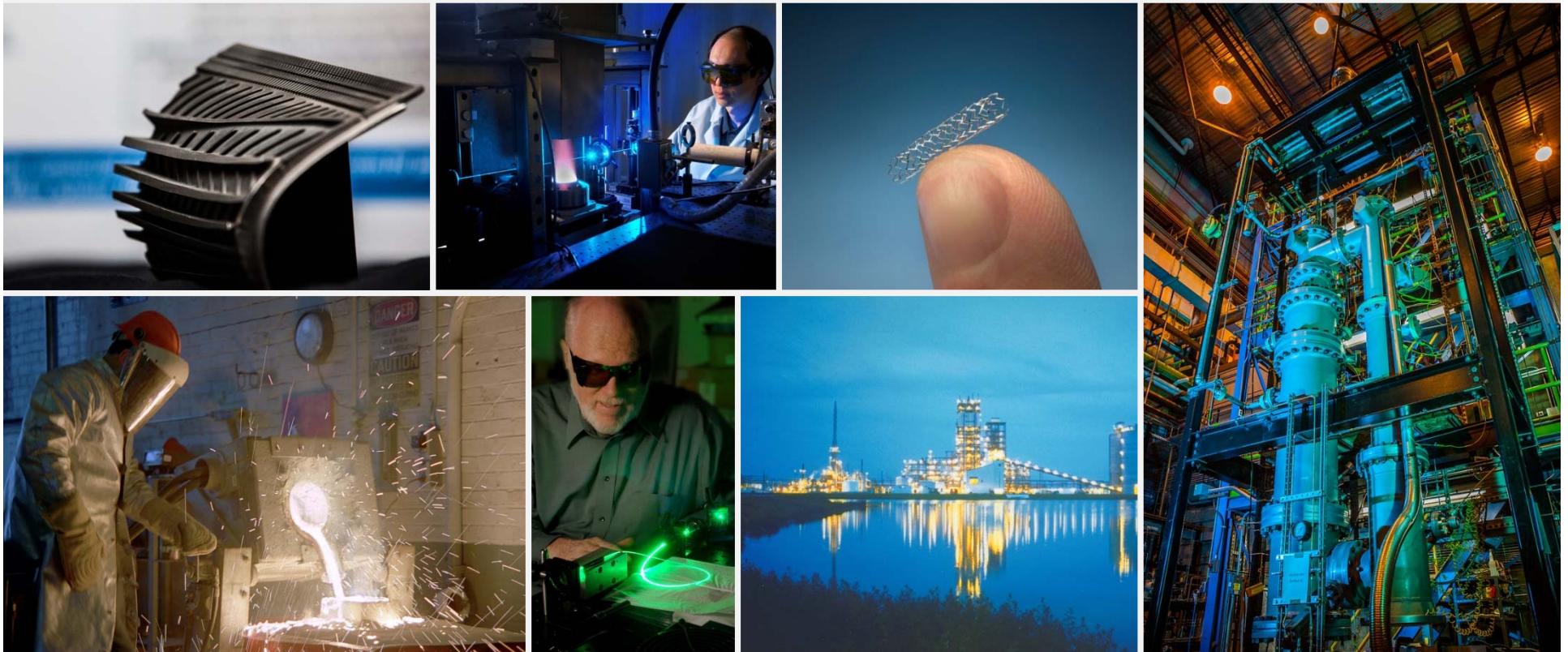




*Driving Innovation ▪ Delivering Results*



## New Advances for Fischer-Tropsch Catalysis

### Christopher Matranga & Dushyant Shekhawat

Office of Research and Development  
August 2015



National Energy  
Technology Laboratory

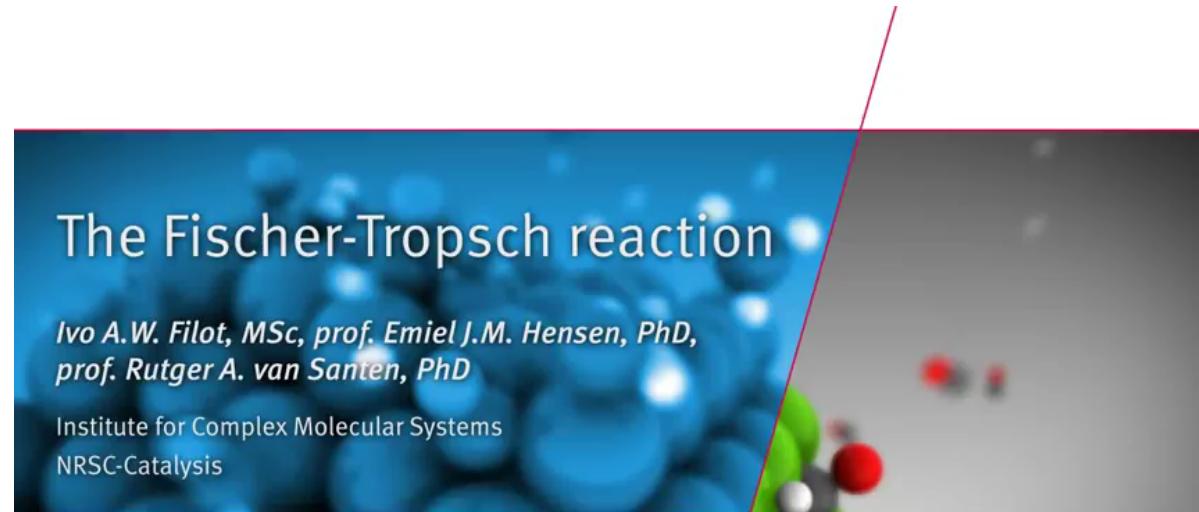
# A Simplified View of Fischer-Tropsch Mechanistic Chemistry



## 6 general rxn steps:

(independent of model, catalysts,  
rxn conditions)

- Adsorption/activation
- Chain initiation
- Chain growth
- Product desorption
- Chain termination
- Readsorption/further rxn



## 3 widely considered mechanisms on Fe:

- Surface carbide (shown)
- Surface enol
- CO insertion

<https://youtu.be/44OU4JxEK4k>

**Youtube Movie Credit:** I. Filot, E. Hensen, R. van Santen

Institute for Complex Molecular Systems, Eindhoven University of Technology



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Figure from Weckhuysen Chem. Soc. Rev., 2008, 37, pgs 2758–2781

