

Optical Thin Films for High Temperature Sensing in Advanced Fossil Energy Applications

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Project Team: Thomas Brown, Congjun Wang, John Baltrus, Michael Buric, Andrew Schultz, Gordon Holcomb, and Joseph Tylczak**OENERG**

Overview of Presentation

- **Motivation and Background**
- **Project Overview**

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- **Recent Research and Development Effort Updates**
- **Summary and Conclusions**

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Motivation and Background

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Advanced Fossil-Based Power Generation

Advanced Fossil-BasedPower Generation Involves High Temperature Gas Streams(Coal or Natural Gas)

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http://www.fossil.energy.gov/programs/powersystems/gasification/howgasificationworks.html

Fossil-Fuel Based Power Plants of the Future are Highly Complex Making Sensors and Controls of Crucial Importance.

Advanced Sensors and Controls Can Also Significantly Impact the Existing Fleet of Utility-Scale Fossil Plants.

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Advanced Fossil-Based Power Generation

Table of Relevant Harsh Environments in Advanced Fossil Energy Technologies

Short Term Focus

A Wide Range of Gas Species are Relevant.

Embedded Sensors Must Operate in Extreme Conditions (T, P, corrosive).

Similar Needs Exist in a Wide Range of Other Industries (Aerospace, Aviation, Manufacturing, etc.).

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Advanced Fossil-Based Power Generation

Example: Solid Oxide Fuel Cells Internal Gas and Temperature

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Distribution

Spatial Distribution in Temperature and Fuel Gas Composition

Incompatible with Traditional Sensing Technologies

1) At Limits of High Temperature Electrical Insulation2) Limited Access Space

- **Parameter Sensing Temperature : 700-800oC Anode Stream : Fuel Gas (e.g. H2-Containing) Cathode Stream : Air or O₂**
	- **3) Only Single-Point, Single-**

Stable Sensors Capable of Embedding in Harsh Environments Would Enable Unprecedented Access to New Process Information.

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Project Overview

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Thin Film Functional Sensor Layers in Harsh Environment Sensing Applications

Example: Pd-Thin Films for H₂-Sensing Applications

Fiber Optic Sensor for Hydrogen Concentrations near the Explosive Limit

Michael A. Butler*

Sandia National Laboratories, Albuquerque, New Mexico 87185

J. Electrochem. Soc., Vol. 138, No. 9, September 1991 (

Optical Fiber

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 PdH_Y Film

Fig. 1. Representation of the micromirror hydrogen sensor. The fiber is 125 μ m in diameter with a 50 μ m core. The palladium film is deposited by evaporation and is typically 10-20 nm thick.

> **1.5 Pages Long, > 84 Citations!**

Fig. 2. Response of the micromirror hydrogen sensor to varying concentrations of H₂ in N₂. Between exposures to hydrogen, the sensor is exposed to air. This palladium film is 10 nm thick and the measurements were made at room temperature.

Fig. 3. Magnitude of the response of the micromirror hydrogen sensor at room temperature to varying concentrations of hydrogen in air. Note that the hydrogen actually produces a decrease in reflectivity as shown in Fig. 2. The absolute reflectivity of this film is about 22%. The vertical line indicates the lower explosive limit.

A Sensor for Leak Detection of H2 in Potentially Explosive Conditions was Proposed by End-coating an Optical Fiber with a Pd Thin Film by Butler in 1991.

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Motivation for Looking at Optical Materials

Chemi-resistive materials : the fundamentals are understood

- -G. Korotcenkov (2007). Materials Science and Engineering: B **139(1): 1-23**
- - Gas Species Interact with Adsorbed Surface Species or Alter Defect Chemistry Changing:

(1) Free Charge Carrier Concentration, (2) Mobility of Free Carriers

Optical materials : fundamentals are poorly understood

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- - How Do Refractive Index and Optical Absorption Depend Upon Defect Chemistry or Concentration of Adsorbed Species?
- - How Can Materials with Useful Responses Be Optimally Integrated into Optical Sensing Devices? **e.g. Evanescent Wave Sensors**

Functional Sensor Material Project Timeline FY 2012 FY 2013FY 2014

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Advanced Functional Sensor Material Project Team Regional University Partnerships

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Collaborative Interactions with the Project Team

2 – Joint Peer-reviewed Publications (U. Pitt, U. Albany)

1 – Joint Patent Application (U. Pitt.)

