



NATIONAL ENERGY TECHNOLOGY LABORATORY



Carnegie Mellon



University of Pittsburgh

VirginiaTech



West Virginia University

URS

SOFC Cathode Infiltration: Performance Enhancements and Processes of Degradation

Kirk Gerdes

U.S. Dept of Energy

National Energy Technology Laboratory

Research Group Leader – Fuel Cells



U.S. DEPARTMENT OF
ENERGY

Outline

- **NETL RUA Fuel Cell Team**
- **Infiltration methods**
 - Performance metrics
 - Technologies
- **Degradation**
 - Techniques
 - Future approach
- **Summary**



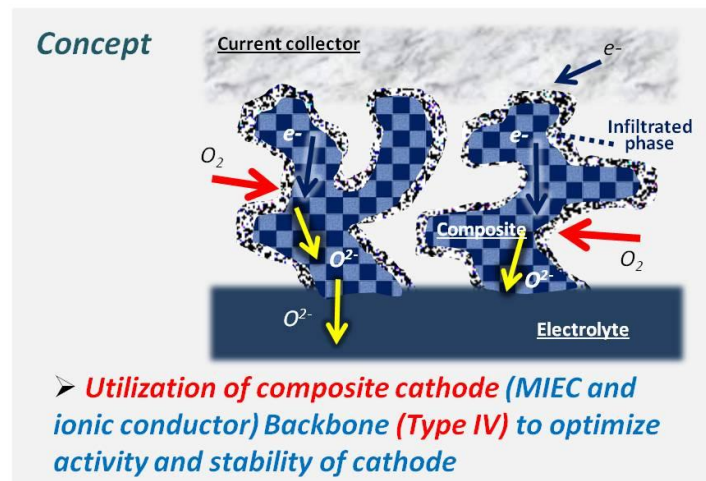
NETL RUA - Solid Oxide Fuel Cells

Support Industrial Development



Operation of NETL Solid Oxide Fuel Cell Multi-Cell Array on direct, coal-derived synthesis gas at the National Carbon Capture Center at Wilsonville, AL in August/Sept 2009.

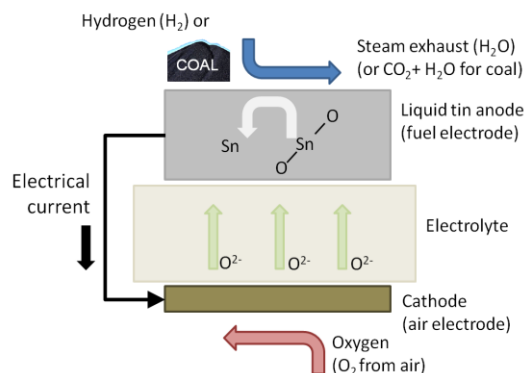
Collected 4,000 + cell-hours of data to support development of gas cleanup systems sufficient for gasifier / fuel cell integration.



Cathode infiltration technology is being developed to enhance the SOFC operating performance. Initial results have demonstrated > 50% performance improvement and acceptable material stability.

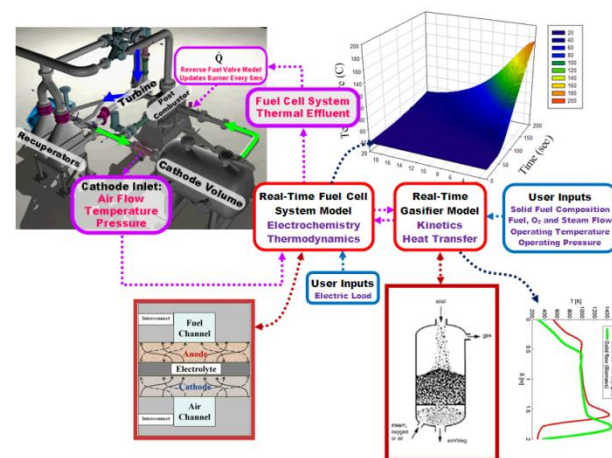
Innovate Technology

Evaluate Advanced Concepts



Liquid tin anode SOFC (at left) is an advanced direct coal to electricity concept. NETL evaluates critical transport and thermodynamic data.

Integrated gasifier / fuel cell / turbine systems (IGFT, at right) support advanced fuel cell demonstrations efforts (2013+). NETL operates a system hardware evaluation and controls development platform.



Cathode Engineering: Metrics of performance

- **Materials/processing cost**
 - \$175/kW stack cost, cathode estimated at 5-10%;
- **Cathode performance improvement**
 - Overpotential $\sim 80\text{mV}$ at 500 mA/cm^2 , 425 mW/cm^2 at 0.85V ;
- **Durability**
 - 0.1-1.0% loss in voltage or power / 1000 hours for full cell; and
- **Thermo-mechanical/chemical stability**
 - verified through relevant analytical tests (XRD, SEM/ESD, TEM).
- Initial performance testing completed under common conditions with standard commercial cell
 - MSRI anode supported LSM or LSCF cathode
 - Operated at 800°C (LSM) or 750° (LSCF) in hydrogen / air
 - Operated at 0.500 A/cm^2 (typically results in 0.90 V operating potential)

