



FARADAY  
TECHNOLOGY, INC.

# *Electrodeposition of SOFC Interconnect Coatings*

<sup>1</sup>H. McCrabb, <sup>1</sup>T.D. Hall, <sup>1</sup>J. Kell, <sup>1</sup>S. Snyder, <sup>2</sup>H. Zhang, <sup>2</sup>X. Liu,  
<sup>1</sup>E.J. Taylor

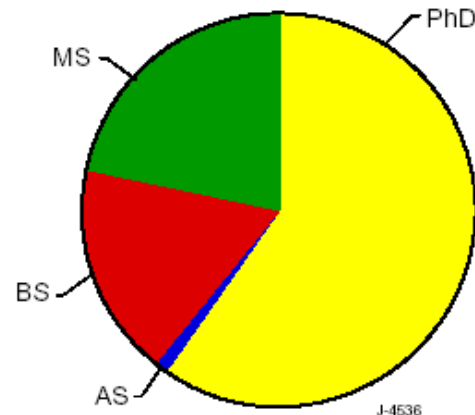
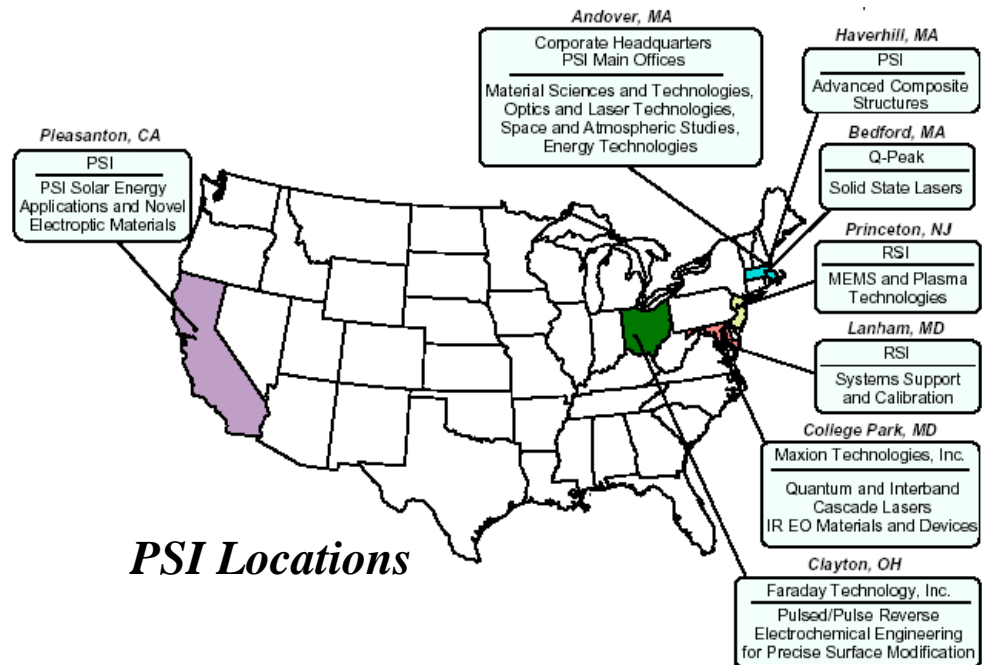
<sup>1</sup>Faraday Technology, Inc. 315 Huls Dr., Clayton, OH 45315

<sup>2</sup>West Virginia University, Dept. of Mechanical Aerospace Eng. ESB, Morgantown, WV  
26506

12<sup>th</sup> Annual SECA Workshop  
July 26 – 28, 2011

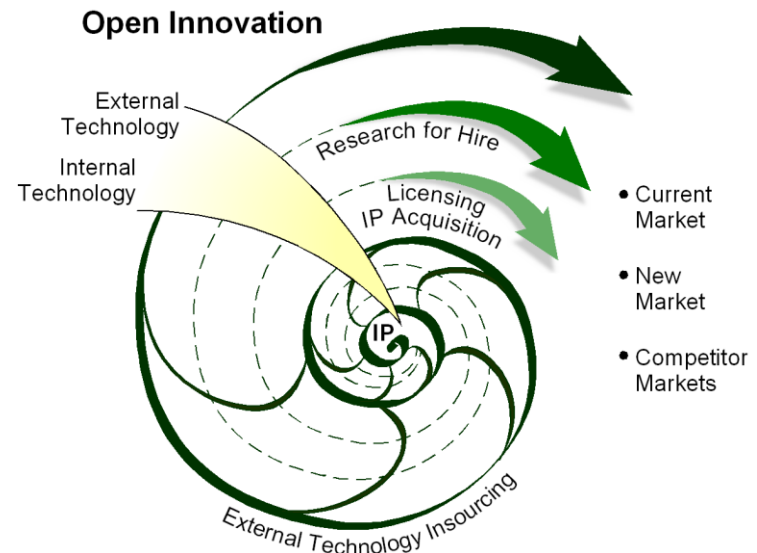
# Faraday Technology, Inc.

- Faraday Technology specializes in electrochemical engineering
  - [www.faradaytechnology.com](http://www.faradaytechnology.com)
- Faraday is a wholly-owned subsidiary of Physical Sciences, Inc. (Boston, MA)
  - [www.psicorp.com](http://www.psicorp.com)
  - Collectively, the company staffs ~185 employees - ~100 with PhDs
  - Annual revenue of ~ \$50M



# *Faraday Embraces Open Innovation*

- Faraday's R&D efforts augment client company R&D needs
- Open innovation leverages government R&D dollars to solve current industrial issues.
- Faraday's strong IP portfolio reinforces our open innovation position and provides a competitive advantage for strategic partners.

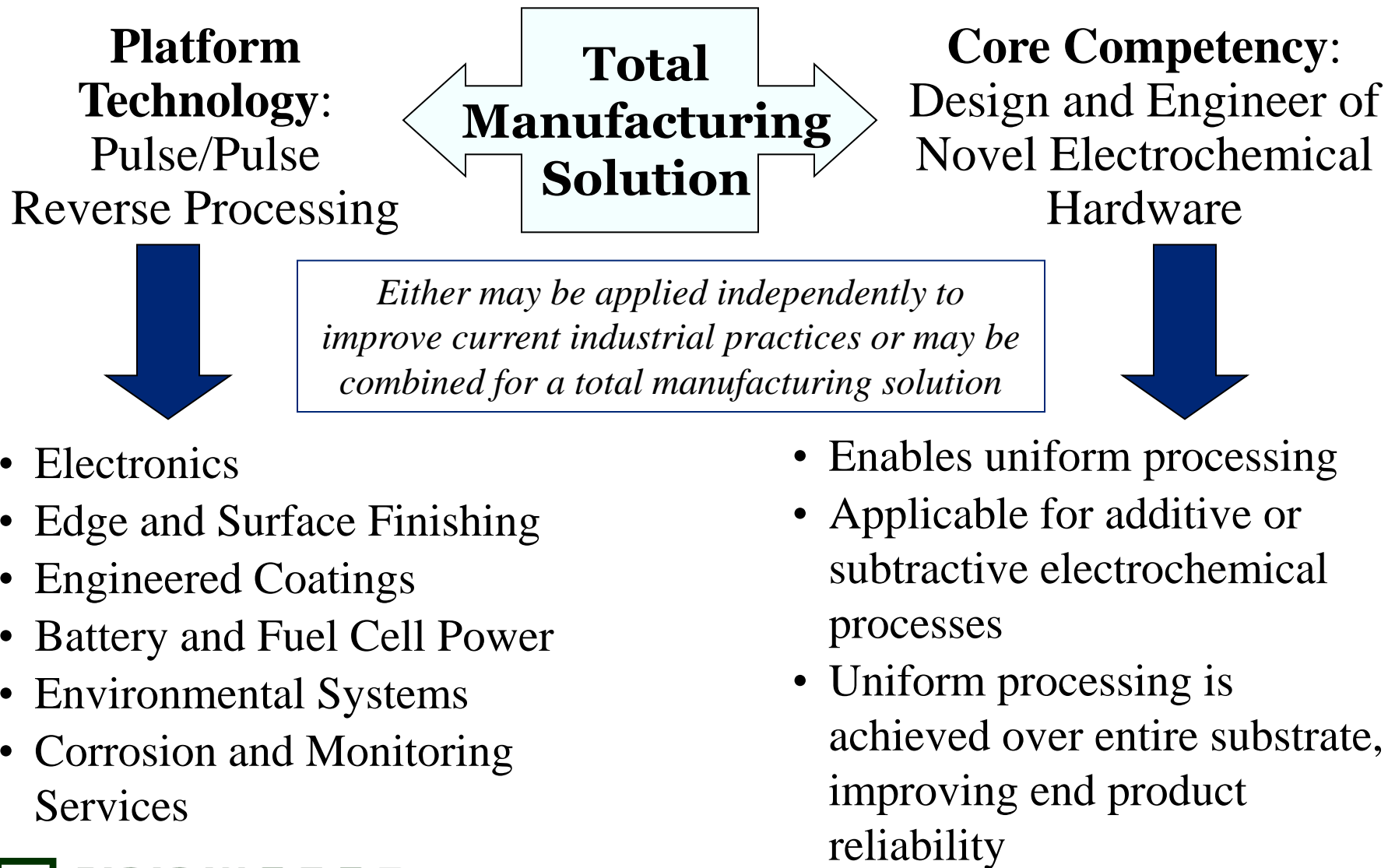


*26 Issued Electrochemical Patents*  
*25 Pending Electrochemical Patents*

**Investment Dollars**  $\xrightarrow[\text{Research}]{\text{Discovery}}$  **Knowledge**  $\xrightarrow{\text{Innovation}}$  **Market Dollars**



# *Faraday Technology, Inc.*



# *Objective of Program*

- Develop, optimize & validate an inexpensive manufacturing process for coating metallic SOFC interconnects with Co and Mn
  - Demonstrate the process's flexibility to 4"x4" and 10" x 10" single and dual-sided patterned interconnect substrates
  - Control coatings nanostructure and composition to prevent T441 exposure to O<sub>2</sub> and Cr diffusion to coating surface
  - Mitigate Cr diffusion by identifying diffusion mechanism via in-situ high temperature XRD and ex-situ XPS depth profiling
  - Develop a comprehensive economic assessment
  - Work closely with our the SOFC industry to enhance the commercialization plan for the program.