1 - Additional Joint Publication Currently in Preparation (U. Conn.)

The Team Also Seeks to Establish New Collaborations with Other

NETL-Funded Projects and Others to Help Promote the Mission of the

Laboratory and the Crosscutting Research Program.

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Unique Capabilities / Facilities of the Project Team

In-Situ Property / Device Measurements In-Situ Characterization Tools

- High Temp. X-ray Diffraction
	- Temperatures up to 1300 $^{\circ}$ C
	- $\,$ Gases Including H $_2$, O $_2$, N $_2$, CO $\,$
- High Temperature Scanning Electron **Microscopy**
	- $~\sim$ 0.5 torr of $\rm~H_2O(g)$ or Air
	- Temperatures up to 900 $\rm ^{\circ}$ C
- X-ray Photoelectron Spectroscopy with a Reaction Chamber
	- Temperatures up to 1300°C

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Gases Including H_2 , O_2 , N_2 , CO

- - Ambient Pressure Reactors:
	- $\,$ H₂, O₂, N₂, CO, CO₂, CH₄
	- Temperatures up to 900 $^{\circ}$ C
	- Full System Automation
- Elevated Pressure Reactor:
	- $\,$ H₂, O₂, N₂, CO, CO₂, CH₄
	- Temperatures up to 900 $\rm ^oC$
	- Pressures up to 900psi
- Film Resistivity
- -Film Transmittance (~300-2500nm)
- -Film Reflectance (~300-2500nm)

The Team Also Has Access to Microscopy, Nanofabrication, and Deposition Capabilities through Carnegie Mellon University.

A Well Equipped Set of Unique Facilities Can Be Leveraged in Collaborations with Extramurally Funded Research Efforts or Others.

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In-Situ Characterization Facilities

Unique Capabilities for Fully Automated Characterization of Electrical and Optical Properties of Thin Films and Sensors at High Temperatures in Various Gas Mixtures Have Been Established

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Peer-Reviewed Publications Published / Accepted Publications, Patents, and Partnership Opportunity

- –P. R. Ohodnicki, Future of Instrumentation International Workshop (FIIW), 2012, 1-4
- –M. P. Buric, P. R. Ohodnicki, J. Duy, SPIE Nanoscience & Engineering, 84630D-84630D-14, 2012
- – P. R. Ohodnicki, T. D. Brown, M. P. Buric, J. P. Baltrus, B. Chorpening, SPIE Nanoscience & Engineering, 845608- 845608-12, 2012
- –PR Ohodnicki, S Natesakhawat, JP Baltrus, B Howard, TD Brown, Thin Solid Films, 520 (19), 6243-6249 (2012)
- – PR Ohodnicki, C Wang, S Natesakhawat, JP Baltrus, TD Brown, Journal of Applied Physics 111 (6), 064320-064320-11, (2012)
- P. R. Ohodnicki et al., Nanoscale, Vol. 5 (19), 9030-9039, 2013.

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- –P. R. Ohodnicki et al, Thin Solid Films, Vol. 539 (31) 327-336 (2013).
- – M. Buric, P. Ohodnicki, and B. Chorpening, Proc. SPIE 8816, Nanoengineering: Fabrication, Properties, Optics, and Devices X, 88160N, 2013 (doi: 10.1117/12.2024167)
- –P. R. Ohodnicki, T. Brown, Nanomaterials and Energy, Vol. 3 (2) 40-46 (2014).
- –P. R. Ohodnicki et al., Sensors and Actuators B, Accepted and in Press, 2014.

