

Proton Conductor Based Solid Oxide Fuel Cells

S. Elangovan, F. Zhao, J. Hartvigsen,
D. Ramirez, and D. Larsen

11th Annual SECA Workshop
July 27, 2010

Pittsburgh, PA

Supported by DOE SBIR Grant: DE-FG02-06ER84595



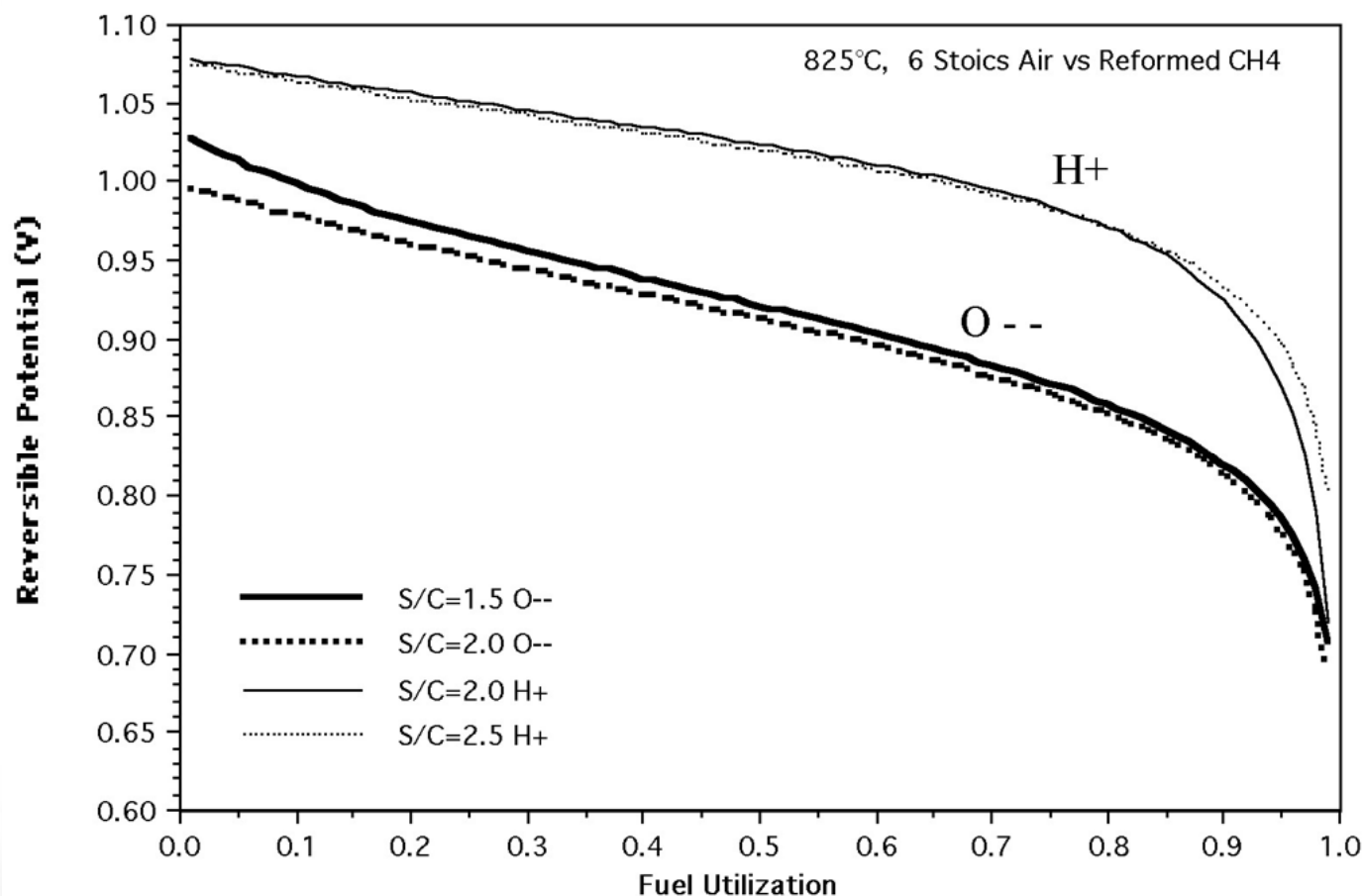
CERAMATEC
TOMORROW'S CERAMIC SYSTEMS

Outline

- Thermodynamic Analysis Shows Higher Efficiency for Proton Cells compared to Oxygen Cells
- Stability addressed by the use of composite electrolyte
- Anode supported composite electrolyte cell shows good performance
- Stability in high CO₂ containing fuel demonstrated