Cathode Infiltration (General)

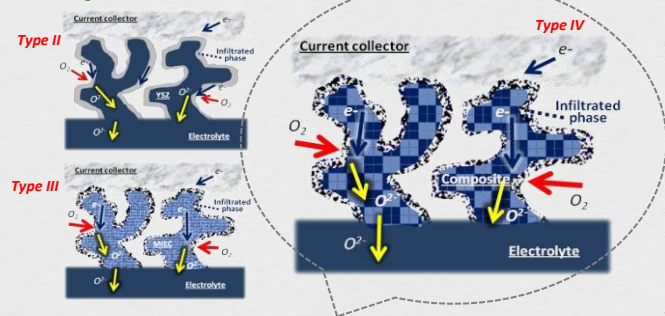
- **Application of discrete active materials into the cathode pore to enhance the electrochemical activity**
 - Electrocatalytic enhancement
 - Surface area enhancement
- **Diverse infiltration methods reported**
 - Colloidal suspension infiltration
 - Aqueous solution infiltration
 - Citrate, glycine, and Pechini methods
 - Modified versions of these methods
 - Control pH, viscosity, hydrophilicity, pre-cursor solubility
- **More combinations of materials, test conditions, and microstructures than can be reasonably tested**
 - Functional processes exist, but innovation is suppressed

MIEC Cathode Infiltration Project

Lee/Manivannan (NETL)

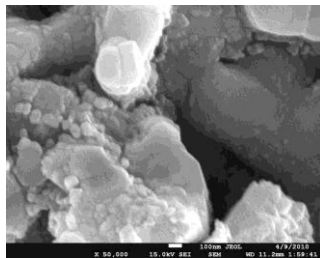
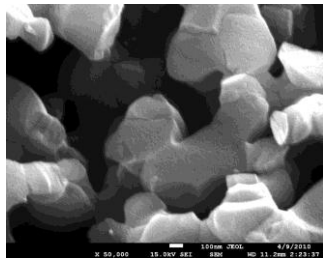
High Performance SOFC with **active** and **stable cathode** by application of controlled **infiltration** process

Concept



➤ Utilization of composite cathode (MIEC and ionic conductor) backbone (Type IV) to optimize activity and stability of cathode.

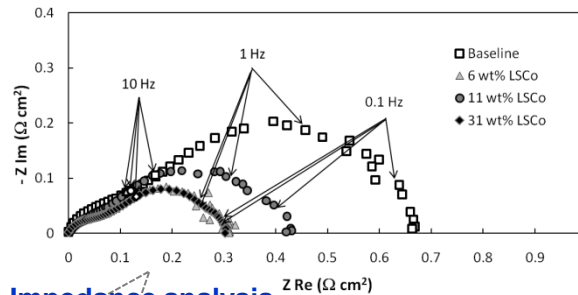
Infiltration Microstructure



Baseline composite cathode (SDC+LSCF)

LSCo Infiltrated Cathode (LSCo: $La_{0.6}Sr_{0.4}CoO_3$)

