DEVELOPMENT OF CERAMIC INTERCONNECT MATERIALS

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NATIONAL LABORATORY

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MOTIVATION

- Challenges of Acceptor-doped Lanthanum Chromite
- Inferior Sintering Behavior
- •Reactivity with YSZ Electrolyte
- (Formation of Lanthanum Zirconate)

<u>GOALS</u>

- **Develop Ceramic Interconnect Materials with**
- Chemical & Dimensional Stability
- •High Electronic & Low Ionic Conductivity
- Improved Sintering Behavior
- •Thermal Expansion Match
- Chemical Compatibility with Other Components







• Chemically Compatible with YSZ, NiO, and LSM at Processing Temperatures.

Through Doping <u>Yttrium Chromite</u> with calcium on A-site and Transition Metals on B-site

CHEMICAL STABILITY



 Single Phase Orthorhombic Perovskite Structure (25°C<T<1100°C, 10⁻²⁰atm<p₀₂<1atm)



stability with Ni-Doping

Reducing Environments

 Co- and Cu-doping has negligible effect on the chemical expansion and the amount of oxygen loss.

ELECTRICAL CONDUCTIVITY







FILM DENSIFICATION



SUMMARY

Calcium- and Transition Metal-doped Yttrium Chromite •Glycine-Nitrate Process

Single Phase Orthorhombic Perovskite Structure Between 25 and 1100°C Over Wide pO₂ Range
Cu-doping significantly improves sinterability.
TEC can be controlled through B-site doping.
Conductivity is improved by Co- and Ni-doping.
Ni-doping improves stability toward reduction.
Oxygen ionic leakage current is sufficiently low.

- 0
 200
 400
 600
 800
 1000

 Temperature (degree C)
 Temperature (degree C)
 Temperature (degree C)
- Dilatometry Measurements
- Co- and Ni-doping improves sinterability.
- Small addition of Cu (~1%) remarkably enhances sinterability.





Chemically compatible with YSZ, NiO, and LSM.

•Full densification can be achieved by addition of infiltration process.

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