lowest cell ASR and ASR degradation rate as a result of slower Cr<sub>2</sub>O<sub>3</sub> scale/RL growth and better scale adhesion.

**Initial ASR ( $R_i$ ), Final ASR after 1000-h Testing ( $R_f$ ), ASR Degradation Rate during Initial 50 h ( $DR_o$ ), ASR Degradation rate during Final 50 h ( $DR_f$ ), and Thicknesses of the Cr<sub>2</sub>O<sub>3</sub> Scale and RL after ASR Testing**

	Alloy 1	Alloy 2	Alloy 3	Co+MnO <sub>2</sub>
$R_o$ (mΩ·cm <sup>2</sup> )	1.8	2.2	2.0	4.6
$R_f$ (mΩ·cm <sup>2</sup> )	4.0	4.6	3.1	7.0
$DR_o$ (μΩ·cm <sup>2</sup> /h)	3.3	3.5	1.5	11.1
$DR_f$ (μΩ·cm <sup>2</sup> /h)	1.4	1.3	1.0	1.4
Cr <sub>2</sub> O <sub>3</sub> scale (μm)	1.0	1.0	0.9	0.6
RL (μm)	1.4	2.2	1.4	4.5

# Effectiveness of the Contact Layer in Mitigating Cr Migration

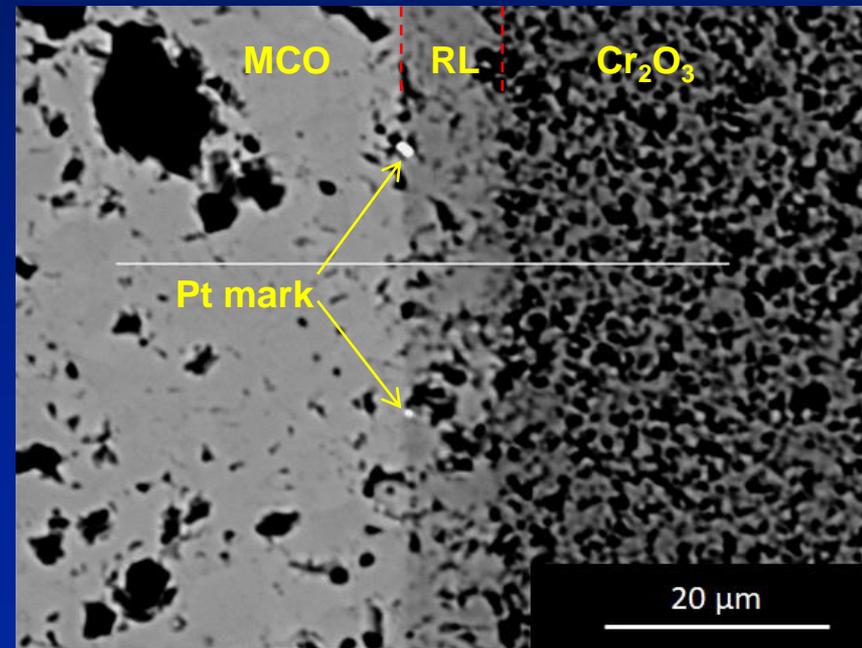


## Cross-sectional View of Tested Cells near the LSM-Contact Interface with Different Contact Precursors: (a) Alloy 1 and (b) Co+MnO<sub>2</sub>

- Minimal interdiffusion was observed between the contact layer and cathode for all tested cells.
- No Cr was detected in the LSM cathode for the cells with the alloy contact precursors, while for the cell with the Co+MnO<sub>2</sub> precursor the Cr level in the porous LSM cathode fluctuated noticeably.