Driving Force Comparison



■ High driving force even at high fuel utilization



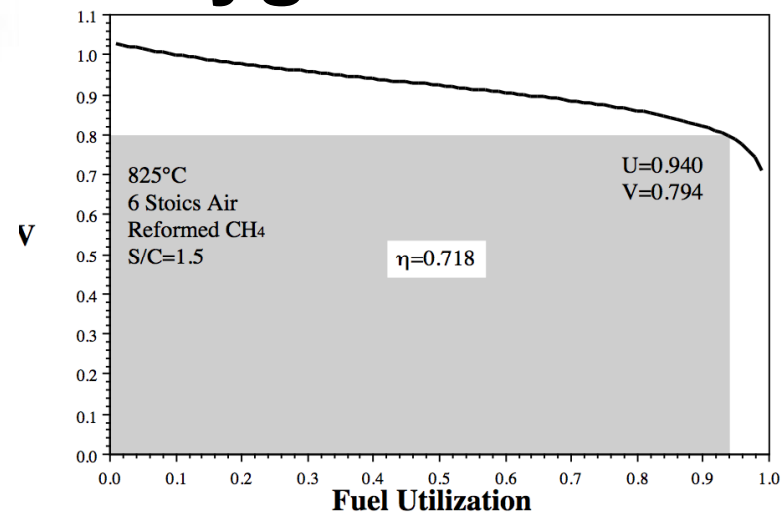
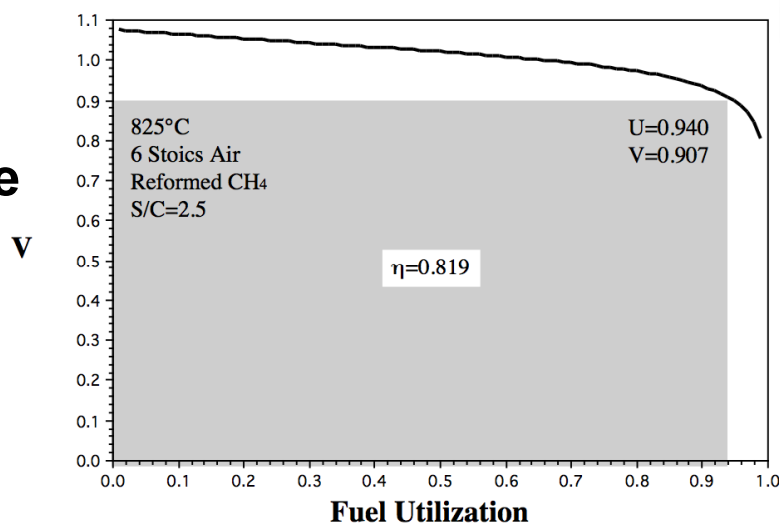
CERAMATEC®
TOMORROW'S CERAMIC SYSTEMS

Max. Efficiency Comparison

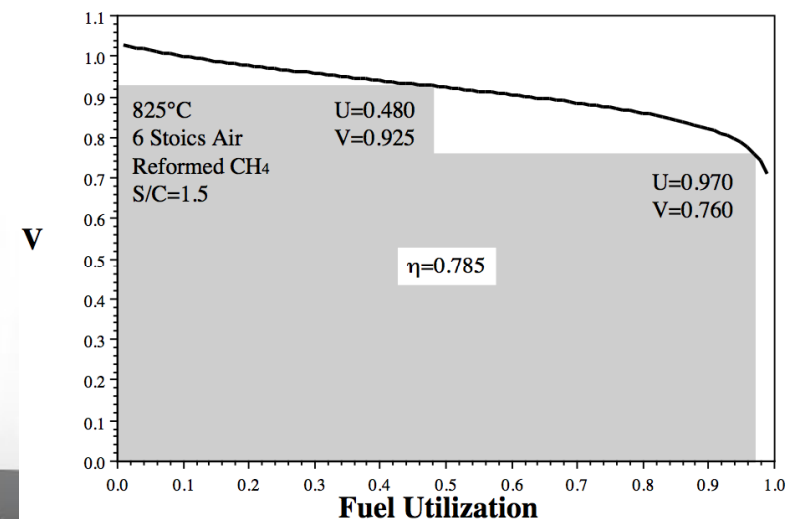
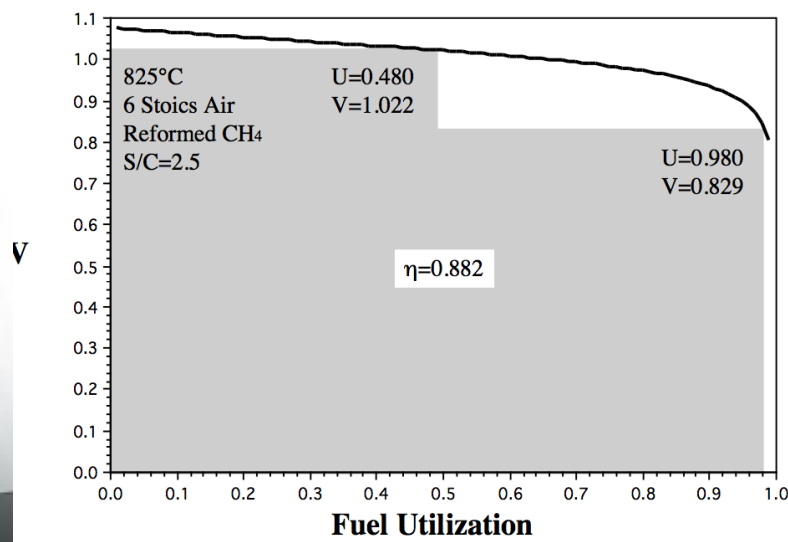
■ Proton Cell