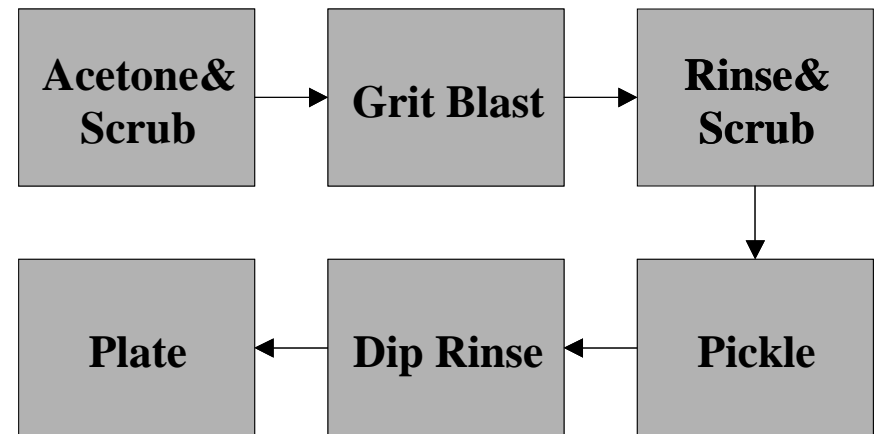


# *Program Summary*

- FARADAYIC Process allows for non-line-of-sight deposition of a wide range of compositions and surface structures using a single plating bath
- Coatings prepared using the FARADAYIC Process have uniform surface composition and thickness on 2" x 2" flat panels
- Coatings exhibit adequate adhesion
- Initial ASR and crystallinity analysis showed that the as-deposited thickness/composition had little effect on performance after a 500 hr heat cycle
- 3 $\mu$ m thickness is capable of minimizing Cr diffusion for 500 hr testing
- Capability to coat dual-sided 1" buttons, 4" x 4", and up to 11" x 14" panels
- Based on batch manufacturing, the DOE's high volume target of 1,600,000 plates per annum at a cost of ~\$1.23 per 25 x 25 cm interconnect

# *Coating Process*

- Surface pretreatment to remove oxide and enhance coating adhesion
- Electrodeposition to coat interconnects with Mn-Co alloy
  - Pulse and pulse reverse electric fields to control deposit properties
- Elevated thermal treatment to convert alloy to spinel



# FARADAYIC Processing

“Tuned” to:

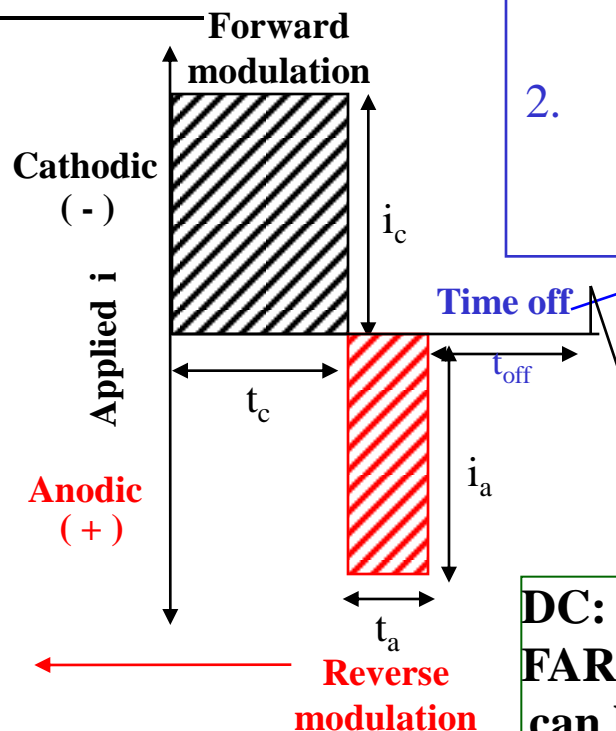
1. enhance mass transfer
2. control current distribution

Controls deposit composition & properties

“Tuned” to remove adverse effects of  $H_2$



Eliminates  $H_2$  embrittlement



“Tuned” to:

1. allow replenishment of reacting species on electrode surface
2. & allow transport of undesirable byproducts from electrode surface

Maintain uniform concentration & hydrodynamic conditions

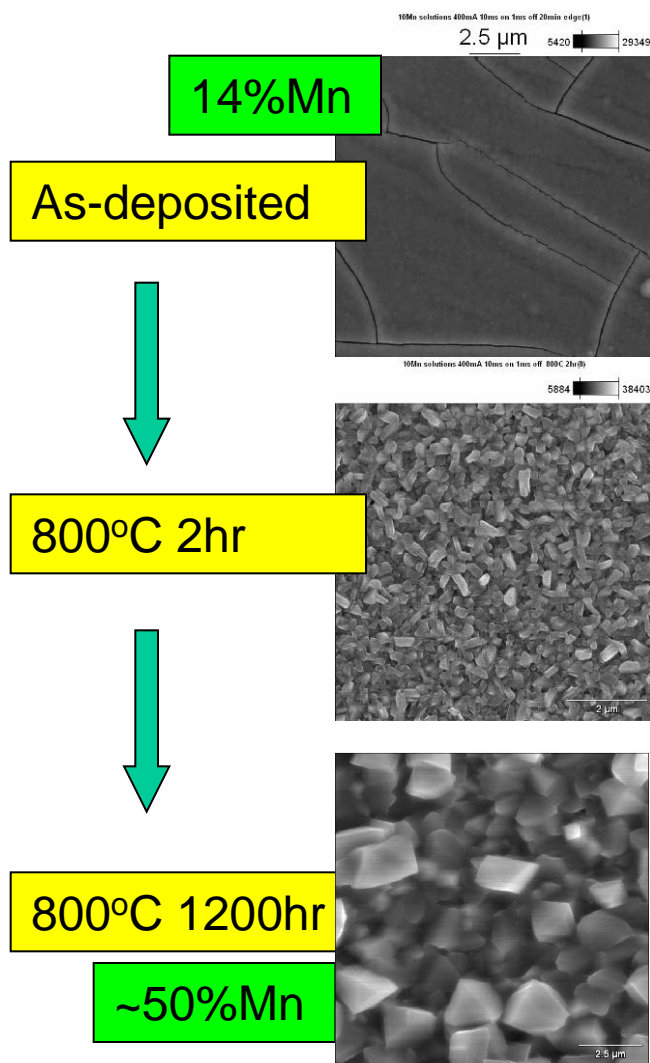
DC: only  $i_{avg}$  can be chosen

FARADAYIC:  $i_a$ ,  $t_a$ ,  $i_c$ ,  $t_c$ ,  $t_{off}$  can be varied independently to achieve a desired rate

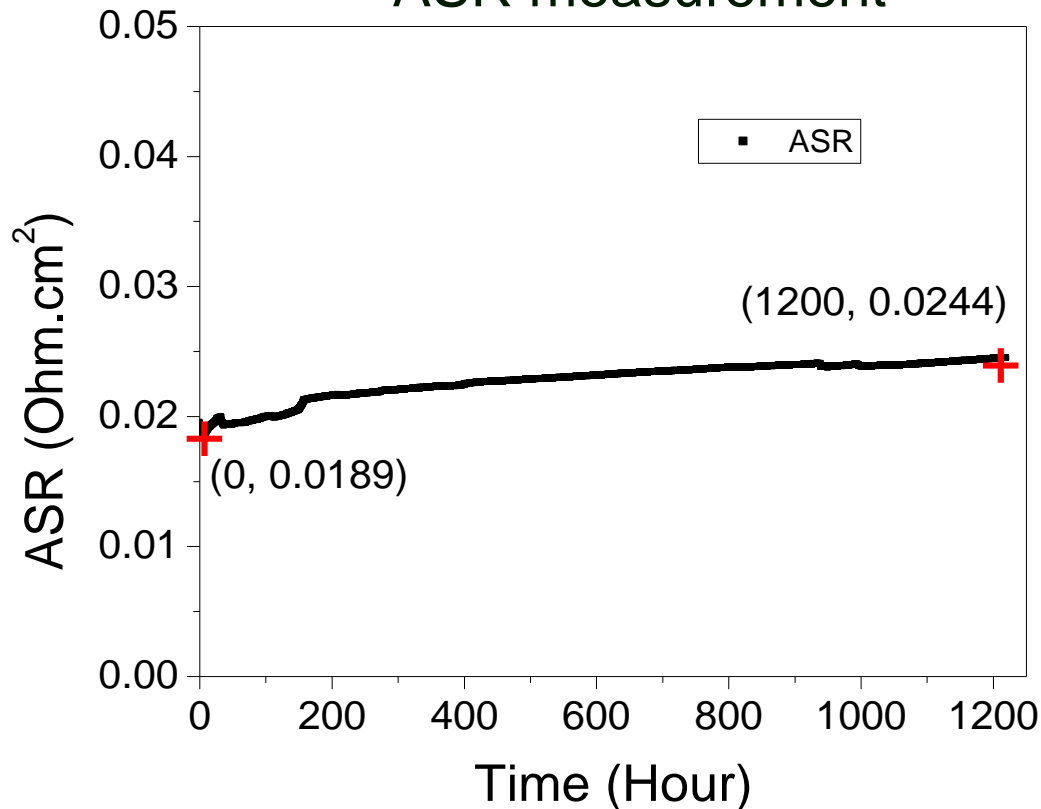




# Initial Coating Development



## ASR measurement



ASR prediction (40,000h, 0.0460 Ohm cm<sup>2</sup>)