# Preliminary Study on Reaction Layer Formation Kinetics/Mechanism

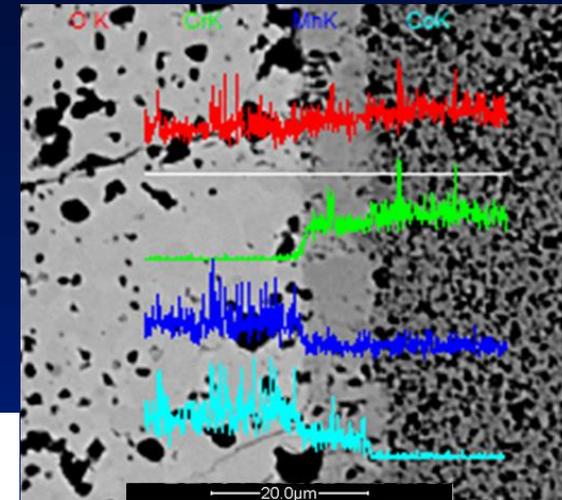
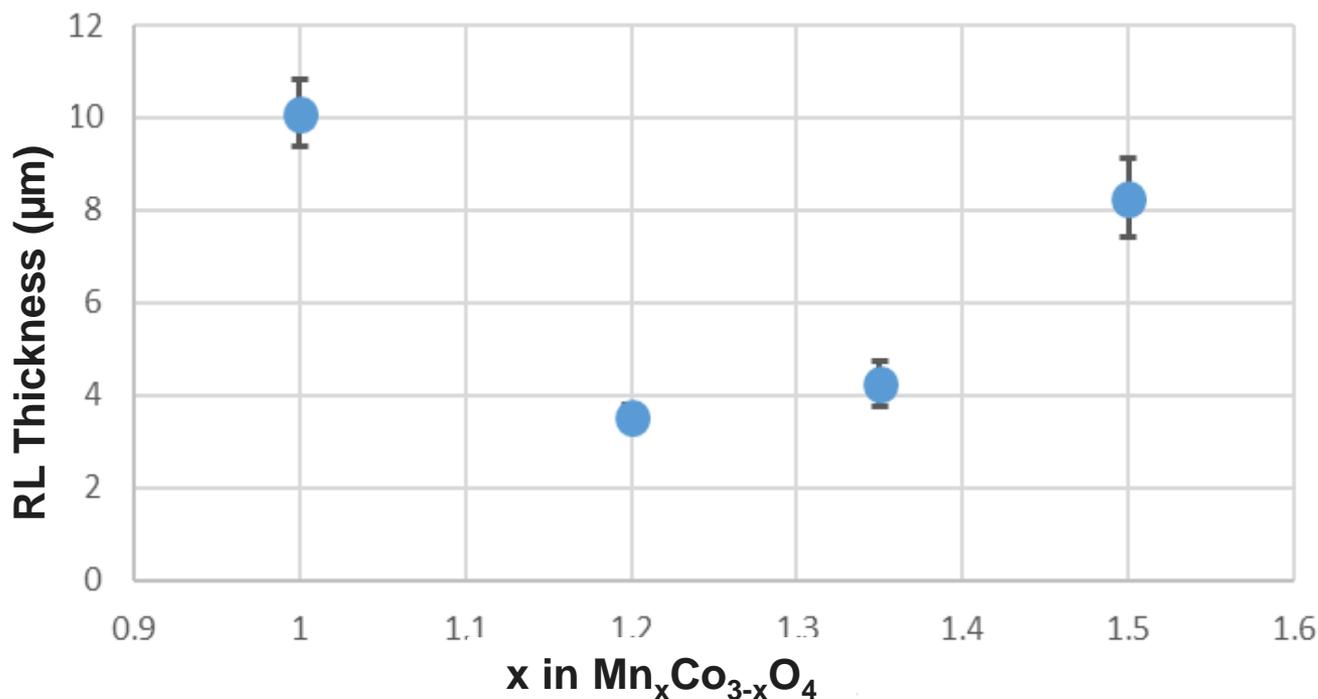
- Since a RL was formed between the MCO layer and the  $\text{Cr}_2\text{O}_3$  scale, it is critical to study the effect of the MCO stoichiometry and additional doping on the LR formation kinetics/mechanism:
  - Contacting faces of  $\text{MCO}/\text{Cr}_2\text{O}_3$  pellets were ground to 800-grit;
  - Pt particles were applied to contacting face of MCO pellet to mark the original interface;
  - The couple was loaded into a vertical furnace and then diffusion annealed at  $900^\circ\text{C}$  for different times,
  - The annealed couples were cross-sectioned and examined with SEM/EDS.



