

Intermediate Temperature Solid Oxide Fuel Cell Development SECA Core Technology Program Review Meeting

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Outline

- Review of intermediate temperature electrolyte options
 - Opportunities and challenges
- Phase I electrode strategy
 - Phase I results
- Phase II plan

Current SOFC Electrolytes

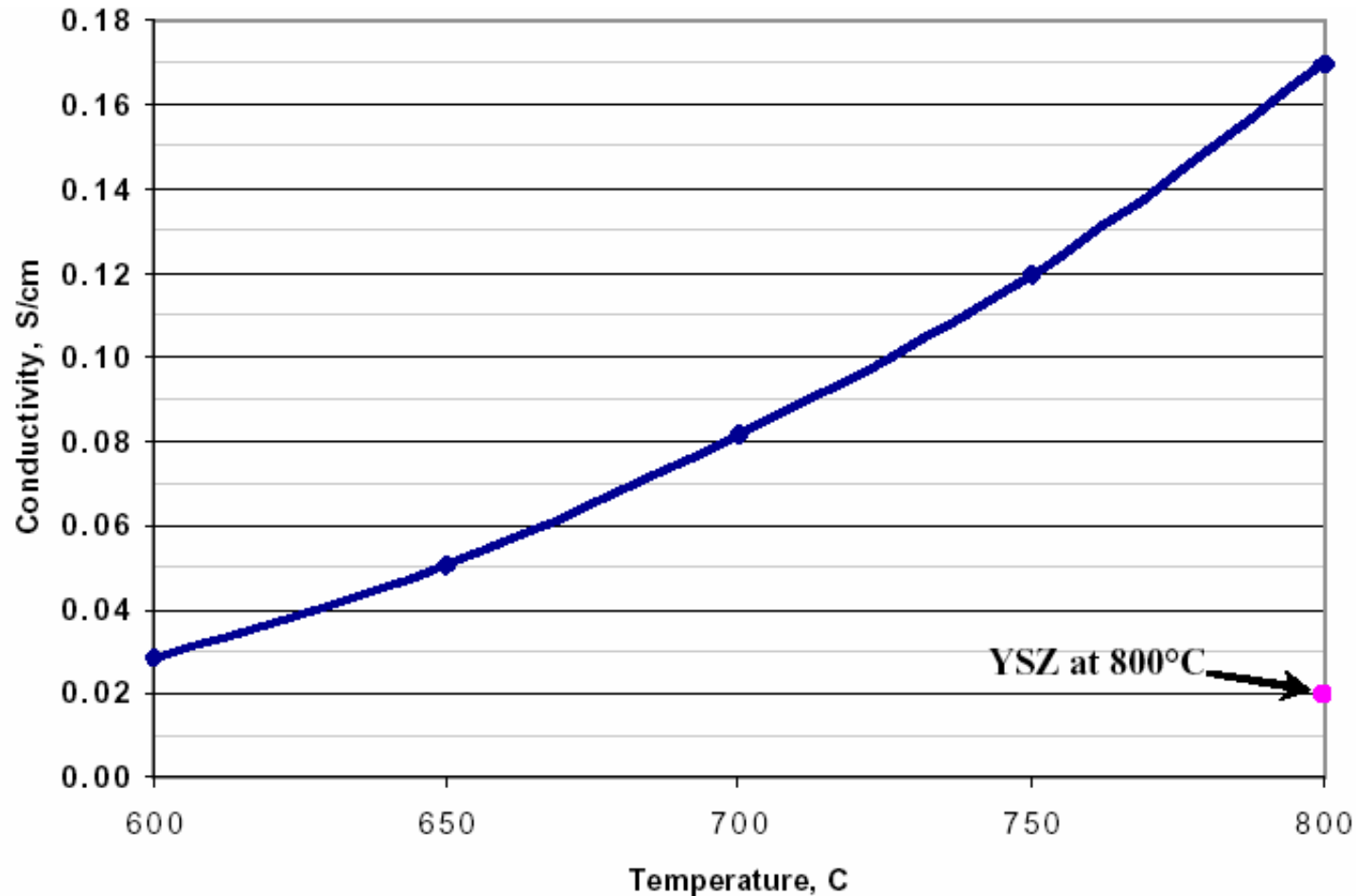
| System | Electrolyte Issues | Anode Material and Issues | Cathode Material and Issues | General Comments |
|------------|--|--|--|--|
| Zirconia | Low conductivity 0.1 S/cm at 1000°C 0.02 S/cm at 800°C | Ni-zirconia Ni coarsening | La(Sr)MnO ₃ zirconate formation at the interface | Demonstrated No electronic leak current 1000°C Operation typical; very thin electrolyte allows 800°C operation; lower than 800°C not practical yet |
| Ceria | High conductivity (0.1 S/cm at 800°C) | Ni-ceria | La(Sr)CoO ₃ CTE mismatch | Electronic short Differential expansion from air to fuel side |
| La gallate | High conductivity (0.1 S/cm at 800°C) | Ni-ceria Formation of La-Ni-O insulating phases | La(Sr)CoO ₃ CTE mismatch | No electronic leak current CTE similar to zirconia Long term cell stability is an issue material cost |

