

Session: Post-Combustion Solvent-Based Capture

Development of Mixed-Salt Technology for CO₂ Capture from Coal Power Plants

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SRI International

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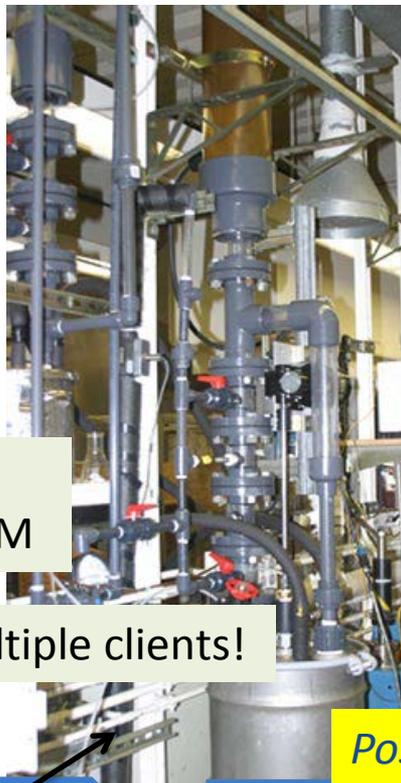
Technology Background

Our Early Experience in Solvent-Based Technology Development from Proof-of-Concept to Pilot-Scale



Energy and Environment Center

Ammonia technology development started in 2004



EPRI, NEXANT
STATOIL & ALSTOM

SRI work with multiple clients!



Post-Combustion CO₂ Capture

Small Bench-Scale

Large Bench-Scale

Pilot-Scale (0.25 MW_{th})
(For ALSTOM)

Key benefits:

- Reduced ammonia emissions
- Enhanced efficiency
- Reduced reboiler duty
- Reduced CO₂ compression energy

A SIGNIFICANT PARASITIC POWER REDUCTION COMPARED TO MEA !

How it works:

Selected composition of potassium carbonate and ammonium salts

- Overall heat of reaction 35 to 60 kJ/mol (tunable)

Absorber operation at 20 - 40° C at 1 atm with 30-40 wt.% mixture of salts

Regenerator operation at 70 - 180° C at 10-20 atm

- Produce high-pressure CO₂ stream

High CO₂ cycling capacity

No Solids

CO₂ Lean

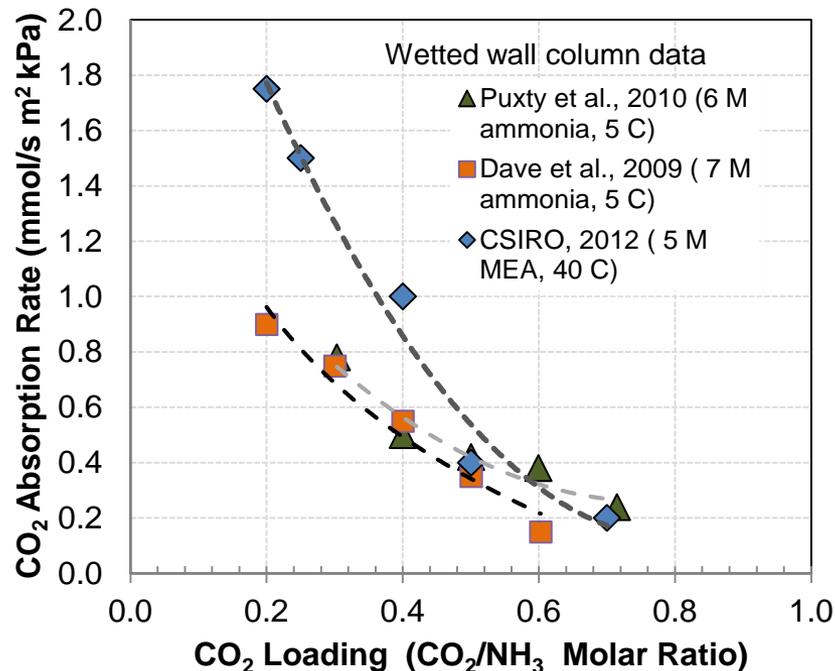
CO₂ Rich

$K_2CO_3-NH_3-xCO_2-H_2O$ system \leftrightarrow $K_2CO_3-NH_3-yCO_2-H_2O$ system

Published Data Showing Favorable Kinetics for CO₂ Absorption in Ammonia Solutions



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Comparison of CO₂ absorption rates for MEA and ammonia

Sources:

- Dave et al., (2009). Energy Procedia 1(1): 949-954
- Puxty et al., (2010). Chemical Engineering Science 65: 915-922
- CSIRO Report (2012). EP116217

Absorber side: Enhanced kinetics

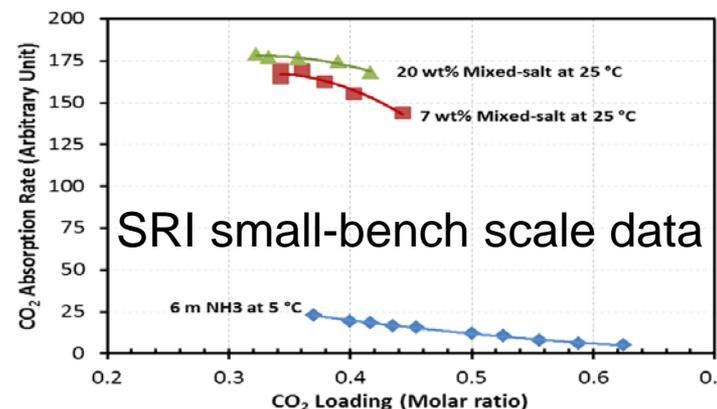
Pseudo first-order rate constants for CO₂ absorption in NH₃, MEA, and MDEA

| Solvent | $k_{app}/10^3 \text{ s}^{-1}$ |
|-------------------------|-------------------------------|
| NH ₃ at 5°C | 0.3 |
| NH ₃ at 10°C | 0.7 |
| NH ₃ at 20°C | 1.4 |
| NH ₃ at 25°C | 2.1 |
| MEA at 25°C | 6 |
| MDEA at 25°C | 0.58 |

Concentration = 1.0 kmol m⁻³

Source:

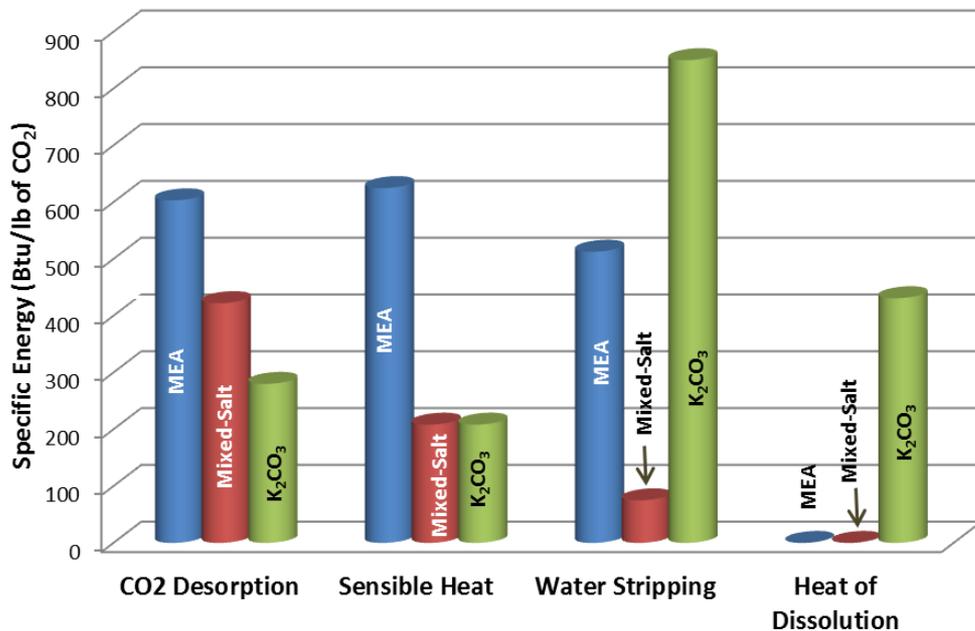
Derks and Versteeg (2009). Energy Procedia 1: 1139-1146



Mixed-Salt has a Low Energy Requirement for CO₂ Stripping

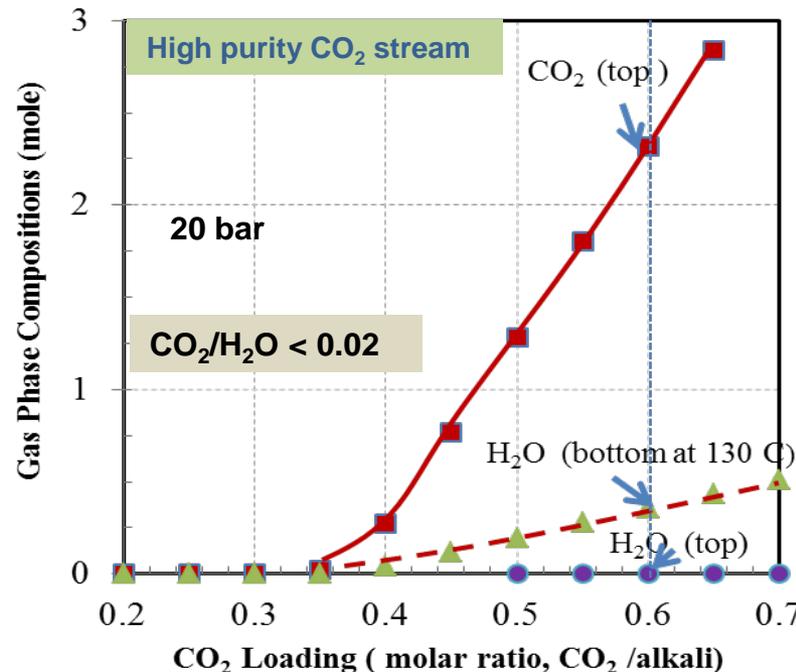


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Estimated regenerator heat requirement for mixed-salt system with 0.2 to 0.6 cyclic CO₂ loading. Comparison with neat K₂CO₃ and MEA is shown.