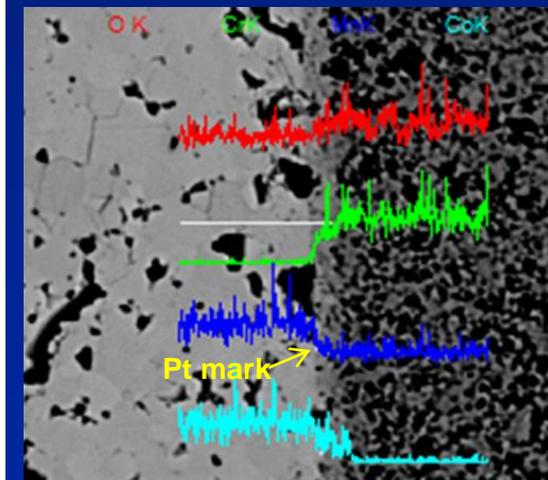
**$\text{Mn}_{1.5}\text{Co}_{1.5}\text{O}_4/\text{Cr}_2\text{O}_3$  Couple after  
Annealing at  $900^\circ\text{C}$  for 300 h**

# Preliminary Study on Reaction Layer Formation Kinetics/Mechanism

- A minimum in RL thickness was observed when  $x = 1.2$  in  $\text{Mn}_x\text{Co}_{3-x}\text{O}_4$ .
- Pt marks were always observed near the MCO/RL interface, indicating the RL formation was via Co/Mn diffusion into  $\text{Cr}_2\text{O}_3$ .
- The RL growth kinetics is being determined.