Lanthanum Gallate

La(Sr)Ga(Mg)O₃ (LSGM)

- High oxygen ion conductivity
 - Higher conductivity at 800°C than zirconia at 1000°C
- Stable in fuel
 - Ionic transference number ~1 over the entire pO₂ range of interest
- Challenges
 - Anode reactivity
 - Strength

LSGM Conductivity



- Ionic conductivity at 750°C ~ zirconia at 1000°C

Phase I Strategy for Electrodes

Anode

Present Anode: Ni-Ceria or Ni-LSGM

- Objective:
 - Eliminate or reduce reaction between La (in electrolyte) and Ni (in anode)
- Approach:
 - Additive to the anode that will reduce the reaction
 - Verify using powder reactivity studies

Phase I Strategy for Electrodes

Cathode

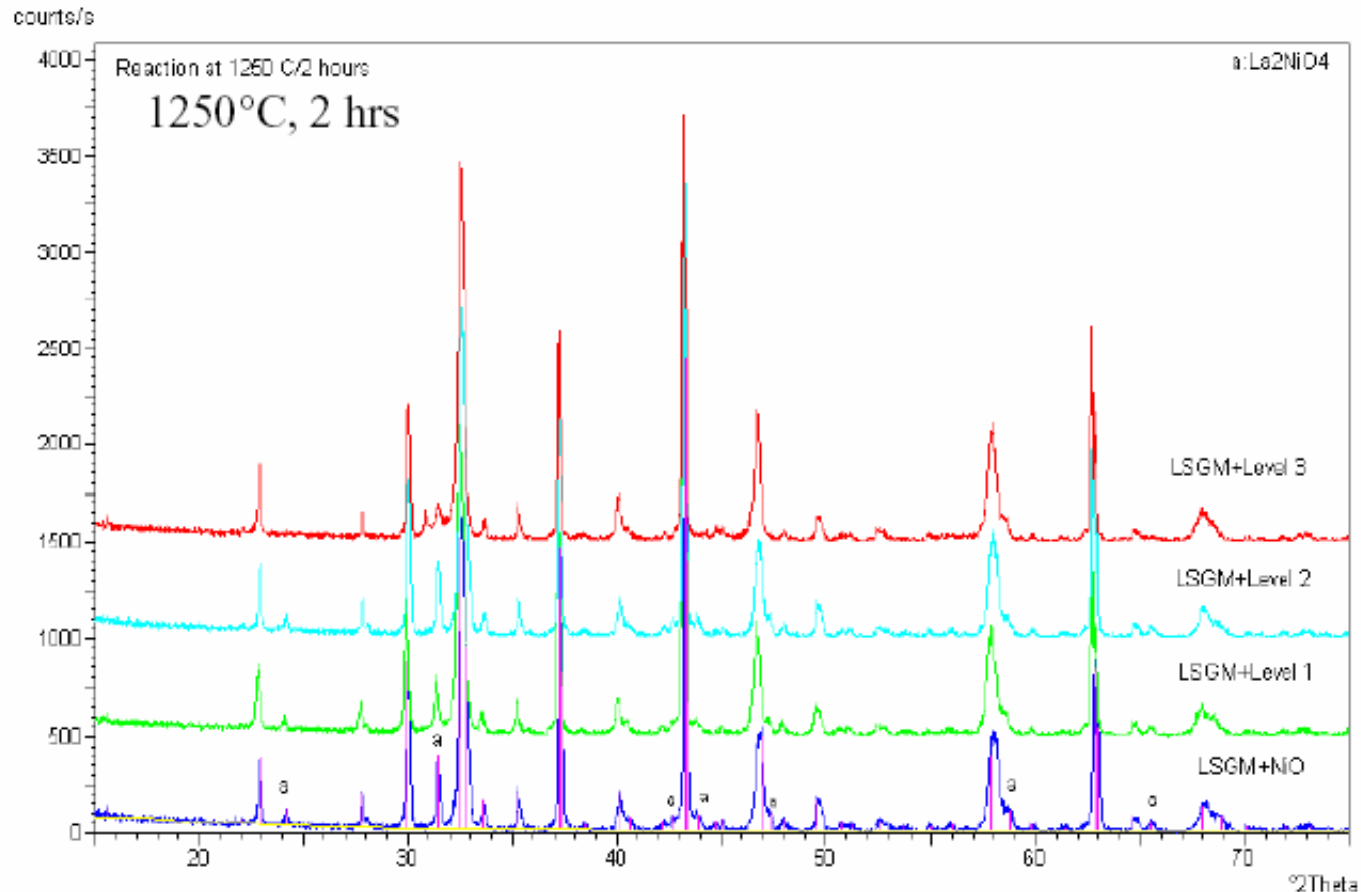
Present Cathode: $\text{La}(\text{Sr})\text{CoO}_3$