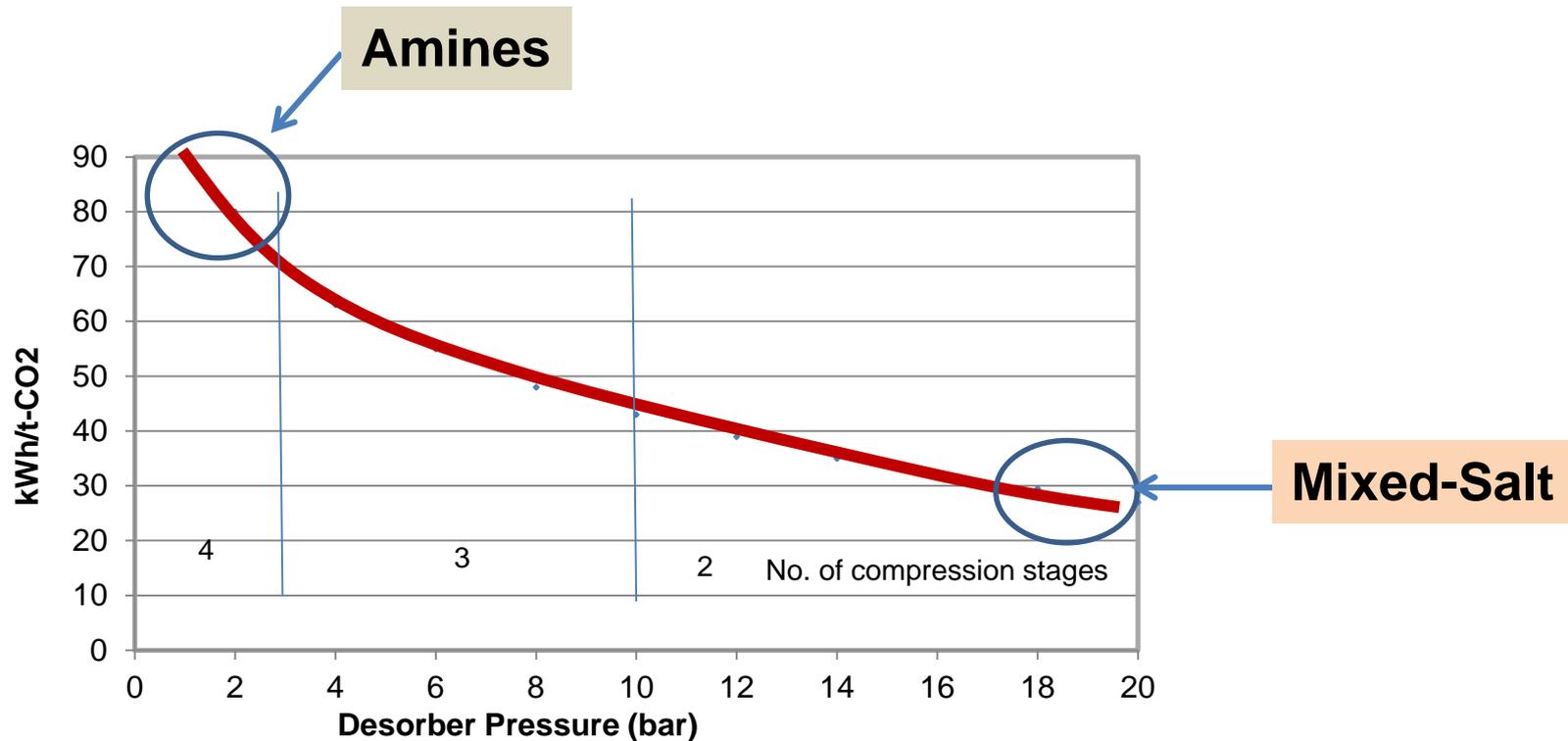
Sources: MEA data: CSIRO report (2012), EP116217
 K₂CO₃ data: GHGT-11; Schoon and Van Straelen (2011), TCCS-6
 Mixed-salt data; SRI modeling



Mixed-salt process requires minimal energy for water stripping

Regenerator side: Reduced water evaporation

Mixed-Salt Requires Less Energy for CO₂ Compression

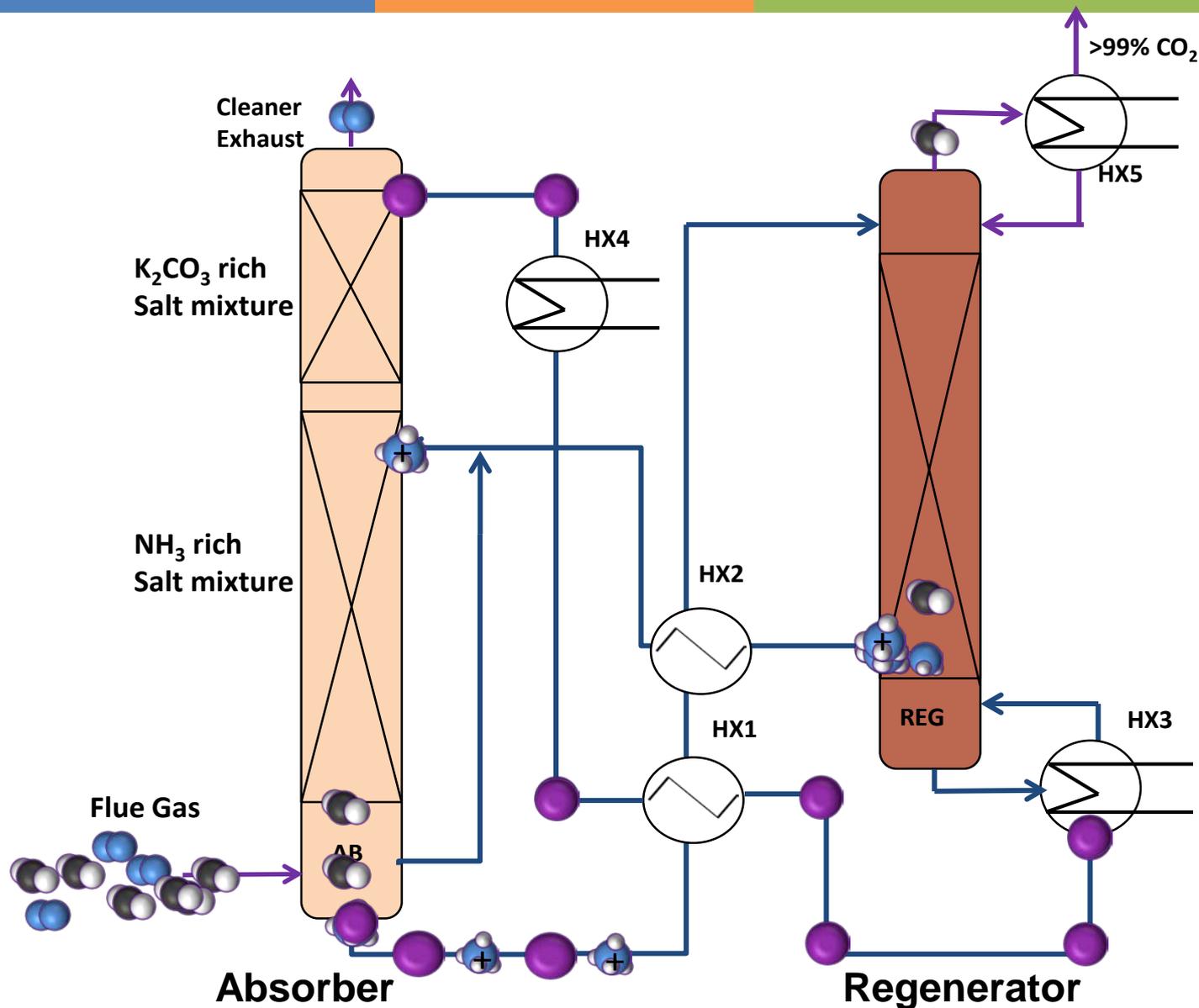


Electricity output penalty of compression to 100 bar as a function of desorber pressure

Source: Luquiaud and Gibbins., Chem Eng Res Des (2011)

CO₂ Compression: High-pressure CO₂ release

Mixed-Salt Process Flow Diagram





DOE Project Overview (Large-Bench Scale Testing)

Budget Period 1:

- Demonstrate the absorber and regenerator processes individually with high efficiency, low NH_3 emissions, and reduced water use compared to state-of-the-art ammonia-based technologies

Budget Period 2:

- Demonstrate the high-pressure regeneration and integration of the absorber and the regenerator
- Demonstrate the complete CO_2 capture system, optimize system operation, and collect data to perform the detailed techno-economic analysis of CO_2 -capture process integration to a full-scale power plant
- Conduct EH&S analysis of the process

The overall project objective is to demonstrate that mixed-salt technology can capture CO_2 at 90% efficiency and regenerate (95% CO_2 purity) at a cost of $\leq \$40/\text{tonne}$ to meet the DOE program goals.