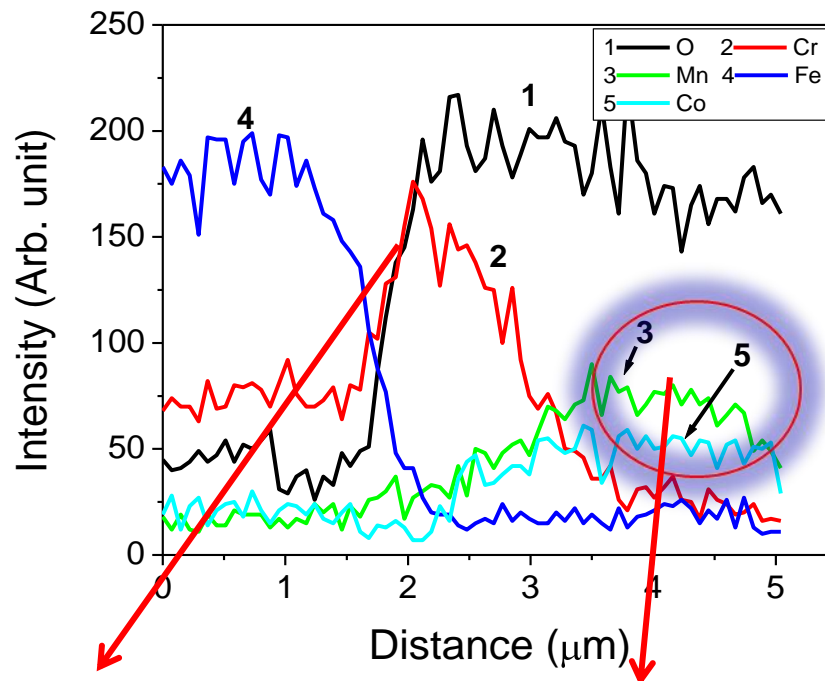
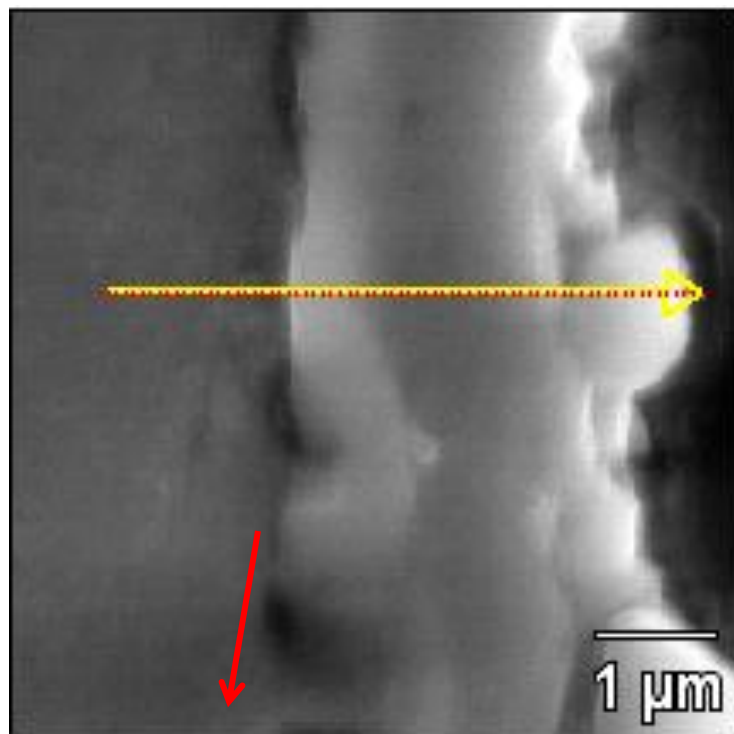
J. Wu, et al., Electrochimica Acta 54 (2008) 793–800



FARADAY TECHNOLOGY, INC.