■ Oxygen Cell

Single Stage

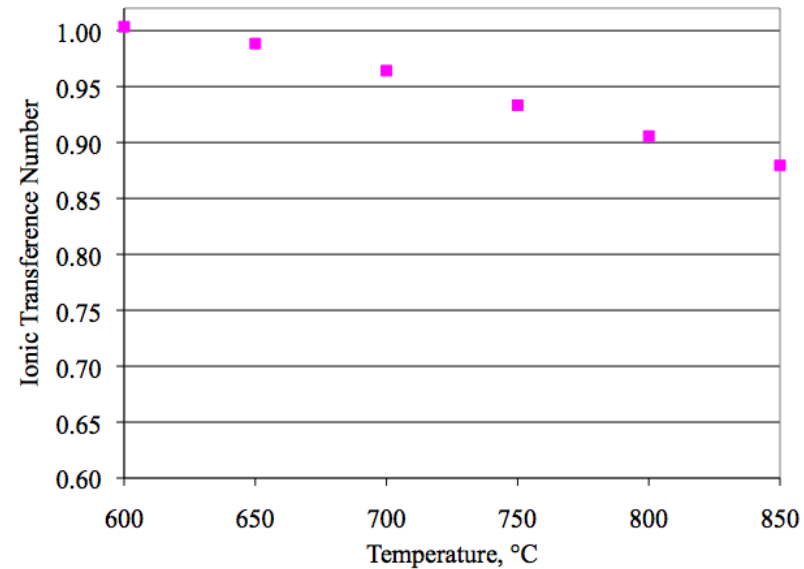
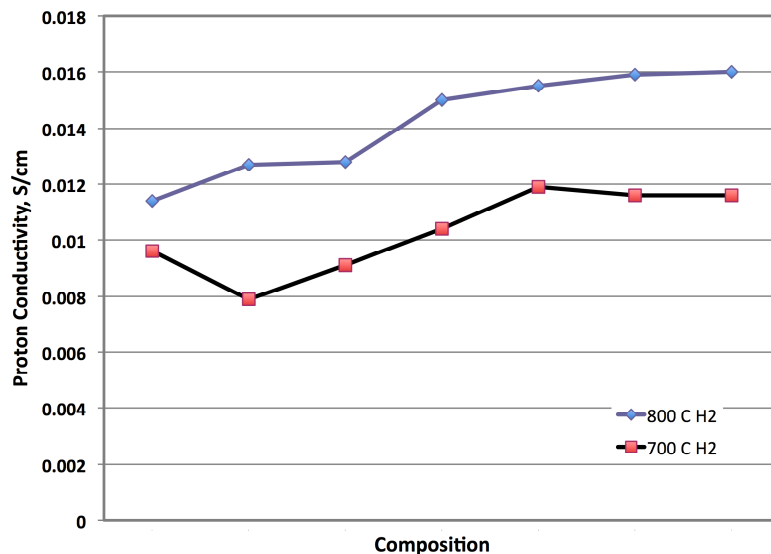


Two Stage

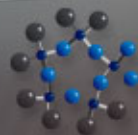


TOMORROW'S CERAMIC SYSTEMS

BaCeO₃ Proton Conductivity and Transference Number

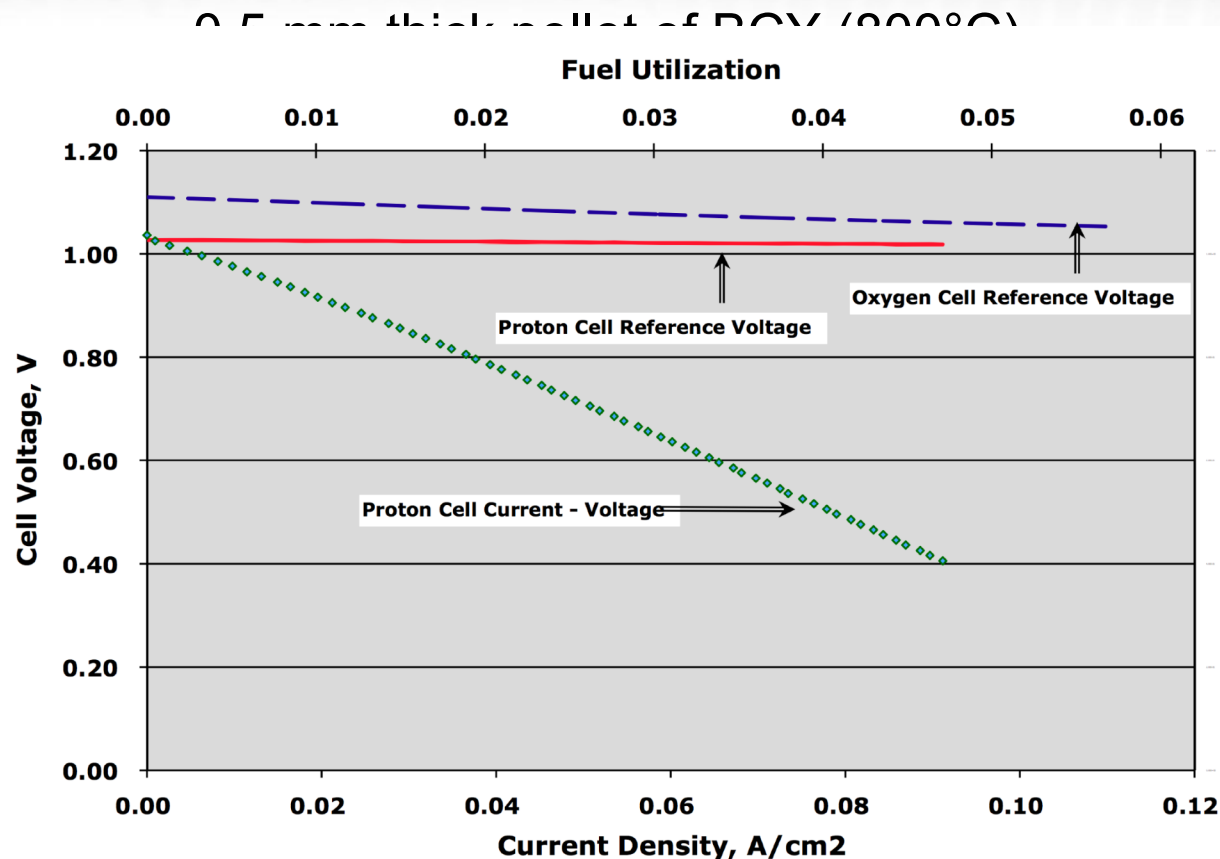


- Highest conductivity range from 0.01 to 0.016 in 700° to 800°C range
- ~ half the oxygen ion conductivity of 8YSZ
- Ionic transference number >0.95 at 700°C

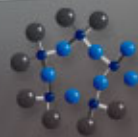


CERAMATEC
TOMORROW'S CERAMIC SYSTEMS

Comparison of Driving Force



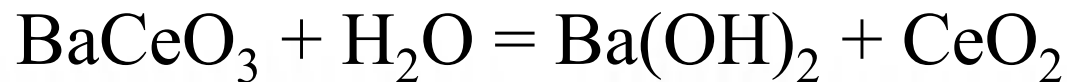
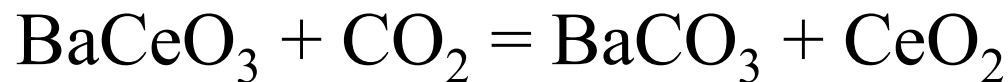
- Proton cell shows negligible change in driving potential compared to Oxygen cell
- Even with lower OCV, the Nernst potential crosses over at utilization of >10%