- Objective:
 - Lower cathode thermal expansion coefficient
- Approach
 - Doping in B-site (replacing Co) may lower CTE

Reactivity Studies

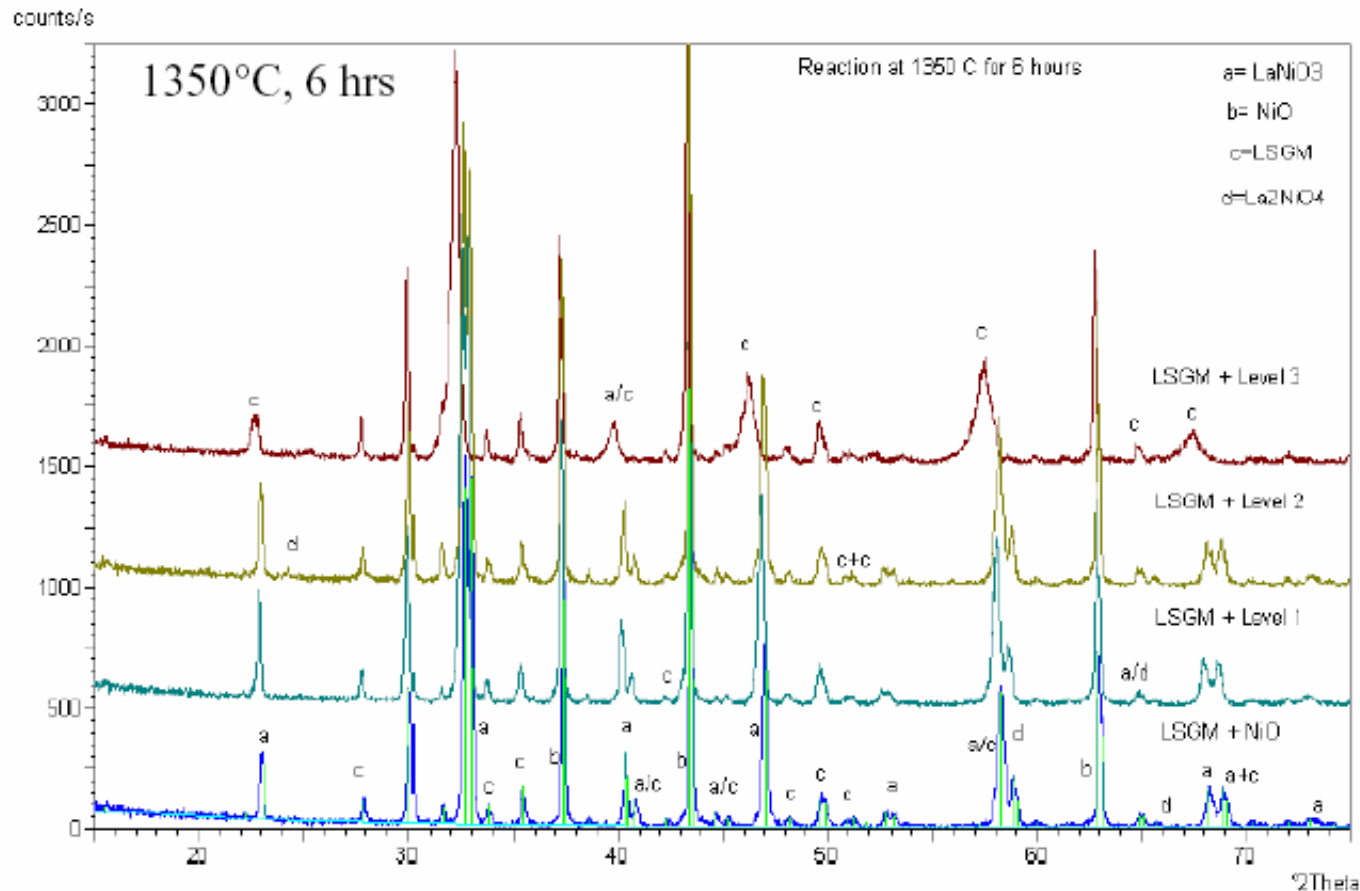
- Common reaction products
 - LaNiO_3 , La_2NiO_4
- Powder mixtures of modified Anode and LSGM
 - 1250°C, 2 hours and 6 hours
 - 1350°C, 6 hours
- 1250°C - potential electrode firing temperature for sintered electrolyte
- 1350°C - potential sintering temperature for an anode supported bi-layer

