

Long-Term Degradation of LSM-Based SOFC Cathodes: Use of a Proven Accelerated Test Regimen

Mark R. De Guire,¹ Arthur H. Heuer,¹ and Zhien Liu²

1) Dept. of Materials Science and Engineering, Case Western Reserve University, Cleveland, OH 44106;
2) LG Fuel Cell Systems Inc., North Canton, Ohio 44720

Overview

The overall goals of this SECA Core Technology Program (CTP) project are to understand the microstructural basis of performance loss in cathodes based on lanthanum strontium manganite (LSM, $(La_{1-x}Sr_x)_{1-y}MnO_{3\pm\delta}$), and to develop strategies for optimizing the long-term performance of LSM-based cathodes for commercial fuel cell systems (Figure A). Here we present results from recent LGFCS-CWRU collaborations as examples of work to be carried out under this CTP project. We will implement an accelerated testing protocol, developed and validated by LGFCS, to gather performance data at CWRU in time frames of e.g. 500 h that are relevant to much longer-term normal cell operation (≥ 5 kh) (Figure B), and compare these findings to results from non-accelerated tests (Figure C). Tested cells will undergo detailed microstructural and microanalytical studies (Figures D, E, F, and G). Analysis of performance histories, microstructural studies, and computational thermodynamic results (Figure H) will lead to design rules for improving the long-term stability and performance of LSM-based cathodes. We will iteratively refine the proposed design rules (GenA – GenD, Figure A), with a view to identifying and quantifying the mechanisms that lead to loss of performance in LSM-based cathodes.

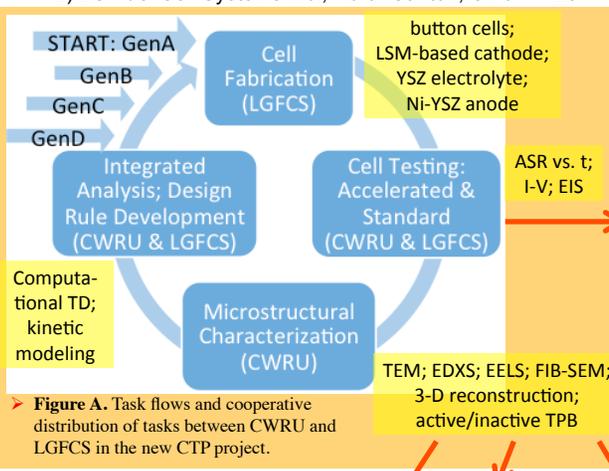


Figure A. Task flows and cooperative distribution of tasks between CWRU and LGFCS in the new CTP project.

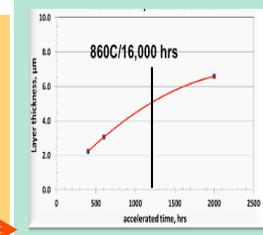


Figure B. Thickness of densified cathode layer vs. duration of accelerated testing. The vertical line indicates that 1.2 kh under the accelerated conditions produces the same thickness of densified cathode (5 µm) observed after 16 kh of testing under simulated system conditions (Figure F).

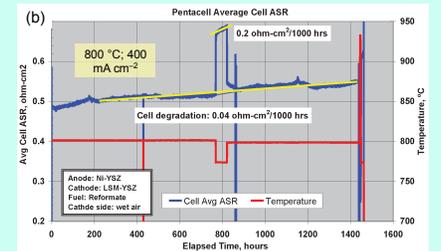
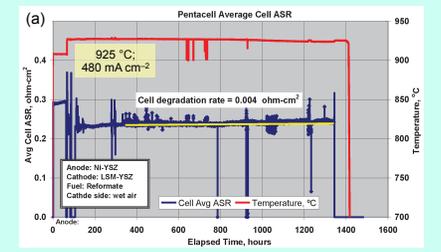


Figure C. Cell ASR vs. time, with H₂O in cathode environment, as a function of operating temperature: (a) 925 °C, 480 mA cm⁻²; (b) 800 °C, 400 mA cm⁻². Both cells were tested with reformate at the anode [1].

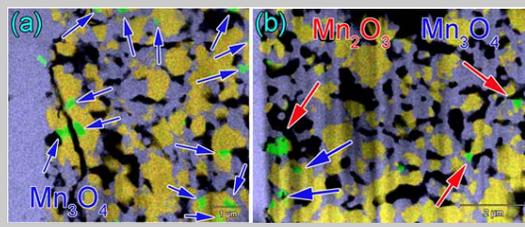


Figure G. EDXS maps of LSM-YSZ cathodes. EELS analysis indicated that Mn-oxides in cells operated at 925 °C were Mn₃O₄ (a). Cells operated at 800 °C contained Mn₂O₃ and Mn₃O₄ (b) [1].