Project Budget

| | Budget Period 1 | Budget Period 2 | Total |
|---------------------------|------------------------|------------------------|-----------------|
| | 10/1/13 - 12/30/14 | 1/1/15 - 3/31/16 | 10/1/13-3/31/16 |
| Total Project Cost | \$1,019,650 | \$1,278,975 | \$2,298,626 |
| DOE Share | \$819,534 | \$1,018,474 | \$1,838,009 |
| Cost Share | \$200,116 | \$260,501 | \$460,617 |

Cost share by SRI, OLI Systems, POLIMI, Aqueous Solutions Aps, Stanford University, and IHI Corporation

Project Manager: Mr. Steven Mascaro, NETL

Prime Contractor: SRI International

US Partners: OLI Systems, Stanford University, Dr. Eli Gal

International Partners: Dr. Kaj Thomsen, POLIMI, IHI Corporation

Project Tasks



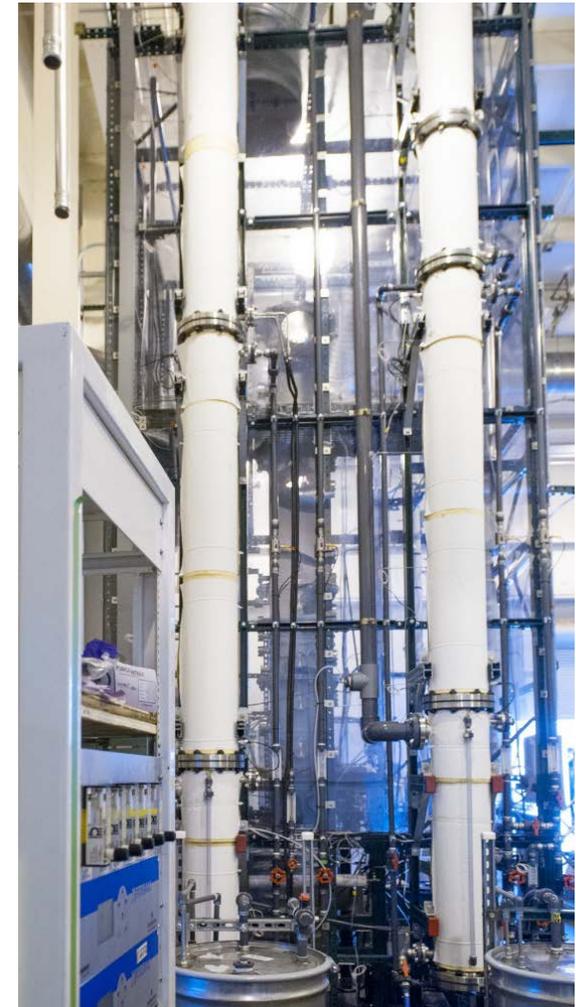
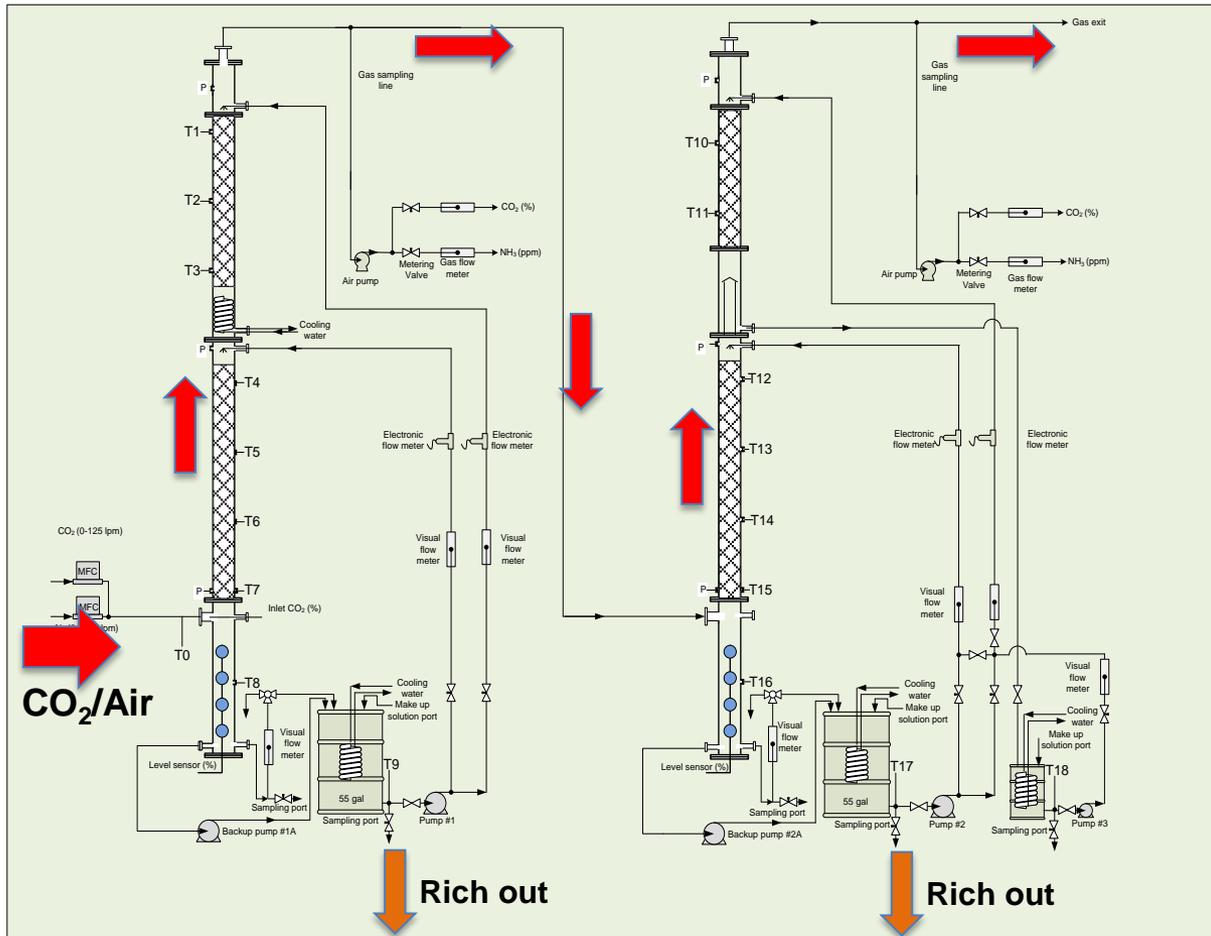
| Task | Start Date | End Date | 2014 | | | | | 2015 | | | | 2016 | | | |
|--|------------|------------|--|----|----|----|----|------|----|----|----|------|----|----|--|
| | | | Q4 | Q1 | Q2 | Q3 | Q4 | Q1 | Q2 | Q3 | Q4 | Q1 | Q2 | Q3 | |
| Mixed-Salt BP1 and BP2 | 10/1/2013 | 3/31/2016 | [Gantt bar spanning from Q4 2013 to Q3 2016] | | | | | | | | | | | | |
| Task 1.0 - Project Management and Planning | 10/1/2013 | 3/31/2016 | [Gantt bar spanning from Q4 2013 to Q3 2016] | | | | | | | | | | | | |
| Task 2.0: Individual Absorber and Regenerator Testing in Semi-Continuous mode | 10/1/2013 | 11/30/2014 | [Gantt bar spanning from Q4 2013 to Q3 2014] | | | | | | | | | | | | |
| Subtask 2.1 - Test Systems Design and Installation | 10/1/2013 | 4/28/2014 | [Gantt bar spanning from Q4 2013 to Q1 2014] | | | | | | | | | | | | |
| Subtask 2.2 - Test Plans | 2/1/2013 | 2/30/2014 | [Gantt bar spanning from Q1 2014 to Q1 2014] | | | | | | | | | | | | |
| Subtask 2.3 - Absorber Tests | 4/30/2014 | 11/30/2014 | [Gantt bar spanning from Q2 2014 to Q3 2014] | | | | | | | | | | | | |
| Subtask 2.4 - Regenerator Tests | 7/1/2014 | 11/3/2014 | [Gantt bar spanning from Q3 2014 to Q3 2014] | | | | | | | | | | | | |
| Subtask 2.5 - Bench-Scale Test Data Analysis | 2/28/2014 | 11/30/2014 | [Gantt bar spanning from Q1 2014 to Q3 2014] | | | | | | | | | | | | |
| Task 3.0 - Preliminary Process Modeling and Techno-Economic Analysis | 3/1/2014 | 12/15/2014 | [Gantt bar spanning from Q1 2014 to Q4 2014] | | | | | | | | | | | | |
| Subtask 3.1 - Process Modeling | 3/1/2014 | 11/30/2014 | [Gantt bar spanning from Q1 2014 to Q3 2014] | | | | | | | | | | | | |
| Subtask 3.2 - Preliminary Economic Analysis | 8/1/2014 | 12/15/2014 | [Gantt bar spanning from Q3 2014 to Q3 2014] | | | | | | | | | | | | |
| Task 4.0 - Budget Period 2 Continuation Application | 9/15/2014 | 9/30/2014 | [Gantt bar spanning from Q4 2014 to Q4 2014] | | | | | | | | | | | | |
| Continuation Report Submission | 9/15/2014 | 9/30/2014 | [Gantt bar spanning from Q4 2014 to Q4 2014] | | | | | | | | | | | | |
| Task 5.0 - Bench-Scale Integrated System Testing | 1/15/2015 | 3/31/2016 | [Gantt bar spanning from Q1 2015 to Q3 2016] | | | | | | | | | | | | |
| Subtask 5.1 - Design of the Bench-Scale Integrated Test System | 1/15/2015 | 3/31/2015 | [Gantt bar spanning from Q1 2015 to Q1 2015] | | | | | | | | | | | | |
| Subtask 5.2 - Installation of the Bench-Scale Continuous, Integrated Test System | 1/15/2015 | 3/31/2015 | [Gantt bar spanning from Q1 2015 to Q2 2015] | | | | | | | | | | | | |
| Subtask 5.3 - Bench-Scale Test Plans | 1/15/2015 | 2/15/2015 | [Gantt bar spanning from Q1 2015 to Q1 2015] | | | | | | | | | | | | |
| Subtask 5.4 - Bench-Scale Tests and Data Analysis | 4/1/2015 | 3/31/2016 | [Gantt bar spanning from Q2 2015 to Q3 2016] | | | | | | | | | | | | |
| Task 6.0 - Process Modeling and Techno-Economic Analysis | 5/1/2015 | 3/31/2016 | [Gantt bar spanning from Q2 2015 to Q3 2016] | | | | | | | | | | | | |
| Subtask 6.1 - Process Modeling | 5/1/2015 | 3/1/2016 | [Gantt bar spanning from Q2 2015 to Q3 2016] | | | | | | | | | | | | |
| Subtask 6.2-Techno-Economic Analysis | 8/1/2015 | 3/30/2016 | [Gantt bar spanning from Q3 2015 to Q3 2016] | | | | | | | | | | | | |
| Subtask 6.3- Technology EH&S Risk Assessment | 9/1/2015 | 3/30/2016 | [Gantt bar spanning from Q3 2015 to Q3 2016] | | | | | | | | | | | | |
| Final Report Submission | 4/30/2016 | 5/30/2016 | [Gantt bar spanning from Q2 2016 to Q2 2016] | | | | | | | | | | | | |