# Initial Coating Development

Cross section after ASR



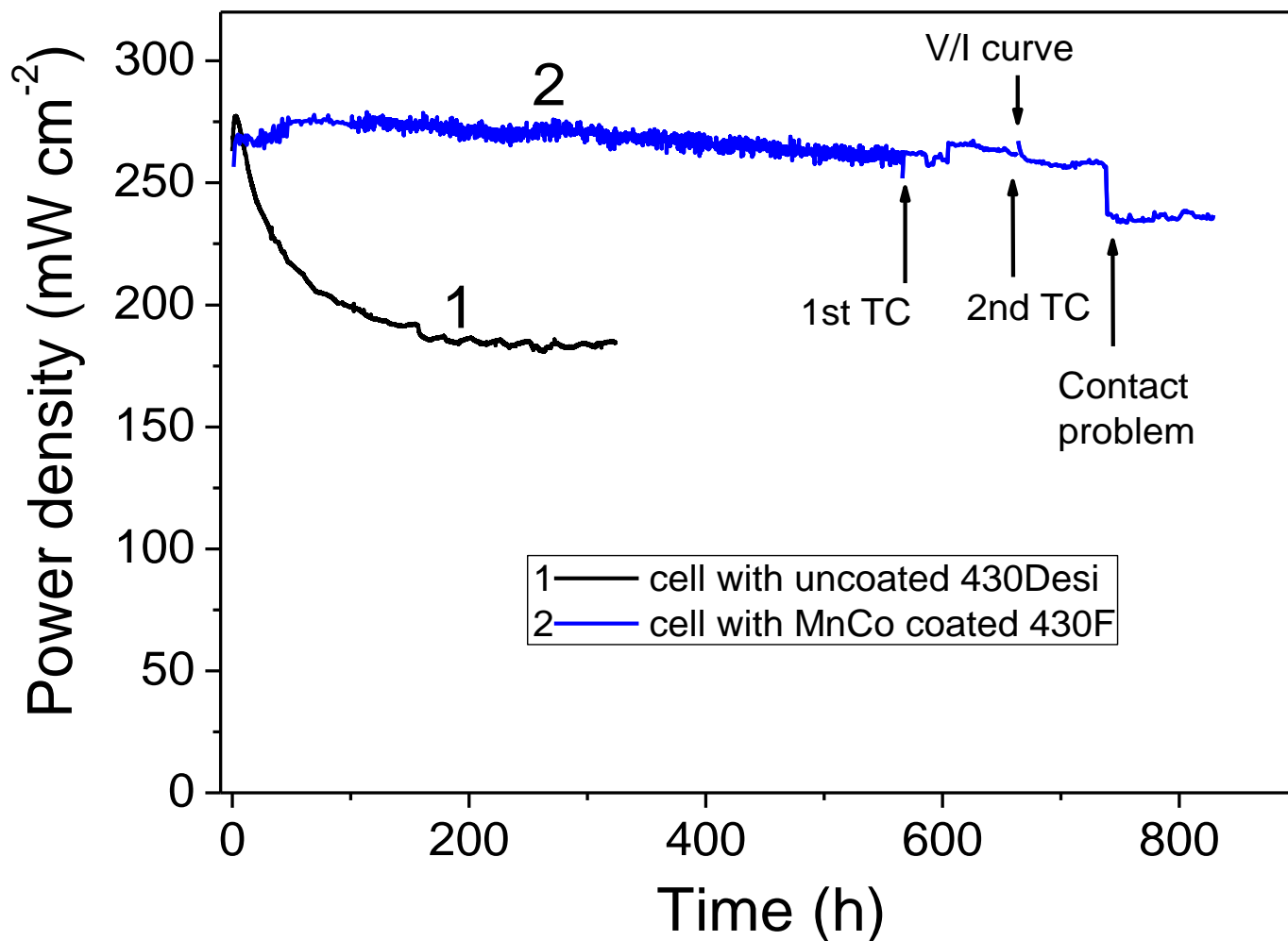
Good adhesion with substrate

No Cr penetrate through the coatings

Significant Mn diffusion from substrate

# Initial Coating Development

Interconnect on button cell test



With electrodeposited MnCo coating, cell performance degradation reduced



**FARADAY** TECHNOLOGY, INC.

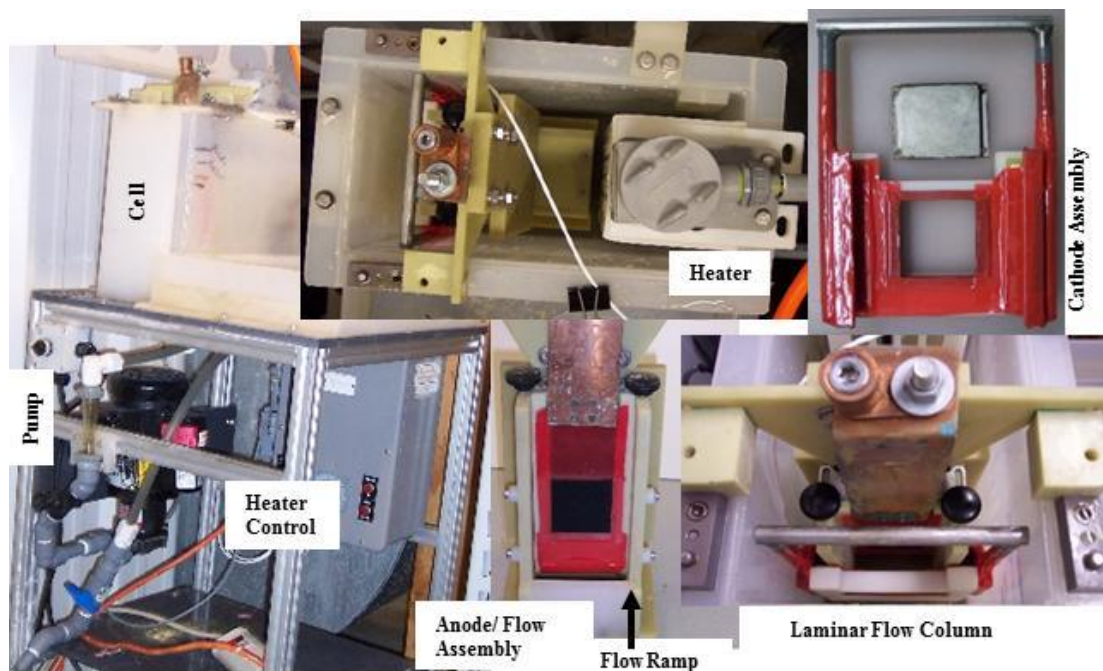
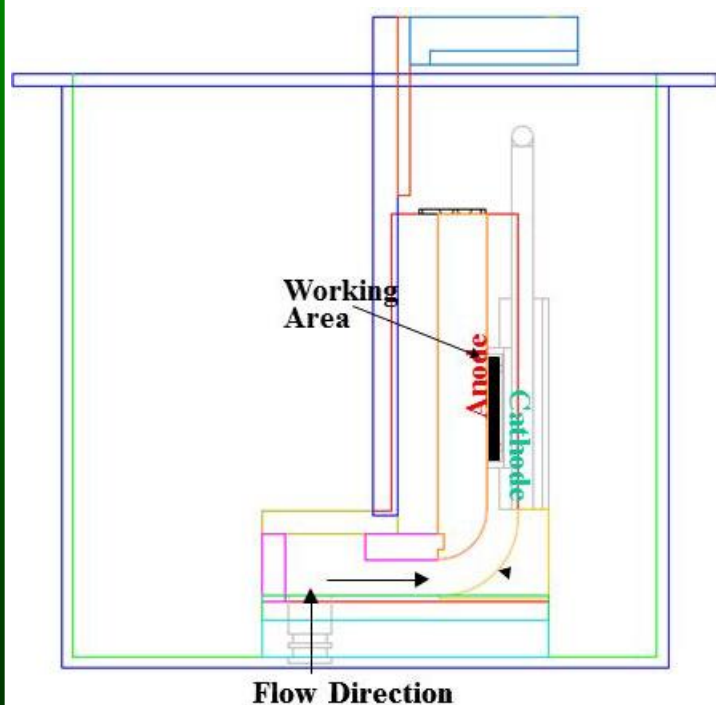
# *Objective of Phase I*

- Demonstrate the technical and economic feasibility of MnCo electrodeposited on SOFC interconnect materials by answering the following questions:
  1. What range of coating percent compositions can the electrodeposition process deposit?
  2. Can the electrodeposition process deposit Mn-Co alloy coatings with a thickness range of 3 – 10  $\mu\text{m}$ ?
  3. How well does the coating perform at varying alloy compositions and coating thickness?
  4. Is the electrodeposition process economically viable?



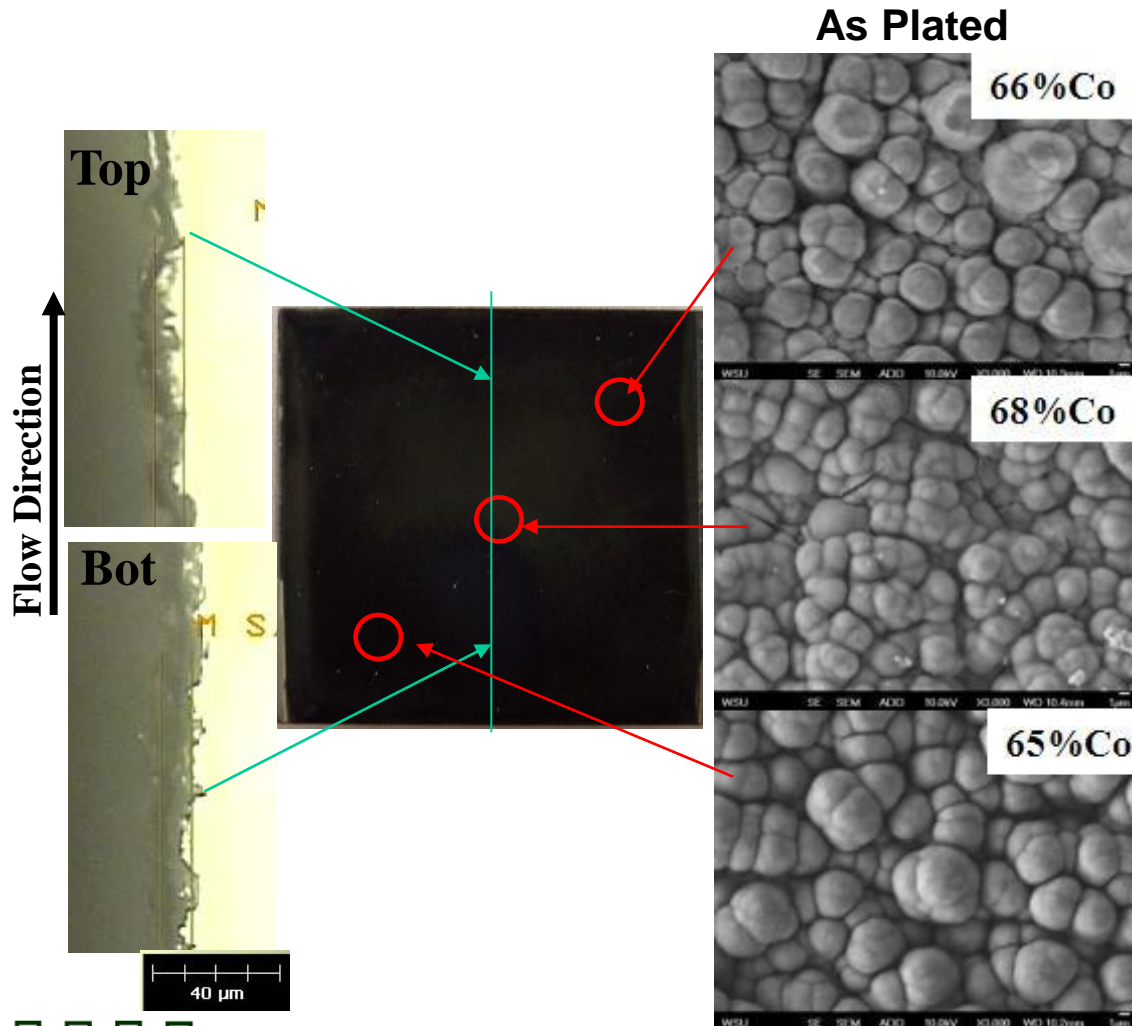
# *Laboratory Scale Electrodeposition Equipment*

Flow cell for plating onto 2"x2" planar T441 substrates



# Coating Uniformity

The electrodeposited coating exhibits a virtually uniform coating composition and thickness across the 2"x2" surface





# Coating Porosity and Adhesion

## *Feroxyl Porosity Test Results (AMS 2460 3.4.4.2)*

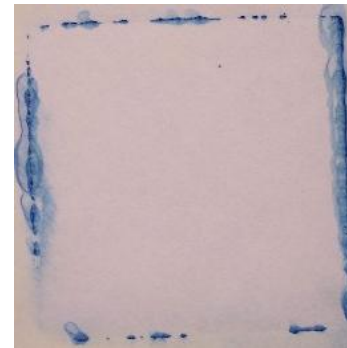
SS Without Coating



Mn-Co Coating As Deposited

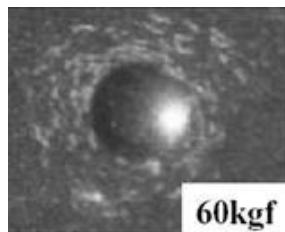


Mn-Co Coating After 72 hour  
Thermal Treatment at 800°C



*\*Rockwell test with 1/16" steel ball used to quantify adhesion after spinel growth after 72 hr. at 800°C*