CERAMATEC
TOMORROW'S CERAMIC SYSTEMS

Instability of Perovskite

- Stability of BaCeO_3 in hydrocarbon based fuel is a major known issue



Composite of BCY + YDC for Improved Stability

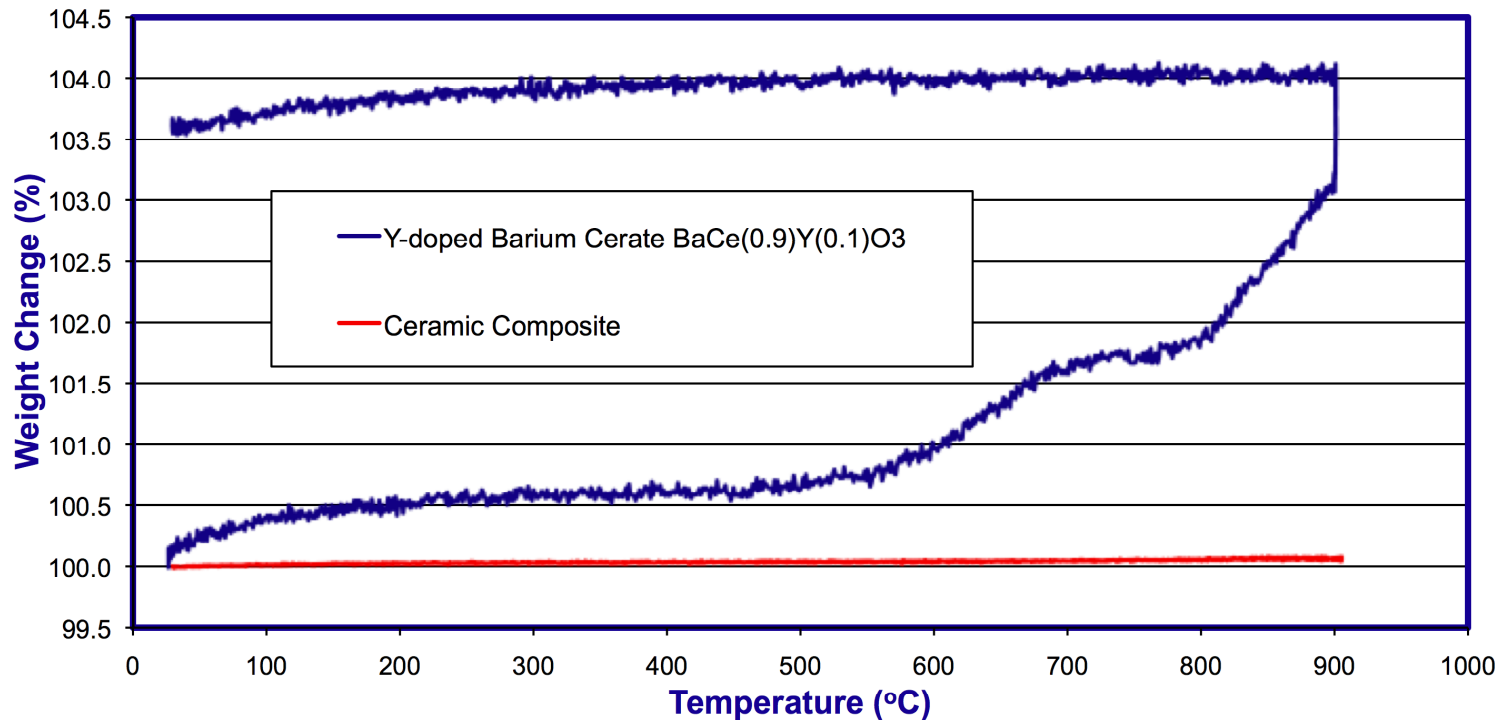


CERAMATEC
TOMORROW'S CERAMIC SYSTEMS

Enhanced Thermochemical Stability

Ceramic Composite over BCY

Thermogravimetric analysis in Air + 5% CO₂

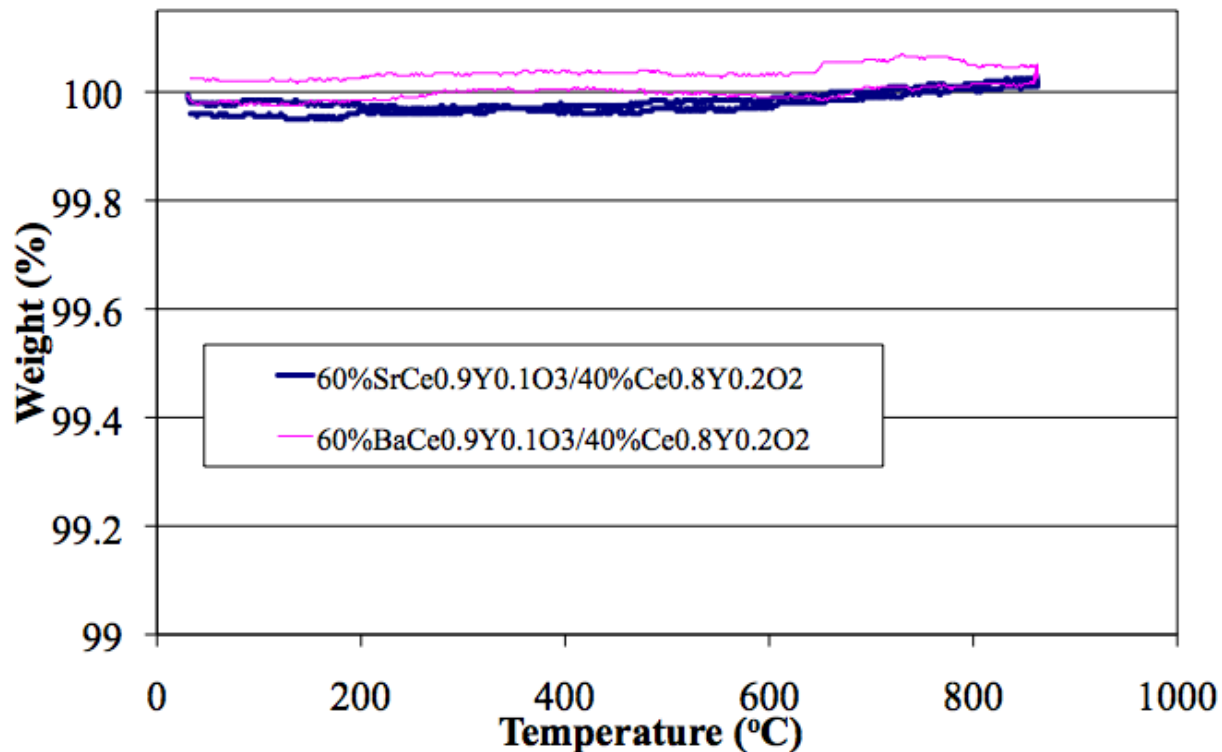


- Stability in CO₂+Air mixture (TGA)
 - BCY + YDC (crushed sintered disk)