$\text{MnCo}_2\text{O}_4/\text{Cr}_2\text{O}_3$  Couple

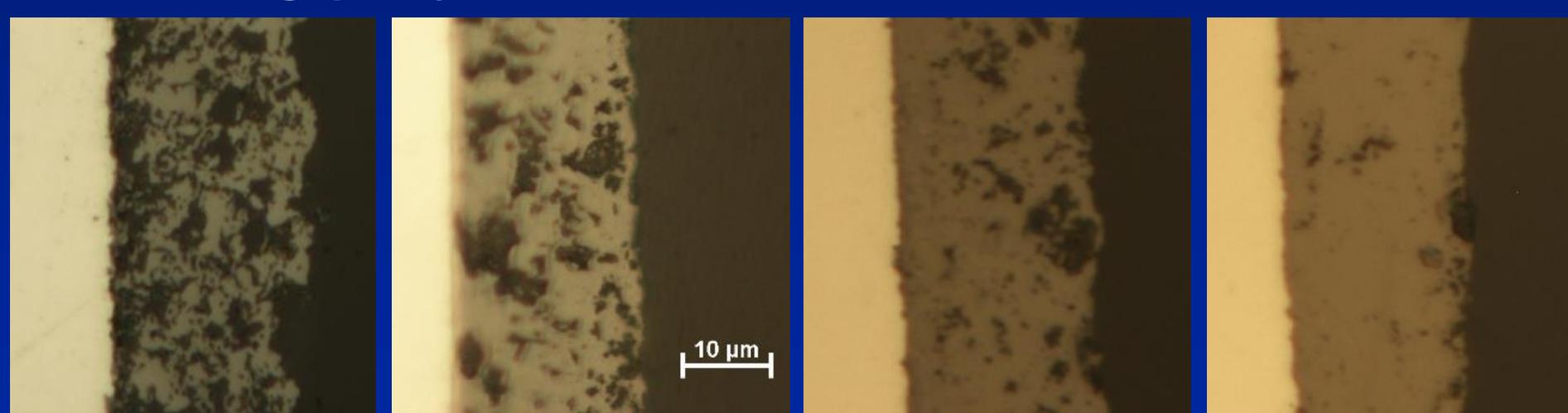


$\text{Mn}_{1.2}\text{Co}_{1.8}\text{O}_4/\text{Cr}_2\text{O}_3$  Couple

RL Thickness after  $900^\circ\text{C} \times 300\text{-h}$  Diffusion Anneal

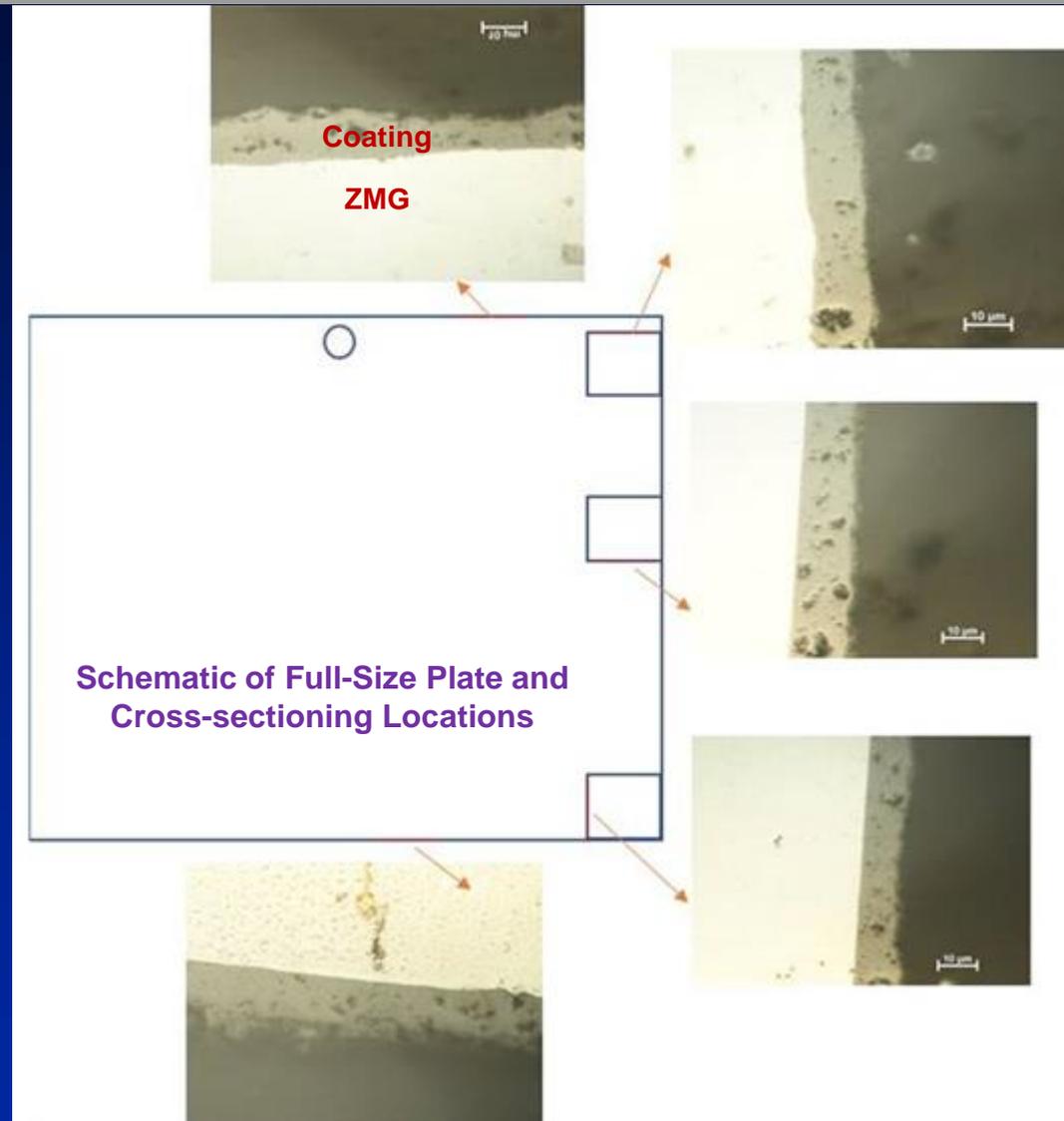
# EARS Processing of Dense MCO Coatings

- Reactive sintering of a dense MCO coating via EARS has been explored and promising results were obtained:
  - By controlling the composition/shape/size/size distribution/initial packing density of the metallic precursor powders, a dense spinel layer was obtained after thermal conversion at 900°C for 2 h in air.
- The EARS-derived coating does not require a reduction treatment or a sintering temperature higher than 900°C, and potentially offers a better MCO coating quality.