Without Grit Blast



With Grit Blast



60kgf

100kgf

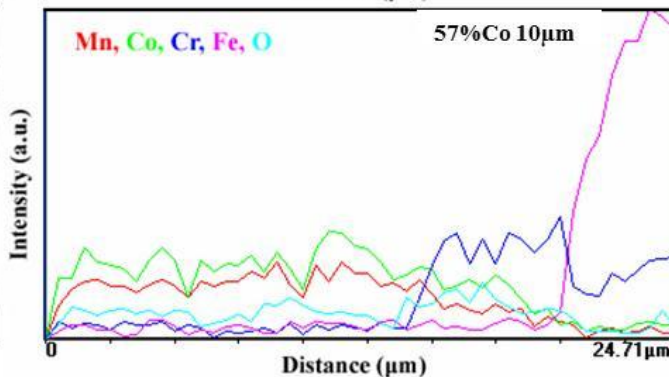
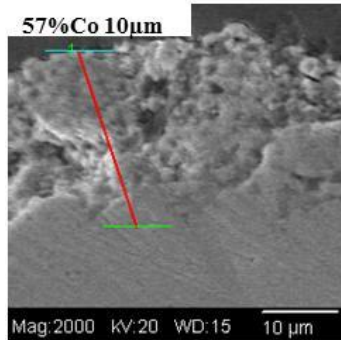
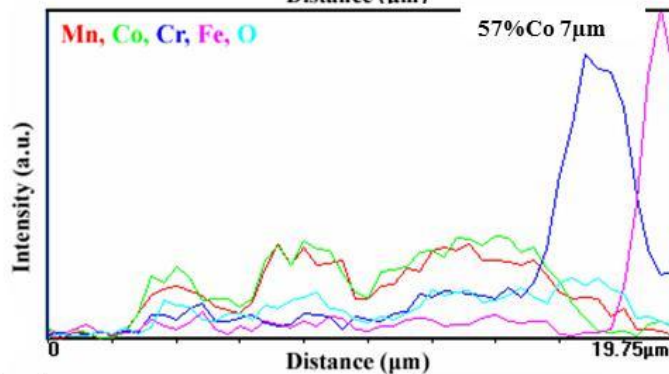
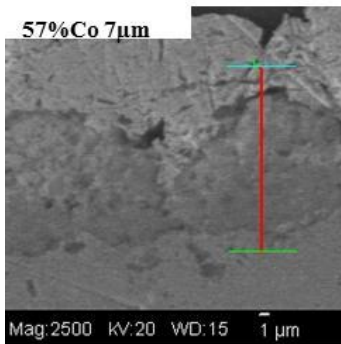
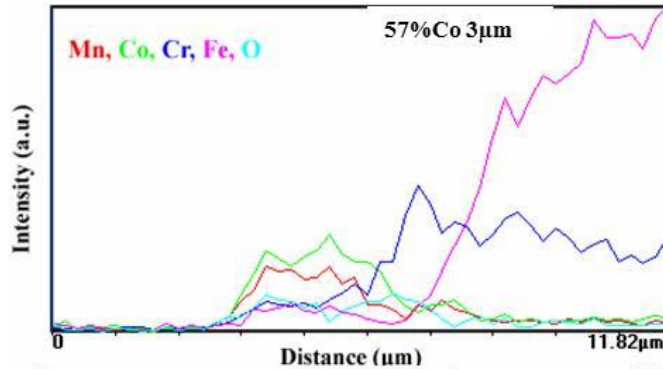
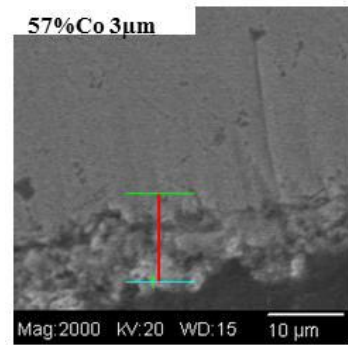
150kgf

\*Sun, X., et. al. JPS, 176 (2008) 167



**FARADAY**   
**TECHNOLOGY, INC.**

# Cr Ion Diffusion and Coating Porosity



Coating surface

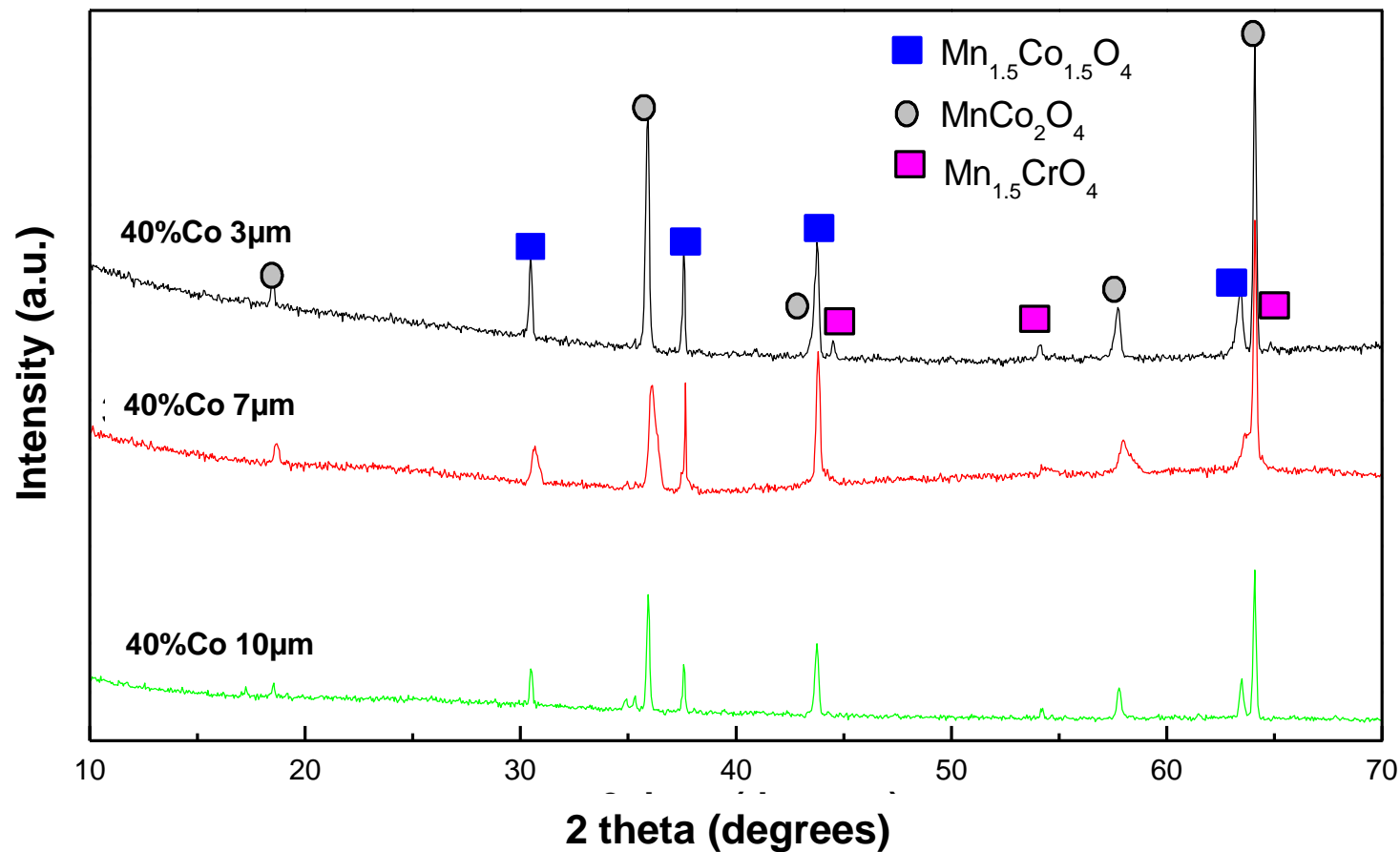
Substrate

- Cross-sections of samples that underwent a soak treatment at 800 C for 500 hrs
  - Coating thickness was as deposited
  - Indicates that a 3 micron layer is adequate to produce Cr complexing and minimize Cr diffusion due to minimal porosity



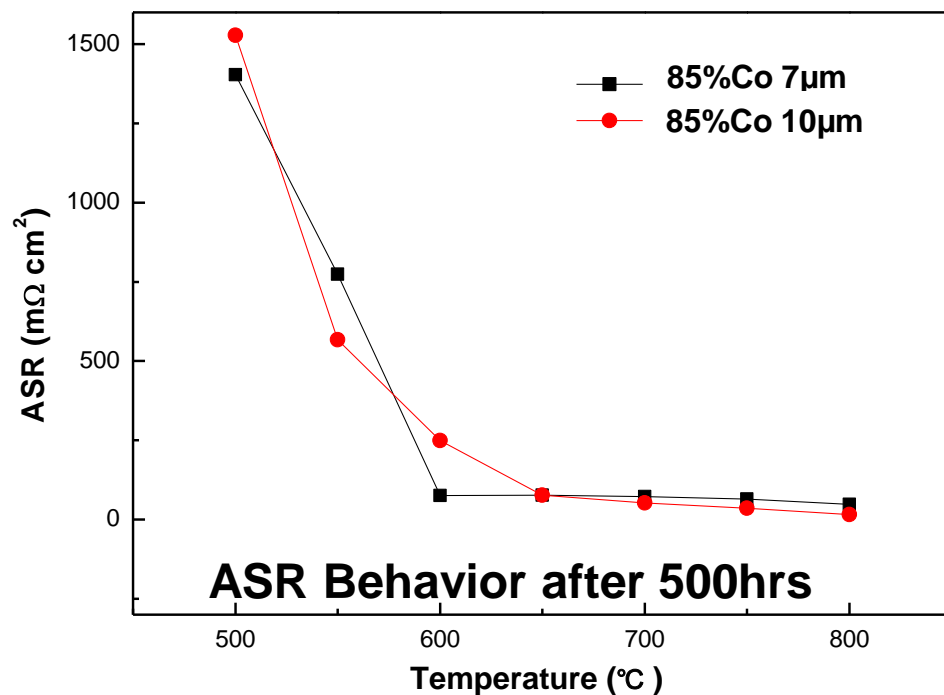
# Coating Crystal Structure

Crystal Structure after 500 hrs at 800°C



# *Effect of Thickness and Composition on Performance*

The ASR is  $\leq 75 \text{ m}\Omega \text{ cm}^2$  regardless of compositions and thickness after 500 hrs at 800 C