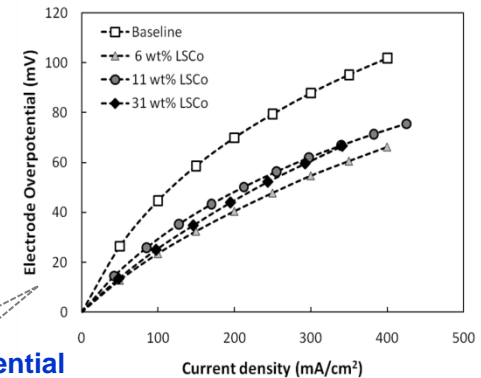
Infiltration Performance



Impedance analysis

Polarization resistance decreased by **38-60%**

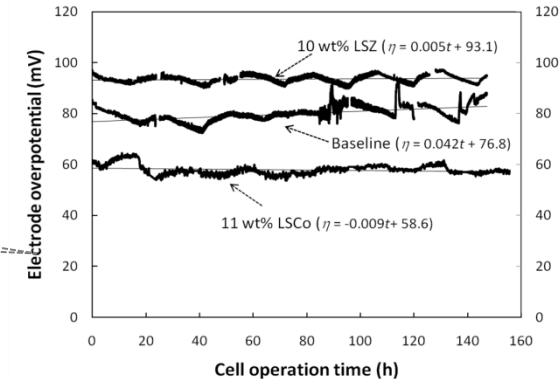
Electrode overpotential



Overpotential reduced to **58%** of that of baseline cell

No degradation in cell voltage for **>150 hrs**

Stability

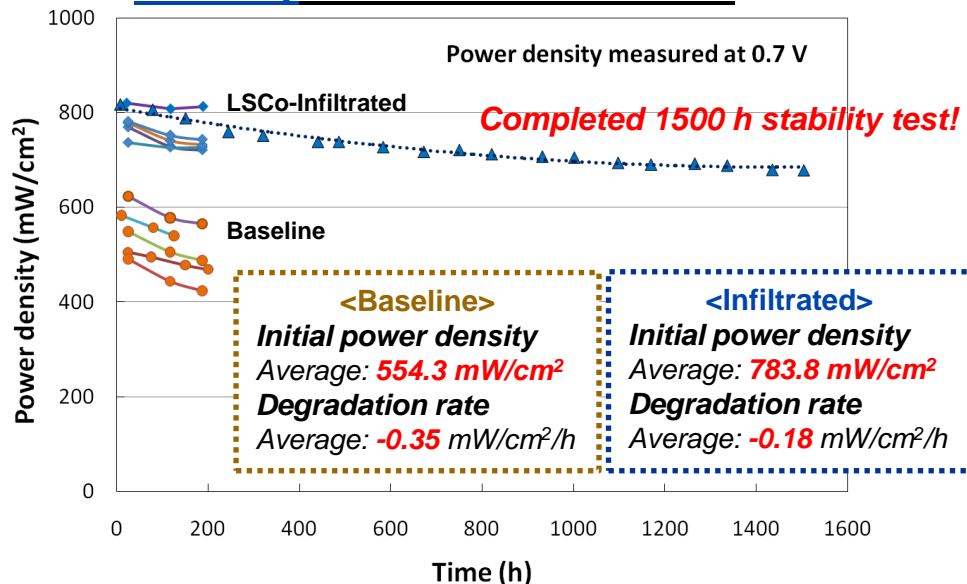


- ✓ A composite cathode of a commercial cell was successfully modified with LSCo infiltration.
- ✓ Infiltration methodology correlating dosage to structure was developed.
- ✓ Effects of electrocatalytic activity of infiltrated materials on cathode performance were demonstrated.

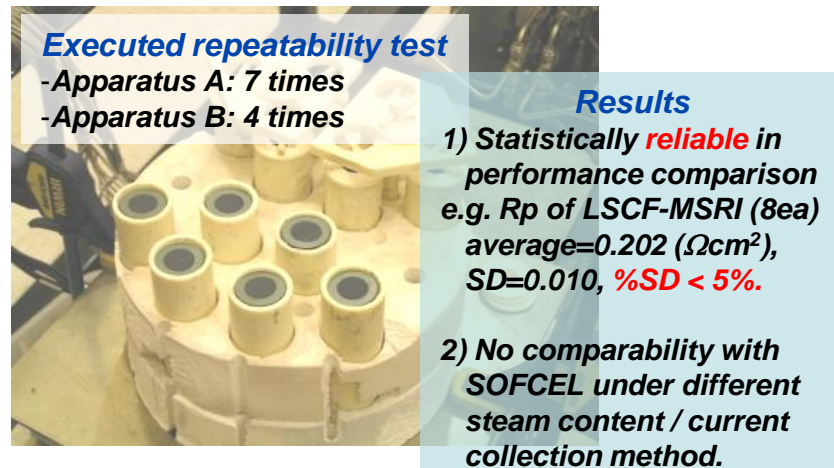
Electrocatalytic Cathode Infiltration

Lee/Manivannan (NETL)

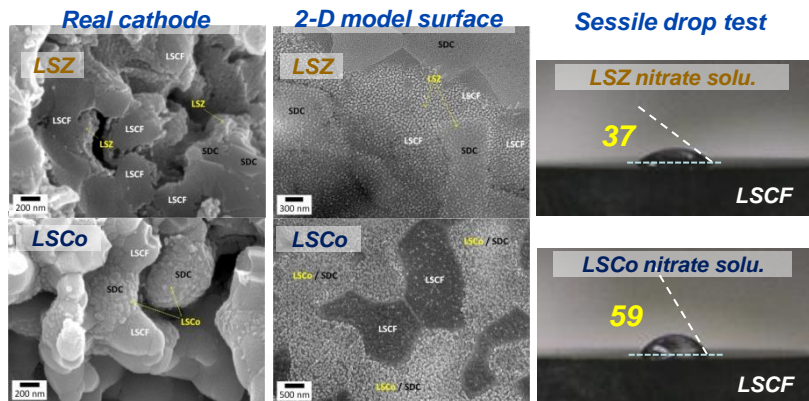
Stability of infiltrated cathode



Repeatability test of MCA (Multi-cell array)



Cathode microstructure control: Wettability



Approach

- 1) Set up model experiments.
- 2) Variables: surface tension of solution & hydrophobicity of backbone.

Publications

1. R. Chao, J. R. Kitchin, K. Gerdes, E. M. Sabolsky, and P. A. Salvador, "Preparation of Mesoporous $\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$ Infiltrated Coatings in Porous SOFC Cathodes Using Evaporation-Induced Self-Assembly Methods" ECS Transactions, 35 (1) 2387-2399 (2011)
2. Shiwoo Lee, Nicholas Miller, Harry Abernathy, Kirk Gerdes, and Ayyakkannu Manivannan "Effect of Sr-doped LaCoO_3 and LaZrO_3 Infiltration on the Performance of SDC-LSCF Cathode." J. Electrochem. Soc., Volume 158, Issue 6, pp. B735-B742 (2011)
3. Shiwoo Lee, Nicholas Miller and A. Manivannan "Microstructural Control of Composite Cathode by Wetting Nature of Infiltrated Solution." ECS Trans., 35 (1) 2401-2407 (2011)