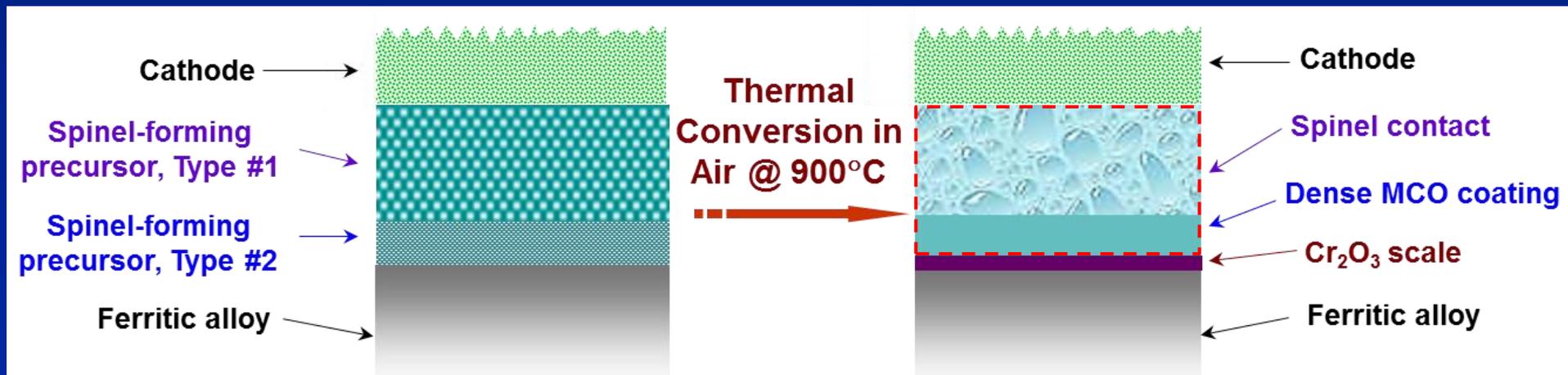
# Processing MCO Coating on Full-Size ZMG 232G10 Alloy Plate

- By optimizing the metallic spinel-forming precursor and thermal conversion conditions, a dense MCO coating was achieved, which was quite uniform in thickness and density throughout the alloy plate surfaces.
- The overall quality of the coating on the full-size alloy plate was comparable to that achieved on the small coupons.



# Co-sintering of Coating/Contact Dual-Layer Structure: Further Cost Reduction and Performance Improvement

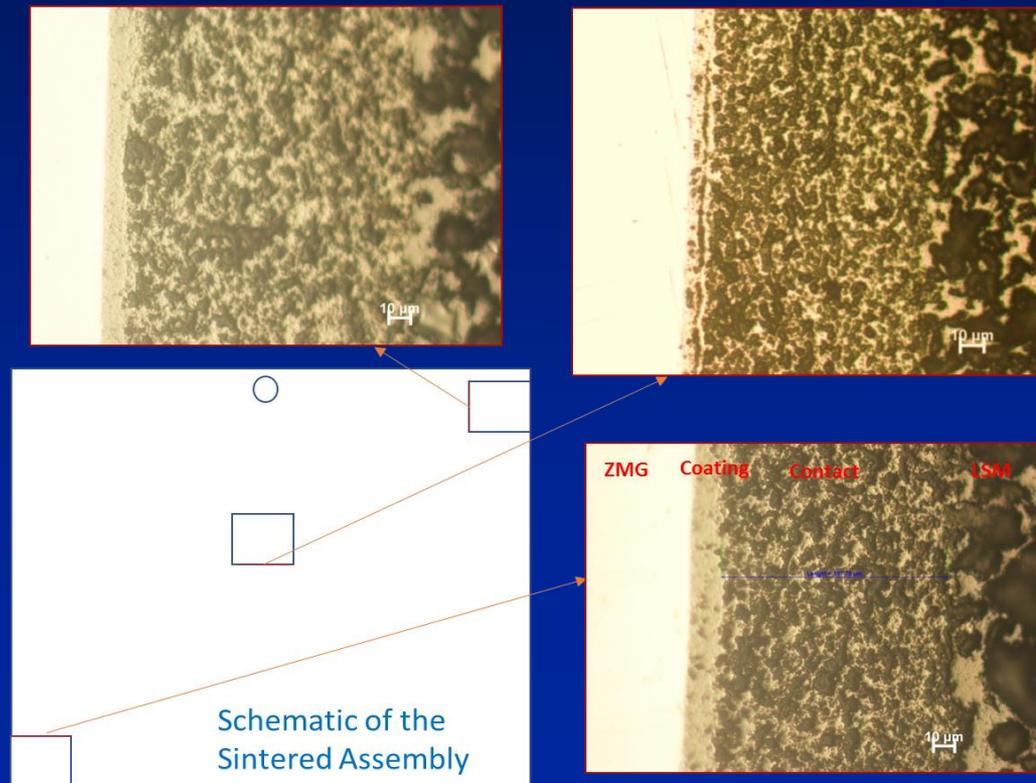
- To lower the interfacial ASR at the alloy/contact interface, improve the contact quality, and reduce the coating and contact processing cost, co-sintering of the coating and the contact layer during initial stack firing/operation is also explored, utilizing two different metallic precursors:
  - Two spinel-forming precursors (Type #1 – for the contact layer and Type #2 – for the dense MCO coating) will be employed;
  - Reactive co-sintering in air at a sintering temperature of 900°C will be utilized for simultaneous formation of the dual-layer structure.