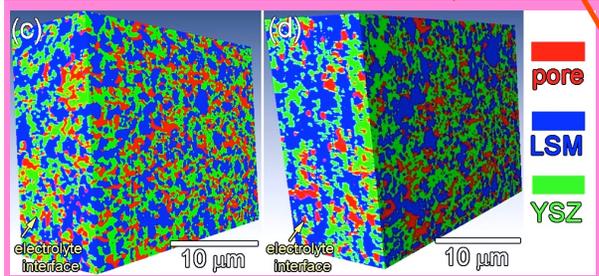


Figure E. 3D reconstructions of the cathode (electrolyte at front left) for cells after testing at 860 °C for (c) 8 kh and (d) 16 kh [2]. (3D studies supported by Ohio Department of Development's Third Frontier Program.)

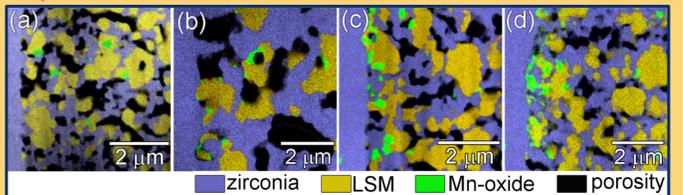


Figure D. EDXS maps acquired near the cathode-electrolyte interface for (a) the as-fired cell; (b) the 800 °C-2 kh cell; (c) the 800 °C-8 kh cell; (d) the 800 °C-16 kh cell. (The electrolyte is at the left edge in each image.) [2]

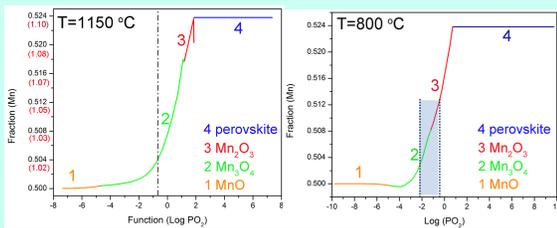


Figure H. Computational thermodynamic predictions of ranges of p_{O_2} over which Mn-containing oxide phases are stable at 1150 °C (left) and 800 °C (right) [3]. Fired at 1150 °C in air during cell fabrication, Mn₃O₄ formation is predicted (left). During operation at 800 °C, a typical atmosphere near the cathode-electrolyte interface should lead to coexistence of Mn₂O₃ and Mn₃O₄ (right). Both predictions are consistent with the TEM/EELS results in Figure G.

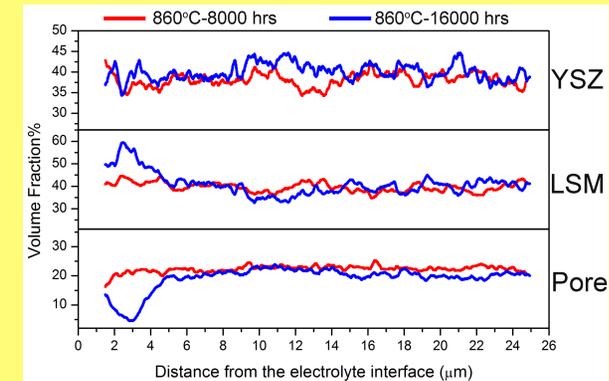


Figure F. Volume fraction profiles of YSZ, LSM, and porosity in cathodes tested at 860 °C (Figure E) for 8 kh (red) and 16 kh (blue) [2].

Summary

The accelerated testing protocol enables the study of microstructural changes and effects on SOFC performance that are characteristic of years of testing, within time frames (~500 h) amenable to applied laboratory-scale SOFC R&D. This brings, within the scope of a three-year program, the ability to modify cathode formulations and test hypotheses for their optimization. Because one of the key challenges in advancing SOFC technology is the difficulty of studying (and understanding) the mechanisms that lead to critical long-term performance loss, the research proposed here has the potential for breakthrough impact in improving the reliability, robustness, and longevity of SOFC technology.

Acknowledgment: This research is based upon work supported by the U. S. Department of Energy, National Energy Technology Laboratory under Award Number DE-FE000303. Disclaimer: This research is based in part upon work supported by an agency of the United States Government. Neither the United States Government nor any agency thereof, nor any of its employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by trade name, trademark, manufacturer, or otherwise does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof, or LG Fuel Cell Systems Inc. Acknowledgment: The presenter acknowledges the support of the State of Ohio, the Ohio Department of Development and the Ohio Third Frontier, in general, for the financial support provided to the research. Disclaimer: The research referenced in this poster was prepared in part with financial support from the State of Ohio. The content reflects the views of Case Western Reserve University and does not purport to reflect the views of the State of Ohio, Ohio Department of Development, the Ohio Third Frontier or LG Fuel Cell Systems Inc.

