Performance and Microstructural Changes in LSM-Based SOFC Cathodes Under Accelerated and Conventional Testing

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Outline

- Project objectives and approach
- Accelerated vs. long-term conventional testing
- LSM compositions: role of A-site deficiency
 - Durability testing
 - Cathode microstructural changes
 - ASR and TPB density vs. time
- Summary & conclusions
- Ongoing & future work



Background observations: long-term conventional testingkh, 860 °C:

Cathode densification layer



Densified layer (between two white lines) is ~5 µm thick (SEM image, courtesy of LGFCS)

(CCC: cathode current collector)

Segregation of Mn oxides*



TEM image with EDXS mapping (LSM: blue; zirconia red; MnOx green) Total of first layer + second layer: 5 μm

*) H.-J. Wang, M. R. De Guire, G. Agnew, R. Goettler, Z. Liu, Z. Xing, A. Heuer, *Met. Mater. Trans. E*, **1** [3] 263-271 (2014). DOI: 10.1007/s40553-014-0026-5.



Project Objectives and Approach

- Implement an accelerated testing protocol to *replicate long-term microstructural changes* in *shorter times*
- Understand *microstructural basis of long-term performance loss* in LSM-based SOFC cathodes
- 4 cycles of cathode formulation, testing, and analysis in 3 years
- Develop strategies for *optimizing LSM-based cathodes* for improved long-term performance and stability





Procedures: button cell specifications

- Fabricated at LGFCS
- Cell details:
 - 8YSZ electrolyte, 32 mm dia.
 - NiO-8YSZ anode (60:40 wt%)
 - Cathodes: A-site deficient LSM + 8YSZ (50:50 wt%)
 - <u>Comp'n A</u>: (La_{0.85}Sr_{0.15})_{0.90}MnO_{3±δ}
 (LSM 85-90)



- <u>Comp'n B</u>: (La_{0.80}Sr_{0.15})_{0.95}MnO_{3±δ} (LSM 80-95)
- Electrodes: screen printed, 9.5 mm dia., fired separately



Procedures: button cell testing

- Pt mesh and wires attached to both electrodes
- Cell sealed to zirconia tube with fired glass paste
- Anode reduction followed by 24-h burn-in at OCV
- Pre-test protocols: details below
- <u>Durability testing</u>
 - H₂, 50 sccm
 - <u>Accelerated tests</u>: 1000 °C, 0.760 A cm⁻²
 - <u>Conventional tests</u>: 900 °C, 0.380 A cm⁻²
 - I-V and EIS scans every 24 or 48 h





Pre-test protocol: temperature parametric study



LSM 80-95 (B) durability testing: reproducibility





Representative V-I & P-I sweeps, 0-624 h





Comparative V-I & P-I sweeps, A vs. B

V-I & P-I Curves





ASR and changes over time: summary

• In durability testing and EIS: *LSM 80-95* (B) had

lower:

- Initial ASR
- Final ASR
- $\triangle ASR$ over time (Ω cm² kh⁻¹)



Procedures: FIB Slice & View for 3DR





LSM 85-90 (A) microstructural evolution: 3D

reconstruction

LSM 85-90 cathode:	as- received	after 200 h accel'd testing	after 493 h accel'd testing	<u>a</u>	<u>as t ↑</u>
pore size (µm)	0.20	0.34	0.42		1
pore tortuosity	2	1.7	1.6		•
normalized pore surface area (µm ⁻¹)	26	17.4	14.2		•
total TPB (µm ⁻²)	17.1	9.6	5.86		•
active TPB (µm ⁻²)	10.3	8.2	5.13		•

Coarsening of pores, loss of pore area and TPB

Other observations —

- Phase fraction profiles: flat across cathode
- Densification at cathode-electrolyte interface? Inconclusive



LSM 85-90 (A) microstructural evolution: TEM

as reduced







MnO_x dispersed across cathode and CCC

Accumulation of MnO_x at cathode-electrolyte interface

Other observations —

- LSM and YSZ composition profiles: flat across cathode
- Densification at cathode-electrolyte interface? Inconclusive