# A Simplified View of Fischer-Tropsch Mechanistic Chemistry

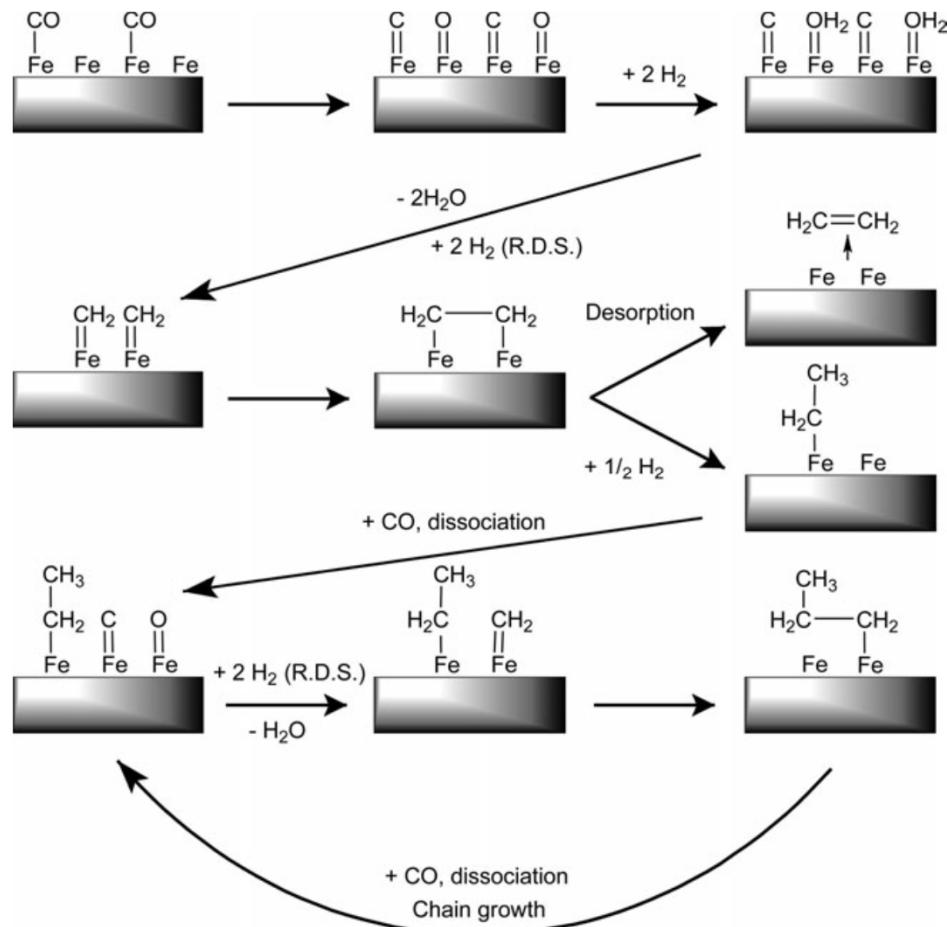


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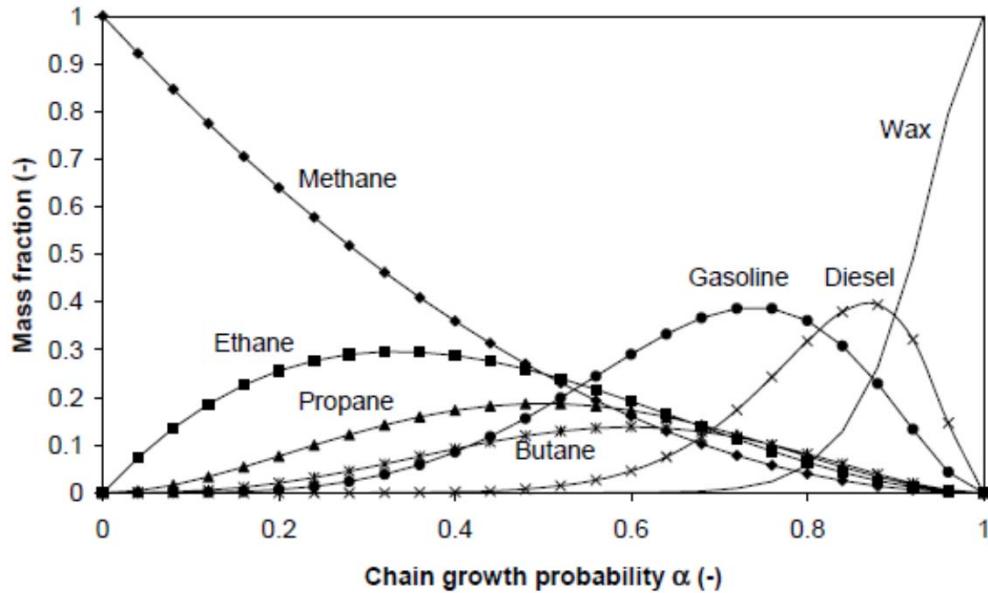
## Surface Carbide Mechanism



## 3 widely considered mechanisms on Fe:

- Surface carbide (shown)
- Surface enol
- CO insertion

# Anderson-Shultz-Flory (ASF) Product Distributions



- Radical polymerization type distro
- Chain termination/progation are critical/rate-determining steps
- Occur for fully thermalized, “equilibrium” or “steady state” conditions