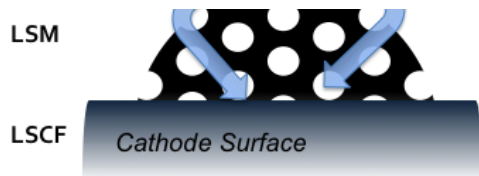
Presentations

1. Shiwoo Lee, Nicholas Miller, Harry Abernathy, Kirk Gerdes, and Ayyakkannu Manivannan "Modification of SDC-LSCF cathode functional layer by solution infiltration." CACC-S3-030-2011, 35th ICACC, Daytona beach (2011)

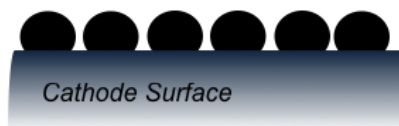
Mesoporous Electrocatalysts Coating

Salvador / Chao (CMU)

**Mesoporous Infiltration:
Inverse Nanoparticles**



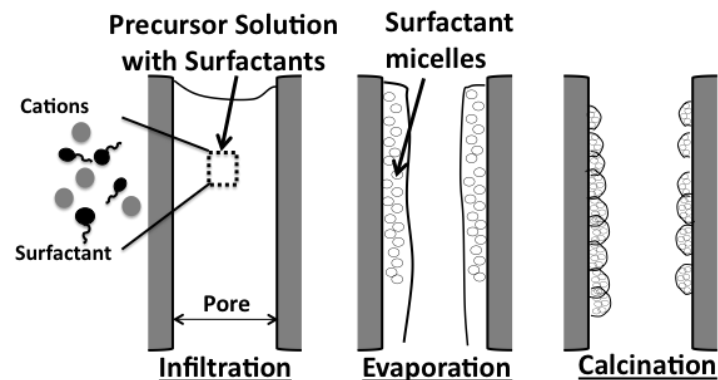
**Typical Infiltration:
Nanoparticles**



Improvements for Mesoporous Infiltrations

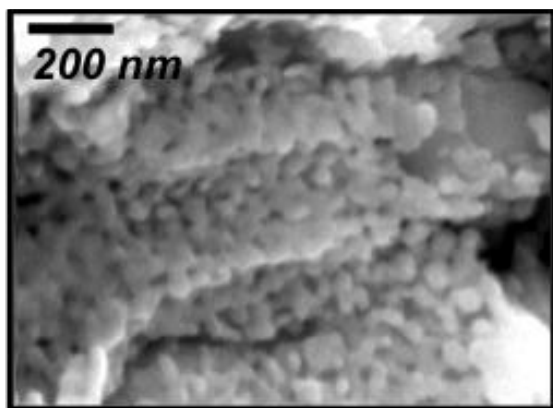
1. Higher surface areas for O_2 adsorption
2. Contiguous solid and pore networks of infiltrates
3. Thermal Stability Demonstrated (wall thickness /pore diameters)
4. Surfactant controls the microstructure
5. Large catalyst loading possible and demonstrated

Evaporation-induced Self-assembly

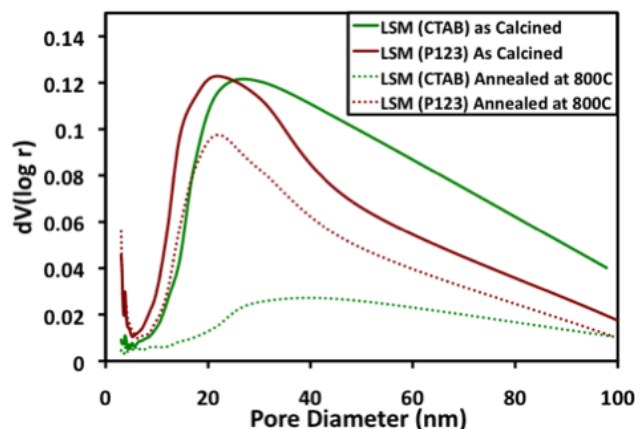


High catalyst loading (10 wt%) can be achieved in **single** infiltration step

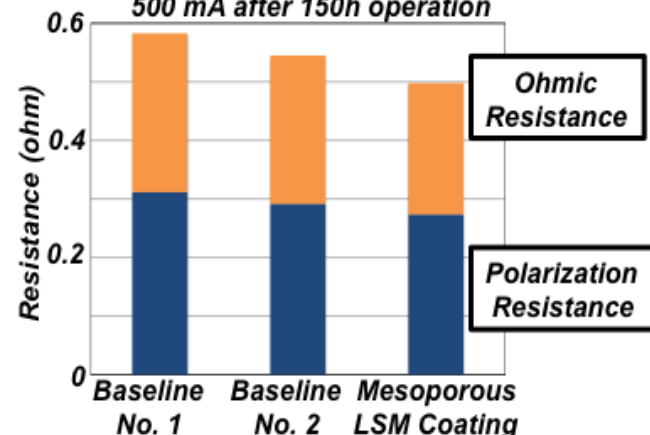
Inverse LSM Nanoparticles
2 mol/L Infiltrated with CTAB in 1 step