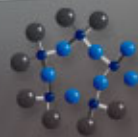


CERAMATEC®
TOMORROW'S CERAMIC SYSTEMS

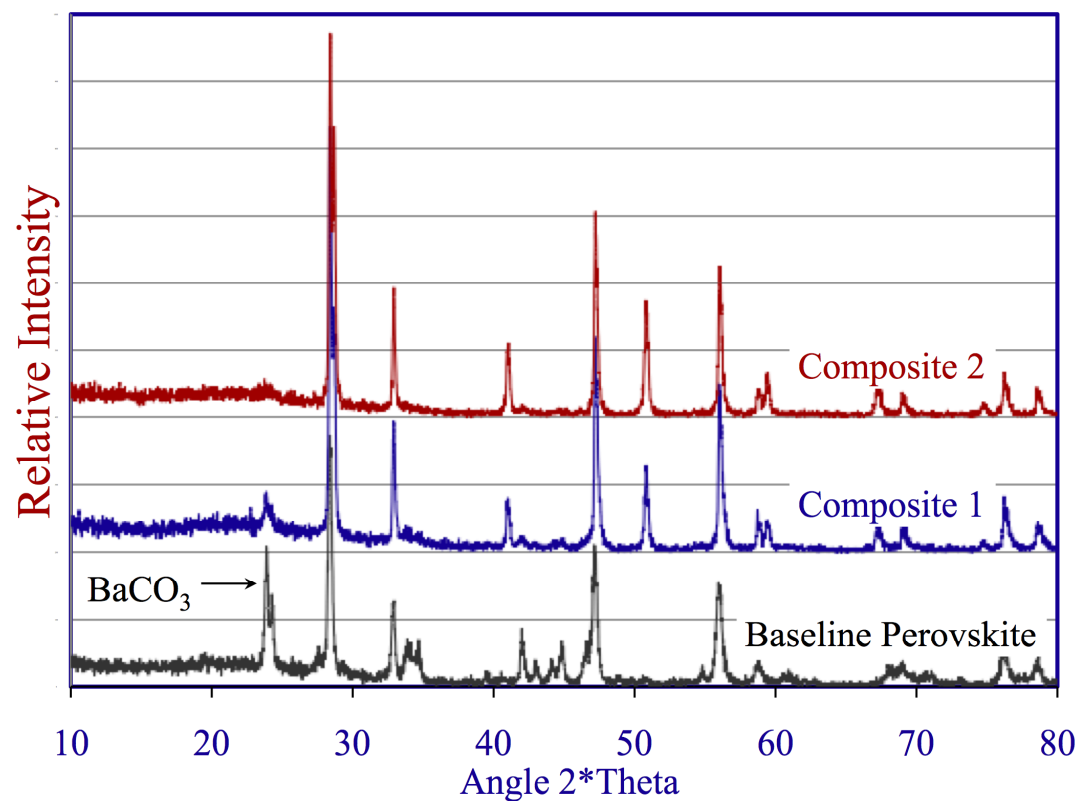
Composite Stability in Syngas



- Stability in CO-CO₂-H₂-H₂O mixture



BaCeO₃ vs Composite Stability

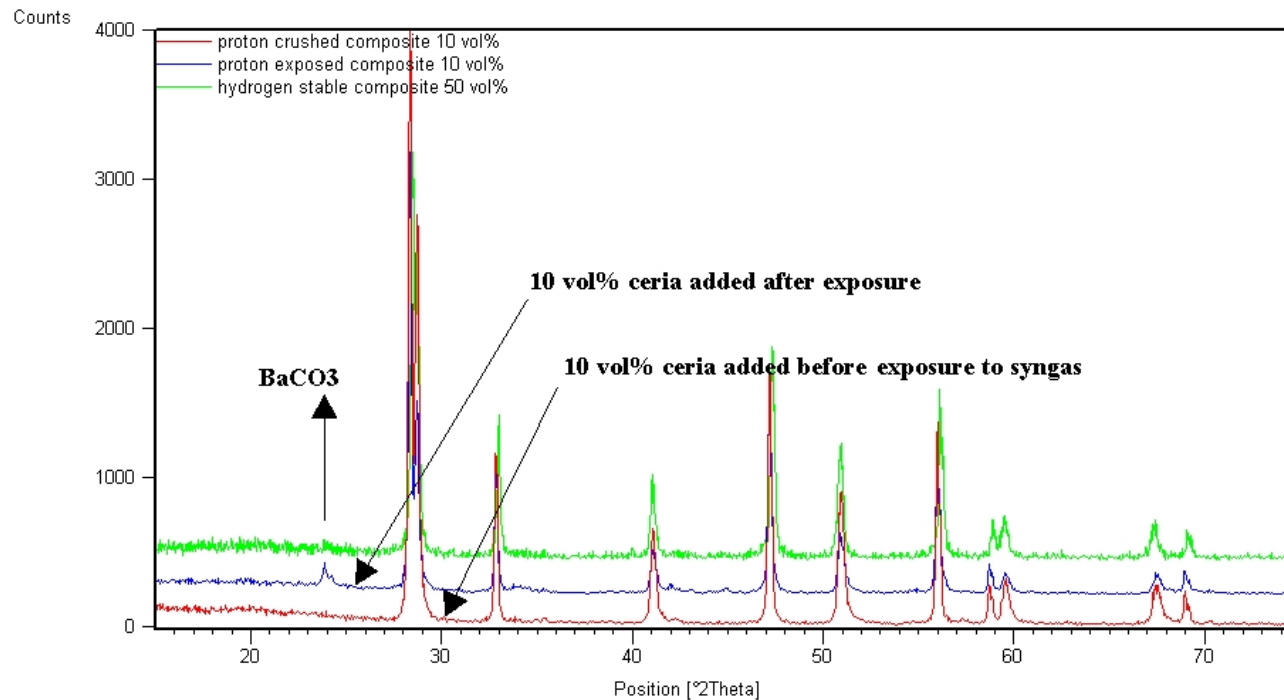


■ Exposure to syngas at 900°C

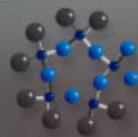


CERAMATEC
TOMORROW'S CERAMIC SYSTEMS

Exposure to Syngas at 700°C