	Chain growth probability	Olefin/paraffin ratio	Carbon deposition	Methane selectivity
Temperature ↑	↓	↓	↑	↑
Pressure ↑	↑	*	*	↓
H <sub>2</sub> /CO ratio ↑	↓	↓	↓	↑
Conversion ↑	*	↓	↑	↑
Space velocity ↑	*	↑	*	↓

$$m_n = (1 - \alpha)\alpha^{n-1}$$

$$\alpha = \frac{R_p}{R_p + R_t}$$

# Are Controlled Deviations from ASF Possible?



## Nano-structured Catalyst Materials

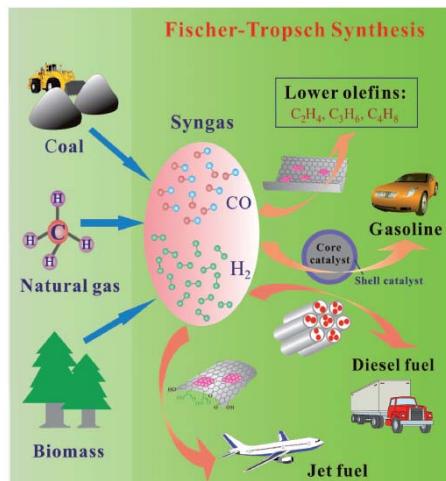
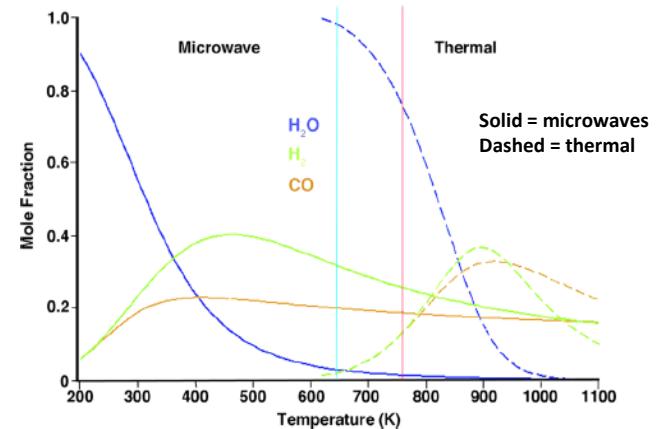


Image from Wang et. al. ChemCatChem Doi: 10.1002/cctc.201000071

### Potential Benefits

- Non Anderson-Shultz-Flory Product Distributions
- Stabilization of active catalyst phase
- Controlled production of oxygenates/aromatics
- Improved reactivity & conversion

## Non-equilibrium Reactors (Microwave-MW)



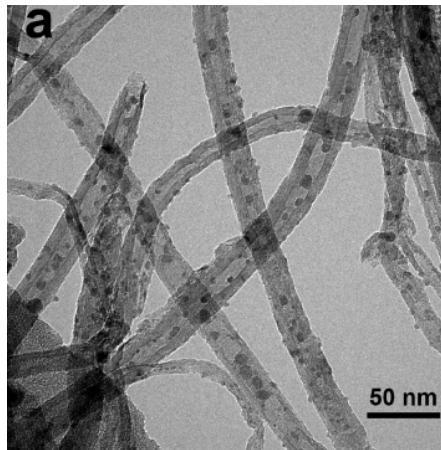
### Potential Benefits

- Non-thermal & Non-ASF Product Distributions
- Lowered reactor temperatures
- Improved kinetics, reactivity & conversion

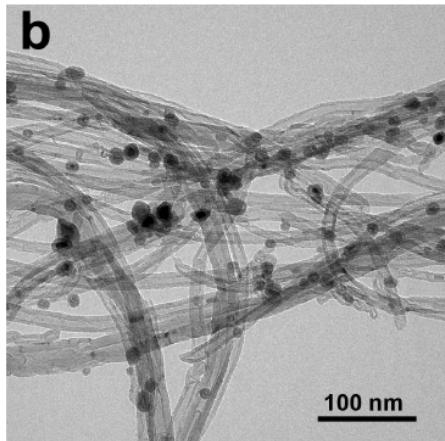
# Motivation from Previous Literature: Process Intensification w/Nano-Catalysis