Schematic of Co-sintering of Spinel-Based Coating/Contact Dual-Layer Structure <sup>23</sup>

# Co-sintering of Coating/Contact Layers between Full-Size Ferritic ZMG 232G10 Alloy and LSM Plates

- Using full-sized alloy and LSM plates supplied by our industrial collaborator, a full-size alloy/coating/contact/LSM assembly with two metallic precursors for the coating and contact layer formation was prepared by co-sintering at 900°C.
  - ✓ A dense coating and a porous contact layer between the ZMG alloy and LSM plates were achieved after co-sintering, via the utilization of two individually-optimized metallic precursors.
  - ✓ The thicknesses of the coating and the contact layer were 10-15 and 100-120  $\mu\text{m}$ , respectively.
  - ✓ ASR testing of the sintered assembly is underway.



# Project Milestones

Milestone Title/Description	Planned Start Date	Planned Completion Date	Actual Completion Date	Verification Method	Status
Revised PMP	10/10/2017	10/30/2017	10/23/2017	PMP file	Completed
Kickoff Meeting	10/30/2017	12/29/2017	11/29/2017	Presentation file	Completed
Compositional optimization of Precursor alloy	11/01/2017	06/30/2018	06/28/2018	Optimal Fe/Ni/Co and other alloy additions are identified.	Completed
Preparation of the alloy powder	11/15/2017	09/31/2018	09/20/2018	Atomization of one alloy powder is completed.	Completed
Demonstration of ASR stability with the new contact	01/01/2018	06/30/2019	02/20/2019	The ASR stability is demonstrated successfully for about 5,000 h.	Completed
Demonstration of stack performance stability with 1-cell stack testing	04/02/2018	09/30/2019	09/27/2019	Stack performance stability testing for $\geq 1,000$ h is completed at industrial site.	Completed
Synthesis of dense coating on full-size current collector	10/20/2019	02/29/2020	12/25/2019	Dense coating on full-size current collector plate is achieved.	Completed
In-stack evaluation with co-sintered coating/contact structure	01/20/2020	08/20/2020		Stack power output stability testing for over 500 h is completed at Saint Gobain.	In progress
Cost analysis and feasibility assess.	06/30/2020	09/30/2020		Cost analysis and scale-up assessment are completed.	Not started

# Concluding Remarks

- The spinel contact-forming alloy precursor composition had a significant effect on the  $\text{Cr}_2\text{O}_3$  scale thickness/quality, reaction layer thickness between the spinel and  $\text{Cr}_2\text{O}_3$ , and cell ASR.
- The MCO spinel stoichiometry drastically affected the thickness of the reaction layer formed between MCO and  $\text{Cr}_2\text{O}_3$ , with the  $\text{Mn}_{1.2}\text{Co}_{1.8}\text{O}_4$  spinel leading to the thinnest reaction layer.
- By controlling the metallic precursor powder composition, size, size distribution, and initial powder packing density, a dense MCO coating has been synthesized on large-size ferritic alloy plate.
- Co-sintering of a dual-layer structure with a dense spinel layer as coating and a porous layer as contact can be achieved by utilizing two tailored metallic precursors between large-size alloy and LSM plates.

# Acknowledgments

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