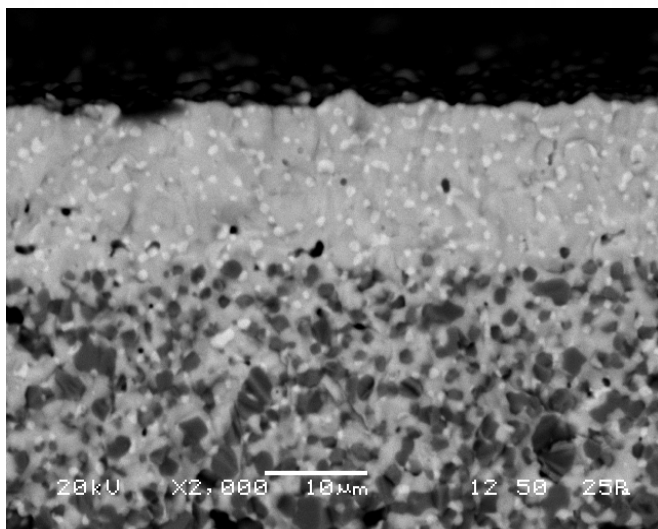
■ As low as 10 vol% Ceria shows improvement in stability



CERAMATEC
TOMORROW'S CERAMIC SYSTEMS

Anode supported thin film cell

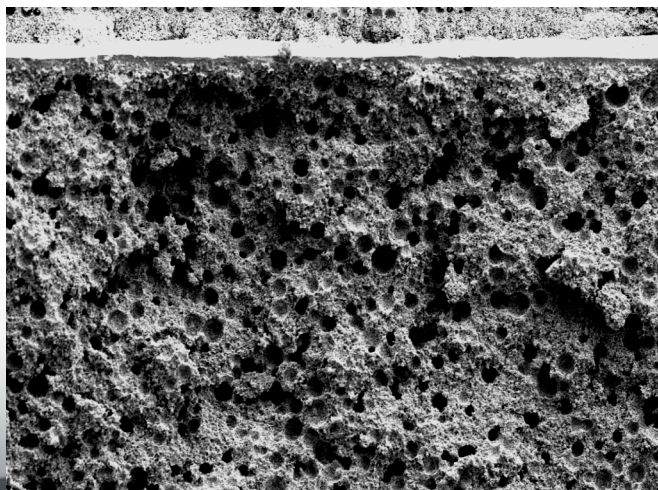
Cell before testing



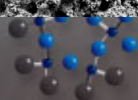
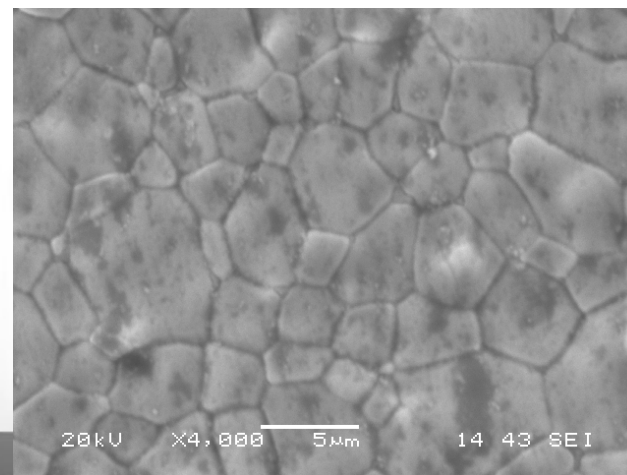
Dense thin film (~15 μm) BCY+YDC
composite electrolyte

Anode: 50 wt% NiO and 50 wt% (BCY+YDC)