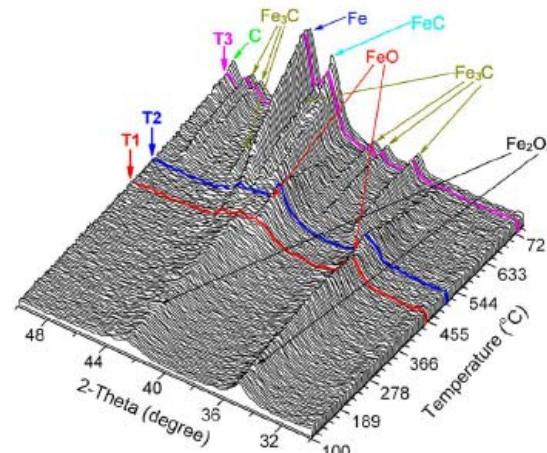
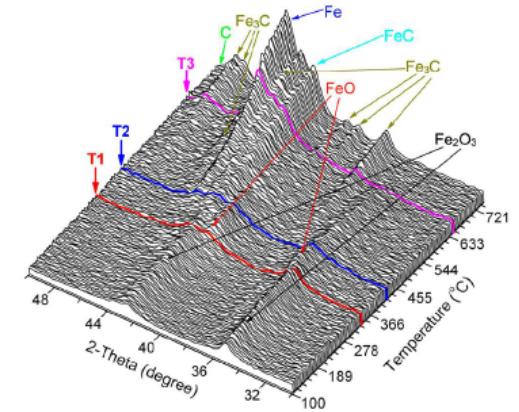
$\text{Fe}_2\text{O}_3$  in Carbon Nanotubes



$\text{Fe}_2\text{O}_3$  outside Carbon Nanotubes



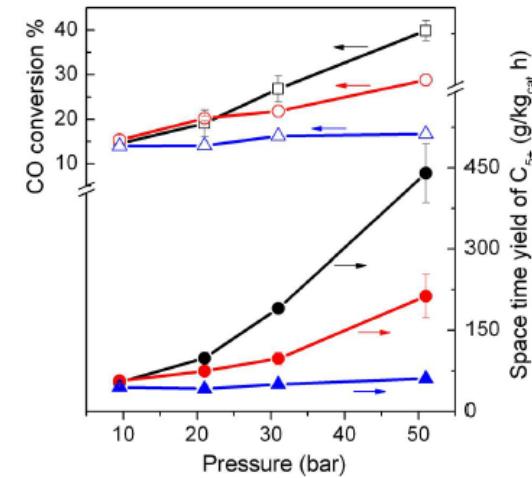
Fe & Fe-carbide form at lower T  
For  $\text{Fe}_2\text{O}_3$  inside CNTs



Altered Product Distros

Table 2. Comparison of the FTS Activity and Product Selectivities at 51 bar

catalyst	CO conversion (%)	yield (g $\text{C}_{\text{sat}}$ /kg <sub>cat</sub> ·h)	$\text{CO}_2$ selectivity (%)	hydrocarbon selectivities (%)		
				$\text{CH}_4$	$\text{C}_2-\text{C}_4$	$\text{C}_5+$
Fe-in-CNT	40	440	18	12	41	29
Fe-out-CNT	29	210	12	15	54	19
Fe/AC	17	61	5	15	71	9



Legend:

- Black = Fe inside CNTs
- Red = Fe outside CNTs
- Blue = Fe on act. carbon



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Data & Images from Bao et. al. JACS, doi: 10.1021/ja8008192