Project is on time and on schedule



Work Performed in BP1

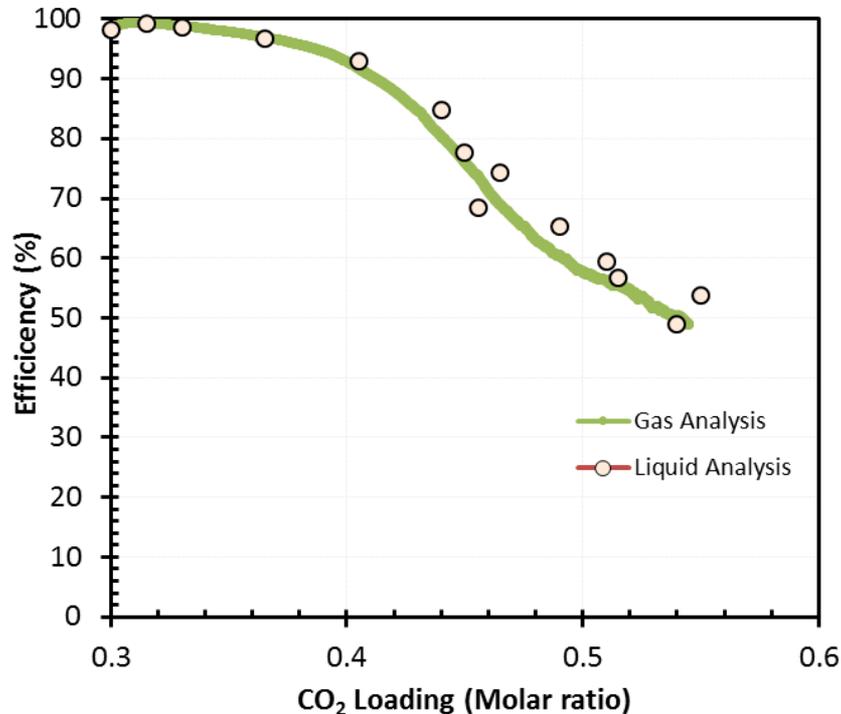
Schematic and Photograph of the Absorber System



0.25 to 1 t-CO₂/day capacity

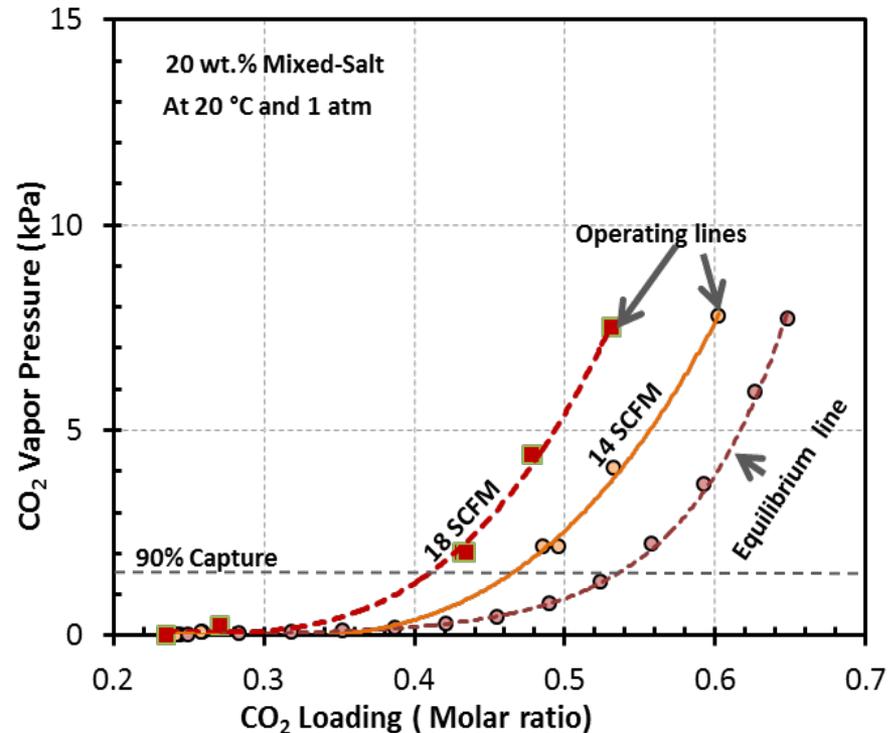
Test Data

Absorber 1 Efficiency



Better than 90% efficiency with incoming lean absorption solution and < 0.4 CO₂ loading