## **ASR at 800 C**

mΩ cm <sup>2</sup>	100 hr	200 hr	500 hr
3 μm 40% Co	35	57	49
7 μm 40% Co	62	7	32
10 μm 40% Co	22	-	36
3 μm 85% Co	31	75	20
7 μm 85% Co	59	40	54
10 μm 85% Co	37	23	22
3 μm 57% Co	-	34	26
7 μm 57% Co	-	-	12
10 μm 57% Co	-	-	12



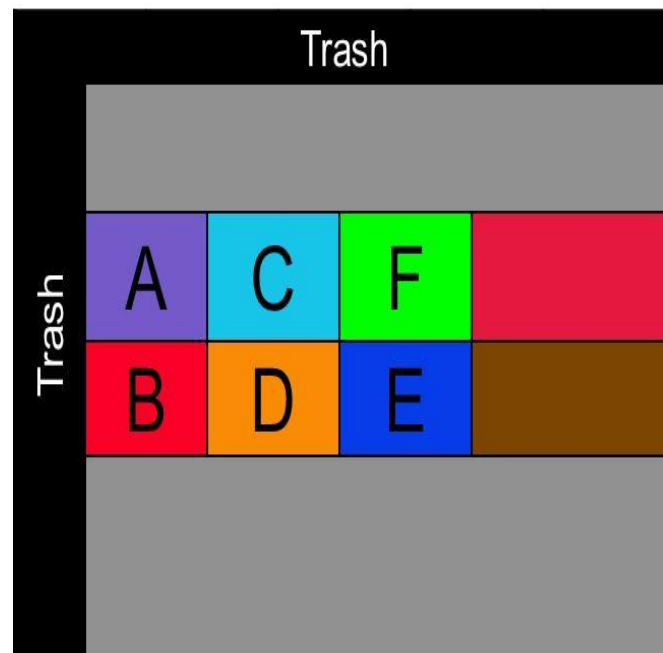
# Phase II Program Milestones

Milestones			
<i>Fiscal Year</i>	<i>Title</i>	<i>Planned Completion</i>	<i>Percent Complete</i>
2011	1. Design/modification of 10" x 10" electrodeposition cell	May 2011	100%
2011	2. Long-term high temperature, thermal evaluation	September 2011	33%
2011	3. Process development for 4"x4" planar interconnects	September 2011	15%
2012	4. Process development for 4"x4" pattern interconnects	June 2012	0%
2012	5. Long-term on-cell performance evaluation	August 2012	0%
2012	6. Qualification/demonstration of IC in single cell test rig	September 2012	0%



# *Extended Thermal Treatment Time Study*

- Six samples were prepared using varying deposition parameters and placed into a tube furnace at 850 C.
  - 2 samples with higher Mn content and 4 with high Co content
  - Target alloy thickness of 5  $\mu\text{m}$
- An air flow of at least 500 sccm flows through the furnace tube (to simulate air in SOFC systems)
- Samples examined after 750, 1500, and 2000 hr

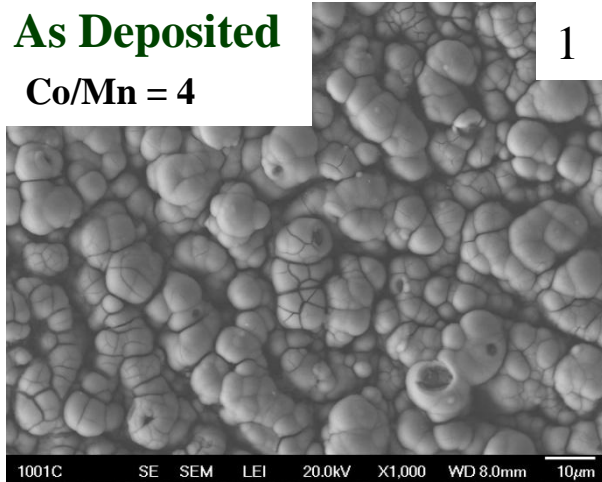


ASR Test order: C( $t=0$ ); D( $t=750$  hrs); E( $t=1500$  hrs); F( $t=2000$  hrs); and A( $t=2000$  hrs) (for Rockwell Test)