# Recent Nano-catalyst results from ORD



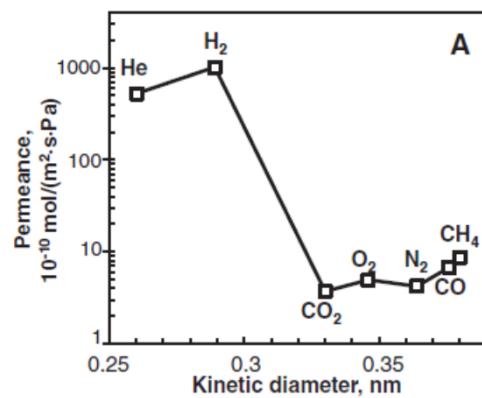
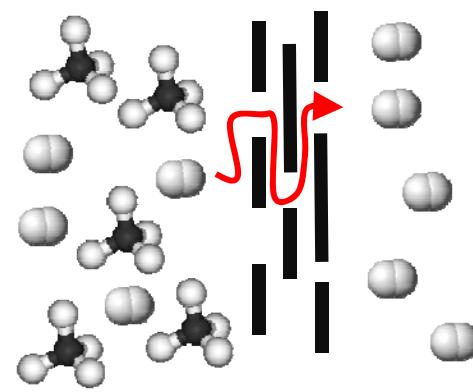
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# Layered Graphene Catalyst Supports for Breaking ASF Distributions

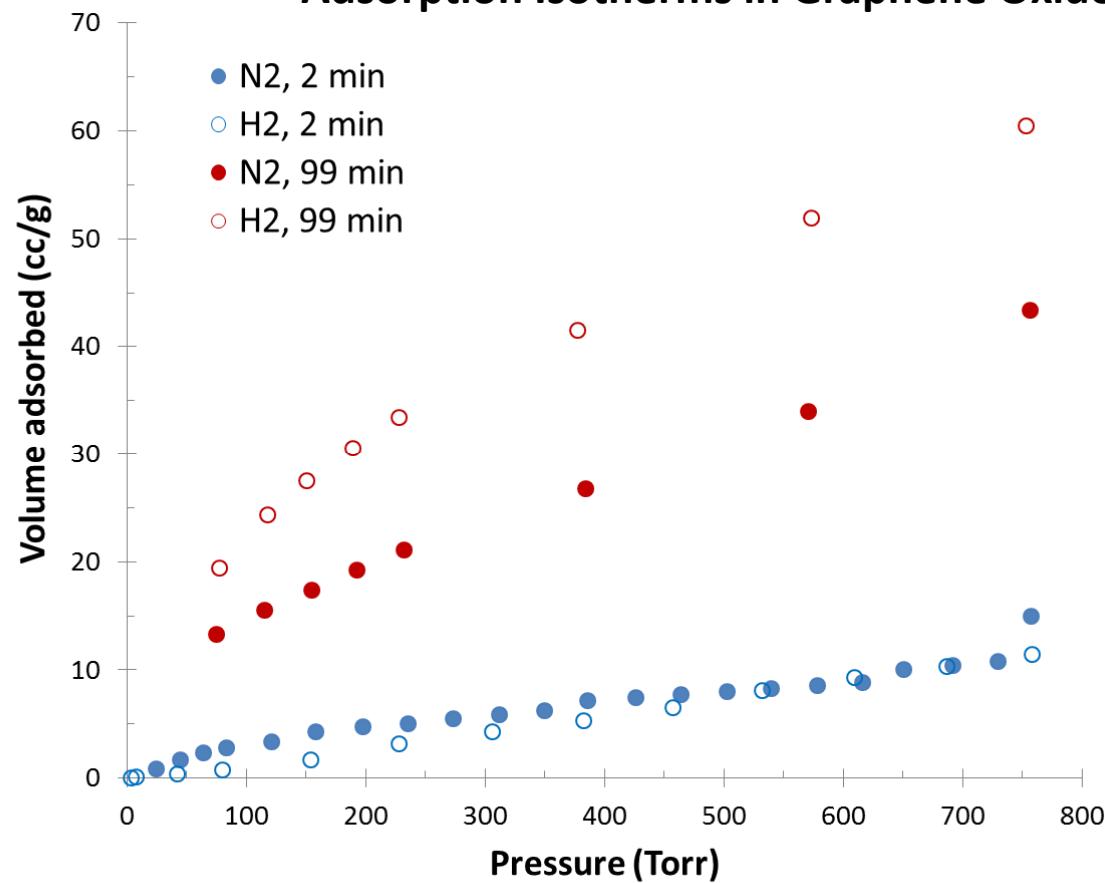


Can this be exploited for FT ?