Cell after testing

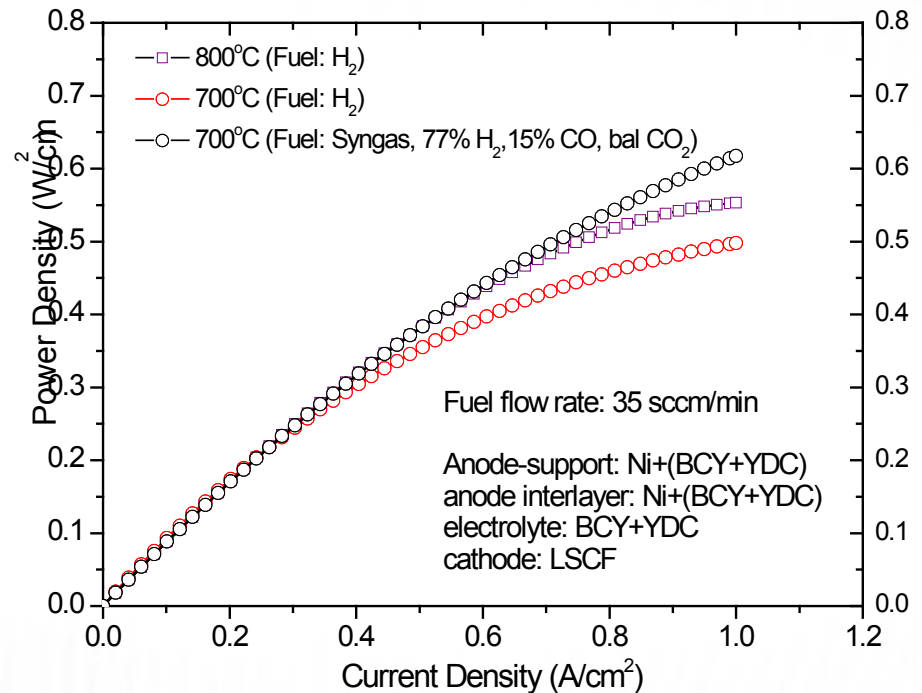
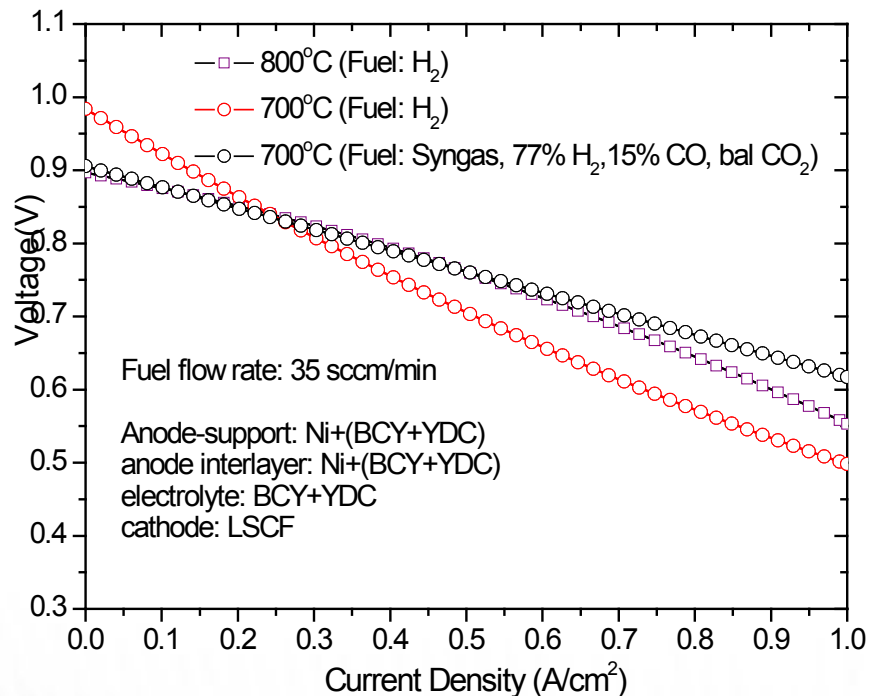