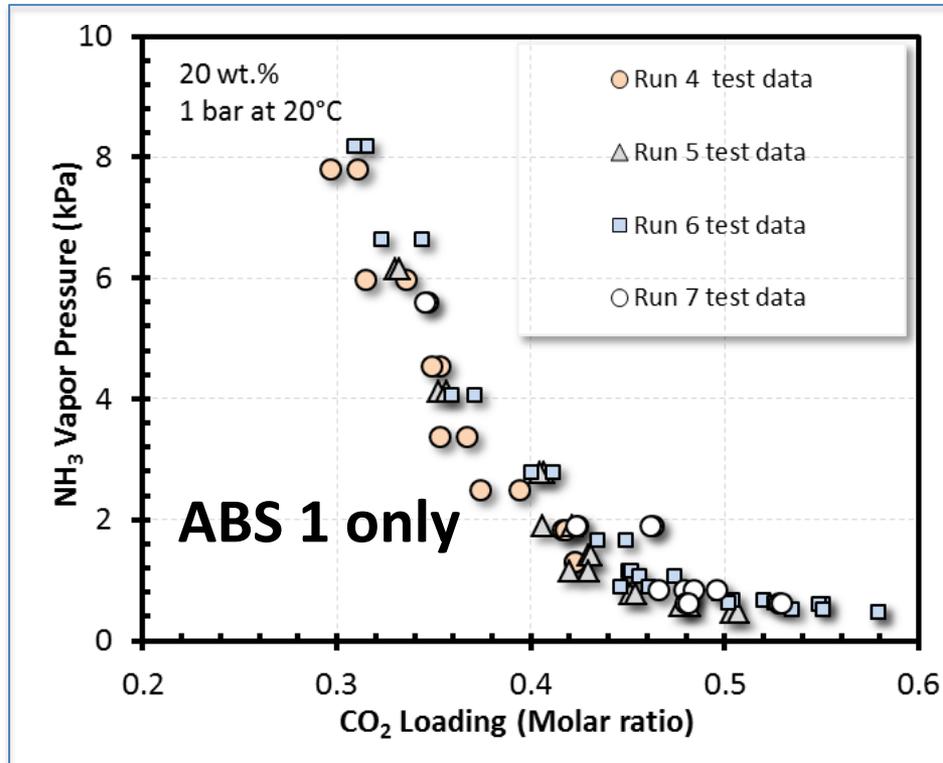
Modeling and Test Data



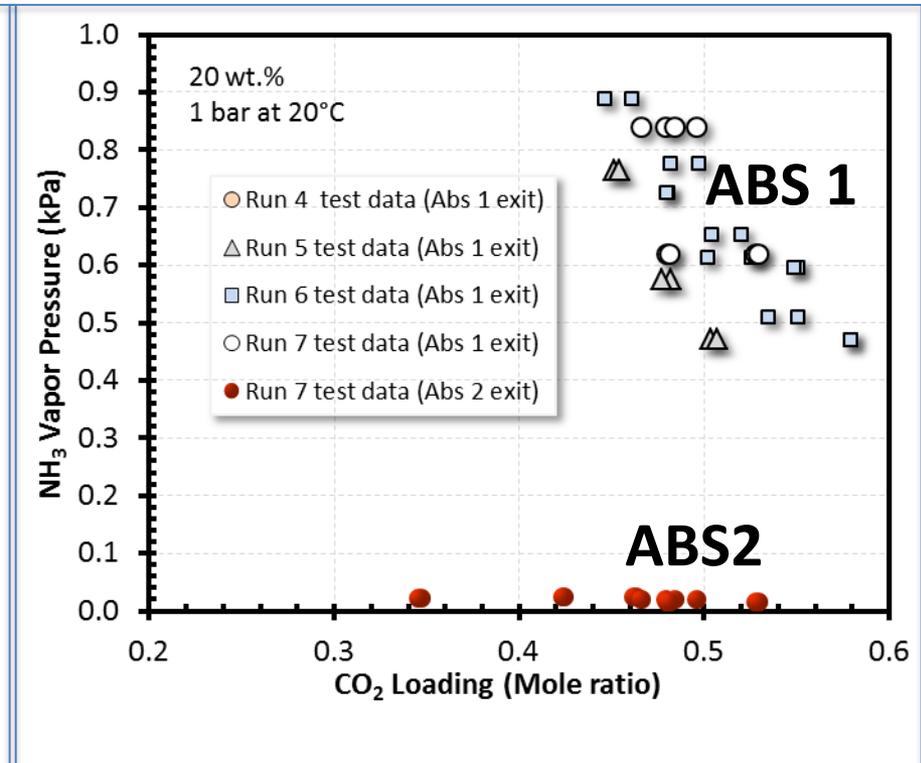
CO₂ vapor pressure at the absorber exit under various CO₂-loading conditions

The observed overall rates for CO₂ absorption are on the same order as those of MEA-based systems and about 5-7x higher than chilled ammonia systems

Test Data

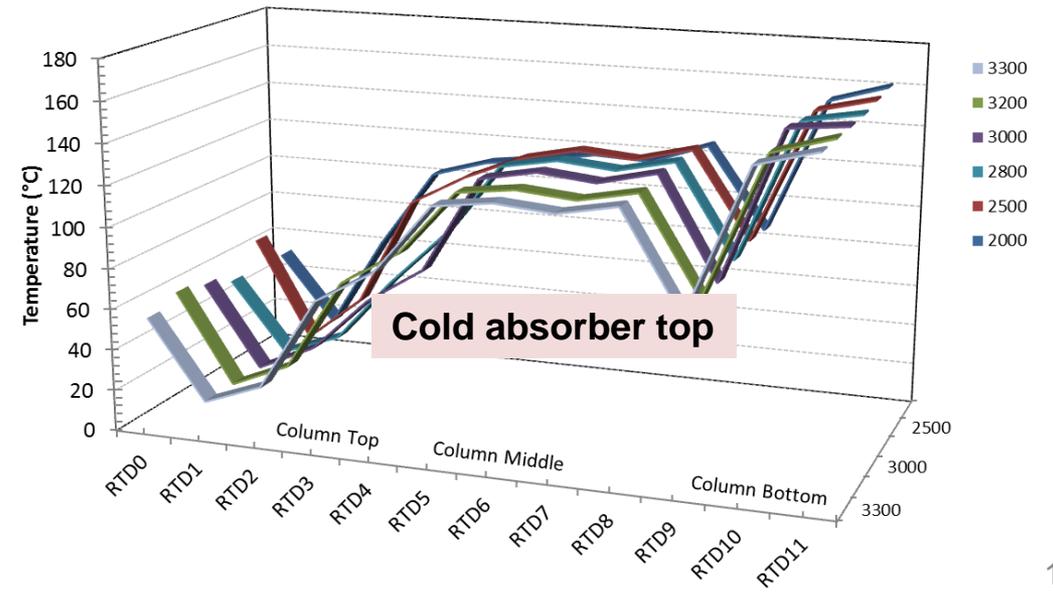
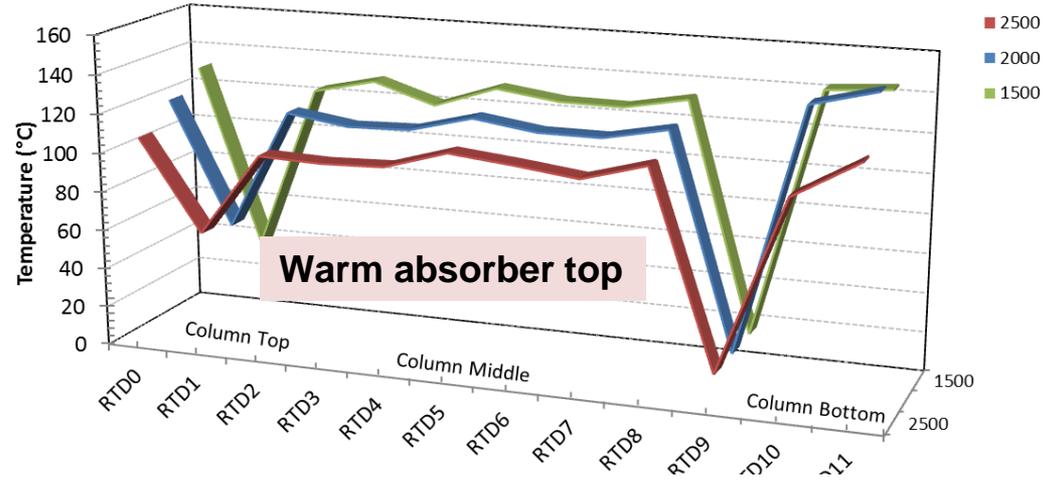
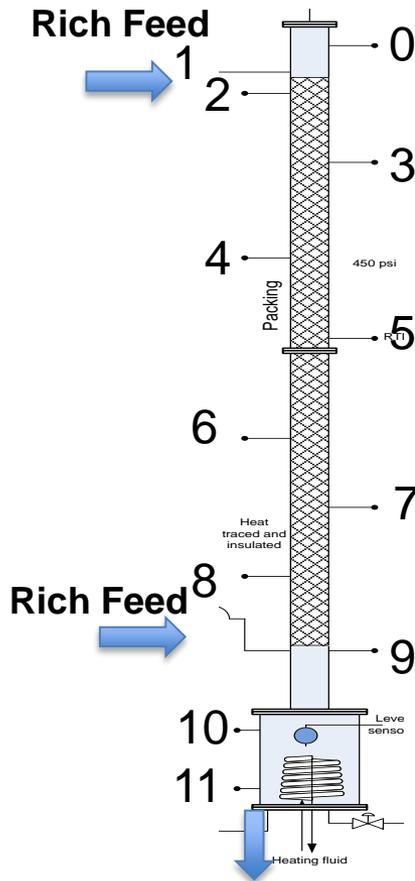


NH₃ vapor pressure at the Absorber 1 exit under various CO₂-loading conditions

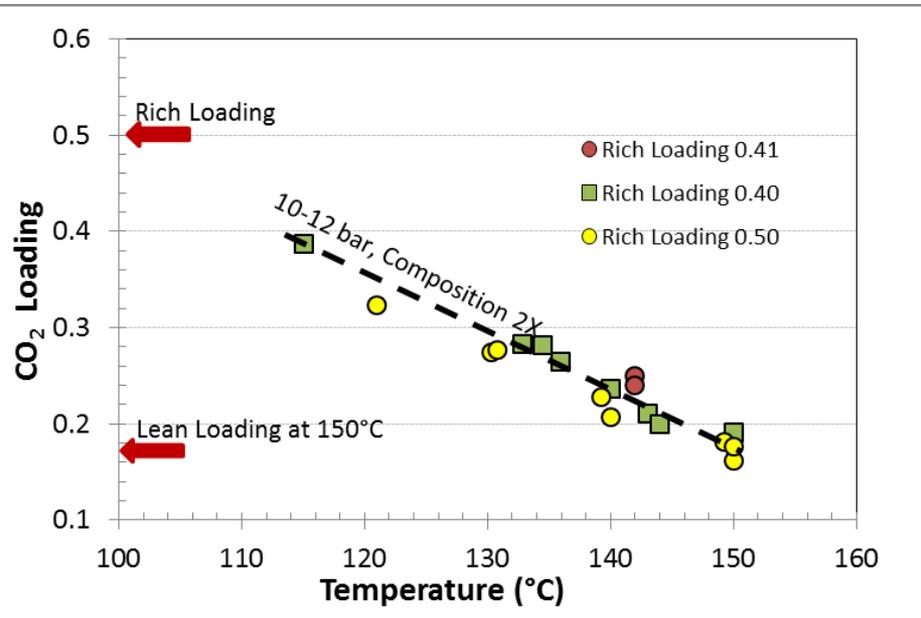


NH₃ vapor pressure at the Absorber 1 and 2 exits under various CO₂-loading conditions

