Zeolite Membrane Reactor for Pre-Combustion Carbon Dioxide Capture

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Overview

Timeline
• Project start date: Oct. 1, 2015
• Project end date: Dec. 31, 2018
• Budget Periods:
  I: 10/1/2015-7/30/2017
  II: 8/1/2017-12/31/2018

Budget
• Total project funding
  – DOE $2,471,557
  – Cost-share: $620,527
  – Total: $3,092,084
• Funding for BP I:
  – DOE $1,274,869

Research Area
2B2: Bench-Scale Pre-Combustion CO₂ Capture Development and Testing

Partners
• Arizona State University (ASU)
• University of Cincinnati (UC)
• Media and Process Technology, Inc (MPT)
• Nexant, Inc.
Principal Investigators

Arizona State University
Jerry Y.S. Lin

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Junhang Dong

Media Processes and Technologies
Richard Ciora

Nexant, Inc
Gerald Choi

NETL Project Manager
Andrew Jones
Project Objectives

To demonstrate a bench-scale zeolite membrane reactor for WGS reaction of coal gasification gas for hydrogen production for integration with IGCC power plant.

To evaluate the performance and cost-effectiveness of this new membrane reactor process for use in 550 MW coal-burning IGCC plant with CO$_2$ capture.
Zeolite Membrane Reactor for Water-Gas Shift Reaction for CO₂ Capture

Zeolite Membrane Requirements:

- Operate at 350-550°C
- Chemically stable in H₂S, thermally stable at ~400°C
- Hydrogen permeance > 1x10⁻⁷ mol/m².s.Pa (>300 GPU)
- Hydrogen selectivity >25
MFI Type Zeolite

Structure of MFI type Zeolite (ZSM-5 or Silicalite)

Highly chemically and thermally stable (up to 700°C)

Surface and cross-section SEM images of (a, b) templated synthesized random oriented MFI membrane, (c, d) template-free synthesized random oriented MFI membranes (from Lin lab)
# Properties of Lab-Scale CVD Modified MFI Zeolite Membranes (Disk Substrates)

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Value</th>
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</thead>
<tbody>
<tr>
<td>$H_2$ Permeance in (mol/m².s.Pa)</td>
<td>$1-4 \times 10^{-7}$</td>
</tr>
<tr>
<td>$H_2$ Permeance in GPU</td>
<td>300-1200</td>
</tr>
<tr>
<td>$H_2$/$CO_2$ selectivity</td>
<td>20-140</td>
</tr>
<tr>
<td>$H_2$/$CO$ selectivity</td>
<td>50-200</td>
</tr>
<tr>
<td>$H_2$/$H_2O$ selectivity</td>
<td>120-180</td>
</tr>
<tr>
<td>$H_2$/$H_2S$ selectivity</td>
<td>100-180</td>
</tr>
<tr>
<td>Tested stability hours in syngas stream at 400 ppm $H_2S$ at 500°C</td>
<td>600</td>
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</tbody>
</table>

With equal-molar feed of $H_2$, $CO_2$, $CO$ and $H_2O$ at 500°C and 2 bar feed (Lin and Dong Labs)
Scope of work

1) Scaling up a zeolite membrane reactor from lab-scale to bench-scale for combined WGS reaction and H₂ separation

2) Conducting a bench-scale study using this zeolite membrane reactor for hydrogen production for IGCC with CO₂ capture.

Goal is to demonstrate effective production of H₂ and CO₂ capture by the bench-scale zeolite membrane reactor from a coal gasification syngas at temperatures of 400-550°C and pressures of 20-30 atm:

- Bench-scale zeolite membrane reactor: 21 zeolite membrane tubes of 3.5 ID, 5.7 OD and 25 cm L(active)

- A system producing H₂ at rate of about 1-10 kg/day, equivalent to a 1-10 kWₜₜ IGCC power plant
General Approach to Scaling up WGS
Zeolite Membrane Reactor

Single-tube zeolite membrane reactor: study WGS up to 30 atm by experiments and modeling

Intermediate-scale membrane reactor: 3-7 tube membrane module, and WGS reaction in the intermediate-scale reactor