LSM 80-95 (B) microstructural evolution: TEM

- As received:
 MnO_x (→)
 only in CCC
- 500 h accel'd testing: occasional small MnO_x grains near electrolyte interface





A – B comparison: TEM

LSM 85-90 (A), accel'd

0 hrs 500 hrs 500 hrs 2.5 µm

LSM 80-95 (B), accel'd

- More MnO_x observed in LSM 85-90 (untested and tested)
- Larger MnO_x particles in LSM 85-90 CCC
- More pores in LSM 80-95 cathode post-testing

A – B comparison: 3D reconstruction

		LSM 85-90 (composition A)			LSM 80-95 (composition B)		
		as received	500 h conv test	493 h accel. test	as received	500 h conv test	500 h accel'd test
sample volume (µm ³)		4350	3700	4525	6300	5000	5096
volume fraction (%)	porosity	17	21.9	18.4	29	26	26
	YSZ	42	42.6	43.2	33	35.5	35
	LSM	41	35.5	38.4	38	38.5	39
particle diameter (μm) LSN	porosity	0.2	0.4	0.42	0.46	0.45	0.38
	YSZ	0.5	0.5	0.46	0.47	0.42	0.51
	LSM	0.6	0.65	0.6	0.67	0.65	0.7
tortuosity YS LSI	porosity	2.0	1.65	1.6	1.34	1.4	1.67
	YSZ	1.5	1.47	1.3	1.32	1.65	1.66
	LSM	1.3	1.45	1.4	1.3	1.5	1.44
normalized surface area (µm ⁻¹)	porosity	26	15.7	14.2	13	13.3	15.9
	YSZ	12	11.5	13	13	14	11.9
	LSM	10	8.9	9.9	8.9	9.3	8.5
Total TPB (µm ⁻²)		17.1	11	5.9	14.5	14.2	14.8
Active TPB (µm ⁻²)		10.3	9.5	5.1	13.0	13	12.5

In contrast to LSM 85-90 (A), LSM 80-95 (B) shows:

- Pore refinement (!?) and increasing area and tortuosity
- Stabler TPB (total and active)

A – B comparison: ASR and TPB density

- LSM 85-90, as t 1:
 - active TPB density
 - ASR 🛧
- LSM 80-95:
 - Higher active TPB density
 - Lower ASR
- Overall: inverse correlation between ASR and TPB density





Summary & Conclusions

- During accelerated testing up to 500 h:
 - *LSM 85-90* (A) cathode:
 - Pore coarsening
 - MnO_x segregation at electrolyte-cathode interface
 - Microstructure-performance trend over time:
 - TPB density ✓
 ASR ↑
 - *LSM 80-95* (B) cathode:
 - Stabler microstructure Less A-site deficient \rightarrow *less MnO_x*
 - *Higher TPB, lower ASR* than LSM 85-90
 - Not yet observed:
 - Cathode densification at electrolyte
 Mn depletion at electrolyte

Inverse TPB - ASR relationship is emerging



Ongoing & Future Work

- Continue reproducibility studies
- 624-h accelerated test: microstructural analysis underway; look for densification layer
- Thermodynamic studies to predict conditions for $\mbox{MnO}_{\rm x}$ formation
- MnO_x formation: symptom, or cause, of degradation?
- Continue to explore relationship between TPB and ASR
 - vs. LSM composition
 Accelerated vs. conventional testing
- Composition C cells fabricated; testing & analysis are underway



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3DR in SOFCs: Triple phase boundaries (TPB)



- YSZ: 100% ionic conductor
- LSM (cathode) and Ni (anode): 100% electronic conductors
- For TPB to be *active electrochemically*, it must have *percolation paths:*
 - Ionic conductor must connect to electrolyte
 - *Electronic conductor* must connect to *current collector*
 - Pore must connect to external atmosphere

3DR can definitively determine whether TPB is *active*, or may be *inactive*



LSM 80-95 durability testing: 624 h





Cathode B: 500-hr Conventional Test

V-I and P-I Sweeps



Representative Bode plots, 0-624 h





Representative Nyquist plots, 24-400 h





Cathode B: 500-hr Conventional Test



Cathode B: 500-hr Conventional Test



Ζ'