Electrolyte surface

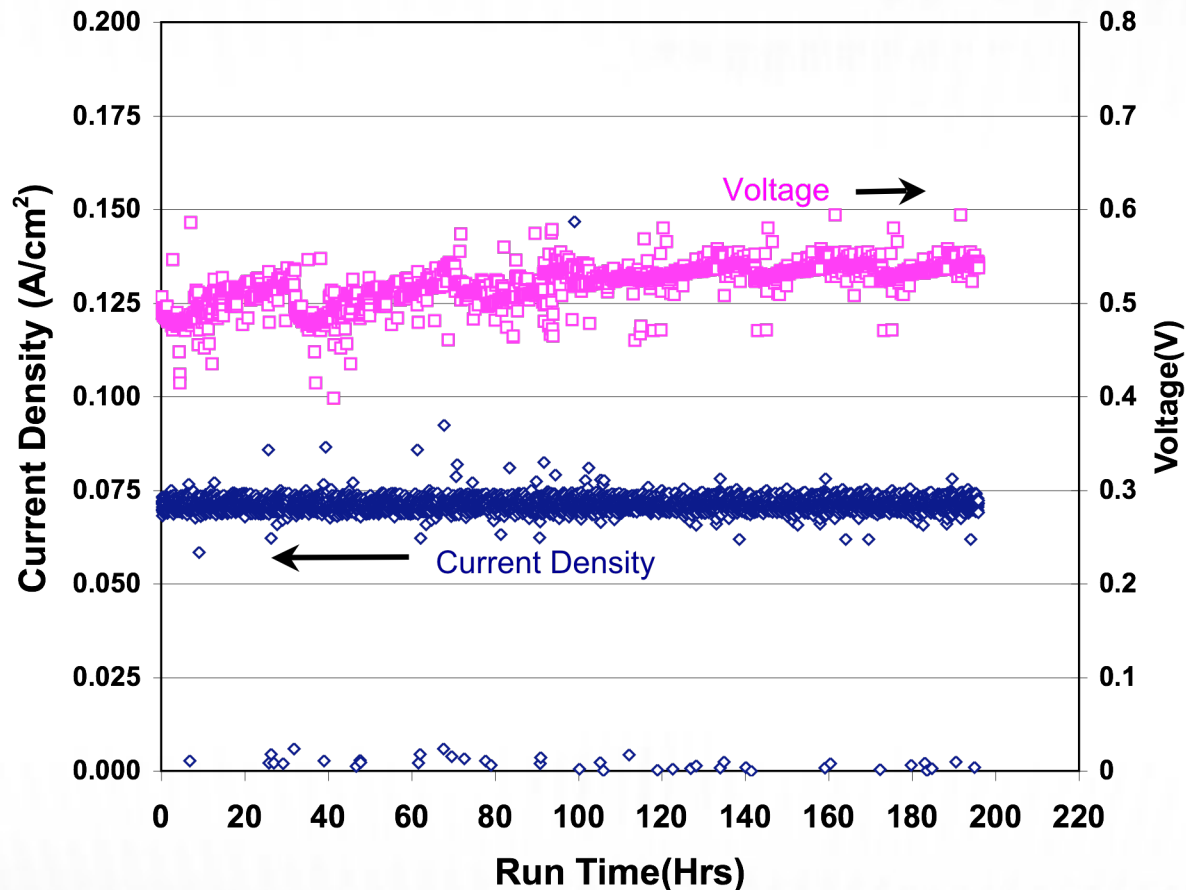


CERAMATEC
TOMORROW'S CERAMIC SYSTEMS

Anode supported P-SOFC



Stability in Syngas

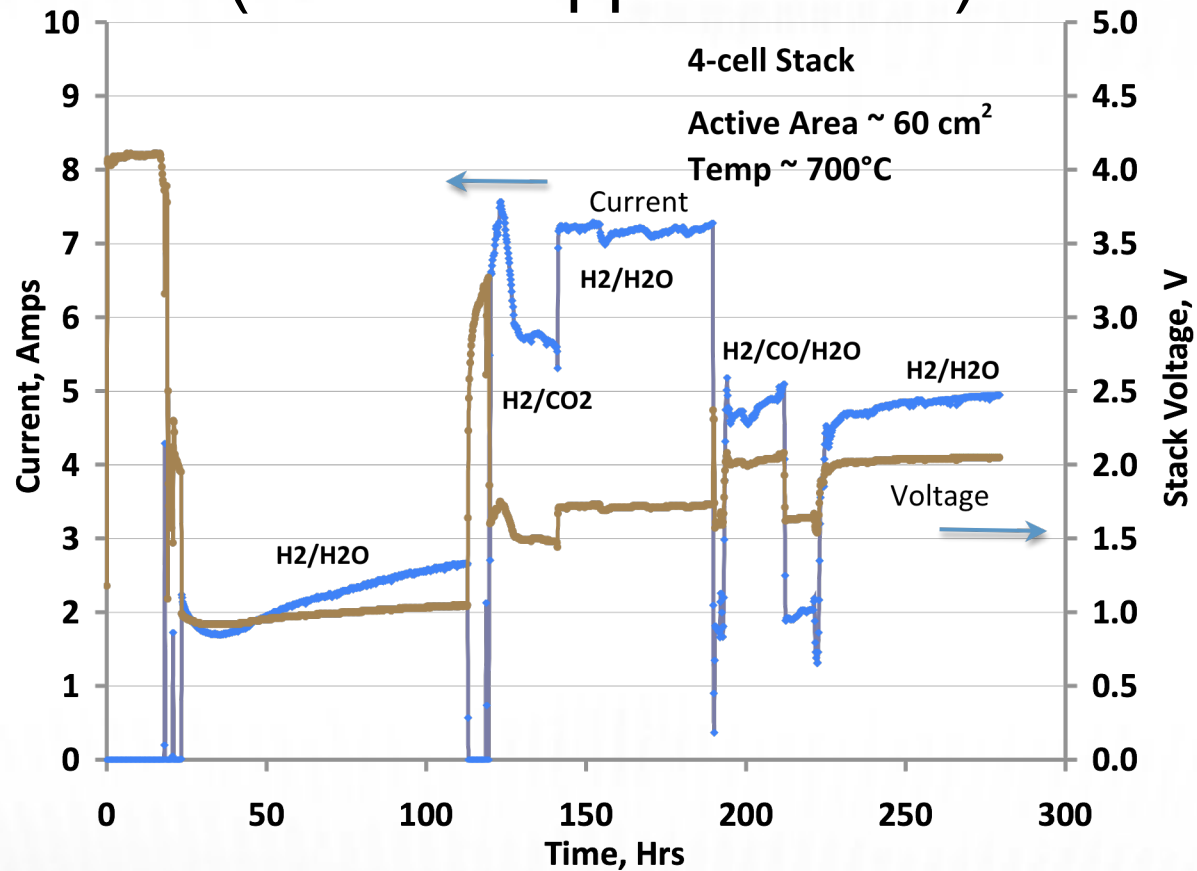


- Fuel: Simulated high utilization (90%CO₂ - balance humidified H₂)

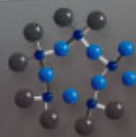


CERAMATEC®
TOMORROW'S CERAMIC SYSTEMS

Short Stack Test (Anode supported cell)



- Good stability demonstrated
- Need to improve cell fabrication process to achieve high performance



CERAMATEC
TOMORROW'S CERAMIC SYSTEMS

Conclusions

- Proton SOFC shows high efficiency possibility
- Practical compositions requires operating temperatures of 700°C or below to realize high t_H
- Thin, supported electrolyte cells demonstrated
- Chemical stability in syngas can be improved by the composite approach
- Cell fabrication process need to be improved to achieve high quality cells (no pin-holes etc.) for performance equivalent to button cells