Stability of Mesoporous Powders
P123 has thicker pore walls



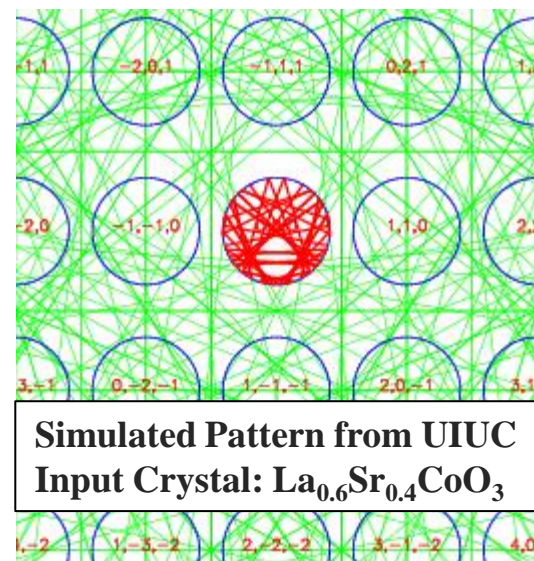
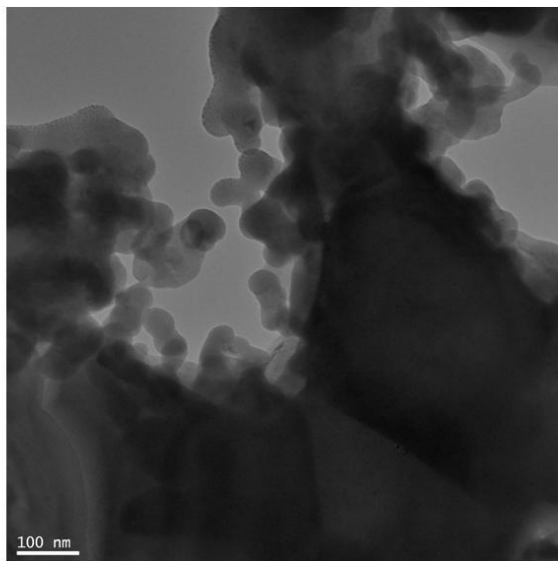
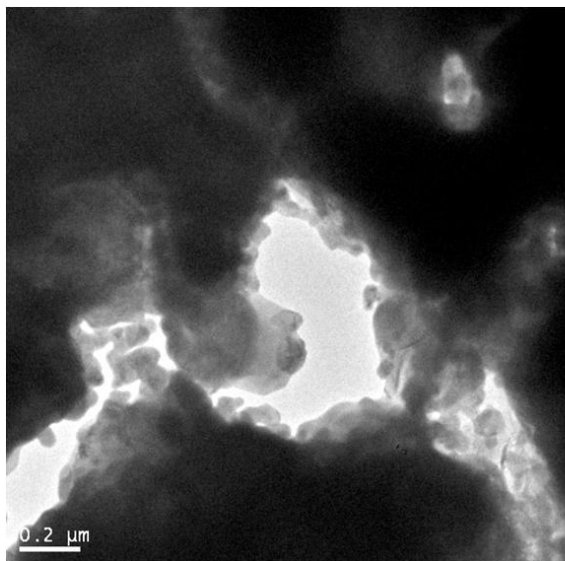
12 % Improvement in Resistance with Mesoporous Infiltration
500 mA after 150h operation



Coating morphology is currently being optimized to enhance the infiltrated cathode performance