30



Microstructural studies

Progress as of 2016-03-08

	GenA bu	itton cell	GenB button cell		
	TEM	3D	TEM	3D	
As-received	\checkmark	\checkmark	\checkmark	\checkmark	
72 h accelerated	\checkmark				
200 h accelerated	\checkmark	\checkmark			
500 h accelerated	\checkmark		\checkmark		
500 h	\checkmark				

TEM w/EDXS mapping

- As reduced (0 h)
 - MnO_x (red arrows) observed sparingly across entire cathode
- 72 h and 493 h accelerated testing
 - MnO_x near cathode/electrolyte interface
 - MnO_x also observed in LSM cathode current collector (CCC) for 500 h 500 h
 - Smaller pores, but no obvious densification layer



e'lyte

SM-8YSZ cathode



e'lvte

LSM-8YSZ cathode





distance from electrolyte interface (μm)

Mn 50

10

Sr 5

0

0

Cation %

distance from electrolyte interface (μm)

- Uniform LSM composition across cathode and CCC
- Same composition as in as-reduced cell (not shown)

TEM w/EDXS of bulk 8YSZ composition

72 h

493 h



- Uniform YSZ composition across cathodes
- 4 5 cat% Mn

Overview: 3D Reconstruction Process

1: Sample Preparation

- Impregnate with epoxy
- Mount with SOFC layers exposed on two sides
- Polish specimen
- Coat with Pd



2: Preparing Area of Interest

- Deposit Pt to protect area of interest
- Focused Ga-ion beam (FIB): prepare two side trenches, one front trench



3: Data Collection

- Iteratively "slice and view":
 - FIB sections, 150 nm thick
 - Each section imaged in SEM



4: Data Processing

- Phase segmentation
- Synthesize stack of 2D images
- Calculate:
 - volume fractions
 - particle diameters
 - tortuosity
 - triple phase boundary (TPB) density



Making steps 1 through 4 appear one at a time could be helpful here. Otherwise the slide is a bit overwhelming at first sight. Before you leave this slide, you should briefly point out what aspects of your sample preparation were not routine, or for which you deviated from standard practice in ways that improved your analyses. Mark De Guire, 10/28/2014

3DR of cathode A — accelerated testing





surfaces near cathodeelectrolyte interface

		Gen A			
		as received	200 h accel.	493 h accel.	
sample volume (µm ³)		≈ 4350	≈ 4620	≈ 4525	
volume fraction (%)	porosity	17	17	18.4	
	YSZ	42	41	43.2	
	LSM	41	42	38.4	
particle diameter (µm)	porosity	0.2	0.34	0.42	
	YSZ	0.5	0.6	0.46	
	LSM	0.6	0.7	0.6	
tortuosity	porosity	2	1.7	1.6	
	YSZ	1.5	1.43	1.3	
	LSM	1.3	1.35	1.4	
normalized	porosity	26	17.4	14.2	
surface area	YSZ	12	10	13	
(µm ⁻¹)	LSM	10	7.6	9.88	
Total TPB (µm ⁻²)		17.1	9.6	5.86	
Active TPB (µm ⁻					

		Ge	en A	Gen B		
		as reduced	493 h accel.	as received	500 h accel	
sample volume (µm ³)		≈ 4350	≈ 4525	≈ 6300	≈ 5096	
volume fraction (%)	porosity	17	18.4	29	26	
	YSZ	42	43.2	33	35	
	LSM	41	38.4	38	39	
particle diameter (µm)	porosity	0.2	0.42	0.46	0.38	
	YSZ	0.5	0.46	0.47	0.51	
	LSM	0.6	0.6	0.67	0.7	
tortuosity	porosity	2	1.6	1.34	1.67	
	YSZ	1.5	1.3	1.32	1.66	
	LSM	1.3	1.4	1.3	1.44	
normalized surface area (µm ⁻¹)	porosity	26	14.2	13	15.88	
	YSZ	12	13	13	11.88	
	LSM	10	9.88	8,9	8.5	
Total TPB (µm ⁻²)		17.1	5.86	14.5	14.8	
Active TPB (μm ⁻²)		10.3	5.13	13.0	12.5	

Total TPB – ASR relationship



Active TPB – ASR relationship