Bench-scale membrane reactor: 21 tube membrane module, and WGS reaction in the bench-scale membrane reactor at NCCC

Membrane reactor in IGCC with CO₂ capture - process design and techno-economic analysis
Progress and Accomplishments

➢ Experimental Study on WGS in Lab-Scale Tubule Zeolite Membrane Reactor (Task 2.0)
➢ Modeling and Analysis for WGS in Zeolite Tubule Membrane Reactor (Task 3.0)
➢ Optimizing Tubule Support Fabrication (Task 4.0)
➢ Optimizing Zeolite Membrane Synthesis Methods (Task 5.0)
➢ Scaling up Synthesis of High Quality Zeolite Membranes (Task 6.0)
➢ Design and Fabrication of Zeolite Membrane Bundles and Modules (Task 7.0)
➢ Testing Zeolite Tube Bundles under Gasifier Conditions Including Membrane Reactor Configuration (Task 8.0)
Experimental Study on WGS in Lab-Scale Tubule Zeolite Membrane Reactor (Task 2.0)

**Subtask 2.1 Setting up high pressure WGS membrane reactor:**

Two single-tube zeolite membrane reactor systems were built (400-550°C, 30 bar), and used to study WGS reaction.

Setup at ASU
Subtask 2.3 Experiments on WGS in lab-scale zeolite membrane reactor

Experimental Study on WGS in Lab-Scale Tubule Zeolite Membrane Reactor (Task 2.0) (Cont’d)

Zeolite membrane:
- \((\alpha_{H_2/CO_2}) = 38,\)
- \(F_{H_2} = 300\) GPU
- Length=8cm

Conditions:
- \(T=500^\circ C,\)
- \(H_2O/CO=3-3.5\)
Morphology of Zeolite Membranes

A 10-cm long tubular MFI zeolite membrane was tested for more than six months under WGS reaction conditions at 500°C and reaction side pressure up to 26.5 bar. It was tested for over one week of WGS operation with feed CO containing 1,000 ppm of H₂S.
Modeling and Analysis for WGS in Zeolite Tubule Membrane Reactor (Task 3.0)

- \( \text{CO} + \text{H}_2\text{O} = \text{H}_2 + \text{CO}_2 \)
- \( \text{H}_2 \) permeation: Shell (reaction) side → Tube side
- Counter-current mode

**Conversion:** 
\[
X_{\text{CO}} = \frac{F_{\text{CO, feed}} - F_{\text{CO, reten}} - F_{\text{CO, perm}}}{F_{\text{CO, feed}}}
\]

**Purity:** 
\[
G_{\text{H}_2} = \frac{F_{\text{H}_2, \text{ perm}}}{F_{\text{total, perm}}}, \quad G_{\text{CO}_2} = \frac{F_{\text{CO}_2, \text{ reten}}}{F_{\text{total, reten}}}
\]

**Recovery:** 
\[
R_{\text{H}_2} = \frac{F_{\text{H}_2, \text{ perm}}}{F_{\text{H}_2, \text{ reten}} + F_{\text{H}_2, \text{ perm}}}
\]

**Capture:** 
\[
R_{\text{CO}_2} = \frac{F_{\text{CO}_2, \text{ reten}}}{F_{\text{CO}_2, \text{ reten}} + F_{\text{CO}_2, \text{ perm}}}
\]
Modeling and Analysis for WGS in Zeolite Tubule Membrane Reactor (Task 3.0) (Cont’d)

H₂ recovery > 92% & CO₂ capture > 90%
Optimizing Tubule Support Fabrication (Task 4.0)

**Ceramic Tube Extrusion**

**Intermediate Layer Deposition**
*Non-porous Tip and End Seals*

**End Seal (Open end)**

**Tip Seal (Closed end)**

**Zeolite Layer Deposition**
*(Silica CCD Modification)*

**MPT 57-tube Bundle**
*(Carbon Molecular Sieve Membrane)*

**Package into Multiple Tube Bundle**
*(Silica CCD Modification)*
Mechanical Strength after Immersion in Zeolite Synthesis Solution

**Challenge Conditions:** $T = 180^\circ C$; up to 48 hours; various NaOH and Zeolite Synthesis solution

- Approach #1: Nominal tube wall thicknesses of 1.1, 1.45, and 1.75mm tested
- Approach #2: Higher alumina content in tube (99%)

**Conclusion**

Thicker wall tube may be appropriate but not required.