Regeneration Operation Modes

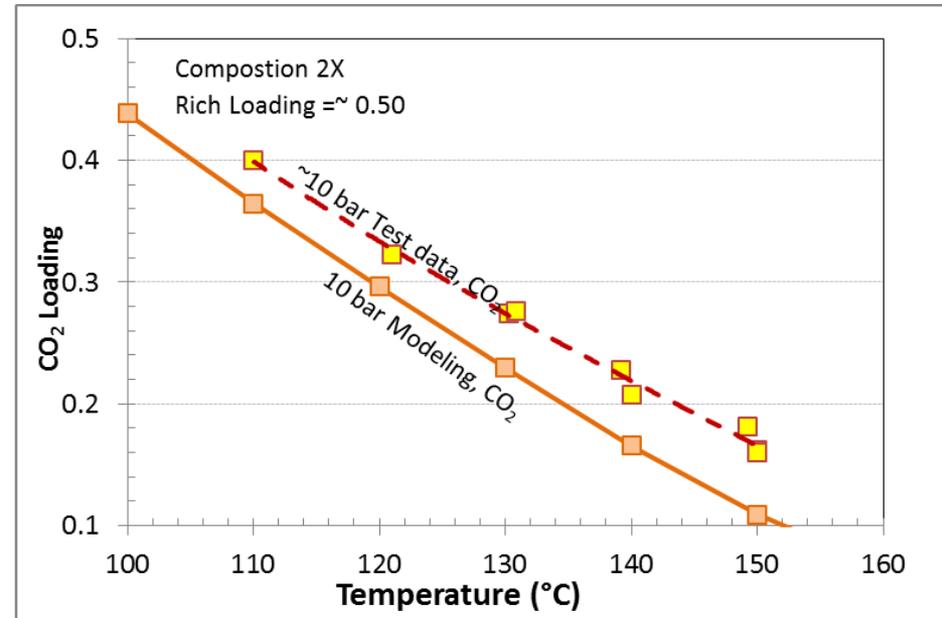


Test Data



Variation of attainable CO₂-lean loading level with temperature for rich loadings of 0.40 to 0.50 at 10-12 bar

Modeling and Test Data



Comparison of measured and modeled attainable CO₂-lean loading at 100 to 150 °C

Process was demonstrated with cyclic loading from 0.2 to (lean) to 0.5 (rich) at 150 °C

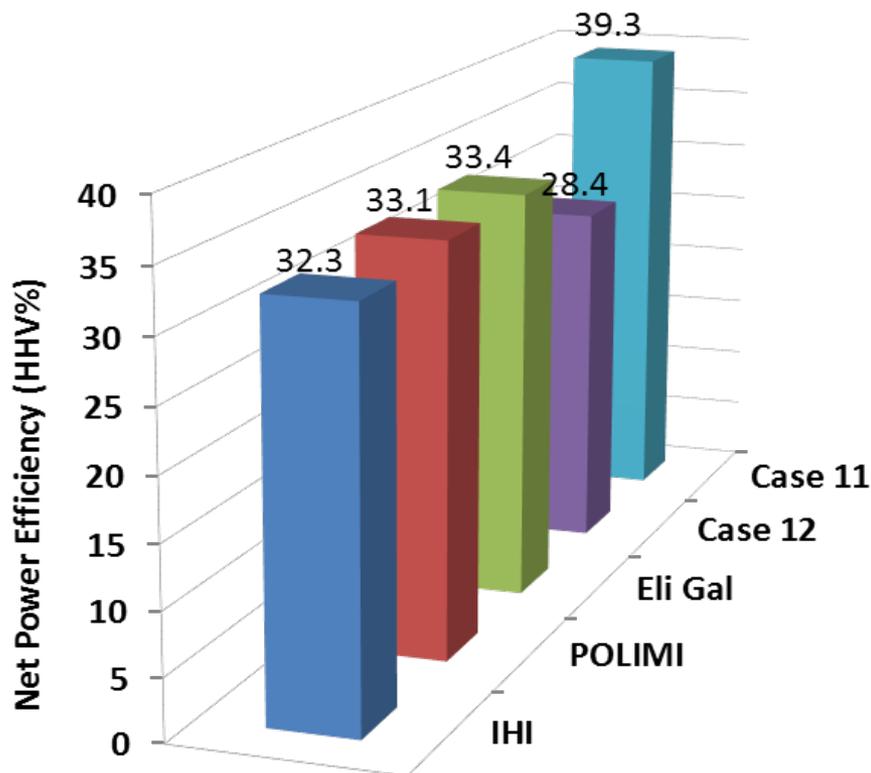
The produced lean loading well exceeds that required for > 90% CO₂ capture from flue gas streams

Modeling Results: Preliminary Net Power Efficiency

Thermodynamic
Model

ASPEN
Modeling

Mass & Energy
Balance



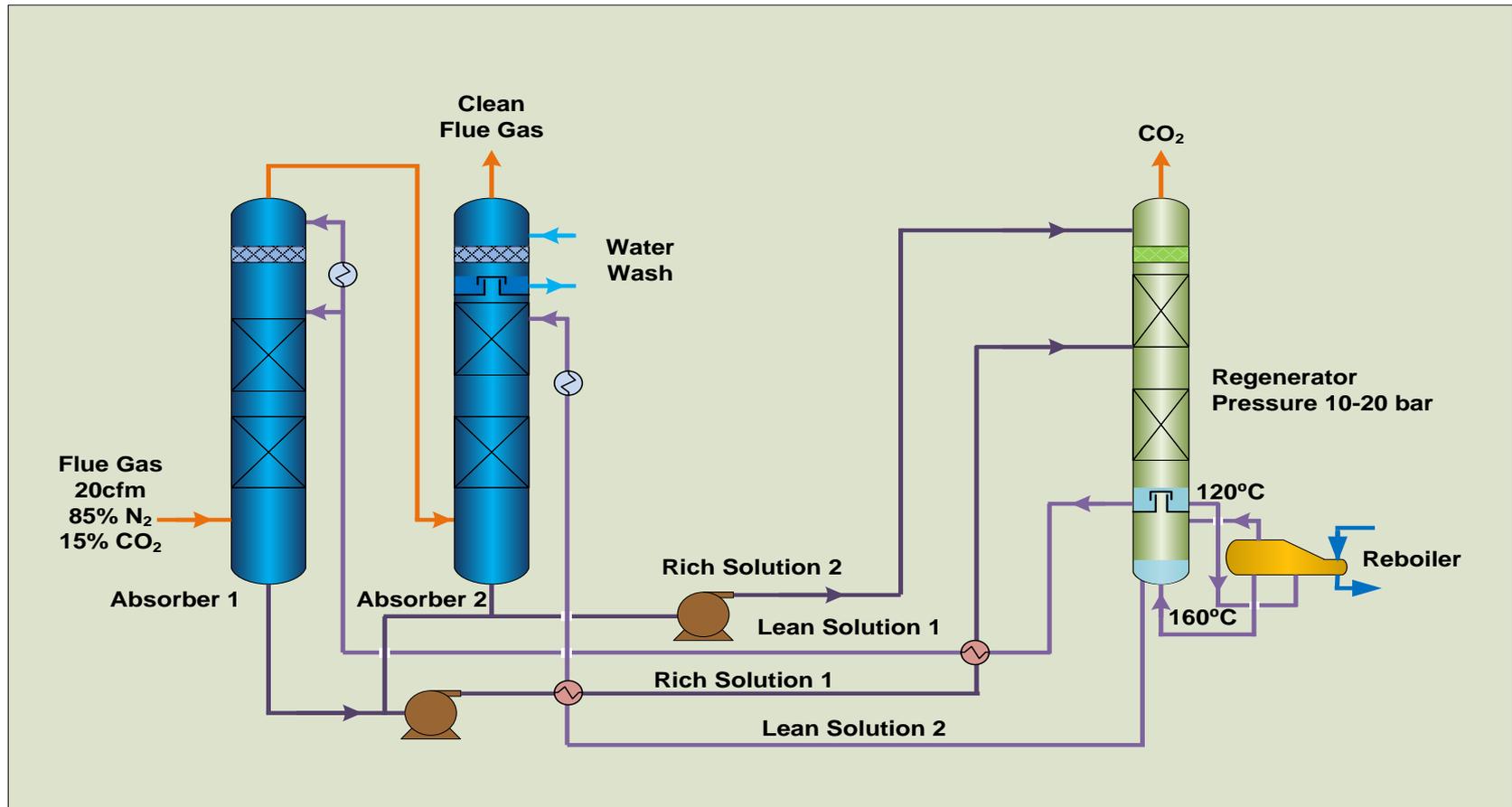
- Software package was developed for thermodynamic modeling of mixed-salt system
- Process layout with two regenerator options were modeled, regenerator energy requirement was in the range 1.8 to 2.2 MJ/kg-CO₂, lowest energy option was chosen for BP2 regenerator design