Mesoporous Electrocatalysts Coating

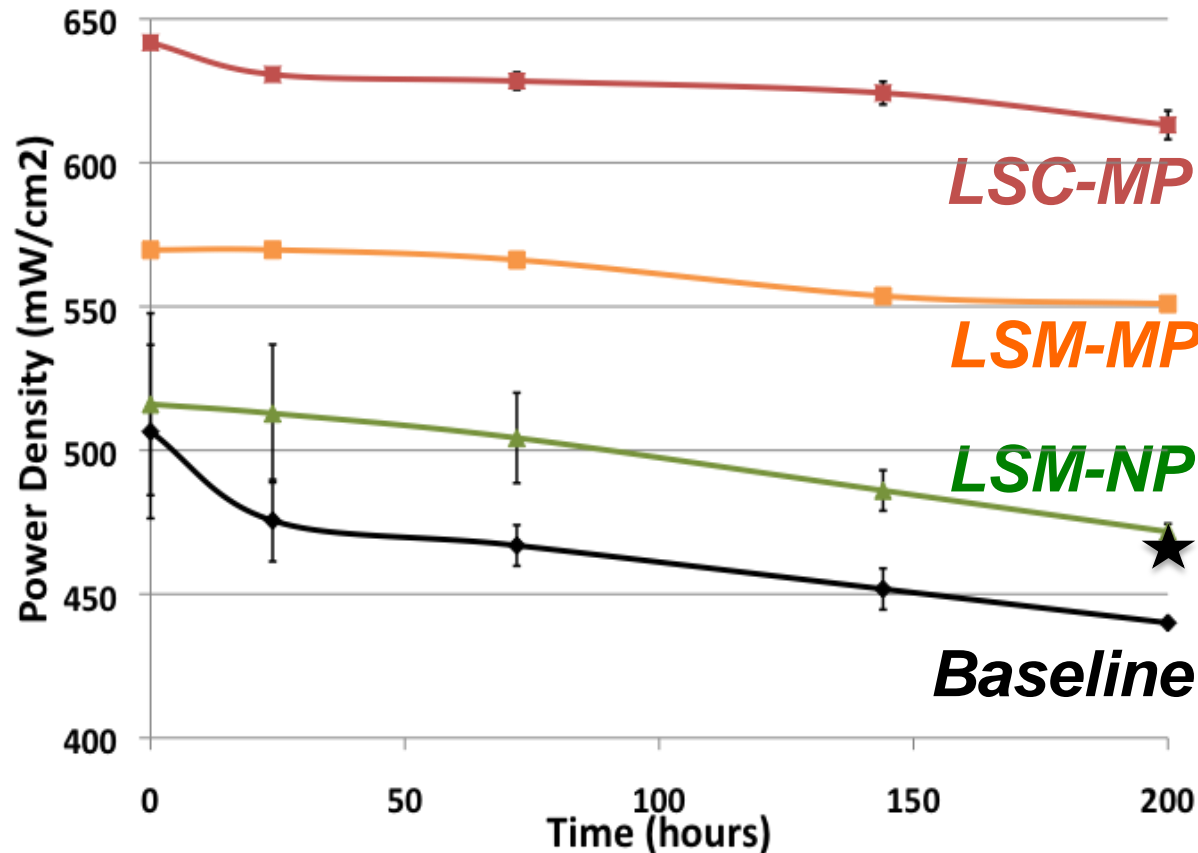
Salvador / Chao (CMU)



- The coatings were uniformly applied to the functional layer pore walls
- The infiltrated coatings have nanostructured porous structure with pore size about 30-50 nm (measured using ImageJ)
- Consistent with cubic perovskite with lattice parameter of 3.854

Average Power Density vs. Time

Salvador/Chao (CMU/NETL)



*Averaged from
2 cells tested*

★ : Baseline avg. from
10 cells: 464 mW/cm²

*Overall Degradation
Rate (mW/hr)*

Baseline #1	0.41
Baseline #2	0.16
LSM-MP-2M #1	0.11
LSM-MP-2M #2	0.12
LSC-MP-2M #1	0.12
LSC-MP-2M #2	0.11
LSM-NP #1	0.39
LSM-NP #2	0.06

*Power Density Improvement
(compared to the baseline)*

At 0 hour At 200 hour

LSM-NP	2%	7%
LSM-MP-2M	12%	25%
LSC-MP-2M	27%	39%

- All infiltrated cells showed **improved performance**
- Mesoporous LSM coatings generate **higher power density** than nanoparticle LSM infiltrates
- Cells infiltrated with LSC showed higher power density than those with LSM coatings
- Mesoporous coating infiltrated cells showed **consistent degradation rate reduction**

Possible Causes of **Infiltrated Cathode** Degradation with Time

Lee (NETL)

1. Particle Coarsening (sintering)

- Loss of active electrochemical reaction area

2. Decomposition (Phase segregation) of infiltrated material

- e.g. Sr exsolution leading to reduced acceptor dopant concentration

3. Interdiffusion at Interfaces

- Chemical reaction with electrolyte (Perovskite-YSZ)
- Ni diffusion cause LSM coarsening and densification

4. Delamination (Spallation)

- Reduction in contact area

5. Formation of resistive phases

- Gaseous impurities

Infiltrated MSRI cathode (Shiwoo Lee, NETL)



Possible Causes of **Infiltrated Cathode** Degradation with Time

Lee (NETL)

1. Particle Coarsening (sintering)

- Loss of active electrochemical reaction area

2. Decomposition (Phase segregation) of infiltrated material

- e.g. Sr exsolution leading to reduced acceptor dopant concentration

3. Interdiffusion at Interfaces

- Chemical reaction with electrolyte (Perovskite-YSZ)
- Ni diffusion cause LSM coarsening and densification

4. Delamination (Spallation)

- Reduction in contact area

5. Formation of resistive phases

- Gaseous impurities

Infiltrated MSRI cathode (Shiwoo Lee, NETL)



Approach to Degradation Modeling

- 1. Characterize the initial microstructure (backbone plus infiltrate)**
- 2. Describe the microstructural evolution**
- 3. Inform the evolution model's kinetic and thermodynamic relationships with targeted experiments**
- 4. Compare computational predictions to final microstructure**

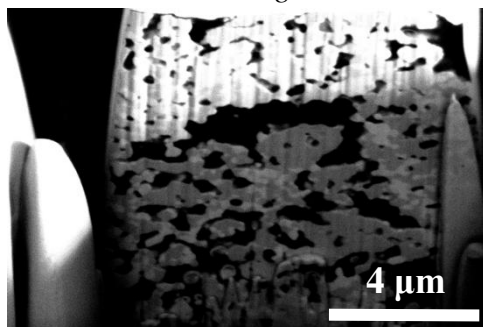
3D microstructural reconstruction

Salvador / Miller / Ohodnicki (CMU/NETL)