Three Patents Awarded and Several Applications Submitted

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Partnership Opportunity Notice:

http://www.netl.doe.gov/business/crada/pdfs/13-443223.pdf

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Recent R&D Updates : Combined Gas and Temperature Optical Sensing Using Au-Nanoparticle Based Plasmonic Oxides

Based largely on an accepted manuscript in press in Sensors and Actuators B

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an Oscillation with an Associated Resonance.

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High Temperature Gas Sensing Using Au-Nanoparticle Incorporated Oxides

Au / TiO₂

Gold Nanoparticle-Doped TiO2 Semiconductor Thin Films: **Gas Sensing Properties****

By Dario Buso, Michael Post, Carlo Cantalini, Paul Mulvaney, and Alessandro Martucci*

Adu. Funct. Moter. 2008, T.S. 3843-3849.

Au / YSZ

Plasmonic Based Kinetic Analysis of Hydrogen Reactions within Au-YSZ Nanocomposites

Nicholas A. Joy, Charles M. Settens, Richard J. Matyi, and Michael A. Carpenter*

Au / WO₃ Optical hydrogen sensitivity of noble metal-tungsten oxide composite films prepared by sputtering deposition

Masanori Ando^{a,*}, Rupert Chabicovsky^b, Masatake Haruta^a

Sensors and Actuators B 76 (2001) 13-17

Fig. 8. Difference spectra obtained by subtracting the absorption spectra of the Au-WO₃ composite film in fresh air from those in air containing 1 vol.% H_2 at temperatures of: (a) 200°C; (b) 250°C.

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dxdblorg/10.1021/jp1122.20h J. Mys. Chem. CXXXX, XXX, 000-000

Figure 2. Example of the LSPR absorption spectra acquired during the experiment and the corresponding Lorentzian fits. The inset illustrates the ~3-5 nm peak shift upon gas exchange.

"Catalytically Active Oxides" Examples: Au / NiO, Au / Co₃O₄, Au / YSZ, Au / TiO2, Au / ZnO, Others

Au Nanoparticles in SiO₂

"Inert Oxides"

*Examples: Au / SiO*₂, *Au / ZrO₂, Au / Al₂O₃*

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In-Situ Optical Measurements on Film Samples

Simple Schematic

We Have Designed and Built a System that Allows us to Monitor Optical Transmission of Film Samples at Elevated Temperatures.

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High Temperature Stable Systems: Au Incorporated "Inert" Oxides

Despite a Low Electronic and lonic Conductivity of SiO₂, Significant **Responses to H₂ Have Been Observed For Au / SiO₂ Films Resulting in an LSPR Absorption Peak Shift to Shorter .**

SiO₂ and Related Systems are Expected to Exhibit Improved Temperature Stability as Compared to Other More Active Oxides (TiO₂, WO₃, etc).

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Theoretical Origin of Au LSPR Temp. Dependence

Wavelength Dependences of the Sensitivity to Changes in Chemical Composition and Temperature of a High Temperature Gas Stream are Different.

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Examples of Au-Incorporated Oxides

A Series of Different Au-Nanoparticle Incorporated Oxides were Prepared Using Standard Wet-Chemistry Based Sol-Gel Recipes.

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Optical Spectrum Dependence on Gas Stream Temperature and Composition

Transmittance Spectra of Synthesized Films are Sensitive to Both the Temperature and the Gas Stream Composition.

The Wavelength Dependences for Each Response are Unique.

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Wavelength Dependent Responses

By Interrogation in Different Wavelength Ranges, One Expects to Identify Responses that are Preferentially Sensitive to Chemical Composition or Temperature of a Gas Stream.

For Oxides with a Well-Resolved Band-Edge in the Wavelength Range of Interest, Band-Edge Monitoring Can Also Be Exploited.

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Optical Temperature Sensing Using Oxides

"Optical temperature sensor based on ZnO thin film's temperature dependent optical properties", S. Chengua et al., Review of Scientific Instruments, 82, 084901 (2011).

 $E_{q} = E_{0} - \alpha T^{2}/(T + \beta)$ Varshni, Physica 34, 149 (1967)

Metal Oxides Such as ZnO Have Been Proposed for Use as Temperature Sensors By Monitoring the Film Transmission in the Vicinity of the Band-Edge.