Progress in BP2

- Absorber modification (completed and continuous operation in progress)
- Novel regenerator design and fabrication (80% complete)
- Integrated system operation (to begin in August)
- Complete system modeling in ASPEN (complete)
- Complete system modeling with rate-based approach (continuing)
- Process technoeconomic analysis (TEA) (to start in August 2015)
- Process EH&S analysis (to start in September 2015)

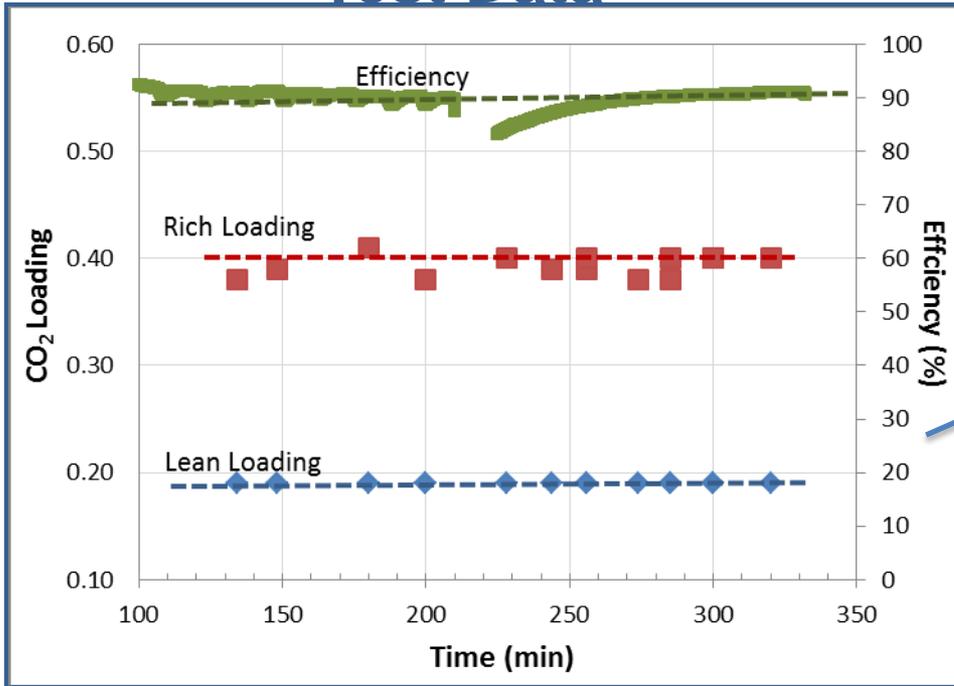
Simplified PFD of the Integrated System



1. System is currently in operation with buffer tanks for lean and rich solution storage
2. Continuous operation of the absorber system was smooth and the observed results were as expected based on the BP1 work

System Testing in Continuous Mode

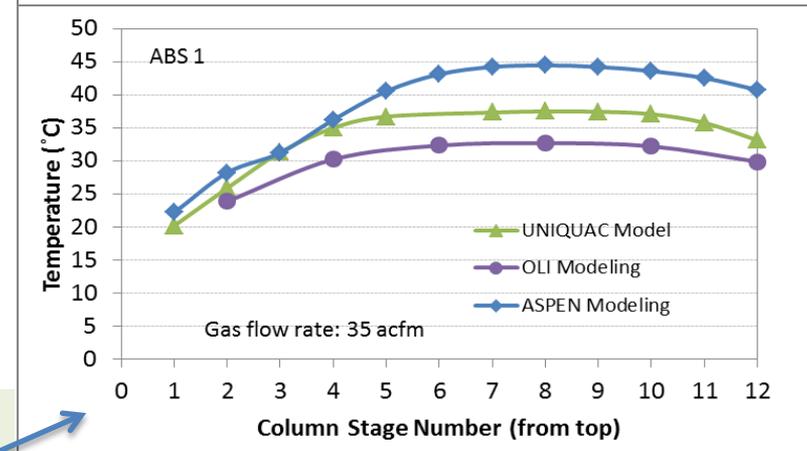
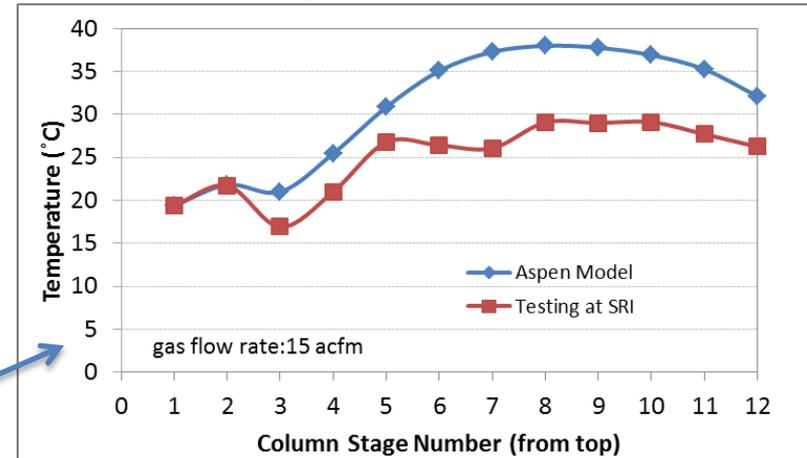
Test Data



90% CO₂ capture efficiency with 0.19 to 0.40 cyclic CO₂ loading in Absorber 1
Gas flow rate = 15 acfm

Process Modeling: SRI (ASPEN) and OLI (ESP)
 Cyclic Loading = 0.18 to 0.58
 Reboiler Duty ~ 1.8 to 1.9 MJ/kg-CO₂
 Ammonia Emission < 10 ppm

Modeling and Test Data



Comparison of observed and modeled temperature profiles for Absorber 1

Process Summary

- **Uses inexpensive, industrially available material (potassium and ammonium salts)**
- **Requires no feedstream polishing**
- **Does not generate hazardous waste**
- **Has the potential for easy permitting in many localities**
- **Uses known process engineering**

Demonstrated Benefits

- **Enhanced CO₂ capture efficiency**
- **High CO₂-loading capacity**
- **High-pressure release of CO₂**

Expected Benefits

- **Reduced energy consumption compared to MEA**
- **Reduced auxiliary electricity loads compared to the chilled ammonia process**
- **Possible flexible carbon capture operation**