Layered graphene controls  
surface mobility of FT-type species  
(Li et. al., Science, 2013)



Adsorption Isotherms in Graphene Oxides

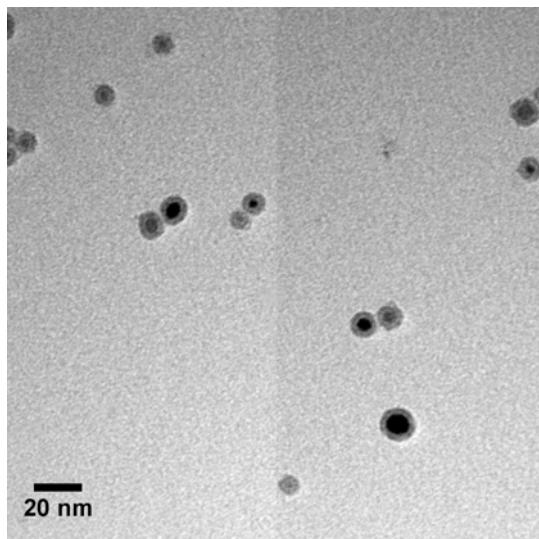


- Surface mobility disrupted (kinetic effect)
- More H<sub>2</sub> adsorbed than N<sub>2</sub> (size exclusion)

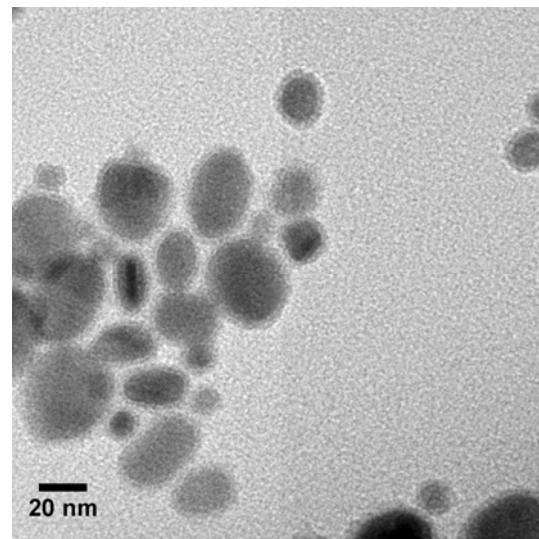
# Nanostructured $\text{Fe}_5\text{C}_2$ "Häggs Phase"