Temperature Dependent Bandgap Arises Due to Two Primary Factors:

- **1) Dilatation of the Lattice (Accounts for a Fraction, Linear at High T.)**
- **2) Temperature Dependent Electron- Lattice Interaction (Major Effect)**

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Gas Sensing at Wavelengths Near Gas

INSERT REFERENCES HERE.

Gas Sensing Responses are Observed at a Wavelength Near the Peak Sensitivity.

Recovery Kinetics after Exposure to a H ²-Containing Atmosphere are Strongly Dependent Upon the Presence of O 2.

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Through Careful Wavelength Selection, One Can Achieve Enhanced Relative Sensitivity to Gas Stream Composition or Temperature.

Band-Edge Interrogation Is Particularly Effective for Temperature Sensing.

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Gas Sensing Responses in High H $_{2}$ **Concentrations**

In All Three Material Systems Investigated, Saturation Effects are Observed for High H2 Concentration Gas Streams at Interrogation Wavelengths Greater than the LSPR Absorption Peak Wavelength.

Short Wavelength Interrogation Below the LSPR Absorption Peak Enables Improved High Level H2 Sensing.

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Fabricated Fiber Sensor Results

P. R. Ohodnicki et al., Nanoscale, Vol. 5 (19), 9030-9039, 2013.

the Temperature Dependent Optical Transmission YFOTASOZY LABOSATOSY

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Recent R&D Updates : Realistic Contaminated Fuel Gas Stream Exposure

Based largely on an accepted manuscript in press in Sensors and Actuators B

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Realistic Fuel Gas Stream Composition Estimated Contaminated Anode Stream Composition: 34.0%H₂, 23.6%CO, 22.2%H₂O, 17.0%CO₂, 3.2%N₂, 0.026%CH₄, 29ppm H₂S

Temperature = 800oC, Approximately 1 Week Exposure

The Oxide Matrix Phases and Au are All Expected to be Stable in the Chosen Fuel Gas Stream.

All Samples were Subjected to a Realistic Fuel Gas Stream Exposure to Gain Preliminary Information About the Sensor Material Stability.

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Some Material Instability was Observed, Au-Nanoparticles Tended to Stabilize Base Oxide Microstructure… But Au Mass Loss was ObservedNATIONAL ENERGY TECHNOLOGY LABORATORY ‹#› *Crosscutting Research Review, Pittsburgh, PA, May 21,2014*

Kinetic Stabilization of Anatase Phase

Au-Nanoparticles Tend to Inhibit the Anatase Rutile Phase Transformation, Likely Due to Grain Boundary Pinning Effects

Locally Obtained Electron Backscatter Diffraction Patterns Illustrate The Large Grains Which Form After the Exposure Step are Rutile Phase in the Au / TiO2 Films

> **No Tendency for Preferred Orientation of the Large Rutile Grains Were Identified**

Additional Studies are Needed of Sensor Material Stability Under Contaminated Fuel Gas Conditions Which are Underway.DGY LA30?ATO?Y

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Summary and Conclusions

- **1) The Team is Actively Working to Develop Embedded Sensors for High Temperature Fossil Applications with an Emphasis on Advanced Sensor Materials**
- **2) The Team Welcomes Opportunities to Discuss New Collaborations and Partnerships.**
- **3) Promising Results for Simultaneous Gas and Temperature Sensing under Application Relevant Temperatures Have Been Demonstrated in Au-Incorporated Oxides But Additional Studies of Stability in Contaminated Fuel Gas Streams are Required**
	- **4) Broadband Wavelength Interrogation Approaches May Enable Multi-Parameter Functionality of Sensors (Temperature, Gas Composition)**
- \rightarrow **Unique Wavelength Dependences of LSPR Absorption on Temperature and Chemical Composition of a Gas Stream**

→ Combined Band-edge and LSPR Sensing

Thank you for your attention!

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Acknowledgements and Disclaimer

This work was funded by the Cross-Cutting Technologies Program at NETL.

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