XRD - Reaction Study



- Additive Level 3 - significant reduction in second phase content at 1250° C

XRD - Reaction Study



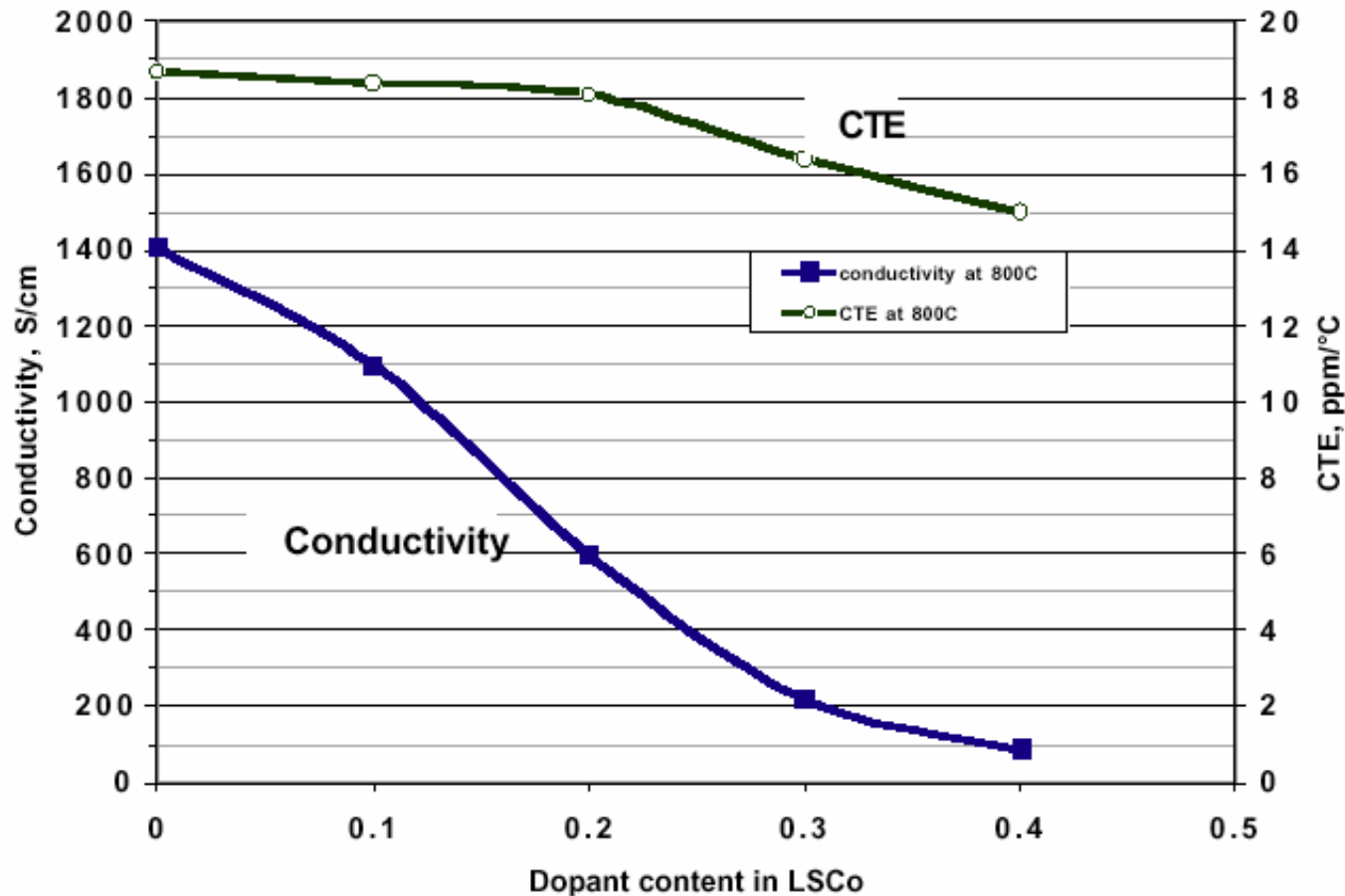
- Additive Level 3 - significant reduction in second phase content even at 1350° C

Conclusion of Anode Study

- Reaction with LSGM
 - Reaction products: 1250°C La_2NiO_4
1350°C LaNiO_3
 - Reaction products below detection limit even at 1350°C, 6 hrs in powder mixture
- The modified anode will allow anode supported thin gallate electrolyte