Intermediate Layer Integrity/ Material Stability in Zeolite Synthesis Solution

**Challenge Conditions:** $T = 180^\circ C$; 48 hours; 2.7% NaOH

**Conclusion**

No impact on intermediate layer quality

Ceramic/Glass Sealant Material Stability in Zeolite Synthesis Solution

**Challenge Conditions:** $T = 180^\circ C$; 48 hours; 2.7% NaOH

**Conclusion**

Glass/Ceramic End Seal

*Post Zeolite Solution Challenge*

Haze and surface roughness development
Gas tight seal remains.
Demonstrate High Temperature Hydrothermal Stability of Membrane Bundle Components and Seals

Single Tube Bundle

Operating Conditions: \( T = 450^\circ C; \ P = 300 \text{ psig}; \ \text{Steam} = 80\% \ (\text{in} \ N_2) \)

Three Primary Scale-up Seal Components
- Ceramic Tip
- Impermeable End Seal
- Ceramic Tube Sheet and Ceramic/Glass Potting

Hydrothermal Stability Testing System

Results/Conclusions

1. No leak development over 185 days of hydrothermal stability challenge testing.
2. All seal component appear to be stable in the testing conditions.
Facility Establishment (design and construction):

- Membrane modification, stability test and WGS reaction system
## Optimizing Zeolite Membrane Synthesis Methods (Task 5.0) (Cont’d)

Best results achieved via in-situ synthesis with solutions of high pH (NaOH) and two-step CCD modification at 450 – 500°C.

<table>
<thead>
<tr>
<th>Method</th>
<th>Hydrothermal precursor</th>
<th>Seed layer</th>
<th>Conditions (hydrothermal &amp; CCD modification)</th>
<th>Quality before CCD</th>
<th>Quality post CCD</th>
</tr>
</thead>
<tbody>
<tr>
<td>In situ</td>
<td>SiO$_2$+H$_2$O+NaOH+TPAOH</td>
<td>No</td>
<td>Hydrothermal: 180°C/8h/rotation CCD: 450 – 500°C; two-step</td>
<td>Excellent</td>
<td>Excellent $\alpha_{\text{H}_2/\text{CO}_2}$ ~ 20 - 50</td>
</tr>
<tr>
<td>In situ</td>
<td>SiO$_2$+H$_2$O+AlCl$_3$+NaOH +TPAOH</td>
<td>No</td>
<td>Hydrothermal: 180°C/6h/rotation CCD: 450; two-step</td>
<td>Good</td>
<td>Good $\alpha_{\text{H}_2/\text{CO}_2}$ ~ 10 - 20</td>
</tr>
<tr>
<td>Secondary growth</td>
<td>TEOS+H$_2$O+AlCl$_3$+NaOH +TPAOH</td>
<td>Yes; Silica-lite</td>
<td>Hydrothermal: 165°C/6h/rotation CCD: 450C; two-step</td>
<td>Excellent</td>
<td>Good $\alpha_{\text{H}_2/\text{CO}_2}$ ~ 12</td>
</tr>
<tr>
<td>Secondary growth</td>
<td>TEOS+H$_2$O+AlCl$_3$+NaOH +TPAOH</td>
<td>Yes; ZSM-5</td>
<td>Hydrothermal: 165°C/6h/rotation CCD: 450C; two-step</td>
<td>Average</td>
<td>Poor $\alpha_{\text{H}_2/\text{CO}_2}$ &lt; 10</td>
</tr>
</tbody>
</table>
Optimizing Zeolite Membrane Synthesis Methods (Task 5.0) (Cont’d)

ZSM-5 Membranes on MPT Tubes
Scaling up Synthesis of High Quality Zeolite Membranes (Task 6.0)

Previous synthesis of 2.5-cm-diameter disc and 2-cm long φ1.0-cm tube (Pall Corp) membranes