Cathode Reconstruction:

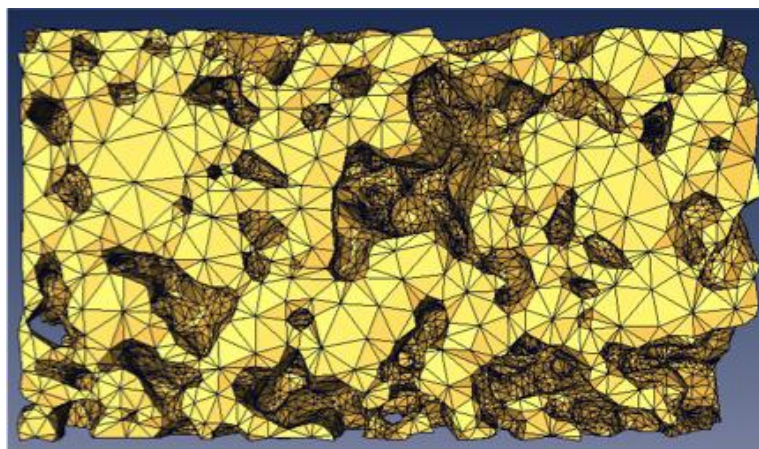
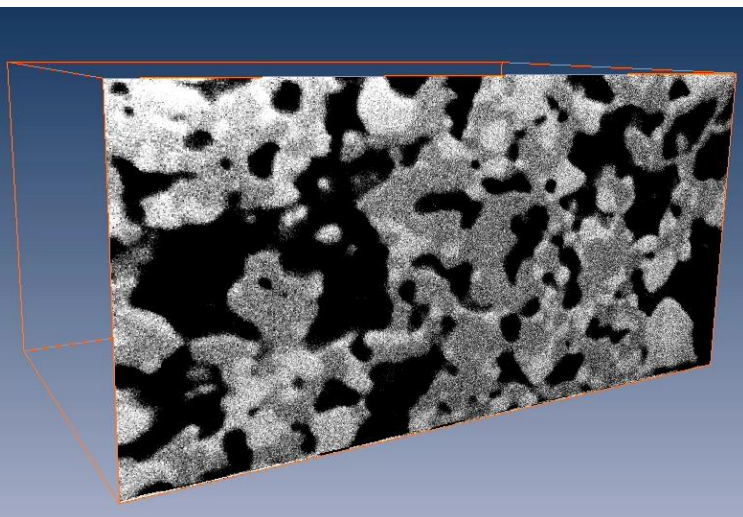
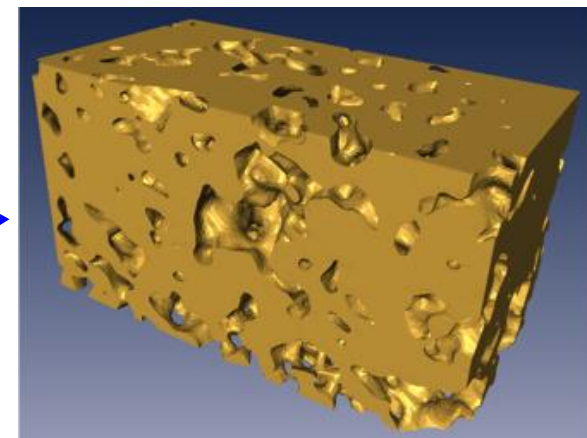
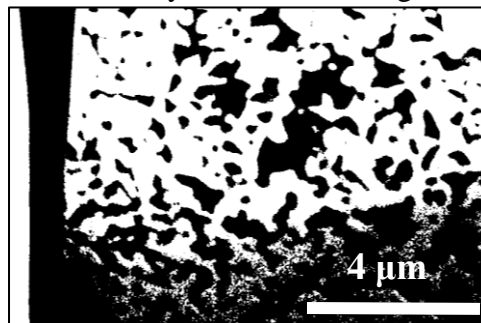
FIB-SEM: TLD

Active Region



Current Collector Region:

binary thresholded image



**Reconstructed
volume = $54 \mu\text{m}^3$**
 $5.65 \mu\text{m}^3 \times 3.15$
 $\mu\text{m}^3 \times 3 \mu\text{m}^3$ (60-
50 nm slices)

- pores = 29 % volume
- pore/LSM surface area per volume = 4.57 l^{-1}

• Continuing to explore 3d x-ray tomography technique – promising first dataset; calibration problems with 3D reconstruction

Reconstruct 3D sections of SOFCS ---- Generate statistical metrics --- input to models --- correlate to performance

Phase Field Method

Chen / Li (PSU)

Description of Project

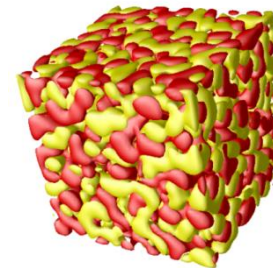
1. Develop a phase-field model for temporally evolving three-phase SOFC cathode microstructures.
2. Couple gas, electronic, and ionic transport equations with phase-field model of microstructure evolution to study the effect of local electrochemical conditions on microstructure degradation.
3. Compare simulation results with experimental microstructures obtained under different thermal/electric conditions such as temperature and average overpotential for validation