LSCo Cathode Evaluation

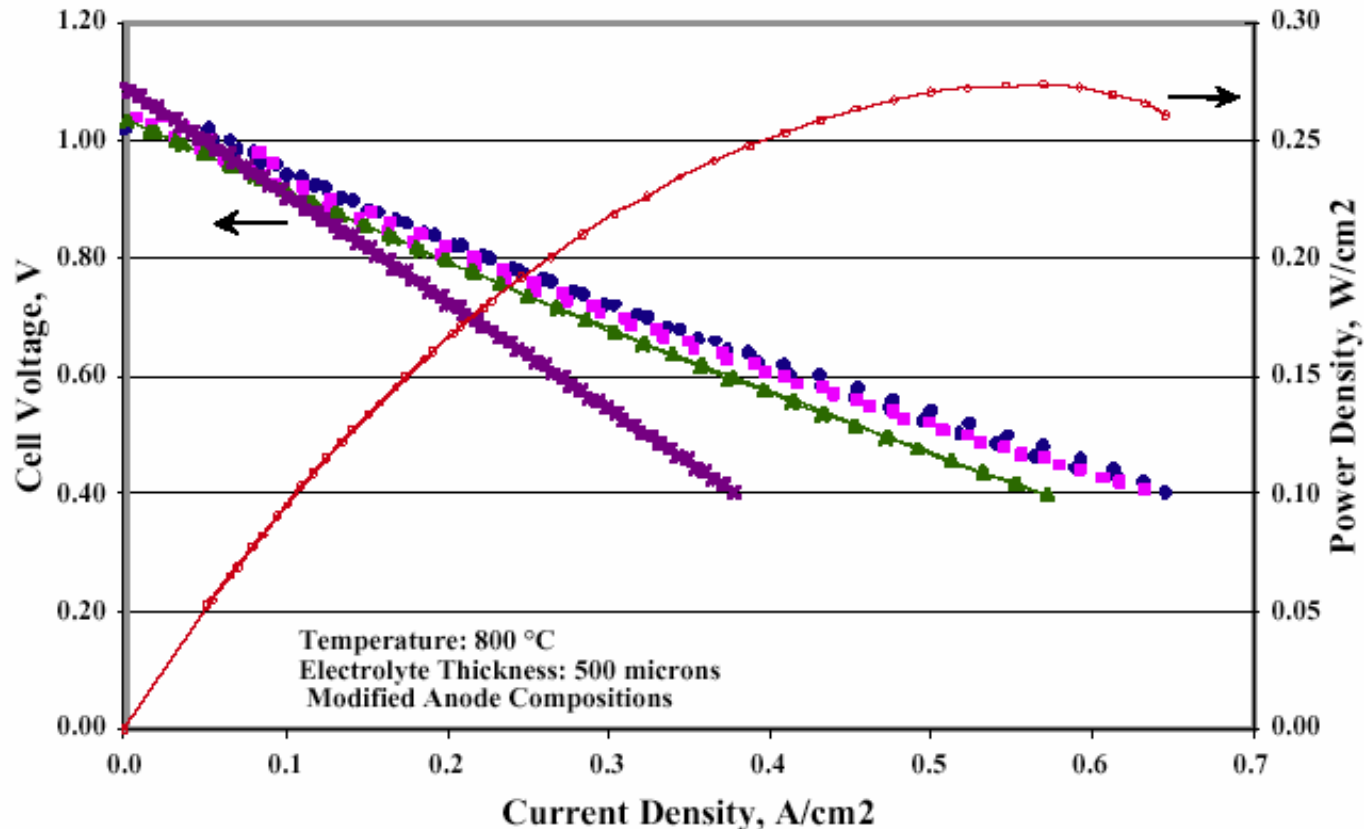
CTE and Conductivity



Conclusion of Cathode Evaluation

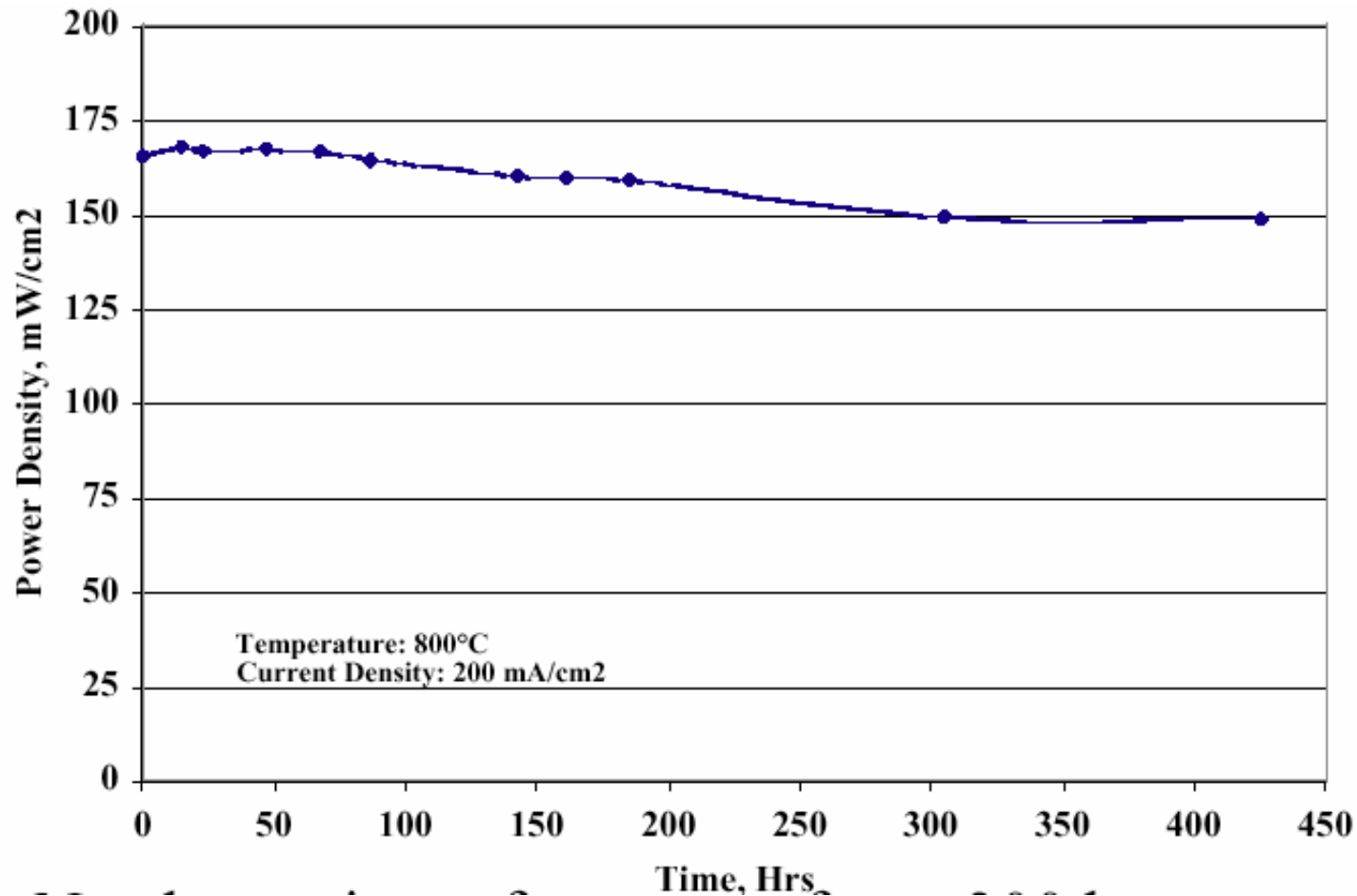
- B-site doping only slightly decreases the CTE
- Decrease in conductivity is rapid with B-site doping
 - The conductivity of doped LSCo is still high even at 600°C: 600 S/cm
 - CTE is still high at 18 ppm/°C
 - Cathode is likely fired at a lower temperature
 - Effect of mismatch would be lower
- No new phase formation detected at 1350°C, 6 hrs