Different size and geometry and surface chemistry of MPT tubes needs changes in: (1) zeolite precursor chemistry, (2) hydrothermal synthesis conditions, (3) calcination conditions, and (4) CCD modification conditions

Synthesis of 10-cm long φ1.0-cm tube (Pall Corp) membranes

Long tube and multi-tube synthesis (dead-ended and open-ended): (1) zeolite precursor chemistry, (2) hydrothermal synthesis conditions, (3) calcination conditions, and (4) CCD modification conditions

Preparation of 35-cm long tube membranes on MPT φ=0.57-cm tubes and scale-up to making multi-tubes in single batch
Performances demonstrated on the modified MFI zeolite membranes of different scale up stages – for separating 50v/50v H₂/CO₂ mixture at 450°C and 1 atm (1GPU=3.35×10⁻¹⁰ mol/m²·s·Pa)

<table>
<thead>
<tr>
<th>Support</th>
<th>Dimensions</th>
<th>Support Maker</th>
<th>Aₘ (cm²)</th>
<th>αₜₕ₂/CO₂</th>
<th>Pₘₜₕ₂ (GPU)</th>
<th>Target</th>
</tr>
</thead>
<tbody>
<tr>
<td>Disc</td>
<td>D = 2.5-cm</td>
<td>UC lab</td>
<td>2.5</td>
<td>62</td>
<td>~390</td>
<td>--</td>
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<tr>
<td>Tube</td>
<td>L=8cm; Lₜₘ =1.5 cm; OD=1cm; ID=0.7 cm</td>
<td>Pall Co.</td>
<td>4.7</td>
<td>&gt;100</td>
<td>~806</td>
<td>--</td>
</tr>
<tr>
<td>Tube</td>
<td>L=8cm; Lₜₘ =6 cm; OD=1cm; ID=0.7 cm</td>
<td>MPT</td>
<td>11.0</td>
<td>45±5</td>
<td>360±50</td>
<td>45</td>
</tr>
<tr>
<td>Tube</td>
<td>L=35cm; Lₜₘ =25 cm; OD=1cm; ID=0.7 cm</td>
<td>MPT</td>
<td>44.8</td>
<td>41±5</td>
<td>725±50</td>
<td>45</td>
</tr>
</tbody>
</table>
Design and Fabrication of Zeolite Membrane Bundles and Modules (Task 7.0)

Single, 3- and 7-tube alumina membrane “bundles” for use in the high temperature hydrothermal pressure testing.
Testing Zeolite Tube Bundles under Gasifier Conditions Including Membrane Reactor Configuration (Task 8.0)

Modify NCCC Test Rig for Gasifier Off-gas Challenge Testing of Zeolite Membrane and Bundle Components and Seals. Target Conditions: 450°C and 300psig; no pretreatment (NCCC max operating conditions available); Single tube, 7-tube, and 21-tube bundles.

**WPI/MTR skid:** Oven and system components approved for use by NCCC under proposed operating conditions.

**MPT Modifications:** Target automated operation and remote monitoring for continuous long term testing of proposed membrane technology.

**MPT Modifications**
Testing of several membrane bundles/modules simultaneously.

**Components and Subsystems:**
1. Original WPI Oven
2. Proposed Main Electronics Control Box
3. Proposed Mass Flow Meter Box
4. Knockout Tanks A, B, C, D
5. Automated Syngas Shut-Off Valve

**Interior dimensions:** 48w x 24h x 24 deep
ZMR Integration Methodology

- **Reference Case IGCC - Case 2 (GE Gasifier with Selexol AGR and GE F-class gas turbines) in the 2013 DOE/NELT Report 1397 on “Bituminous Coal and Natural Gas to Electricity, Rev 2a”**
- **IGCC Design to NETL’s QGESESS Guidelines**
- **Cost Estimation and Financial Modeling Methodology**
  - For process systems associated with the ZMR WGS and CO₂ capture technologies, Nexant will carry out preliminary process design to establish system performances and develop major equipment-factored capital costs. Costs for proprietary equipment will be provided by technology licensors.
  - For process and support systems that are unrelated to the ZMR WGS and CO₂ capture technology, performances and capital costs will be scaled from the NETL Reference Case 2 according to capacity factors established by process H&MB, and by overall utility and commodity material balances.
Plan for Budget Period 2 Work
### Budget Period 2 Project Tasks & Schedule

