

H₂S Resistant Steam Reforming Catalysts for Upgrading Coal Derived Syngas for Coal-to-Liquids Production

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Project Background

DOE NETL Cooperative Agreement DE-FE0012054

- Southern Research in cooperation with US Department of Energy-NETL is developing a high concentration H₂S resistant reforming catalyst for increasing syngas yields derived from low rank coal and biomass gasification.
- Conventional low rank coal and biomass gasified syngas streams contain sulfur, residual methane, tars and ammonia which are usually removed via gas clean up processes before Fischer-Tropsch reactor to produce liquid fuels.
- Highly adaptable sulfur resistant catalyst technology for coal and coal biomass syngas upgrading potentially allows lowering the capital costs, GHG emissions (via biomass utilization) and is an easy retrofit to existing gasifiers. It allows US coal industry to further improve process carbon efficiencies.

Project Goals

- □ To increase the yield and possibly H₂:CO ratio of a raw low-rank coal syngas by high temperature reforming prior to sulfur removal
- \Box Develop catalysts for simultaneously reforming of tars, methane, and C₂-C₆ hydrocarbons, and decomposing NH₃ in the presence of H₂S
- Demonstrate the performance of the catalyst(s) in a lab-scale reactor version of the catalytic reforming process at commercially viable conditions
- Preliminary techno-economic model that will be used to evaluate the commercial viability of this process relative to the current state-of-the art

Potential Technical Advantages

- Tar, methane, light hydrocarbons, NH₃ conversion to additional CO and H₂ in one step
- Reduce or eliminate requirements for a water-gas shift step
- Potential for Fischer-Tropsch (FT) tail gas to provide heat for the reforming and increase syngas yield to FT liquids
- Sulfur tolerant reforming catalyst allows use at any stage of cleanup
- Adaptable to all coal or coal-biomass gasification technologies
- Increasing the energy efficiency of gasification using hightemperature gas clean up



- Lab scale micro-reactor system constructed for tests
- 12 reforming catalysts synthesized, characterized and screened for performance using simulated TRIG and Lurgi gasifier syngas with up to 500 ppm H.S.
- Three catalysts showed high reforming activity in the presence of H₂S for extended periods of time



Process Flow Diagram and Results

- Steam reformer prior to cooling syngas
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- POX on recycled gas or natural gas to generate heat for reformer
 No tar and ammonia cleanup processes needed
- High-activity liquid-selective FT Catalyst that does not produce wax
 (SR/NETL Project DE-FE0010231)
- · Little to no product upgrading

Feed Compositions: Protein R F and R F Web (2006) Prince

robachi, e. r. and e. z. fittas (2000), 51 deal 5. (2000).								
<u>TRIG -</u> lignite	<u>29.2%</u>	<u>34.3%</u>	<u>13.6%</u>	<u>2.5%</u>	<u>18.9%</u>	<u>100-250 ppm</u>	.28%	.10%
Lurgi's FBDB -	18.7%	7.5%	15.7%	5.2%	51.6%	.29%	.58%	.41%

Catalytic Studies Results:

 \square Results obtained over Southern's proprietary catalysts show resilience to H₂S with high activity as compared to five comparable literature/commercial catalysts





• 112 + 00 % Change _ 12/00 Ratio



additional steam.

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Catalyst Stability and Preliminary

Economics



- No effect of 90 ppm H₂S on methane conversion
- Reduced conversion in 500 ppm H₂S, but
- Minimal reduction of performance over 200 hours of exposure to high concentration of H₂S

Preliminary Economics for CTL with Iron catalyst



Summary

- I. Promising catalyst candidates synthesized and tested at lab scale with 35 to 500 ppm $\rm H_2S$
- II. Catalyst studied showed no significant reduction in activity for ~ 220 hours for 35 ppm, 50 hours for 90 ppm H_2S , and significant tolerance to 500 ppm H_2S
- III.Significant increase in syngas yields using raw syngas feed, however no appreciable changes in H₂:CO ratio without additional steam
- IV.Preliminary TEA modeling of 50,000 bpd plant indicates greatly improved FT yield for given coal feed rate (less coal needed for given FT output), but near identical costs for FT production due to much less power produced. CO₂ emission costs and power price have large impacts on economics. 50% reduction in carbon emissions per barrel of FT liquids.

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