Single Cell Performance



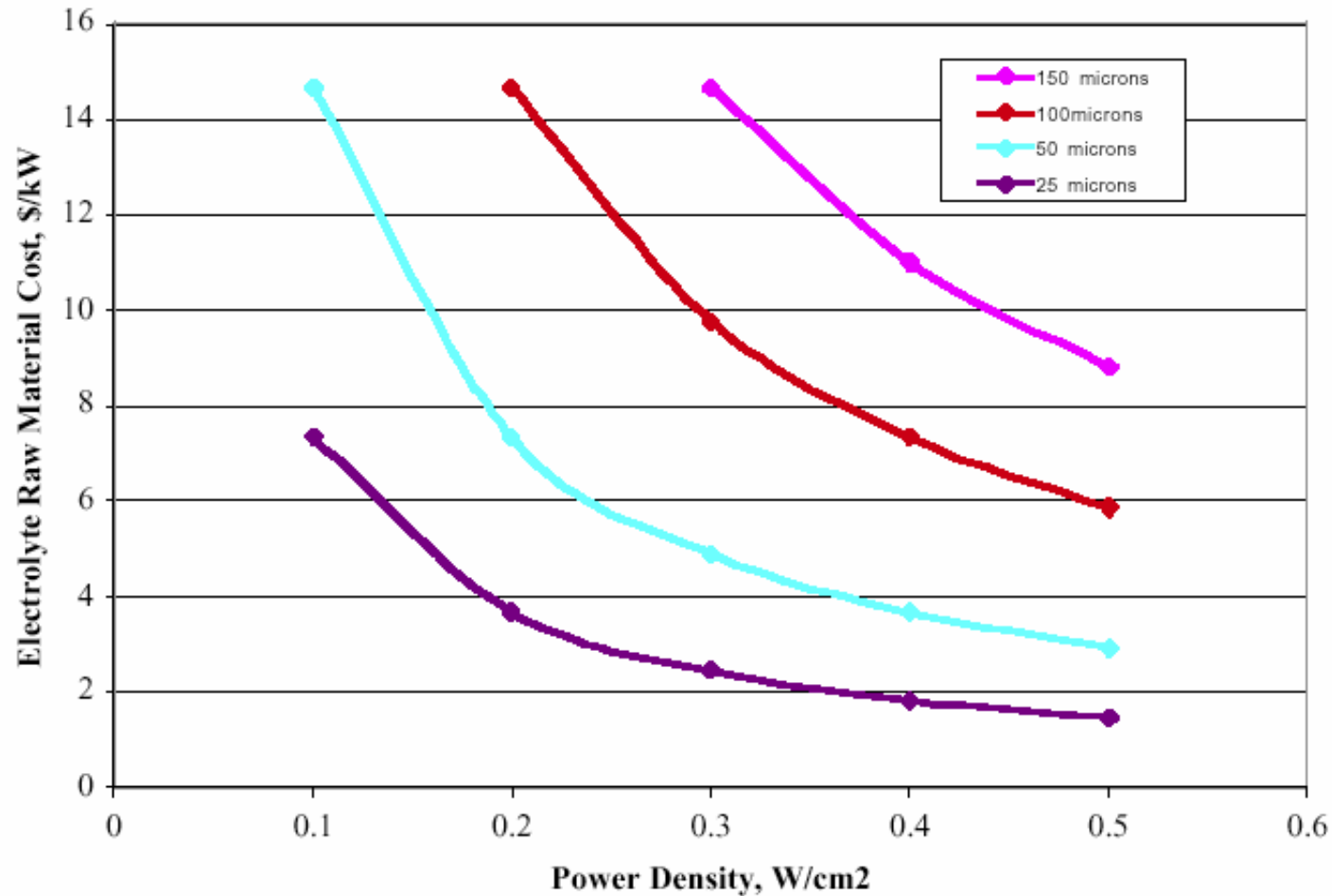
- > 1 V OCV indicating good ionic transference number
- ASR 1 ohm.cm² with newer electrodes and 500 micron electrolyte

Initial test on performance stability



- No change in performance after ~ 300 hours

LSGM Raw Material Cost Estimate



Summary of Phase I

- LSGM is a promising electrolyte for low temperature operation
- Reactivity with anode was addressed in Phase I
 - Anode composition with the promise of bi-layer compatibility developed
- No reaction product with LSCo cathode
 - Cathode less likely to be an issue

Technical Directions

- Cell Fabrication
 - Electrolyte thickness
 - Electrode microstructure
 - Fabrication conditions
- Electrolyte Strength
 - Literature Strength Data: 160 MPa at room temp. and 55 MPa at 900°C
 - Approach: Anode supported

Phase II Plan

- Technical Issues
 - Electrolyte sintering temperature
 - Strength of electrolyte
 - Bi-layer fabrication development
- Phase II Team
 - Ceramatec - Materials, Fabrication, Testing
 - Sandia National Labs. - Electrolyte Strength
 - New Mexico Tech. - Ceramic Process Development