<table>
<thead>
<tr>
<th>Task</th>
<th>Description</th>
<th>Quarter 1</th>
<th>Quarter 2</th>
<th>Quarter 3</th>
<th>Quarter 4</th>
<th>Quarter 5</th>
<th>Quarter 6</th>
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<tbody>
<tr>
<td>Task 10.0</td>
<td>Modeling and Analysis of WGS in Bench Scale Zeolite Membrane Modules for WGS (ASU)</td>
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<tr>
<td>Subtask 10.1</td>
<td>Modeling and analysis of WGS in multi-tube membrane reactor module</td>
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<tr>
<td>Subtask 10.2</td>
<td>Optimization of operation conditions for WGS in zeolite membrane module</td>
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<td>Task 11.0</td>
<td>Fabrication of Large Quality Tubular Supports (MPT)</td>
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<td>Task 12.0</td>
<td>Preparation of Large Quantity MFI Zeolite Tube Membranes for Bench-Scale Module (UC)</td>
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<tr>
<td>Subtask 12.1</td>
<td>Identifying conditions for fabrication of large quantity of zeolite membrane tubes</td>
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<tr>
<td>Subtask 12.2</td>
<td>Fabrication of 200-300 zeolite membrane tubules with desired quality</td>
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<tr>
<td>Task 13.0</td>
<td>Design and Fabrication of Bench-Scale Zeolite Membrane Module Housing with Seals (MPT/UC/ASU)</td>
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<td>Task 14.0</td>
<td>Building Bench-Scale Zeolite Membrane Reactors (MPT/ASU/UC)</td>
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<tr>
<td>Subtask 14.1</td>
<td>Fabrication and evaluation of WGS catalyst for bench-scale WGS reaction (ASU)</td>
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<td>Subtask 14.2</td>
<td>Assembling and testing bench-scale zeolite membrane reactor (MPT/UC/ASU)</td>
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<td>Subtask 14.3</td>
<td>Modification and installation of the membrane reactor testing skid (MPT/ASU)</td>
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<td>Task 15.0</td>
<td>Testing WGS in Bench-Scale Membrane Reactor (MPT)</td>
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<tr>
<td>Task 16.0</td>
<td>Process Design, Techno-Economic and EH&amp;S Analyses (MPT)</td>
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<td>Subtask 16.1</td>
<td>Design of Commercial Scale WGS Zeolite Membrane Reactor and Process (MPT/Nexant)</td>
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<td>Subtask 16.2</td>
<td>Techno-Economic Analysis (TEA) of IGCC Plant (Nexant)</td>
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<td>Subtask 16.3</td>
<td>Preliminary Technology EH &amp; S Assessment (MPT)</td>
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</table>
Bench-Scale Zeolite Membrane Reactors

21 Tube, $H_2$ Production Rate
1-10 kg/Day
Modeling of 21 Tube Bench-Scale ZMR: 3-D CFD Simulation Approach

Software: ANSYS Fluent

Geometry creation

Mesh generation

3-D model and Boundary conditions

Solution parameters

Calculate a solution

Converged?

Yes

Accurate?

Yes

Stop

No

Modify Boundary conditions or solution parameters

No

Modify Boundary conditions or solution parameters

Yes

Calculate a solution

Converged?

No

Modify Boundary conditions or solution parameters

No

Modify Boundary conditions or solution parameters

Yes

Converged?

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Converged?

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Modify Boundary conditions or solution parameters

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Yes

Converged?
Conclusions

- Zeolite membranes fabricated on cost-effective industrial alumina substrates
- CCD modified MFI zeolite membrane synthesis scaled up to longer alumina substrates
- High pressure and temperature intermediate scale zeolite membrane modules built and tested
- WGS catalysts studied and model for WGS zeolite membrane reactors established
- TEA model for zeolite membrane reactor –IGCC process established
- The project positioned to move to the second phase (Budget Period 2).