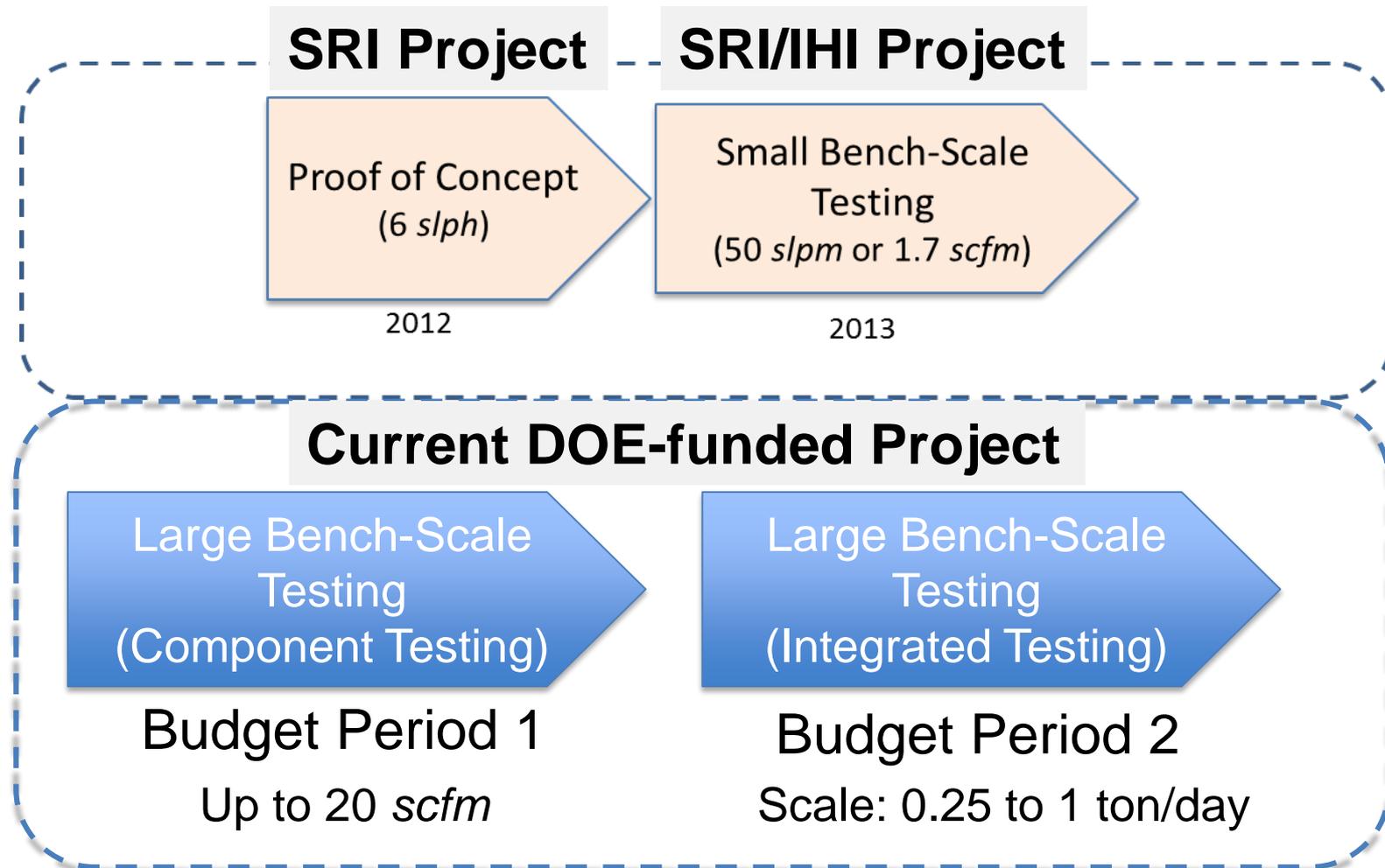


Plans for Future Testing and Commercialization

Bench-scale Technology Development Timeline



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Scale-up Plan of Mixed-Salt Process for CO₂ Capture from Coal Power Plants

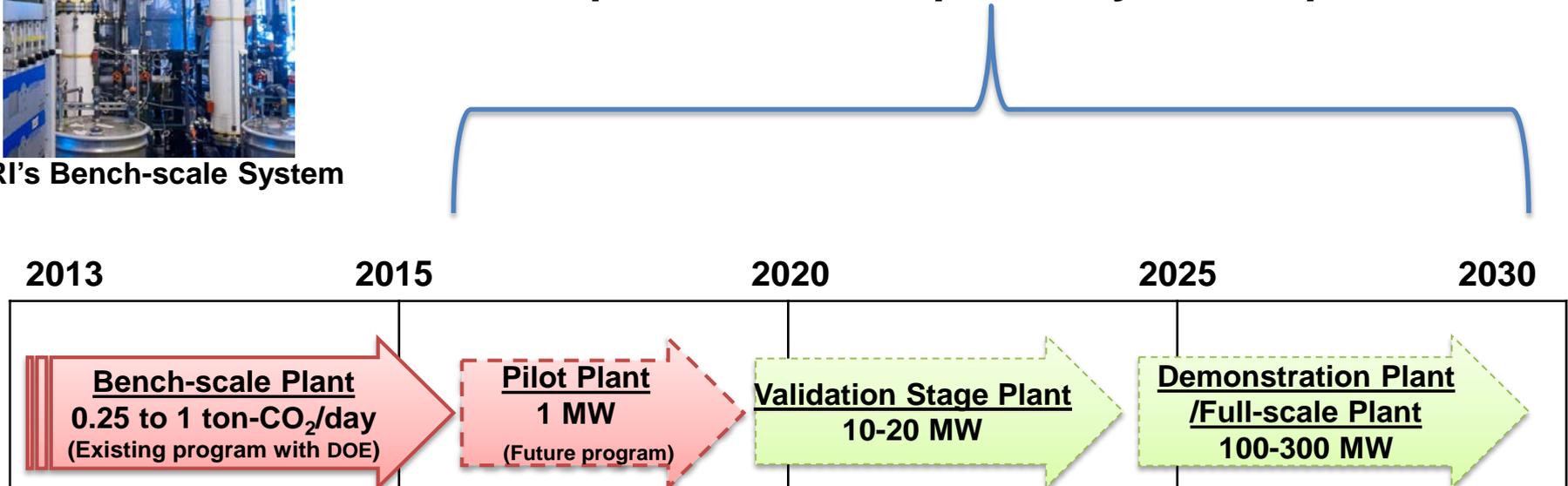


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SRI's Bench-scale System

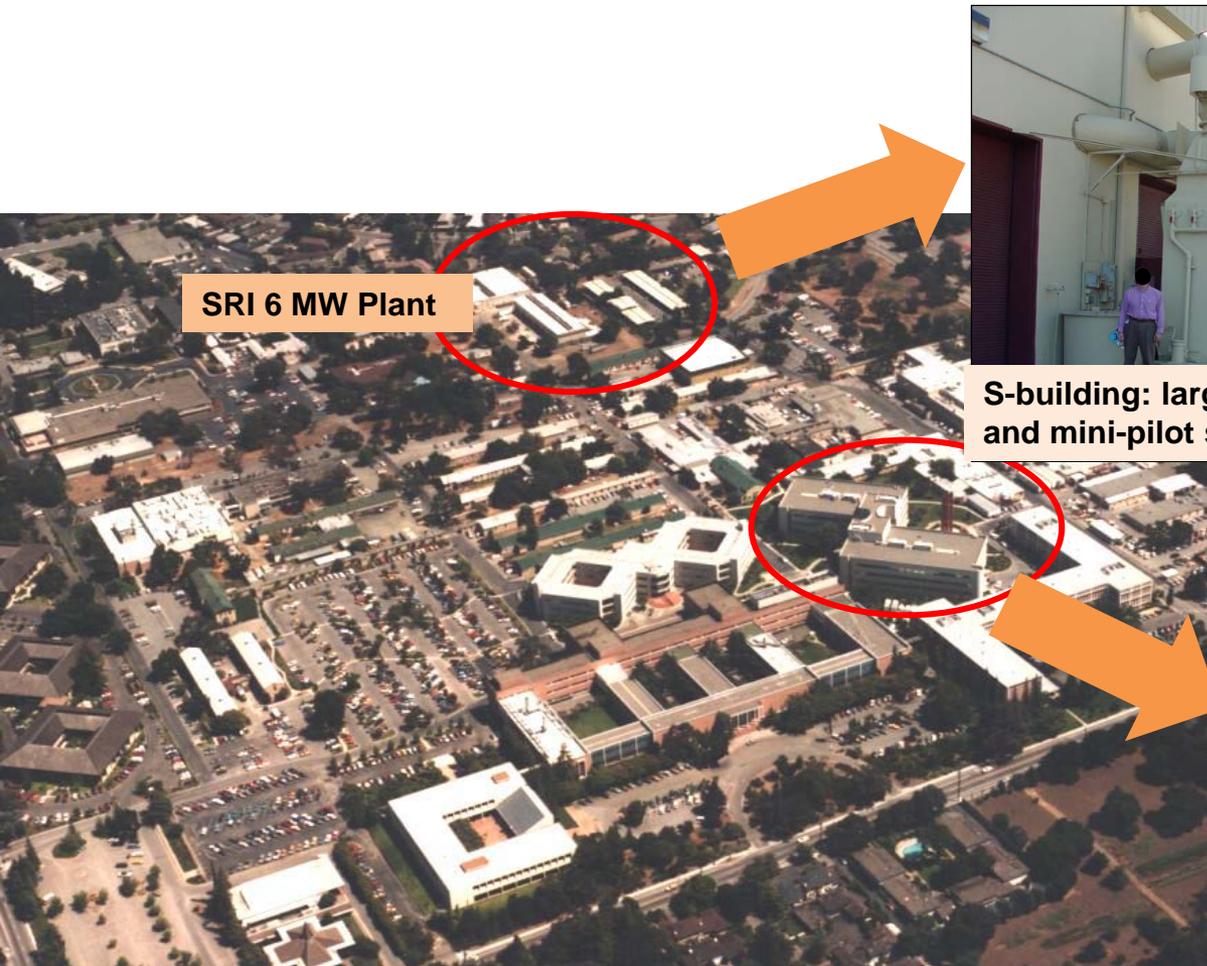
Development Plan Proposed by IHI Corporation



Current Project Location



Energy and Environment Center



SRI 6 MW Plant



S-building: large bench and mini-pilot studies



CO₂ yard for mini-pilot testing (up to 100 acfm)



P-building: lab-scale tests

SRI's site in Menlo Park, CA (~ 65 acres)
SRI also has a test site near Livermore, CA (480 acres)

NETL (DOE)

- Mr. Steve Mascaro, Ms. Lynn Bricket, and other NETL staff members

SRI Team

- Dr. Indira Jayaweera, Dr. Palitha Jayaweera, Dr. Jianer Bao, Ms. Regina Elmore, Dr. Srinivas Bhamidi, Mr. Bill Olsen, Dr. Marcy Berding, Dr. Chris Lantman, and Ms. Barbara Heydorn

Collaborators

- OLI Systems (Dr. Prodip Kondu and Dr. Andre Anderko), POLIMI (Dr. Gianluca Valenti and others), Stanford University (Dr. Adam Brant and Mr. Charles Kang), Dr. Eli Gal, and Dr. Kaj Thomsen

Industrial Partner

- IHI Corporation (Mr. Shiko Nakamura, Mr. Okuno Shinya, Dr. Kubota Nabuhiko, Mr. Yuichi Nishiyama, and others)

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