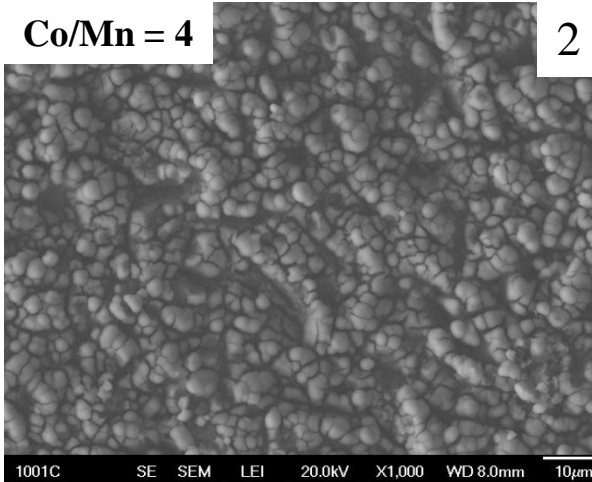


# *Extended Thermal Treatment Time Study*

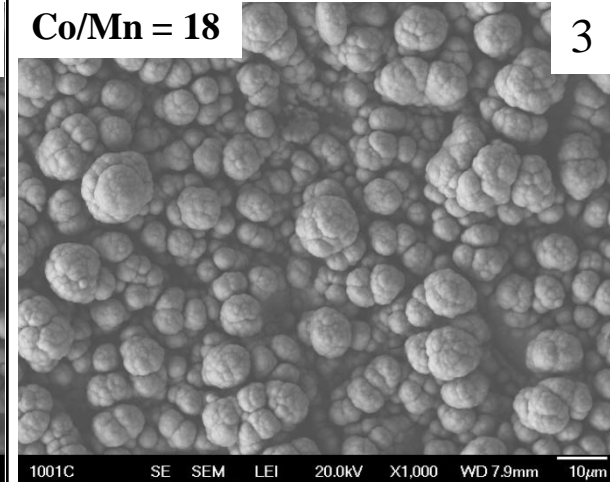
**As Deposited**  
**Co/Mn = 4**



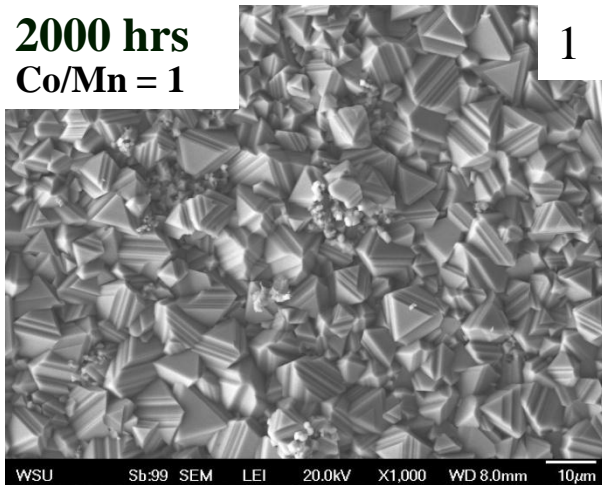
**Co/Mn = 4**



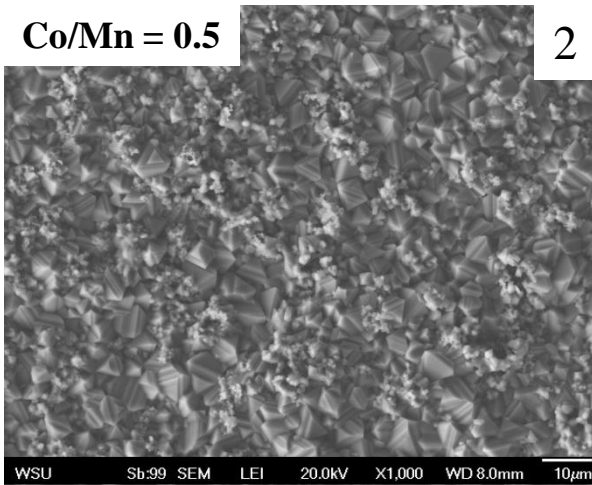
**Co/Mn = 18**



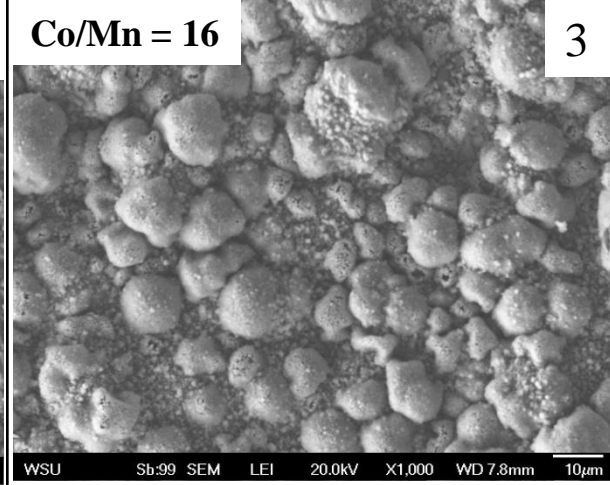
**2000 hrs**  
**Co/Mn = 1**



**Co/Mn = 0.5**

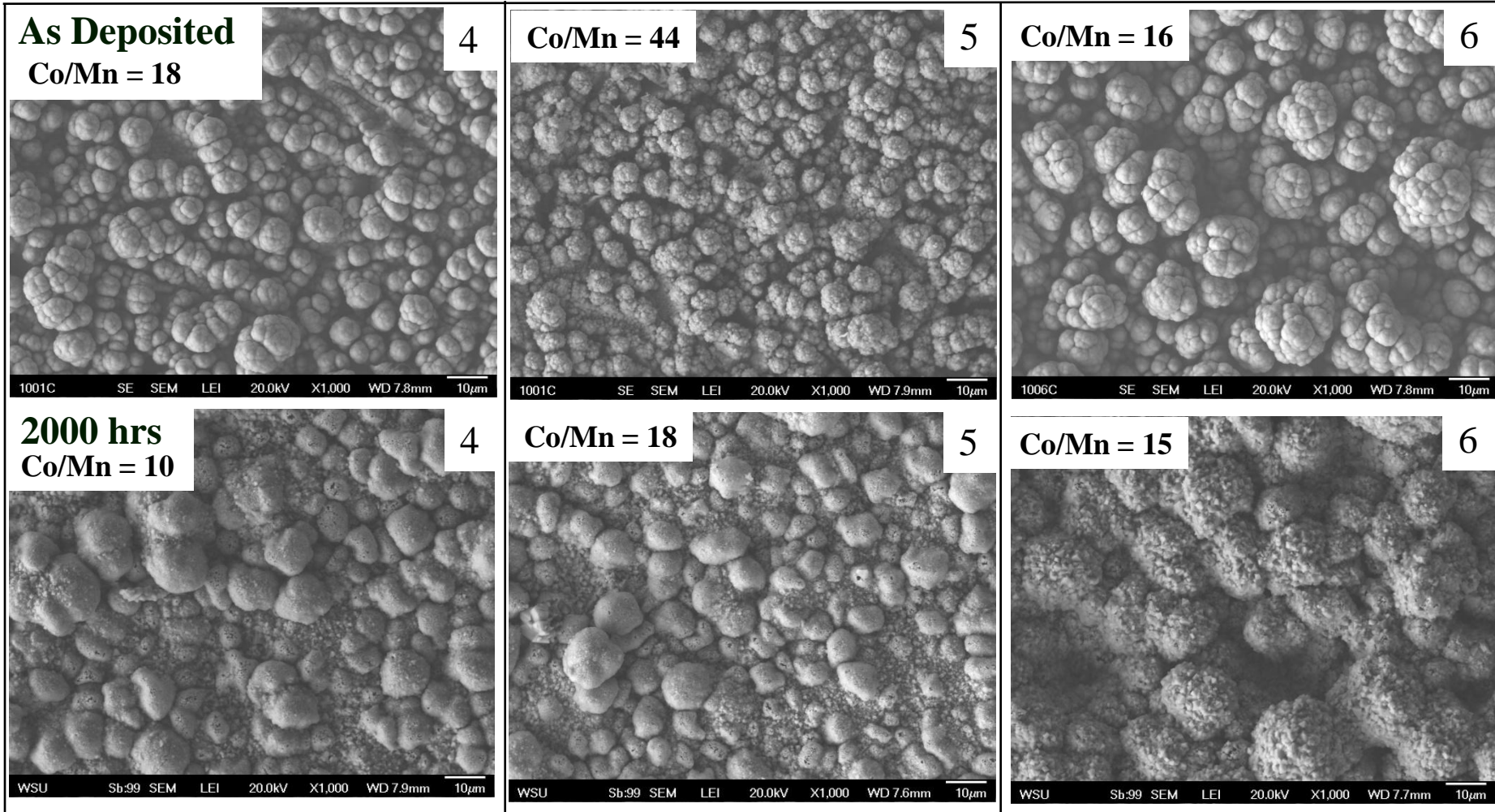


**Co/Mn = 16**



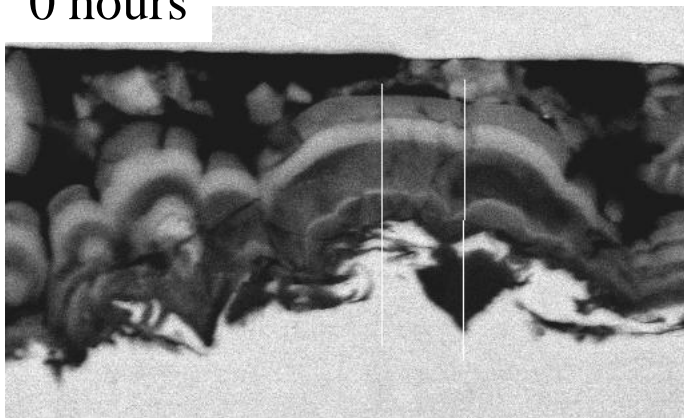


# *Extended Thermal Treatment Time Study*

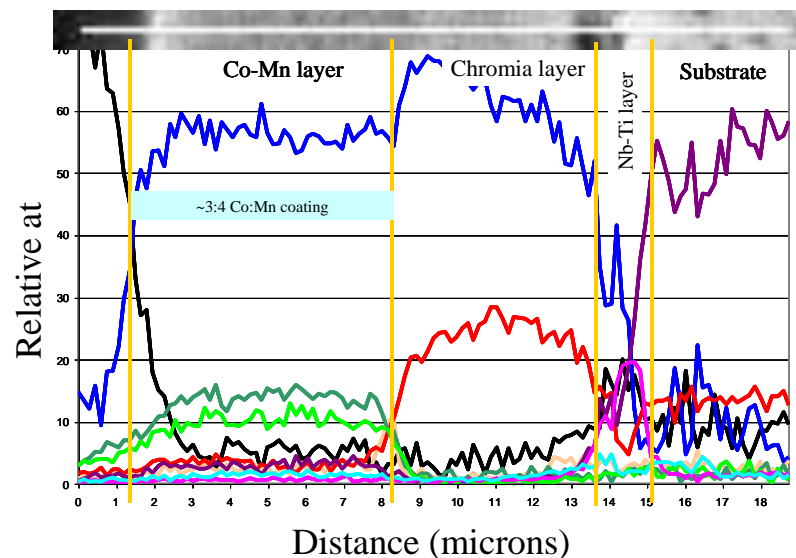
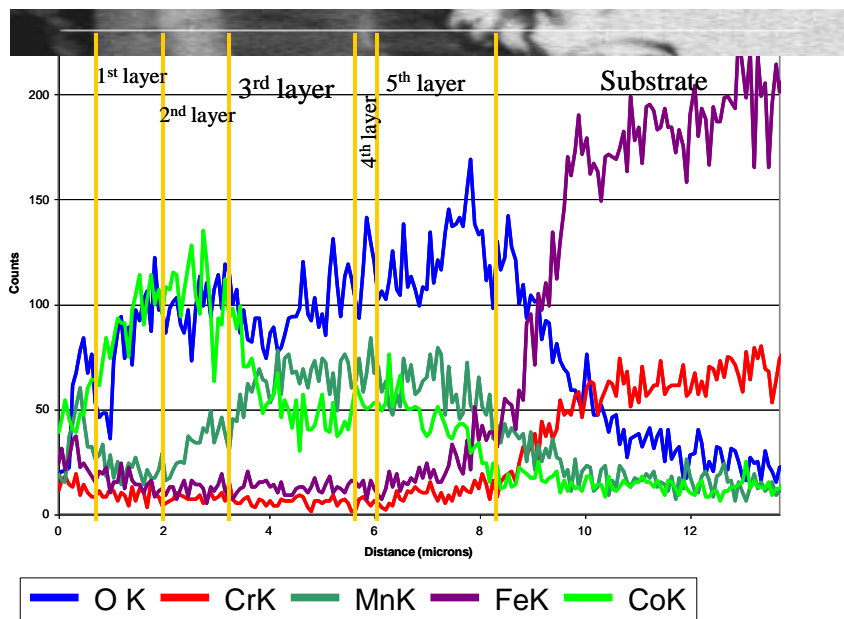
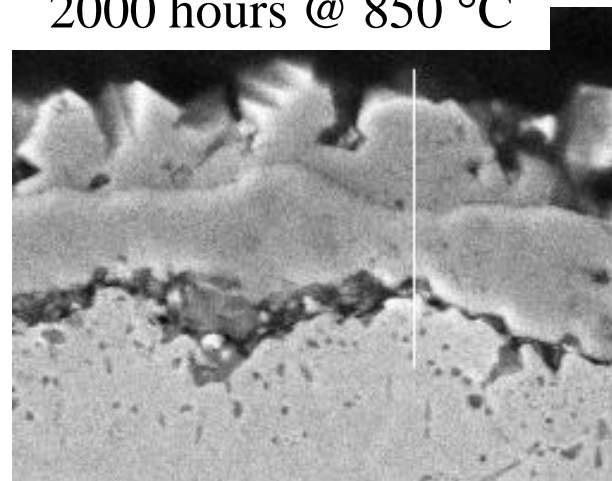


