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## University of Nevada, Las Vegas Department of Chemistry

information on

•effect of heat

and sputter

treatment to

composition of

•valence level

information

surface vs. "bulk"

(compare to STS)

surface

•chemical

surface

structure

## Objectives

## MIT

(local surface structure) (sample growth)

CMU

•samples •information about growth conditions

•feedback on sample quality •identification of contamination sources

**UNLV** 

(electronic structure, surface chemical composition)

•find a surface that allows study of occupied and unoccupied states with respect to a common reference (Fermi energy, vacuum level) •investigate effect of temperature and  $pO_2$  on surface chemical composition and electronic structure correlation with theory

## **Experimental Methods**

- (i) XPS (X-ray Photoelectron Spectroscopy)
- (ii) UPS (UV Photoelectron Spectroscopy)
- (iii) AES (Auger Electron Spectroscopy)
- (iv) XES\* (X-ray Emission Spectroscopy)
- (v) IPES (Inverse Photoemission Spectroscopy)
- (vi) XAS\* (X-ray Absorption Spectroscopy) performed at the Advanced Light Source Lawrence Berkeley National Laboratory





<u>(а</u>



## Spectroscopic investigation of the impact of film thickness and postgrowth treatment on the electronic structure of $La_xSr_{1-x}MnO_3$ films

### Ex-situ treatment



- Decrease of SrCO<sub>3</sub>-related C 1s/O 1s peak
- •Shift of about 0.2 eV for O 1s main peak and Sr 3p to higher E<sub>B</sub>
- •C 1s main peak shifts to lower E<sub>B</sub>
- $\rightarrow$  No complete reduction of C 1s Combined with mild sputtering



- •La/Sr ratio stays constant up to 500°C, then it increases by 3-4%. (corroborates earlier annealing experiment to 800°C in air, 20% increase)
- •Mn 2p and La 3d increase due to C 1s decrease and corresponding decrease in energy dependent signal attenuation due to carbon overlayer.



## Mn oxidation state





648 652 656 660 664 668

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Photon Energy (eV)



 $(1.25 \pm 0.2) \text{ eV}$ 

**IPES** 

- various oxidation states of Mn and comparison
- $\rightarrow$  measurements corroborate literature data
- $\rightarrow$  Mixture of Mn<sup>3+</sup> and Mn<sup>4+</sup>

S. Krause, T. Hofmann, and C. Heske, Department of Chemistry, University of Nevada, Las Vegas, NV L. Yan, H. Du, B. Kavaipatti, and P. Salvador, Carnegie Mellon University, Pittsburgh, PA H. Jalili, K. Katsiev, and B. Yildiz, Massachusetts Institute of Technology, Cambridge, MA

- First combined UPS and IPES spectrum
- •Smallest gap found:  $1.25 (\pm 0.20)$ eV
- •More measurements for different thicknesses, preparations and substrates needed



#### LSMO/YSZ (as-received) ----- 0° — 70° 534 532 530 528 Binding Energy (eV) Binding Energy (eV O 1s LSMO/NGO – normal emission – 70° off-normal (one cleaning cycle) 70° off-normal 280 275 270 Binding Energy (eV) 534 532 530 528 Binding Energy (eV normal emissior - 70° off-normal normal emissior – 70° off-normal LSMO/STO (as-received) • All surfaces terminated with a strontium oxide, except for NGO substrate with strontium carbonate Different geometrical surface structure might be responsible $\rightarrow$ LEED