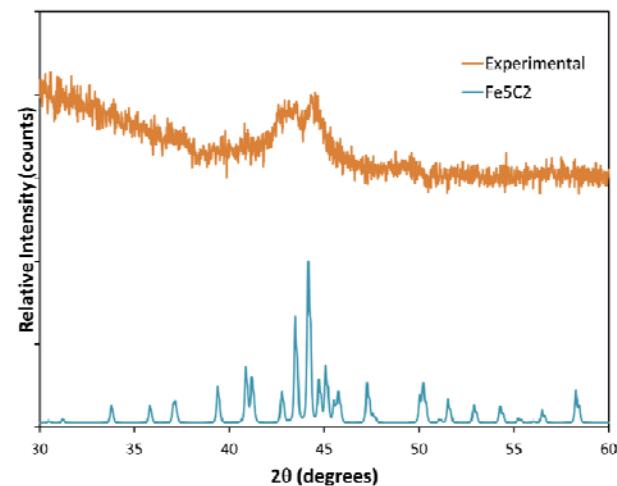
10 nm  $\text{Fe}_5\text{C}_2$



20 nm  $\text{Fe}_5\text{C}_2$

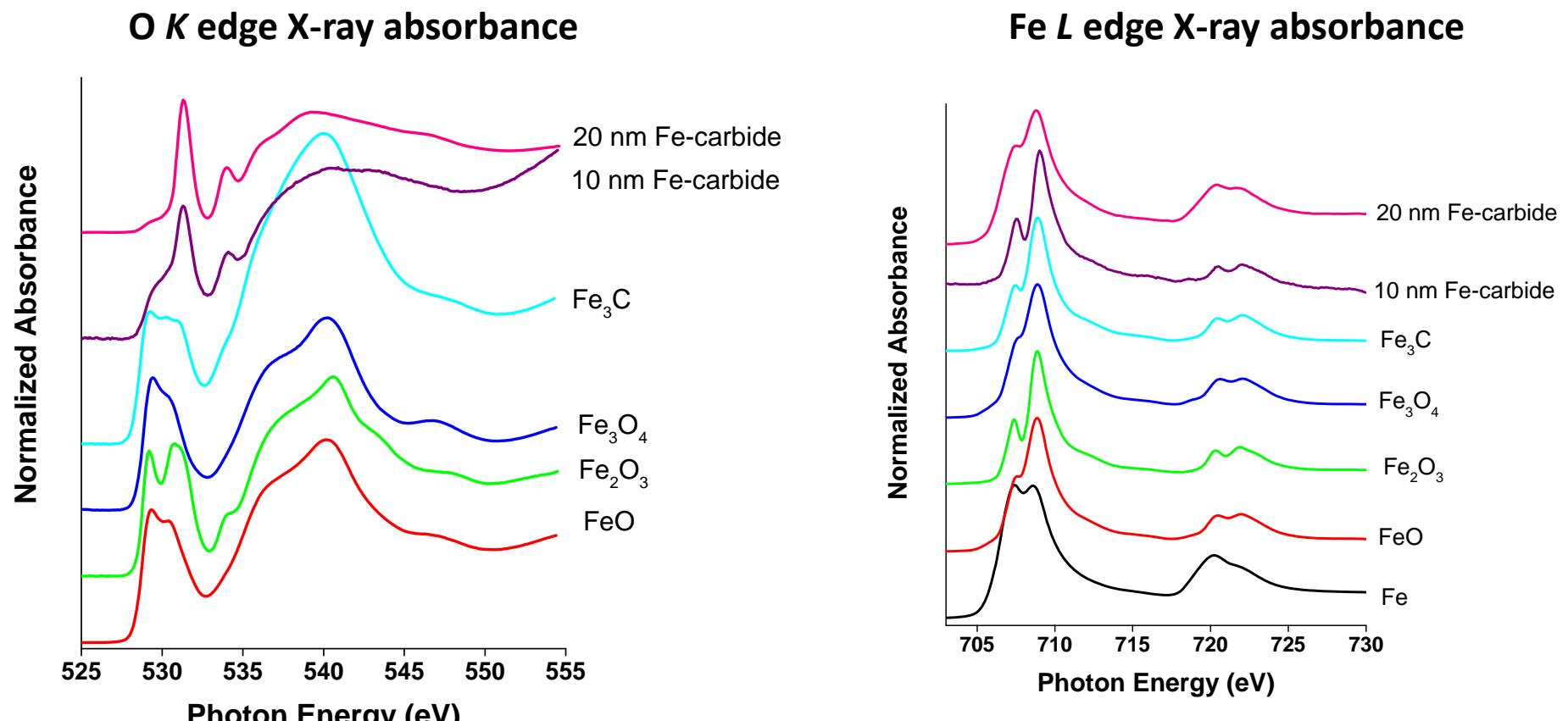


X-ray diffraction confirms  
nano- $\text{Fe}_5\text{C}_2$  structure



- $\text{Fe}_5\text{C}_2$  Häggs phase one of most active phases for FT
- ORD synthesis produces high yield, gram, batches of nearly pure  $\text{Fe}_5\text{C}_2$
- Nanoparticle shell is a mixed amorphous Fe-carbide/oxide
- Future work will incorporate into layered graphene and/or carbon nanotube supports

# Synchrotron X-ray characterization of nano- $\text{Fe}_5\text{C}_2$



# Summary



- ASF arises from a radical chain polymerization process at thermal equilibrium
- Deviations from ASF require disrupting molecular processes on catalyst/support surface (adsorption, diffusion, etc)
- Microwave reactors offer additional opportunities to deviate from ASF
- Nano-structured Graphene and  $\text{Fe}_5\text{C}_2$  have been synthesized and initial characterization started.