# Cross Section Results - 1001

0 hours

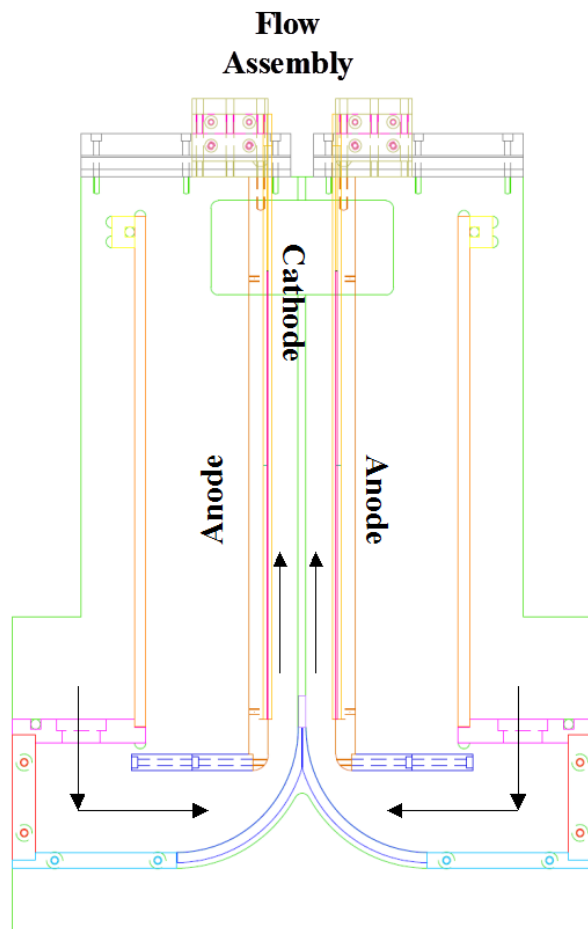


2000 hours @ 850 °C



**FARADAY**  
**TECHNOLOGY, INC.**

# *Pilot Scale Electrodeposition Equipment*



Based upon Faraday's electrochemical cell design that facilitates uniform flow across the surface of a flat substrate (US patent #7,553,401)



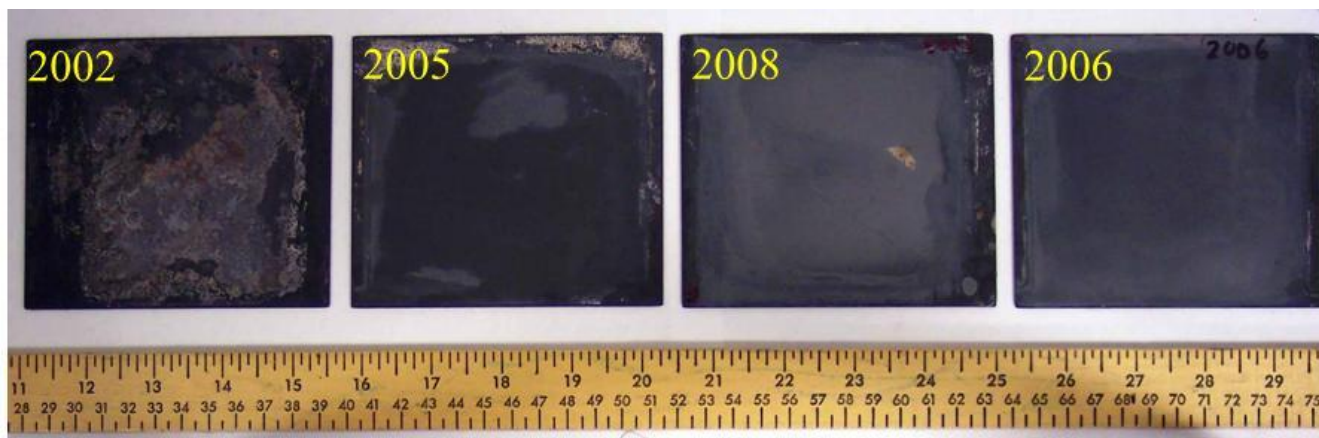
**FARADAY**   
**TECHNOLOGY, INC.**



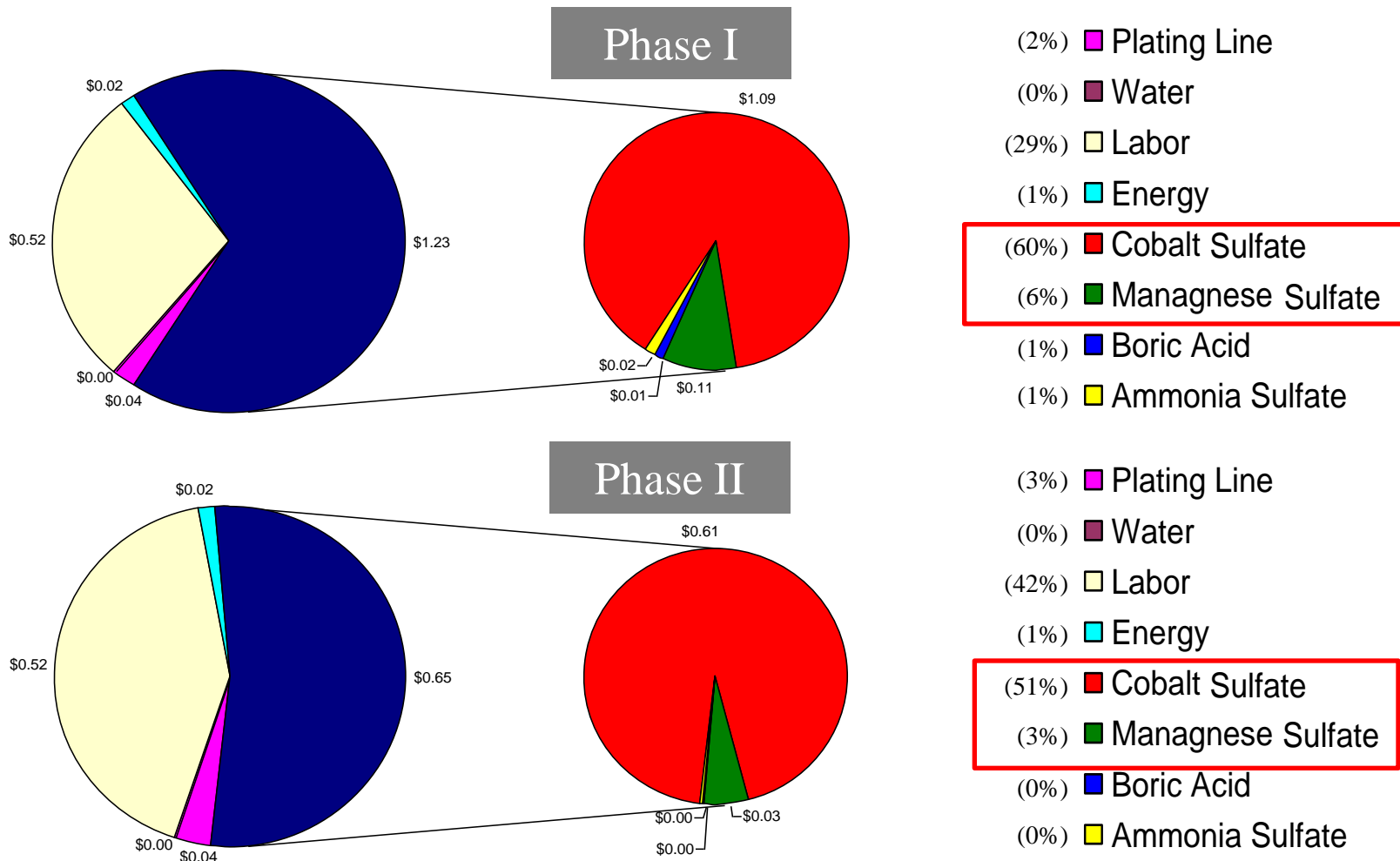
# *Initial Pilot Scale Experiments*

**FARADAYIC Electrodeposition parameters varied for experiments conducted in the newly modified FARADAYIC Electrodeposition Cell**

<b>Trial No</b>	<b>Electrode Spacing (inches)</b>	<b>Electrolyte Flow (PSI)</b>	<b>Vibration</b>	<b>Oscillation</b>
2001	3	16	-	-
2002	1.5	16	-	-
2003	4.5	16	-	-
2004	4.5	10	50	50
2005	4.5	16	50	50
2006	3	16	50	50
2008	3	5	50	50
2009	3	10	50	50



# Electrodeposition Cost Analysis



High volume manufacturing of 1,600,000 plates per annum at a cost of  
~\$1.23 per 25 x 25 cm interconnect

# *Future Direction*

- Determine plating parameters effect on chromium and oxygen diffusion
- Continue scale-up development for large area planar T441 substrates
- Begin scale-up development for large area pattern interconnects
  - Demonstrate coating uniformity and composition
- Testing in single cell and short stack SOFC systems

# *Acknowledgments*

- Briggs White and the entire SECA program management team
- This material is based upon work supported by the Department of Energy under Award Nos. DE-SC0001023 and DE-FE0006165. Any opinions, findings, conclusions and recommendations expressed in this material are those of the authors and do not necessarily reflect the views of the DOE.
- Contact Information:  
Heather McCrabb  
Ph: 937-836-7749  
Email: [heathermccrabb@faradaytechnology.com](mailto:heathermccrabb@faradaytechnology.com)

**Thank You!**

