

Simulations and reduced order modeling of CO₂-cement systems



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Cement and CO₂ Interactions

- Cement in damaged wells will be in contact with brine and CO₂.
- Reaction between cement and carbonated brine results in altered cement layers with different physical and chemical properties.
- Reactions can also result in calcite precipitation within the fracture.



Coupled chemical, mechanical, transport model

- Two-phase flow model
 - Mass balance for brine and CO₂.
 - Extension of Darcy's law to multiphase flow:

$$\boldsymbol{v}_i = -\frac{kk_{r,i}}{\mu_i} (\nabla p_i - \boldsymbol{g}\rho_i)$$

- Reaction front model
 - Transport between the fronts is via diffusion:

$$\frac{\partial}{\partial x} \left(D_{eff} \frac{\partial [\mathrm{E}]}{\partial x} \right) = 0.$$

 Front movement is controlled by diffusion or reaction based on which phenomenon is slower:

$$\left[\left[c_{\rm E}(1-\phi) \right] \right] v_{front} = -\left[\left[D_{eff} \frac{\partial [{\rm E}]}{\partial x} \right] \right] or - r_{\rm E}.$$

- Mechanical model
 - Altered cement has lower stiffness, which may also lead to fracture sealing. This is captured by coupling the mechanical response of the fracture to the extent of reaction.



Logarithmic inputs and outputs yield reduced order models that capture the simulated CO₂ fluxes



- Quasi-Newton sampling and multivariate adaptive regression splines are used to construct dynamic ROM for CO₂ leakage rate
- Our results suggest that, for dynamic ROM, temporal coordinate might need to be separated out from other inputs
- In the future work, we will compute ROM coefficients for each time step then construct ROM for these time-varying coefficients

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