A preliminary Phase Field model of 3D 3-phase cathode microstructure is developed

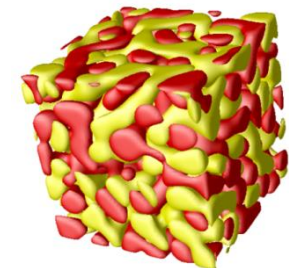
$C(r) = C_\beta \quad \eta_i = 1$ $C'(r) = C_\alpha \quad \eta_i' = 0$ YSZ	$C(r) = C_\alpha \quad \eta_i = 0$ $C'(r) = C_\beta \quad \eta_i' = 1$ LSM
Pore $C(r) = C'(r) = C_\alpha \quad \eta_i = \eta_i' = 0$	

Two concentration fields C and C' are used to define compositions of LSM and YSZ. The temporal evolution of the field variables is described by Cahn-Hilliard Equations

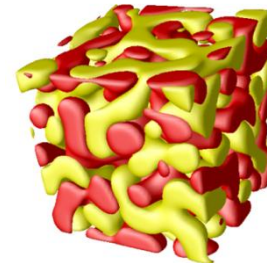
$$\frac{\partial C_i(\mathbf{r}, t)}{\partial t} = \nabla \left(M_i \nabla \frac{\partial F}{\partial C_i(\mathbf{r}, t)} \right)$$



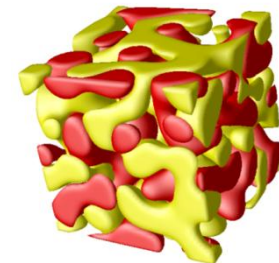
Time step:1000



Time step:5000



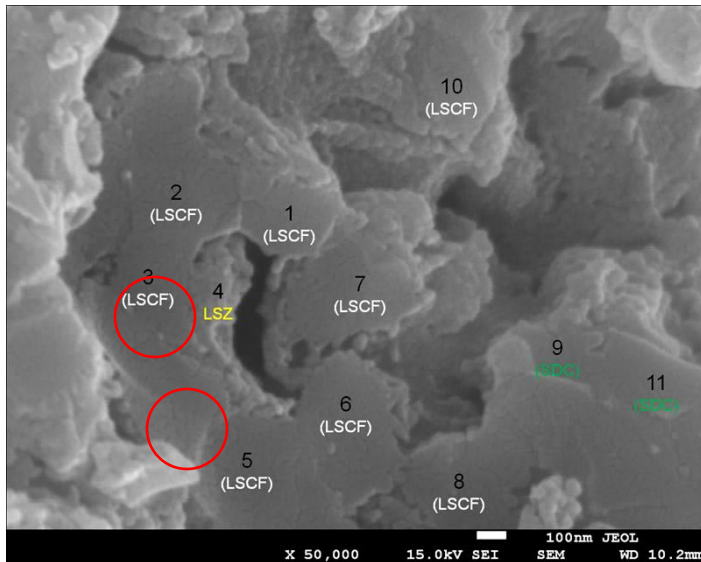
Time step:10000



Time step:20000

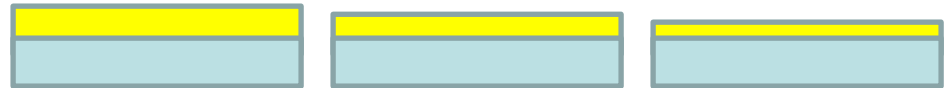
Creep Microbeam Bending Messing/Kupp/H.Lee (PSU)

- **Creep microbeam bending measurement**
 - Model stress evolution of infiltrated system during processing and operation
 - Multilayer structures used to approximate thickness and density variations of the infiltrated system
 - Data used to calculate stress evolution in the infiltrated system

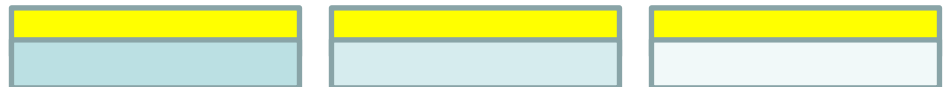


LSCF/SDC cathode infiltrated with LSZ material (Shiwoo Lee, NETL)

→ *Thickness dependence*



→ *Density dependence*



Summary

- **Cathode infiltration is possible and performance has been demonstrated for single chemistry**
- **Degradation issues have been initially investigated but more thorough characterization (1+ khr) is required**
- **Degradation can be simulated through a computational framework supported with direct experiments**
- **Computational/modeling framework has been formulated and initial results generated for each framework component**

Acknowledgements

- **NETL RUA**
 - Shiwoo Lee, A. Manivannan (NETL)
 - Paul Salvador, Robin Chao (CMU)
 - Gary Messing, Libby Kupp, Haijoon Lee (PSU)
 - LongQing Chen, Qun Li, Liangjun Li (PSU)
 - Nicholas Miller (URS)